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## Supplementary materials

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### **Developing a slow-release permanganate composite for degrading aquaculture antibiotics**

Chainarong Sakulthaew <sup>1</sup>, Chanat Chokejaroenrat <sup>2\*</sup>, Sidaporn Panya <sup>2</sup>, Apisit Songsasen <sup>3</sup>, Kitipong Poomipuen <sup>1</sup>, Saksit Imman <sup>4</sup>, Nopparat Suriyachai <sup>4</sup>, Torpong Kreetachat <sup>4</sup> and Steve Comfort <sup>5</sup>

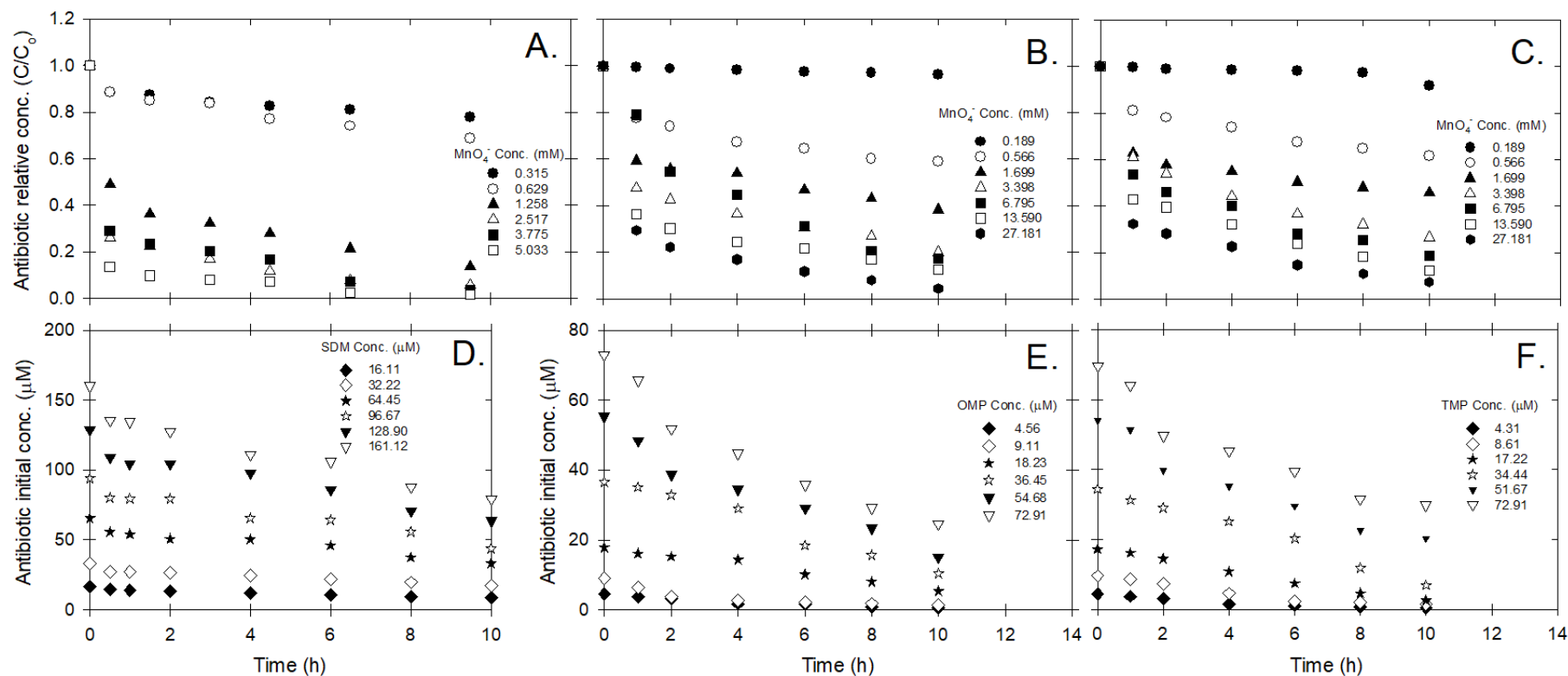
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10 Figures

