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The Version of Record is available online at:

<https://doi.org/10.1007/s00128-022-03545-z>

Zhou, B., Xing, M., Liao, H. et al. Assessing Heavy Metal Pollution of the Largest Nature Reserve in Tianjin City, China. *Bull Environ Contam Toxicol* (2022).

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1 **Assessing heavy metal pollution of the largest nature reserve in Tianjin City,**  
2 **China**

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13 **Abstract** Beidagang Wetland (BW) Nature Reserve is centrally situated in Tianjin City,  
14 experiencing an extreme industrial development. This study uses index characteristic  
15 analysis systems for assessing the individual and combined heavy metal pollution  
16 loading in the water during the spring and autumn seasons. By combining the pollution  
17 level of single pollutant, a more comprehensive evaluation of water quality in BW was  
18 achieved. Water quality was worst during autumn due to high level of Cd and Pb, which  
19 indicate the type of anthropogenic activities have a serious effect on heavy metal  
20 pollution in BW. In addition, high exchangeable amounts of Cd (>40%) were found in  
21 the sediments of BW, indicating Cd pollution has emerged. There is a need for  
22 appropriate abatement actions curbing heavy metal loading and improving water  
23 quality of the BW Nature Reserve, thereby ensuring a sustainable management of its  
24 ecosystem services.

25 **Keywords:** Assessment; Water quality; Sediment; Wetland

26 Along with economic development and population growth, the problem of water  
27 pollution has increased (Tang et al. 2022). Most of the pollutants, such as nutrients and  
28 heavy metals, discharged into water bodies eventually end up and accumulate in the  
29 sediments (Hwang et al. 2019; Fang et al. 2019), making especially wetlands a sink of  
30 these pollutants. Subsequently, wetlands may also act as a source of these pollutants,  
31 by releasing their pool of pollutants back into the overlying water (Huo et al. 2013),

32 sustaining the level of these contaminants in the water body. The level of contaminants  
33 in the sediments is thus a good measure for the current state of the wetland environment.

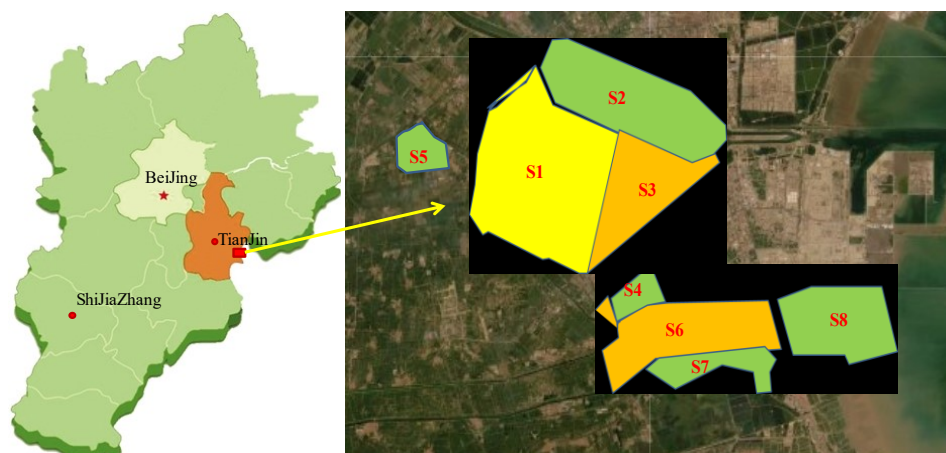
34 Development of the Bohai sea economic zone is a major national strategy in order  
35 to promote the economy of the hinterland. The aim for the coordinated plan of the  
36 Beijing-Tianjin-Hebei region is to achieve perfect urban layout and form. However,  
37 along with the rapid economic development in this region, there has been a rapid  
38 increase in local industrial activities producing lead-containing batteries, non-metallic  
39 minerals, metal surface treatment, chemical raw materials and other chemical products.  
40 Their emissions of wastewater to the local environment poses a great threat to the BW  
41 ecosystem (Li et al. 2021).

42 Previous studies have focused on nutrients in Beidagang Wetland (BW), but lack  
43 of attention to heavy metals (Perez and Anderson 2009; Garnier et al. 2015). This lack  
44 of a knowledge poses a main restriction for achieving sustainable management of the  
45 valuable BW (Li et al. 2021). In this study, water quality is assessed based on the levels  
46 of seven heavy metals. These parameters were selected for assessment based on their  
47 current level of contamination of the BW water environment, as well as their ecological  
48 effect. From the sediments, heavy metals were extracted using the three-step  
49 Community Bureau of Reference (BCR) sequential extraction procedure. The  
50 combined levels of these contaminants were assessed using comprehensive pollution  
51 index (Larner et al. 2006; Rauret et al. 1999; Pueyo et al. 2008). Knowledge gained  
52 from this study is a prerequisite for selecting optimal abatement and conservation  
53 strategies ensuring a sustainable water management and water security.

