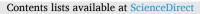
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# Al-Mn-based decagonal quasicrystal in AZ magnesium alloys and its nucleation on Al<sub>8</sub>Mn<sub>5</sub> during solidification

Di Wang<sup>a</sup>, Liuqing Peng<sup>a,b,\*</sup>, Christopher M. Gourlay<sup>a,\*</sup>

<sup>a</sup> Department of Materials, Imperial College London, London SW7 2AZ, UK

<sup>b</sup> National Engineering Research Centre for Magnesium Alloy, Chongqing University, Chongqing 400044, China

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Keywords: Quasicrystal Analytical electron microscopy Intermetallic compounds Heterogeneous nucleation of phase transformation Multicomponent solidification	Manganese is added to many magnesium alloys to control impurity iron, but its effects on phase transformations and microstructure formation remain incompletely understood. Here we show that an Al-Mn-based decagonal quasicrystal (d-QC) forms in the late stages of solidification in AZ31 and AZ91 magnesium alloys at relatively slow cooling rates, here down to 0.1 K/s. The d-QC has a periodicity of $\sim 12$ Å, grew as decagonal rods and commonly shared interfaces with Al <sub>8</sub> Mn <sub>5</sub> and eutectic Mg <sub>17</sub> Al <sub>12</sub> . A reproducible orientation relationship (OR) was measured with Al <sub>8</sub> Mn <sub>5</sub> only, indicating that the d-QC nucleated on Al <sub>8</sub> Mn <sub>5</sub> . The OR is consistent with the structural relationship between quasicrystals and gamma brasses and gives the d-QC nucleation advantages over LT-Al <sub>11</sub> Mn <sub>4</sub> during solidification. A subsequent heat treatment at 410 °C caused both Al <sub>8</sub> Mn <sub>5</sub> and the d-QC to transform into LT-Al <sub>11</sub> Mn <sub>4</sub> .

Many magnesium alloys, including the AM- and AZ- series, contain aluminium and a small (0.1 - 0.5 wt%) manganese addition [1]. The latter reduces the impurity iron content in the melt prior to casting [2] and ensures the iron remaining in the liquid crystallises into Al-(Mn,Fe) intermetallic compounds (IMCs) instead of Al-Fe IMCs which prevents excessive micro-galvanic corrosion of the α-Mg phase [3]. However, the Mn addition also adds significant complexity to the solidification sequence and results in multiple Al-Mn IMCs whose formation remains incompletely understood. For example, consider the Mg-rich corner of the Mg-Al-Mn liquidus projection in Fig. 1(a). The Scheil solidification paths are superimposed for Mg-3Al-0.5Mn, Mg-9Al-0.2Mn and Mg-9Al-0.06Mn (wt.%). The first two have Al<sub>8</sub>Mn<sub>5</sub> as the primary phase while the latter has α-Mg as the primary phase. Then, in all three alloys, the liquid composition follows the same eutectic grooves and passes two quasi-peritectic points involving  $\alpha$ -Mg and various Al-Mn IMCs (Al<sub>8</sub>Mn<sub>5</sub>, LT-Al<sub>11</sub>Mn<sub>4</sub> and then  $\mu$ -Al<sub>4</sub>Mn) until it reaches the ternary eutectic point of  $\alpha$ -Mg+Mg<sub>17</sub>Al<sub>12</sub>+ $\mu$ -Al<sub>4</sub>Mn. The results in Fig. 1(a) are only subtly different for alloys with a dilute Zn addition such as AZ31 and AZ91 since Zn has negligible solubility in these Al-Mn intermetallics [4].

Past experimental work on cast AZ31 and AZ91 has presented clear evidence (with diffraction) for  $Mg_{17}Al_{12}$  [5],[6],  $Al_8Mn_5$  [7] [8,9,10,11, 12], and LT-Al<sub>11</sub>Mn<sub>4</sub> [10],[11], after solidification but  $\mu$ -Al<sub>4</sub>Mn has not

been identified conclusively in these alloys. Additionally, LT-Al<sub>11</sub>Mn<sub>4</sub> is usually only present in trace amounts after solidification unless very low cooling rates are applied (0.1 K  $s^{-1}$  in [10]). This indicates that the Al-Mn intermetallics formed in the last stages of AZ31 and AZ91 solidification may exhibit nucleation and growth difficulties and meta-stability. Therefore, there is a need for a study to better understand Al-Mn phase formation in the last stages of solidification in AZ31 and AZ91.

