



## Effect of bovine feeding system (pasture or concentrate) on the oxidative and sensory shelf life of whole milk powder

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### ABSTRACT

Correlating volatile compounds with the sensory attributes of whole milk powder (WMP) is fundamental for appreciating the effect of lipid oxidation (LO) on sensory perception. LO compounds can adversely affect the sensory perception of WMP by imparting rancid, metallic, and painty notes. Whole milk powders derived from milk produced by cows maintained on a pasture diet (grass and grass-clover mix) versus a nonpasture diet [total mixed ration (TMR); concentrates and silage] were stored at room temperature 21°C (ambient storage) and 37°C (accelerated storage) and analyzed for volatile compounds and sensory attributes every 2 mo for a total of 6 mo. Thirteen volatile compounds originating from LO were chosen to track the volatile profile of the WMP during storage. Color, composition, total fatty acid, and free fatty acid profiling were also carried out. Significant variations in the concentrations of 14 fatty acids were observed in WMP based on diet. Concentrations of free fatty acids increased in all sample types during storage. Similar trends in sensory attributes were observed with an increase in painty attributes, corresponding to an increase in hexanal. Buttery/toffee attributes were found to be more closely correlated with TMR WMP. Those WMP derived from pasture diets were found to be more susceptible to LO from a volatile perspective, particularly in relation to aldehyde development, which is likely due to increased concentrations of conjugated linoleic acid and  $\alpha$ -linolenic acid found in these samples.

**Key words:** pasture, total mixed ration, sensory, volatile, whole milk powder

### INTRODUCTION

Whole milk powder (WMP) is an important dairy commodity that is largely produced in countries with an abundant supply of fresh milk and exported to be reconstituted and consumed directly as a nutritious beverage or used in soups and sauces, or in baking and confectionary (Early, 2012). Spray drying enables milk to be easily transported and stored as WMP for extended periods of time. However, the spray drying process can also facilitate oxidative changes as the high fat content is exposed to elevated temperatures, resulting in reduced shelf life due to off-flavor development. Moreover, WMP can also be subjected to extreme temperature fluctuations during transport and storage, further affecting oxidative stability. Lipid oxidation is a major cause of quality deterioration in fat-containing foodstuffs, which results in alterations to taste and odor through the creation of oxidation compounds, such as aldehydes, ketones, and alcohols.

Dairy products containing increased levels of specific PUFA may be more susceptible to lipid oxidation (LO; Hedegaard et al., 2006). Bovine diet is known to influence many aspects of milk composition, but especially the fatty acid (FA) profile (Liu et al., 2016; O'Callaghan et al., 2019). Quantifying both total fatty acids (TFA) and free fatty acids (FFA) in dairy products is important (Mannion et al., 2016, 2019), to understand the potential susceptibility of dairy products to LO and to know the abundance of specific FFA that can directly contribute to flavor. Bovine milk also contains various natural oxidants and antioxidants that can also be affected by feeding system, but to date very little research has been published on the susceptibility of milk or WMP to LO dependent on bovine feeding system.

Headspace solid phase-microextraction (HS-SPME) GC-MS is a widely used technique for the identifica-

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tion and quantification of volatile compounds in dairy products (Tunick et al., 2013). A validated HS-SPME GC-MS method (Clarke et al., 2019) was used to quantify 13 compounds from 3 chemical classes (aldehydes, ketones, and alcohols) associated with LO. Quantitative descriptive analysis has previously been applied to a variety of dairy products (Stone and Sidel, 1998), and is based on the ability of trained panelists to measure specific attributes of a product (Chapman et al., 2001). This technique was employed to track the development of taste and odor attributes in WMP as LO progressed at 2 storage temperatures (21°C and 37°C) in opened lidded cans. Panelists (n = 9) were trained to rate the WMP samples based on selected relevant sensory attributes (Drake et al., 2003; Lloyd et al., 2009a,b; Park et al., 2016), which were defined and agreed upon by all panelists before final scoring. Analyses of TFA, FFA content, color, composition, sensory attributes, and volatile profile of the WMP were undertaken at 3 time points over a predefined period of 6 mo.

## MATERIALS AND METHODS

### Milk Production

Fifty-four lactating Friesian cows were divided into 3 groups, namely, grass-only cows (**GRS**), grass-clover cows (**CLV**), and TMR cows (n = 18). The GRS cows were maintained outdoors on perennial ryegrass (*Lolium perenne* L.) and received approximately 2 kg of concentrate and 15 kg DM of grass per cow; CLV cows were also maintained outdoors on perennial ryegrass and white clover mix (*Trifolium repens* L.) and received 2 kg of concentrate and 15 kg DM of grass-clover per cow; TMR cows were housed indoors and received 9 kg DM of maize silage + 4.5 kg DM of grass silage + 8.5 kg DM of concentrate throughout the study. Cows within the TMR group were fed daily in electronically controlled Griffith Elder Mealmaster individual feed bins (Griffith Elder and Company Ltd.), and feed was available ad libitum. The CLV sward contained ~20% white clover, as outlined by O'Callaghan et al. (2016). Cows in the GRS and CLV groups received a mineral supplement in the form of a liquid mineral preparation injected into the water supply (Terra Liquid Minerals), giving a mean intake (mg/cow per day) of Na, Mg, Zn, Cu, Se, and Co of 5.0, 1.2, 219, 106, 3.8, and 3.0, respectively. The concentrate portion of the TMR feed was supplemented with a commercial mineral balancer (Dairy Hi-Phos; McDonnell Bros. Agricultural Suppliers Ltd.) to give added Ca, Na, P, Zn, Cu, Mn, I, Co, and Se of 3,340, 2000, 1200, 140, 100, 70, 10, 2, and 0.8 mg/kg, respectively (Gulati et al., 2018).

