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Research on imbibition effect of surfactant fracturing fluid in offshore reservoirs with low permeability and high temperature

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Surfactant imbibition can effectively improve the development effect of fracturing technique. In order to enhance oil recovery, a temperature resistant surfactant was selected using interfacial tension and contact angle testers based on the offshore reservoir conditions of high temperature and low permeability. Experiments on spontaneous and forced imbibition with artificial columnar cores were performed to optimize parameters of imbibition processes during fracturing in high temperature and low permeability reservoirs. Results show that among the three surfactants, amphoteric Gemini surfactant had a stable molecular structure with non-compressible hydrophobic groups which significantly reduced oil-water interfacial tension, enhanced rock wettability, and demonstrated good resistance to high temperatures. The maximum recovery increase during amphoteric Gemini surfactant imbibition is 16.22%. The higher the core permeability, the higher the core inside pressure, and the longer the well shut-in time, the greater the forced imbibition effect. A proper well shut-in time and injection volume should be considered when using surfactant solution as fracturing liquid.

KEYWORDS

surfactants, imbibition effect, offshore oilfields, high temperature, low permeability, physical simulation

1 Introduction

China's demand for oil and gas resources increases annually and offshore oilfields occupy an important position in the oil and gas supply field. Continuous development in China's offshore oilfields has resulted in a greater proportion of crude oil reserves in low-permeability reservoirs, as well as increasing development difficulties (Cao et al., 2022; Han et al., 2022; Lin et al., 2022). Block Z of the Bohai Oilfield is characterized by high-temperature and low-permeability. Its oil has low density, low viscosity, low sulfur content, high wax content, and a high solidification point (Wei and Liu, 2019; Zhong et al., 2020; Xu et al., 2021). Besides, this reservoir is a structural layered reservoir with a complex oil-water relationship influenced by fault cutting, along with poor inter-well connectivity (Abbassi et al., 2022). During the initial stage of oil production, crude oil has good properties, a strong self-jet ability, a high production ability, and a long period of low water-cut (Dontsov, 2022;

Ma et al., 2022). But, when a well encounters water, its production capacity falls off significantly and quickly. A fracturing technique is an effective way to decrease the seepage resistance (Tan et al., 2023; Huang et al., 2023a), and then increase the production of low permeability oil wells (Tan et al., 2021; Huang et al., 2023b). However, hundreds of cubic meters of guar gum or gel fracturing fluid can not only supply the formation energy, but also reduce the matrix permeability near the fracture due to its wall building property. Therefore, the current offshore fracturing technique has limited oil-increasing effects because of reservoir physical properties. The method to use surfactant solution as fracturing liquid to enhance imbibition function has shown to be an effective strategy to further improve oil recovery.

Due to the effectiveness of imbibition of surfactant in the fracturing fluid to develop a low permeability reservoir, petroleum technology professionals have studied spontaneous imbibition extensively under the condition of unconnected wells. Wang et al. developed a static imbibition mathematical model for the numerical simulation of spontaneous imbibition, which considered the influence of imbibition driving forces such as capillary force and gravity on imbibition oil recovery (Wang et al., 2022; Marzhan et al., 2023). Through spontaneous imbibition experiments, Xu et al. investigated the mechanism of core imbibition oil displacement under different permeabilities, and concluded that better pore structure can enhance spontaneous imbibition (Xu and Ma, 2015). Li et al. used numerical modeling to explore the influence of fracturing fluid interfacial tension on spontaneous imbibition in low permeability reservoirs (Li et al., 2018). Based on low field nuclear magnetic resonance testing, Wang et al. carried out on-line scanning of core nuclear magnetic resonance under five boundary conditions. Zhu et al. studied the effects of different hydrophilic and hydrophobic structures on spontaneous imbibition in high temperature, high salinity and low permeability reservoirs (Zhu et al., 2021). Cai et al. analyzed the discriminant parameters of spontaneous imbibition mechanism including interfacial tension and wetting angle (Cai and Yu, 2012).