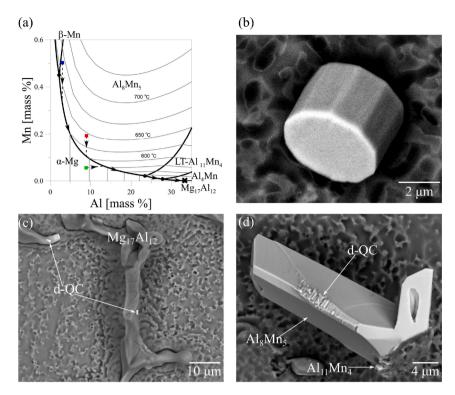
Al-Mn intermetallics have been studied intensively in binary Al-Mn alloys and Al-Mn-TM alloys (TM = another transition metal) [13,14, 15,16], and icosahedral quasicrystals (i-QCs) [17] and decagonal quasicrystals (d-QCs) [18,19], were first discovered in this system. In the latter stages of Mg-(3–9)Al-xMn alloy solidification (Fig. 1(a)) the calculated equilibrium Al-Mn compounds (LT-Al<sub>11</sub>Mn<sub>4</sub> and then  $\mu$ -Al<sub>4</sub>Mn [4]) have compositions in the range 73 – 80 at.% Al which is similar to multiple known equilibrium and metastable Al-Mn phases including an i-QC and d-QC [17,18,20,21,22,23,13,24]. However, an interesting difference is that the 'low temperature' Al<sub>8</sub>Mn<sub>5</sub> and LT-Al<sub>11</sub>Mn<sub>4</sub> phases crystallise in magnesium-rich liquid in Mg-Al-Mn-based alloys [25] whereas they do not exist in equilibrium with liquid in the binary Al-Mn system [13,26,27]. This makes magnesium-aluminium-manganese alloy solidification an interesting space to gain new insights into the competition and cooperation

\* Corresponding authors. *E-mail addresses:* liuqing.peng@cqu.edu.cn (L. Peng), c.gourlay@imperial.ac.uk (C.M. Gourlay).

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**Fig. 1.** (a) Calculated liquidus projection of the Mg-rich corner of the Mg-Al-Mn system from ThermoCalc database TCMG4. Scheil solidification paths of Mg-3Al-0.5Mn (blue), Mg-9Al-0.2Mn (red), and Mg-9Al-0.06Mn (green) are superimposed. X=the ternary eutectic point. (b)-(d) intermetallics in deep-etched Mg-9Al-0.7Zn-0.2Mn. (b) d-QC rod; (c) d-QC rods on Mg<sub>17</sub>Al<sub>12</sub> and (d) d-QC rods on Al<sub>8</sub>Mn<sub>5</sub>.

 Table 1

 Compositions (wt%) of AZ31–0.5Mn, AZ91–0.06Mn and AZ91–0.2Mn.

	Mg	Al	Zn	Mn	Fe	Cu	Si	Ni
AZ31-0.5Mn	Bal.	3.09	1.06	0.49	0.003	0.002	0.067	< 0.001
AZ91-0.06Mn	Bal.	9.24	0.76	0.056	0.0013	0.0008	0.035	0.0002
AZ91-0.2Mn	Bal.	8.95	0.72	0.19	< 0.001	0.001	0.039	< 0.001

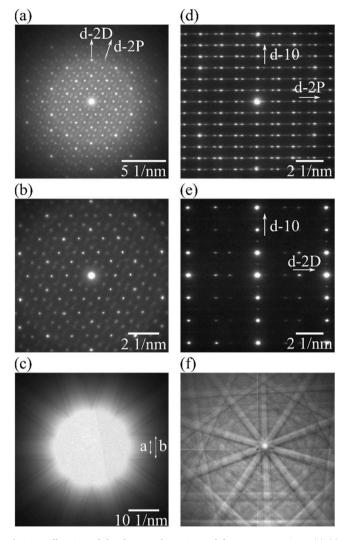
amongst complex Al-Mn IMCs.

