

A data analytics approach for rational design of nanomedicines with programmable drug release

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3 ABSTRACT
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7 Drug delivery vehicles can improve the functional efficacy of existing antimicrobial therapies by
8 improving biodistribution and targeting. A critical property of such nanomedicine formulations is
9 their ability to control the release kinetics of their payloads. The combination of (and interactions
10 between) polymer, drug, and nanoparticle properties gives rise to nonlinear behavioral
11 relationships and a large data space. These factors complicate both first-principles modeling and
12 screening of nanomedicine formulations. Predictive analytics may offer a more efficient
13 approach toward rational design of nanomedicines by identifying key descriptors and correlating
14 them to nanoparticle release behavior. In this work, antibiotic release kinetics data were
15 generated from polyanhydride nanoparticle formulations with varying copolymer compositions,
16 encapsulated drug type, and drug loading. Four antibiotics, doxycycline, rifampicin,
17 chloramphenicol, and pyrazinamide, were used. Linear manifold learning methods were used to
18 relate drug release properties with polymer, drug, and nanoparticle properties, and key
19 descriptors were identified that are highly correlated with release properties. However, these
20 linear methods could not predict release behavior. Non-linear multivariate modeling based on
21 graph theory was then used to deconvolute the governing relationships between these properties,
22 and predictive models were generated to rapidly screen lead nanomedicine formulations with
23 desirable release properties with minimal nanoparticle characterization. Release kinetics
24 predictions of two drugs containing atoms not included in the model showed good agreement
25 with experimental results, validating the model and indicating its potential to virtually explore
26 new polymer and drug pairs not included in training data set. The models were shown to be
27 robust after inclusion of these new formulations in that the new inclusions did not significantly
28 change model regression. This approach provides the first steps towards development of a

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3 framework that can be used to rationally design nanomedicine formulations by selecting the
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5 appropriate carrier for a drug payload to program desirable release kinetics.
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8 9 1. INTRODUCTION

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11 Intracellular bacterial infections are challenging to treat using traditional antimicrobial
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13 therapies due to the difficulty in achieving high enough local drug concentration for
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15 antimicrobial activity without inducing host cell toxicity.¹ Elimination of soluble drugs through
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17 host metabolism and excretion pathways act to reduce bioavailable amounts of antimicrobials
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19 requiring repeated dosing to maintain therapeutic concentrations to mitigate the development of
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21 antibiotic resistant in pathogens.^{2,3} Drug delivery vehicles can improve the efficacy and potency
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23 of antimicrobials by altering the drug biodistribution with improved intracellular localization and
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25 delivery of cargo to the pathogen's intracellular niche within host cells.^{4,5} Biodegradable
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27 polyanhydride nanoparticles show passive targeting and payload stabilization properties that
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29 make them uniquely suited for antibiotic delivery for intracellular infections.^{4,6} These particles
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31 are internalized efficiently by phagocytic cells using multiple mechanisms, and have been used
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33 to deliver antibiotics to kill intracellular *Brucella abortus*.⁷ Additionally, polyanhydride
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35 nanoparticles mediated efficient killing of filarial parasites by co-delivering an antiparasitic with
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37 an antibiotic targeting an intracellular endosymbiotic bacterium that supports parasite health and
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39 reproduction.⁸
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46 A key feature of the effectiveness of these nanomedicine formulations is their ability to control
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48 payload release rate, however rationally designing nanomedicines with programmable release
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50 remains elusive. Release kinetics are influenced by drug distribution within a device and/or a
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52 particle, which is in turn influenced by polymer-drug thermodynamic interactions.⁹⁻¹¹ These
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54 interactions give rise to nonlinear release behavior which is difficult to predict *a priori*.
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3 Screening nanomedicine formulations is challenging as polymer and nanoparticle properties
4 (e.g., polymer chemistry, nanoparticle size, polydispersity, release kinetics, and encapsulation
5 efficiency) yield a large number of additional variables beyond drug-specific properties. This
6 large dataspace, coupled with the multiple length scales at play, poses difficulties for
7 generalizing conclusions to other nanoparticle systems and impedes first principles modeling of
8 nanoparticle behavior.^{12,13} Hierarchical modeling may be a more efficient approach for such
9 systems, wherein key descriptors are identified and correlated to performance parameters.

19 Informatics methods encompass several tools for such hierarchical modeling. Data mining
20 techniques can deconvolute complex behavior, unraveling relationships that lie on non-Euclidian
21 surfaces,¹⁴ which enables pattern recognition and prediction through the development of
22 quantitative structure-property relationships (QSPRs).¹⁵ To this end, previous informatics
23 analyses from our laboratories has enabled identification of polyanhydride chemistry and
24 structural factors that influence protein release from films¹⁶ and enable pathogen-mimicking
25 nanoparticle processing by immune cells.^{17–20}

35 The focus of this work was to develop an informatics-based framework that determines how
36 polymer, drug, and nanoparticle characteristics influence drug encapsulation efficiency and
37 release kinetics. We sought to generate predictive models that can virtually test potential new
38 polymer and drug combinations for desirable release kinetics. Our long-term goal is to develop a
39 predictive analytics framework to enable rational design of nanomedicine formulations for
40 different types of therapeutic and prophylactic applications.

49 2. MATERIALS AND METHODS

51 **2.1. Materials.** Sebacic acid (SA) was purchased from Sigma Aldrich (St. Louis, MO).
52 Triethylene glycol, 4-*p*-hydroxybenzoic acid, 1- methyl-2-pyrrolidinone, and 1,6-dibromohexane
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3 were purchased from Sigma Aldrich for 1,8-bis(p-carboxyphenoxy)-3,6-dioxaoctane (CPTEG)
4 and 1,6-bis(p-carboxyphenoxy)hexane (CPH) monomer synthesis. Potassium carbonate,
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6 dimethyl formamide, toluene, acetonitrile, acetic acid, sulfuric acid, N,N-dimethylacetamide, and
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8 acetic anhydride were purchased from Fisher Scientific (Fairlawn, NJ) for monomer and polymer
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10 synthesis. 4-*p*-fluorobenzonitrile was purchased from Apollo Scientific (Cheshire, UK) for use in
11
12 monomer synthesis. Methylene chloride, pentane, and hexanes were purchased from Fisher
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14 Scientific for polymer purification and nanoparticle synthesis. Doxycycline (DOX), rifampicin
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16 (RIF), and pyrazinamide (PZA) were purchased from Sigma Aldrich, and chloramphenicol
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18 (CAM) was purchased from Fisher Scientific. Meropenem (MEM) was purchased from Ark
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20 Pharm, Inc. (Arlington Heights, IL) and ceftazidime (CAZ) was purchased from Acros Organics
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22 (NJ). ¹H NMR analysis used deuterated chloroform purchased from Cambridge Isotope
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24 Laboratories (Andover, MA). Drug quantification used UV-transparent microplates from Greiner
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26 Bio-One (Kremsmünster, Austria), HPLC grade acetonitrile, methanol, and tetrahydrofuran from
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28 Fisher Scientific, and phosphoric acid from Sigma Aldrich.

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31 **2.2. Polymer and nanoparticle synthesis.** CPTEG and CPH diacid were synthesized as
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33 described previously.^{10,21,22} CPTEG:CPH and CPH:SA copolymers were synthesized as
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35 described previously.^{10,21} Briefly, monomers were weighed in appropriate molar ratios and added
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37 to a round bottom flask. The monomers were acetylated in excess acetic acid at 125°C for 30
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39 min, and rotary evaporation was used to remove excess solvent from the resulting prepolymer.
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41 CPTEG:CPH was reacted for six hours at 140°C at <0.1 Torr and CPH:SA was reacted for 30
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43 min at 180°C at <0.3 Torr. Polymers were purified by precipitation in chilled hexanes.
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52 Copolymer composition and number average molecular weight (M_n) was confirmed from ¹H
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3 NMR spectra acquired on a Varian MR-400 (Varian, Inc. Palo Alto, CA) and thermal properties
4 of the copolymers was characterized by DSC (Q2000, TA Instruments, New Castle, DE).
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7 Antibiotic-loaded nanoparticles were synthesized as described previously.^{7,8} Polymer and drug
8 were weighed in separate scintillation vials at appropriate %w/w ratios. Enough methylene
9 chloride to dissolve the polymer at 20 mg/mL was added to the drug vial to dissolve/disperse the
10 drug, then transferred to the polymer vial. The combined drug and polymer solution was poured
11 into a pentane antisolvent bath at room temperature (CPH:SA) or -10°C (CPTEG:CPH) at a
12 solvent:anti-solvent ratio of 1:250 and nanoparticles were recovered by vacuum filtration.
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14 CPTEG:CPH nanoprecipitation was carried out in a cold room at 4°C. A total of 68
15 nanoformulations were tested, spanning drug loadings between 1% and 20% loading (% w/w).
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17 All drugs were tested in 20:80 CPH:SA and 20:80 CPTEG:CPH nanoparticles, and rifampicin
18 was additionally tested in 10:90, 30:70, and 50:50 CPTEG:CPH. ¹H NMR spectra of empty
19 20:80 CPTEG:CPH and 20:80 CPH:SA nanoparticles indicated undetectable amounts of
20 methylene chloride and trace amounts of pentane (data not shown).
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35 To validate the informatics analysis, nanoparticles encapsulating meropenem or ceftazidime
36 were synthesized using a high-throughput method adapted from Goodman et al.²³ Briefly,
37 polymer and drugs were dissolved/dispersed in methylene chloride and dispensed via high-
38 throughput robot into 10 mL borosilicate tubes at a final polymer concentration of 20 mg/mL.
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40 The robot sonicated and dispensed the combined polymer and drug solution into 50 mL conical
41 polypropylene tubes containing 45 mL pentane (1:18 solvent:anti-solvent ratio) at the
42 temperatures listed above. Multiple particle batches were pooled and recovered by vacuum
43 filtration. Scanning electron microscopy (SEM, FEI Quanta 250, Hillsboro, OR) was used to
44 image all nanoparticles, and size distributions were calculated using Fiji image analysis
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3 software²⁴ and the ParticleSizer plugin script for Fiji. Nanoparticle zeta potential was measured
4 using a Zetasizer Nano (Malvern Instruments Ltd., Worcester, UK).
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7 **2.3. Drug release kinetics.** Nanoparticles (9-11 mg) were dispersed in 0.5 mL PBS, pH 7.4 and
8 suspended by sonication (VCX 130 PB, Sonics & Materials, Inc., Newtown, CT). At each time
9 point, the nanoparticles were pelleted by centrifugation and supernatant was collected for drug
10 quantification. Fresh PBS was added to maintain perfect sink conditions and the nanoparticles
11 were dispersed by sonication. At the end of the release experiment, 40 mM sodium hydroxide
12 was added to accelerate polymer degradation and extract the remaining encapsulated drug as
13 described previously.²⁵
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23 The drug mass released at each time point was determined by spectrophotometry (SpectraMax
24 M3, Molecular Devices, San Jose, CA) and UV-HPLC (1200 series, Agilent Technologies, Santa
25 Clara, CA). Doxycycline, rifampicin, and chloramphenicol were quantified by absorbance in
26 UV-transparent 96-well plates at 350, 333, and 293 nm, respectively. Pyrazinamide release and
27 base extraction samples were separated using a Phenomenex Kinetex 2.6-micron C18 100Å
28 100x4.6mm column and a 30:5:65 acetonitrile:methanol:water mobile phase adjusted to pH 5.2
29 with phosphoric acid.²⁶ The flow rate was 0.6 mL/min and pyrazinamide was quantified at 268
30 nm. Meropenem and ceftazidime release and base extraction samples were separated using a
31 Zorbax Eclipse XDB-C8 5-micron 4.6x150 mm column, monitoring at 299 nm and 246 nm,
32 respectively. Meropenem release samples used a mobile phase gradient ramping from 0.1/99.9
33 (%v/v) methanol/water to 50/50 over 15 min. Meropenem base extraction samples used a
34 gradient ramping from 0.1/99.9 acetonitrile 0.1% trifluoroacetic acid/ water 0.1% trifluoroacetic
35 acid to 50/50 over 15 min. Ceftazidime release samples used a mobile phase protocol with an
36 isocratic step at 0.1/99.9 methanol/water for 5 minutes followed by a gradient ramping to 50/50
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3 over 10 min. Ceftazidime base extraction samples used an isocratic step at 15/85 from 0.1/99.9
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5 acetonitrile 0.1% trifluoroacetic acid/ water 0.1% trifluoroacetic acid for 1 min followed by a
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7 gradient ramping to 40/60 over 5 min. All meropenem and ceftazidime HPLC protocols used a
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9 flow rate of 1 mL/min.