Currently, spontaneous imbibition is the main experimental technique used to explore imbibition in low-permeability cores, and the experiment temperature is often not more than 100°C. There are few studies considering external fluid pressure on the imbibition effect in high-temperature and low-permeability reservoirs (Gushchin et al., 2018; Bai et al., 2021; Liu et al., 2023). Therefore, in view of the characteristics of high temperature, insufficient productivity and disconnection between wells in Block Z of Bohai Oilfield, this paper studied the imbibition development effect of high temperature resistant surfactant through spontaneous and forced imbibition experiments. The type and concentration of imbibition agent suitable for target reservoir were optimized by static imbibition and forced imbibition physical simulation experiments. The influence mechanism of factors such as shut-in pressure, shut-in time and core permeability on the forced imbibition effect. The related study can provide theoretical basis for the practical construction in Block Z.

2 Experimental section

2.1 Experimental materials

Experimental agents include: lauryl alcohol sodium sulfate (anionic), effective content of 30%; lauric acid (anionic), effective

content of 40%; Gemini surfactant (amphoteric), content of 40%; heavy water (D₂O), deuterium element could effectively shield nuclear magnetic signal, effective content of 99.9%.

The water used in spontaneous imbibition and forced imbibition is simulated injection water, and is prepared by using distilled water and heavy water, respectively. The ionic composition of injection water is shown in Table 1. The experimental oil was prepared by mixing degassed crude oil and light hydrocarbon in Bohai oilfield, with a viscosity of 5.0 mPa s at 138°C.

In this study, the effects of the surfactant imbibition ratio on spontaneous imbibition and forced imbibition were investigated using artificial hydrophilic columnar cores cemented by quartz sands (Ø2.5 × 5.0 cm) (Han et al., 2016; Xie et al., 2018). The permeabilities of these cores were approximately 20mD, 60mD, respectively, and their pore structures and permeabilities were comparable to reservoir cores in the Bohai oilfields. The procedures to prepare the artificial cores can be found in the Chinese patent “Quartz sand epoxy resin cementation heterogeneity model making method”.

2.2 Experimental instruments

The IFT between the oil and water was measured at 138 °C using a TX500HP high temperature interfacial tension meter. The contact angle was measured at 138 °C with an OCA20 optical contact angle measuring instrument using a sessile drop method. Additionally, spontaneous imbibition recovery was measured using an imbibition cell made of high-temperature and pressure-resistant glass (as shown in Figure 1). The smallest division values of this imbibition cell were 0.005 mL. The forced imbibition test equipment mainly includes ISCO injection pump, confining pressure pump, intermediate vessels, pressure sensor and core online displacement nuclear magnetic resonance analyzer (MacroMr12-150H-I, Niumag Analytical Instrument, Suzhou, China), with the experimental temperature of 138 °C. The nuclear magnetic pictures of T₂ spectrum before and after forced imbibition were compared. Changes in T₂ spectrum are used to determine the relative content change of crude oil at different pores scales in the core. The experimental equipment and procedures are shown in Figure 2.

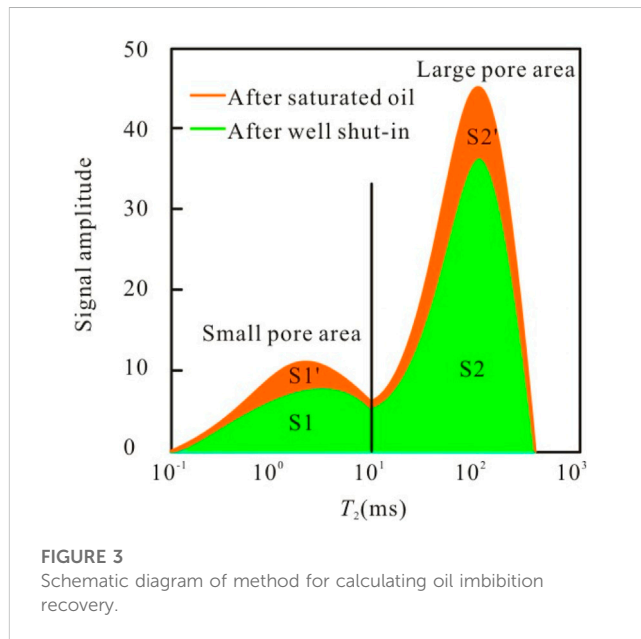
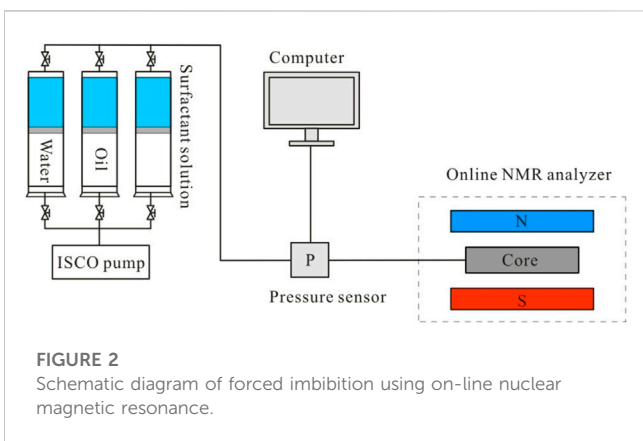
2.3 Experimental instruments