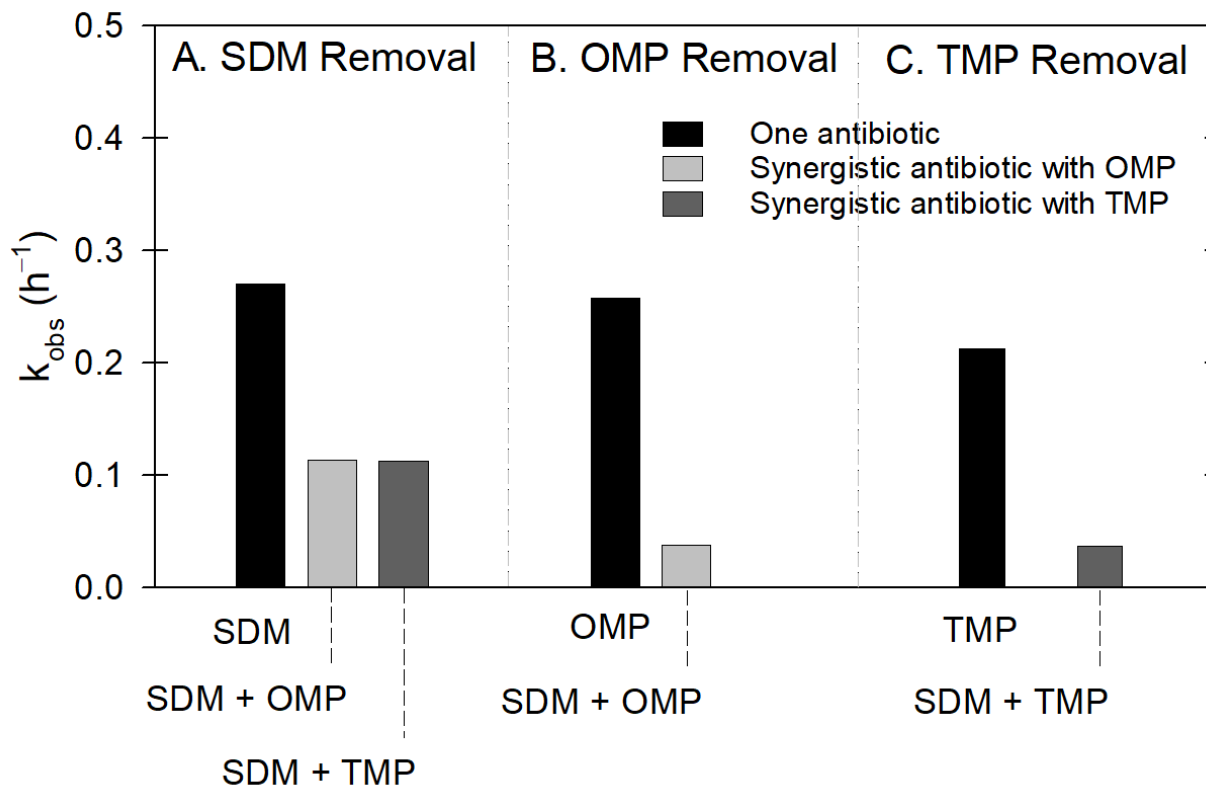
4 Tables

- 15 pages –

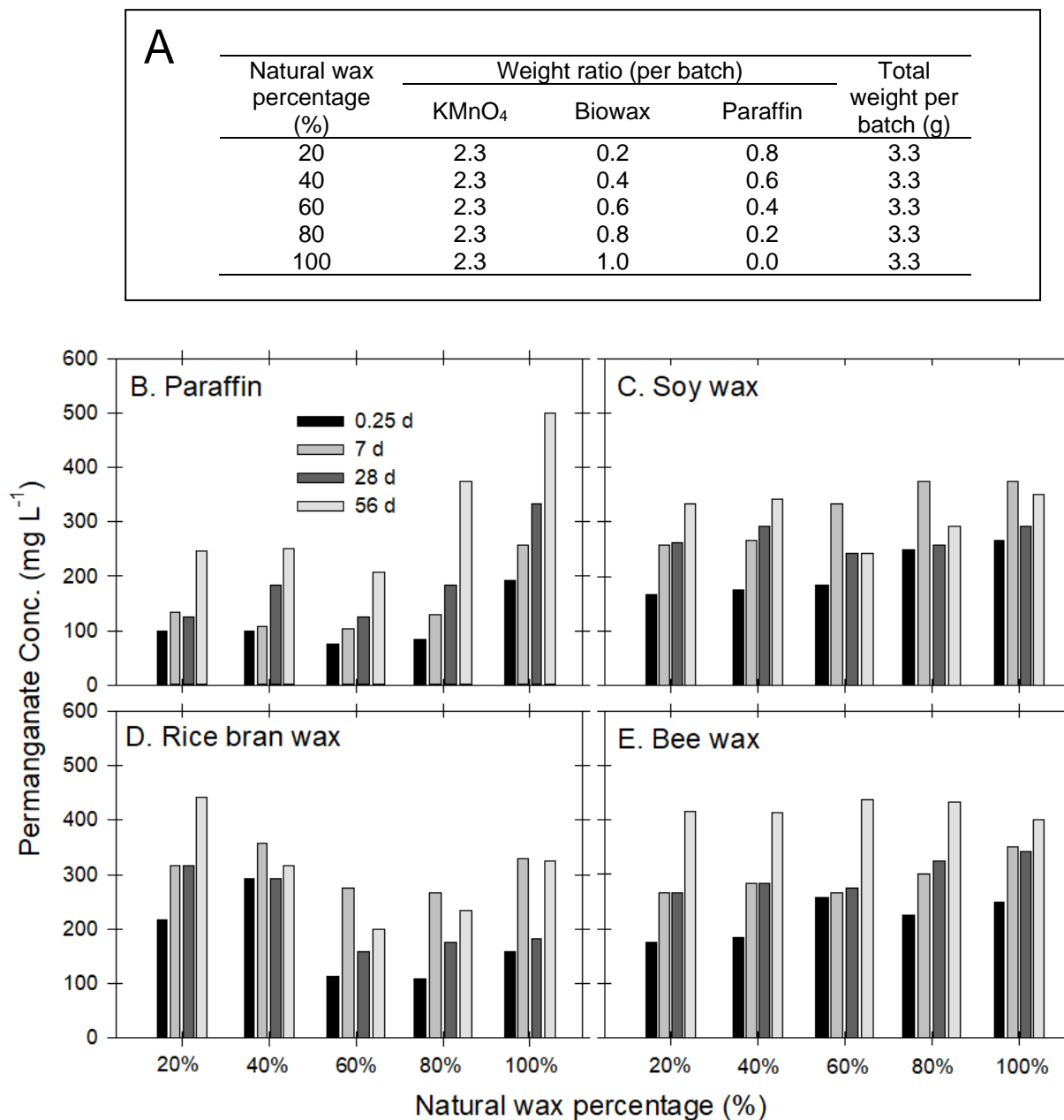
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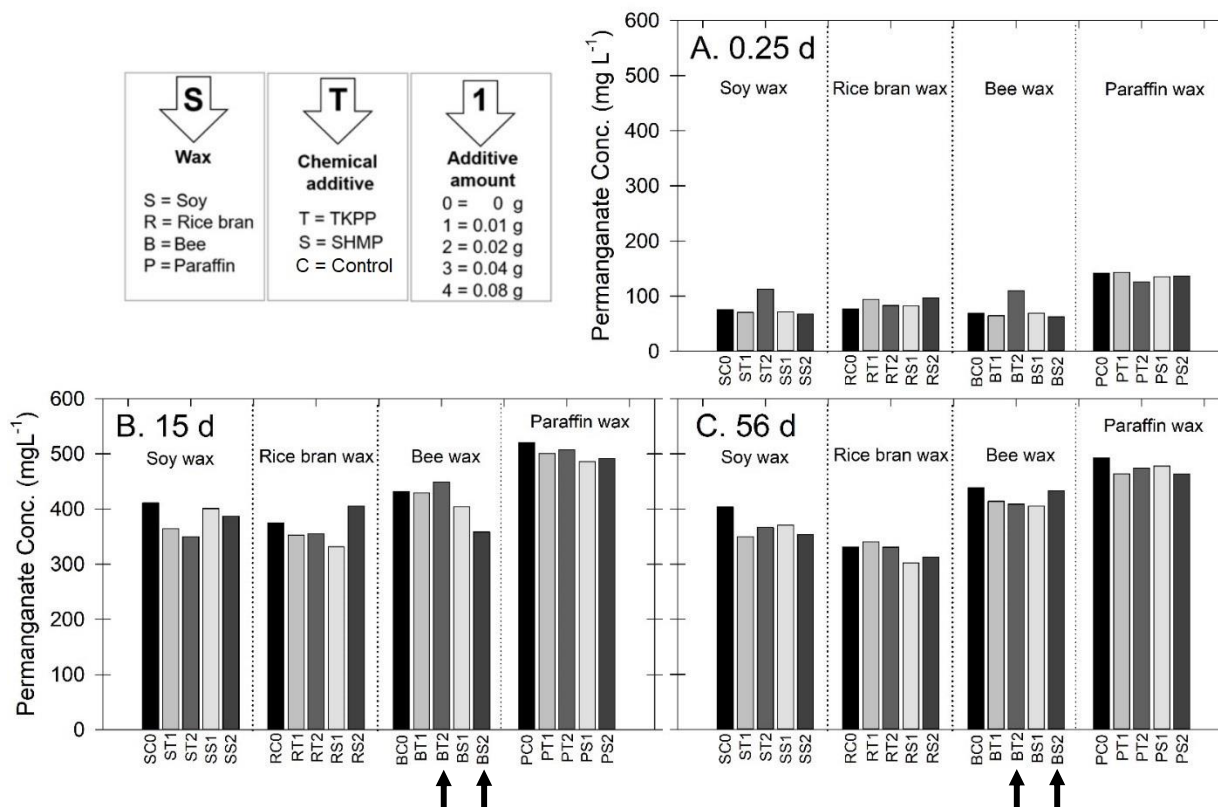
**Figure S1.** Temporal changes in antibiotic relative concentrations (A; Sulfadimethoxine, SDM 161.12 μM, B; Ormetoprim, OMP 36.45 μM, C; Trimethoprim, TMP 34.44 μM) following treatment with varying MnO<sub>4</sub><sup>-</sup> concentrations (0.315 to 5.033 mM for SDM or 0.189 to 27.181 mM for OMP or TMP) and loss of initial concentrations of antibiotics (D; SDM 16.11 to 161.12 μM, E; OMP 4.56 to 72.91 μM, F; TMP 4.56 to 72.91 μM) when treated with MnO<sub>4</sub><sup>-</sup> at 1.133 mM.



