

Contents and benthic fluxes of nutrients in sediment pore water of the Southern Ocean

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Abstract The concentration and profile characters of nutrients in sediment pore water of the South Ocean, sampled during the cruise of CH NARE-18 of the austral summer 2001/02 were determined. The results show that the content of $\text{SiO}_3\text{-Si}$ and $\text{NH}_4\text{-N}$ were much higher than other nutrients. The profile of $\text{SiO}_3\text{-Si}$ was characterized with sharp gradients near the sediment-water interface, profile distribution of silicate show that during the diagenetic reaction of silicate, dissolution was the main part. The organic matter decomposition was occurred under the anaerobic condition. Benthic fluxes of $\text{SiO}_3\text{-Si}$, $\text{NH}_4\text{-N}$ were from pore water to overlying water, and regeneration of silicate and ammonium were the main part of the nutrient regeneration from the pore water of the study area.

Key words nutrient, interstitial water, sediment, the South Ocean

1 Introduction

Sediments and their pore water play an important role in the cycling of many chemical substances, such as nutrients and organic contaminants. Both chemical and microbiological transformation processes are responsible for cycling elements between water and sediments. Below the sediment and water interface, mineralization of organic matter is the main part of early diagenetic processes^[1], during which nutrients that preserved in sediments could release to pore water, and exchanged with overlying water by gradient of concentration, and by physical and biological stirring. Therefore, study on pore water compositions of nutrients in the sediment is very important to understanding nutrients biochemical cycling^[2].

The data set presented in this study is mainly focused on the pore water concentrations of nutrients and their fluxes across the sediment-water interface at the stations in the Prydz Bay, east Antarctica, and adjacent area of the Great Wall Station, west Antarctica, during the cruise of CH NARE-18, the austral summer of 2001/02.

2 Material and methods

Four sediment cores (G-1, N-8, N-10, D-1) were collected by no-perturbation- sed-

iment-collector with four tubes, during the cruise of CHNARE-18 of the austral summer 2001/2002. Fig 1 and table1 shows the investigation area and the sampling stations. The diameter and length of the collector tube is 10 cm and 60 cm, respectively. And the lengths of our cores are from 10cm to 30 cm. The sediment in the upper 10 cm of each core was sliced into 1 cm intervals, and below 10 cm then into 2 cm intervals. Samples for analysis of nutrients concentrations in the bottom water were also collected from the cores by drawing water from 5 cm above the undisturbed sediment-water interface. The separation of the pore water from the sediment slices was operated by centrifugation ((at 4000 g for 30 minutes). All the water samples filtered through 0.45 μm pore size cellulose acetate filters and HgCl_2 were added in all filtered samples and stored for further analysis. Nutrients concentrations in pore water were analyzed in the laboratory using an autoanalyzer (Brown Lubbi), and the methods recommended by the "Standards of Marine Survey" (The STSA, 1991) were used. The standard solutions for analysis of the content of nutrients were provided by the Second Institute of Oceanography, SOA. Sediment-water flux incubations were held by using one tube of sediment in G-1 and N-10, and fluxes of nutrients across the sediment-water interface were calculated from the time-varying concentration of them in the overlying water of the incubated tube. All sediment processing was carried out at in situ temperature ($2\text{ }^\circ\text{C}$) to avoid temperature artifacts.

Table 1 Sampling stations

Station	Longitude	Latitude	Depth(m)	Date
Great wall station G-1	58 $^\circ$ 55. 0 W	62 $^\circ$ 13. 0 S	6	2001. 12. 17
Prydz Bay N-8	72 $^\circ$ 59. 1 E	66 $^\circ$ 51. 2 S	510	2002. 02. 23
Prydz Bay N-10	72 $^\circ$ 56. 5 E	67 $^\circ$ 30. 0 S	596	2002. 02. 23
Davis station D-1	77 $^\circ$ 53. 0 E	68 $^\circ$ 33. 0 S	30	2002. 02. 25

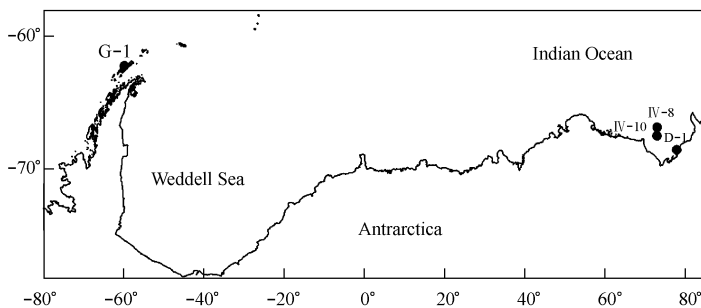


Fig 1 Sampling stations in the Southern Ocean

3 Result and discussion

3.1 Dissolved silicate (DSi) down core profiles

Fig 2 shows the concentrations of silicate in the overlying water and pore water. The pore water concentrations of DSi display a larger range of variation, from 229.00-759.57 $\mu\text{mol}/\text{dm}^3$. The average concentration of N-8, N-10, G-1, D-1 was 566.79, 617.01,