## 54 **Materials and Methods**

55 The BW Nature Reserve (Fig.1), covering a total area of 44,240 hectares, was  
56 established in 2001 (Li et al. 2021). The wetland area is an important ecological barrier  
57 between the industrial area and the new Binhai residential area in the north. BW is  
58 divided into eight areas: I.e., S1 - the main Beidagang (BDG<sub>w</sub>) Reservoir area, situated  
59 in the west; S2 - along the lower reaches of DuliuJian River, including WanMu (WM)  
60 fish farming pond; S3 - an additional part of the Beidagang (BDG<sub>e</sub>) reservoir area on  
61 the east side; S4 - east ShaZiJing (SJZ) Reservoir; S5 - QianQuan (QQ) reservoir  
62 furthest to the west; S6 - LiErWan (LEW); S7 - lower part of LiErWan reservoir (LEW<sub>s</sub>);  
63 and S8 - MaPengKou (MPK) along the coast. The BDG reservoir is the main part of

64 the BW Nature Reserve, centrally located in terms of neighboring the Dagang oilfield  
65 to the southeast. It plays a central role in the Dagang entire ecosystems by mediating  
66 the filtration of the industrial pollution and by generating a local microclimate.



67

68 **Fig.1** Map of the study area and sampling sites (S1 - BDG<sub>w</sub>; S2 - WM; S3 - BDG<sub>e</sub>; S4 -  
69 SJZ; S5 - QQ; S6 - LEW; S7 -LEW<sub>s</sub>; and S8 - MPK).

70 Water samples were collected in the spring and autumn of 2020 from between 4  
71 and 14 sites in each area (Table S1 – Table S8). Generally, water samples were collected  
72 at 20 cm depth in the reservoirs. Where water depths exceeded 2 m, samples from 20  
73 cm were merged with water collected at 1.5 m below water surface. Samples were  
74 collected in triplicate doubles in order to ensure accuracy and precision of the data. Half  
75 of the 500 mL water sample was filtrated through 0.45µm membrane filters (Covelli  
76 and Fontolan 1997).

77 Sediment samples were collected in 2020 from the same sites as water samples  
78 (Table S1 – Table S8), using a bottom sediment grab sampler. Sediment samples were  
79 placed in self-sealing bags and immediately transported to the laboratory for cooling.  
80 In the laboratory the samples were freeze-dried using FD-1A-50 freeze drier and passed  
81 through 100-mesh screens, prior to storage until analysis (Song et al. 2020). All plastic  
82 ware was cleaned by soaking in 10% (v/v) HNO<sub>3</sub> for more than 24 h and then rinsed  
83 with Milli-Q water (>18 MΩ cm). Heavy metal fraction is crucial to the level of heavy  
84 metal pollution (Rauret et al. 1999), so BCR sequential extraction scheme was used as  
85 described in S2 to extract fractions of heavy metals in the sediments. All supernatants  
86 were filtered through a 0.45µm cellulose filter prior to storage for analysis (Larner et  
87 al. 2006; Rauret et al. 1999).

88 The Ca, Mg, Na and K analyses were carried out using inductively coupled plasma

89 (ICP)-optical emission spectrometer (OPTIMA 2000 DV, Perkin Elmer Inc., Waltham,  
90 MA). The samples were analyzed for Cd, Cr, Cu, Ni, Pb and Zn using an Inductively  
91 Coupled Plasma Mass Spectrometer (ICP-MS) according to standard methods. pH in  
92 sediment samples was conducted as described in standard method. All chemicals were  
93 analytical-reagent-grade or equivalent analytical purity. Relative standard deviation  
94 (RSD) of parallel samples were lower than 10% implying good data precision. The  
95 accuracy of heavy metal data was assured using standard reference material (GBW-  
96 07309).

97 Water quality was assessed using both the single factor pollution index ( $P_i$ ) and the  
98 comprehensive pollution index method (P). Following the Chinese environmental  
99 quality standard for surface waters (GB3838, 2002)  $P_i$  relate the measured  
100 concentration of contaminants to the Chinese governments limit values, while the P is  
101 the average  $P_i$  values for each area. P values exceeding 1 indicate high level of pollution,  
102 values greater than 2 imply serious pollution (Desaules et al. 2012; Cresswell et al.  
103 2012). These methods are described in detail in the supplementary data (S3-S4). The  
104 potential ecological risk (PER), posed by the levels of pollutants in the sediments, was  
105 determined as described in S5 (Håkanson 1980; Abraham and Parker 2008; Cappuyns  
106 2008). The PER method provides index values ( $E^i_r$ ) for each heavy metal (i) based on  
107 its concentration relative to background values and a toxicity factor (Guo et al. 2010).  
108 The comprehensive PER index (RI) is the sum of the PER indexes. Statistical  
109 assessment of the data was performed using SPSS 16.0. Visualization of sample sites  
110 distribution was achieved using Google Earth. Origin Pro 8.5 software was used to plot  
111 the data.