### Manufacture of WMP

Medium-heat WMP was produced in triplicate from milk from each group (n = 18) at Moorepark Technology Limited BioFunctional Food Engineering pilot plant (Teagasc, Moorepark, Fermoy, Co. Cork, Ireland) over a 3-wk period in May 2019. Raw whole milk was collected both morning and evening from each group (GRS, CLV, and TMR) for 3 consecutive days, and each batch (~1,000 kg) was preheated to 50°C and pasteurized at 90°C for 30 s in an APV plate heat-exchanger (SPX Flow Technology), before homogenization using an APV-Gaulin 2-stage homogenizer at first- and second-stage pressures of 150 and 50 bar, respectively. The homogenized milk was evaporated to ~40% TS in a single-effect recirculating evaporator (Scheffers). The concentrate was preheated to 65°C in a plate heat-exchanger and transferred to an Anhydro 3-stage spray dryer (SPX Flow Technology Denmark A/S; air inlet temperature 170°C; air outlet temperature 65°C). First and second fluid bed temperatures were set at 65°C and 25°C, respectively. Fines were returned to the top of the spray dryer from the second fluid bed and the cyclone, yielding an agglomerated WMP of approximately 97% TS. Production of WMP from each feeding system was carried out in triplicate from 3 independent raw milk collections outlined previously (Magan et al., 2019). All equipment was cleaned thoroughly between batches. All WMP were packed in 400-g aluminum cans, flushed with N<sub>2</sub>, and sealed immediately. The WMP produced from grass-only, grass-clover, and TMR milk were denoted as GRS, CLV, and TMR, respectively. All results reported are the averages of triplicate analysis of the 3 production batches for GRS, CLV, and TMR powders (n = 9) unless otherwise stated.

### Shelf Life Study Design

The fresh WMP produced from each feeding system were split into 2 storage groups, 21°C and 37°C; this resulted in 6 samples in total, denoted GRS 21°C, GRS 37°C, CLV 21°C, CLV 37°C, TMR 21°C, and TMR 37°C. All sample cans were opened at **T0**, and initial sensory and volatile measurements were taken. The opened cans were used for all subsequent analysis throughout the study, and plastic lids were placed on the open cans (closed but not sealed) while in storage (Cesa et al., 2015), to best reflect typical use in a domestic environment. The time points chosen for analysis of WMP color, volatile profile, and sensory perception were 0 (**T0**), 2 (**T2**), 4 (**T4**), and 6 (**T6**) mo. The FFA content was assessed at T0, T4, and T6, whereas TFA content and WMP composition were assessed at T0 only.

### Color Measurements

Color measurements were performed on the GRS, CLV, and TMR WMP stored at 21°C and 37°C at each time point according to the CIE Lab system (CIE, 1978;  $L^*$  is a measure of lightness;  $a^*$  is a measure of green-to-red color on a negative-to-positive scale, respectively, and  $b^*$  is a measure of blue-to-yellow color on a negative-to-positive scale, respectively), using a Minolta colorimeter (Minolta Camera). All WMP were reconstituted to 12% TS using distilled water ( $dH_2O$ ) 24 h before analysis and chilled at 4°C. Approximately 2 mL of sample was placed in a spectrophotometric cuvette and allowed to stabilize at room temperature for 30 min before analysis, as per Faulkner et al. (2018). Results were expressed as the average of triplicate measurements of each liquid sample.

### Powder Composition

Each WMP was analyzed for fat, protein, lactose, true protein, and casein content directly after manufacture (T0) using a Bentley DairySpec FT (Technopath Distribution). The WMP samples were reconstituted to 3.5% fat using  $dH_2O$ , as per the equation outlined by the International Dairy Federation (IDF, 1997) 24 h before analysis. Results were expressed as the averages of 2 replicates.

### Fatty Acid Profiling

**Free Fatty Acid Profiling.** Analysis of FFA was carried out on the GRS, CLV, and TMR WMP at T0, T4, and T6 mo of storage at 21°C and 37°C. Lipid extraction, butyl ester derivatization of triglycerides, solid-phase extraction and gas chromatography instrument conditions were performed as per Mannion et al. (2019). Each powder (4 g) was analyzed in duplicate, and the extracts were pooled for solid-phase extraction.

