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Theoretical spectra and energetics for c - C_3HC_2H , l - C_5H_2 , and bipyramidal D_{3h} C_5H_2

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The recent astronomical detection of c - C_3HC_2H and l - C_5H_2 has led to increased interest in C_5H_2 isomers and their relative stability. The present work provides the first complete list of anharmonic vibrational spectral data with infrared intensities for three such isomers as well as including the first set of rotational data for the bipyramidal C_5H_2 isomer allowing for these molecules to serve as potential tracers of interstellar carbon. All three isomers have fundamental vibrational frequencies with at least one notably intense fundamental frequency. The l - C_5H_2 isomer has, by far, the highest intensities out of the three isomers at 2076.3 cm^{-1} (738 km mol^{-1}) and 1887.5 cm^{-1} (182 km mol^{-1}). The c - C_3HC_2H isomer has one intense peak at 3460.6 cm^{-1} (84 km mol^{-1}), and the bipyramidal C_5H_2 isomer has one intense peak at 489.3 cm^{-1} (78 km mol^{-1}). The relative intensities highlight that while l - C_5H_2 is not the lowest energy isomer, its notable intensities should make it more detectable in the infrared than the lower energy c - C_3HC_2H form. The bipyramidal isomer is firmly established here to lie $44.98\text{ kcal mol}^{-1}$ above the cyclic form. The explicitly correlated coupled cluster rovibrational spectral data presented herein should assist with future laboratory studies of these C_5H_2 isomers and aid in detection in astronomical environments especially through the newly operational James Webb Space Telescope.

KEYWORDS

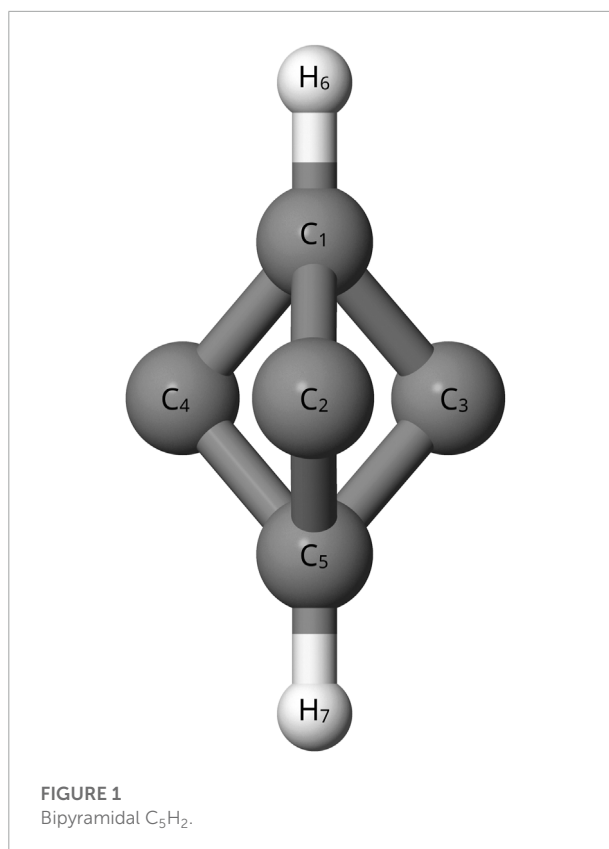
rotational spectroscopy, vibrational spectroscopy, infrared observations, coupled cluster theory, astrochemistry, carbon chemistry

1 Introduction

The unidentified infrared bands (UIRs) are unattributed spectral features observed in virtually all types of astronomical objects whose provenance is not firmly established (Peeters et al., 2002). The UIRs were first recognized in 1973 in the 8 – $13\text{ }\mu\text{m}$ region and have since been detected in nebulae and in the diffuse interstellar medium (ISM) among other astronomical environments (Peeters et al., 2002). The UIRs were first believed to originate from mostly grain mantle molecules (Allamandola et al., 1979). However, since 1984 most of the UIR spectral features have been hypothesized to be caused

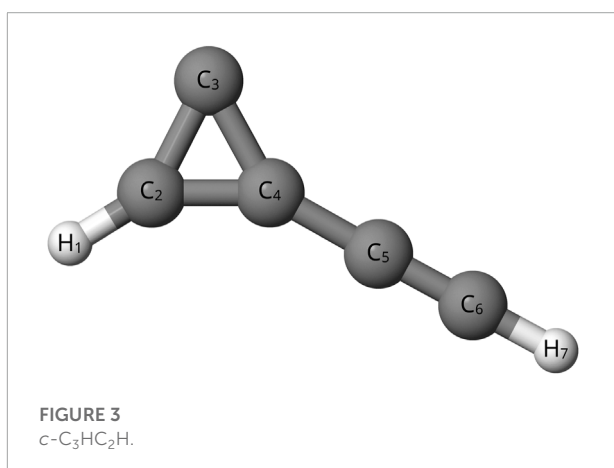
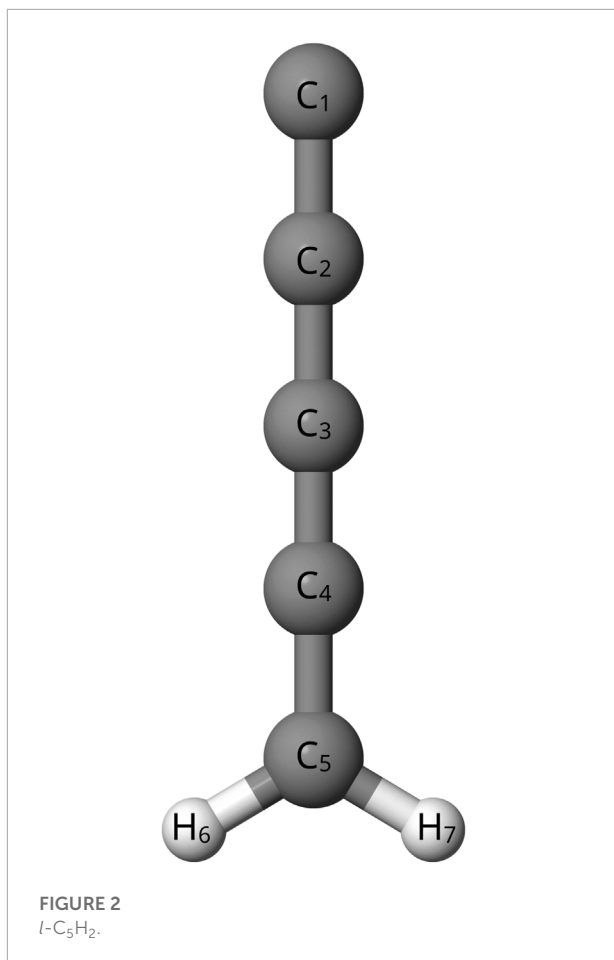
by polycyclic aromatic hydrocarbons (PAHs) (Leger and Puget, 1984; Sellgren, 1984; Molster et al., 2001). One difficulty with supporting this so-called PAH hypothesis is that most PAHs are not rotationally active at worst or weakly dipolar at best. This leads many PAHs to be poor candidates for rotational astronomical detection, which has been the primary vehicle for most astrochemical molecular observation including the recent detection of PAH-like molecules (Cabezas et al., 2021; Cernicharo et al., 2021). However, some small PAH-like molecules, like isomers of C_5H_2 , have a lower symmetry which can allow for a nonzero dipole moment and possible rotational activity. Recently, two isomers of C_5H_2 , *l*- C_5H_2 and ethynyl cyclopropenylidene (*c*- C_3HC_2H), have been independently detected in Taurus Molecular Cloud-1 and pathways of formation have been determined experimentally and supported theoretically (He et al., 2020; Cabezas et al., 2021; Cernicharo et al., 2021; Karton and Thimmakond, 2022). The detection of C_5H_2 isomers could allude to other rotationally dark PAHs lurking in the interstellar medium. Insight into the presence, and identity, of PAHs in various astronomical environments can shed light on the chemical and physical aspects of these environments and how they contribute to the UIRs (Tielens, 2008). Additionally the recently launched *James Webb Space Telescope* (JWST) will allow for more detections through vibrational/infrared spectroscopy which can be used for the detection of PAHs that previously could not be detected through rotational spectroscopy.

