Supporting Information for Article

Minimal Auxiliary Basis Set Approach for the Electronic Excitation Spectra of Organic Molecules

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2D structures of the molecules in EXTEST42 and TUNE20 are shown in Fig. S1. The coordinate files for all geometries are available online.^{1,2}



Figure S1: 2D structures of molecules in EXTEST42 set. TUNE20 set is composed of molecules with red names.

The detailed results from scanning θ from 0.05 to 10 for each individual molecule in TUNE20 are shown in Fig. S2



Figure S2: The E_{ene} and E_{spe} against θ for molecules in TUNE20 with def2-SVP and the TPSSh, PBE0, BH&H-LYP, PBE and ω B97X-D density functionals. The system-averaged values are shown on the right. The vertical red line indicates the chosen global value of $\theta = 0.2$.

Table SI lists the S_1 energies of the EXTEST42 predicted by PBE0 with standard TDDFT/def2-TZVP, and the S_1 deviation of sTDDFT/def2-TZVP, TDDFT-ris/def2-TZVP and TDDFT/def2-SVP. In the case of state flipping, a star mark denotes that instead of the S_1 state, the state closest to TDDFT S_1 state is compared (the state with largest association value j(i) in Eq. (12) in main paper). Table SI also reports E_{ene} for sTDDFT and TDDFT-ris for each individual molecule in the last two columns. The last row reports the mean absolute error (MAE) of S_1 energies and E_{ene} , only including molecules where both sTDDFT and TDDFT-ris have correct S_1 state (no star mark).

Table SI: With PBE0, the *S*₁ energies of the EXTEST42 by standard TDDFT/def2-TZVP, and the *S*₁ deviation of sTDDFT, TDDFT-ris and TDDFT/def2-SVP. Star mark demotes a state flipping.

			S_1 (eV)	S	1 error (eV)		Eene	(eV)
Classes	No. Molecules	N _{atm}	TDDFT s	TDDFT 1	ГDDFT-ris de	f2-SVP	sTDDFT '	TDDFT-ris
natural compound &bioluminescent	1 firefly luciferin	26	3.82	-0.19	-0.10	0.11	0.204	0.101
	2 coumarin 153	36	3.30	-0.23	-0.06	0.07	0.234	0.054
	3 fluorescein	37	2.83	-0.19	-0.10	-0.10	0.241	0.046
	4 fluoresccein Na ³	37	2.24	-0.25	0.00	-0.07	0.252	0.033
	5 coelenterazine ⁴	53	2.72	-0.25	-0.08	0.00	0.264	0.043
	6 cypridina luciferin	57	3.52	-0.25	-0.04	0.08	0.255	0.057
	7 DFL luciferin	83	2.87	-0.26	0.00	0.08	0.269	0.052
biomolecule	8 Cytosine	13	4.74	-0.06	-0.12	0.06	0.285	0.073
	9 Adenine	15	5.02	-0.06	-0.11	-0.02	0.224	0.130
	10 Tyrosine	24	4.95	-0.13	-0.15	0.10	0.249	0.048
	11 Uracil	29	4.78	-0.08	-0.12	-0.01	0.248	0.066
	12 Guanine	32	4.87	-0.09	-0.15	0.10	0.258	0.085
	13 Ser-Tryp	38	4.56	-0.18	-0.09	0.11	0.267	0.063
	14 retinal	49	2.93	-0.21	-0.06	0.05	0.229	0.047
	15 phospholipid	52	5.55	0.11*	-0.15	-0.01	0.366	0.047
	16 provitamin D_3	72	4.32	-0.14	-0.10	0.10	0.244	0.039
	17 protoporphyrin	76	2.18	-0.32	-0.06	0.02	0.246	0.036
	18 1'-OH- γ -carotene	99	1.90	-0.22	-0.06	0.04	0.272	0.033
organic dyes &chromophore	19 [Al–MeG] ⁻⁵	28	3.97	-0.18	-0.06	0.08	0.263	0.054
	20 SUSU Me ⁶	28	4.23	-0.18	-0.11	0.10	0.209	0.086
	21 DAPI	36	3.54	-0.17	-0.05	0.08	0.246	0.059
	22 Rpet ^o	37	3.05	-0.13	-0.13	0.10	0.209	0.068
	23 tri-BrPh ⁹	42	4.19	-0.20	-0.07	0.06	0.232	0.041
	24 PyrImid TMIO5 ¹⁰	43	3.47	-0.20	-0.07	0.08	0.239	0.050
	25 organoboron6 ¹¹	44	3.11	-0.16	-0.08	0.02	0.246	0.049
	$26 \text{ azobenzene } 3a^{12}$	46	2.83	-0.27	-0.03	0.02	0.224	0.085
	27 DBF C5 ¹³	47	2.74	-0.20	-0.03	0.02	0.246	0.030
	28 triphenylamino 5 ¹⁴	49	2.85	-0.22	-0.05	0.07	0.226	0.072
	29 DPP1 15	55	2.06	-0.27	-0.03	0.04	0.240	0.040
	30 PP1 10	69 72	4.35	-0.16	-0.10	0.13	0.233	0.079
	31 Betaine 30 22 pc (WG2) 17	72	1.57	-0.29	-0.04	0.03	0.275	0.022
	32 BF ₂ (WS3) ¹⁷	92	1.84	-0.30	-0.03	0.02	0.258	0.035
drug	33 azidopy ¹⁸	14	3.44	-0.17	-0.07	0.02	0.211	0.073
	34 gout7g^{19}	35	3.68	-0.33	-0.00*	0.05	0.251	0.092
	35 chloroquine	48	3.88	-0.21	-0.08	0.09	0.232	0.073
	36 merocyanine	70	2.95	-0.17	-0.07	0.07	0.267	0.048
other systems from literature	37 naphthalene ²⁰	18	4.37	0.05	-0.04*	0.09	0.180	0.068
	38 octatetraene ²⁰	18	3.98	0.01	0.08	0.09	0.207	0.066
	39 diphenylbutadiene	30	3.46	0.08	0.04	0.07	0.223	0.042
	40 indigo ²⁰	30	1.99	0.19	0.02	0.07	0.201	0.050
	41 perylimid	40	2.05	0.22	0.00	0.06	0.236	0.038
	42 (N-cbz)-anthracene ²⁰	44	2.88	0.23	0.00	0.02	0.224	0.049
	MAE (eV)			0.188	0.068	0.063	0.242	0.058