For this paper, we studied one AZ31 and two AZ91 alloys with compositions in Table 1, which are similar to the three alloys in Fig. 1(a). Three solidification conditions were applied: (i) gravity casting, (ii) melting ~2 g samples within Al<sub>2</sub>O<sub>3</sub> cups in quartz tubes under Ar before placing the tube into a cylindrical hole in a room-temperature steel mould to solidify, and (iii) the same as (ii) but with furnace cooling. Detailed conditions for method (i) are given in ref. [28] and for methods (ii) and (iii) in ref. [10]. The cooling rates in the liquid prior to  $\alpha$ -Mg nucleation were measured to be ~90 K  $s^{-1}$ , 4 K  $s^{-1}$  and 0.1 K  $s^{-1}$  by a K-type thermocouple, either within the cavity of the gravity casting mould or on the outside of the quartz tube. For the slowest cooling rate (method (iii)), samples were switched to cooling method (ii) from 410 °C to minimise any solid-state transformations.

Samples of each alloy were examined in the as-cast condition and the two AZ91 alloys were additionally examined after solution heat treatment in quartz tubes under Ar at 410 °C for 1 day and 15 days followed by quenching in water. For microstructural characterisation, some samples were prepared as polished cross-sections and others were deep etched using 10 % nitric acid in ethanol for 5–10 min to enable imaging of the 3D morphology of IMCs.

A ThermoFisher Scientific Helios 5CX dual-beam scanning electron microscope and focused ion beam (SEM-FIB) instrument was used for site-specific lift out of thin lamellae. These were then investigated by transmission electron microscopy (TEM) in a JEOL 2100F fieldemission-gun (FEG) TEM. SEM imaging and electron back-scattered diffraction (EBSD) were conducted using two Zeiss Sigma-300 FEG-SEMs, one with a Bruker  $e^{-}$ FlashHR detector and the other with an EDAX Clarity detector.

In the three alloys and under all cooling conditions, it was found that an Al-Mn IMC was present with the morphology of decagonal rods as summarised in Fig. 1(b)-(d). Commonly, the rods shared interfaces with Mg<sub>17</sub>Al<sub>12</sub> and Al<sub>8</sub>Mn<sub>5</sub> as shown in the examples in Fig. 1(c)-(d). Structural characterisation of the Al-Mn rod phase by TEM selected area electron diffraction (SAED) confirmed that it is a decagonal quasicrystal (d-QC). Diffraction patterns of d-QC rods are shown in Fig. 2. FIB lamellae prepared such that the long rod axis was in the TEM incident beam direction gave SAED patterns along the 10-fold (d-10) axis, Fig. 2 (a)-(b), in which the directions of the two distinct 2-fold axes are labelled as 2D and 2P [29]. Both Fig. 2(a) and 2(b) show 10-fold rotational symmetry. Note that the SAED patterns in Fig. 2(a) and 2(b) were taken along the same orientation but at different camera length. Fig. 2(c) is the convergent beam electron diffraction (CBED) pattern from the same orientation as the SAED patterns, exhibiting 10-fold symmetry. Each Kikuchi band in Fig. 2(c) consists of two colinear bands with band width ratio=1.62 ~  $\tau$  (the golden mean ratio), arising from its quasi-periodic nature consistent with [30]. SAED patterns along the 2D and 2P directions are shown in Fig. 2(d) and Fig. 2(e) respectively, which were FIB milled to be 90° away from the rod axis (i.e. the 10-fold axis) and were 18° away from each other. SAED patterns along the 2D and 2P directions are periodic along the d-10 axis with a periodicity of  $\sim$ 12.3 Å, similar to the  $\sim$ 12 Å periodicity of the Al-Mn-type d-QC found



**Fig. 2.** Diffraction of the decagonal quasicrystal from AZ91–0.06Mn. (a)-(c): SAED and CBED taken along the longitudinal direction of a rod such as Fig. 1 (b). (b) is zoomed in of (a). All three show clear 10-fold symmetry. In (c) the ratio of band widths  $b/a = 1.62 \sim \tau$ . (d) and (e): SAED from 2D and 2P respectively, 18° a part to each other. (f): EBSD pattern with 10-fold axis near the centre.