2.3.1 Spontaneous imbibition

- 1) The dried cores were weighed before and after being saturated with simulated injection water prepared by distilled water. The pore volume was calculated through measuring the weight difference of the core sample.
- 2) Each core was placed into the core holder, saturated with crude oil, and aged with crude oil at 138 °C for 24 h to prepare it for use.
- 3) For spontaneous imbibition, an oil-aged core was placed inside an imbibition cell at 138 °C. The imbibition cell was filled with a surfactant solution to a desired height.
- 4) The volume of produced oil was monitored and recorded as often as appropriate. The imbibition recovery was calculated based on the volume of produced oil.

TABLE 1 Ion composition of injection water (mg/L).

Cations			Anions				Total dissolved solids (TDS)
Ca ²⁺	Mg ²⁺	Na ⁺	CO ₃ ²⁻	HCO ₃ ⁻	Cl ⁻	SO ₄ ²⁻	
193.12	10.71	3196.54	0	846.51	4879.12	4.30	9130.30



- 4) The surfactant solution was injected at 0.2 mL/min into the core holder, the outlet was shut at the same time. When the injection pressure reaches the set pressure, close the inlet for a designed shut-in time to simulate forced imbibition during well shut;
- 5) Outlet valve was opened until no liquid flows out after the shut-in time reached the designed time, the T₂ spectrum after forced imbibition was tested.
- 6) The forced imbibition effect of core was analyzed by comparing the change of T₂ spectrum in different size pore-throat.

2.3.2 Forced imbibition

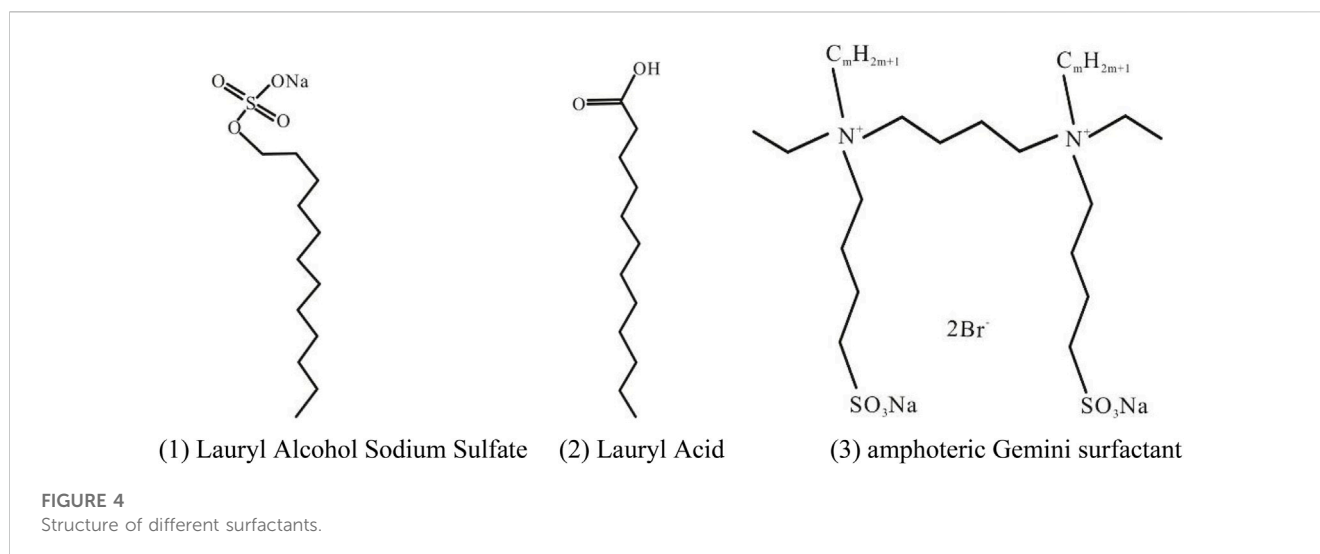
The specific experimental procedures of forced imbibition are as follows:

- 1) The dried core was weighed before and after being saturated with simulated water prepared by heavy water. The pore volume could be calculated by measuring the weight difference.
- 2) Core was placed into a core holder and was saturated with oil at reservoir temperature until it comes out at the outlet. The crude oil injection speed was 0.02 mL/min. After the oil saturation process, the core was aged with the crude oil at 138 °C for 24 h to prepare it for use. The core was placed into the core holder of NMR analyzer, and then the initial T₂ spectrum was tested;
- 3) The experimental surfactant solution prepared by heavy water was heated to the temperature of 80 °C in advance.

2.3.3 Experimental theory

During NMR testing, the signal amplitude of transverse relaxation time T₂ spectrum reflects the relative saturation of residual oil in pores of different sizes. Additionally, the decreasing amplitude of the T₂ spectrum at various time intervals is used to quantify crude oil recovery. . . Regions with T₂ relaxation times less than 10 m are defined as small pores, and regions with T₂ relaxation times greater than 10 m are defined as large pores (Xie et al., 2022). The T₂ spectrum of the initial stage and stage after well shut was entered into a computer image recognition program. The relative movable oil of different size pores was obtained by comparing the area of T₂ spectrum between initial stage and stage after well shutting. According to Figure 3, the oil recovery can be calculated using following formula:

$$R_1 = \frac{S'_1}{(S_1 + S'_1)} \tag{1}$$

**TABLE 2** Interfacial tension test results (mN/m).

Surfactant type	Concentration (%)		
	0.1	0.2	0.3
Sodium lauryl alcohol sulfate	2.27×10^{-1}	1.89×10^{-2}	1.15×10^{-3}
Lauric acid	1.51×10^{-3}	1.33×10^{-3}	1.19×10^{-3}
Amphoteric Gemini surfactant	1.20×10^{-3}	9.70×10^{-4}	6.80×10^{-4}

$$R_2 = \frac{S_2'}{(S_2 + S_2')} \quad (2)$$

$$R = \frac{(S_1' + S_2')}{(S_1 + S_1' + S_2 + S_2')} \quad (3)$$

where R_1 , R_2 , R refers to the oil recovery of small pore, large pore, and total pore. S_1 , S_1' refers to the final area of T_2 spectrum and differential area between beginning and ending time in small pore. S_2 , S_2' refers to the final area of T_2 spectrum and differential area between beginning and ending time in small pore.

3 Results and discussion

3.1 Forced imbibition system performance evaluation

3.1.1 Interfacial tension

Three surfactants at concentrations of 0.1%, 0.2% and 0.3% were prepared using simulated injection water at 138°C reservoir temperature. Surfactant molecular structures and interfacial tension test results are shown in Figure 4 and Table 2 respectively.

As shown in Figure 4 and Table 2, the concentrations of the surfactant solutions and the interfacial tension are negatively correlated. At a concentration of 0.2%, the interfacial tension follows the trend: sodium lauryl alcohol sulfate > lauric acid > amphoteric Gemini surfactant. A greater oil-water interface layer

results from a change in the structure of diffusion electric double layers due to high temperature, which also causes the hydrophobic group of chain structures to curl. As a result of the surfactants' low temperature tolerance, the interfacial tension rises. In contrast with the previous two surfactants, the amphoteric surfactant has a higher degree of hydrophobic chain branching and a more stable bond energy (C-C bond 332 kJ/mol). It also has a temperature-resistant sulfonic acid group ($-\text{SO}_3^{2-}$) in its hydrophobic chain. The surfactant's hydrophobic chain is capable of maintaining a stretch state at high temperatures due to its stable molecular structure. Additionally, the hydrophobic group is difficult to be compressed which is good for high temperature resistance.

3.1.2 Temperature resistance

The standing time of different surfactants of 0.3% concentration under the temperature of 138°C was changed, and then the interfacial tension between them and crude oil was tested by interfacial tension meter can be used in high temperature and high pressure. The test results are shown in Table 3.

As shown in Table 3, the standing time of the surfactant solutions at high temperature has an effect on the oil-water interfacial tension. After standing for 5 days, the interfacial tension of amphoteric Gemini surfactant can still obtain an oil-water interfacial tension of 10^{-3} mN/m, showing the best temperature resistance.