Fig. S3 shows the two-dimensional scans for naphthalene with PBE0/def2-SVP against auxiliary basis exponent of C and H.



Figure S3: Two-dimensional scan of E_{ene} as a function of α_C and α_H for naphthalene for the PBE0 functional and def2-SVP basis set. Errors are in eV.



UV absorption spectra for each individual molecule in EXTEST42 with the PBE0 density functional and the def2-TZVP basis set are shown in S4.

Figure S4: The UV spectra simulated by TDDFT(black solid line), TDDFT-ris (red solid line) and sTDDFT (blue dashed line) for EXTEST42 with PBE0/def2-TZVP. Spectra were plotted using Lorentzian broadening with FWHM = 0.1 eV.

References

- Zhou, Z. Escfrisprep: a plugin for the TDDFT-ris and TDDFT-ris+p method in TURBO-MOLE, Github. 2022; https://github.com/John-zzh/TDDFT-ris.
- (2) Zhou, Z.; Della Sala, F.; Parker, S. M. Minimal auxiliary basis set approach for the electronic excitation spectra of organic molecules, OSF, Dataset. 2022; doi: 10.17605/OSF.IO/X5BSV.
- (3) Barbero, N.; Barni, E.; Barolo, C.; Quagliotto, P.; Viscardi, G.; Napione, L.; Pavan, S.; Bussolino, F. A study of the interaction between fluorescein sodium salt and bovine serum albumin by steady-state fluorescence. *Dyes Pigm.* 2009, *80*, 307–313.
- (4) Nishihara, R.; Suzuki, H.; Hoshino, E.; Suganuma, S.; Sato, M.; Saitoh, T.; Nishiyama, S.; Iwasawa, N.; Citterio, D.; Suzuki, K. Bioluminescent coelenterazine derivatives with imidazopyrazinone C-6 extended substitution. *Chem. Commun.* 2014, *51*, 391–394.
- (5) Zhang, L.; Liu, R.; Gung, B. W.; Tindall, S.; Gonzalez, J. M.; Halvorson, J. J.; Hagerman, A. E. Polyphenol–Aluminum complex formation: Implications for aluminum tolerance in plants. *J. Agric. Food Chem.* **2016**, *64*, 3025–3033.
- (6) Wenska, G.; Taras-Goslinska, K.; Filipiak, P.; Hug, G. L.; Marciniak, B. Photochemical reactions of 4-Thiouridine disulfide and 4-Benzylthiouridine—the involvement of the 4-Pyrimidinylthiyl radical. *Photochem. Photobiol. Sci.* 2008, 7, 250–256.
- (7) Kapuscinski, J. DAPI: A DNA-Specific fluorescent probe. *Biotech. Histochem.* 1995, 70, 220–233.
- (8) Clarke, T. M.; Gordon, K. C.; Kwok, W. M.; Phillips, D. L.; Officer, D. L. Tuning from π,Π^* to charge-transfer excited states in styryl-substituted terthiophenes: An ultrafast and steady-state emission study. *J. Phys. Chem. A* **2006**, *110*, 7696–7702.
- (9) Iyer, V. S.; Wehmeier, M.; Brand, J. D.; Keegstra, M. A.; Müllen, K. From hexa-perihexabenzocoronene to "Superacenes". *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1604–1607.