## 112 **Results and Discussion**

113 Physical and chemical characteristics of sediments are presented in Table 1. The mean  
114 value of pH was 7.4, indicating the sediments were neutral or slightly alkaline.  
115 Moreover, the mean values of K, Na, Ca and Mg were 23.3, 14.1, 36.9 and 16.3 g/kg,  
116 respectively. Mean  $P_i$  values of the studied heavy metals in spring and autumn water  
117 from the eight areas (S1-8 in Supporting data) of the BW are shown in Table 2. Mean  
118 level of Cd exceeded the Chinese standard limit at all sites during both seasons, while  
119 Pb was above the limit in most of the study areas. It was reported that the pollution of  
120 heavy metals in river sediments from Haihe River Systems was ubiquitous because of  
121 rapid urbanization and agricultural intensification in this region (Tang et al. 2013). The

122 results of this study has indicated heavy metal pollution of the BW, implying that urgent  
 123 abatement actions are required to curb the pollution from the neighboring industries. In  
 124 addition, the mean  $P_i$  of Cr, Cu, Zn, and Zn meet their standard, implying no pollution  
 125 by these contaminants in the BW.

126 **Table 1** Physical and chemical properties of sediments from the BW (SD,  
 127 standard deviation).

	Item	Range	Mean	SD
<b>Sediment</b>	pH	7.0~8.4	7.4	0.32
	K (g/kg)	18.9~34.1	23.3	2.6
	Na (g/kg)	13.3~15.1	14.1	0.40
	Ca (g/kg)	29.3~48.5	36.9	4.1
	Mg (g/kg)	10.7~28.2	16.3	3.4

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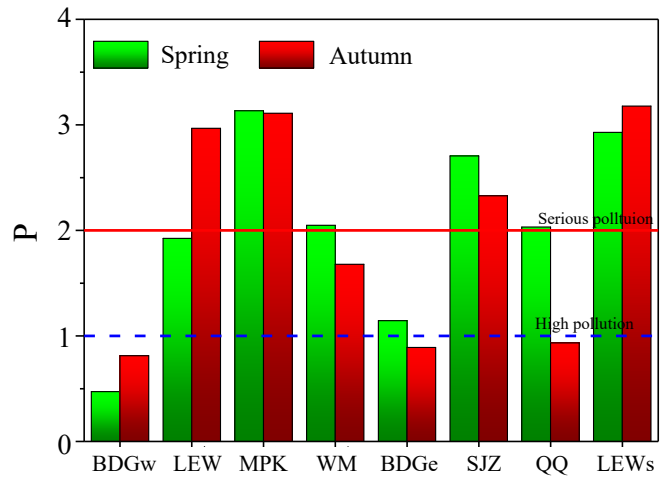
129 **Table 2** Single factor pollution index ( $P_i$ ) of overlying water from the BW (Yellow:  
 130 exceeds the standard; Orange: serious pollution degree).

<b>Spring</b>	Cd	Cr	Cu	Ni	Pb	Zn
BDGw	1.9	<1	<1	<1	<1	<1
LEW	10	<1	<1	<1	4.3	<1
MPK	18	<1	<1	<1	9.0	<1
WM	10	<1	<1	<1	4.3	<1
BDGe	6.3	<1	<1	<1	1.1	<1
SJZ	14	<1	<1	<1	5.7	<1
QQ	9.6	<1	<1	<1	2.4	<1
LEWs	16	<1	<1	<1	6.9	<1
<b>Autumn</b>	Cd	Cr	Cu	Ni	Pb	Zn
BDGw	5.0	<1	<1	<1	<1	<1
LEW	15	<1	<1	<1	8.9	<1
MPK	15	<1	<1	<1	11	<1
WM	8.0	<1	<1	<1	4.6	<1
BDGe	5.0	<1	<1	<1	<1	<1
SJZ	11	<1	<1	<1	6.9	<1

QQ	5.0	<1	<1	<1	<1	<1
LEW <sub>s</sub>	17	<1	<1	<1	10	<1

131

132 A main aim when assessing the state of an environment is to identify risk areas that  
 133 require specific attention. The comprehensive pollution indexes (P) for each of the eight  
 134 areas of BW are shown in Fig. 2. It is clear from the figure that all sites, with the  
 135 exception of BDG<sub>w</sub>, have high pollution level in the spring. The conditions are slightly  
 136 better in the autumn, were BDG<sub>w</sub>, BDG<sub>e</sub> and QQ fall slightly below the high pollution  
 137 classification. Areas with P values greater than 2, having serious pollution, are at  
 138 especially high ecological risk. These were MPK, WM, SJZ, QQ and LEW<sub>s</sub> in the  
 139 spring, and LEW, MPK, SJZ and LEW<sub>s</sub> in the autumn. Overall, the results show that  
 140 LEW, MPK, SJZ and LEW<sub>s</sub> are the areas with higher water pollution by heavy metals,  
 141 though there are concerns regarding pollution levels in different seasons at LEW and  
 142 QQ.

