### SHORT COMMUNICATION

# Inhibition of the alpha- and beta-carbonic anhydrases from the gastric pathogen *Helycobacter pylori* with anions

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#### Abstract

The gastric pathogen *Helicobacter pylori* encodes two carbonic anhydrases (CAs, EC 4.2.1.1), an  $\alpha$ - and a  $\beta$ -class one, hp $\alpha$ CA and hp $\beta$ CA, crucial for its survival in the acidic environment from the stomach. Sulfonamides, strong inhibitors of these enzymes, block the growth of the pathogen, *in vitro* and *in vivo*. Here we report the inhibition of the two *H. pylori* CAs with inorganic and complex anions and other molecules interacting with zinc proteins. hp $\alpha$ CA was inhibited in the low micromolar range by diethyldithiocarbamate, sulfamide, sulfamic acid, phenylboronic acid, and in the submillimolar one by cyanide, cyanate, hydrogen sulfide, divanadate, tellurate, perruthenate, selenocyanide, trithiocarbonate, iminodisulfonate. hp $\beta$ CA generally showed a stronger inhibition with most of these anions, with several low micromolar and many submillimolar inhibitors detected. These inhibitors may be used as leads for developing anti-*H. pylori* agents with a diverse mechanism of action compared to clinically used antibiotics.

Keywords: Carbonic anhydrase, anion, alpha/beta-class enzyme, inhibitor, *Helicobacter pylori* 

# Introduction

Carbonic anhydrases (CAs, EC 4.2.1.1) are metalloenzymes which catalyze the hydration of carbon dioxide to bicarbonate and protons<sup>1-3</sup>. Many pathogenic bacteria encode such enzymes belonging to the  $\alpha$ -,  $\beta$ -, and/or  $\gamma$ -CA families<sup>1</sup>. In the last decade, the  $\alpha$ -CAs from Neisseria spp. and Helicobacter pylori as well as the  $\beta$ -class enzymes from *Escherichia coli*, *H. pylori*, Mycobacterium tuberculosis, Brucella spp., Streptococcus pneumoniae, Salmonella enterica and Haemophilus influenzae have been cloned and characterized in detail<sup>1-13</sup>. Most of these pathogenic enzymes show a very high catalytic efficiency for the physiological reaction. Recent studies detected various classes of CA inhibitors (CAIs) targeting these enzymes, most of which belong to the sulfonamide/sulfamate class. They were critical to establish the roles of these CAs in the pathogen life cycle, and whether CA inhibition may constitute and alternative pathway for finding novel types of antibiotics<sup>1,14-16</sup>. For some of these enzymes, the X-ray crystal structures

HIM & SHING were also reported, which can be helpful for drug design purposes<sup>12,13</sup>. As resistance to antibiotics belonging to several different classes is escalating and represents a worldwide problem<sup>17,18</sup>, it is essential to explore alternative classes of compounds which inhibit crucial steps in pathogen's life cycles. Indeed, a high number of strains of Gram-negative/positive bacteria (such as Staphylococcus aureus, Mycobacterium tuberculosis, Helycobacter pylori, Brucella suis, Streptococcus pneumoniae, and so on) no longer respond to some classical antibiotics<sup>17,18</sup>. Cloning of the genomes of bacteria offers, however, the possibility to explore alternative pathways for inhibiting virulence factors or proteins essential for the pathogens. Among the many, new such possible drug targets recently explored, there are several CAs<sup>1</sup>.

These metalloenzymes are found in various organisms all over the phylogenetic tree, as five different, genetically distinct families, the  $\alpha$ -,  $\beta$ -,  $\gamma$ -,  $\delta$ - and  $\zeta$ -CAs<sup>1-7</sup>. The metal ion from the enzyme active site (which may be Zn(II); Fe(II); Cd(II) or Co(II) among others) is essential for the

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<sup>(</sup>Received 31 October 2011; revised 06 December 2011; accepted 09 December 2011)

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catalytic reaction and also for the binding of most (but not all) classes of CA inhibitors (CAIs) investigated so far<sup>1-11</sup>.

