

Supporting Information

Long-term degradation mechanisms in application-implemented radical thin films

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Table of contents

- 1) XPS fit parameters for the freshly evaporated films.
- 2) Time dependent core level signals.
- 3) Time dependent NEXAFS signals.
- 4) C 1s core level spectra after air exposure.
- 5) XPS fit parameters after air exposure.
- 6) Spin polarized density of states of the Blatter-pyr.
- 7) Spin polarized density of states (DOS) of the diBlatter.
- 8) Simulated N 1s core level spectra of Blatter-pyr upon adsorption of characteristic H⁺, H₂, Ar and OH species and as a function of the adsorption sites (S1).
- 9) References.

1) Fit parameters for the freshly evaporated Blatter-pyr films as in Figure 1. The fit has been performed as described in references ¹⁻⁵.

The expected stoichiometric values for the C 1s and N 1s elemental analyses are:

C-C = 24%, C-H = 62%, C-N = 14%, and N_{radical} = N_{imino} = N_{amino} = 33%

Table S1

	Energy (eV)	Lorentzian	Gaussian	Intensity (%)
C-C	284.4	0.08	0.90	17.7
C-H	284.8	0.08	0.90	47.9
S ₁	285.4	0.08	0.90	1.2
C-N	285.9	0.08	0.90	11.6
S ₂	286.6	0.08	0.90	4.3
S ₃	287.5	0.08	0.90	1.1
S ₄	288.4	0.08	1.44	2.0
S ₅	291.6	0.08	5.14	14.2

C-C + S₁+S₃+S₄=22.0%, C-H+ S₅=62.1%, C-N + S₂=15.9%

Table S2

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
N _{rad}	398.3	0.1	0.8	28.6
N ₂	399.2	0.1	0.8	28.1
S ₁	400.1	0.1	0.8	5.2
N ₁	401.0	0.1	0.8	29.7
S ₂	402.1	0.1	0.8	6.1
S ₃	403.0	0.1	0.8	2.3

N_{rad}+S₁= 33.8%, N₂+S₂= 34.2 %, N₁+S₃=32.0%

Fit parameters for the freshly evaporated diBlatter films as in Figure 2.

The expected stoichiometric values for the C 1s and N 1s elemental analyses are:

C-C = 11.1%, C-H = 66.7%, C-N = 22.2%, and N_{radical} = N₁ = N₂ = 33%

Table S3

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
C-C	284,9	0.08	1.0	11.9
C-H	285,3	0.08	1.0	52.7
S ₁	285,9	0.08	1.0	1.6
C-N	286,3	0.08	1.0	19.9
S ₂	287,0	0.08	1.0	8.8
S ₃	287,7	0.08	1.5	1.8
S ₄	292,2	0.08	3.1	3.3

C-C + S₁ = 13.5%, C-H + ½ S₂ + S₃ + S₄ = 62.2%, C-N + ½ S₂ = 24.3%

Table S4

	Energy (eV)	Lorentzian	Gaussian	Intensity (%)
N _{rad}	398.9	0.1	0.95	26.1
N _{1,3}	399.9	0.1	0.95	28.5
N _{2,4}	401.7	0.1	0.95	29.9
S ₁	400.6	0.1	0.95	5.9
S ₂	402.2	0.1	0.95	3.0
S ₃	402.4	0.1	0.95	3.0
S ₄	403.3	0.1	0.95	2.1
S ₅	404.3	0.1	0.95	1.5

$$N_{\text{rad}} + \frac{1}{2}(S_1 + S_2) + S_4 + S_5 = 34.1 \text{ \%}$$

$$N_{1/3} + \frac{1}{2}(S_1 + S_2) = 33.0 \text{ \%}$$

$$N_{2/4} + S_3 = 32.9 \text{ \%}$$

2) Time dependent core level signals.