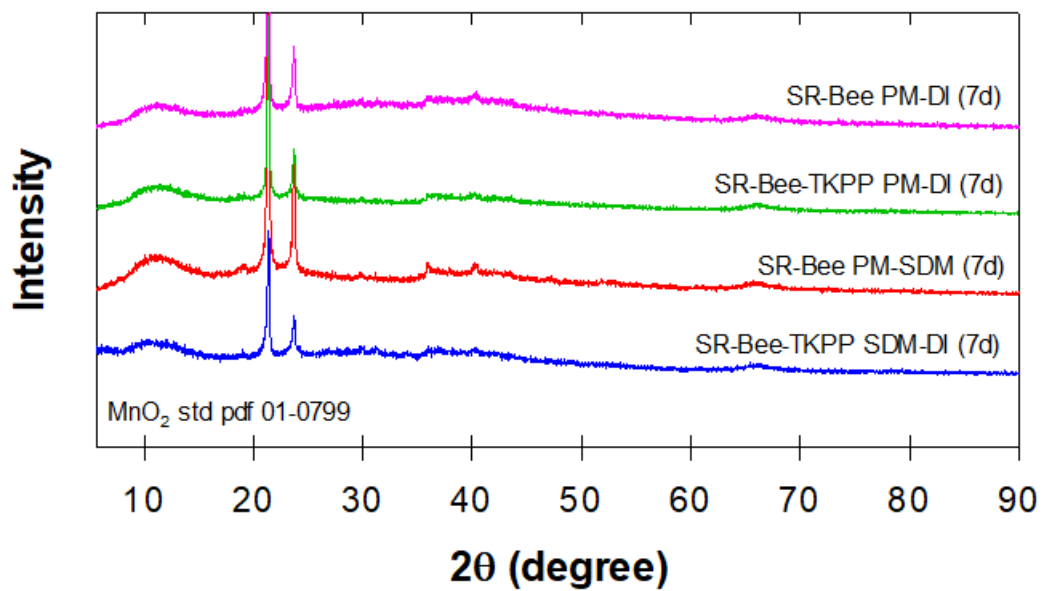
**Figure S2.** Observed kinetic rate constant ( $k_{obs}$ ) of each antibiotic degradation (A; Sul-fadimethoxine, SDM, B; Ormetoprim, OMP, or C; Trimethoprim, TMP) with presence of different synergetic antibiotics (as individual, SDM+OMP, or SDM+TMP) following treatment with  $MnO_4^-$  at  $180\text{ mg L}^{-1}$ .