In addition to the astronomical interest, previous theoretical works have disagreed about the relative energy of C_5H_2 isomers. These have been quantum chemically computed using coupled cluster theory at the CCSD/DZP, CCSD/TZ2P, and CCSD(T)/cc-pVTZ levels, but there are still discrepancies in the relative energies (Balaji and Michl, 1988; Seburg et al., 1997; Veis et al., 2008). Specifically, the bipyramidyl C_5H_2 isomer, shown in Figure 1, has an inconsistent magnitude of relative energy when compared to other C_5H_2 isomers depending on the level of theory utilized (Balaji and Michl, 1988; Seburg et al., 1997; Veis et al., 2008). A previous study at the HF/6-31G* level (Pople et al., 1993) places bipyramidyl C_5H_2 100 kcal mol⁻¹ higher than other C_5H_2 isomers while another study utilizing MBPT(2)/6-31G* puts the bipyramidyl C_5H_2 isomer only 30 kcal mol⁻¹ higher (Balaji and Michl, 1988; Pople et al., 1993; Seburg et al., 1997). Furthermore, previous density functional and coupled cluster theory computations also place the bipyramidyl C_5H_2 isomer 47–70 kcal mol⁻¹ higher than other isomers (Seburg et al., 1997; Karton and Thimmakond, 2022). Despite the different levels of theory agreeing that the bipyramidyl C_5H_2 isomer is not the lowest energy isomer, there is a clear disagreement on the magnitude of energy difference for the isomer (Balaji and Michl, 1988; Pople et al., 1993; Seburg et al., 1997; Karton and Thimmakond, 2022). Determining more accurate relative



energies is necessary because the ambient energy in the ISM allows for the formation of non-global-minimum structures. However, the energy can not be too much higher; a higher energy difference closer to 100 kcal mol⁻¹ is much more unreasonable for formation in the ISM (Seburg et al., 1997). Newer, higher-level computations could help to settle the disagreements as to the relative energy of the bipyramidal C_5H_2 structure compared to the other isomers of C_5H_2 . A further look into the *l*- C_5H_2 isomer, shown in Figure 2, and the *c*- C_3HC_2H isomer, shown in Figure 3, can provide a common reference to determine the relative energy of the bipyramidal C_5H_2 isomer. Additionally the *c*- C_3HC_2H isomer and the *l*- C_5H_2 isomer have been detected in the laboratory by Fourier transform spectrometry which provides a benchmark to the theoretical methods utilized herein (McCarthy et al., 1997; Travers et al., 1997).

The present work utilizes high level quantum chemical computations to compute molecular structures, energies, and spectroscopic data for the bipyramidal C_5H_2 isomer shown in Figure 1, the *l*- C_5H_2 isomer shown in Figure 2, and the *c*- C_3HC_2H isomer shown in Figure 3. The spectroscopic data is computed using quartic force fields (QFFs) which are fourth-order Taylor series expansions of the internuclear potential energy portion of the Watson Hamiltonian (Fortenberry and Lee, 2019). QFFs based on coupled cluster theory at



the singles, doubles, and perturbative triples level within the explicitly correlated F12b construction [CCSD(T)-F12b] (Adler et al., 2007; Knizia et al., 2009) and a triple- ζ basis set can frequently achieve agreement to within 5–7 cm⁻¹ of gas-phase experimental vibrational frequencies (Gardner et al., 2021). However a shortcoming of this method is that the rotational constants are only accurate to about 60 MHz

(Agbaglo et al., 2019; Agbaglo and Fortenberry, 2019a,b; Watrous et al., 2021). An improvement would be to utilize a composite method combining core-correlated CCSD(T)-F12b with the cc-pCVTZ-F12 basis set and a correction for scalar relativity (Watrous et al., 2021). This method is referred to as F12-TZ-cCR and offers closer agreement, compared to previous methods, to experimental rotational constants on the order of 7.5 MHz which is necessary for accurate detection (Westbrook et al., 2021b; Watrous et al., 2021). Thus the fundamental vibrational frequencies and rotational constants at this level of theory provided here will lend theoretical support to experiments that will help NASA missions such as the *Stratospheric Observatory for Infrared Astronomy* (SOFIA) and JWST to facilitate the detection of these isomers in space.

2 Computational details

All four levels of theory in the present work utilize CCSD(T)-F12b (Raghavachari et al., 1989; Adler et al., 2007; Knizia et al., 2009; Shavitt and Bartlett, 2009). The first two QFFs use the cc-pVXZ-F12 basis sets (Peterson et al., 2008; Yousaf and Peterson, 2008; Knizia et al., 2009) where X is D or T. This combination is abbreviated as F12-XZ in the following. F12-DZ and F12-TZ QFFs are utilized for all three isomers. The other two levels of theory are based on core-correlated CCSD(T)-F12b with the cc-pCVXZ-F12 basis sets, (Watrous et al., 2021), where X is again D or T. Additionally this method includes a correction for scalar relativity computed with canonical CCSD(T) with a cc-pVTZ-DK basis set (Douglas and Kroll, 1974; Jansen and Hess, 1989). This combination is hereafter referred to as F12-XZ-cCR. F12-DZ-cCR and F12-TZ-cCR QFFs are only utilized for the *c*-C₃HC₂H isomer due to the substantial increase in computational cost. The QFF for the *c*-C₃HC₂H isomer is composed of 20,641 displacements from the equilibrium geometry along the symmetry-internal coordinates (SICs) shown below with atom labels corresponding to Figure 3.

$$S_1(a') = r(H_1 - C_2) \quad (1)$$

$$S_2(a') = r(C_2 - C_3) \quad (2)$$

$$S_3(a') = r(C_3 - C_4) \quad (3)$$

$$S_4(a') = r(C_4 - C_5) \quad (4)$$

$$S_5(a') = r(C_4 - C_6) \quad (5)$$

$$S_6(a') = r(C_4 - H_7) \quad (6)$$

TABLE 1 Relative energy of C₅H₂ isomers in kcal mol⁻¹ compared to previous experiment.

Isomer	F12-DZ	F12-TZ	CCSD(T)/cc-pVTZ ^a	CCSDT(Q)/CBS ^b
c-C ₃ HC ₂ H	0.00	0.00	2.01	0.66
l-C ₅ H ₂	13.86	13.79	13.82	13.53
bipyramidal C ₅ H ₂	45.90	44.98		47.51

^aFrom Ref. (Seburg et al., 1997) where energies are relative to a lower energy isomer.

^bFrom Ref. (Karton and Thimmakonda, 2022) where energies are relative to a lower energy isomer.

TABLE 2 Rotational constants of l-C₅H₂ compared to previous experiment^a.