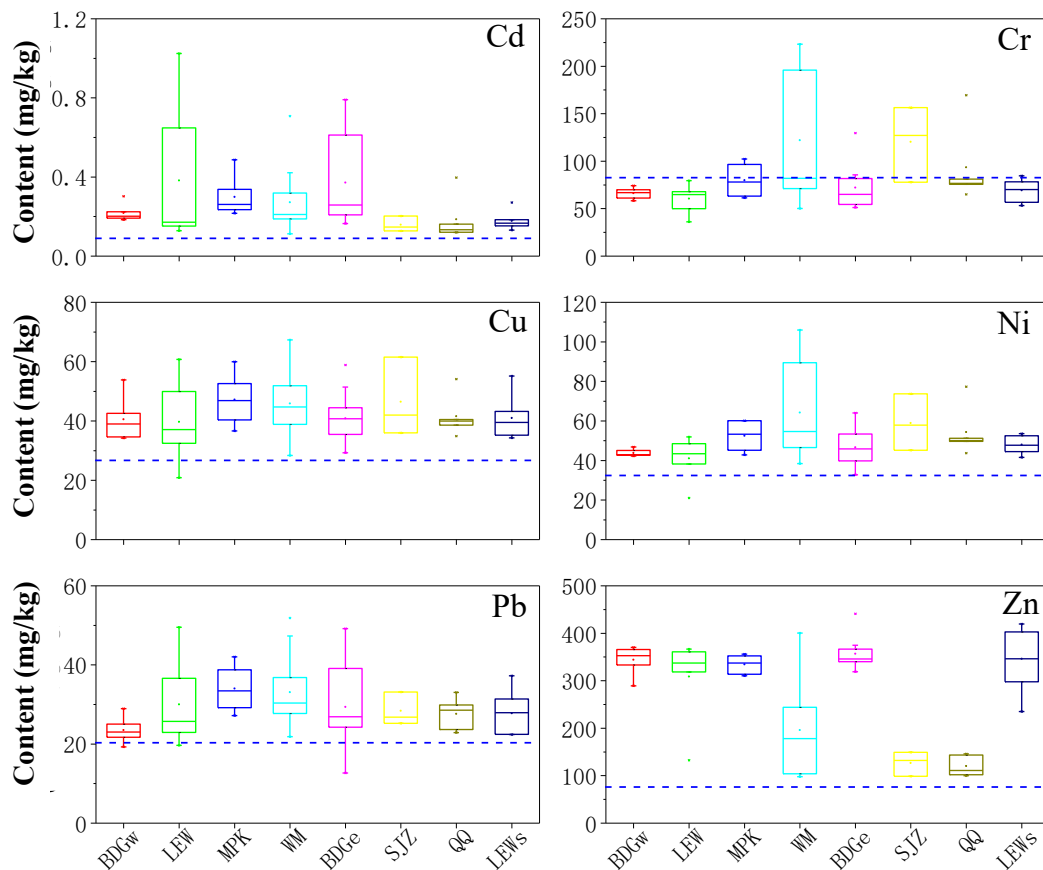


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144 **Fig.2** The comprehensive pollution indexes (P) of water from the BW.

145 In this study, all sediment samples had levels of Cd, Ni and Zn above the Chinese  
 146 soil environmental quality standard (Fig. 3). For Cu and Pb, the levels were below the  
 147 limit value in only two samples, respectively. However, in regards to Cr only 20% of  
 148 the samples were above the limit value. Therefore, Cd, Ni, Pb, Cu and Zn could be  
 149 regarded as pollutants in BW sediments, and the sediment samples were mainly  
 150 contaminated by high Cd and Zn contents. Some previous studies indicated that the  
 151 residual fractions of heavy metals have lithogenic sources but that non-residual  
 152 fractions are mainly related to anthropogenic inputs (Gao and Chen, 2012; Islam et al.,

153 2015). The relative amounts of exchangeable ( $B_1$ ), ferro-manganese bond or reducible  
 154 ( $B_2$ ), organic matter-bound or oxidisable ( $B_3$ ), and residual ( $B_4$ ) heavy metals pools are  
 155 shown in Fig. 4. Regarding Cd the exchangeable fraction ( $B_1$ ) was the highest  
 156 comprising more than 40% of the total Cd content. The mean percentage of organic  
 157 matter-bound or oxidisable pools ( $B_3$ ) of Cr, Cu, Ni and Zn accounted for more than  
 158 26%. Regarding the ferro-manganese bond or reducible pools ( $B_2$ ), the Pb comprised  
 159 more than 65%. These results show that the heavy metals differ widely in the way they  
 160 are bound to the sediments (Nolan et al. 2005; Filgueiras et al. 2002): The Cd is mainly  
 161 found in the exchangeable pool ( $B_1$ ); Pb is mainly found to be adsorbed to iron ( $B_2$ );  
 162 Zn, Ni and Cu are mainly bound to organic matter ( $B_3$ ); while Cr has the highest residual  
 163 fraction ( $B_4$ ). These differences are partly caused by the form in which these heavy  
 164 metals are emitted to the BW environment, the levels of the contaminants in the  
 165 environment, and partly governed by their chemical properties, such as the ionic- and  
 166 covalent index. As shown in Table 2, the high contamination by Cd in the water were  
 167 likely mainly due to the high exchangeable amounts of Cd in the sediments.