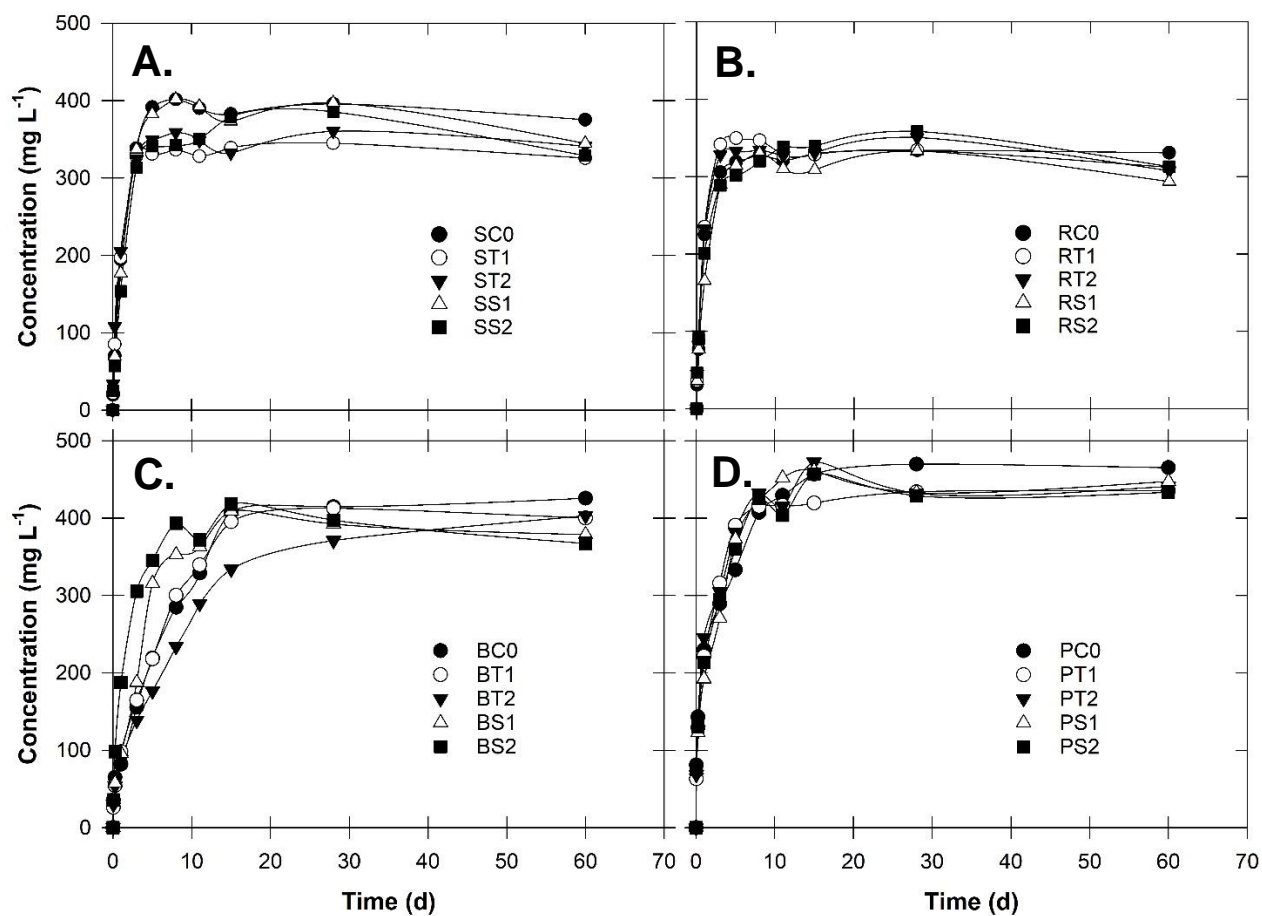
**Figure S3.** Permanganate release concentration [Mixing ratio: Set A]; **(A)** Weight composition of SR-MnO<sub>4</sub> per batch (3.3 g) at different natural wax percentages. **(B–E)** Permanganate concentrations of each type of SR-MnO<sub>4</sub> at different natural wax percentages (20–100%) and at different timelines (0.25, 7, 28, and 56d). Graphs **(B–E)** represent different types of natural wax in the mixture: **(B)** synthetic paraffin, **(C)** soy wax, **(D)** rice bran wax, and **(E)** beeswax.