The genome of the human carcinogenic pathogen H. pylori contains at least two CAs, an  $\alpha$ -CA<sup>12</sup> (denominated hp $\alpha$ CA)<sup>14-16</sup> and a  $\beta$ -class enzyme, denominated hpßCA<sup>14-16,19,20</sup>. These two CAs show a different subcellular localization: a periplasmic one for the  $\alpha$ -class CA<sup>19</sup> and a cytoplasmic one for  $hp\beta CA^{15}$ . These enzymes were also shown to be catalytically efficient, with almost identical activity to that of the human (h) isoform hCA I, for the CO<sub>2</sub> hydration reaction, and highly inhibited by many sulfonamides/sulfamates, including acetazolamide, ethoxzolamide, topiramate and sulpiride, all clinically used drugs<sup>14-16</sup>. Furthermore, certain CAIs, such as the clinically used sulfonamides acetazolamide and methazolamide, were shown to inhibit the bacterial growth in cell cultures<sup>15</sup>. Since the efficacy of *H. pylori* eradication therapies currently employed has been decreasing due to drug resistance and side effects of the commonly used drugs<sup>15,20,21</sup>, the dual inhibition of  $\alpha$ - and/or  $\beta$ -CAs of H. pylori may serve as an alternative therapy in patients with H. pylori infection or for the prevention of gastroduodenal diseases provoked by this widespread pathogen<sup>1,15</sup>. In fact, a pilot study has demonstrated the efficacy of acetazolamide in the treatment of gastric ulcer<sup>21</sup>. Moreover, this compound (as well as ethoxzolamide) was widely used clinically as antiulcer agents in the 1970s and 1980s by Puscas's group, although its mechanism of action was not properly understood at that time as being due to the pathogen CA inhibition<sup>20</sup>.

All these data show that hp $\alpha$ CA and hp $\beta$ CA are promising drug targets and that sulfonamide CAIs may have clinical applications. However, their inhibition has been scarcely investigated to date, apart our earlier work on sulfonamide inhibitors<sup>14-16</sup>. Exploring alternative chemotypes to the sulfonamides as possible inhibitors of the  $\alpha/\beta$ -CAs from this bacterial pathogen is thus of great interest. Here, we report an inhibition study of the two enzymes with a wide range of simple inorganic anions, as well as various small molecule compounds known to target the metal ion in metalloenzymes like CAs, such as among others sulfamides, sulfamic acid, boronic and arsonic acids, and so on.

#### **Materials and methods**

#### Chemistry

All anions/small compounds used here were commercially available, highest purity reagents, from Sigma-Aldrich (Milan, Italy).

#### Enzymology

 $hp\alpha CA$  and  $hp\beta CA$  were recombinant enzymes obtained as described earlier  $^{14\text{--}16}$ .

#### CA catalytic activity and inhibition assay

An Applied Photophysics stopped-flow instrument has been used for assaying the CA catalysed CO<sub>2</sub> hydration activity<sup>22</sup>. Phenol red (at a concentration of 0.2 mM) has been used as indicator, working at the absorbance maximum of 557 nm, with 10-20 mM Hepes (pH 7.5, for  $\alpha$ -CAs) or TRIS (pH 8.3 for  $\beta$ -CAs) as buffers, and 20 mM  $Na_2SO_4$  (for  $\alpha$ -CAs) or 20 mM  $NaBF_4$  – for  $\beta$ -CAs (for maintaining constant the ionic strength), following the initial rates of the CA-catalyzed CO<sub>2</sub> hydration reaction for a period of 10–100 s. The CO<sub>2</sub> concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters and inhibition constants. For each inhibitor, at least six traces of the initial 5%-10% of the reaction have been used for determining the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of inhibitor (10 mM) were prepared in distilleddeionized water and dilutions up to 0.01 nM were done thereafter with distilled-deionized water. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature prior to assay, in order to allow for the formation of the E-I complex. The inhibition constants were obtained by non-linear least-squares methods using PRISM 3, whereas the kinetic parameters for the uninhibited enzymes from Lineweaver-Burk plots, as reported earlier<sup>23-30</sup>, and represent the mean from at least three different determinations.

# **Results and discussions**

Inhibition data of two human CA isoforms, hCA I and II (highly abundant proteins with important physiological functions)<sup>2-4</sup> as well as three bacterial CAs, PCA (a  $\beta$ -CA from *S. pneumoniae*) and hp $\alpha$ CA and hp $\beta$ CA investigated here are shown in Table 1. The data of hCA I, hCA II and PCA were reported earlier by this group<sup>9,25-27</sup>, and are provided here for comparisons reasons.