### Table 2

Composition (at.%) of d-QC, LT-Al<sub>11</sub>Mn<sub>4</sub> and Al<sub>8</sub>Mn<sub>5</sub> measured from TEM-EDS.  $^{1}$ =AZ91–0.06Mn,  $^{2}$ =AZ91–0.2Mn,  $^{3}$ =AZ31–0.5Mn.

	Al	Mn	Mg	Fe	
d-QC <sup>1</sup> d-QC <sup>2</sup> d-QC <sup>3</sup>	76.2	19.1	4.4	0.3	
d-QC <sup>2</sup>	77.8	20.1	2.0	0.1	
d-QC <sup>3</sup>	74.6	21.5	3.8	0.1	
LT-Al <sub>11</sub> Mn <sub>4</sub>	76.0	23.5	0.5	_	
Al <sub>8</sub> Mn <sub>5</sub>	61.2	38.6	0.3	_	

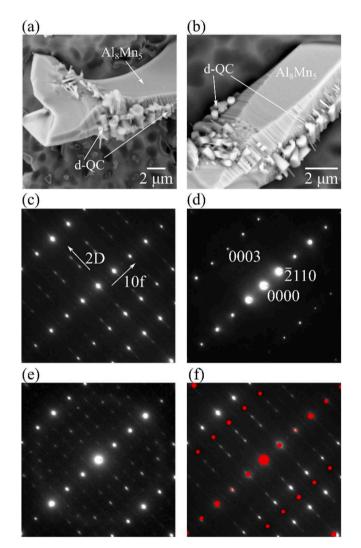
by Bendersky [18]. This  $\sim$ 12 Å periodicity can also be expressed as a D<sub>3</sub> structure based on the D<sub>i</sub> designation from [31]. An EBSD pattern from the d-QC with the 10-fold axis near the centre is shown in Fig. 2(f).

The chemical compositions of the Al-Mn IMCs measured from TEM energy dispersive X-ray spectroscopy (EDS) are shown in Table 2. The Al/Mn ratio of the d-QC is close to Al<sub>4</sub>Mn, which is similar to prior studies on decagonal and approximant phases in Al-Mn alloys [18,32,33, 34]. In addition, around 3 at.% Mg and 0.15 at.% Fe were found in the d-QC in this work. The d-QC in AZ31–0.5Mn has slightly less Al and more Mn than the d-QC in AZ91.

### Table 3

Summary of Al-Mn IMCs in three AZ magnesium alloys with three different solidification conditions.  $Al_{11}Mn_4$  refers to LT-Al<sub>11</sub>Mn<sub>4</sub> [35].

Alloy	Gravity casting	Moderate cooling	Slow cooling
Wt.%	$\sim$ 90 K $s^{-1}$	4 K $s^{-1}$	$0.1 \text{ K} \text{ s}^{-1}$
Mg-3Al-1.0Zn- 0.5Mn Mg-9Al-0.7Zn- 0.2Mn Mg-9Al-0.7Zn- 0.06Mn	$\begin{array}{l} Al_8Mn_5+d-\\ QC\\ Al_8Mn_5+d-\\ QC\\ Al_8Mn_5+d-\\ QC\\ QC \end{array}$	$\begin{array}{l} Al_8 Mn_5 + Al_{11} Mn_4 + \\ d\text{-QC} \\ Al_8 Mn_5 + Al_{11} Mn_4 + \\ d\text{-QC} \\ Al_8 Mn_5 + Al_{11} Mn_4 + \\ d\text{-QC} \end{array}$	$\begin{array}{l} Al_8 Mn_5 + Al_{11} Mn_4 \\ + \ d\text{-}QC \\ Al_8 Mn_5 + Al_{11} Mn_4 \\ + \ d\text{-}QC \\ - \end{array}$



**Fig. 3.** Orientation relationship between Al<sub>8</sub>Mn<sub>5</sub> and d-QC. (a)-(b) Al<sub>8</sub>Mn<sub>5</sub> rods with d-QC on the surface from AZ91 with 0.2Mn sample. (c) SAED of d-QC along d-2P axis; (d) SAED of Al<sub>8</sub>Mn<sub>5</sub> along [0110]; (e) SAED of both and (f) SAED of d-QC superimposed with simulated pattern of Al<sub>8</sub>Mn<sub>5</sub>. Two red circles are drawn in (f) to indicate the positions of simulated weak (4220) and (4220) spots of Al<sub>8</sub>Mn<sub>5</sub>.