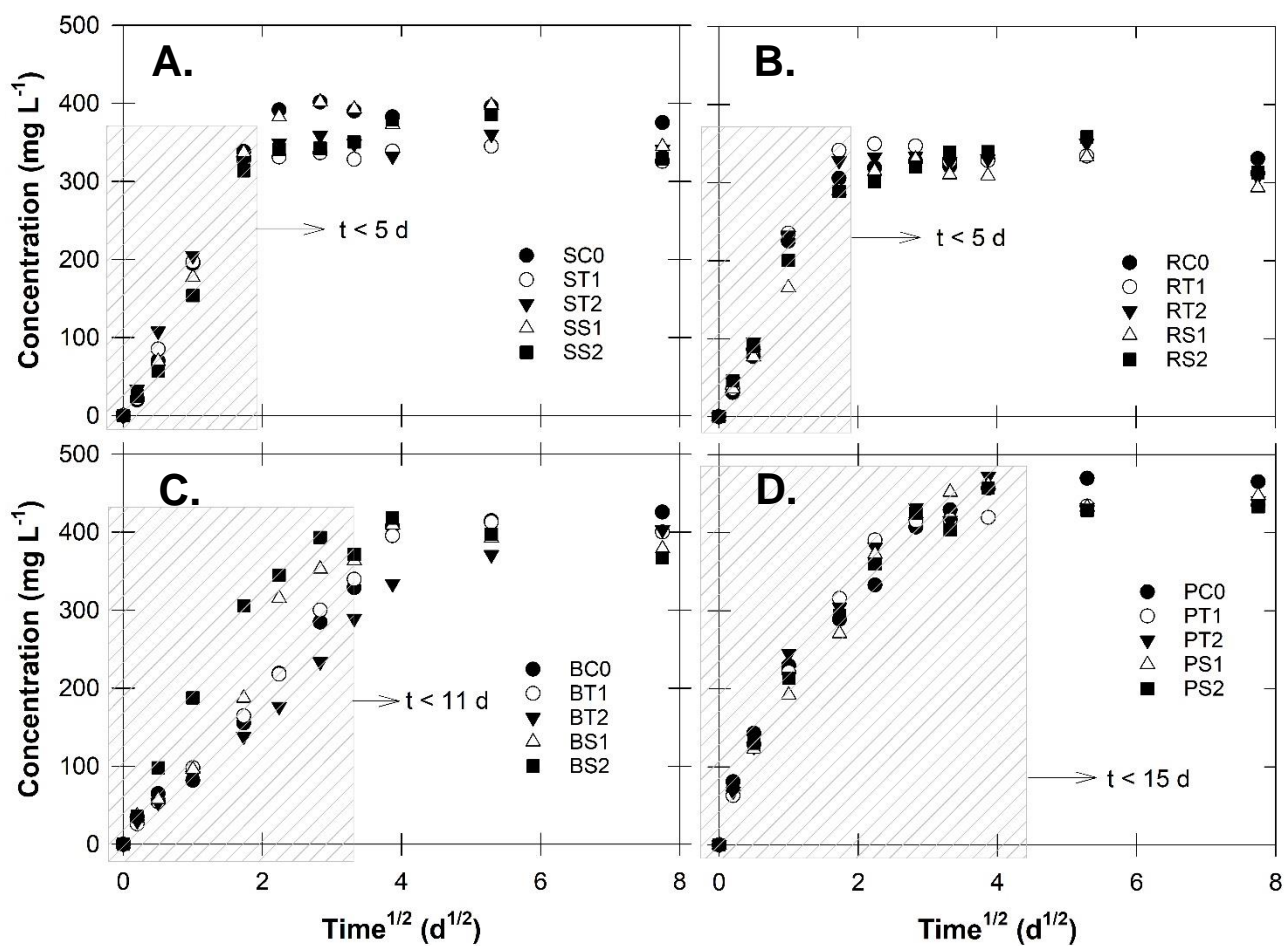
**Figure S4.** Permanganate concentrations of each type of SR-MnO<sub>4</sub><sup>-</sup> with different formulations of natural wax, synthetic paraffin, and chemical addition (TKPP or SHMP) for different timelines: (A) 0.25 day, (B) 15 day, and (C) 56 day.



**Figure S5.** X-ray diffraction analysis of different types of slow-release beewax (SRB) before and after soaking in SDM solution for 7 d. (Bee: beewax, PM: permanganate, TKPP: Tetrapotassium pyrophosphate, SDM: Sulfadimethoxine, DI: distilled water)

