## STUDIES ON PHOTOCATALYTIC ACTIVITY OF ONE DIMENSIONAL COORDINATION POLYMERS BASED ON SCHIFF BASE LIGAND

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In recent decades, design and synthesis of new materials with photocatalytic properties draw attention in scientific research. Water pollution caused by the presence of non-biodegradable dyes, widely used in the textile and paper industry, has expanded with the increase of industrialization and population [1].

Here, we report the investigation of photocatalytic properties of cadmium(II) and copper(II) coordination polymers:  ${}^{1}_{\infty}$ [Cd<sub>3</sub>L(CH<sub>3</sub>COO)<sub>4</sub>]·H<sub>2</sub>O (**CP1**),  ${}^{1}_{\infty}$ [Cd<sub>2</sub>L(NO<sub>3</sub>)<sub>2</sub>]·CHCl<sub>3</sub> (**CP2**)  ${}^{1}_{\infty}$ [Cd<sub>5</sub>L<sub>2</sub>(CH<sub>3</sub>COO)<sub>6</sub>] (**CP3**),  ${}^{1}_{\infty}$ [Cu<sub>3</sub>L<sub>2</sub>(NO<sub>3</sub>)](NO<sub>3</sub>)·2MeOH·2H<sub>2</sub>O (**CP4**),  ${}^{1}_{\infty}$ [Cu<sub>3</sub>L<sub>2</sub>(N<sub>3</sub>)](CH<sub>3</sub>COO) (**CP5**) and  ${}^{1}_{\infty}$ [Cu<sub>3</sub>L<sub>2</sub>(H<sub>2</sub>O)](ClO<sub>4</sub>)<sub>2</sub> (**6**) based on the Schiff base H<sub>2</sub>L (N,N'-bis[(2-hydroxybenzilideneamino)propyl]piperazine) [2] for degradation of industrial dyes. Congo Red (CR), Acid Orange 7 (AO7), Methyl Orange (MO), and Direct Green 6 (DG6) dyes were selected as potential colored pollutants. Photocatalytic experiments were carried out under visible irradiation at 546 nm, and room temperature using 1 g/L compound (**CP1** ÷ **CP6**).

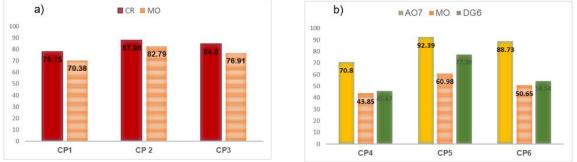


Figure 1. Degradation efficiency of investigated dyes using (a) Cd<sup>II</sup>-CP and (b) Cu<sup>II</sup>-CP

Under irradiation with visible light, catalytic efficiency for the degradation of investigated dyes increased in order: CP1 < CP3 < CP2 using  $Cd^{II}$ -CP and CP4 < CP6 < CP5 for  $Cu^{II}$ -CP. The highest photocatalytic activities were obtained using CP2: 87.98% for CR, and 82.79% for MO, and CP5: 92.39% for AO7, 40.98% for MO, and 77.38% for DG6.

The kinetic of dyes photodegradation was investigated using the Langmuir-Hinshelwood model and the obtained results showed that the degradation process followed a pseudo-first-order kinetic.

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## References

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