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Light-driven benzylic azolation

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ORIGIN

Selective activation of C(sp³)–H bonds has been reported by Musacchio *et al.*, who developed a photoredox-catalyzed coupling between benzylic positions and azoles to afford *N*-benzylated products. Critically, *N*-alkoxypyridinium salts are used for enabling C(sp³)–H azolation via a combination of hydrogen atom transfer (HAT) and oxidative radical polar crossover (RPC). This platform has been demonstrated with a broad series of heterocycles and various benzylic coupling partners.

REACTION MECHANISM

In the mechanistic proposal, the photophysical properties of Ir(dFppy)₃ as a photocatalyst confirmed its strong photoreduction $\{E_{1/2}^{exc}[*Ir(III)/Ir(IV)] = -1.23 \text{ V}$ s. Ag/AgCI in MeCN} and results in oxidative quenching via a single electron transfer (SET) with N-methoxypyridinium $\mathbf{1}(E_{1/2}^{ecd} = -0.44 \text{ V}$ s. Ag/AgCI in MeCN]. This process has been explored by cyclic voltammetry and Stern–Volmer fluorescence quenching studies \mathbf{A} , which reveals efficient oxidative quenching of the excited state photocatalyst by electron withdrawing substituted pyridinium salt due to a preferable reduction potential \mathbf{B} . The so obtained methoxy radical $\mathbf{2}$ is competent in an intermolecular HAT step with the benzylic proton of substrate $\mathbf{3}$ to form a carbon-centered radical intermediate $\mathbf{4}$ and simultaneously liberates pyridine, as low nucleophilic byproduct \mathbf{D} . The initial formation of an electron donor-acceptor (EDA) complex with pyridinium and benzylic reagent via π – π stacking is plausible as observed by a bathochromic shift in the UV-Vis study \mathbf{C} . Such an EDA complex could facilitate the reduction of $\mathbf{1}$, therefore enhancing the release of radical $\mathbf{2}$. Respectively, electron-rich substrate shows more competent stacking with electron-poor pyridinium salt. Subsequent SET oxidation of $\mathbf{4}$ with Ir(IV) results in an RPC to yield benzylic carbocation $\mathbf{5}$, which enhances the azolation in a nucleophilic addition to yield the product $\mathbf{6}$. This platform has been demonstrated for a series of N-heterocycles with diverse electronic properties, such as halides, cyano, ester, or phenyl substituents, which could couple with benzylic C(sp³)</sup>–H bonds. Gratifyingly, this methodology even tolerates benzylic chlorides, a modular functional group for follow-up derivatization.



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IMPORTANCE

This study reveals a combined photoredox-catalyzed HAT and RPC mechanism utilizing *N*-alkoxypyridinium salts for direct access to 2° and 3° benzylic carbocations. Enabling ubiquitous C(sp³)–H bonds in organic compounds is an advantage of this approach and this protocol tolerates common organic functional groups on the N-heterocycles and is even mild enough to leave benzylic electrophiles untouched, leading to potential late-stage functionalization of pharmaceutical targets.

Declaration of interests

No interests have been declared.

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