Constant	Units	F12-DZ		F12-TZ		Exp. ^a
		0.005	0.010	0.005		
A _e	MHz	290162.5	290162.5	290123.6		
B _e	MHz	2290.1	2290.1	2296.4		
C _e	MHz	2272.1	2272.1	2278.4		
A ₀	MHz	286248.3	286245.9	287832.8	277600	
B ₀	MHz	2294.6	2294.5	2307.8	2304.7844	
C ₀	MHz	2275.7	2275.7	2286.9	2285.8053	
D _J	Hz	92.822	92.831	91.191	104	
D _{JK}	kHz	70.773	71.920	2,642.5	46.4	
D _K	MHz	22.182	22.180	19.676		
d ₁	mHz	-771.6	-771.6	-775.9		
d ₂	mHz	-285.6	-282.0	1,081.6		
H _J	nHz	-558.98	-516.46	-216,800		
H _{JK}	mHz	238.14	232.78	11,321		
H _{KJ}	kHz	-1.8080	-1.4656	-478.98		
H _K	kHz	6.9698	6.6280	484.19		
h ₁	nHz	-50.873	-105.28	-36970		
h ₂	mHz	1.1503	1.0644	-0.015785		
h ₃	nHz	121.91	115.04	-2042.6		
μ	D			5.94		
κ		-0.99987	-0.99987	-0.99985		

^aFrom McCarthy et al., 1997.

$$S_7(a') = \angle(H_1 - C_2 - C_3) \quad (7)$$

$$S_8(a'') = \angle(C_2 - C_3 - C_4) \quad (8)$$

$$S_9(a'') = \angle(C_3 - C_4 - C_5) \quad (9)$$

$$S_{10}(a'') = \angle(C_3 - C_4 - C_6) \quad (10)$$

$$S_{11}(a'') = \angle(C_3 - C_4 - H_7) \quad (11)$$

$$S_{12}(a'') = \tau(H_1 - C_2 - C_3 - C_4) \quad (12)$$

$$S_{13}(a') = \tau(C_2 - C_3 - C_4 - C_5) \quad (13)$$

$$S_{14}(a'') = \tau(C_2 - C_3 - C_4 - C_6) \quad (14)$$

$$S_{15}(a') = \tau(C_2 - C_3 - C_4 - H_7) \quad (15)$$

The QFFs for the l-C₅H₂ and bipyramidal C₅H₂ isomers are run directly in Cartesian coordinates composed of 46,064 points and 38,968 points respectively. This increase in the number of points for Cartesians instead of SICs exacerbates the increase in computational cost that would be needed for the F12-DZ-cCR and F12-TZ-cCR methods. For the SIC-based QFFs, the geometry is first optimized at each of the four levels of theory, then each geometry is displaced by 0.005 Å or 0.005 radians along the SICs to form the QFF. For the c-C₃HC₂H isomer, in addition to the previous displacements, the out-of-plane bends (OPB) are instead displaced by 0.010 Å or 0.010 radians to help combat the well-known OPB problem (Westbrook et al., 2020; Lee and Fortenberry, 2021). At each of the displaced geometries, single-point energies are calculated. The resulting energies are then fit by a least-squares procedure to yield the optimized geometry. A refit to the function minimum zeroes the gradients and yields the new equilibrium geometry as well as the corresponding force constants. The force constants are then converted from SICs to Cartesian coordinates by using the INTDER program (Allen and coworkers, 2005). For the Cartesian QFFs, each geometry is first optimized at the F12-DZ and F12-TZ levels of theory. Then, each geometry is displaced by 0.005 Å along the Cartesian x, y, and z axes. For l-C₅H₂ at the F12-DZ level of theory, each geometry was displaced by 0.010 Å in the out-of-plane coordinates. The force constants are computed numerically by central finite differences (Westbrook et al., 2021b).

For all of the QFFs, the rovibrational spectral data is computed using the SPECTRO software package (Gaw et al., 1991) which uses the Cartesian force constants in vibrational perturbation theory at second order (Watson, 1977; Papoušek and Aliev, 1982; Franke et al., 2021) and second-order rotational perturbation theory (Mills, 1972). Type-1 and -2 Fermi resonances, Fermi polyads (Martin and Taylor, 1997) and Coriolis resonances are taken into account to further increase the accuracy of the rovibrational data (Martin et al., 1995; Martin and Taylor, 1997). These are shown in Supplementary Tables S1–S3 of the Supplemental Information (SI) along with the optimized

TABLE 3 Rotational constants of *c*-C₃HC₂H compared to previous experiment^a.

Constant	Units	F12-DZ		F12-DZ-cCR		F12-TZ	F12-TZ-cCR	Exp. ^a
		0.005	0.010	0.005	0.010	0.010	0.010	
<i>A_e</i>	MHz	34593.7	34593.7	34768.1	34768.1	34629.3	34835.3	
<i>B_e</i>	MHz	3410.1	3410.1	3424.5	3424.5	3413.1	3429.6	
<i>C_e</i>	MHz	3104.1	3104.1	3117.5	3,117.5	3106.7	3122.3	
<i>A₀</i>	MHz	34386.9	34398.1	34578.6	34578.7	34449.4	34654.1	34638.701
<i>B₀</i>	MHz	3406.0	3406.2	3420.9	3420.7	3409.9	3426.0	3424.8768
<i>C₀</i>	MHz	3095.8	3096.1	3,109.8	3109.6	3099.6	3114.6	3113.6386
<i>D_J</i>	Hz	245.00	245.00	246.72	246.71	244.68	246.14	288
<i>D_K</i>	kHz	29.813	29.801	29.780	29.788	29.659	29.795	29.5
<i>D_{JK}</i>	kHz	105.97	105.80	107.72	106.72	105.79	106.57	
<i>d₁</i>	Hz	-28.524	-28.523	-28.662	-28.663	-28.472	-28.589	
<i>d₂</i>	Hz	-19.625	-19.617	-19.581	-19.586	-19.524	-19.570	
<i>H_J</i>	μHz	-173.88	-173.55	-172.98	-172.73	-170.73	-171.01	
<i>H_{JK}</i>	mHz	106.73	106.531	106.24	105.97	104.63	105.22	
<i>H_{KJ}</i>	Hz	-4.0028	-3.9978	-3.9473	-3.9491	-3.8877	-39.167	
<i>H_K</i>	Hz	4.0870	4.0827	4.0363	4.0384	3.9812	40.157	
<i>h₁</i>	μHz	-13.184	-13.165	-13.022	-13.040	-12.987	-12.952	
<i>h₂</i>	μHz	90.284	90.145	89.750	89.589	88.667	88.905	
<i>h₃</i>	μHz	18.237	18.222	18.092	18.098	17.999	18.007	
<i>μ</i>	D					3.59		
<i>κ</i>		-0.98018	-0.98019	-0.98023	-0.98023	-0.9802	-0.98025	

^aFrom Travers et al., 1997.TABLE 4 Rotational constants of bipyramidal C₅H₂.