547.88 and 454.98 $\mu\text{mol}/\text{dm}^3$, respectively. We can see that the average at IV-8 and G-1 were similar and that at D-1 was the lowest. Sharp gradients appeared near the sediment-water interface in all the profiles and are more pronounced at the station IV-8 and IV-10 in the Prydz Bay. The DSi concentration increased from 74.35 μM , which represents the mean DSi concentration in the bottom water, to the value of 349.08 $\mu\text{mol}/\text{dm}^3$ at 0 cm, and with depth increased the concentration of DSi in each pore water except in the G-1, exponentially approached to a stable asymptotic concentration of C_d in upper 10 cm of the core, which closed to the typical distribution of sediment-water interface undergoing low physical and biological stirring^[3,4]. In G-1 further increase in DSi below ~ 10 cm are obviously, and under ~ 15 cm the increase is small ranging between 599.47-719.57 $\mu\text{mol}/\text{dm}^3$.

Archer *et al* (1993) assumed that the pore water concentration of DSi regulate biogenic silica dissolution rates as it increases down-core until reaching saturation for the biogenic opal present, and C_d is assumed to be the solid phase solubility^[5]. However other researchers did not think the C_d was the solubility^[6], because the early diagenetic processes include not only the dissolution process but also other chemistry processes. Moreover the adsorption by clay made the pore water concentration of DSi lower than the solubility^[7], so the C_d is a concentration of equilibrium. Fig 3 is the "dissolution-diffusion-reaction" mode of DSi in pore water raised by Song (1991). According to the mode, dissolution is the main part of the early diagenetic processes in IV-8, IV-10 and D-1. However, in G-1, except the dissolution process there also adsorption and precipitation processes in the early diagenetic process. The C_d value of IV-8, IV-10 and D-1 is 624, 651 and 488 $\mu\text{mol}/\text{dm}^3$, respectively, which are less than the solubility of biogenic silica measured in the laboratory (1100 μM at 2 $^{\circ}\text{C}$ ^[8]), but higher than that in Northeast Atlantic, which varied between 180 and 230 $\mu\text{mol}/\text{dm}^3$ ^[3].

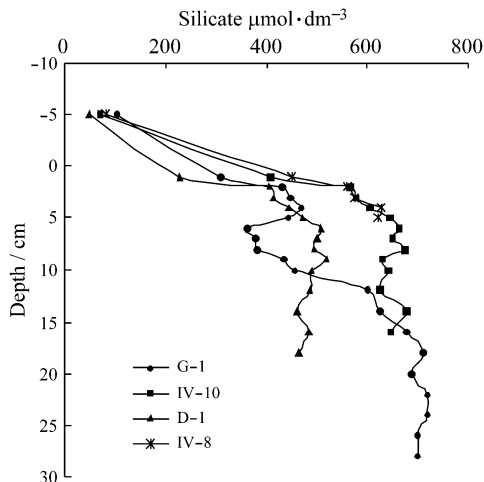


Fig 2 Profile distribution of silicate in sediment pore water ($\mu\text{mol}/\text{dm}^3$).

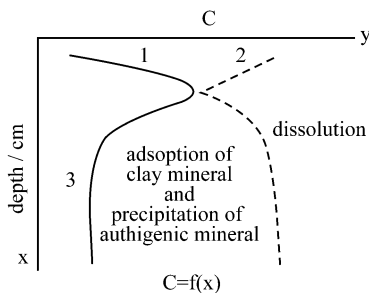


Fig 3 Mode of “diffusion-advection-reaction” of dissolved silicate in sediment interstitial water^[9].