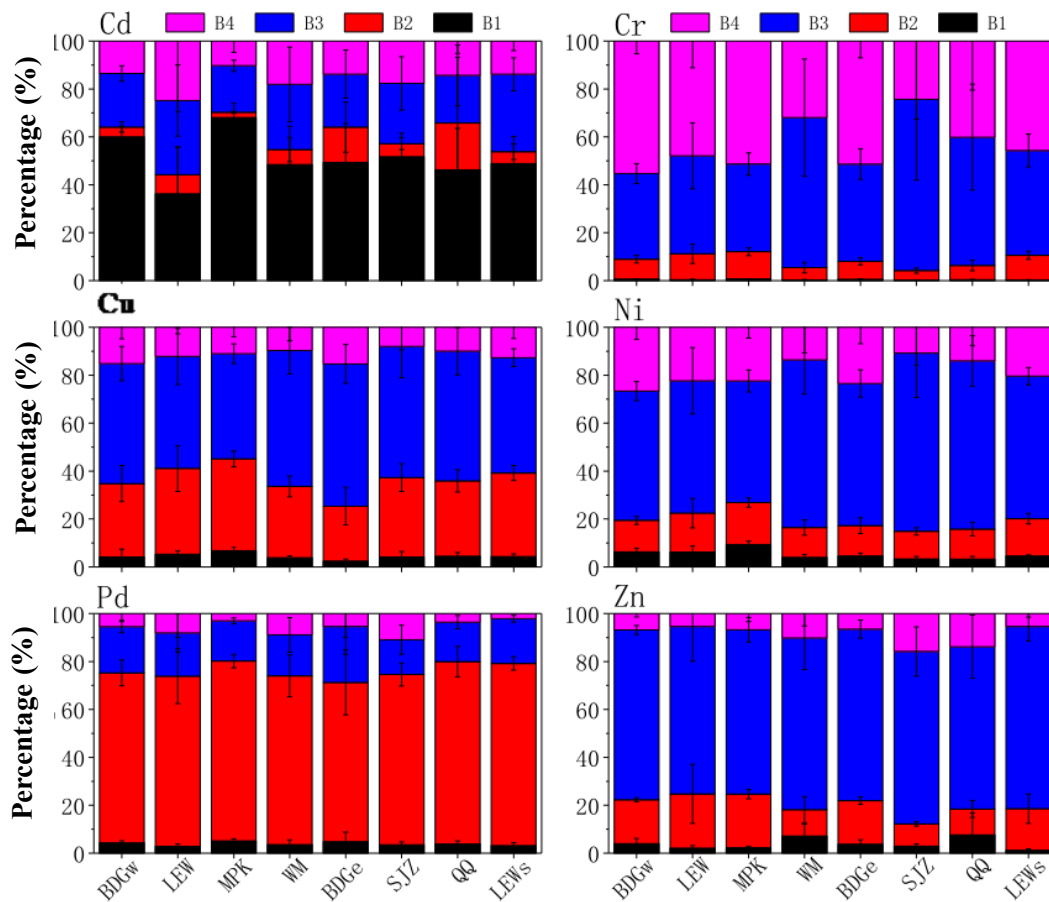


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169

**Fig.3** The distribution of heavy metal contents in the sediments from the BW.





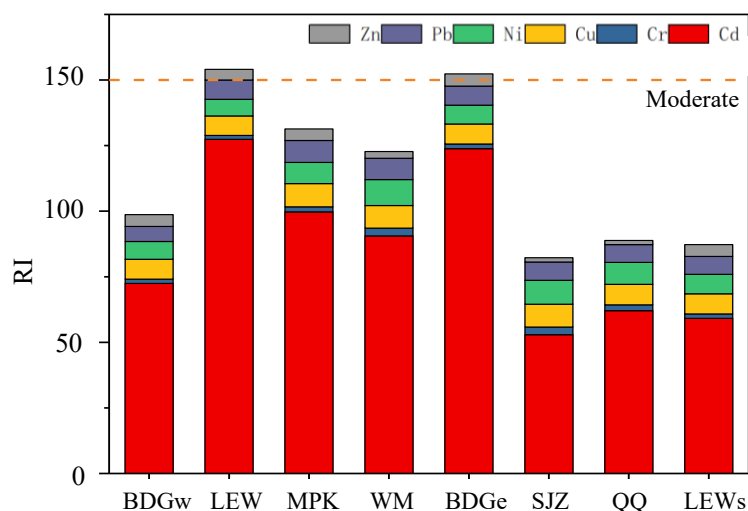
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171 **Fig.4** The distribution of fraction percentage of heavy metals in the sediments from  
 172 the BW.