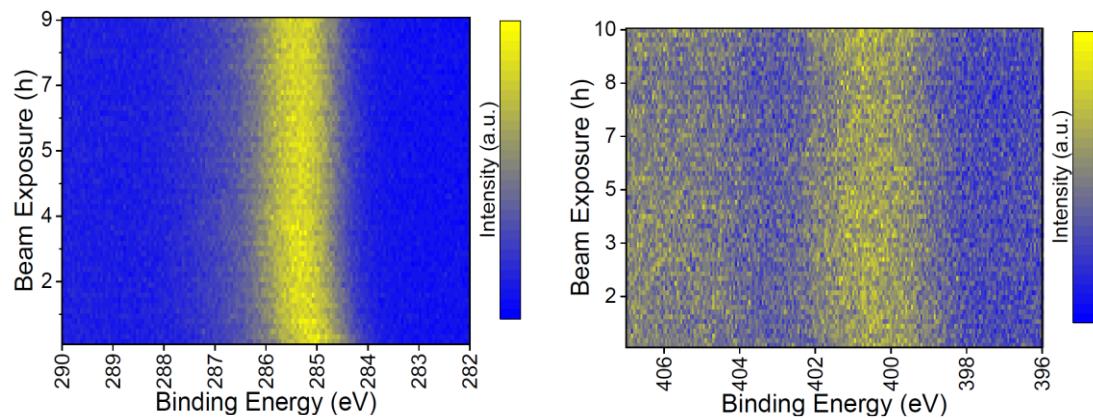


Figure S1. diBlatter thin films. (Left) Time-dependent C 1s and (right) N 1s core level signals.

Color scale: Blue represents the background signal; yellow is the peak intensity (photon energy: 700 eV).

3) Time-dependent NEXAFS signals.

The unoccupied states and the molecular arrangement adopted by the molecules in the thin films can be investigated by NEXAFS spectroscopy. NEXAFS features and their intensities are sensitive to intermolecular interactions, and NEXAFS dichroism allows determining the orientation of the molecules with respect to the substrate.^{6, 7} In particular, the main resonance at around 286.0 eV can be identified as due to the pyrene substituent, leading to a calculated average orientation of the pyrene plane of 30°. The observed dichroic behavior is the typically dichroic expected one for the herringbone structure,⁸⁻¹⁰ due to the compromise between the strength of the C-H and the $\pi-\pi$ interactions.¹¹

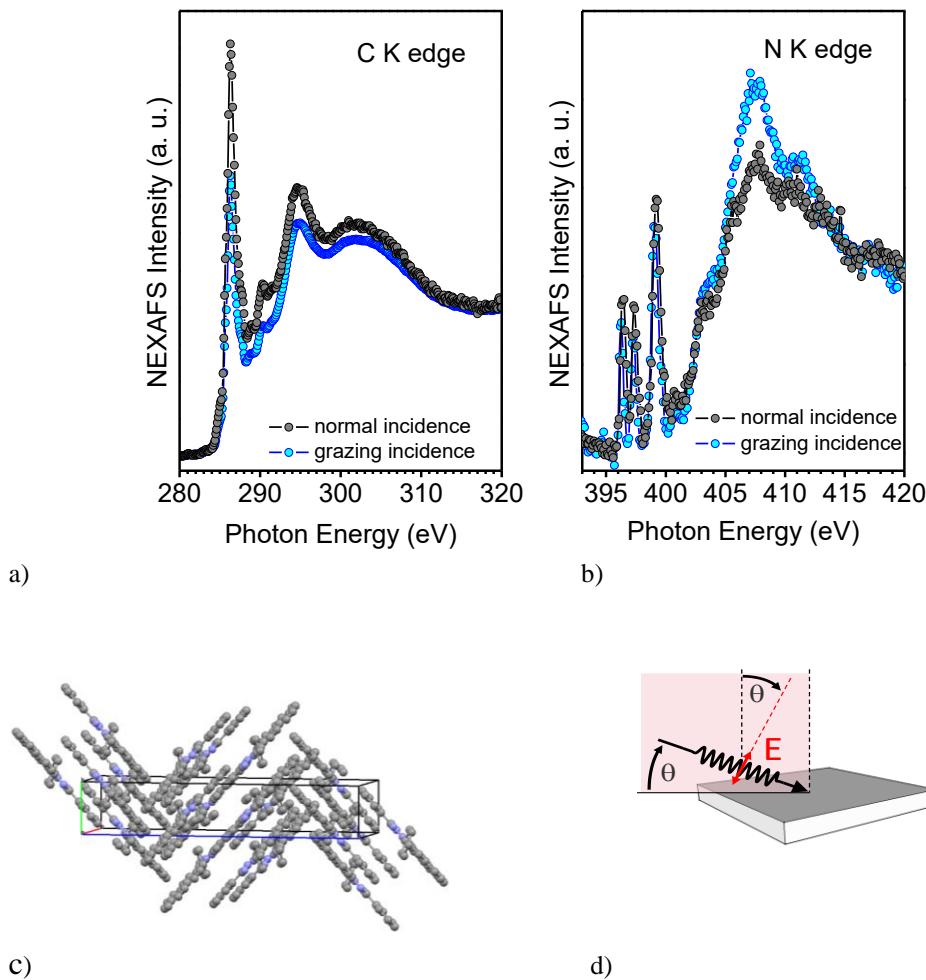


Figure S2. Blatter-pyr. a) C K edge and b) N K edge NEXAFS spectra of a nominally 3.5 nm thick film. c) Crystal structure. d) Geometry of the experiment.