The following should be noted regarding the inhibition data of Table 1:

1. The  $\alpha$ -class enzyme hp $\alpha$ CA was sensitive to this class of inhibitors, which showed inhibition constants in the range of 4.9 µM-10.1 mM. The only anion which did not show any notable inhibitory properties was tetrafluoroborate ( $K_1 > 200 \text{ mM}$ ), known for its lack of interaction with most metal ions from metalloenzyme active sites9. The halogenides, thiocyanate, and perchlorate were the least effective inhibitors of hp $\alpha$ CA, with K<sub>1</sub>s in the range of 2.41–10.1 mM. This is rather amazing, considering that perchlorate does not inhibit appreciably any other  $\alpha$ - or  $\beta$ -CA investigated earlier<sup>1-9</sup>. The remaining anions were submillimolar hp $\alpha$ CA inhibitors, with K<sub>1</sub>s in the range of 0.27-0.99 mM, except for four derivatives which were micromolar inhibitors (Table 1). Indeed, diethyldithiocarbamate, sulfamide, sulfamic acid and phenylboronic acid were the most effective inhibitors of this enzyme, with K<sub>i</sub>s in the range 4.9–97  $\mu$ M. It is interesting to observe that sulfate is a weak inhibitor (K, of 0.82 mM) but replacing one or two oxygen

Table 1. Inhibition constants of anion inhibitors against  $\alpha$ -/ $\beta$ -CAs from mammals (hCA I, and II, human isoforms) and bacteria: PCA (from *S. pneumoniae*), and hp $\alpha$ CA/hp $\beta$ CA (from *H. pylori*) for the CO<sub>2</sub> hydration reaction, at 20 °C<sup>22</sup>.

$\begin{array}{ c c c c c c c c c c c c c c c c c c c$			K <sub>I</sub> [mM] <sup>b</sup>			
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Inhibitor <sup>a</sup>	hCA I <sup>d</sup>	$hCAII^{d}$	PCA <sup>e</sup>	$hp\alpha CA^{c}$	hpβCA <sup>c</sup>
Br4630.0462.410.38I0.3260.0546.050.63CNO'0.00070.030.0980.600.37SCN'0.21.600.384.100.68CN'0.00050.020.0410.760.54N_3^-0.00121.510.350.830.80HCO_3^-12850.330.750.50CO_3^215730.530.660.42NO_2^-8.4630.660.930.67HS'0.00060.040.350.690.58HSO_3^-18890.570.990.63SnO_3^2-0.570.830.0660.550.48SeO_4^2-1181120.0440.720.65TeO_4^2-0.660.920.0490.340.45Oso5_2^2-0.990.970.480.710.61P_2Q_7^4-0.540.570.0380.270.18B_4Q_7^2-0.640.950.320.560.68ReQ_4^-0.1100.750.0390.880.82RuO_4^-0.1010.690.0360.361.10S_2Q_8^{2-}0.6970.0880.2210.380.21SeCN'0.0850.0860.220.730.97CS_3^20.00870.00880.210.380.21SeCN'0.0850.8660.220.73<	F-	> 300	>300	0.85	4.08	0.67
I0.3260.0546.050.63CNO'0.00070.030.0980.600.37SCN'0.21.600.384.100.68CN'0.00050.020.0410.760.54 $N_3^-$ 0.00121.510.350.830.80HCO_3^-12850.330.750.50CO_3^2-15730.530.660.42NO_3^-7350.390.810.78HS'0.00060.040.350.690.58HSO_3^-18890.570.990.63SnO_3^{2-}0.570.830.0660.550.48SeO_4^{2-}1181120.0440.720.65TeO_4^{2-}0.660.920.0490.340.45OSO_5^20.920.950.0600.480.89 $S_2O_7^{2-}$ 0.990.970.0480.710.61 $P_2O_7^{4-}$ 25.7748.500.450.660.75 $V_2O_7^4-$ 0.540.570.380.270.18 $ReO_4^-$ 0.1100.750.390.880.82RuO_4^-0.1010.690.360.361.10 $S_2O_8^{2-}$ 0.6970.0840.0220.730.97CS_3^2^-0.0870.0880.0210.380.21SeCN'0.0850.860.0220.730.97CS_2^20.793.1	Cl⁻	6	200	0.052	2.70	0.56
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Br-	4	63	0.046	2.41	0.38
$\begin{array}{llllllllllllllllllllllllllllllllllll$	I-	0.3	26	0.054	6.05	0.63
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	CNO-	0.0007	0.03	0.098	0.60	0.37
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		0.0005	0.02		0.76	0.54
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0	0.0012	1.51	0.35	0.83	0.80
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		12	85	0.33	0.75	0.50
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	CO <sub>3</sub> <sup>2-</sup>	15	73	0.53	0.66	0.42
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		7	35	0.39	0.81	0.78
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$NO_2^-$	8.4	63	0.66	0.93	0.67
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	HS-	0.0006	0.04	0.35	0.69	0.58
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	HSO <sub>3</sub> <sup>-</sup>	18	89	0.57	0.99	0.63
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	SnO <sub>3</sub> <sup>2-</sup>	0.57	0.83	0.066	0.55	0.48
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	SeO <sub>4</sub> <sup>2-</sup>	118	112	0.044	0.72	0.65
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	TeO <sub>4</sub> <sup>2-</sup>	0.66	0.92	0.049	0.34	0.45
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		0.92	0.95	0.060	0.48	0.89
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		0.99	0.97	0.048	0.71	0.61
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		25.77	48.50	0.45	0.66	0.75
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	V <sub>2</sub> O <sub>7</sub> <sup>4-</sup>	0.54	0.57	0.038	0.27	0.18
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$B_4 O_7^{2-}$	0.64	0.95	0.32	0.56	0.68
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	ReO <sub>4</sub> -	0.110	0.75	0.039	0.88	0.82
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	RuO <sub>4</sub> -	0.101	0.69	0.036	0.36	1.10
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		0.107	0.084	0.046	0.92	0.93
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	SeCN <sup>-</sup>	0.085	0.086	0.022	0.73	0.97
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$CS_{3}^{2-}$	0.0087	0.0088	0.021	0.38	0.21
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Et <sub>2</sub> NCS <sub>2</sub> <sup>-</sup>	0.79	3.10	0.61	0.0049	0.0074
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	SO4 <sup>2-</sup>	63	>200	4.15	0.82	0.57
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		>200	>200	>200	10.1	6.50
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	-	>200	>200	>200	>200	>200
$\begin{array}{llllllllllllllllllllllllllllllllllll$	•	0.79	0.46	0.060	0.91	0.75
$H_2 NSO_2 NH_2$ 0.31 1.13 4.25 0.073 0.072		0.31	0.76	28.1	0.54	0.70
	0 2	0.31	1.13	4.25	0.073	0.072
2 J		0.021	0.39	6.68	0.080	0.094
Ph-B(OH), 58.6 23.1 6.47 0.097 0.073	2 0	58.6	23.1	6.47	0.097	0.073
Ph-AsO <sub>3</sub> H <sub>2</sub> 31.7 49.2 5.86 0.44 0.092	2	31.7	49.2	5.86	0.44	0.092