3.1.3 Wettability

Columnar cores with a permeability of 20 mD were cut into several thin slices of 2 mm thickness using a mechanical cutting machine, and the initial contact angle is 65.2° which means a weaker water-wet. The cores were immersed in three surfactants of 0.1%, 0.2% and 0.3% concentrations for 24 h at 138°C. The variation of contact angle on the core surface in relation to the type and concentration of surfactant is shown in Table 4.

As shown in Table 4, the contact angle of the rock surface decreases with the increase of surfactant concentration. When the surfactant concentration is 0.2%, the contact angle shows the following trend: sodium lauryl sulfate > lauric acid > amphoteric

TABLE 3 Interfacial tension after different standing time (mN/m).

Type	Time(d)			
	0	1	3	5
Sodium lauryl sulfate	1.15×10^{-3}	2.30×10^{-3}	8.80×10^{-3}	2.33×10^{-2}
Lauric acid	1.19×10^{-3}	3.15×10^{-3}	3.25×10^{-3}	3.15×10^{-3}
Amphoteric Gemini surfactant	6.80×10^{-4}	2.22×10^{-3}	2.39×10^{-3}	2.46×10^{-3}

TABLE 4 Relationship between contact angle, contact angle difference and surfactant concentration (°).

Concentration (%)		Surfactant type		
		Sodium lauryl sulfate	Lauric acid	Amphoteric gemini surfactant
0.1	Before soaking	65.03	65.88	64.64
	After soaking	60.91	59.42	56.29
	Difference	4.12	6.46	8.35
0.2	Before soaking	65.03	65.88	64.64
	After soaking	47.55	42.65	38.03
	Difference	17.48	23.23	26.61
0.3	Before soaking	65.03	65.88	64.64
	After soaking	43.73	40.66	35.66
	Difference	21.30	25.22	28.98

TABLE 5 Core spontaneous imbibition recovery.

Surfactant type	Concentration (%)	Permeability K_g (mD)	Interfacial tension (mN/m)	Oil saturation (%)	Recovery (%)
Sodium lauryl sulfate	0.1	19	2.27×10^{-1}	62.70	7.69
	0.2	20	1.89×10^{-2}	61.94	14.22
	0.3	21	1.15×10^{-3}	62.89	10.65
Lauric acid	0.1	20	1.51×10^{-3}	62.42	8.78
	0.2	22	1.33×10^{-3}	62.83	15.33
	0.3	21	1.19×10^{-3}	62.86	11.34
Amphoteric Gemini surfactant	0.1	19	1.20×10^{-3}	62.97	10.36
	0.2	22	9.70×10^{-4}	62.62	16.22
	0.3	21	6.80×10^{-4}	62.74	12.74

Gemini surfactant. Compared with the two anionic surfactants sodium lauryl sulfate and lauric acid, amphoteric Gemini surfactant can effectively improve the wettability of the core surface and make the core surface more water-wet. It is challenging to determine the imbibition effectiveness solely based on physicochemical properties since the influencing factors during the surfactant imbibition process are complex. Therefore, spontaneous imbibition experiments are used to further compare and optimize the types and concentrations of surfactants for offshore oilfields.

3.2 Optimization of technological parameters of forced imbibition system

3.2.1 Spontaneous imbibition

Columnar cores were immersed in the surfactant solution in the high temperature imbibition bottle placed in the temperature of 138°C. The results of core spontaneous imbibition recovery experiments are shown in Table 5.

In static spontaneous imbibition, amphoteric Gemini surfactant had the maximum oil recovery among the three surfactants, as

TABLE 6 Forced imbibition recovery under different core permeability.

Permeability K_g (mD)	Core inside pressure (MPa)	Shut-in time (h)	Oil saturation (%)	Recovery (%)		
				R_1	R_2	R
20	15	24	62.16	28.70	27.02	27.42
60			61.98	38.89	35.51	36.24