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12 The small mass of drug and large volume of antisolvent used in nanoparticle synthesis render the
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14 non-encapsulated drug concentration below the limit of detection of the analytical methods used
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16 in this study. Therefore encapsulation efficiency (EE) was calculated from the cumulative sum of
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18 detected drug mass released in PBS and base extraction samples using Eq. 1.²⁵ In a minority of
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20 formulations >100% EE was observed, which could arise from the presence of drug
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22 nanocrystals²⁷ (which was not detected on nanoparticle surfaces by SEM), gravimetric
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24 inaccuracies due to static charge of the nanoparticles, or residual error in the drug concentration
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26 quantification assays. Drug release kinetics are presented as fraction released, where the
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28 cumulative drug mass release is normalized by the total encapsulated drug mass. Prism 7
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30 (Graphpad Software, La Jolla, CA) was used to generate release kinetics figures.

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33 Eq. 1)

$$34 \quad \frac{EE}{100\%} = \frac{\text{cumulative drug mass from release \& base extraction}}{\text{nanoparticle mass} * \text{drug loading fraction} \left(\frac{wt}{wt} \right)}$$

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43 **2.4. Informatics analysis.** Release behavior parameters, along with polymer, drug, and
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45 nanoparticle properties, were normalized and mean-centered. Three different informatics
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47 approaches were integrated and applied to analyze the data in this work. Linear manifold
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49 learning approaches, such as principal component analysis (PCA),²⁸⁻³⁰ permit us to identify the
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51 right projection of data from which meaningful features associated with the input data can be
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53 identified. PCA performs an eigenvector decomposition and defines a new set of linear
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3 combinations of descriptors which maximize the amount of unique information in a minimal set
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5 of orthogonal axes, termed principal components (PCs). The original data is decomposed into
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7 two matrices of interest for this work: the scores and loadings. The scores describe the different
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9 conditions (i.e., nanoparticle and drug chemistry), while the loadings describe the different
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11 descriptors and properties. The interpretation of these matrices is provided here with the relevant
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13 results, and an additional term called the variable importance projection (VIP) is calculated from
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15 the loadings matrix using Eq. 2.
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20 Eq. 2)

$$21 \quad 22 \quad 23 \quad 24 \quad 25 \quad 26 \quad 27 \quad 28 \quad 29 \quad 30 \quad 31 \quad 32 \quad 33 \quad 34 \quad 35 \quad 36 \quad 37 \quad 38 \quad 39 \quad 40 \quad 41 \quad 42 \quad 43 \quad 44 \quad 45 \quad 46 \quad 47 \quad 48 \quad 49 \quad 50 \quad 51 \quad 52 \quad 53 \quad 54 \quad 55 \quad 56 \quad 57 \quad 58 \quad 59 \quad 60$$
$$VIP = \frac{PC_x^T * PC_x^i}{\sum PC_x^T * PC_x^i}$$

In this case, $x = 5$ because 5 PCs captured >90% of the variance in the data. The analysis was performed for $T =$ encapsulation efficiency, drug released @2 h, and fraction released/day (Figure 6). Partial Least Squares (PLS) is a multi-linear regression approach which accounts for co-linearity in the data, and therefore limits bias and develops more robust quantitative relationships.³¹⁻³⁴ PLS performs separate PC analyses on the predictor variables (i.e., descriptors) and the predicted variables (i.e., properties). These therefore represent linear manifold learning approaches which provide qualitative and quantitative design relationships.

In order to model the drug release properties accurately and robustly, we found that non-linearity needed to be accounted for in the modeling. Therefore, we first developed non-linear parameterization of the data through non-linear manifold learning, based on graph theory, using the Isomap algorithm.^{14,35} This approach generates a graph connecting data points on a high dimensional space to their nearest neighbors, mapped out in the high dimensional space, and then

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3 fit to a low dimensional manifold. The assumption here is that the graph Euclidean distance
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5 between the points in high dimensions closely approximates the curvilinear distances along the
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7 low dimensional manifold. Through dimensionality reduction the manifold unravels into two or
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9 three dimensions allowing it to be visualized. The result of such dimensionality reduction is a
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11 weighted graph of the original data points where the edges are weighted according to the
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13 geodesic distances. Like in PCA, we develop a set of parameters for each set of conditions,
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15 although in this case the parameters are based on a non-linear combination of descriptors.
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20 3. RESULTS