<sup>a</sup>As sodium salt.

<sup>b</sup>Errors were in the range of 3%–5 % of the reported values, from three different assays;

<sup>c</sup>This work.

<sup>d,e</sup>From references<sup>9,25,26</sup>.

atoms from it with  $NH_2$  moieties, as in sulfamic acid or sulfamides, leads to a dramatic increase of the inhibitory power. The same may be noted regarding trithiocarbonate, which is only slightly more inhibitory than carbonate (or bicarbonate). However, using this zinc-binding group and adding the diethylamino fragment to it, as in diethyldithiocarbamate, leads to a low nanomolar inhibitor. It is thus obvious that the dithiocarbamates, a class of recently discovered CAIs<sup>24,28</sup>, constitute a novel class of inhibitors also for hp $\alpha$ CA. Among the other simple inorganic anions investigated here it should be noted also that tellurate, divanadate and perruthenate were among the most efficient submillimolar inhibitors, with K<sub>1</sub>s in the range of 0.27–0.36 mM.

- 2. hp $\beta$ CA was generally even more sensitive to this class of CAIs compared to the  $\alpha$ -class enzyme discussed above, these compounds showing inhibition constants in the range of 7.4  $\mu$ M–6.4 mM, again, except tetrafluoroborate which was not inhibitory,  $K_r > 200 \text{ mM}$  (Table 1). The most effective hp $\beta$ CA inhibitors were diethyldithiocarbamate, sulfamide, sulfamic acid, phenylboronic and phenylarsonic acid, with K,s in the range 7.4-94 µM. All the remaining anions, except perchlorate and perruthenate (K<sub>i</sub>s of 1.1–6.4 mM) were submillimolar inhibitors with K<sub>1</sub>s in the range of 0.18–0.97 mM (Table 1). Among the effective anions were again trithiocarbonate and divanadate, with K<sub>i</sub>s in the range 180-210  $\mu$ M. Generally, most of these anions showed a slightly better inhibitory capacity against the  $\beta$ - over the  $\alpha$ -class enzyme of *H. pylori*, although few anions (diethyldithiocarbamate, tellurate and perosmate among others) were better hp $\alpha$ CA than hp $\beta$ CA inhibitors.
- 3. The inhibition profiles of the two *H. pylori* enzymes are very different both from those of the host enzymes considered here, hCA I and II, as well as from those of the bacterial enzyme used for comparison, PCA (Table 1). This is a very encouraging result being also a proof that it may be possible to develop CAIs which specifically target the *H. pylori* enzymes, without interfering with the human isoforms, highly abundant in many tissues and involved in critical physiological processes.