Constant	Units	F12-DZ	F12-TZ
<i>B_e</i>	MHz	12733.1	12742.1
<i>C_e</i>	MHz	9145.2	9153.0
<i>B₀</i>	MHz	12647.4	12656.6
<i>C₀</i>	MHz	9055.4	9063.7
<i>D_J</i>	KHz	2.979	2.979
<i>D_{JK}</i>	kHz	6.770	6.718
<i>D_K</i>	kHz	1.676	1.751
<i>d₁</i>	mHz	48.6	2.06
<i>d₂</i>	mHz	8.42	1.21
<i>H_J</i>	mHz	1.736	1.746
<i>H_{JK}</i>	mHz	-10.886	-11.033
<i>H_{KJ}</i>	mHz	9.011	9.935
<i>H_K</i>	mHz	21.254	20.669
<i>h₁</i>	μHz	0.702	-1.215
<i>h₂</i>	μHz	1.282	-2.913
<i>h₃</i>	μHz	263.914	257.523

geometrical parameters and *T*₁ diagnostics shown in [Supplementary Tables S4–S6](#). The anharmonic zero point vibrational energy (ZPVE) from the refit geometry, refit energy (only for the SIC QFF), and energy from the optimized minima are added together to calculate the relative energy.

The geometry optimizations, harmonic frequencies, single point energies, and dipole moments are computed using the Molpro 2020.1 software package (Werner et al., 2020). The dipole moments are computed at the F12-TZ level of theory. Double-harmonic and anharmonic infrared intensities are computed using Gaussian16 (Frisch et al., 2016) with the MP2/aug-cc-pVTZ (Møller and Plesset, 1934; Kendall et al., 1992), B3LYP/aug-cc-pVTZ (Yang et al., 1986; Lee et al., 1988; Becke, 1993), and B3LYP/N07D (Barone et al., 2008) levels of theory. Previous work has shown semi-quantitative accuracy for harmonic infrared intensities for the MP2 level of theory for a substantial decrease in cost relative to higher-level computations (Yu et al., 2015; Finney et al., 2016; Westbrook et al., 2021a).

3 Results and discussion

The relative energies of *l*-C₅H₂, *c*-C₃HC₂H, and bipyramidal C₅H₂ are compared at the F12-DZ and the F12-TZ levels of theory as shown in [Table 1](#). The relative energies of the *c*-C₃HC₂H and *l*-C₅H₂ isomers are similar to previous (Seburg et al., 1997) relative energy computations with the *l*-C₅H₂ isomer being 13.79 kcal mol⁻¹ higher in energy than the *c*-C₃HC₂H isomer. However, the bipyramidal C₅H₂ isomer is lower in relative energy

TABLE 5 F12-DZ and F12-TZ harmonic and fundamental vibrational frequencies and B3LYP/N07D intensities (in km mol⁻¹) of *l*-C₅H₂ compared to previous theory computed at the fc-CCSD(T)/cc-pVTZ level of theory^b.

Description	F12-DZ		F12-TZ	B3LYP/N07D		Previous theory ^b		
	0.005	0.010	0.005	Freq	<i>f</i>	Freq	<i>f</i>	
$\omega_1(b_2)$	C-H Anti-Sym Stretch	3,219.4	3219.4	3215.5	3,199.1	1	3,215.1	0.0
$\omega_2(a_1)$	C-H Sym Stretch	3127.7	3127.7	3125.1	3117.5	1	3127.3	0.6
$\omega_3(a_1)$	(C ₁ -C ₂) - (C ₂ -C ₃) + (C ₃ -C ₄) - (C ₄ -C ₅)	2145.3	2,145.3	2,153.5	2212.7	806	2,151.6	677.1
$\omega_4(a_1)$	(C ₁ -C ₂) - (C ₂ -C ₃) - (C ₃ -C ₄) + (C ₄ -C ₅)	1924.4	1924.4	1935.2	1974.6	206	1921.8	176.6
$\omega_5(a_1)$	H-C-C Sym Bend	1503.9	1503.9	1506.9	1513.8	9	1,503.5	7.7
$\omega_6(a_1)$	(C ₁ -C ₂) + (C ₂ -C ₃) - (C ₃ -C ₄) - (C ₄ -C ₅)	1349.5	1349.5	1355.8	1369.9	7	1346.9	18.0
$\omega_7(b_2)$	H-C-C Anti-Sym Bend	1033.7	1,033.7	1030.7	1031.0	1	1030.3	0.4
$\omega_8(b_1)$	C ₅ OPB	936.0	936.3	941.4	968.6	34	943.4	31.6
$\omega_9(a_1)$	(C ₁ -C ₂) + (C ₂ -C ₃) + (C ₃ -C ₄) + (C ₄ -C ₅)	749.0	749.0	752.2	769.8	1	745.5	0.4
$\omega_{10}(b_1)$	C ₃ -C ₄ OPB	534.0	534.3	536.1	656.4	7	576.2	2.8
$\omega_{11}(b_2)$	C ₂ -C ₃ IPB	398.4	398.4	411.7	527.6	4	445.4	0.9
$\omega_{12}(b_1)$	C ₂ -C ₄ OPB	244.8	244.7	239.9	314.8	7	270.9	8.5
$\omega_{13}(b_2)$	C ₂ -C ₄ IPB	231.1	231.1	230.9	286.2	7	254.9	5.7
$\omega_{14}(b_2)$	C ₁ -C ₅ IPB	133.3	133.3	123.2	148.3	1	139.3	0.0
$\omega_{15}(b_1)$	C ₁ OPB	110.8	110.7	104.3	143.7	2	122.3	3.1
ZPVE		8758.4	8812.5	8948.0				
$\nu_1(b_2)$	C-H Anti-Sym Stretch	3073.7	3066.1	3078.2	3047.4	1		
$\nu_2(a_1)$	C-H Sym Stretch	3019.2	3011.6	2,955.4	3013.1	2		
$\nu_3(a_1)$	(C ₁ -C ₂) - (C ₂ -C ₃) + (C ₃ -C ₄) - (C ₄ -C ₅)	2098.9	2090.6	2076.3	2170.4	738		
$\nu_4(a_1)$	(C ₁ -C ₂) - (C ₂ -C ₃) - (C ₃ -C ₄) + (C ₄ -C ₅)	1898.0	1894.9	1887.5	1944.1	182		
$\nu_5(a_1)$	H-C-C Sym Bend	1472.0	1467.4	1,43.8	1483.5	6		
$\nu_6(a_1)$	(C ₁ -C ₂) + (C ₂ -C ₃) - (C ₃ -C ₄) - (C ₄ -C ₅)	1340.7	1334.1	1318.9	1350.5	6		
$\nu_7(b_2)$	H-C-C Anti-Sym Bend	1022.5	1018.7	1155.4	1011.6	1		
$\nu_8(b_1)$	C ₅ OPB	924.6	923.8	769.0	944.8	35		
$\nu_9(a_1)$	(C ₁ -C ₂) + (C ₂ -C ₃) + (C ₃ -C ₄) + (C ₄ -C ₅)	741.0	730.1	756.4	772.4	1		
$\nu_{10}(b_1)$	C ₃ -C ₄ OPB	534.0 ^a	534.3 ^a	536.1 ^a	365.8	7		
$\nu_{11}(b_2)$	C ₂ -C ₃ IPB	398.4 ^a	398.4 ^a	411.7 ^a	285.8	2		
$\nu_{12}(b_1)$	C ₂ -C ₄ OPB	224.8 ^a	244.7 ^a	239.9 ^a	244.3	4		
$\nu_{13}(b_2)$	C ₂ -C ₄ IPB	231.1 ^a	231.1 ^a	230.9 ^a	247.5	9		
$\nu_{14}(b_2)$	C ₁ -C ₅ IPB	133.3 ^a	133.3 ^a	123.2 ^a	151.6	1		
$\nu_{15}(b_1)$	C ₁ OPB	110.8 ^a	110.7 ^a	104.3 ^a	133.7	2		

^aHarmonic values reported.^bPrevious theory from Karton and Thimmakonda, 2022.

than some previous computations (Balaji and Michl, 1988; Pople et al., 1993; Seburg et al., 1997). The relative energy of the bipyramidal C₅H₂ isomer at the F12-TZ level of theory is 44.98 kcal mol⁻¹ compared to the *c*-C₃HC₂H isomer. This relative energy is much closer to the previous MBPT(2)/6-31G* calculations, at about 30 kcal mol⁻¹, as opposed to previous calculations at the HF/6-31G* level of theory where the relative energy was roughly 100 kcal mol⁻¹ higher than other C₅H₂ isomers (Balaji and Michl, 1988; Pople et al., 1993; Seburg et al., 1997). While this highly-symmetric isomer gives indication of possessing multi-reference effects due to its large *T*₁ diagnostic (0.059), the extent to which this will affect the relative energies is left for future work.