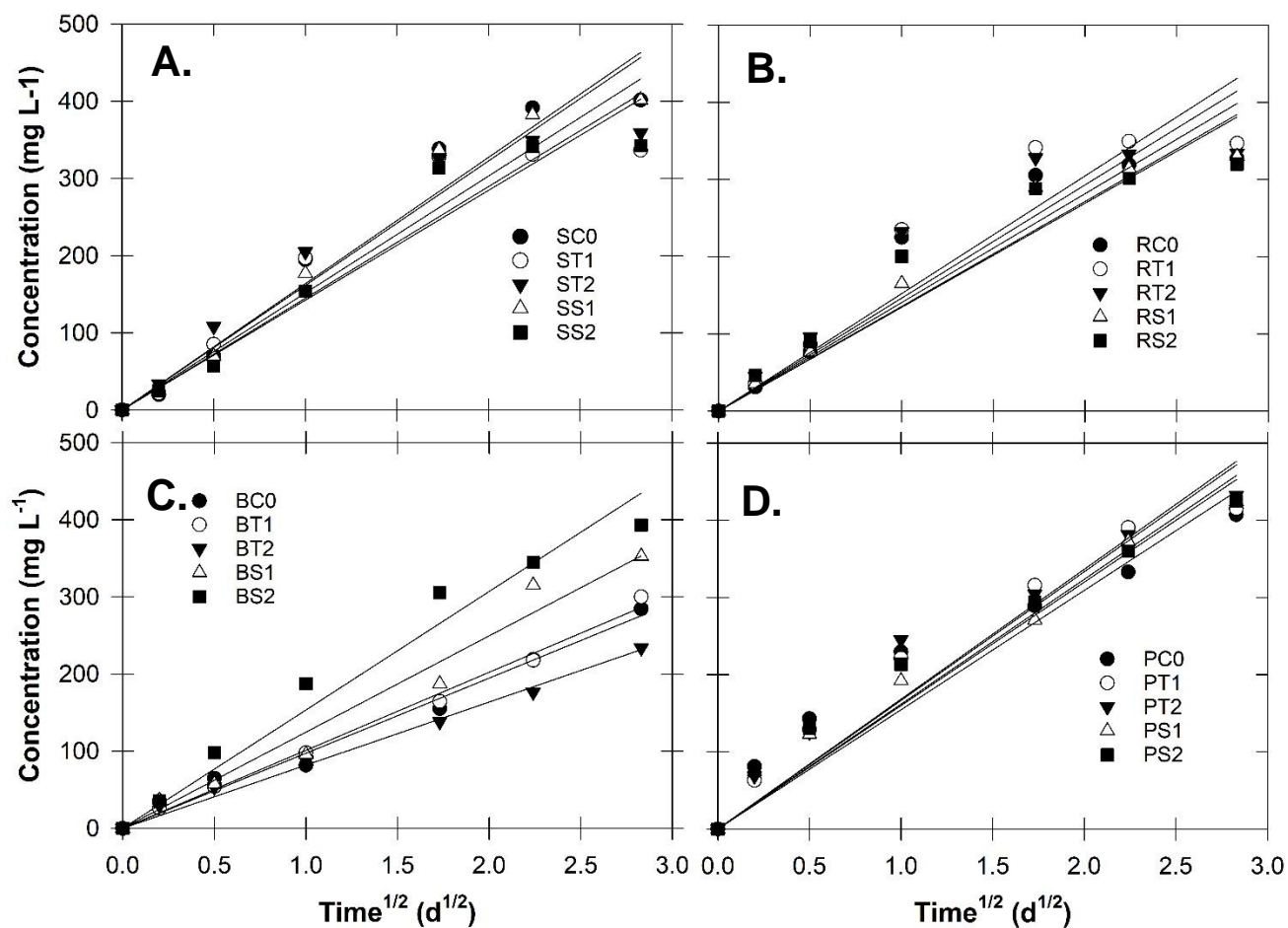


**Figure S6.** Temporal changes in  $MnO_4^-$  concentration from each type of SR- $MnO_4^-$ : (A) soy wax; (B) rice bran wax (C) beeswax; and (D) paraffin. The explanation of slow-release sample abbreviation (e.g., SC0, ST1, etc.) is provided in the Figure S4.

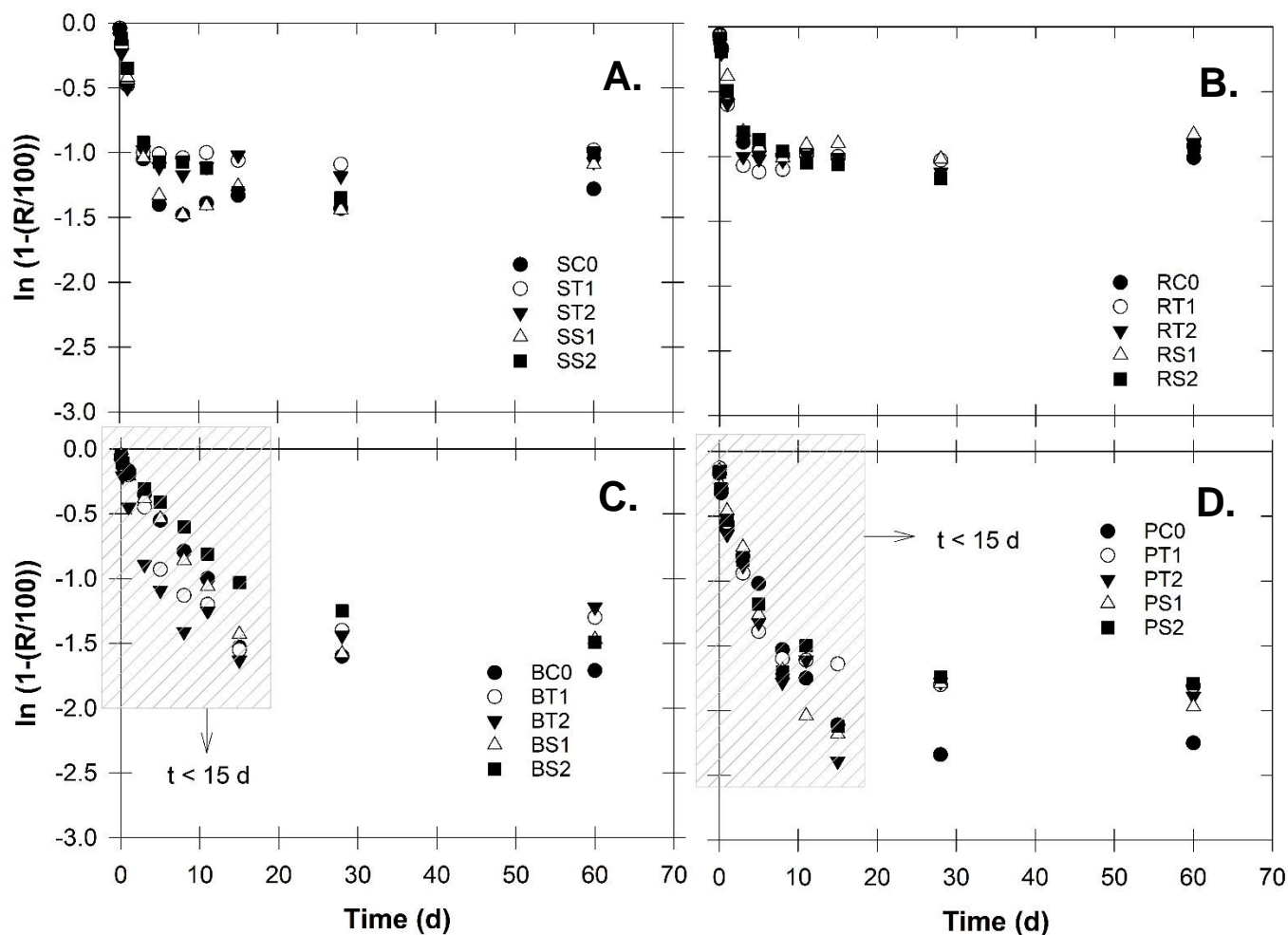


**Figure S7.** Release pattern of MnO<sub>4</sub><sup>-</sup> concentration of each type of SR-MnO<sub>4</sub><sup>-</sup> plotted for Higuchi releasing model with all selected data (t < 60 d): (A) soy wax; (B) rice bran wax (C) beeswax; and (D) paraffin. Hatched boxes represented range of time that data may be fitted in linear regression. The explanation of slow-release sample abbreviation (e.g., SC0, ST1, etc.) is provided in the Figure S4.