3.2 Dissolved inorganic nitrogen down core profiles

The pore water concentration range of nitrate, nitrite and ammonium for all samples of the present study were 2.86-56.20, 0.14-2.43 and 10.70-623.60 $\mu\text{mol}/\text{dm}^3$, respectively. Concentrations of ammonium were much higher than the former two. And of the three forms of inorganic nitrogen nutrients, ammonium nitrogen took up about 98% of the total in G-1 (average concentration: 305.97 $\mu\text{mol}/\text{dm}^3$) and D-1 (average concentration: 263.72 $\mu\text{mol}/\text{dm}^3$). Whereas the nitrate concentrations in pore water at G-1 and D-1 were only 5.77 and 6.73 $\mu\text{mol}/\text{dm}^3$, respectively. On the contrary, nitrate was the main part of the three forms of inorganic nitrogen nutrients in N-8 and N-10, 35.26 and 29.16 $\mu\text{mol}/\text{dm}^3$, and they contained the lowest average concentration of ammonium, 29.73 $\mu\text{mol}/\text{dm}^3$ at N-10 and 42.90 $\mu\text{mol}/\text{dm}^3$ at N-8.

The forms and concentrations of inorganic nitrogen nutrients in the pore water are mainly controlled by the early diagenetic process. In G-1 and D-1, ammonium was the main form of inorganic nitrogen nutrients, which indicate that degradation occurred under anoxic conditions induced by reducing bacteria. Fig 4 is the profile distribution of Eh in sediments of G-1 and N-10. The value of Eh decreased with depth, which ranged from -45 to 194 mv in G-1 and 268-497 mv in N-10, means that sediment in G-1 was under weak anoxic conditions and increased with depths sediment became much more reductive, and the top sediment in N-10 was under oxidative in some extent than the sediment under 9 cm, based on the correlation between reductive-oxidative characteristic and Eh of sediment^[10,11].

Ammonium concentrations in the pore water increased with depth, especially it was much higher than that of the surface layer below 10 cm (Fig 5-a). However, the nitrate profiles were contrary to the ammonium profiles, which were decreased with depth (Fig 5-b), which we infer that is because of nitrate acting as oxidation agent^[12-16]. In surface sediments, where under oxidation condition, ammonium could be oxidized into nitrate, but with depth increasing, sediment became much more reductive and under anoxic condition, nitrate acted as oxidation agent. Thus, the nitrate concentration in the top surface sediment was relatively high, but rapidly decreased with depth.

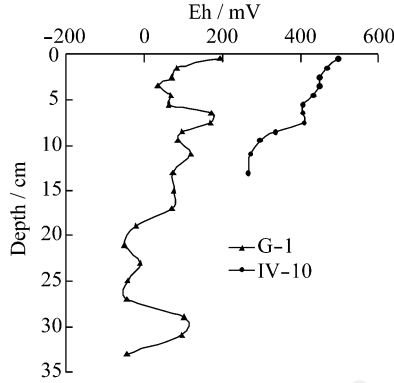


Fig 4 Profile distribution of Eh in sediments(mv).

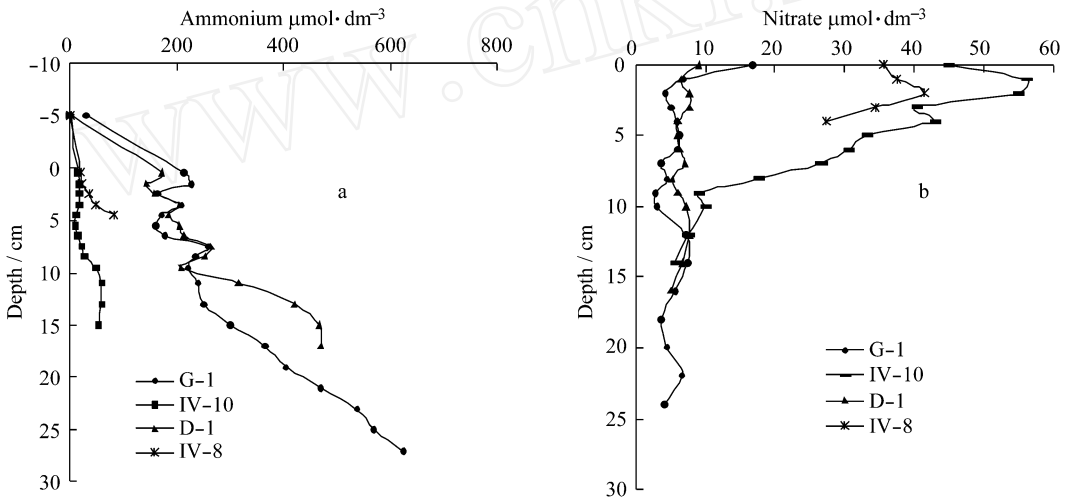


Fig 5 Profile distribution of ammonium and nitrate in sediment pore water ($\mu\text{mol}/\text{dm}^3$).