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22 **3.1. Building descriptor library.** To generate the data set, we focused on nanoparticles
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24 composed of CPTEG, CPH, and SA copolymers (Figure 1 a,b). Nanoparticles synthesized from
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26 these polyanhydride copolymers have been shown to kill intracellular bacteria because of their
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28 high internalization rates by phagocytic cells,^{7,36} localization in intracellular compartments that
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30 harbor these bacteria,^{17,36} and improved antimicrobial activity of encapsulated drugs.⁶⁻⁸ In
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32 addition to the structural descriptors defined by Li et al.,¹⁶ we included molecular weight and
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34 compositional data from ¹H NMR and thermal characterization from DSC analysis. The release
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36 kinetics of four antibiotics, doxycycline (Figure 1c), rifampicin (Figure 1d), chloramphenicol
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38 (Figure 1e), and pyrazinamide (Figure 1f), were studied. The choice of the drug library was
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40 motivated by multiple factors. These drugs were selected due to their diversity of molecular
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42 weight, chemical structure, and hydrophobicity, among other physicochemical differences. All
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44 are FDA-approved drugs and belong to separate antibiotic classes, target distinct bacterial
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46 structures, and have well-characterized pharmacokinetics/pharmacodynamics. Experimental and
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48 predicted physicochemical properties for each of these drugs were gathered from the Drugbank
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3 database.³⁷ Predicted drug properties from this database were calculated by ALGOPS and
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5 ChemAxon methodologies.
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8 These drugs were encapsulated in polyanhydride nanoparticles by flash nanoprecipitation, and
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10 zeta potential, size distributions, and polydispersity index were obtained. Release profiles and
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12 encapsulation efficiencies were obtained from *in vitro* experiments in PBS, pH 7.4 (Figure 2).
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14 Figure 2 shows representative release kinetics data for multiple drugs, selected from a total of 68
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16 nanoformulations that were tested. The formulations depicted in Figure 2 were selected to show
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18 the diversity of release behavior in the data set. CPH:SA-doxycycline nanoformulations tended
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20 to show a higher burst than CPTEG:CPH nanoformulations, and the lower loading in the
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22 CPH:SA nanoformulations tended to have a greater sustained release slope (Figure 2a). The
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24 chemistry trend was reversed in the rifampicin nanoformulations, where the CPTEG:CPH
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26 chemistries tended to show a higher burst release than the CPH:SA chemistries, and increasing
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28 the loading increased the burst (Figure 2b). For chloramphenicol, both 20:80 CPH:SA and
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30 CPTEG:CPH nanoformulations tended to generate a large burst release followed by a slow rate
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32 of drug release (Figure 2c). Pyrazinamide formulations generated a large burst from the 20:80
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34 CPH:SA nanoparticles followed by a steady rate of drug release (Figure 2d). In contrast, the
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36 20:80 CPTEG:CPH nanoparticles encapsulating pyrazinamide showed a small burst and slow
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38 rate of drug release and did not release more than 20% of the payload in one week. These results
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40 add to the body of literature^{9-11,38} that indicates that copolymer chemistry, drug type, and drug
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42 loading influence drug release kinetics from biodegradable particles and other devices.
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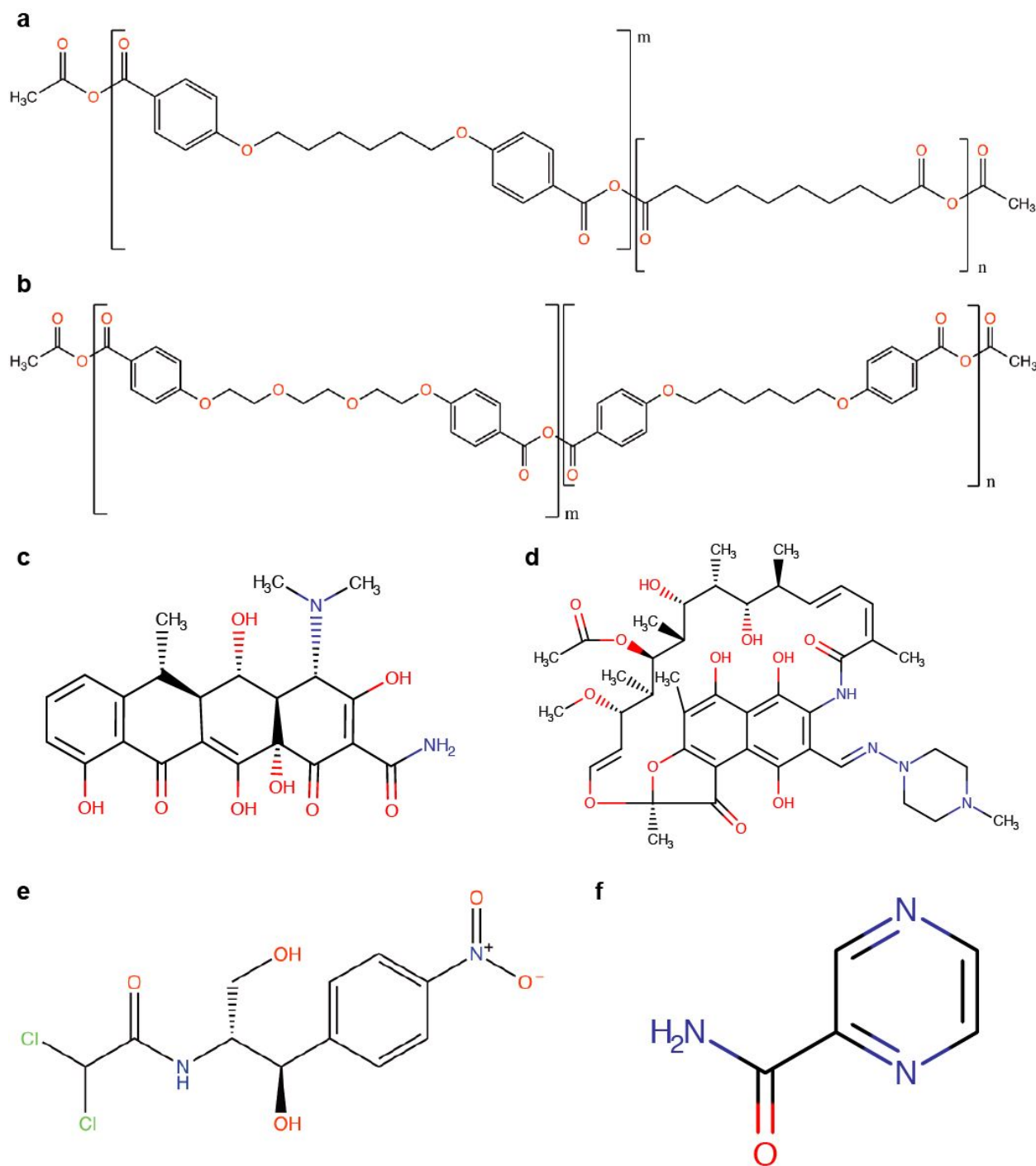


Figure 1. Polymer and antibiotic chemical structures. (a,b) Structures of CPH:SA (a) and CPTEG:CPH (b) copolymers, where m and n are the number of repeats for each unit. (c-f)

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3 Structure of doxycycline (DOX, a), rifampicin (RIF, b), chloramphenicol (CAM, c), and
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5 pyrazinamide (PZA, d).
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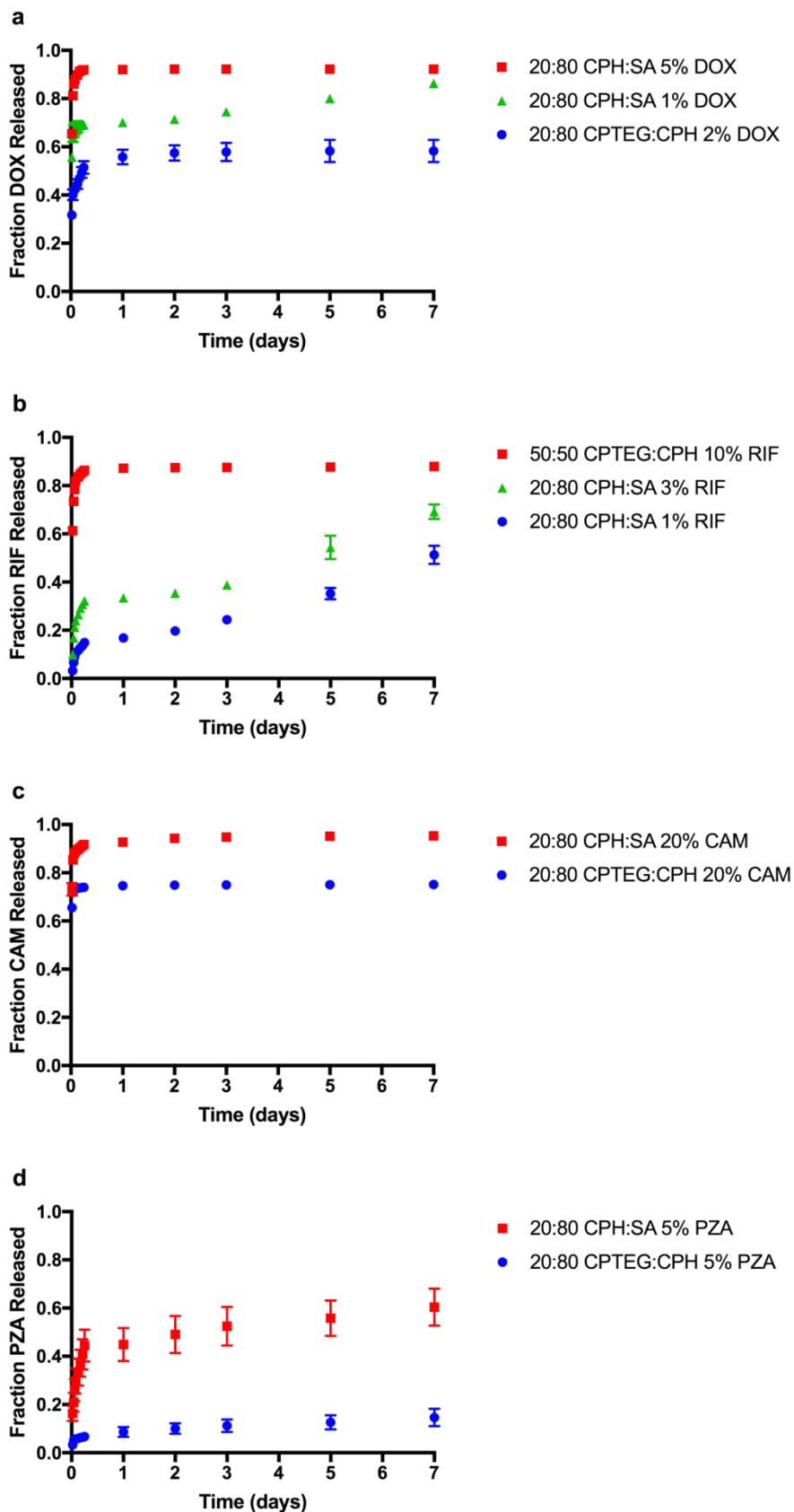


Figure 2. Representative antibiotic release kinetics from nanoparticles encapsulating doxycycline (DOX, a), rifampicin (RIF, b), chloramphenicol (CAM, c), and pyrazinamide (PZA, d). The depicted nanoformulations represent a subset of the 68 formulations tested and were selected to display the diversity of release behavior in the data set. Data are presented as mean \pm SD. Error bars are not depicted in cases where the error bar height is smaller than the symbol. Release profiles were parameterized into two-hour burst, one day burst, and 2-7 day sustained release slope.

Table 1. Representative antibiotic release properties. FR (2h) fraction released in two-hour burst, FR (24h) fraction released in 24-hour burst. Data are presented as mean \pm SD.