The Al-Mn IMCs found for each combination of composition and solidification condition are listed in Table 3.  $Al_8Mn_5$ , LT-Al<sub>11</sub>Mn<sub>4</sub> and d-QC were all present at the lowest cooling rate (0.1 K  $s^{-1}$ ). With increasing cooling rate, the fraction of d-QC increased and the fraction of LT-Al<sub>11</sub>Mn<sub>4</sub> deceased until, in gravity castings, no LT-Al<sub>11</sub>Mn<sub>4</sub> was found. No  $\mu$ -Al<sub>4</sub>Mn was detected in this work. The fraction of d-QC also increased with increasing Al/Mn ratio in the alloy; the highest fraction

## Table 4

Approximant phases used for (mis)indexing EBSD patterns from the d-QC. Cross-correlation coefficients (CCC) quantify the match between one d-QC pattern and the best-matching simulated pattern for each approximant.  $Al_8Mn_5$  and  $LT-Al_{11}Mn_4$  are included to show their lattice parameters and EBSD pattern similarity to that of d-QC.

Phase Space group	Space group	Pearson symbol	Lattice parameter (nm and°)						CCC	Ref.
		а	b	с	α	β	γ			
T-Al <sub>3</sub> Mn	Pnma	oP156	1.48	1.24	1.26	90	90	90	0.478	[48]
R-Al <sub>4</sub> Mn	Cmcm	oS156	2.36	1.24	0.77	90	90	90	0.427	[49]
Al <sub>3</sub> Pd	Pna2 <sub>1</sub>	-	2.34	1.67	1.23	90	90	90	0.380	[50]
$\mu$ -Al <sub>4</sub> Mn	P6 <sub>3</sub> /mmc	hP563	2.00	2.00	2.47	90	90	120	0.376	[20]
$\lambda$ -Al <sub>4</sub> Mn	P6 <sub>3</sub> /m	hP568	2.84	2.84	1.24	90	90	120	0.362	[21]
Al <sub>13</sub> Fe <sub>4</sub>	C2/m	mS102	1.55	0.80	1.25	90	107.7	90	0.371	[51]
Al <sub>8</sub> Mn <sub>5</sub>	R3m	hR26	1.27	1.27	0.79	90	90	120	0.355	[37]
LT-Al11Mn4	P-1	aP15	0.51	0.89	0.51	89.4	100.5	105.1	< 0.1	[35]

of d-QC was in Mg-9Al-0.7Zn-0.06Mn (Al/Mn = 150) and the lowest in Mg-3Al-1Zn-0.5Mn (Al/Mn = 6).

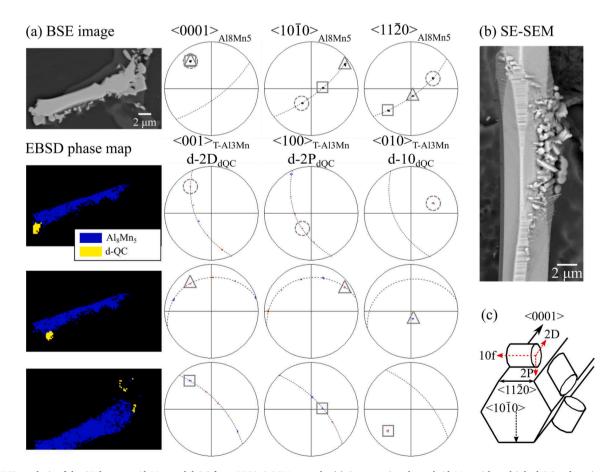
The d-QC and (where present) LT-Al<sub>11</sub>Mn<sub>4</sub> usually grew on the Al<sub>8</sub>Mn<sub>5</sub> phase (e.g. Fig. 1(d)) which formed earlier in the solidification sequence, Fig. 1(a). Micrographs of typical microstructures of the Al-Mn IMCs at each combination of composition and cooling condition are given as Supplementary Information (Figure SI- 1). d-QC rods on the surfaces of Al<sub>8</sub>Mn<sub>5</sub> were usually aligned in certain common directions as shown in Fig. 3(a)-(b), indicating an orientation relationship (OR) between the two phases. Typical TEM SAED results of the OR are shown in Fig. 3(c)-(f). Fig. 3(c)-(e) are SAED patterns along the d-QC d-2P direction, the Al<sub>8</sub>Mn<sub>5</sub> [0 $\overline{110}$ ] direction and a region containing both phases. Fig. 3(f) is the combination of the measured d-QC pattern (Fig. 3(c))