- (10) Chalmers, B. A.; Saha, S.; Nguyen, T.; McMurtrie, J.; Sigurdsson, S. T.; Bottle, S. E.; Masters, K.-S. TMIO-PyrImid hybrids are profluorescent, site-directed spin labels for nucleic acids. *Org. Lett.* **2014**, *16*, 5528–5531.
- (11) Chan-Navarro, R.; Jiménez-Pérez, V. M.; Muñoz-Flores, B. M.; Dias, H. V. R.; Moggio, I.; Arias, E.; Ramos-Ortíz, G.; Santillan, R.; García, C.; Ochoa, M. E. et al. Luminescent organoboron compounds derived from salicylidenebenzohydrazide: Synthesis, characterization, structure, and photophysical properties. *Dyes Pigm.* **2013**, *99*, 1036–1043.
- (12) Yoshino, J.; Kano, N.; Kawashima, T. Synthesis of the most intensely fluorescent azobenzene by utilizing the B–N interaction. *Chem. Commun.* 2007, *0*, 559–561.
- (13) Shang, X. S.; Li, D. Y.; Li, N. T.; Liu, P. N. A concise synthesis of tunable fluorescent 1,3-Dihydroisobenzofuran derivatives as new fluorophores. *Dyes Pigm.* 2015, *114*, 8–17.
- (14) Tao, T.; Ma, B.-B.; Peng, Y.-X.; Wang, X.-X.; Huang, W.; You, X.-Z. Asymmetrical/Symmetrical D-π–A/D-π–D thiazole-containing aromatic heterocyclic fluorescent compounds having the same triphenylamino chromophores. J. Org. Chem. 2013, 78, 8669–8679.
- (15) Warnan, J.; Favereau, L.; Pellegrin, Y.; Blart, E.; Jacquemin, D.; Odobel, F. A compact diketopyrrolopyrrole dye as efficient sensitizer in titanium dioxide dye-sensitized solar cells. *J. Photochem. Photobiol. A* 2011, 226, 9–15.
- (16) Kabe, R.; Adachi, C. Organic long persistent luminescence. Nature 2017, 550, 384–387.
- (17) Jimenez, J. C.; Zhou, Z.; Rheingold, A. L.; Parker, S. M.; Sauvé, G. Tuning the properties of azadipyrromethene-based near-infrared dyes using intramolecular BO chelation and peripheral substitutions. *Inorg. Chem.* **2021**, *60*, 13320–13331.
- (18) Hostetler, K. J.; Crabtree, K. N.; Poole, J. S. The photochemistry of 4-Azidopyridine-1-Oxide. J. Org. Chem. 2006, 71, 9023–9029.

- (19) Huang, J.; Zhou, Z.; Zhou, M.; Miao, M.; Li, H.; Hu, Q. Development of benzoxazole deoxybenzoin oxime and acyloxylamine derivatives targeting innate immune sensors and xanthine oxidase for treatment of gout. *Bioorg. Med. Chem.* **2018**, *26*, 1653–1664.
- (20) Grimme, S. A simplified tamm-dancoff density functional approach for the electronic excitation spectra of very large molecules. *J. Chem. Phys.* **2013**, *138*, 244104.