#### Conclusions

We evaluated a series of inorganic anions and similar small molecules known to bind to metalloenzymes, for the inhibition of hp $\alpha$ CA and hp $\beta$ CA, the two CAs from the bacterial pathogen *H. pylori*. As other enzymes from these classes investigated earlier, they are highly sensitive to anion inhibitors. hp $\alpha$ CA was inhibited in the low micromolar range by diethyldithiocarbamate (K, of 4.9  $\mu$ M), sulfamide, sulfamic acid, phenylboronic acid, and in the submillimolar one by a wide range of anions including, cyanide, cyanate, hydrogen sulfide, divanadate, tellurate, perruthenate, selenocyanide, trithiocarbonate, and iminodisulfonate. hp $\beta$ CA generally showed an even stronger inhibition with most of these anions compared to the  $\alpha$ -class enzyme, with several low micromolar and many submillimolar inhibitors detected (among which diethyldithiocarbamate, sulfamide, sulfamic acid, phenylboronic/phenylarsonic acid). These new CA inhibitors detected here may be used as leads for developing anti-*H. pylori* agents with a diverse mechanism of action compared to clinically used antibiotics for which many strains exhibit a wide range of drug resistance.

# **Declaration of interest**

The authors report no conflict of interest. This work was supported by an EU FP7 research grant (Metoxia project).

# References

- 1. Supuran CT. Bacterial carbonic anhydrases as drug targets: toward novel antibiotics? Front Pharmacol 2011;2:34.
- 2. Supuran CT. Carbonic anhydrase inhibitors. Bioorg Med Chem Lett 2010;20:3467-3474.
- 3. Supuran CT. Carbonic anhydrases: novel therapeutic applications for inhibitors and activators. Nat Rev Drug Discov 2008;7:168–181.
- 4. Neri D, Supuran CT. Interfering with pH regulation in tumours as a therapeutic strategy. Nat Rev Drug Discov 2011;10:767-777.
- 5. Pastorekova S, Parkkila S, Pastorek J, Supuran CT. Carbonic anhydrases: current state of the art, therapeutic applications and future prospects. J Enzyme Inhib Med Chem 2004;19:199–229.
- 6. Supuran CT. Carbonic anhydrase inhibitors and activators for novel therapeutic applications. Future Med Chem 2011;3:1165–1180.
- 7. Supuran CT, Scozzafava A, Casini A. Carbonic anhydrase inhibitors. Med Res Rev 2003;23:146–189.
- 8. Nishimori I, Minakuchi T, Vullo D, Scozzafava A, Innocenti A, Supuran CT. Carbonic anhydrase inhibitors. Cloning, characterization, and inhibition studies of a new  $\beta$ -carbonic anhydrase from *Mycobacterium tuberculosis*. J Med Chem 2009;52:3116-3120.
- 9. Burghout P, Vullo D, Scozzafava A, Hermans PW, Supuran CT. Inhibition of the  $\beta$ -carbonic anhydrase from *Streptococcus pneumoniae* by inorganic anions and small molecules: Toward innovative drug design of antiinfectives? Bioorg Med Chem 2011;19:243–248.
- 10. Winum JY, Köhler S, Supuran CT. *Brucella* carbonic anhydrases: new targets for designing anti-infective agents. Curr Pharm Des 2010;16:3310-3316.
- 11. Vullo D, Nishimori I, Minakuchi T, Scozzafava A, Supuran CT. Inhibition studies with anions and small molecules of two novel  $\beta$ -carbonic anhydrases from the bacterial pathogen *Salmonella enterica* serovar Typhimurium. Bioorg Med Chem Lett 2011;21:3591–3595.
- Chirică LC, Petersson C, Hurtig M, Jonsson BH, Borén T, Lindskog S. Expression and localization of alpha- and betacarbonic anhydrase in *Helicobacter pylori*. Biochim Biophys Acta 2002;1601:192–199.
- Elleby B, Chirică LC, Tu C, Zeppezauer M, Lindskog S. Characterization of carbonic anhydrase from *Neisseria gonorrhoeae*. Eur J Biochem 2001;268:1613-1619.
- 14. Nishimori I, Minakuchi T, Morimoto K, Sano S, Onishi S, Takeuchi H et al. Carbonic anhydrase inhibitors: DNA cloning and inhibition

studies of the alpha-carbonic anhydrase from *Helicobacter pylori*, a new target for developing sulfonamide and sulfamate gastric drugs. J Med Chem 2006;49:2117-2126.