3.1 Rotational constants

Out of the three isomers, the *l*-C₅H₂ isomer has the largest dipole moment of 5.94 D and its equilibrium (ν_e) and vibrationally-averaged (ν_0) rotational constants, shown in

Table 2, demonstrate near prolate character ($\kappa = -0.99987$ for the F12-TZ level of theory, where κ is defined as $(2B_0 - A_0 - C_0)/(A_0 - C_0)$). This large dipole moment has helped with detection of the molecule through rotational spectroscopy. The F12-DZ and F12-TZ rotational constants agree well with a maximum relative difference of 0.57% in the *B*₀ rotational constant. Compared to experimental Fourier transform microwave spectroscopy (McCarthy et al., 1997), the F12-TZ *B*₀ and *C*₀ rotational constants differ from experiment by only 3.0 MHz (0.13%) and 1.1 MHz (0.05%) respectively. This benchmarking constrains the expected accuracy for other, unknown spectral observables provided in this work. The *A*₀ rotational constant differs from experiment by 10,200 MHz (3.67%), but the error for the *A*₀ rotational constant is known to have a higher error for near prolate molecules (Gardner et al., 2021). The complete set of the vibrationally excited rotational constants are reported in Supplementary Table S7 of the SI. The quartic and sextic distortion constants in the *S*-reduced Hamiltonian are reported

TABLE 6 Harmonic and fundamental vibrational frequencies of *c*-C₃HC₂H compared to previous theory computed at the fc-CCSD(T)/cc-pVTZ level of theory^b.

Description	F12-DZ		F12-DZ-cCR		F12-TZ	F12-TZ-cCR	Previous Theory ^b
	0.005	0.010	0.005	0.010	0.010	0.010	
$\omega_1(a')$ Acetylene C-H Stretch	3454.7	3454.7	3459.8	3459.8	3,454.2	3460.6	3457.0
$\omega_2(a')$ Ring CH Stretch	3261.1	3261.1	3266.4	3266.4	3,259.9	3266.7	3261.4
$\omega_3(a')$ C ₅ In-Plane Motion	2166.4	2166.4	2172.6	2172.6	2,167.9	2176.2	2163.8
$\omega_4(a')$ C ₄ -C ₅ Out-of-Plane Motion	1713.4	1713.4	1719.3	1719.3	1715.1	1722.9	1708.0
$\omega_5(a')$ (C ₂ -C ₃) + (C ₃ -C ₄) Stretch	1278.5	1,278.6	1,283.9	1283.9	1279.7	1286.5	1269.8
$\omega_6(a')$ (C ₂ -C ₃) - (C ₃ -C ₄) Stretch	1117.4	1117.4	1122.5	1122.5	1119.8	1126.1	1107.9
$\omega_7(a')$ C ₃ -C ₂ -H ₁ IPB	949.9	949.9	953.8	953.8	948.5	952.3	945.8
$\omega_8(a'')$ C ₃ -C ₂ -H ₁ OPB	892.9	893.2	895.9	895.9	893.7	898.2	891.2
$\omega_9(a'')$ C ₅ -C ₆ -H ₇ OPB	710.0	710.0	714.8	714.8	713.9	717.8	705.4
$\omega_{10}(a')$ C ₄ -C ₅ Stretch	696.0	696.0	698.6	698.6	697.5	700.9	692.1
$\omega_{11}(a')$ C ₅ -C ₆ -H ₇ IPB	604.4	604.4	608.7	608.7	611.3	615.6	601.8
$\omega_{12}(a'')$ C ₄ -C ₅ Out-Of Plane Motion	510.4	510.4	511.6	511.6	517.7	521.7	520.4
$\omega_{13}(a')$ C ₄ -C ₅ In Plane Motion	498.8	498.8	502.4	502.4	504.3	508.9	500.1
$\omega_{14}(a'')$ C ₅ -C ₆ Out-Of-Plane Motion	201.8	201.8	202.8	202.7	202.6	203.6	212.9
$\omega_{15}(a')$ C ₅ -C ₆ In-Plane Motion	189.2	189.2	190.7	190.7	190.8	192.0	193.0
ZPVE	7869.7	8908.0	9383.5	9115.5	9234.5	9069.4	
$\nu_1(a')$ Acetylene C-H Stretch	3412.6	3341.7	3311.3	3328.8	3346.4	3347.8	
$\nu_2(a')$ Ring CH Stretch	3034.4	3,102.4	3137.1	3139.1	3145.4	3,127.1	
$\nu_3(a')$ C ₅ In-Plane Motion	2168.0	2131.8	2,123.4	2,128.7	2,124.3	2135.5	
$\nu_4(a')$ C ₄ -C ₅ Out-of-Plane Motion	1673.6	1675.5	1675.2	1681.4	1685.3	1706.8	
$\nu_5(a')$ (C ₂ -C ₃) + (C ₃ -C ₄) Stretch	1138.0	1,199.5	1,262.1	1,265.1	1,285.6	1256.3	
$\nu_6(a')$ (C ₂ -C ₃) - (C ₃ -C ₄) Stretch	1194.6	1,113.6	1073.5	1090.0	1073.9	1061.0	
$\nu_7(a')$ C ₃ -C ₂ -H ₁ IPB	949.9 ^a	891.7	944.2	958.9	948.5 ^a	916.2	
$\nu_8(a'')$ C ₃ -C ₂ -H ₁ OPB	892.9 ^a	848.5	895.9 ^a	865.1	893.7 ^a	898.2 ^a	
$\nu_9(a'')$ C ₅ -C ₆ -H ₇ OPB	710.0 ^a	684.7	711.6	689.8	705.2	646.6	
$\nu_{10}(a')$ C ₄ -C ₅ Stretch	696.0 ^a	698.2	675.1	690.3	674.7	700.9 ^a	
$\nu_{11}(a')$ C ₅ -C ₆ -H ₇ IPB	555.1	574.6	574.4	585.6	595.0	577.9	
$\nu_{12}(a'')$ C ₄ -C ₅ Out-Of Plane Motion	510.4 ^a	510.4 ^a	511.6 ^a	511.6 ^a	517.7 ^a	521.7 ^a	
$\nu_{13}(a')$ C ₄ -C ₅ In Plane Motion	498.8 ^a	476.6	502.1	506.6	504.3 ^a	500.7	
$\nu_{14}(a'')$ C ₅ -C ₆ Out-Of-Plane Motion	201.8 ^a	201.8 ^a	202.8 ^a	202.7 ^a	202.6 ^a	203.6 ^a	
$\nu_{15}(a')$ C ₅ -C ₆ In-Plane Motion	189.2 ^a	189.2 ^a	190.7 ^a	190.7 ^a	190.8 ^a	169.5	

^aHarmonic values reported.^bPrevious theory from Karton and Thimmakondy, 2022.

in Table 2. The D_J constant is about 13 Hz lower than experiment while the D_{JK} constant differs from experiment by around 24 kHz at the F12-DZ level of theory, but has a much higher difference at the F12-TZ level of theory at around 2,596 kHz. While D_{JK} often exhibits minor effects on the rotational spectrum of near-prolate molecules (Stein et al., 2015), this erroneous l -C₅H₂ D_{JK} value for F12-TZ (and, in fact, all of its sextic distortion constants, as well) presently stems from a compounding error brought about by the step size and the resulting QFF improperly representing these data. The F12-DZ data are consistent within step size and experimental comparison implying that these data are reliable as are the F12-TZ data behaving in similar ways.