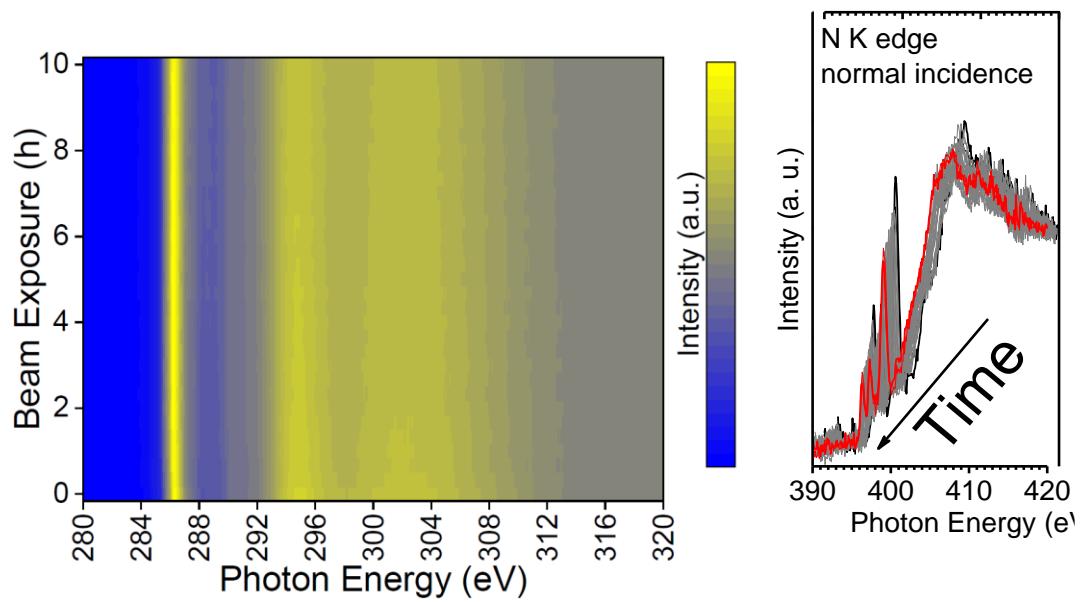


Figure S3. Blatter-pyr thin films. NEXAFS intensity versus beam exposure.

4) C 1s core level spectra after air exposure.