- 15. Nishimori I, Onishi S, Takeuchi H, Supuran CT. The  $\alpha$  and  $\beta$  classes carbonic anhydrases from *Helicobacter pylori* as novel drug targets. Curr Pharm Des 2008;14:622–630.
- 16. Nishimori I, Minakuchi T, Kohsaki T, Onishi S, Takeuchi H, Vullo D et al. Carbonic anhydrase inhibitors: the  $\beta$ -carbonic anhydrase from *Helicobacter pylori* is a new target for sulfonamide and sulfamate inhibitors. Bioorg Med Chem Lett 2007;17:3585-3594.
- 17. Furtado GH, Nicolau DP. Overview perspective of bacterial resistance. Expert Opin Ther Pat 2010;20:1273–1276.
- Ginsberg AM. Emerging drugs for active tuberculosis. Semin Respir Crit Care Med 2008;29:552–559.
- 19. Marcus EA, Moshfegh AP, Sachs G, Scott DR. The periplasmic alpha-carbonic anhydrase activity of *Helicobacter pylori* is essential for acid acclimation. J Bacteriol 2005;187:729–738.
- 20. Puscas I. Treatment of gastroduodenal ulcers with carbonic anhydrase inhibitors. Ann N Y Acad Sci 1984;429:587-591.
- 21. Shahidzadeh R, Opekun A, Shiotani A, Graham DY. Effect of the carbonic anhydrase inhibitor, acetazolamide, on *Helicobacter pylori* infection in vivo: a pilot study. Helicobacter 2005; 10:136-138.
- 22. Khalifah RG. The carbon dioxide hydration activity of carbonic anhydrase. I. Stop-flow kinetic studies on the native human isoenzymes B and C. J Biol Chem 1971;246:2561–2573.
- Supuran CT. Carbonic anhydrase inhibition with natural products: novel chemotypes and inhibition mechanisms. Mol Divers 2011;15:305–316.
- 24. Ozensoy O, Arslan M, Supuran CT. Carbonic anhydrase inhibitors: purification and inhibition studies of pigeon (*Columba livia* var. *domestica*) red blood cell carbonic anhydrase with sulfonamides. J Enzyme Inhib Med Chem 2011;26:749-753.
- 25. Innocenti A, Scozzafava A, Supuran CT. Carbonic anhydrase inhibitors. Inhibition of cytosolic isoforms I, II, III, VII and XIII with less investigated inorganic anions. Bioorg Med Chem Lett 2009;19:1855-1857.
- 26. Innocenti A, Scozzafava A, Supuran CT. Carbonic anhydrase inhibitors. Inhibition of transmembrane isoforms IX, XII, and XIV with less investigated anions including trithiocarbonate and dithiocarbamate. Bioorg Med Chem Lett 2010;20: 1548–1550.
- 27. Temperini C, Scozzafava A, Supuran CT. Carbonic anhydrase inhibitors. X-ray crystal studies of the carbonic anhydrase II – trithiocarbonate adduct – an inhibitor mimicking the sulfonamide and urea binding to the enzyme. Bioorg Med Chem Lett 2010;20:474–478.
- 28. Kolayli S, Karahalil F, Sahin H, Dincer B, Supuran CT. Characterization and inhibition studies of an α-carbonic anhydrase from the endangered sturgeon species *Acipenser gueldenstaedti*. J Enzyme Inhib Med Chem 2011;26:895–900.
- 29. Schlicker C, Hall RA, Vullo D, Middelhaufe S, Gertz M, Supuran CT et al. Structure and inhibition of the CO2-sensing carbonic anhydrase Can2 from the pathogenic fungus *Cryptococcus neoformans*. J Mol Biol 2009;385:1207–1220.
- 30. Pacchiano F, Carta F, Vullo D, Scozzafava A, Supuran CT. Inhibition of  $\beta$ -carbonic anhydrases with ureido-substituted benzenesulfonamides. Bioorg Med Chem Lett 2011;21:102–105.