**Total Fatty Acid Profiling.** Profiling of TFA was carried out on the GRS, CLV, and TMR WMP at T0. Lipid extraction and methyl ester derivatization of triglycerides were carried out as per De Jong and Badings (1990) and O'Callaghan et al. (2019). The WMP were reconstituted to 12% TS using  $dH_2O$  1 h before analysis, and 10 mL of reconstitute was used for analysis. The GC conditions were also outlined by O'Callaghan et al. (2019). Briefly, analysis was performed on an Agilent 7890A GC, equipped with an Agilent 7693 autosampler (Agilent Technologies Ltd.) and flame ionization detector. The column was a Select FAME capillary column (100 m × 250- $\mu$ m internal diameter, 0.25- $\mu$ m phase thickness; part number CP7420; Agilent Technologies Ltd.). The injector was held at 250°C for the entire

run and was operated in split mode using a split ratio of 1:10. The column oven was held at 80°C for 8 min, raised to 200°C at 8.5°C/min, and held for 55 min. The total runtime was 77.12 min. The flame ionization detector was operated at 300°C. Results were processed using OpenLab CDS Chemstation edition software version Rev.C.01.04 (35) (Agilent Technologies Ltd.).

A 37-component FAME reference mix containing C4:0 to C24:0 (part number 35077; Thames Restek UK Ltd.) was analyzed as an in-run quality control sample, with the FAME present at concentrations of 60 to 180 mg/kg. This was used to ensure that accurate quantification was being achieved throughout sample analysis. The FAME mix was analyzed once every 5 samples in the sequence. Accuracy was monitored by comparing the measured concentration of this FAME mix against its true concentration.

### Sensory Evaluation

Quantitative descriptive analysis was carried out on the WMP in Teagasc Ashtown (Dublin, Ireland). An external trained sensory panel consisting of 9 members was recruited, based on their ability to perceive a wide variety of attributes and their continued availability. Panelists had between 3 and 4 yr experience working as descriptive panelists on a weekly basis. Panel training for WMP evaluation consisted of 2 attribute generation sessions (3 h duration each), wherein the panelists evaluated a variety of volatile compounds on cotton wool (Supplemental Table S1, <http://hdl.handle.net/11019/2424>) in addition to the use of 12 Sniffing Sticks (cardboard, rancid, butter, soapy, musty/cellar, cheesy/sweaty, mushroom, earthy, malty, cabbage, animal/stable, and fishy; Dohler GmbH) designed specifically for this study. A further 4 sessions of panel training were carried out using a variety of product standards to create aroma, texture, flavor, and aftereffect scales for each sensory descriptor that was subsequently applied to the GRS, CLV, and TMR WMP (Supplemental Table S2, <http://hdl.handle.net/11019/2424>). Panelists used the consensus list of descriptors for scoring, measured on a scale of 1 to 10 with 1 to 3 considered low intensity, 4 to 7 considered medium intensity, and 8 to 10 considered high intensity. Triangle tests were used before the final scoring to ensure that panelists were performing within expectations. The WMP were reconstituted 24 h before scoring based on the fat and protein content (IDF, 1997) using  $dH_2O$  and were stored at 2°C until approximately 1 h before each training and scoring session. Samples were allowed to reach 11 to 12°C before serving. The reconstituted WMP were gently stirred and poured into 20-mL clear plastic cups, which were labeled with random 3-digit codes. Panelists were given

water and plain crackers or green apples to cleanse the palate between samples. The project was set up as a complete block design using Compusense 5.6 (sensory data capture package, <https://compusense.com/>). The WMP were scored in triplicate for each trial replicate (GRS, CLV, and TMR), for each descriptor, and the results were averaged ( $n = 9$ ). Analysis of color was also carried out by the panelists on each WMP.

### Volatile Analysis

Thirteen volatile aromatic compounds including 7 aldehydes [hexanal, pentanal, heptanal, octanal, (E)-2-nonenal, 2,4-decadienal, undecanal], 4 ketones (2-heptanone, 2-nonanone, 2-pentanone, 3-octen-2-one), and 2 alcohols (1-heptanol and 1-pentanol) known to be important to the sensory perception of dairy products were selected for quantification based on current literature (Van Aardt et al., 2005; Faulkner et al., 2018; Kilcawley et al., 2018). Authentic standards for each of the compounds were obtained from Merck Ireland and stored at room temperature. All standard solutions for HS-SPME GCMS analysis were prepared at 0.1% (wt/vol) in methanol and stored at  $-18^{\circ}\text{C}$  until required for analysis, but for no longer than 6 mo. For all calibration curves the external standard mixture was prepared at 0.004% (wt/vol; 4 mL in 100 mL of  $\text{dH}_2\text{O}$ ) and the internal standard mixture (2-methyl-3-heptanone, 4-methyl-2-pentanol, and isovaleraldehyde) was prepared at 0.001% (wt/vol; 1 mL in 100 mL of  $\text{dH}_2\text{O}$ ). For the preparation of calibration curves, varying levels of the standard mixture were prepared in 10-mL volumetric flasks with  $\text{dH}_2\text{O}$ .

The HS-SPME GCMS analysis was carried out as described by Clarke et al. (2019) at T0, T2, T4, and T6 sampling points. Briefly, WMP (2.40 g) was weighed out directly into an amber La-Pha-Pack head-space vials (20 mL) with magnetic caps and silicone/polytetrafluoroethylene 1.3-mm  $45^{\circ}$  Shore A septa (Apex Scientific Ltd.). Then  $\text{dH}_2\text{O}$  (2.50 g) and 250  $\mu\text{L}$  of 0.001% (wt/vol) internal standard were added to each sample. A calibration curve was also prepared by spiking a set of the reconstituted WMP samples with varying levels of the external standard mixture. Matrix (control) samples (WMP sample +  $\text{dH}_2\text{O}$  only) were also included in each run. Samples were incubated for 45 min at a temperature of  $43^{\circ}\text{C}$  using a CombiPal agitator/heater module (Elementec Ltd.), followed by a 10-min pre-extraction incubation time with pulsed agitation of 5 s at 500 rpm. Each sample was analyzed in triplicate at each time point.