Nano-formulation	EE	FR (2h)	FR (24h)	FR slope (FR/day)
20:80 CPH:SA 5% DOX	64.9% \pm 5.8%	0.878 \pm 0.005	0.921 \pm 0.007	0.00003 \pm 0.00016
20:80 CPH:SA 1% DOX	159.0% \pm 9.3%	0.675 \pm 0.025	0.702 \pm 0.010	0.02963 \pm 0.97909
20:80 CPTEG:CPH 1% DOX	71.6% \pm 5.8%	0.329 \pm 0.023	0.462 \pm 0.159	0.00082 \pm 0.00014
50:50 CPTEG:CPH 10% RIF	86.7% \pm 2.6%	0.818 \pm 0.008	0.872 \pm 0.004	0.00087 \pm 0.20502
20:80 CPH:SA 3% RIF	29.8% \pm 2.4%	0.240 \pm 0.013	0.335 \pm 0.009	0.06993 \pm 0.96020
20:80 CPH:SA 1% RIF	49.3% \pm 2.8%	0.106 \pm 0.011	0.168 \pm 0.010	0.06285 \pm 0.96084
20:80 CPH:SA 20% CAM	59.8% \pm 0.4%	0.886 \pm 0.009	0.927 \pm 0.006	0.00192 \pm 0.47169
20:80 CPTEG:CPH 20% CAM	123.5% \pm 13.0%	0.766 \pm 0.020	0.782 \pm 0.018	0.00055 \pm 0.00821
20:80 CPH:SA 5% PZA	34.9% \pm 4.8%	0.298 \pm 0.053	0.449 \pm 0.069	0.02168 \pm 0.30458
20:80 CPTEG:CPH 5% PZA	89.7% \pm 36.6%	0.057 \pm 0.013	0.086 \pm 0.020	0.00879 \pm 0.33933

3.2. Identifying Factors that Influence Drug Release. The drug release profiles were parameterized using three attributes: (i) fraction released at two hours (FR (2h)); (ii) fraction released in one day (FR (24h)), both of which characterized the burst effect; and (iii) the slope of the release profile between 2 and 7 days to characterize the sustained release (Table 1). The normalized and mean-centered data are represented in the form of a heat map to provide an overview and to ensure that no outliers are biasing the results (Figure 3, Table S1-2). In this step,

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3 no data specific to particle chemistry was included so as to not bias the analysis. A clustering
4 analysis, based on Euclidian distance, was used to visualize broad trends in the data set between
5 descriptors and nanoformulations, and is represented in Figure 3 by dendrograms which define
6 the correlative indices. The clustering along the y-axis of Figure 3 can be visualized as plotting
7 each nanoformulation in multidimensional space, where each dimension is a different descriptor.
8 Encompassing n-dimensional “spheres” are defined at the locations of the nanoformulations, and
9 as the radii of the spheres increase additional nanoformulations are encompassed. The relative
10 sphere size needed to encompass multiple descriptors is comparable to the height of the branch
11 in the dendrogram. Nanoformulations or descriptors grouped lower in the dendrograms are likely
12 to show relatively strong, positive correlations. Branches higher in the dendrograms are more
13 likely to show weak, positive correlations or inverse correlations.
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29 From Figure 3, we find that the primary difference is between CPH:SA and CPTEG:CPH, given
30 that the two chemistries branch off at the lowest correlation node. Therefore, particle chemistry
31 is the key discriminator for nanoformulation behavior. Within each node, the compounds then
32 group based on drug type, and then finally branch off based on theoretical drug loading and
33 molar monomer ratios within the copolymer. This defines the order of importance on release
34 properties with CPH:SA versus CPTEG:CPH as the most important and the theoretical drug
35 loading having less importance. For CPH:SA nanoparticles, rifampicin and pyrazinamide
36 grouped together strongly, whereas doxycycline and chloramphenicol grouped together within
37 the CPTEG:CPH chemistries. The CPTEG:CPH-chloramphenicol and -pyrazinamide
38 nanoformulations clustered together, and diverged from the CPTEG:CPH-rifampicin and -
39 doxycycline nanoformulations. Considering correlations to the release properties, the fraction
40 released at two hours and 24 hours are strongly correlated with the polymer melting point (T_m)
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3 and zeta potential. The fraction released/day clustered with nanoparticle diameter and PDI
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5 polymer DOP and M_n . The relatively low branching of these properties in the dendrogram
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7 indicates moderate to strong correlation. Encapsulation efficiency (EE) was most strongly
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9 correlated with water solubility, followed by fraction released/day. This (weak) correlation to
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11 water solubility is expected, as incompatibility between polymer and drug
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13 hydrophobicity/hydrophilicity can result in drug partitioning more strongly in the antisolvent
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15 than the polymer matrix. The drug release properties (burst release, slope of release, and
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17 encapsulation efficiency) appeared relatively isolated from each other within the dendrogram,
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19 suggesting potential for independent control of these properties in designing nanoformulations.
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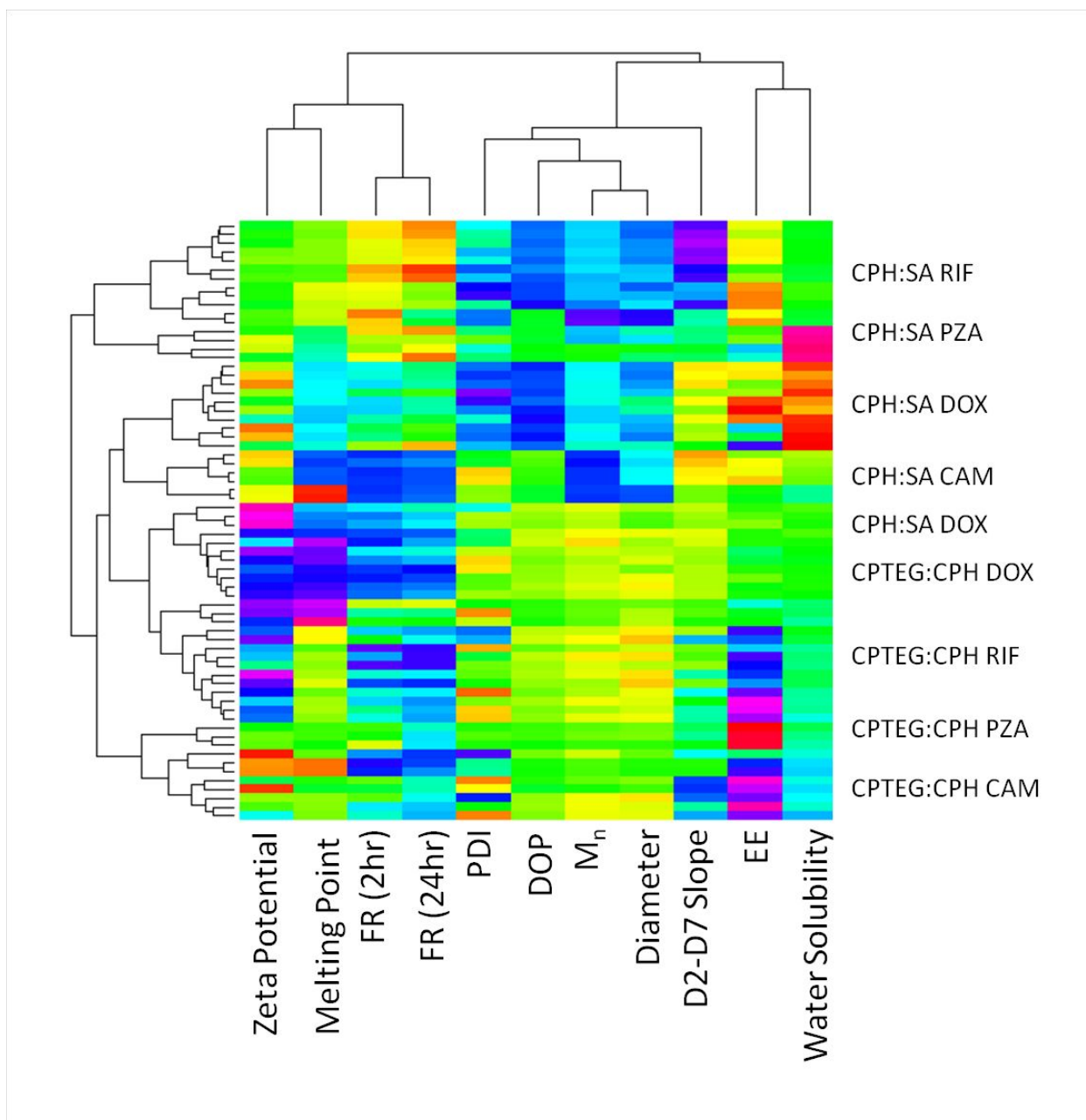


Figure 3. Representation of correlations in data using Euclidian distance-based clustering, with the dendrograms defining the degree of correlation (i.e., branches at the bottom of the dendrogram have high correlation and correlation decreases as moving along the branches). From the dendrograms, the key discriminators among nanoformulations (vertical axis) in the order of importance are carrier chemistry (CPTEG:CPH versus CPH:SA), drug type, and

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3 theoretical drug loading. Concerning correlations between drug release properties and descriptors
4 (horizontal axis), nanoparticle burst release (FR (2hr) and FR (24 hr)) was most strongly
5
6 correlated with zeta potential and drug melting point. The release slope was most strongly
7
8 correlated with the nanoparticle and polymer size properties. Encapsulation efficiency was most
9
10 strongly correlated with the water solubility of the drug. That these data fall within comparable
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12 ranges demonstrates the robustness of the method and data set, enabling interrogation of
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14 nanoformulation behavior. Drug abbreviations: doxycycline (DOX), rifampicin (RIF),
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16 chloramphenicol (CAM), and pyrazinamide (PZA). Raw and processed input data are included in
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18 Tables S1-2.
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25 A dimensionality reduction analysis, specifically principal component analysis (PCA), was then
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27 applied to the data of Figure 3, with descriptors specific to the particle and drug chemistry added
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29 to the data set (Table S3). Plots of formulation mapping and descriptor mapping within the
30
31 dimensionally-reduced space are shown in Figures 4 and 5, respectively. In these figures, the
32
33 principal components (PC) are ordered in terms of decreasing variability captured. PC1, the most
34
35 important PC, captured particle chemistry properties (41.3%); therefore, differences in particle
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37 chemistry descriptors explain more variance in release behavior than other descriptor sets. The
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39 next most important PC, PC2, captured differences in drug-specific descriptors (27.9%). The
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41 scores plot (Figure 4), which maps individual nanoparticle formulations onto these PC's (which,
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43 between them, allow us to reliably capture correlations in those two dimensions), shows a clear
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45 separation between CPTEG:CPH and CPH:SA particle chemistries. Within each polymer,
46
47 doxycycline, chloramphenicol, and pyrazinamide clustered together, whereas rifampicin
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49 formulations formed a cluster isolated from the other drugs, indicating potentially different types
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51 of interactions with the particle carriers.
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The loadings plot (Figure 5), which maps the descriptor variables onto the PC's shows that the role of the particle and the drug bank descriptors have been isolated (i.e., particle data lies along the PC1 axis and drug bank data is along the PC2 axis). Given that PC1 is the most important axis, we are capturing that the particle chemistry is the critical characteristic for predicting particle release behavior. The drug release properties do not adhere exclusively to either PC1 or PC2 axes, indicating that they are influenced by both polymer and drug characteristics. The ability to isolate the different controls allows us to assess, model, and design by the material characteristics.