The *c*-C₃HC₂H isomer has a smaller dipole moment than the *l*-C₅H₂ isomer at 3.59 D. The *c*-C₃HC₂H isomer also shows near prolate character ($\kappa = -0.98026$ at the F12-TZ-cCR level of theory) as exhibited by the rotational constants B_0 and C_0 being almost the same as shown in Table 3. The F12-TZ-cCR data agrees extremely well with previous

Fourier transform spectrometry results (Travers et al., 1997), differing by only 1.1 MHz (0.03%) for the B_0 rotational constant, 0.96 MHz (0.03%) for the C_0 rotational constant, and even 15 MHz (0.04%) for the A_0 rotational constant. Both the F12-DZ-cCR and F12-TZ-cCR methods produce rotational constants closer to experiment than F12-DZ and F12-TZ which in line with previous work (Watrous et al., 2021). Doubling the step size for the OBPs has little-to-no effect in the rotational constants as expected. The S -reduced Hamiltonian quartic and sextic distortion constants are reported in Table 3, and the D_K constant compares well with experiment, only differing by 0.2 kHz at the F12-TZ-cCR level of theory. The experimental D_J rotational constant is 288 Hz while the computed D_J rotational constant is 246.14 Hz. However, the computed A -reduced Hamiltonian Δ_J , determined in this work, has a value of 285.28 Hz which aligns closely with the reported experimental value, suggesting a potential misattribution in the experimental work.

TABLE 7 Intensities in km mol^{-1} of $c\text{-C}_3\text{HC}_2\text{H}$ compared to previous theory computed at the fc-CCSD(T)/cc-pVTZ level of theory^a.

Description	MP2/TZ		B3LYP/TZ		B3LYP/N07D		Previous theory ^a	
	Freq	<i>f</i>	Freq	<i>f</i>	Freq	<i>f</i>	<i>f</i>	
$\omega_1(a')$	Acetylene C-H Stretch	3474.6	85	3461.2	90	3478.7	100	71.1
$\omega_2(a')$	Ring CH Stretch	3283.4	2	3234.4	1	3250.2	1	1.1
$\omega_3(a')$	C ₅ In-Plane Motion	2145.1	6	2192.6	3	2193.7	3	10.6
$\omega_4(a')$	C ₄ -C ₅ Out-of-Plane Motion	1703.2	4	1719.7	5	1724.9	6	5.0
$\omega_5(a')$	(C ₂ -C ₃) + (C ₃ -C ₄) Stretch	1286.9	32	1280.3	38	1275.1	39	41.2
$\omega_6(a')$	(C ₂ -C ₃) - (C ₃ -C ₄) Stretch	1128.9	3	1,108.7	8	1103.3	7	6.5
$\omega_7(a')$	C ₃ -C ₂ -H ₁ IPB	944.6	3	951.9	2	943.8	3	2.3
$\omega_8(a'')$	C ₃ -C ₂ -H ₁ OPB	902.0	13	899.9	15	880.8	17	17.1
$\omega_9(a'')$	C ₅ -C ₆ -H ₇ OPB	687.5	31	756.5	27	744.4	36	28.7
$\omega_{10}(a')$	C ₄ -C ₅ Stretch	704.2	2	708.8	1	706.5	1	1.7
$\omega_{11}(a')$	C ₅ -C ₆ -H ₇ IPB	614.5	45	610.1	54	595.4	65	49.2
$\omega_{12}(a'')$	C ₄ -C ₅ Out-Of Plane Motion	531.9	2	549.5	5	562.2	2	1.9
$\omega_{13}(a')$	C ₄ -C ₅ In Plane Motion	503.0	3	530.7	4	538.2	2	2.4
$\omega_{14}(a'')$	C ₅ -C ₆ Out-Of-Plane Motion	206.5	1	218.6	1	238.4	1	0.9
$\omega_{15}(a')$	C ₅ -C ₆ In-Plane Motion	188.0	7	202.3	7	211.9	6	5.7
$\nu_1(a')$	Acetylene C-H Stretch	3,351.6	78	3,335.6	81	3353.6	84	
$\nu_2(a')$	Ring CH Stretch	3,156.9	1	3106.0	1	3117.9	1	
$\nu_3(a')$	C ₅ In-Plane Motion	2113.8	3	2152.0	2	2152.1	2	
$\nu_4(a')$	C ₄ -C ₅ Out-of-Plane Motion	1667.0	3	1,677.7	4	1681.0	4	
$\nu_5(a')$	(C ₂ -C ₃) + (C ₃ -C ₄) Stretch	1253.1	29	1,249.3	36	1245.1	34	
$\nu_6(a')$	(C ₂ -C ₃) - (C ₃ -C ₄) Stretch	1107.2	1	1078.6	3	1075.4	3	
$\nu_7(a')$	C ₃ -C ₂ -H ₁ IPB	921.5	4	930.7	3	922.3	4	
$\nu_8(a'')$	C ₃ -C ₂ -H ₁ OPB	886.1	13	885.1	15	871.0	20	
$\nu_9(a'')$	C ₅ -C ₆ -H ₇ OPB	673.0	31	742.5	28	691.3	30	
$\nu_{10}(a')$	C ₄ -C ₅ Stretch	703.3	1	696.9	1	696.1	1	
$\nu_{11}(a')$	C ₅ -C ₆ -H ₇ IPB	613.0	43	592.7	55	555.0	52	
$\nu_{12}(a'')$	C ₄ -C ₅ Out-Of Plane Motion	532.8	3	545.0	4	508.1	1	
$\nu_{13}(a')$	C ₄ -C ₅ In Plane Motion	497.4	4	526.4	4	491.7	9	
$\nu_{14}(a'')$	C ₅ -C ₆ Out-Of-Plane Motion	210.5	1	220.2	1	212.0	1	
$\nu_{15}(a')$	C ₅ -C ₆ In-Plane Motion	192.2	7	201.7	7	195.7	5	

^aPrevious theory from [Karton and Thimmakonda, 2022](#).

Unlike the other two isomers, the bipyramidal C₅H₂ isomer does not have a dipole due to its D_{3h} symmetry. Even so, the rotational constants for the bipyramidal C₅H₂ isomer are given in [Table 4](#) for completeness. Additionally, the molecule is oblate and has only B₀ and C₀ rotational constants. The bipyramidal C₅H₂ molecule has not been detected in astrophysical observations or even observed experimentally. The F12-DZ and the F12-TZ methods have good agreement differing by only 9.2 MHz for the C₀ rotational constant. Hence, the results presented here provide the most accurate predictions of its rotational constants which may be helpful for the accurate kinetic rate determination or the thermochemical analysis that requires the full rotational partition function.