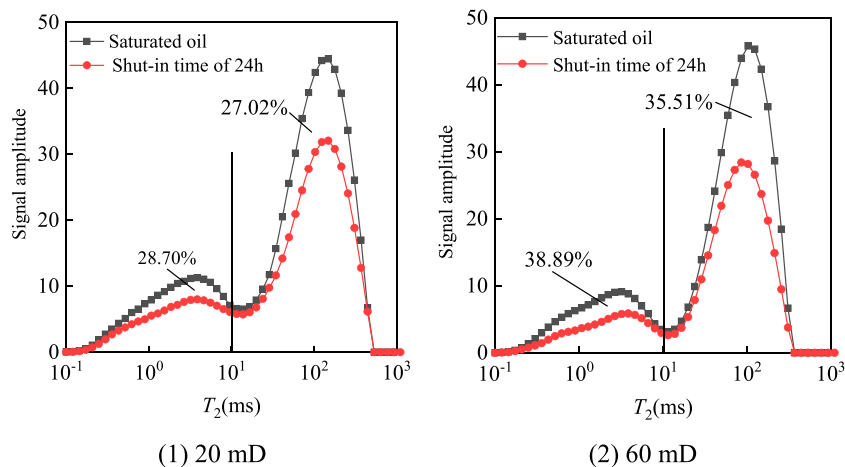


FIGURE 5 T_2 spectrum distribution under different core permeability.

indicated in Table 5. When the surfactant concentration was 0.2%, the oil recovery of the amphoteric Gemini surfactant solution was 16.22%. A mechanism study shows that improving imbibition oil recovery can be achieved by properly increasing the surfactant concentration and reducing interfacial tension. However, too low interfacial tension will limit the capillary force, which is detrimental for imbibition. In addition, when the surfactant concentration exceeds the critical micelle concentration, the effective concentration decreases and the imbibition effect deteriorates. Further analysis revealed that surfactants can change the wettability of the rock pore surface to more water-wet resulting in the reduction of the adhesion of oil phase on the rock surface. This causes an even force field to form around the oil droplets in the pore-throat, which facilitates the oil film's easier separation from the rock's surface and increases imbibition. On the other hand, surfactants have the ability to significantly lower the capillary resistance and adherence of the rock surface to oil droplets by lowering the interfacial tension between oil and water. This increases the number of capillaries, enhancing the deformability of oil droplets, improving the flow ability of the oil phase, and thus increasing the imbibition oil recovery. Based on the above results, 0.2% amphoteric Gemini surfactant is recommended for subsequent experimental studies taking into account the technical and economic factors.

3.2.2 Forced imbibition

1) Influence of core permeability

0.2% of amphoteric Gemini surfactant was prepared by simulated formation water using heavy water. The inlet was

closed when the core inside pressure reached a pressure of 15MPa, and then outlet was opened after a shut-in time of 24 h. When the outlet has no fluid to flow, the T_2 spectrum was tested. The effect of core permeability on the forced imbibition and corresponding T_2 spectrum are shown in Table 6 and Figure 5.

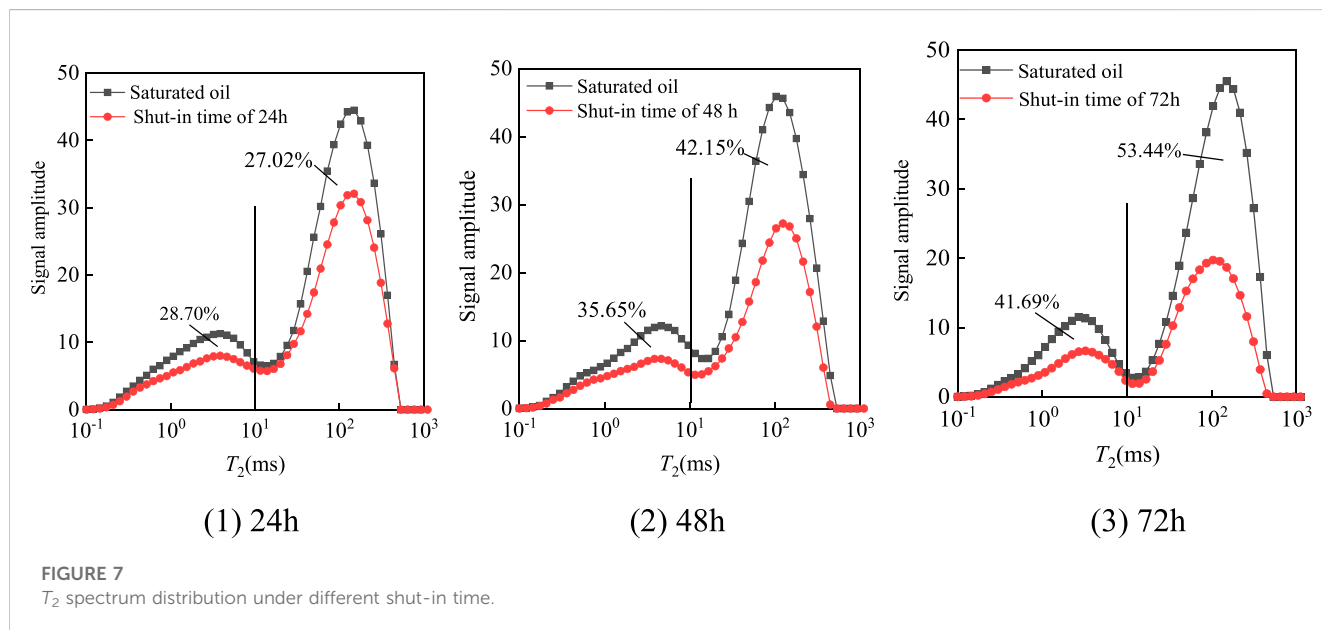
As shown in Table 6 and Figure 5, as core permeability increases, the pore throat radius increases, oil has a lower flow resistance, more oil can be produced through imbibition under the function of capillary and displacement, and the forced imbibition recovery rate increases. According to the analysis, oil in large pore areas is primarily dependent on the displacement of surfactant, so the small pore area has a higher oil recovery than large pore areas, and the oil recovery in large pore areas decreases with the decrease of core permeability. The analysis also shows that small pore areas have a stronger capillary force, which is helpful to enhance imbibition function. Because the initial saturated oil volume is larger and the large pore area has a larger pore volume, oil recovery from the large area accounts for the majority of overall oil recovery.

