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Addition/Correction

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Correction to "New Measurements of the Apparent Thermal Conductivity of Nanofluids and Investigation of Their Heat Transfer Capabilities"

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In Tables 3−6, the values given for the "mass %" column, inserted during the review process at the request of a referee, have been incorrectly calculated from the values of the adjacent "vol %" column. It is important to stress that this does not affect

Table 3. Measurements of the Thermal Conductivity of CuO in Ethylene Glycol, TiO_2 in Water and TiO_2 in Ethylene Glycol, at 298.15 K[a]

			transient hot-wire instrument		hot-disk thermal analyzer		
size/nm	mass %	vol %	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	dispersant or stabilizer[b]
				CuO + $C_2H_6O_2$			
<80	23.28	5.00	323	29.6			none/1 h sonication
	15.84	3.16	293	17.7			none/1 h sonication
	10.53	2.00	276	10.7			none/1 h sonication
	5.50	1.00	264	6.0			none/1 h sonication
40	23.28	5.00	308	23.8			none/1 h sonication
	15.13	3.00	288	15.7			none/1 h sonication
	9.80	1.85	270	8.5			none/1 h sonication
	5.13	0.93	262	5.0			none/1 h sonication
				TiO_2 + H_2O			
30−50	13.98	4.00	658	8.4			1 mass % PVP
	10.76	3.00	651	7.4			1 mass % PVP
	8.17	2.23	636	4.9			1 mass % PVP
	3.38	0.89	619	2.1			1 mass % PVP
5−15	13.98	4.00	666	9.7	665	9.6	1 mass % PVP
	10.76	3.00	660	8.7	659	8.6	1 mass % PVP
	7.37	2.00	640	5.5	644	6.1	1 mass % PVP
5	19.93	6.00	648	6.8			none/2 h sonication
	17.03	5.00	637	5.0			none/2 h sonication
	13.98	4.00	639	5.4			none/2 h sonication
	10.76	3.00	626	3.2			none/2 h sonication
	7.37	2.00	618	1.9			none/2 h sonication
				TiO_2 + $C_2H_6O_2$			
15	19.99	6.64	351	41.0			1 mass % PVP
	15.61	5.00	325	30.7			1 mass % PVP
	10.77	3.32	291	17.1			1 mass % PVP
	5.60	1.66	268	7.5			1 mass % PVP
5	12.77	4.00	269	8.1			none/1 h sonication
	9.80	3.00	266	6.7			none/1 h sonication
	6.69	2.00	258	3.8			none/1 h sonication
	3.43	1.00	253	1.6			none/1 h sonication

[a]Transient hot-wire $U_c(\lambda)$ = 2% (level of confidence = 0.95); hot-disk $U_c(\lambda)$ = 5% (level of confidence = 0.95). [b]PVP poly(vinylpyrrolidone).

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Table 4. Measurements of the Thermal Conductivity of Al_2O_3 in Water and in Ethylene Glycol at 298.15 K[a]

			transient hot-wire instrument		hot-disk thermal analyzer		
size/nm	mass %	vol %	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	dispersant or stabilizer[b]
				$Al_2O_3 + H_2O$			
30	19.89	6.00	724	19.4			1 mass % PVP
	16.99	5.00	706	16.4			1 mass % PVP
	13.95	4.00	683	12.6			1 mass % PVP
	10.74	3.00	664	9.5			1 mass % PVP
	5.59	1.50	637	5.0			1 mass % PVP
10	19.89	6.00	716	18.1	721	18.9	1 mass % PVP
	16.99	5.00	703	15.9	705	16.3	1 mass % PVP
	10.74	3.00	672	10.8	668	10.2	1 mass % PVP
	5.59	1.50	646	6.5	645	6.4	1 mass % PVP
5	16.99	5.00	661	8.9			none/1h sonication
	10.74	3.00	641	5.7			none/1h sonication
	7.35	2.00	629	3.6			none/1h sonication
	5.59	1.50	626	3.1			none/1h sonication
	3.78	1.00	621	2.5			none/1h sonication
				$Al_2O_3 + C_2H_6O_2$			
15	15.57	5.00	304	22.1			1 mass % PVP
	12.74	4.00	294	17.9			1 mass % PVP
	6.67	2.00	278	11.5			1 mass % PVP
	3.42	1.00	260	4.2			1 mass % PVP
5	15.57	5.00	285	14.5			none/1h sonication
	9.78	3.00	271	8.9			none/1h sonication
	6.67	2.00	264	6.0			none/1h sonication
	3.42	1.00	261	4.8			none/1h sonication

[a]Transient hot-wire $U_c(\lambda)$ = 2% (level of confidence = 0.95); hot-disk $U_c(\lambda)$ = 5% (level of confidence = 0.95). [b]PVP poly(vinylpyrrolidone).