and the simulated Al<sub>8</sub>Mn<sub>5</sub> pattern (simulated in the [0110] direction of Fig. 3(d)) to help interpret the experimental pattern from both phases in Fig. 3(e). Note the overlapping spots along d-10 of the d-QC and {1120} of Al<sub>8</sub>Mn<sub>5</sub> in Fig. 3(f) which is associated with an excellent planar d-spacing match with ~0.16 % misfit. This OR can be expressed as Eq. (1) in the hexagonal setting and Eq. (2) in the body centred rhombohedral (BCR) setting of Al<sub>8</sub>Mn<sub>5</sub>:

$$\frac{[2110]_{Al8Mn5}}{[0003]_{Al8Mn5}}/d - 10$$

$$\frac{[0003]_{Al8Mn5}}{[0110]_{Al8Mn5}}/d - 2D$$
(1)

and



**Fig. 4.** EBSD analysis of the OR between  $Al_8Mn_5$  and d-QC from AZ91–0.06Mn sample. (a) Cross-section through  $Al_8Mn_5$  with multiple d-QC rods on its surface. Pole figures are shown for selected regions of d-QC highlighting one OR with three variants. The d-QC is indexed using the crystalline approximant T-Al<sub>3</sub>Mn. (b) Multiple d-QC rods on  $Al_8Mn_5$  after deep etching (c) schematic showing the d-QC on  $Al_8Mn_5$  with the three variants of the OR.

$$\frac{[\bar{1}10]_{BCR}}{[111]_{BCR}} / d - 10$$

$$\frac{[111]_{BCR}}{[11\bar{2}]_{RCR}} / d - 2D$$

$$\frac{[11\bar{2}]_{RCR}}{[12]_{RCR}} / d - 2P$$

$$(2)$$

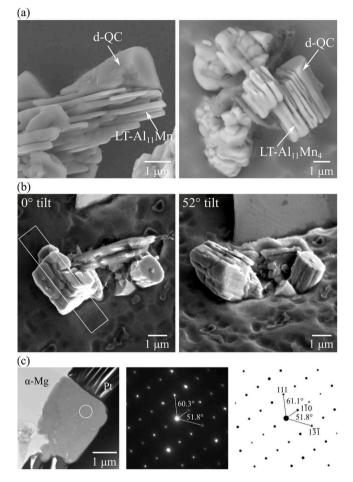
Al<sub>8</sub>Mn<sub>5</sub> is a rhombohedral gamma brass [24,36–38]. Previous studies have discussed gamma brasses as quasicrystal approximants [39,40,41, 42], and, from structural considerations, the expected orientation relationship between cubic gamma brasses and d-QCs was predicted by Dong [39] to be:  $[1\overline{10}]$  // d-10; [001], [112] and [221] // d-2P; and [115], [111] and [110] // d-2D. Rhombohedral Al<sub>8</sub>Mn<sub>5</sub> is related to the cubic gamma brasses by a small rhombohedral distortion ( $\alpha \sim 89.1^{\circ}$ when using a BCR unit cell) and small atomic shifts [36,37,24,38]. The OR measured here between Al<sub>8</sub>Mn<sub>5</sub> and the d-QC is consistent with the OR predicted by Dong [39] for cubic gamma brasses when considering the reduction of symmetry to rhombohedral Al<sub>8</sub>Mn<sub>5</sub>.

