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Phase segregation and nanoconfined fluid O_2 in a lithium-rich oxide cathode

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Oxygen-redox cathodes

- Lithium-rich: $Li_{1+x}M_{1-x}O_2$ compositions
- Capacity from cationic and anionic redox
- Voltage hysteresis and voltage fade
- Structural changes during cycling

Modelling O-redox

- How can we obtain realistic structures?
- Kinetically-viable structural rearrangements
- Thermodynamic ground-state structures
- Disorder & nanoscale structural changes

thermodynamics







$Li_{1,2}Mn_{0,8}O_2$

• O2 stacking, ribbon superstructure



Phase segregation $Mn_{0.8}O_2 \rightarrow two-phase mixture of MnO_2 + O_2$

Cluster expansion

- Lattice representation of O₂ molecules
- Equivalent lattice configurations: sample for a ground-state structure

Mapping relaxed structures to lattice



Thermodynamic ground state structure: **lattice Monte Carlo**

- Simulated annealing: obtain ground-state structures
- Nanovoids containing confined O₂ molecules

Nanoconfined fluid O₂

Confined O_2 in the voids is a supercritical fluid and can diffuse over long distances, through the bulk

Methods

AIMD

- AIMD with PBE+U (VASP) to sample structural changes
- 300K, 600K, 900K simulations in 120 atom unit cell
- HSE06 relaxations (CRYSTAL) along AIMD trajectory

Cluster expansion

- CE fit: *ICET* with ~400 HSE06-relaxed structures
- AIMD step prevents ambiguous structure mapping \bullet
- LASSO+RFE fit: CV RMSE of <22 meV atom⁻¹

Conclusions

Combination of AIMD and cluster expansion

GENERATION LI-ION CATHODE MATERIALS

obtain kinetically viable and thermodynamically stable O-redox structures • Phase segregation forms nanovoids • Fluid O_2 can diffuse through the structure, linking bulk O_2 formation and surface O_2 loss

Preprint

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