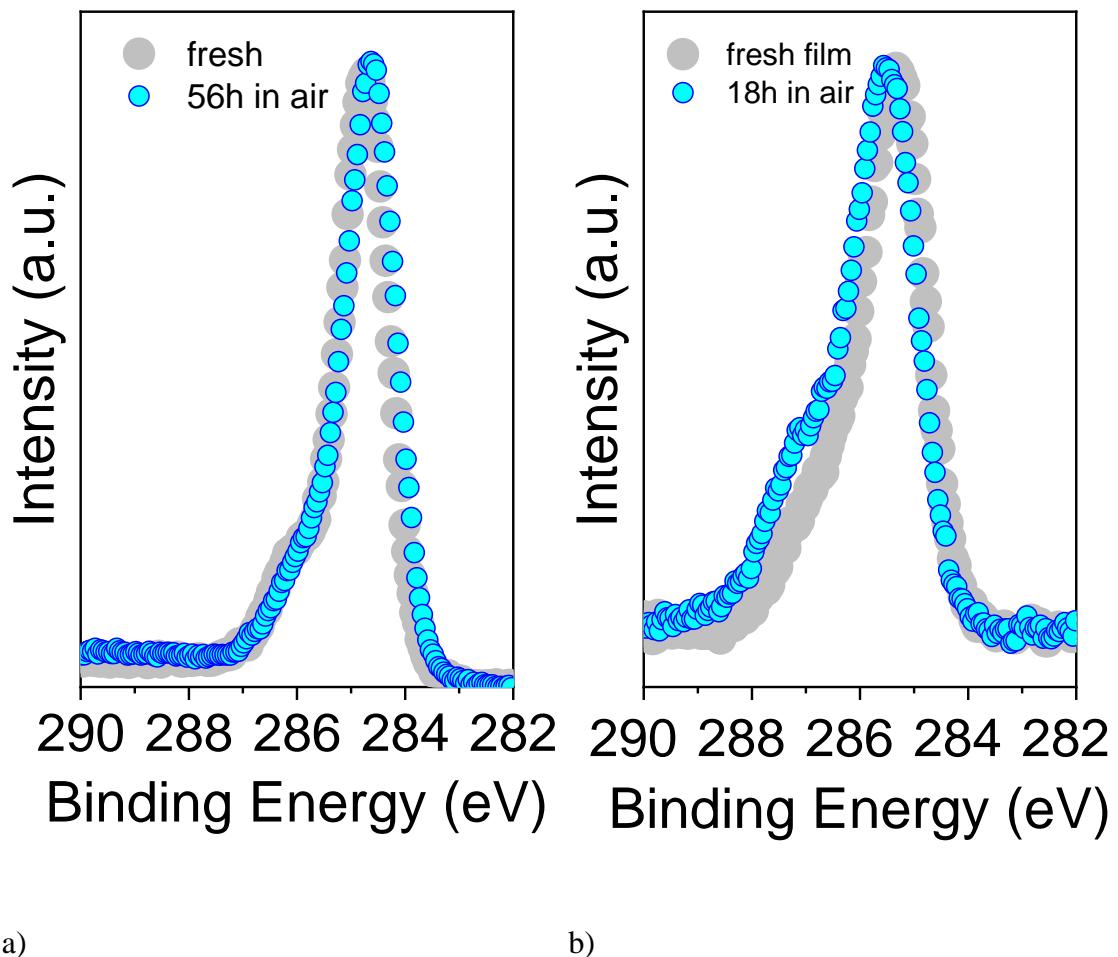


Figure S4. C 1s core level spectra of a) Blatter-pyr and b) diBlatter thin films, after air exposure, as indicated. The fresh film core level curves (light grey) are also shown for comparison (photon energy: 1486.6 eV).

5) Fit parameters for Blatter-pyr films as in Figure 3 after 45 hours of air exposure

Table S5

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
C-C	284.4	0.08	0.90	23.9
C-H	284.8	0.08	0.90	41.0
S ₁	285.6	0.08	0.90	2.3
C-N	285.8	0.08	0.90	10.1
S ₂	286.6	0.08	0.90	3.8
S ₃	287.5	0.08	0.90	0.5
S ₄	288.3	0.08	2.45	4.4
S ₅	291.6	0.08	5.14	14.0

Table S6

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
N _{rad}	398.3	0.1	0.8	27.0
N ₂	399.2	0.1	0.8	28.6
S ₁	400.1	0.1	0.8	6.4
N ₁	401.0	0.1	0.8	29.1
S ₂	402.0	0.1	0.8	5.8
S ₃	403.0	0.1	0.8	3.1

Fit parameters for diBlatter films as in Figure 3 after 72 hours of air exposure.

Table S7

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
C-C	284.9	0.08	1.0	11.2
C-H	285.4	0.08	1.0	55.9
S ₁	285.9	0.08	1.0	1.7
C-N	286.3	0.08	1.0	17.9
S ₂	287.0	0.08	1.0	7.7
S ₃	287.7	0.08	1.3	2.8
S ₄	292.2	0.08	3.8	2.8

Table S8

	Energy	Lorentzian	Gaussian	Intensity
	(eV)	Width (eV)	Width (eV)	(%)
N _{rad}	398.9	0.1	0.95	16.3
N _{1,3}	399.9	0.1	0.95	28.6
N*	400.8	0.1	0.95	25.0
N _{2,4}	401.7	0.1	0.95	20.4
S ₂	402.2	0.1	0.95	1.3
S ₃	402.4	0.1	0.95	4.3
S ₄	403.4	0.1	0.95	2.5
S ₅	404.3	0.1	0.95	1.6

6) Spin-polarized density of states of the Blatter-pyr.