### Statistical Analysis

Statistical analysis for data relating to color and composition was carried out using one-way ANOVA with post-hoc Tukey tests using SPSS software, version 24 (IBM Corp.). Pearson correlation analysis was carried out on the sensory and volatile data using SPSS. Principal component analysis biplots of the volatile versus sensory data were constructed using the factextra and FactoMineR packages within R (v. 3.4.1; R Core Team, 2013). All sensory and volatile data were averaged before analysis.

## RESULTS AND DISCUSSION

### Milk Color

The TMR WMP scored significantly ( $P < 0.05$ ) lower for  $a^*$  and  $b^*$  values compared with the GRS and CLV WMP at each time point (Table 1). The TMR WMP scored significantly ( $P < 0.05$ ) higher for  $L^*$  values at T0 and T2, and, although not statistically different at T4 and T6,  $L^*$  values remained higher than GRS and CLV WMP. These results were in agreement with previous studies on dairy products produced from pasture versus concentrate-based feeding systems (Faulkner et al., 2018; Clarke et al., 2020a). The significant differences in color between pasture-derived dairy products and those derived from concentrate has previously been attributed to the abundance of  $\beta$ -carotene in pasture forages, which results in a more yellow or creamy color (Martin et al., 2005). This study corroborates these results with TMR samples scoring significantly ( $P < 0.05$ ) higher for white color, whereas pasture-derived products (GRS and CLV) were scored higher for creamy color by the sensory panel. The TMR WMP also scored significantly ( $P < 0.05$ ) lower for  $a^*$  and  $b^*$  values compared with the GRS and CLV powders at each time point. Lightness also varied significantly ( $P < 0.05$ ) between the 3 WMP at T0 and T2. Although lightness was not statistically significant at T4 and T6, similar trends in lightness were observed between the different types of WMP, with slight increases observed. The  $b^*$  values (blue-to-yellow) also increased slightly in all 3 WMP from T0 to T6, possibly due to Maillard browning reactions occurring during storage (Bastos et al., 2012). The rate of Maillard browning is known to be affected by several factors, including the chemical nature of the reactants (type of amine and carbonyl groups), water activity, pH, temperature, heating time, and protein-to-sugar ratio (Labuza and Baiser, 1992; Rozycki et al., 2007).

**Table 1.** Results of color measurements ( $n = 3$ ) taken of the 3 reconstituted whole milk powders derived from milk from cows fed perennial ryegrass (GRS), perennial ryegrass and white clover (CLV), or TMR at times (T) 0, 2, 4, and 6 mo of storage at 21°C and 37°C<sup>1</sup>

Sample	L*	a* (-)	b*
T0			
GRS	80.97 <sup>c</sup>	3.24 <sup>a</sup>	7.24 <sup>a</sup>
CLV	82.16 <sup>b</sup>	3.24 <sup>a</sup>	6.96 <sup>a</sup>
TMR	83.26 <sup>a</sup>	2.84 <sup>b</sup>	4.86 <sup>b</sup>
T2			
GRS 21°C	86.02 <sup>b</sup>	3.51 <sup>a</sup>	8.11 <sup>a</sup>
CLV 21°C	85.94 <sup>b</sup>	3.32 <sup>a</sup>	7.75 <sup>a</sup>
TMR 21°C	86.56 <sup>a</sup>	2.86 <sup>b</sup>	6.37 <sup>b</sup>
GRS 37°C	85.40 <sup>b</sup>	3.51 <sup>a</sup>	8.02 <sup>a</sup>
CLV 37°C	85.88 <sup>b</sup>	3.49 <sup>a</sup>	7.84 <sup>a</sup>
TMR 37°C	86.14 <sup>a</sup>	2.91 <sup>b</sup>	5.62 <sup>b</sup>
T4			
GRS 21°C	85.92	3.39 <sup>a</sup>	8.04 <sup>a</sup>
CLV 21°C	85.58	3.35 <sup>a</sup>	7.38 <sup>a</sup>
TMR 21°C	86.02	2.59 <sup>b</sup>	5.28 <sup>b</sup>
GRS 37°C	85.72	3.41 <sup>a</sup>	8.01 <sup>a</sup>
CLV 37°C	82.64	3.35 <sup>a</sup>	7.73 <sup>a</sup>
TMR 37°C	85.67	2.58 <sup>b</sup>	5.38 <sup>b</sup>
T6			
GRS 21°C	85.49	3.40 <sup>a</sup>	7.77 <sup>a</sup>
CLV 21°C	82.52	3.28 <sup>a</sup>	7.55 <sup>a</sup>
TMR 21°C	85.87	2.72 <sup>b</sup>	5.41 <sup>b</sup>
GRS 37°C	84.90	3.52 <sup>a</sup>	7.55 <sup>a</sup>
CLV 37°C	84.97	3.43 <sup>a</sup>	7.51 <sup>a</sup>
TMR 37°C	85.06	2.78 <sup>b</sup>	5.21 <sup>b</sup>

<sup>a-c</sup>Mean values in the same column (analyzed by time point: T0, T2, T4, T6, representing 0, 2, 4, and 6 mo) with different superscripts differ ( $P < 0.05$ ) based on feeding system.