The variants of the OR in Fig. 3 were studied by SEM EBSD. One approach to index quasicrystal EBSD patterns is to (mis)index them to a closely related crystalline approximant phase and then convert back to d-QC indices using the decagonal pseudosymmetry of the approximant [43,44]. To identify a suitable approximant for the Al-Mn-(Mg) d-QC, we compared experimental EBSD patterns from the d-QC with the range of d-QC approximants in Table 4 using Bruker DynamicS. The approach involved finding the dynamical simulated pattern of each approximant that best matched the measured d-QC EBSD pattern and then finding the closest matching approximant by comparing the cross-correlation coefficients. As shown in Table 4 the best match was to T-Al<sub>3</sub>Mn (also known as HT-Al<sub>11</sub>Mn<sub>4</sub>) closely followed by R-Al<sub>4</sub>Mn, (also known as  $\pi$ -Al<sub>4</sub>Mn) and these were substantially better than the other candidate approximants. Consistent with this, both T-Al<sub>3</sub>Mn [34,45,46], and R-Al<sub>4</sub>Mn [14,47], are known to be approximants of the Al-Mn d-QC.

Past work [46] has shown that the Al-Mn d-QC and T-Al<sub>3</sub>Mn are related by d-2P // <100><sub>Al3Mn</sub>, d-2D // <001><sub>Al3Mn</sub> and d-10 // <010><sub>Al3Mn</sub>. In Fig. 4, the <001><sub>Al3Mn</sub> pole figure contains five spots, each with a different colour, rotated 36° about <010> from the neighbouring spot, representing the ten d-2D directions (five in the northern hemisphere). The <100><sub>Al3Mn</sub> pole figure also contains five spots, each rotated 36° about <010> from the neighbour, representing the ten d-2D directions (five in the northern hemisphere). The <100><sub>Al3Mn</sub> pole figure also contains five spots, each rotated 36° about <010> from the neighbour, representing the ten d-2P directions, where each is rotated 18° about <010> from the nearest <001>. In the <010><sub>Al3Mn</sub> pole figure, all five spots overlap in the d-10 direction. Considering the five pseudo-T-Al<sub>3</sub>Mn orientations together gives the d-QC orientation.

In Fig. 4(a), multiple d-QC rods surrounded a single  $Al_8Mn_5$  rod. There were three regions of d-QC rods and each region had the OR in Eq. (1) to the  $Al_8Mn_5$  single crystal. The schematic in Fig. 4(c) highlights that the three variants arise because the d-10 axis of the d-QC can be parallel to any one of the three  $<11\bar{2}0>$  axes of the  $Al_8Mn_5$  rod, with a d-2D axis parallel to the  $<0001>_{Al8Mn5}$  and a d-2P axis parallel to the  $<10\bar{1}0>_{Al8Mn5}$  in each case. Examining the SE image of a deep etched  $Al_8Mn_5$  rod in Fig. 4(b), notice that the d-QC rods mostly have one of three alignments relative to the  $Al_8Mn_5$  rod, associated with the three variants of the OR.

During solidification, both the d-QC and LT-Al<sub>11</sub>Mn<sub>4</sub> mostly grew on Al<sub>8</sub>Mn<sub>5</sub> (e.g. Fig. 1(d)), the fraction of LT-Al<sub>11</sub>Mn<sub>4</sub> decreased as the cooling rate increased (Table 4), and  $\mu$ -Al<sub>4</sub>Mn was not detected (Table 4). Thus, it is probable that the d-QC formed during solidification because it outcompeted the LT-Al<sub>11</sub>Mn<sub>4</sub> (and possibly  $\mu$ -Al<sub>4</sub>Mn) phase(s) in terms of nucleation and/or growth on Al<sub>8</sub>Mn<sub>5</sub>. Note that, while the d-QC is best approximated by the T-Al<sub>3</sub>Mn phase (Table 4), it also shares similar icosahedral building blocks with the LT-Al<sub>11</sub>Mn<sub>4</sub> [35] and  $\mu$ -Al<sub>4</sub>Mn [33,52], phases and so the d-QC phase is outcompeting phases with which it shares structural similarities. We have previously reported ORs for LT-Al<sub>11</sub>Mn<sub>4</sub> growing on Al<sub>8</sub>Mn<sub>5</sub> during solidification at low cooling rate [10]. Those LT-Al<sub>11</sub>Mn<sub>4</sub>/Al<sub>8</sub>Mn<sub>5</sub> ORs were more complex than the d-QC/Al<sub>8</sub>Mn<sub>5</sub> OR found here and, often, the measured planes in the LT-Al<sub>11</sub>Mn<sub>4</sub>/Al<sub>8</sub>Mn<sub>5</sub> oR were only nearly parallel [10]. From the simple and highly reproducible OR between the d-QC and Al<sub>8</sub>Mn<sub>5</sub> in