3. 3 Nutrients fluxes at the sediment-water interface

3. 3. 1 Fluxes estimated from pore water gradients

Considering only the state of molecular diffusion, the bottom flux of nutrients in sediment pore water can be calculated from Fick's First Law as:

$$F_s = D_s \frac{\partial C}{\partial Z} \tag{1}$$

Where F_s is the flux from sediment to overlying water ($\text{mmol}/(\text{m}^2 \cdot \text{d})$); θ denotes the porosity of the sediment ($0.78^{[31]}$); D_s is the coefficient of molecular diffusion in sediment (cm^2/s), from Li (1974)^[17]; $(\partial C/\partial Z)$ is the concentration gradation of nutrients in interstitial water of the sediment; Considering $D_s = \theta^2 D_o$ (D_o molecule diffusion coefficient in seawater, NH_4^+ : $9.8 \times 10^{-6} \text{ cm}^2 \text{ s}^{-1}$, Si: $5.5 \times 10^{-6} \text{ cm}^2/\text{s}$, NO_3^- : $9.78 \times 10^{-6} \text{ cm}^2/\text{s}^{[17]}$) thus the equation is:

$$F_s = D_s \frac{\partial C}{\partial Z} = D_o \theta^3 \frac{C}{Z} \tag{2}$$

The DSi profiles were fitted by Eq (3)^[4]:

$$C_x = C_d - (C_d - C_0) \exp(-x) \tag{3}$$

C_x is the DSi concentration at the depth x , C_d is the asymptotic DSi concentration, is the second fitting parameter, the curvature of the profile^[4]. The flux is then determined using Eq (4), which is the first derivative of Eq (3) at the interface, substituted into Eq (2):

$$F_s = -D_o (C_d - C_0) \tag{4}$$

Nutrients in sediment interstitial water except nitrate have the upward fluxes from sediment to overlying water, and nitrate has the downward fluxes from overlying water to sediment in G-1 and D-1 sediment cores. Fluxes of DSi fluctuated between 0.70 and 1.22 mmol/(m² · d), and NH₄ between 0.13 and 2.03 mmol/(m² · d) (Tab 2, 3). Although nitrate has the downward fluxes, compared with silicate and ammonium it was much lower (Tab 4). So we can conclude that in study areas, regeneration of silicate and ammonium are the main part of nutrients regeneration in the sediments.

3.3.2 Fluxes measured in situ incubation

The changes in the gradients of the DSi concentration in the tube incubation were used to calculate the flux of silicic acid coming from the dissolution of biogenic silica, either at the sediment-water interface or within the sediment column. Fluxes were estimated using linear regressions applied to the data presented in Fig 6. The measured fluxes of G-1 and N-10 were 1.74 and 1.22 mmol/(m² · d) (Table 2), respectively. The measured flux of N-10 was very similar to the fluxes calculated using pore water gradients, which was based

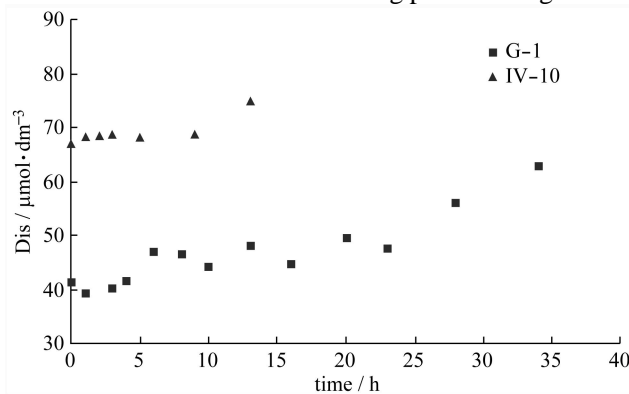


Fig 6 DSi concentration in the tube incubation as a function of time

Table 2 The benthic diffusion fluxes of nutrients in sediment pore water in the study stations

Station	C_0 SiO_3 -Si $\mu\text{mol}/\text{dm}^3$	C_d SiO_3 -Si $\mu\text{mol}/\text{dm}^3$	cm^{-1}	Calculated fluxes $\text{mmol}/(\text{m}^2 \cdot \text{d})$	Measured fluxes $\text{mmol}/(\text{m}^2 \cdot \text{d})$
G-1	102.75	309.71*		0.93**	1.74
N-10	68.74	651.30	0.82	1.19	1.22
N-8	79.53	624.30	0.90	1.22	
D-1	46.36	487.80	0.64	0.70	

* : the content of DSi in the surface pore water, not the C_d ; ** : the flux of G-1 was calculated based on the Eq (1)

on the Eq (4). It can be presumed that most of the diffusive fluxes are result from dissolution of BSi in the upper 5 cm of the sediment column in N-10.