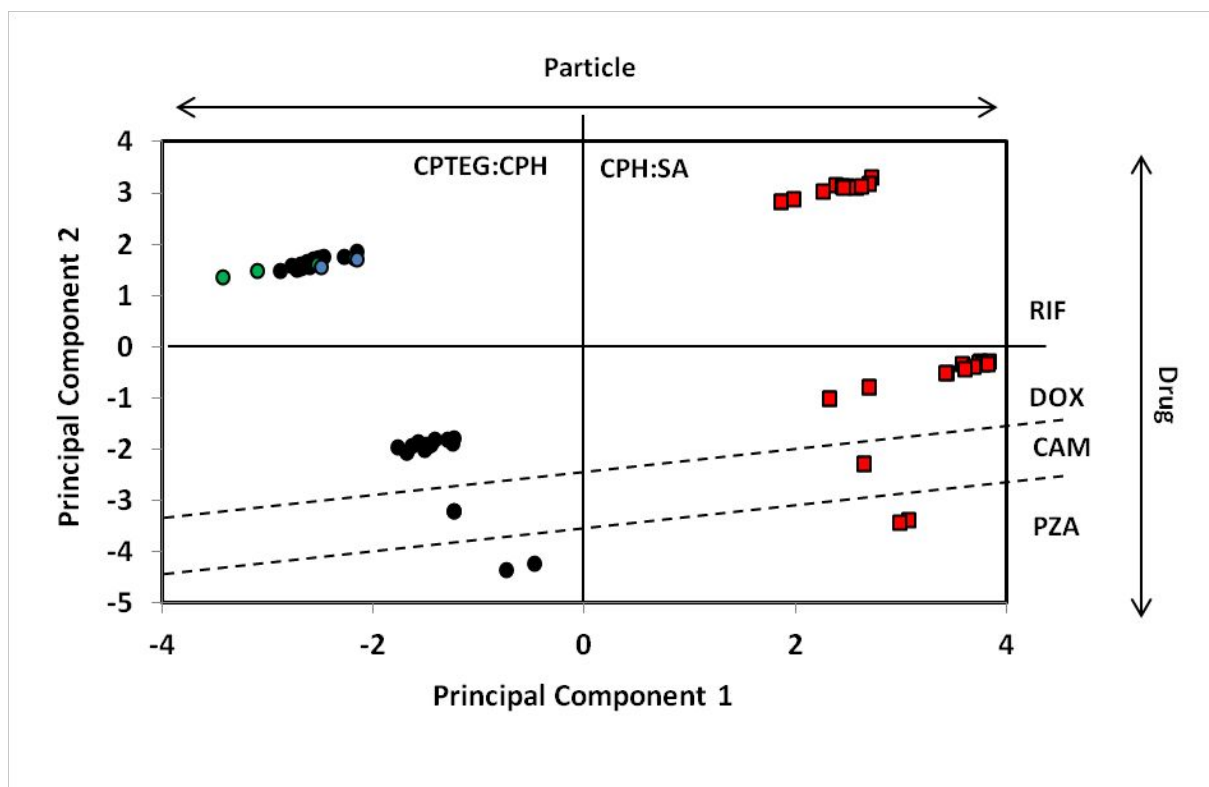
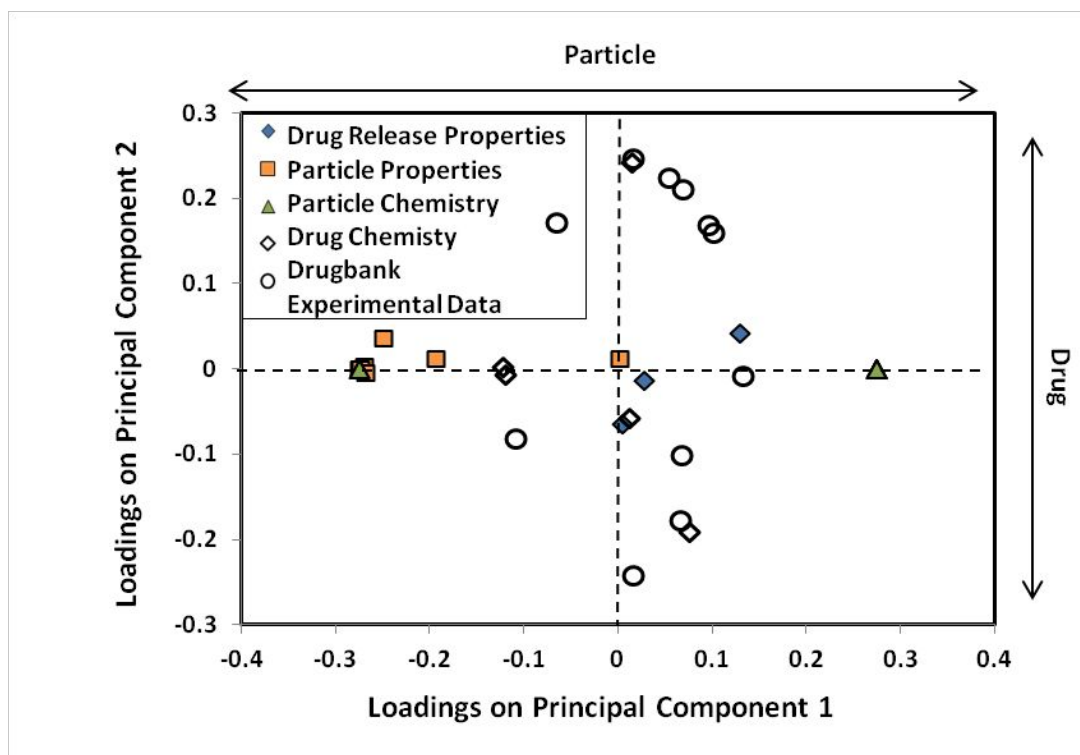


Figure 4. PCA scores plot. PC1 captures differences due to particle chemistry (10:90 CPTEG:CPH in blue, 20:80 CPTEG:CPH in black, and >30:<70 CPTEG:CPH in green) and PC2 captures the differences due to drug. There is a clear separation of formulations due to different

1
2
3 chemistries, with a demonstrated capability to isolate the effects of particle chemistry from drug
4 properties. PC1 and PC2 captured 43.1% and 27.9% of variability, respectively. Drug
5
6 abbreviations: doxycycline (DOX), rifampicin (RIF), chloramphenicol (CAM), and
7
8 pyrazinamide (PZA). Raw input data are provided in Table S3.
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39 **Figure 5.** PCA loadings plot. Particle descriptors lie along the PC1 axis and drug descriptors lie
40 along the PC2 axis. Drug release properties lie along both axes, indicating some dependence on
41 both particle and drug descriptors. PC1 and PC2 captured 43.1% and 27.9% of variability,
42
43 respectively. Raw input data are provided in Table S3.
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49 To further quantify the correlation between descriptors and release properties, we calculated the
50 VIP. In all, a total of 36 descriptors were used in the VIP analysis (as shown in Table 2),
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52 describing nanoparticle (1-3), polymer (4-13), and drug properties (14-36). The encapsulation
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3 efficiency was most strongly correlated with zeta potential (-), % Cl (drug) (+), % O (drug) (+),
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5 $T_{m,drug}$ (-), predicted water solubility (-), pKa (strongest base) (-), and drug rotatable bond count
6
7 (+). As seen in Figure 3, the two-hour burst and slope of release were highly correlated with each
8
9 other, and showed similar correlations with the descriptors. Both the two-hour burst and slope of
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11 release were most strongly correlated with % Cl (drug) (+), % O (drug) (+), water solubility (-),
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13 % N (drug) (-), pKa (strongest acid) (-), and rotatable bond count (+). The identification of
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15 several highly correlated descriptors allows for reduction of the descriptor space to a minimum
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17 number and defines the number of descriptors necessary for performing high throughput
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19 calculations. This minimization is an important objective in computational modeling to improve
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21 model robustness. The purpose of VIP analysis is to assess the descriptors that contribute
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23 significant information as well as to identify correlated descriptors. While we identify the drug-
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25 related descriptors as having the highest individual impact, the particle-related descriptors
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27 collectively contribute the largest amount of information as seen in Figure 4.
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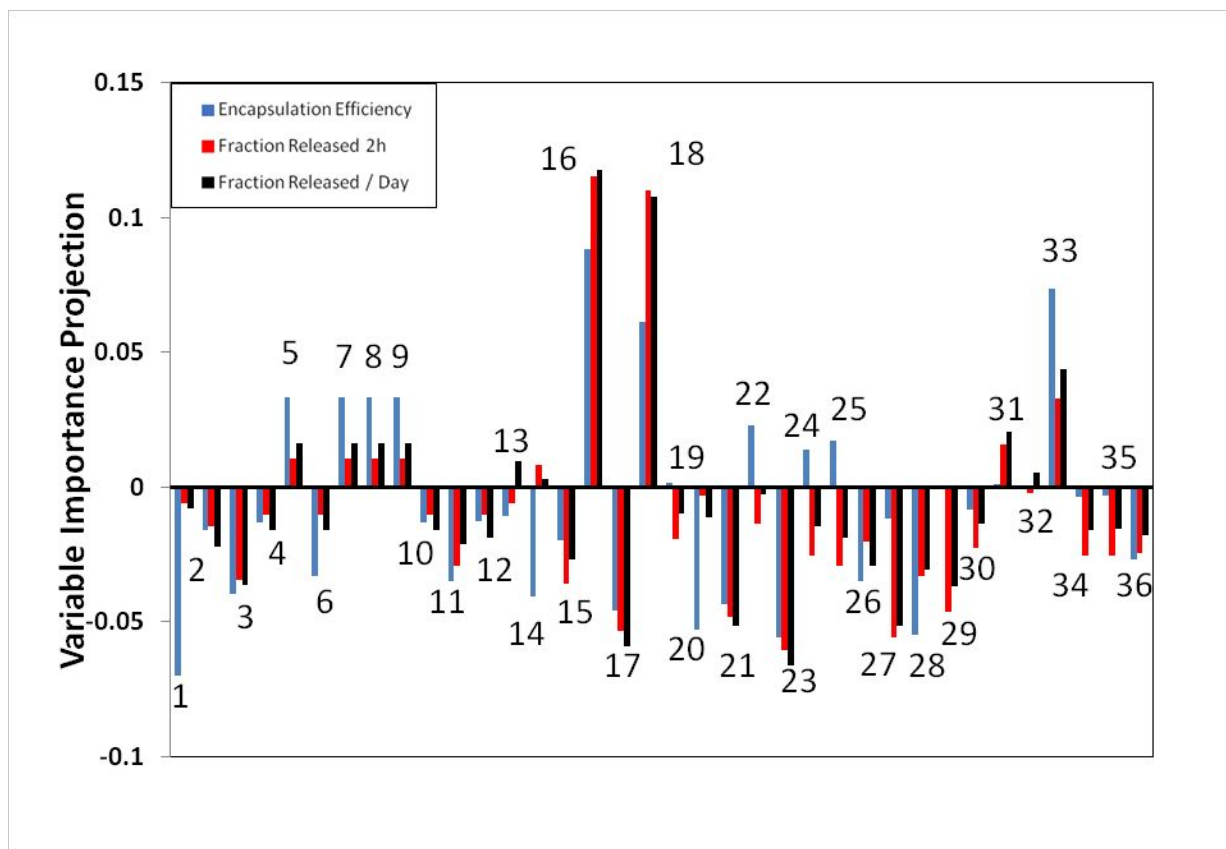