**Fig. 5.** Solid-state transformation of d-QC to LT-Al<sub>11</sub>Mn<sub>4</sub> at 410 °C. (a) d-QC rods partially transformed to layers of LT-Al<sub>11</sub>Mn<sub>4</sub> from the AZ91–0.06Mn sample held at 410 °C for 1 day and 15 days respectively; (b) SEM image of such a transformed rod in (a) from the AZ91–0.06Mn sample held at 410 °C for 15 days. A TEM lamella was lifted out by FIB in the white rectangle region. (c) ADF-STEM of the FIB lamellae in (b) and experimental and simulated SAED from the white circle region, viewed along LT-Al<sub>11</sub>Mn<sub>4</sub> [112]. N.B. LT-Al<sub>11</sub>Mn<sub>4</sub> = triclinic P-1 [35].

Fig. 3 and Fig. 4, it is likely that there is easier nucleation kinetics for the d-QC on  $Al_8Mn_5$  than for LT- $Al_{11}Mn_4$  on  $Al_8Mn_5$ , which ultimately seems to arise from the structural relationship between quasicrystals and gamma brasses [39].

AZ91 alloys are often given a solution heat treatment of 16–24 h at 413  $\pm$  6 °C [53] to dissolve the  $Mg_{17}Al_{12}$  phase. Microstructures of the d-QC particles after 1 day and 15 days at 410 °C are shown in Fig. 5(a). The d-QC rods are in the process of transforming into parallel plates of LT-Al\_{11}Mn\_4, with a morphology similar to past work when  $Al_8Mn_5$  transforms into LT-Al\_{11}Mn\_4 at 410 °C [10]. FIB lift-out and TEM analysis of a plate from a partially transformed d-QC rod is overviewed in Fig. 5 (b)-(c). The SAED pattern in Fig. 5(c) is along the [112] zone axis of LT-Al\_{11}Mn\_4 and corresponding TEM EDS data are shown in Table 3. This transformation from d-QC to LT-Al\_{11}Mn\_4 shows that the d-QC is metastable at 410 °C.

In a similar AZ91 alloy, Zeng et al. [54] reported a decagonal Al-Mn-(Mg) phase of 20–200 nm size and Al:Mn ratio of 6.5 within the  $\alpha$ -Mg phase, that was stable during solution treatment at 420 °C for 72 h. There are significant differences between the d-QC reported in that study [54] and the present paper. Here, the Al-Mn-(Mg) d-QC was ~2  $\mu$ m, had an Al:Mn ratio of ~4.0, grew in the liquid in the latter stages of solidification, was usually attached to Al<sub>8</sub>Mn<sub>5</sub>, and was unstable at 410 °C where it transformed into LT-Al<sub>11</sub>Mn<sub>4</sub>. Therefore, further work is

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required to study Mn-bearing IMCs within the  $\alpha$ -Mg matrix in AZ91 and understand how they relate to the d-QC that formed by solidification on Al<sub>8</sub>Mn<sub>5</sub> in the present work.

In summary, an Al-Mn-based decagonal quasicrystal with composition Al<sub>76.2</sub>Mn<sub>20.2</sub>Mg<sub>3.4</sub>Fe<sub>0.2</sub> (at.%) has been found in AZ-series alloys solidified at cooling rates from 0.1 K s<sup>-1</sup> - ~90 K s<sup>-1</sup>. It is an Al-Mn-type d-QC with periodicity around 12.3 Å. The d-QC was commonly found on the surfaces of Al<sub>8</sub>Mn<sub>5</sub> with an OR involving three variants linked to the similarities between gamma brasses and quasicrystals. The d-QC formed in the latter stages of solidification where it outcompeted the equilibrium LT-Al<sub>11</sub>Mn<sub>4</sub> phase (and possibly the  $\mu$ -Al<sub>4</sub>Mn phase) due to favourable nucleation on Al<sub>8</sub>Mn<sub>5</sub> that forms earlier in the solidification sequence. The d-QC is a metastable phase at the typical solution heat treatment temperature of 410 °C where it transformed into LT-Al<sub>11</sub>Mn<sub>4</sub>.

## **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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#### Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.scriptamat.2023.115886.

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