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EVSC 610-101: Environmental Chemical Science

Lijie Zhang

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EVSC 610: Environmental Chemical Science-Water

Fall 2023 Course Syllabus

NJIT Academic Integrity Code: All Students should be aware that the Department of Chemistry & Environmental Science (CES) takes the University Code on Academic Integrity at NJIT very seriously and enforces it strictly. This means that there must not be any forms of plagiarism, i.e., copying of homework, class projects, or lab assignments, or any form of cheating in quizzes and exams. Under the University Code on Academic Integrity, students are obligated to report any such activities to the Instructor.

COURSE INFORMATION

Course Description: This course will cover environmental water chemistry, focusing on acid-base and metal-ligand equilibria, oxidation-reduction reactions, and chemical reaction kinetics and thermodynamics. There is some emphasis on equilibria governing inter-phase (gas-liquid, solid-liquid) chemical distribution.

Course Objectives and Student Learning Outcomes:

Learning Outcomes: By the end of this course,

- 1) Students will understand with the equilibrium reactions in close and open systems, the interaction between different phases (liquid, gas and solid), and the redox reactions in environmental water systems.
- 2) Students will learn to predict the chemical composition, pH and redox condition of an aquatic system.
- 3) Students will develop the tools to solve problems with complex chemical reactions in natural and engineer systems

Number of Credits: 3

Course-Section and Instructors

Section	Location	Instructor
R 6 pm – 8:50 pm	FMH 310	Lijie Zhang

Office Hours: Monday 5-6 pm.

Required Textbook:

Title	Water Chemistry
Author	Mark M. Benjamin
Edition	2 nd
Publisher	Waveland Press, Inc
ISBN#	978-1-4786-2308-3

Canvas: There is a course Canvas site that will include significant resources and updates of importance to this course. Please check it frequently, and also make sure to check or forward your NJIT email in order to receive important announcements.

Grading Policy: The final grade in this course will be determined as follows:

Homework	20
Quizzes	10
Group project	15
Midterm Exam	25
Final Exam	30

Your final letter grade in this course will be based on the following tentative curve:

A	90+	C	60+
B+	85+	D	50+
B	80+	F	<50
C+	70+		

Attendance: Attendance is mandatory. Students who miss class due to a valid personal or medical reason must contact the Dean of Students office with the valid excuse and get the excuse approved. The Dean of Students will then contact me and any other professors whose classes were missed. Missed worksheets, quizzes, or exams without a Dean of Students approved excuse will result in a zero grade. The instructor does not approve excuses of any sort. Students walking out of class early or low attendance early in a class may result in a roll call being taken and loss of points. When you miss a class, it will be your responsibility to find out what was discussed.

Homework: Homework assignments must be completed and submitted on Canvas on the date specified by the instructor.

Exams: Exams are **open-book** and **open-note**. There will be a midterm exam and a final exam, and the final exam will cover all materials throughout the semester. Normally **NO MAKE-UP EXAMS** are allowed during the semester. In the event that a student has a legitimate reason for missing an exam, the student should contact the Dean of Students office and present written verifiable proof of the reason for missing the exam, e.g., a doctor's note, police report, court notice, etc. clearly stating the date AND time of the mitigating problem. The student must also notify the CES Department Office/Instructor prior to the exam that the exam will be missed so that appropriate steps can be taken to make up the grade.

Group Project: 3-4 students will form a group and work together on a term paper. Students are encouraged to consult with the instructor about term paper topic. The term paper needs to be at least 5 pages (font 12, 1.15× spaced) in length excluding figures and references, though inclusion of figures and tables is recommended to facilitate the delineation. For the presentation, please plan your talk for about 15 minutes, leaving 5 minutes for questions and discussions. Both paper and presentation are weighed equally for grading.

ADDITIONAL RESOURCES

Accommodation of Disabilities: Office of **A**ccessibility **R**esources and **S**ervices (*formerly known as Disability Support Services*) offers long term and temporary accommodations for undergraduate, graduate and visiting students at NJIT.

If you need accommodations due to a disability please contact Chantonette Lyles, Associate Director at the Office of Accessibility Resources and Services at [973-596-5417](tel:973-596-5417) or via email at lyles@njit.edu. The office is located in Fenster Hall Room 260. A Letter of Accommodation Eligibility from the Office of Accessibility Resources Services office authorizing your accommodations will be required.

For further information regarding self-identification, the submission of medical documentation and additional support services provided please visit the Accessibility Resources and Services (OARS) website at:

<http://www5.njit.edu/studentsuccess/disability-support-services/>

Important Dates (See <https://www.njit.edu/registrar/fall-2023-academic-calendar>)

Date	Event
September 4	Labor Day. University Closed.
September 5	First Day of Classes
September 11	Last Day to Add/Drop a Class Last Day for 100% Refund, Full or Partial Withdrawal
September 12	W Grades Posted for Course Withdrawals
September 18	Last Day for 90% Refund, Full or Partial Withdrawal No Refund for Partial Withdrawal after this date
October 2	Last Day for 50% Refund, Full Withdrawal
October 24	Last Day for 25% Refund, Full Withdrawal
November 13	Last Day to Withdraw
November 21	Thursday Classes Meet
November 22	Friday Classes Meet
November 23	Thanksgiving Recess Begins
November 26	Thanksgiving Recess Ends
December 13	Last Day of Classes
December 14	Reading Day 1
December 15	Reading Day 2
December 17	Final Exams Begin
December 23	Final Exams End
December 25	Final Grades Due

Tentative Course Outline

Week	Date	Topic	Readings
1	Sep. 5	Introduction, general chemistry concepts, chemical reactivity	Chapter 1 Chapter 2
2	Sep. 12	Reaction Kinetics (<i>Term project group formed</i>)	Chapter 3
3	Sep. 19	Reaction thermodynamic and equilibrium	Chapter 2 Chapter 4
4	Sep. 26	Introduction to acid-base chemistry (<i>Term project Title and Objective due</i>)	Chapter 5
5	Oct. 3	Graphical solutions for acid-base chemistry	Chapter 6
6	Oct. 10	Chemical equilibrium software Titrations and buffers	Chapter 7 Chapter 8
7	Oct. 17	Titrations and buffers	Chapter 8
8	Oct. 24	Midterm Exam	
9	Oct. 31	Gas-liquid equilibrium	Chapter 9
10	Nov. 7	Chemistry of metals in aqueous solutions: solution phase reactions	Chapter 10
11	Nov. 14	Precipitation and dissolution reactions (<i>Term paper first draft due</i>)	Chapter 11
12	Nov. 28	Redox chemistry 1	Chapter 12
13	Dec. 5	Redox chemistry 2	Chapter 12
14	Dec. 12	Adsorption reactions	Chapter 13
15	Apr. 27	Student presentations (<i>Term paper due</i>)	
16	TBD	Final Exam	