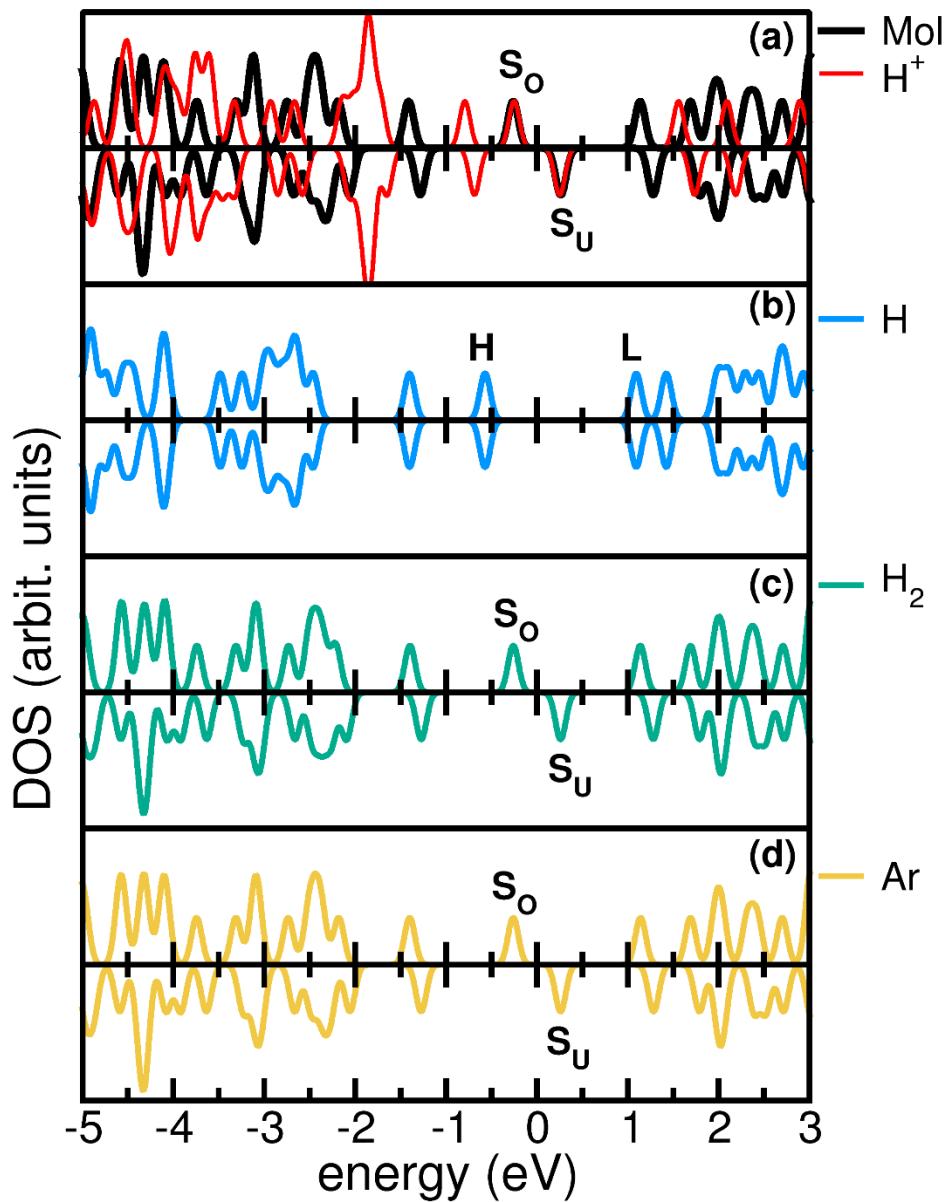


Figure S5. Spin-polarized density of states (DOS) of the Blatter-pyr without (black line, panel a) and with selected adsorbed impurities: H^+ (red line, panel a), H (cyan line, panel b), H_2 (green line panel c), and Ar (yellow line, panel d). Negative/positive energies indicate occupied/empty electronic states: highest Single Occupied (S_O) and lowest Single Unoccupied (S_U) molecular orbitals are indicated (panels a,c,d) , along with double occupied HOMO (H) and LUMO (L) states (panel b). All elements/molecules are adsorbed on site S1.

7) Spin-polarized density of states (DOS) of the diBlatter.