173 The toxicities and bioavailabilities of heavy metals are dependent on the chemical  
 174 speciation and concentrations of the metals (Islam et al., 2015). Especially, the pool of  
 175 exchangeable (B<sub>1</sub>) heavy metals is the fraction of main contemporary ecological  
 176 concern, as these ions are in rapid equilibrium with the water and thus readily  
 177 bioavailable. The other fractions are rather stable forms that are not readily available,  
 178 though may become available over time. The large exchange pool of Cd at LEW and  
 179 BDG<sub>e</sub> are thus of prime ecological concern, while the contamination by Ni, Cr, Pb, Cu,  
 180 and Zn are expected to have less ecological impact (Jonge MD et al. 2012a, b). The  
 181 mean PER value for Cd ( $E^{Cd}_r$ ) at LEW, MPK, WM and BDG<sub>e</sub> were higher than 80,  
 182 implying a considerable degree pollution. Ignoring Cd, the RI adds up to less than 40,  
 183 implying a low ecological risk in the BW regarding the other heavy metals. Adding Cd  
 184 ( $E^{Cd}_r$ ), which had mean values above 50, raised the RI to moderately polluted in LEW  
 185 and BDG<sub>e</sub> (Fig. 5). However, except for at LEW and BDG<sub>e</sub>, there is according to the  
 186 RI low ecological risk in other areas of BW.

187 Beijing-Tianjin-Hebei is the largest urbanized region in northern China, comprising  
188 the economic region surrounding the cities of Beijing, Tianjin, and Hebei, along the  
189 coast of the Bohai Sea (Wang et al. 2014). The BW is the largest wetland nature reserve  
190 in Tianjin City, which is served as an important habitat for the migratory shorebirds in  
191 the East Asian-Australasian flyway. It has been listed as an international important  
192 wetland because of abundant waterbirds resources. The quality of water and sediments  
193 in BW is very important to the migratory shorebirds, especially heavy metal pollution  
194 (Chen et al. 2018). Soils in China have been contaminated with heavy metals to varying  
195 degrees, and Cd and Hg have been identified as priorities for control due to their higher  
196 concentrations in soils and higher public health risks (Zhang et al. 2015). At the same  
197 time, these heavy metals also entered in the water environment and posed risk  
198 (Pavageau et al. 2002). In the BW, LEW, MPK, SJZ and LEWS were found with high  
199 water pollution by Cd and Pb, meanwhile, moderate ecological risk associated with  
200 heavy metals was observed in at LEW and BDGe.

201 In 2021, there are more than 70 species of migratory birds in BW, and the total  
202 number is close to 450,000. A bird's choice of habitat during migration must be  
203 ecologically safe, and there is plenty of food. If these conditions are not available, they  
204 will not settle down and would rather continue to migrate (Xie et al. 2011). The habitat  
205 quality of migratory birds in BW has been of particular concern. In addition to  
206 polycyclic aromatic hydrocarbons (Wang et al. 2020), heavy metals in the water and  
207 sediments of BW also present ecological risks in this study. Toxic metals could pose a  
208 serious threat to aquatic ecosystems because of their toxicity, recalcitrant nature and  
209 persistence, accumulation and potential to enter the food chain (Nobi et al. 2010, Zhang  
210 et al. 2014). Eventually, it will threaten the safety of migratory shorebirds. Therefore,  
211 appropriate abatement actions curbing heavy metal loading should be taken to improve  
212 water and sediment quality of the BW nature reserve.



213  
214 **Fig.5** The RI values of heavy metals in sediments from the BW.

215 This study assessed the levels of heavy metal concentrations in the water, during  
216 spring and autumn, and in the sediments of the BW Nature Reserve situated in the heavy  
217 industrialized Tianjin City. Levels of Cd in the surface waters, during both spring and  
218 autumn, exceeded the Chinese limit values. The levels of Pb in the water posed also a  
219 cause for concern. The sediment samples were moderately polluted in regards to their  
220 total Cd levels. Moreover, the Cd is mainly found in the exchangeable fraction (B1),  
221 which is a fraction that poses contemporary ecological risk. There is therefore an urgent  
222 need to determine the source of the Cd pollution and enforce appropriate abatement  
223 actions to curb this pollution. Although the sediments generally had a low degree  
224 pollution, the water quality was poor. This will eventually cause the environmental  
225 conditions of the sediments to also deteriorate. These findings imply that more attention  
226 is required regarding the ecological impact of heavy metal pollution in these wetland  
227 systems.

228 **Acknowledgments** This research was supported by Tianjin Science and technology  
229 project (Grant No.18ZXSZSF00130).