4 Conclusion

From the resulted data, it is possible to draw the following conclusions

(1) High silicate and ammonium concentration was the main characteristics of Nutrients composition in sediment pore water of study areas, which concentration rang from 229.00-759.57 $\mu\text{mol}/\text{dm}^3$ and 10.70-623.60 $\mu\text{mol}/\text{dm}^3$, respectively

(2) Ammonia nitrogen comprised up to 98% of the three forms of total inorganic nitrogen. Profile distribution of ammonium increased with depth. However the profile distribution pattern of nitrate was contrary to ammonium. For ammonium and nitrate, the profile distributions basically agree with the anoxic process. Profile distributions of silicate indicate that the main diagenetic reactions of silicate were dissolution of silicate.

(3) Nutrients in sediment pore water except nitrate have the upward fluxes from sediment to overlying water. Regeneration of nutrients was mainly achieved by silicate and ammonium. Most of the diffusive fluxes are result from dissolution of BSi in the upper 5 cm of the sediment column.

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References

- [1] Bemer RA (1980): Early diagenesis: A theoretical approach. Princeton series in Geochemistry, Princeton, 241.
- [2] Zhou WH, Wu YH, Chen SY (2001): Study on nutrients (N, P, Si) in sediment interstitial waters in the district of Nansha Islands Sea area. *Tropic Oceanology*, 20 (4): 49-55.
- [3] Ragueneau O, Gallinari M, Corrin L *et al* (2000): The benthic silica cycle in the Northeast Atlantic: annual mass balance, seasonality, and importance of non-steady-state processes for the early diagenesis of biogenic opal in deep-sea sediments. *Progress in Oceanography*, 50: 171-200.
- [4] Mcmanus J, Hammond DE, Berelson WM *et al* (1995): Early diagenesis of biogenic opal: Dissolution rates, kinetics, and paleoceanographic implications. *Deep-Sea Research II*, 42: 871-903.
- [5] Archer D, Lyle M, Rodgers K *et al* (1993): What controls opal preservation in tropical deep-sea sediments?. *Paleoceanography*, 8: 7-21.
- [6] Cappellen V, Qiu PL (1996): Biogenic silica dissolution in sediments of the Southern Ocean. I. Solubility. *Deep-Sea Research II*, 44 (5): 1109-1128.
- [7] Marckin JE (1987): Boron and silica behavior in saltmarsh sediments: Implications for pale-boron distributions and the early diagenesis of silica. *American Journal of Science*, 287: 197-241.
- [8] Hurd DC, Birdwhistell S (1983): On producing a more general model for biogenic silica dissolution. *American Journal of Science*, 283: 1-28.

- [9] Song JM, Li Y (1991): Studies on models of " diffusion-advection-reaction " of silicate and sulfate in sediment interstitial water of the East China Sea Marine Sciences, 27-34.
- [10] Song JM (1997): Chemistry of sediment-sea water interface of the China Seas Beijing: China Ocean Press 6-8
- [11] Gu deyu N, Trefry H (1989): Pore water characteristic of a reductive sediment Acta Oceanologica Sinica, 11 (2): 171-175.
- [12] Pan JM, Zhou HY, Hu CY (2002): Nutrient profiles in interstitial water and flux in water-sediment interface of Zhujiang Estuary of China in summer, Acta Oceanologica Sinica, 24 (3): 52-59.
- [13] Chen SY, Han W Y (1994): The study on N and Si in sediment interstitial water of the Northeast South China Sea Tropic Oceanology, 0256-0260.
- [14] Ye XW, Liu SM, Zhang J (2002): Nutrients in Sediment Pore Water in Tidal Flat Area in Yalujiang Estuary Chinese Journal of Environmental Science, 0392-0396.
- [15] Ni JY, Michael M, Liu XY (2005): Pore water distribution and quantification of diffusive benthic fluxes of silicate, nitrate and phosphate in surface sediments of the equatorial northeastern Pacific Geochimica, 0649-0656.
- [16] Zhang DR, Chen FR, Yang YQ *et al* (2005): Benthic fluxes of nutrients in neritic zone outside Zhu Jiang River estuary in summer Journal of Tropical Oceanography, 0655-0662.
- [17] Li Y, Gregory S (1974): Diffusion of ions in sea water and deepsea sediment Geochimica et Cosmochimica Acta, 38: 703-714.