Figure 6. Variable importance projection of descriptors with respect to drug release properties.

Descriptors are listed in Table 2. Positive VIP values correspond to positive correlation, and negative values to inverse correlation.

Table 2. List of descriptors used in VIP analysis.

1	Zeta potential	13	Tg (°C)	25	logP (predicted, ChemAxon)
2	Diameter	14	% C (Drug)	26	logS (predicted, ALGOPS)
3	PDI	15	% H (Drug)	27	pKa (Strongest Acidic)
4	Water contact angle	16	% Cl (Drug)	28	pKa (Strongest Basic)
5	Backbone O	17	% N (Drug)	29	Physiological charge
6	Aliphatic C	18	% O (Drug)	30	Hydrogen acceptor count
7	Aromatic C	19	molar mass	31	Hydrogen donor count
8	% O	20	Tm (°C, Drug)	32	Polar surface area
9	% H	21	water solubility (experimental)	33	Rotatable bond count
10	% C	22	logP (experimental)	34	Refractivity
11	DOP	23	water solubility (predicted, ALGOPS)	35	Polarizability
12	Mn (Da)	24	logP (predicted, ALGOPS)	36	Number of rings

3.3. Modeling release behavior. Beyond only observing the correlation of data, we wanted to identify similarities and design pathways between the various nanoformulations. This connectivity defines samples which have the most similar behavior and can provide information on potential replacements and design. In order to accomplish this, we performed a graph theory analysis (Figure 7). For the CPH:SA particle chemistries, there is a high connectivity (illustrated by black lines) and tight clustering within individual drugs. For the CPTEG:CPH particle chemistries, chloramphenicol, doxycycline, and pyrazinamide showed a high internal connectivity, but rifampicin branched out significantly. Each drug showed some degree of connectivity between the CPH:SA and CPTEG:CPH particle chemistries and doxycycline appeared to be the most interconnected across particle chemistries.

The degree of similarity can be defined by the number of connections required to connect two points. The distance along the two-dimensional projection also indicates the similarity of formulations. Pyrazinamide and chloramphenicol generated the least similarity in release behavior, as they required 4-6 connections, and lie far from each other along the projection.

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3 Within each particle chemistry, doxycycline showed the most similarity to rifampicin and
4 pyrazinamide, and rifampicin showed the most similarity to doxycycline and chloramphenicol.
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6 The branched region of the CPTEG:CPH-rifampicin nanoformulations indicates some
7
8 dissimilarity from the other rifampicin nanoformulations and some unique behavior that will
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10 need to be explored more systematically using experiments. Of note, the rifampicin formulations
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12 with altered molar composition of CPTEG:CPH (from the 20:80 that makes up most of the data
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14 set) showed high similarity to the 20:80 CPTEG:CPH nanoformulations within the cluster. This
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16 would suggest that nano-carrier copolymer compositions can be interchanged within these
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18 rifampicin-loaded formulations without major impact.
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25 This graph theory mapping in Figure 7 yielded notably different drug clustering within each
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27 nano-carrier chemistry compared to PCA (Figure 4). Rifampicin and chloramphenicol
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29 formulations are closely related in this map, while they were distant from each other in the PCA
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31 scores plot. Strikingly, chloramphenicol and pyrazinamide are most distant in the graph theory
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33 map, while they were clustered closely in the PCA scores plot. These clustering differences are
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35 likely due to PCA's limited ability to capture non-linear relationships. Non-linear modeling
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37 techniques like graph theory are better equipped to capture the non-linear release behavior
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39 arising from interactions between polymer and drug properties. In summary, the graph theory
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41 mapping defined similarity and connectivity between different nanoparticle formulations, while
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43 capturing non-linearity in relationships that can be lost in linear analysis.
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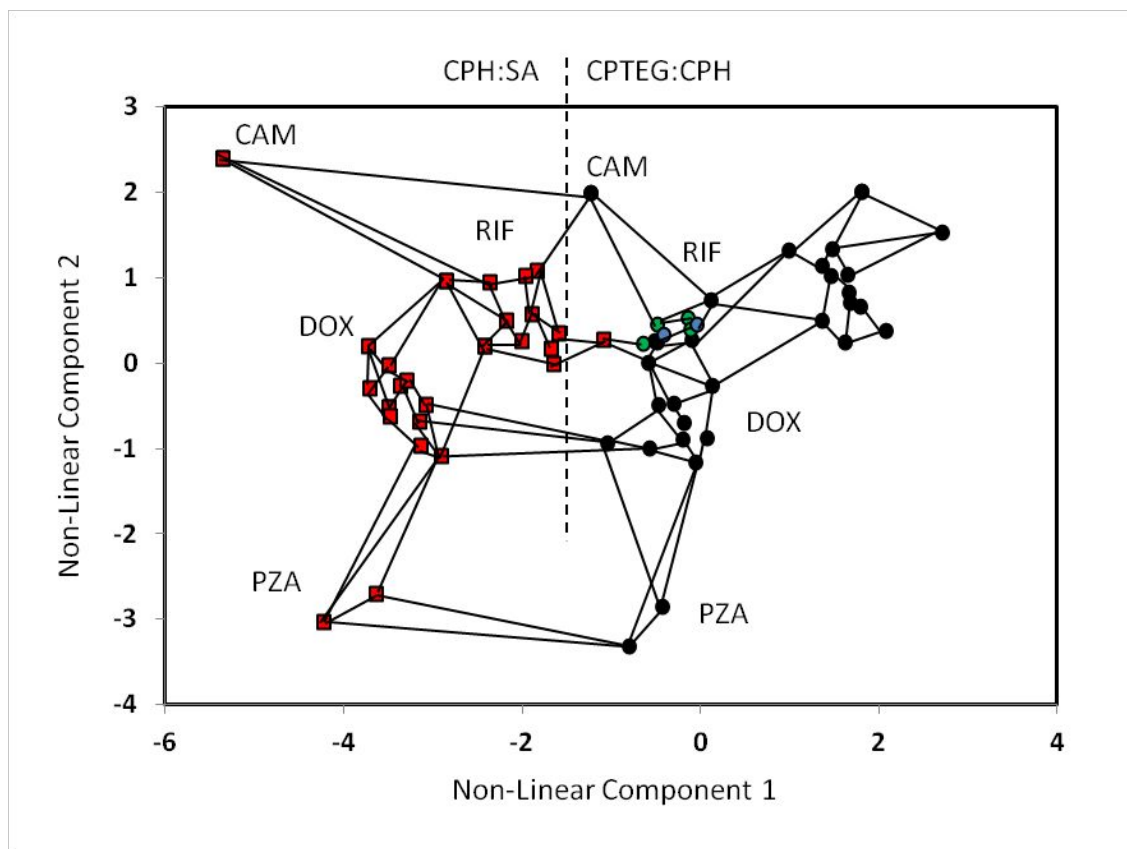
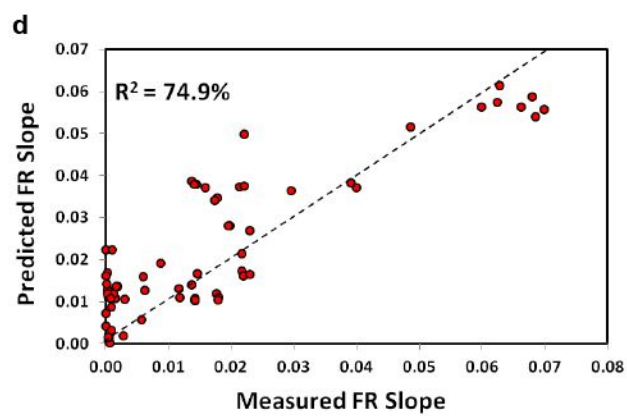
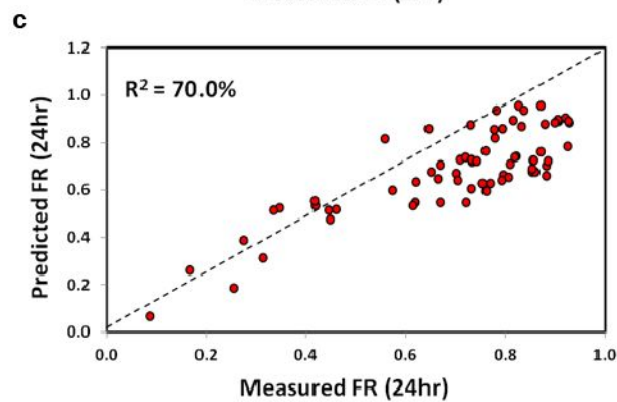
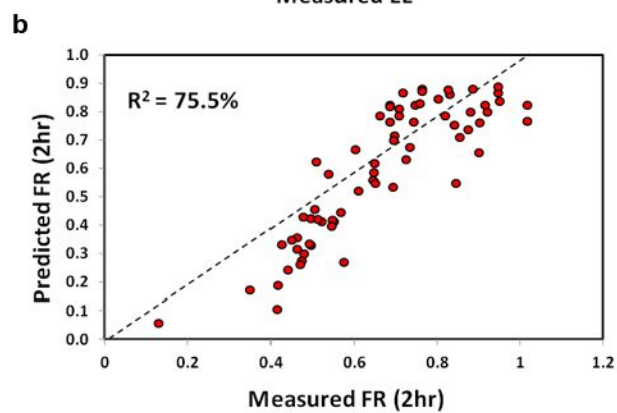
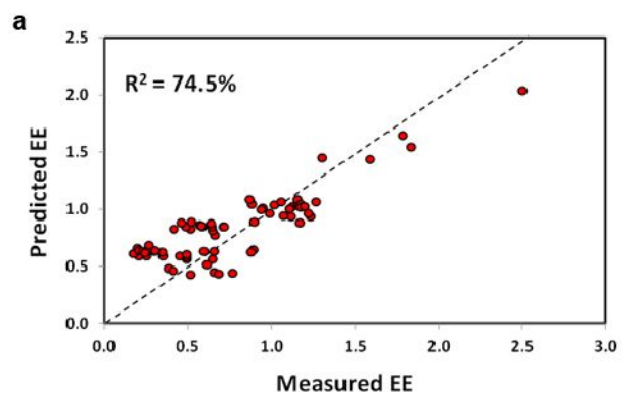