<sup>1</sup>Each result is the average of triplicate analysis of WMP derived from 3 production trials for GRS-, CLV-, and TMR-based feeding systems ( $n = 9$ ). L\* is a measure of lightness; a\* is a measure of green-to-red color on a negative-to-positive scale, respectively; b\* is a measure of blue-to-yellow color on a negative-to-positive scale, respectively. All a\* values are negative, indicated by (-).

### Milk Powder Composition

No significant differences were observed between the fat, protein, lactose, true protein, and casein contents of the GRS, CLV, and TMR WMP analyzed directly after manufacture at T0 (Supplemental Table S3).

### Fatty Acid Profiling

**Free Fatty Acid Profiling.** Levels of FFA in the WMP were quantified at T0, T4, and T6. Results showed that C16:0 (palmitic acid) was the most abundant FFA in all sample types at T0 (62–67 mg/kg), followed by C18:0 (stearic acid; 33–38 mg/kg), and C18:1 (oleic acid; 26–32 mg/kg; Table 2). The concentrations of FFA in the GRS, CLV, and TMR WMP did not vary significantly ( $P < 0.05$ ) at T0. However, increases in C16:0, C18:0, and C18:1 were observed in all 3 WMP from T0 to T6. Significant ( $P < 0.001$ ) increases in

the concentrations of C18:0 were observed in CLV and TMR WMP from T4 to T6. Increases in the total FFA content were also observed in all WMP from T0 to T6. The levels of individual FFA in the GRS, CLV, and TMR WMP were comparable to those reported by Páez et al. (2006).

Certain FA are directly responsible for off-flavors such as rancid, astringent, and butyric (Deeth 2006), but, perhaps more importantly from a sensory standpoint, FA are precursors of oxidation reactions resulting in the production of aldehydes and ketones, which are responsible for oxidized, metallic, and tallowy off-flavors often observed in milk powders (Muir, 1996; Páez et al., 2006). It has been speculated that powders derived from pasture may be more susceptible to LO due to the increased presence of PUFA such as arachidonic acid and docosahexaenoic acid (Kilcawley et al., 2018); however, it has been noted that the presence or concentration of natural antioxidants and pro-oxidants are also major factors (Romeu-Nadal et al., 2004).

**Total Fatty Acid Profiling.** A total of 29 FA were quantified (g/100 g of milk fat) in the WMP. It was evident that feeding system had a significant ( $P = 0.05$ ) effect on the concentrations of 14 of the 29 FA quantified in the WMP, which is in agreement with previous studies on milk and WMP (Semeniuc et al., 2008; O'Callaghan et al., 2016). The 14 FA that varied significantly ( $P < 0.05$ ) based on feeding system were undecanoic acid (C11:0), tridecanoic acid (C13:0), myristoleic acid (C14:1), pentadecanoic acid (C15:0), palmitic acid (C16:0), *trans*-9,12-octadecadienoate (C18:2 n6t),  $\alpha$ -linolenic acid (C18:3 n3),  $\gamma$ -linolenic acid (C18:3 n6), eicosanoic acid (C20:0), eicosenoic acid (C20:2), eicosapentaenoic acid (C20:5), heneicosanoic acid (C21:0), CLA (c10t12), and CLA (c9t11; Table 3).

As previously reported in milk (O'Callaghan et al., 2016), palmitic acid (C16:0) and oleic acid (C18:1 n9c) were found to be the most abundant FA in all WMP analyzed. Average palmitic acid content was highest in TMR powders ( $29.94 \pm 2.98$ ), significantly ( $P < 0.05$ ) higher than in GRS ( $24.96 \pm 0.90$ ) and CLV ( $25.06 \pm 0.28$ ) WMP. The concentration of oleic acid (C18:1 n9c) did not vary significantly between the WMP in this study at T0. Higher proportions of CLA have previously been reported in milk from cows consuming significant quantities of grazed grass than cows whose diet primarily consists of conserved forages and concentrates (Kelly et al., 1998). The significantly ( $P < 0.05$ ) higher proportions of CLA observed in pasture-derived WMP (GRS and CLV) agrees with studies on milk and cheese from the same feeding systems used in this study (O'Callaghan et al., 2016, 2017).