3.2 Vibrational frequencies

The *l*-C₅H₂ isomer has the largest anharmonic intensities out of the three C₅H₂ isomers, as shown in [Table 5](#). The two most intense modes are ν_3 with an intensity of 738 km mol^{-1} at 2076.3 cm^{-1} (4.81 μm) and ν_4 with an intensity of 182 km mol^{-1} at 1887.5 cm^{-1} (5.30 μm). These intensities represent the

movements in the linear carbon chain and are up to ten times larger than the anti-symmetric stretch in water, which has an intensity of about 70 km mol^{-1} at the same level of theory. The large intensities of these two frequencies imply that *l*-C₅H₂ should be observable right at the edge of the NIRspec instrument onboard JWST in a region often bereft of known transitions for organic molecules. Out of the three levels of theory in this work used to compute intensities, only the B3LYP/N07D results are reported for the *l*-C₅H₂ isomer as it is the only method with all real frequencies in line with the present high-level computations. The previous theory used to compute the intensities agrees qualitatively with the order of intensities computed with B3LYP/N07D. Additionally, the N07D basis set is designed to handle hydrocarbon systems like this and has been used to produce anharmonic frequencies in reasonable agreement with experiment ([Barone et al., 2008](#); [Barone et al., 2013](#)). However in this case, the anharmonic frequencies produced by B3LYP/N07D fall short in comparison with the fundamental frequencies produced by the F12-TZ level of theory. In contrast, the F12-DZ and F12-TZ anharmonic frequencies have relatively close agreement. The exceptions to this are the modes involving an in-plane bend (IPB) or OPB of

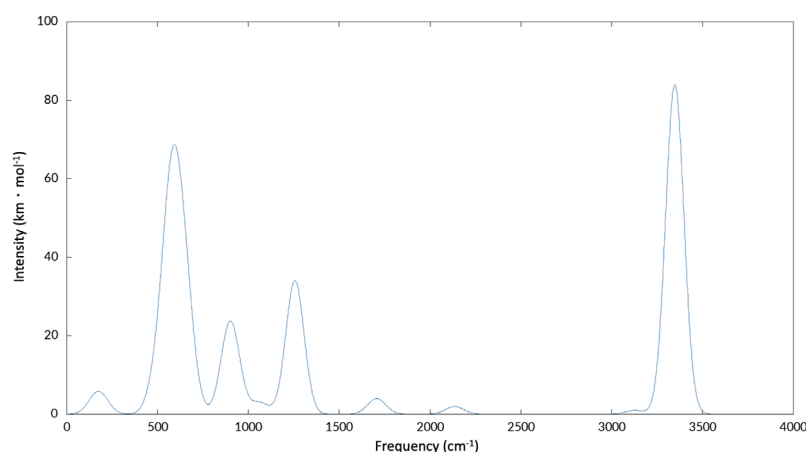


FIGURE 4
Simulated spectrum of $c\text{-C}_3\text{H}_2$.

TABLE 8 F12-DZ and F12-TZ harmonic and fundamental frequencies of bipyramidal C_5H_2 with intensities in km mol^{-1} compared to previous theory computed at the fc-CCSD(T)/cc-pVTZ level of theory^a.

Description	F12-DZ	F12-TZ	MP2/TZ		B3LYP/TZ		B3LYP/N07D		Previous theory ^a	
			Freq	f	Freq	f	Freq	f	Freq	f
$\omega_1 (a_1')$ C-H Sym Stretch	3331.6	3329.5	3376.0	–	3311.0	–	3319.9	–	3331.6	0.0
$\omega_2 (a_2'')$ C-H Anti-Sym Stretch	3330.0	3327.8	3,374.7	30	3309.4	11	3318.5	9	3330.0	12.3
$\omega_3 (a_1')$ C ₁ -C ₅ Stretch	1307.1	1308.7	1,302.2	–	1314.5	–	1329.5	–	1299.7	0.0
$\omega_4 (a_2'')$ C ₁ -C Ring - C ₅ -C Ring Stretch	1288.3	1290.4	1,298.0	29	1,289.5	28	1302.8	24	1276.2	19.6
$\omega_5 (e')$ H-CC Sym Bend	1128.2	1126.8	1079.6	5	1129.2	10	1136.8	10	1129.0	8.3
$\omega_6 (e'')$ H-CC Anti-Sym Bend	1010.9	1008.3	978.1	–	1017.8	–	1,024.1	–	1011.5	0.0
$\omega_7 (a_1')$ C ₁ -C ₂ ,C ₃ ,C ₄ -C ₅ Stretch	852.2	851.5	913.1	–	837.5	–	834.4	–	846.8	0.0
$\omega_8 (e')$ C ₂ -C ₃ - C ₂ -C ₄ Stretch	842.5	844.1	813.4	30	850.4	15	857.5	16	839.4	20.4
$\omega_9 (e'')$ C ₁ -C ₂ - C ₂ -C ₅ Stretch	783.8	786.2	807.0	–	734.2	–	771.1	–	777.1	0.0
$\omega_{10} (e')$ C ₂ -C ₃ + C ₂ -C ₄ Stretch	488.7	489.3	382.4	82	525.9	48	530.7	46	490.7	49.9
ZPVE	9228.0	9187.4								
$\nu_1 (a_1')$ C-H Sym Stretch	3190.5	3190.4	3,240.7	1	3187.4	1	3189.1	1		
$\nu_2 (a_2'')$ C-H Anti-Sym Stretch	3146.4	3139.3	3239.6	30	3187.0	10	3188.8	7		
$\nu_3 (a_1')$ C ₁ -C ₅ Stretch	1250.9	1253.0	1271.2	1	1262.9	1	1283.3	1		
$\nu_4 (a_2'')$ C ₁ -C Ring - C ₅ -C Ring Stretch	1273.4	1273.6	1269.6	27	1318.5	27	1334.3	23		
$\nu_5 (e')$ H-CC Sym Bend	1126.4	1108.7	1048.7	5	1103.8	9	1108.7	11		
$\nu_6 (e'')$ H-CC Anti-Sym Bend	1009.1	995.2	953.7	–	1001.5	–	1001.2	–		
$\nu_7 (a_1')$ C ₁ -C ₂ ,C ₃ ,C ₄ -C ₅ Stretch	834.4	832.4	914.3	1	850.4	3	847.0	1		
$\nu_8 (e')$ C ₂ -C ₃ - C ₂ -C ₄ Stretch	830.1	829.5	800.2	30	836.9	12	842.3	14		
$\nu_9 (e'')$ C ₁ -C ₂ - C ₂ -C ₅	756.3	759.2	770.8	–	719.7	–	752.9	–		
$\nu_{10} (e')$ C ₂ -C ₃ + C ₂ -C ₄	452.6	426.1	344.9	78	524.2	50	526.6	47		

^aPrevious theory from [Karton and Thimmakonda, 2022](#).

the carbon chain, ν_{10} - ν_{15} . The portions of the QFF involving these modes have problems with the cubic and quartic force constants. Those modes are reported as the harmonic values, and most anharmonic corrections are within the margin of error for the levels of theory. A larger step size is also used at the F12-DZ level of theory to combat the OPB problem ([Fortenberry et al., 2018](#); [Westbrook et al., 2020](#); [Lee and Fortenberry, 2021](#); [Lee and Fortenberry, 2021](#)). However there is no significant improvement

of the issue. As a result, the larger step size is not repeated for the F12-TZ level of theory. This will not affect possible detection *via* JWST because the NIRSpec and MIRI instruments on the JWST only capture spectra to about 333 cm^{-1} , but the resolution drops for the lower frequencies below 2000 cm^{-1} since the larger range MIRI instrument does not have the same capabilities as NIRSpec. The harmonic frequencies for both the F12-DZ and F12-TZ levels of theory agree well with previous theoretical harmonic

frequencies for most modes (Karton and Thimmakonda, 2022), but vary more for the OPB and IPB which is likely due to the aforementioned OPB problem (Westbrook et al., 2020; Lee and Fortenberry, 2021).