230 **Appendix A. Supplementary data**

231 Supplementary data to this article can be found online at .

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## Supplementary data for

### Assessing heavy metal pollution of the largest nature reserve in Tianjin City, China

#### S1.1 Main Beidagang Reservoir (BDGw)

**Table S1** Coordinate of sampling sites in BDGw.

No.	Name	Longitude	Latitude
0	BDGw1	117.390557020	38.767444830
1	BDGw2	117.382883830	38.765102480
2	BDGw3	117.344893930	38.780070110
3	BDGw4	117.329913330	38.787721650
4	BDGw5	117.310310320	38.800997130
5	BDGw6	117.281882160	38.786062950
6	BDGw7	117.264735296	38.757758398
7	BDGw8	117.278374730	38.743390110
8	BDGw9	117.299231800	38.734495600
9	BDGw10	117.254118030	38.710676810
10	BDGw11	117.306728430	38.692476460
11	BDGw12	117.346104640	38.683563120
12	BDGw13	117.352355000	38.679533740

#### S1.2 Lower Reaches of DuliuJian River (WM)

**Table S2** Coordinate of sampling sites in WM

No.	Name	Longitude	Latitude
0	WM1	117.452535340	38.790061360
1	WM2	117.445092890	38.792587470
2	WM3	117.431506020	38.796778380
3	WM4	117.400416057	38.800134557
4	WM5	117.373080610	38.802829202

5	WM6	117.350041310	38.806177613
6	WM7	117.375644042	38.790576465
7	WM8	117.390058698	38.785621511
8	WM9	117.428156090	38.780492110
9	WM10	117.466910890	38.785648330
10	WM11	117.473585050	38.781419510
11	WM12	117.456580060	38.790605110
12	WM13	117.445869240	38.775804090
13	WM14	117.443676210	38.772255330
14	WM15	117.445461520	38.768847430

### S1.3 Additional Beidagang Reservoir (BDGe)

**Table S3** Coordinate of sampling sites in BDGe

No.	Name	Longitude	Latitude
0	BDGe1	117.4009455	38.76279668
1	BDGe2	117.3841357	38.69526438
2	BDGe3	117.3847358	38.69516349
3	BDGe4	117.4198303	38.71431155
4	BDGe5	117.4396821	38.73207277
5	BDGe6	117.4768625	38.74440317
6	BDGe7	117.4767763	38.74507204

### S1.4 ShaJingZi Reservoir (SJZ)

**Table S4** Coordinate of sampling sites in SJZ

No.	Name	Longitude	Latitude
0	SJZ1	117.42516155	38.67802394
1	SJZ2	117.40292275	38.66675919
2	SJZ3	117.38834166	38.65683151
3	SJZ4	117.39160152	38.64672878



4	SJZ5	117.43211479	38.66391744
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### S1.5 QianQuan Reservoir (QQ)

**Table S5** Coordinate of sampling sites in QQ

No.	Name	Longitude	Latitude
0	QQ1	117.211975970	38.770522166
1	QQ2	117.224171446	38.763291812
2	QQ3	117.221707034	38.747179411
3	QQ4	117.201356285	38.747379612
4	QQ5	117.197446014	38.770691560

### S1.6 LiErWan (LEW)

**Table S6** Coordinate of sampling sites in LEW

No.	Name	Longitude	Latitude
0	LEW1	117.391132300	38.627460790
1	LEW2	117.407303290	38.639422200
2	LEW3	117.415462670	38.615596400
3	LEW4	117.428670710	38.620635870
4	LEW5	117.443523270	38.622719200
5	LEW6	117.458322820	38.622866920
6	LEW7	117.484371570	38.622667590
7	LEW8	117.494278500	38.623031940
8	LEW9	117.510216890	38.623682830
9	LEW10	117.529167320	38.624406690

### S1.7 LiErWan Reservoir (LEWs)

**Table S7** Coordinate of sampling sites in LEWs

No.	Name	Longitude	Latitude
0	LEWs1	117.4421054670	38.6180405530

1	LEWs2	117.4259222360	38.6111418290
2	LEWs3	117.4424805250	38.6073800470
3	LEWs4	117.4678236620	38.6113247870
4	LEWs5	117.4883246900	38.6184931530
5	LEWs6	117.5156203840	38.6193927460
6	LEWs7	117.5304424430	38.6125619930

**S1.8 MaPengKou (MPK)**

**Table S8** Coordinate of sampling sites in MPK

<b>No.</b>	<b>Name</b>	<b>Longitude</b>	<b>Latitude</b>
0	MPK1	117.552967101	38.632146149
1	MPK2	117.572620190	38.634076690
2	MPK3	117.581223709	38.641160645
3	MPK4	117.578982480	38.650621213
4	MPK5	117.562907296	38.653141986
5	MPK6	117.546242178	38.651206902

## S2. Extraction procedure

**First step** (Exchangeable and weak acid soluble fraction, B<sub>1</sub>): 1 g soil sample was extracted with 40 mL of 0.11 mol L<sup>-1</sup> acetic acid (Merck Suprapur) solution by shaking in a mechanical, end-over-end shaker at 30 ± 10 rpm at 22 ± 5 °C for 16 h. The extract was separated by centrifugation at 3000 ×g for 20 min, collected in polyethylene bottles and stored at 4 °C until analysis. The residue was washed by shaking for 15 min with 20 mL of doubly deionized water and then centrifuged, discarding the supernatant.