**Table 2.** Concentrations (mg/kg) of the individual free fatty acids (FFA: C4, C6, C8, C10, C12, C14, C16, C18, C18:1, C18:2, and C18:3,  $\pm$  SD) quantified in whole milk powders (WMP) derived from milk from cows fed perennial ryegrass (GRS), perennial ryegrass and white clover (CLV), or TMR at times (T) 0, 4, and 6 mo of storage at 37°C<sup>1</sup>

Fatty acid	GRS T0	CLV T0	TMR T0	GRS T4	CLV T4	TMR T4	GRS T6	CLV T6	TMR T6	TMR T6	P-value
Butyric acid C4:0	16 $\pm$ 7.11	10 $\pm$ 7.09	10 $\pm$ 7.64	13 $\pm$ 3.34	18 $\pm$ 18.42	16 $\pm$ 2.03	16 $\pm$ 8.53	13 $\pm$ 3.05	9 $\pm$ 1.59	9 $\pm$ 1.59	NS 0.71
Caproic acid C6:0	14 $\pm$ 7.43	9 $\pm$ 7.68	8 $\pm$ 7.65	9 $\pm$ 0.13	9 $\pm$ 9.34	9 $\pm$ 1.11	10 $\pm$ 2.80	10 $\pm$ 1.38	8 $\pm$ 1.06	8 $\pm$ 1.06	NS 0.95
Octanoic acid C8:0	14 $\pm$ 7.83	8 $\pm$ 8.14	8 $\pm$ 7.92	6 $\pm$ 0.57	7 $\pm$ 0.43	6 $\pm$ 1.22	6 $\pm$ 1.81	6 $\pm$ 1.22	4 $\pm$ 1.21	4 $\pm$ 1.21	NS 0.75
Decanoic acid C10:0	18 $\pm$ 9.21	12 $\pm$ 9.50	11 $\pm$ 9.20	8 $\pm$ 1.39	10 $\pm$ 0.55	10 $\pm$ 3.01	11 $\pm$ 3.07	11 $\pm$ 1.36	9 $\pm$ 2.64	9 $\pm$ 2.64	NS 0.84
Lauric acid C12:0	22 $\pm$ 11.07	14 $\pm$ 11.81	14 $\pm$ 10.84	12 $\pm$ 4.20	21 $\pm$ 0.15	17 $\pm$ 6.96	18 $\pm$ 8.38	16 $\pm$ 2.80	12 $\pm$ 3.40	12 $\pm$ 3.40	NS 0.88
Myristic acid C14:0	28 $\pm$ 10.52	21 $\pm$ 11.74	21 $\pm$ 10.54	19 $\pm$ 4.57	27 $\pm$ 2.84	24 $\pm$ 10.36	34 $\pm$ 6.84	36 $\pm$ 4.20	30 $\pm$ 7.54	30 $\pm$ 7.54	NS 0.42
Palmitic acid C16:0	67 $\pm$ 3.07	62 $\pm$ 7.64	67 $\pm$ 12.20	64 $\pm$ 17.13	88 $\pm$ 18.38	88 $\pm$ 37.90	113 $\pm$ 16.46	121 $\pm$ 8.05	117 $\pm$ 24.58	117 $\pm$ 24.58	0.02*
Stearic acid C18:0	38 $\pm$ 6.46	33 $\pm$ 7.31	33 $\pm$ 6.86	15 $\pm$ 10.86	9 $\pm$ 1.12	8 $\pm$ 2.60	36 $\pm$ 17.66	55 $\pm$ 3.97	52 $\pm$ 7.22	52 $\pm$ 7.22	<0.001*
Oleic acid C18:1	32 $\pm$ 5.11	26 $\pm$ 7.09	27 $\pm$ 8.28	29 $\pm$ 11.91	45 $\pm$ 14.21	42 $\pm$ 22.68	57 $\pm$ 11.62	54 $\pm$ 12.77	46 $\pm$ 10.22	46 $\pm$ 10.22	NS 0.11
Linoleic acid C18:2	13 $\pm$ 8.86	6 $\pm$ 8.46	6 $\pm$ 8.86	2 $\pm$ 1.37	0 $\pm$ 0.00	0 $\pm$ 0.00	3 $\pm$ 2.23	5 $\pm$ 0.67	7 $\pm$ 1.40	7 $\pm$ 1.40	NS 0.37
$\alpha$ -Linolenic acid C18:3	10 $\pm$ 7.35	5 $\pm$ 7.09	5 $\pm$ 6.75	1 $\pm$ 1.93	0 $\pm$ 0.00	2 $\pm$ 2.72	0 $\pm$ 0.00	1 $\pm$ 2.02	0 $\pm$ 0.00	0 $\pm$ 0.00	NS 0.31
Total FFA	272 $\pm$ 30.93	205 $\pm$ 45.42	210 $\pm$ 41.21	179 $\pm$ 20.72	235 $\pm$ 17.18	222 $\pm$ 38.38	305 $\pm$ 13.36	329 $\pm$ 8.28	294 $\pm$ 19.16	294 $\pm$ 19.16	NS 0.338

<sup>1</sup>Each result is the average of duplicate analysis of WMP derived the from 3 production trials for GRS-, CLV-, and TMR-based feeding systems (n = 6). NS = not significant. \*Significant differences in FFA composition when  $P = 0.05$ .

## Sensory Evaluation

White color was more closely associated with TMR WMP, whereas creamy color was associated with pasture-derived (GRS and CLV) WMP. At T0, significant ( $P < 0.001$ ) differences were observed for color, dairy sweet aroma, buttery/toffee aroma, and buttery/toffee flavor between the different WMP samples. The attribute dairy sweet aroma was higher in GRS and CLV WMP, with buttery/toffee aroma and flavor higher in TMR WMP. The white color association with the TMR WMP observed by panelists is in agreement with the instrumental color analysis. At T2, the differences between the WMP were more apparent for both storage temperatures (21°C and 37°C). At T2 cooked milk aroma, dairy sweet flavor, cooked milk flavor, and dairy sweet aftereffect varied significantly ( $P < 0.05$ ) between the different WMP.