Table 5. Measurements of the Thermal Conductivity of MWCNT in Water at 298.15 K[a]

			transient hot-wire instrument		hot-disk thermal analyzer		
size/nm	mass %	vol %	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	λ/mW·m^{-1}·K^{-1}	$(\lambda-\lambda_{BF})/\lambda_{BF}$/%	dispersant or stabilizer[b]
50−80[b]	3.00	1.45	748	23.3	751	23.8	2 mass % nonionic
	2.08	1.00	696	14.8	704	16.1	2 mass % nonionic
	1.04	0.50	652	7.6			2 mass % nonionic
	0.52	0.25	633	4.4	621	2.4	2 mass % nonionic
5−15[c]	2.08	1.00	639	5.4			2 mass % nonionic
	1.51	0.725	628	3.6			2 mass % nonionic
	1.04	0.50	626	3.3			2 mass % nonionic
	0.52	0.25	619	2.1			2 mass % nonionic

[a]Transient hot-wire $U_c(\lambda)$ = 2% (level of confidence = 0.95); Hot-disk $U_c(\lambda)$ = 5% (level of confidence = 0.95). [b]10−20 μm length. [c]50 μm length.

Table 6. Measurements of the Viscosity of Nanofluids at 298.15 K[a]

			AR550 rotational rheometer			
size/nm	mass %	vol %	η_{pl}[b]/mPa·s	η_{pl0}[c]/mPa	dispersant or stabilizer[d]	shear rate/s^{-1}
				$CuO + C_2H_6O_2$		
<80	19.37	4.00	18.2	105.3	none/1 h sonication	100−1500
	5.50	1.00	17.0	83.1	none/1 h sonication	100−1500
40	19.37	4.00	22.8	1539.4	none/1 h sonication	100−2000
	10.53	2.00	18.3	623.4	none/1 h sonication	100−2000
	5.50	1.00	17.0	89.2	none/1 h sonication	100−1500
				$TiO_2 + H_2O$		
30−50	10.10	2.80	4.9	6451.0	1 mass % PVP	100−2500
	3.38	0.89	0.9	0.0	1 mass % PVP	100−1000
5−15	13.98	4.00	15.3	35216.0	1 mass % PVP	100−2000
	10.76	3.00	6.9	11740.0	1 mass % PVP	100−2500
	7.37	2.00	3.7	2354.5	1 mass % PVP	100−2000
	3.79	1.00	1.1	0.0	1 mass % PVP	100−1500
5	13.98	4.00	1.2	27.0	none/2h sonication	100−1500
	3.79	1.00	0.98	4.0	none/2h sonication	100−1500

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Table 6. continued

			AR550 rotational rheometer			
size/nm	mass %	vol %	η_{pl}[b]/mPa·s	η_{pl0}[c]/mPa	dispersant or stabilizer[d]	shear rate/s^{-1}
			$TiO_2 + C_2H_6O_2$			
5	12.77	4.00	33.8	13603.0	none/1 h sonication	100−2500
	6.69	2.00	26.0	3204.0	none/1 h sonication	100−2500
	3.43	1.00	21.7	584.9	none/1 h sonication	100−2000
			$Al_2O_3 + H_2O$			
30	13.95	4.00	12.1	2499.9	1 mass % PVP	100−2000
	5.59	1.50	23.0	206.2	1 mass % PVP	100−1500
10	15.49	4.50	27.7	9695.8	1 mass % PVP	100−2000
	5.59	1.50	2.2	308.1	1 mass % PVP	100−1500
5	13.95	4.00	1.9	474.4	none/1 h sonication	100−2000
	7.35	2.00	1.2	288.0	none/1 h sonication	100−2000
	3.78	1.00	1.1	0.0	none/1 h sonication	100−1500
			$Al_2O_3 + C_2H_6O_2$			
15	12.74	4.00	81.8	30186.0	1 mass % PVP	100−1500
	6.67	2.00	35.1	2327.3	1 mass % PVP	100−2000
5	12.74	4.00	25.1	1147.3	none/1 h sonication	100−2000
	6.67	2.00	19.8	288.0	none/1 h sonication	100−1500
	3.42	1.00	18.2	119.1	none/1 h sonication	100−1500
			MWCNT + H_2O			
50−80[e]	2.08	1.00	1.6	48.2	2 mass % nonionic	100−1500
	0.52	0.25	1.0	0.0	2 mass % nonionic	100−1500
5−15[f]	2.08	1.00	4.4	1165.0	2 mass % nonionic	100−2500
	1.04	0.50	1.6	128.4	2 mass % nonionic	100−1500
	0.52	0.25	1.3	153.8	2 mass % nonionic	100−1500

[a]$U_c(\eta)$ = 5% (level of confidence = 0.95). [b]η_{pl} viscosity or plastic viscosity (Newtonian or Bingham plastic fluids, respectively). [c]η_{pl0} yield point (Bingham plastic fluids). [d]PVP poly(vinylpyrrolidone). [e]10−20 μm length. [f]50 μm length.

any other column in the tables, or any other calculation in the paper, because everything was based on the correct "vol %" concentration values. Corrected tables are attached. The authors would like to apologize for any inconvenience this may have caused.

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