**Second step** (Ferro-manganese bond or reducible fraction, B<sub>2</sub>): 40 mL of 0.5 mol L<sup>-1</sup> hydroxylammonium chloride (Merck pro-analysis) solution was added to the residue from the first step, and the mixture was shaken 30 ± 10 rpm at 22 ± 5 °C for 16 h. The acidification of this reagent is by the addition of a 2.5% (v/v) 2 mol L<sup>-1</sup> HNO<sub>3</sub> solution (prepared by weighing from a suitable concentrated solution). The extract was separated and the residue was washed as in the first step.

**Third step** (Organic bound or oxidisable fraction, B<sub>3</sub>): 10 mL of 8.8 mol L<sup>-1</sup> hydrogen peroxide (Merck Suprapur) solution was carefully added to the residue from the second step. The mixture was digested for 1 h at 22 ± 5 °C and for 1 h at 85 ± 2 °C, and the volume was reduced to less than 3 mL. A second aliquot of 10 mL of H<sub>2</sub>O<sub>2</sub> was added, the mixture was digested for 1 h at 85 ± 2 °C, and the volume was reduced to about 1 mL. The residue was extracted with 50 mL of 1 mol L<sup>-1</sup> ammonium acetate (Merck pro-analysis) solution, adjusted to pH 2.0, at 30 ± 10 rpm and 22 ± 5 °C for 16 h. The extract was separated and the residue was washed as in previous steps.

**Residue from the third step** (Residual fraction, B<sub>4</sub>): the residue from step 3 was digested with aqua regia, following the ISO 11466 (ISO, 1995b). In this case, the amount of acid used to attack 1 g of sample was reduced to keep the same volume/mass ratio: 6.0 mL of HCl (37%) and 2.0 mL of HNO<sub>3</sub> (70%) were added, and then add 2 mL hydrofluoric acid by microwave digestion.

The quality of the analytical data for the sequential extraction procedure was assessed by carrying out analyses of the certified reference materials GBW07438. Three independent replicates were performed for each sample and blanks were measured in parallel for each set of analyses using BCR procedure. The moisture content of each sample was determined by drying a separate 1 g sample in an oven (105±2 °C) to constant mass.

### **S3. Single factor pollution index**

The single factor pollution index method is the ratio of the measured value of a single water quality index to the evaluation standard value, which is used to evaluate whether the water quality index meets the requirements of the corresponding standard. We here use the Chinese governments Environmental quality standards for surface waters (GB 3838, 2002). The expression is as follows:

$$P_i = C_i/C_0$$

Where  $P_i$  is the single factor pollution index;  $C_i$  is the actual concentration measurement value of item  $i$ ;  $C_0$  is the evaluation standard limit of the index  $i$ . When  $P_i \leq 1$ , it indicates that this index meets the standard; When  $P_i > 1$ , it indicates that the index exceeds the standard. The larger  $P_i$  and more serious pollution degree of the index are.

### **S4. Comprehensive pollution index**

The comprehensive pollution index is a statistical analysis on the basis of single factor pollution index, and a comprehensive evaluation of water pollution degree is made according to arithmetic average, weighted average or other mathematical results of selected water quality index  $P_i$ . The mean comprehensive pollution index formula is as follows:

$$P = \frac{1}{n} \sum_{i=1}^n P_i$$

Where  $P$  is the mean comprehensive pollution index, and  $P_i$  is the single factor pollution index.  $P$  is the arithmetic mean of the single factor pollution index of  $n$  indices. The pollution degree of the mean comprehensive pollution index is divided into:  $P \leq 0.20$ , good water quality;  $0.21 \sim 0.40$ , the water quality is good;  $0.41 \sim 0.70$ , light pollution;  $0.71 \sim 1.00$ , moderate pollution;  $1.01 \sim 2.00$ , high pollution;  $P \geq 2.01$ , serious pollution (Desaules et al. 2012; Cresswell et al. 2012).

### **S5. Potential ecological risk**

The potential ecological risk (PER) method was developed by Hakanson (Håkanson 1980). The PER index was introduced to assess the degree of contamination of trace

metals in the soils. The equations for calculating the PER indexes are as follows:

$$E_r^i = T_r^i \times C_f^i = T_r^i \times (C_s^i / C_n^i),$$

$$RI = \sum_{i=1}^n E_r^i,$$

where  $C_s^i$  is the content of the element in samples,  $C_n^i$  is the background value of the element,  $C_f^i$  is the single element pollution factor,  $E_r^i$  is the PER index of an individual element, and  $T_r^i$  is the biological toxicity factor of an individual element, which are defined as Cd=30, Cr=2, Cu=Ni=Pb=5, and Zn=1 (Guo et al. 2010).  $RI$  is the comprehensive PER index, which is the sum of  $E_r^i$ .

**Table S9** shows the factor standard of different levels.

<b>Potential ecological risk (PER) index</b>			
$E_r^i$	PER of individual elements	$RI$	Comprehensive PER
<40	Low	<150	Low
40–80	Moderate	150–300	Moderate
80–160	Considerable	300–600	High
160–320	High	$\geq 600$	Serious
$\geq 320$	Very high		

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