For the *c*-C₃HC₂H isomer, the vibrational frequencies are in Table 6, and the double-harmonic intensities are in Table 7. The *c*-C₃HC₂H isomer has two intense peaks with the most intense peak being the acetylene C-H stretch, ν_1 , at 3,347.8 cm⁻¹ (2.99 μ m) with an anharmonic intensity of 84 km mol⁻¹ at the B3LYP/N07D level of theory. The second most intense peak is the C₅-C₆-H₇ in-plane bend (IPB), ν_{11} , at 577.9 cm⁻¹ (17.3 μ m) with an intensity of 52 km mol⁻¹. The intensity changes depending on which level of theory is used to calculate the double-harmonic intensities, but all three methods computed for the intensities agree on the qualitative order of intensities for the modes, as does the previous theory (Karton and Thimmakonda, 2022). The intensities for *c*-C₃HC₂H are not as strong as the intensities for *l*-C₅H₂, but are still strong enough for detection.

For the *c*-C₃HC₂H, there are differences in the anharmonic frequencies between the levels of theory and step sizes as shown in Table 6. Fundamental frequencies ν_{12} , ν_{14} , and ν_{15} produce questionable positive anharmonicities, although they are below or on the lower edge of the range to be detected through the aforementioned instruments on JWST. Additionally, these three modes have a maximum intensity of 5 km mol⁻¹. Both considerations make these frequencies unlikely to be detected. Increasing the level of theory to F12-DZ-cCR improves the anharmonic correction for the ν_9 , ν_{10} , and ν_{13} fundamentals. The doubled step size in the OPB coordinates further improves the anharmonic correction for F12-DZ and F12-DZ-cCR. As a result only the doubled step size is calculated at the higher levels of theory due to the increased computational cost. Similarly, this doubled step size helps the anharmonic correction for ν_7 and ν_8 with F12-DZ and F12-DZ-cCR. There is much less variation in the harmonic frequencies between the levels of theory. Additionally, the harmonic frequencies compare well to previous theory (Karton and Thimmakonda, 2022). Figure 4 shows a simulated spectrum of the F12-TZ-cCR anharmonic frequencies and the B3LYP/N07D intensities with all bands incorporated; only those with notable intensities are present.

For the bipyramidal C₅H₂ isomer, the vibrational frequencies for the F12-DZ and F12-TZ QFFs, shown in Table 8, have close agreement for most of the frequencies. The bipyramidal C₅H₂ isomer has one intense frequency in which is ν_{10} , the symmetric C₂-C₃ + C₂-C₄ stretch, with an intensity of 78 km mol⁻¹ at 426.1 cm⁻¹ (23.5 μ m) which is slightly smaller than the *c*-C₃H₂H isomer and much smaller than the most intense peaks for the *l*-C₅H₂ isomer. Unlike the other isomers, the bipyramidal C₅H₂ isomer only has three other fundamentals that have intensities over 10 km mol⁻¹ at the B3LYP/N07D level of theory. These are ν_4 at 1,273.6 cm⁻¹ (7.85 μ m), ν_5 at 1,108.7 cm⁻¹ (9.02 μ m), and ν_8 at 829.5 cm⁻¹ (12.0 μ m). While the different methods differ in the

quantitative value of the intensities, they again agree qualitatively with each other and with previous theory on what will be the most and least intense peaks (Karton and Thimmakonda, 2022). However only having one fundamental with a large intensity could increase the difficulty for the detection of the bipyramidal C₅H₂ isomer compared to the other two isomers. The F12-DZ and the F12-TZ QFFs show good agreement for the harmonic and anharmonic fundamental frequencies and the harmonic frequencies further show good agreement to previous theory (Karton and Thimmakonda, 2022). The largest discrepancy arises in ν_{10} , with a difference of 26.5 cm⁻¹, but a majority of the frequencies agree to within 3 cm⁻¹. However, given the greater computational rigor of F12-TZ, its data should be considered the more reliable of the two.

4 Conclusion

The large intensities for some of the fundamental frequencies in all of the isomers may facilitate the detection of these C₅H₂ structures *via* JWST. Such spectral data could lead to the identification for some UIRs as well as unknown transitions of known interstellar molecules such as *l*-C₅H₂ and *c*-C₃HC₂H. The bipyramidal C₅H₂ isomer has some of the lowest intensities of the three isomers but still has one peak at 426.1 cm⁻¹ (23.5 μ m) with an intensity of 78 km mol⁻¹. The *c*-C₃HC₂H isomer has one intense fundamental frequency at 3,347.8 cm⁻¹ with an intensity of 84 km mol⁻¹. The *l*-C₅H₂ isomer has the two highest intensities out of the three isomers for its ν_4 at 1887.5 cm⁻¹ with a large intensity of 182 km mol⁻¹ and ν_3 at 2076.3 cm⁻¹ with a very large intensity of 738 km mol⁻¹ making this fundamental the most likely of the group especially since, again, *l*-C₅H₂ has been previously detected *via* radioastronomy. The F12-TZ methodology produces rotational constants for *l*-C₅H₂ within 3 MHz (0.13%) for the *B*₀ and *C*₀ rotational constants. Improving upon this high level of accuracy, the F12-TZ-cCR methodology for *c*-C₃HC₂H produces *B*₀ and *C*₀ rotational constants within 1.1 MHz (0.03%) of experiment. This close agreement to experiment lends supports to the accuracy of the other theoretical values, notably the fundamental vibrational frequencies, reported herein. Out of the three C₅H₂ isomers presented in this work, the *c*-C₃HC₂H isomer has the lowest relative energy, with *l*-C₅H₂ 13.79 kcal mol⁻¹ higher in line with previous work (Seburg et al., 1997). The bipyramidal C₅H₂ isomer is 44.98 kcal mol⁻¹ higher in energy compared previous computations which places this *D*_{3h} isomer 100 kcal mol⁻¹ or more higher than other C₅H₂ isomers (Seburg et al., 1997). The dipole moments for the *c*-C₃HC₂H and *l*-C₅H₂ isomers are large at 3.59 D and 5.94 D, respectively, which has facilitated the previous detections of the isomers in space through rotational spectroscopy. Overall, the anharmonic vibrational frequencies and rotational constants presented herein may help

to facilitate the laboratory or astronomical detection of unknown transitions of known molecules. In particular, the high-intensity fundamental frequencies of each molecule that fall within the high-resolution range of JWST may also help to disentangle some of the heretofore unidentified features in the UIRs.

Data availability statement

The original contributions presented in the study are included in the article/[Supplementary Material](#), further inquiries can be directed to the corresponding author.

Author contributions

AW, BW, and RF contributed to conception and design of the study. AW performed the computations. AW and BW analyzed the data. AW wrote the first draft of the manuscript, and BW and RF edited the manuscript. RF secured funding for the research and provided administration for the project. All authors contributed to manuscript revision, read, and approved the submitted version.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary Material

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fspas.2022.1051535/full#supplementary-material>

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