2) Influence of core inside pressure

The reservoir inside pressure has an obvious effect on imbibition for it will influence the crossflow between large pore area and small pore area. Different volumes of fracturing fluid will lead to a different reservoir inside pressure. The core inside pressure was changed in the experiments to study the proper injection volume of fracturing fluid. A concentration of 0.2% amphoteric Gemini surfactant was prepared by simulated formation water using

TABLE 8 Forced imbibition recovery under different shut-in time.

Shut-in time (h)	Core inside pressure (MPa)	Permeability K_g (mD)	Oil saturation (%)	Recovery (%)		
				R_1	R_2	R
24	15	21	62.16	28.70	27.02	27.42
48		22	61.84	35.65	42.15	40.64
72		21	62.23	41.69	53.44	50.88



small and large pore areas. The oil recovery increasing amplitude of the big pore area is greater than that of the small pore area because the oil produced by displacement is more than the oil produced by imbibition. Therefore, a proper shut-in time should be considered when designing a well shut-in time.

4 Conclusion

The surfactant used in fracturing liquid can obviously improve the effect of increasing oil in offshore oilfields with low permeability and high temperature, the surfactant type, construction, spontaneous imbibition effect and forced imbibition effect should be considered at the same time.

- 1) As the concentration of the three surfactant increases, the IFT decreases, and the water-wet ability becomes stronger. Amphoteric Gemini surfactant has superior temperature resistance compared to sodium lauryl alcohol sulphate and lauric acid because of its stable molecular structure and the difficulty compressing its hydrophobic group.
- 2) In spontaneous imbibition, it was found that amphoteric Gemini surfactant showed by far the highest oil recovery of the three surfactants tested. At a concentration of 0.2% of the surfactant, amphoteric Gemini can achieve spontaneous oil recovery of 16.22% of the oil.

- 3) Imbibition oil recovery can be improved by appropriately increasing the surfactant concentration and decreasing interfacial tension, however, too low interfacial tension will limit the capillary force, which is undesirable for imbibition.
- 4) The forced imbibition effect is influenced by the core permeability, the core inside pressure and the well shut-in time. The high core permeabilities, high core inside pressures, and long well shut-in times results into better forced imbibition effect. When employing surfactant solutions as a fracturing liquid, proper well shut-in time and injection volume should be taken into account.

Data availability statement

The original contributions presented in the study are included in the article/supplementary material, further inquiries can be directed to the corresponding author.

Author contributions

YL: Conceptualization, Formal Analysis, Investigation, Methodology, Writing—original draft. JZ: Project administration,

Supervision, Writing–review and editing. XL: Conceptualization, Formal Analysis, Writing–original draft, Writing–review and editing. SG: Methodology, Writing–review and editing. LZ: Writing–review and editing. XH: Writing–review and editing.

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Conflict of interest

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