For the pasture-derived WMP (GRS and CLV), creamy flavor and dairy sweet flavor dominated at T0. However, a barnyard/cow aroma and flavor, cooked milk aroma, and hay-like flavor and aroma were to the fore at T2. An increase in painty flavor and astringency were observed in GRS and CLV WMP stored at 37°C at T4 and T6, which corresponded with an increase the concentration of all volatile compounds. The increases in painty flavor and painty flavor aftereffect were more pronounced in CLV WMP compared with GRS WMP samples (Figure 1A and B). In TMR WMP, the pleasant attributes that are commonly associated with fresh reconstituted WMP (creamy and dairy sweet) were dominant at T0. At T2, buttery/toffee and hay-like flavors became more pronounced in both TMR 21°C and TMR 37°C. At T4, dairy sweet aroma remained, but metallic off-flavors began to be perceived. At T6, painty flavor and solvent-like aroma dominated in TMR WMP stored at 37°C, whereas metallic flavor and cooked milk aroma were to the fore in TMR WMP stored at 21°C (Figure 1C). When comparing the 3 WMP (GRS, CLV, and TMR) dairy sweet flavor, creamy flavor, creaminess, and viscosity were associated with all 3 WMP samples at T0. Barnyard/cow aroma, hay-like aroma, and hay-like flavor were more closely correlated with the GRS and CLV WMP at T2, whereas painty aroma, painty flavor, and painty flavor aftereffect were more correlated with TMR samples stored at 37°C at T6.

The ability of the sensory panelists to identify and rate the intensity of a painty flavor and aroma in some WMP suggests that the levels of volatile aromatic compounds responsible for this attribute had increased above their odor thresholds over the 6-mo storage period. For example, hexanal, derived primarily from linoleic acid, has been shown to be responsible for a painty off-flavor in milk powders (Lloyd et al., 2009a; Clarke et al.,

**Table 3.** Mean fatty acid composition of whole milk powders (g/100 g of milk fat  $\pm$  SD) produced from milk from cows fed perennial ryegrass (GRS), perennial ryegrass and white clover (CLV), or TMR<sup>1</sup>

Fatty acid <sup>2</sup>	GRS	CLV	TMR	<i>P</i> -value <sup>3</sup>	
Butyric acid C4:0	6.02 $\pm$ 0.34	5.81 $\pm$ 0.25	5.20 $\pm$ 0.55	NS	0.24
Caproic acid C6:0	2.62 $\pm$ 0.10	2.55 $\pm$ 0.10	2.28 $\pm$ 0.20	NS	0.16
Octanoic acid C8:0	1.53 $\pm$ 0.06	1.50 $\pm$ 0.06	1.41 $\pm$ 0.05	NS	0.17
Decanoic acid C10:0	3.70 $\pm$ 0.18	3.69 $\pm$ 0.30	3.25 $\pm$ 0.24	NS	0.132
Undecanoic acid C11:0	0.12 $\pm$ 0.02 <sup>a</sup>	0.13 $\pm$ 0.03 <sup>a</sup>	0.08 $\pm$ 0.00 <sup>b</sup>	*	0.002
Lauric acid C12:0	4.15 $\pm$ 0.18	4.25 $\pm$ 0.37	3.80 $\pm$ 0.27	NS	0.246
Tridecanoic acid C13:0	0.14 $\pm$ 0.02 <sup>a</sup>	0.16 $\pm$ 0.03 <sup>a</sup>	0.10 $\pm$ 0.00 <sup>b</sup>	*	0.002
Myristic acid C14:0	10.98 $\pm$ 0.09	11.40 $\pm$ 0.30	10.45 $\pm$ 0.89	NS	0.363
Myristoleic acid C14:1 c9	0.84 $\pm$ 0.01 <sup>ab</sup>	0.95 $\pm$ 0.06 <sup>a</sup>	0.76 $\pm$ 0.04 <sup>b</sup>	*	0.006
Pentadecanoic acid C15:0	1.77 $\pm$ 0.07 <sup>a</sup>	1.98 $\pm$ 0.08 <sup>a</sup>	1.44 $\pm$ 0.13 <sup>b</sup>	*	<0.001
Palmitic acid C16:0	24.96 $\pm$ 0.90 <sup>b</sup>	25.06 $\pm$ 0.28 <sup>b</sup>	29.94 $\pm$ 2.98 <sup>a</sup>	*	0.003
Palmitoleic acid C16:1 c9	1.26 $\pm$ 0.06	1.20 $\pm$ 0.08	1.24 $\pm$ 0.07	NS	0.561
Heptadecanoic acid C17:0	0.52 $\pm$ 0.01	0.52 $\pm$ 0.01	0.50 $\pm$ 0.02	NS	0.561
Stearic acid C18:0	7.71 $\pm$ 0.46	7.52 $\pm$ 0.29	7.91 $\pm$ 0.50	NS	0.477
Oleic acid C18:1n-9 c	13.35 $\pm$ 0.41	13.03 $\pm$ 0.28	13.13 $\pm$ 0.11	NS	0.657
Elaidic acid C18:1n-9 t	6.55 $\pm$ 0.16	6.48 $\pm$ 0.15	6.62 $\pm$ 0.48	NS	0.943
Linoleic acid C18:2n-6 c	0.84 $\pm$ 0.02 <sup>b</sup>	0.85 $\pm$ 0.02 <sup>b</sup>	1.29 $\pm$ 0.08 <sup>a</sup>	*	<0.001
<i>Trans</i> -9,12-octadecadienoate C18:2n-6 t	10.54 $\pm$ 0.68	10.35 $\pm$ 0.95	9.43 $\pm$ 1.05	NS	0.407
$\alpha$ -Linolenic acid C18:3n-3	0.50 $\pm$ 0.02 <sup>a</sup>	0.52 $\pm$ 0.03 <sup>a</sup>	0.26 $\pm$ 0.02 <sup>b</sup>	*	<0.001
Gamma linolenic acid C18:3n-6	0.05 $\pm$ 0.00 <sup>b</sup>	0.05 $\pm$ 0.00 <sup>ab</sup>	0.06 $\pm$ 0.00 <sup>a</sup>	*	0.013
Eicosanoic acid C20:0	0.07 $\pm$ 0.00 <sup>b</sup>	0.07 $\pm$ 0.00 <sup>b</sup>	0.12 $\pm$ 0.00 <sup>a</sup>	*	<0.001
<i>Cis</i> -11-eicosenoic acid C20:1	0.03 $\pm$ 0.00	0.03 $\pm$ 0.01	0.03 $\pm$ 0.00	NS	0.79
Eicosenoic acid C20:2	0.01 $\pm$ 0.01 <sup>b</sup>	0.01 $\pm$ 0.01 <sup>b</sup>	0.03 $\pm$ 0.00 <sup>a</sup>	*	0.02
Eicosadienoic acid C20:3n-6	0.04 $\pm$ 0.00	0.04 $\pm$ 0.00	0.06 $\pm$ 0.01	NS	0.18
Eicosapentaenoic acid C20:5	0.05 $\pm$ 0.00 <sup>b</sup>	0.06 $\pm$ 0.00 <sup>b</sup>	0.00 $\pm$ 0.00 <sup>a</sup>	*	<0.001
Heneicosanoic acid C21:0	0.05 $\pm$ 0.01 <sup>a</sup>	0.06 $\pm$ 0.02 <sup>a</sup>	0.01 $\pm$ 0.00 <sup>b</sup>	*	<0.001
Tricosanoic acid C23:0	0.01 $\pm$ 0.01	0.01 $\pm$ 0.02	0.00 $\pm$ 0.00	NS	0.28
CLA (c10t12)	1.02 $\pm$ 0.06 <sup>a</sup>	1.15 $\pm$ 0.13 <sup>a</sup>	0.25 $\pm$ 0.11 <sup>b</sup>	*	<0.001
CLA (c9t11)	0.54 $\pm$ 0.04 <sup>a</sup>	0.56 $\pm$ 0.06 <sup>a</sup>	0.33 $\pm$ 0.03 <sup>b</sup>	*	<0.001