Figure 7. Graph theory map of formulation connectivity of release properties. Similarity between points is defined as the number of connections (solid lines) required to connect points. CPH:SA chemistries are represented by orange squares, while CPTEG:CPH chemistries are represented by circles (10:90 CPTEG:CPH in blue, 20:80 CPTEG:CPH in black, and >30:<70 CPTEG:CPH in green) This represents an approach for building a set of non-linearly derived parameters for performing high-throughput predictions. This approach was applied to a reduced descriptor set in order to develop a parameterization of the data, which ensures robustness by minimizing the number of input parameters, while incorporating non-linear relationships and maximizing variance in the data. Drug abbreviations: doxycycline (DOX), rifampicin (RIF), chloramphenicol (CAM), and pyrazinamide (PZA).

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3 As PCA projects data onto a linear manifold, it has difficulty explaining non-linear relationships.
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5 To this end, PCA demonstrated insufficient capability to accurately predict release properties in
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7 this data set. By contrast, graph theory can be used to project the data onto a non-linear manifold.
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10 This provides high-throughput modeling that accounts for non-linearity without requiring so
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12 many terms as to reduce the robustness of analysis. Therefore, the input into the predictions
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14 defines the graph theory values of Figure 7, which reflects a non-linear combination of
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16 descriptors, and a multi-linear regression between these values and the drug release properties
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18 was developed. It should be noted that the drug release properties were not included in the non-
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20 linear parameterization used for the prediction input, because that would result in predicting a
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22 property as a function of itself. The result of the high throughput modeling is shown in Figure 8.
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25 This represents a model with non-linear parameters that are a function of nanoparticle chemistry
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27 and theoretical drug loading and is defined generally so as to be applied to a wide range of
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29 chemistries. These models are fairly accurate, with R^2 values ranging between 70.0% and 75.5%.
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32 Cross validation was applied to ensure an even trade-off between robustness and accuracy. Since
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34 these methods are based on descriptors that can be generated for potential new nanoparticle
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36 formulations, the models provide a method to virtually explore a large search space. This method
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38 can guide experimentation by predicting target properties for a desired release profile, suggesting
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40 chemistries that match the targeted properties for testing.
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3 **Figure 8.** Graph theory high-throughput modeling of drug release properties. The horizontal axis
4 is the experimental measurements. The vertical axis is the predicted encapsulation efficiency (a),
5 2-hour burst release (b), 24-hour burst release (c), and the d2-d7 release profile slope values from
6 our model based on the reduced descriptor set. These calculations are based on a new hybrid
7 informatics approach where non-linear manifold projections serve as the input, thereby
8 accounting for greater complexity in descriptor-property relationships while also increasing the
9 robustness of the models. The models are reasonably accurate for all tested release properties.
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20 **3.4. Model validation.** To evaluate the robustness and accuracy of the multilinear models,
21 nanoparticles encapsulating two new antibiotic drugs (not included in the training data set),
22 meropenem and ceftazidime, were synthesized and characterized. Importantly, these drugs
23 contain sulfur atoms (Figure S4) which are not present in the four drugs used in the original
24 model training. The models in Figure 8 were used to predict the release properties for these new
25 formulations (Table 3). Based on these predictions, it is expected that all eight formulations
26 would show a high (>80%) burst release at 2h and 24h and minimal sustained release over d2-d7.
27 With the exception of the 20:80 CPH:SA–meropenem formulation, all other nanoformulations
28 are expected to show near-100% encapsulation efficiency.
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42 Strikingly, these predictions match experimental results closely. These new nano-formulations
43 displayed similar release profiles characterized by a >90% burst release within two hours,
44 followed by small amounts of drug released over the following two weeks (Figure S5, Table 3).
45 For this data set, the models tended to under-predict the burst release and over-predict the
46 sustained release behavior of the nano-formulations. The EE model was relatively accurate for
47 20:80 CPH:SA formulations, within 5%-20% of the measured EE's. The EE model showed more
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deviation from measured values for 20:80 CPTEG:CPH formulations, at circa 15-35% differences from the experimental values.

Table 3. Parameterized release properties of nanoparticles encapsulating two drugs (meropenem and ceftazidime) not included in training data set. FR (2h) fraction released in two-hour burst, FR (24h) fraction released in 24-hour burst.

Nano-formulation	EE		FR (2h)		FR (24h)		FR slope (FR/day)	
	predicted	measured	predicted	measured	predicted	measured	predicted	measured
20:80 CPTEG:CPH 5% meropenem	159%	195.0%	0.83	0.9406	0.84	0.9408	0.0033	0.000853
20:80 CPTEG:CPH 10% meropenem	128%	140.9%	0.81	0.9616	0.86	0.9617	0.0026	0.000271
20:80 CPH:SA 5% meropenem	29%	34.5%	0.95	0.9933	0.90	0.9933	0.0011	0.000656
20:80 CPH:SA 10% meropenem	49%	61.7%	0.98	0.9965	0.91	0.9967	0.0014	0.000168
20:80 CPTEG:CPH 5% ceftazidime	175%	156.8%	0.92	0.9997	0.88	1.0000	0.0019	0.000001
20:80 CPTEG:CPH 10% ceftazidime	190%	159.9%	0.91	0.9997	0.87	1.0000	0.0041	0.000004
20:80 CPH:SA 5% ceftazidime	114%	91.8%	0.97	0.9915	0.91	0.9952	0.0031	0.000771
20:80 CPH:SA 10% ceftazidime	129%	112.7%	0.94	0.9910	0.91	0.9941	0.0051	0.000686

To test the robustness of the models when adding new, untrained chemistries, eight nano-formulations were included in the models (compositional percentages were calculated including sulfur atoms, but without a separate descriptor for sulfur) and new regressions were calculated. We found R^2 values for EE, FR (2h), FR (24h) and d2-d7 slope after these inclusions to be 74.3%, 75.5%, 69.9%, and 74.6%, respectively. The small changes in regression from the original model data in Figure 8 indicates that the analytics methodology was able to incorporate new drug chemistries with minor impacts on the models. This confirms the robustness of the model and its capability to screen drug and polymer chemistries not included in the model development.

4. DISCUSSION

Due to the wide diversity of microbial infections, nanomedicines need to be customizable. Infections that are responsive to antibiotics may benefit from sustained release-skewing