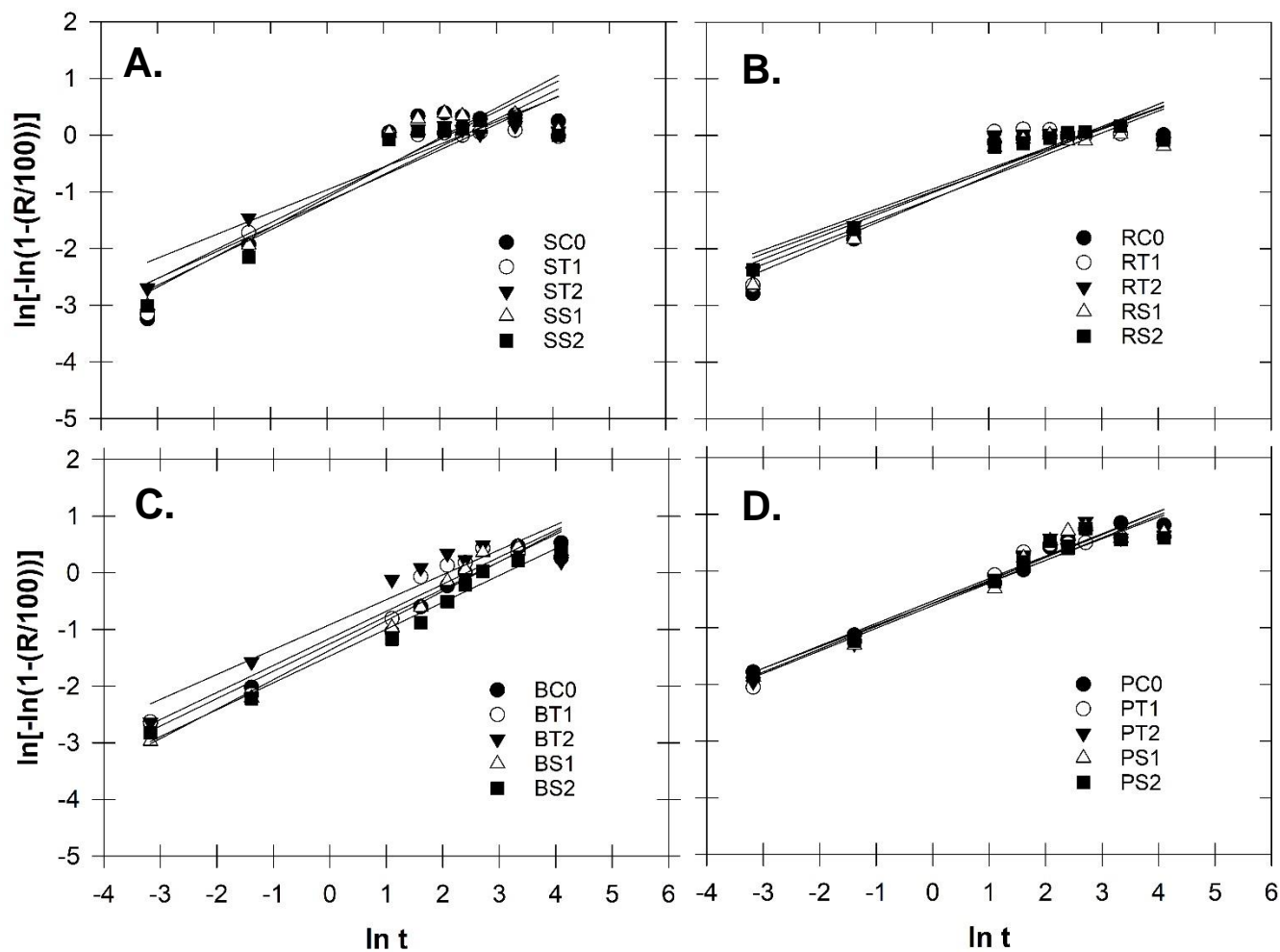




**Figure S8.** Linear regression of each type of SR-MnO<sub>2</sub> using Higuchi releasing model with selected data from  $t < 8$  d: **(A)** soy wax; **(B)** rice bran wax **(C)** beeswax; and **(D)** paraffin. The explanation of slow-release sample abbreviation (e.g., SC0, ST1, etc.) is provided in the Figure S4.



**Figure S9.** Release pattern of  $\text{MnO}_4^-$  concentration of each type of SR- $\text{MnO}_4^-$  plotted for Noyes-Whitney releasing model: **(A)** soy wax; **(B)** rice bran wax **(C)** beeswax; and **(D)** paraffin. Hatched boxes represent range of time that data may be fitted in linear regression. The explanation of slow-release sample abbreviation (e.g., SC0, ST1, etc.) is provided in the Figure S4.



**Figure S10.** Linear regression of each type of slow-release permanganate using Weibul releasing model: (A) soy wax; (B) rice bran wax (C) beeswax; and (D) paraffin. The explanation of slow-release sample abbreviation (e.g., SC0, ST1, etc.) is provided in the Figure S4.



Table S1. Physicochemical characteristics of antibiotics.