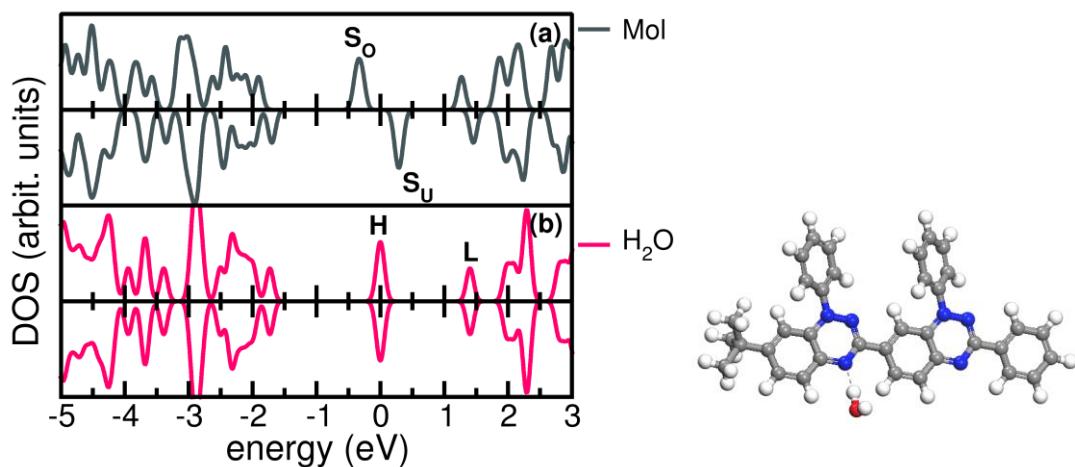


Figure S6. Spin-polarized density of states (DOS) of diBlatter radical (black line, panel a) and of the same molecule with H_2O molecule adsorbed in site S1' (pink line, panel b). Negative/positive energies indicate occupied/empty electronic states: double degenerate highest Single Occupied (S_O) and lowest Single Unoccupied (S_U) molecular orbitals are indicated (panels a), along with double degenerate/double occupied HOMO (H) and LUMO (L) states (panel b). Relaxed geometry upon water molecule adsorption is shown on right.

8) Simulated N 1s core level spectra of Blatter-pyr upon adsorption of characteristic H⁺, H₂, Ar, and OH species and as a function of the adsorption sites (S1).

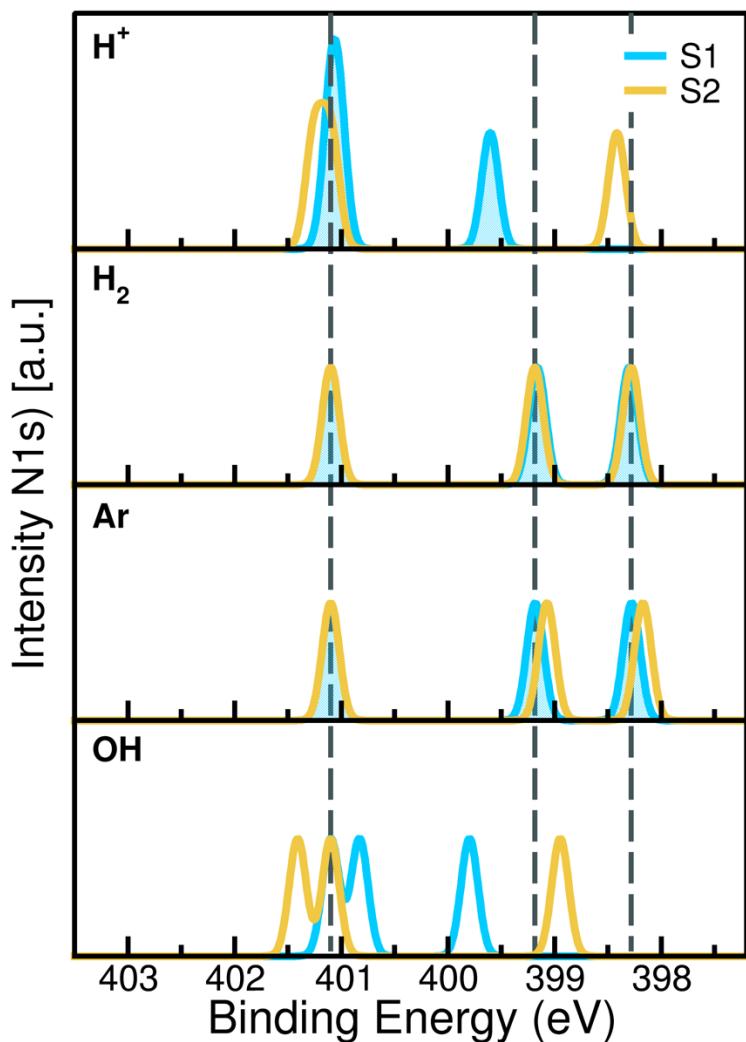


Figure S7. Simulated N 1s core level spectra of Blatter-pyr upon adsorption of characteristic H⁺, H₂, Ar, and OH species and as a function of the adsorption sites (S1). Vertical dashed lines refer to the core-level features of the pristine molecules. The N₃ peak is assumed as the energy reference for all spectra and aligned to the experimental value for a direct comparison. Labels and colors refer to Figure 5, main text.

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