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3 formulations by leveraging the dose sparing properties, limiting the risk of off-target effects,
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5 reducing the number of administrations, and enhancing patient compliance.^{4,8,39} Polyanhydride
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7 nanoparticles represent an attractive and adaptable nanomedicine platform by virtue of their
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9 tunable degradation and payload release rates,^{25,40} high biocompatibility,^{41,42} and efficient
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11 internalization by phagocytic cells.⁷
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15 Predictive analytics approaches have the potential to accelerate nanomedicine clinical
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17 translation, but the application of such informatics and data mining techniques to nanomedicine
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19 design has been slow to develop.¹³ To date, the majority of such efforts has focused on either
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21 linear dimensionality reduction through PCA and regression through PLS,²⁷ which provides
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23 insight into relationships between formulations and variables but has limited capacity to capture
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25 nonlinear behavior, or else artificial neural network “black box” models,^{43–45} which can capture
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27 nonlinear behavior but obscure interpretation of the structure of the model and dataspace. As the
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29 long-term goal of this research is to facilitate rational design of nanomedicine formulations,
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31 interpretation of the relationships between formulations is important. Accordingly, the
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33 dimensionality reduction approach was selected for this research, and paired with graph theory
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35 mapping to overcome the linearity limitations of PCA.
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42 A hybrid data mining approach was employed to deconvolute the complex polymer and drug
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44 relationships and develop QSPRs that describe release kinetics and encapsulation efficiency. We
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46 correlated antibiotic release properties from varying polyanhydride chemistries, encapsulated
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48 drug types, and drug loading within the nanoparticles. Through PCA analysis, we showed that
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50 release properties are dependent on both copolymer chemistry properties and drug properties,
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52 with polymer properties being more important. VIP analysis identified key polymer and drug
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3 descriptors that predicted drug release and encapsulation properties, but PCA was insufficient to
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5 predict release behavior from these formulations.
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9 Graph theory was used to characterize the multilinear connectedness and similarity of
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11 formulations, which can guide selection of replacement formulations with similar release
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13 behavior. For example, it is expected that 20:80 CPH:SA rifampicin-loaded nanoparticles
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15 (Figure 2b) would demonstrate similar release behavior (including burst release, slope of release
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17 between days 2-7, and encapsulation efficiency) as 20:80 CPH:SA doxycycline-loaded
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19 nanoparticles (Figure 2a) based on their close connections and proximal distance on the map
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21 (Figure 7). Similarly, 20:80 CPTEG:CPH pyrazinamide-loaded nanoparticles (Figure 2d) would
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23 be expected to show large differences in release behavior from 20:80 CPH:SA chloramphenicol-
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25 loaded nanoparticles (Figure 2c) due to the large number of lines needed to connect them and far
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27 distance on the map (Figure 7). The descriptors identified by VIP analysis were paired with the
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29 multilinear mapping from graph theory to generate predictive models for *a priori* screening of
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31 nanoparticle formulations with desired release kinetics and high encapsulation efficiency.
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37 The physicochemical properties of compounds influence their distribution, either in blood
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39 plasma or a polymer matrix. To this effect, VIP analysis (Figure 6) indicated that the descriptors
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41 most strongly correlated with release properties were both polymer and drug properties. This is
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43 expected, as favorable mixing thermodynamics allows distribution of the drugs inside the
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45 polymer device.^{10,46} In polyanhydride nanoparticles, such a distribution allows an erosion-
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47 controlled release profile, which tends toward sustained release.^{9,11} In contrast, poor mixing
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49 between the polymer and drug induces thermodynamic partitioning of the drug into polymer
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51 microdomains and/or localization at the particle surface, which skews the release profile toward
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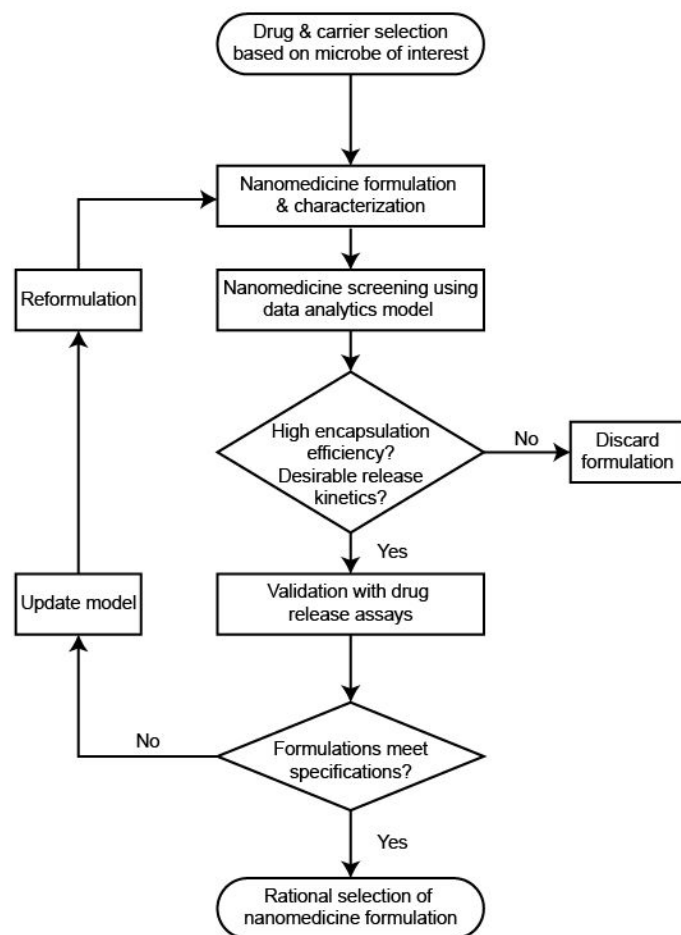
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3 a high-burst, diffusion-dominated regime.¹¹ Many of these same drug properties were correlated
4 with encapsulation efficiency, supporting the notion that polymer-drug mixing influences the
5 carrying capacity of delivery devices. As empty polyanhydride nanoparticles have a moderately
6 negative zeta potential,²³ the strong negative correlation between zeta potential and encapsulation
7 efficiency could reflect a strong surface localization of positively charged drugs. If this were the
8 case, however, we would expect a strong positive correlation between zeta potential and the two-
9 hour burst release, which was not observed. Regardless, the predictive power of this descriptor
10 could support the use of zeta potential as a quality control metric to ensure consistent
11 encapsulation efficiencies of lead formulations. While it is not surprising that these drug
12 properties affect encapsulation and release kinetics, this informatics analysis provides a sense of
13 their relative impact. Reducing the data space in this way can help guide rational selection of
14 antibiotic and polymer carrier pairs for nanomedicine formulations. These observations underline
15 the complexity of these relationships and provide support for the use of data analytics approaches
16 to enable rational design of nanomedicines.

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19 It should be noted that we can only confidently make quantitative predictions in chemical spaces
20 represented in our training data. While the additional testing of drugs containing sulfur, which
21 was not represented in our training data, resulted in approximately no change in accuracy,
22 materials that have unique behavior but with chemistries outside our training data may not be
23 quantitatively described by this approach. However, even in these cases, our approach has
24 significant impact. While the objective for the systems described by our training data is to predict
25 properties with high accuracy, the objective for systems containing groups and elements not in
26 our training data is to identify polymer and drug combinations which have the most promising

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3 characteristics and identify where additional experiments are needed. This leads to an iterative
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5 approach where necessary experiments are identified, thus feeding back to the analysis.
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9 From all of these results, we propose a framework for rational design and rapid testing of
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11 nanomedicine formulations (Scheme 1). In the first step, selected antibiotic drug candidates are
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13 encapsulated within nanoparticles of various polymer chemistries (potentially using high
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15 throughput techniques,²³ as demonstrated in section 3.4), and characterization of size distribution
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17 by SEM and zeta potential is obtained. These nanoparticle characteristics, along with polymer
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19 properties and drug properties, can then be fed into the multilinear graph theory model to predict
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21 encapsulation efficiencies and release kinetics. Nanomedicine candidates that demonstrate
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23 insufficient encapsulation and/or undesirable drug release profiles can be discarded. The *in vitro*
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25 performance of the lead nanomedicine candidates that emerge from this step can then be
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27 validated using drug release kinetics assays. A feedback reformulation loop allows gradual
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29 optimization of nanomedicine formulations and iterative updates to the models when release
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31 behavior deviates from predictions. In theory, this framework could be expanded to include other
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33 performance metrics, including internalization by appropriate cells and biological efficacy. As
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35 this methodology uses standard polymer and nanoparticle characterization techniques used in
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37 nano-carrier drug delivery research and publicly available drug information, this approach could
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39 be expanded to include other types of polymeric materials and other classes of small molecule
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41 drugs. This data analytics framework constitutes the first steps toward the rational design of
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43 nanomedicine formulations for antimicrobial therapies.
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51 **Scheme 1.** Data analytics framework for rapid nanomedicine design and screening.
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5. CONCLUSIONS

A multivariate data analytics approach was used to correlate drug release profiles from nanomedicine formulations based on different polyanhydride chemistries, encapsulated antibiotic drug type, and varying drug loading. We showed that both drug and polymer properties influence the drug encapsulation efficiency within the nanoparticles, the prevalence of burst in the drug release profile, and the slope of post-burst release. Polymer and drug properties that significantly impacted drug encapsulation efficiency and release kinetics were identified and defined a minimum descriptor set. The informatics analysis captured and preserved non-linear behavior governing relationships between drug type, polymer chemistry, and nanoparticle release properties, enabling interrogation of nanomedicine design pathways. We developed predictive

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3 models for drug release kinetics of untested drugs, using data from the Drugbank database and
4 nano-carrier characterization as inputs. Release kinetics predictions of two drugs containing
5 atoms not included in the model showed good agreement with experimental results, validating
6 the model and indicating its potential to virtually explore new polymer and drug pairs not
7 included in training data set. The models were shown to be robust after inclusion of these new
8 formulations in that there were no significant changes in the model regressions. This multilinear
9 modeling approach provides the first steps towards development of a framework that can be used
10 to rationally design nanomedicine formulations by selecting the appropriate carrier for a drug
11 payload to program desirable release kinetics profiles.
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23 6. SUPPORTING INFORMATION

24 Dendrogram raw and processed input data; Principal component analysis raw input data;
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26 Chemical structures of drugs in validation data set; Release kinetics of drugs in validation data
27 set.
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Author Contributions

ASM, BN, AMB, NPB, and BHB designed the experiments. ASM performed the experiments. ASM, SB, KR, and BN designed the informatics approach. SB and KR performed the informatics analysis. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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ABBREVIATIONS

CAM, chloramphenicol; CAZ, ceftazidime; CPH, 1,6-bis(*p*-carboxyphenoxy)hexane; CPTEG, 1,8-bis(*p*-carboxyphenoxy)-3,6-dioxaoctane; DOX, doxycycline; DSC, dynamic scanning calorimetry; EE, encapsulation efficiency; FR, fraction released; MEM, meropenem; NMR, nuclear magnetic resonance; PCA, principal component analysis; PZA, pyrazinamide; RIF, rifampicin; QSPR, quantitative structure-property relationship; SA, sebacic acid; SEM, scanning electron microscopy; T_g , glass transition temperature; T_m , melting point; VIP, variable importance projection

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