Antimicrobial; Abbreviation (CAS Number)	Chemical Structure	Molecular Formula	MW (g/mol)	Water Solubility at 25°C (mg/L)	Henry's Con- stant (atm·m <sup>3</sup> ·mol <sup>-1</sup> )	Log K <sub>ow</sub>	pK <sub>a1</sub> (pK <sub>a2</sub> )	Wavelength (nm)
Sulfadimethox- ine;SDM <sup>a</sup> (122-11-2)		C <sub>12</sub> H <sub>14</sub> N <sub>4</sub> O <sub>4</sub> S	310.33	343	1.30 × 10 <sup>-14</sup>	1.63	2.4 (6.0)	270 <sup>e</sup>
Ormetoprim; OMP <sup>a</sup> (6981-18-6)		C <sub>14</sub> H <sub>18</sub> N <sub>4</sub> O <sub>2</sub>	274.32	1,540	4.45 × 10 <sup>-13</sup>	1.23	7	270 <sup>e</sup>
Trimethoprim; TMP <sup>b</sup> (738-70-5)		C <sub>14</sub> H <sub>18</sub> N <sub>4</sub> O <sub>3</sub>	290.32	400	1.32 × 10 <sup>-6</sup> (Vapor pres- sure) <sup>c</sup>	0.91	3.2 (6.7) <sup>d</sup>	230 <sup>e</sup>

References: <sup>a</sup>Sanders et al. [1]; <sup>b</sup>Straub, [2]; <sup>c</sup>Gros et al. [3]; <sup>d</sup>Qiang and Adams [4]; <sup>e</sup>Samuelsen et al. [5]

Table S2. Properties of TKPP and SHMP (Chokejaroenrat et al. [6])

Chemical	Molecular Structure	Molecular Formula	Description	M.W. (g mol <sup>-1</sup> )	Density (g cm <sup>-3</sup> )	Solubility (mg L <sup>-1</sup> )	Manufacturer
Sodium hex- ameta-phos- phate(SHMP)		(NaPO <sub>3</sub> ) <sub>6</sub>	Dispersing agent	611.77	2.484	Soluble	Sigma Aldrich
Tetrapotassium pyrophosphate (TKPP)		K <sub>4</sub> P <sub>2</sub> O <sub>7</sub>	Dispersing agent	330.34	-	Highly sol- uble	Carus Corpo- ration



Table S3. Physicochemical properties of aquaculture discharge wastewater.

Parameter	Unit	Value
pH	-	7.91
Turbidity	NTU	84
Conductivity	$\mu\text{S cm}^{-1}$	141
Total kjeldahl nitrogen (TKN)	$\text{mgL}^{-1}$	7.68
Nitrate	$\text{mgL}^{-1}$	0.1
Sulfate	$\text{mgL}^{-1}$	6.03
Chloride	$\text{mgL}^{-1}$	110.83
Phosphate	$\text{mg-P L}^{-1}$	7.45
Salinity	PSU	0.7
Total phosphorus	$\text{mgL}^{-1}$	0.46
Ammonia-nitrogen (mg/l)	$\text{mgL}^{-1}$	0.89
Dissolved oxygen	$\text{mgL}^{-1}$	8.8
Total alkalinity	$\text{mgL}^{-1}$ as $\text{CaCO}_3$	31.6
Total suspended solid	$\text{mgL}^{-1}$	282
Total dissolve solid (TDS)	$\text{mgL}^{-1}$	147
Total organic carbon (TOC)	$\text{mg-C L}^{-1}$	62.4
Biological oxygen demand (BOD)	$\text{mgL}^{-1}$	22.6
Chemical oxygen demand (COD)	$\text{mgL}^{-1}$	225



**Table S4.** Weight composition of SR-MnO<sub>4</sub><sup>-</sup> with chemical addition (Tetrapotassium pyrophosphate, TKPP, or Sodium hexametaphosphate, SHMP) per SR (0.75 g) [Mixing ratio: Set B].

SR-MnO <sub>4</sub> <sup>-</sup>	Weight ratio (per slow-release) (g)				
	KMnO <sub>4</sub>	Biowax	Paraffin	TKPP	SHMP
XC0	0.525 (70.0%)	0.135 (18.0%)	0.090 (12.0%)	-	-
XT1	0.525 (70.0%)	0.125 (16.7%)	0.090 (12.0%)	0.010 (0.0%)	-
XT2	0.525 (70.0%)	0.115 (15.3%)	0.090 (12.0%)	0.020 (1.3%)	-
XT3	0.525 (70.0%)	0.095 (12.7%)	0.090 (12.0%)	0.040 (2.7%)	-
XT4	0.525 (70.0%)	0.055 (7.3%)	0.090 (12.0%)	0.080 (5.3%)	-
XS1	0.525 (70.0%)	0.135 (18.0%)	0.090 (12.0%)	-	0.010 (0.0%)
XS2	0.525 (70.0%)	0.125 (16.7%)	0.090 (12.0%)	-	0.020 (1.3%)
XS3	0.525 (70.0%)	0.115 (15.3%)	0.090 (12.0%)	-	0.040 (2.7%)
XS4	0.525 (70.0%)	0.095 (12.7%)	0.090 (12.0%)	-	0.080 (5.3%)

**Remark**

1. First letter of each slow-release PM (X) represents the type of wax (P = paraffin; S = soy wax; R = rice bran, wax; B = Beeswax)
2. Percentages in parentheses were calculated based on amount of component in one slow-release PM (0.75 g).



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## Reference

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- [6] Chokejaroenrat, C.; Comfort, S.; Sakulthaew, C.; Dvorak, B. Improving the treatment of non-aqueous phase TCE in low permeability zones with permanganate. *J. Hazard. Mater.* **2014**, *268*, 177-184, https://doi.org/10.1016/j.jhazmat.2014.01.007.