<sup>a,b</sup>Mean values in the same row with different superscripts differ ( $P = 0.05$ ) based on feeding system.

<sup>1</sup>Each result is the average of duplicate analysis of WMP derived from 3 production trials for GRS-, CLV-, and TMR-based feeding systems ( $n = 6$ ). Statistical analysis was carried out by one-way ANOVA.

<sup>2</sup>c = *cis*; t = *trans*.

<sup>3</sup>NS = not significant.

\*Significant differences in fatty acid composition when  $P = 0.05$ .

2020b) and was present in GRS, CLV, and TMR WMP at highest concentrations (1,957, 2,092, and 1,791 mg/kg, respectively) at T6 stored at 37°C. Concentrations of pentanal and heptanal were also significantly ( $P < 0.05$ ) higher in the GRS and CLV WMP compared with the TMR WMP, possibly due to the greater amount of linoleic acid present in the GRS and CLV WMP samples. The odor threshold (mg/kg) in an oil matrix for hexanal, pentanal, and heptanal has been reported as 320, 240, and 3,200 mg/kg, respectively (Decker et al., 2010). Thus, hexanal and pentanal were present at >4 times their odor threshold in all WMP stored at 37°C at T6, which may likely explain the increase in painty attributes. Heptanal increased above its odor threshold in CLV WMP stored at 37°C at T4 and T6 (4,325 and 4,409, respectively). Levels of heptanal were also more abundant in GRS WMP stored at 37°C at T4 and T6 (2,937 and 2,994, respectively) but likely below its odor threshold.

Panelists identified a dominant buttery/toffee note in all WMP samples, which decreased over time, as

a painty aroma and flavor became more dominant (Figure 2A–F). This trend was particularly evident in TMR WMP, where the increase in the painty aroma and flavor was strongly correlated with increases in the concentrations of hexanal (Figure 2E and 2F). Hexanal, heptanal, and octanal have previously been found to be good predictors of painty and grassy flavors in WMP (Lloyd et al., 2009a). Hexanal reached similar concentrations across all 3 WMP, but the associated painty flavor was more readily identifiable in TMR WMP at T6, as the buttery/toffee notes declined. Thus, it appears that other volatile compounds must also contribute to painty flavor or possibly are masking painty flavor. Maillard and caramelization reactions have been shown to produce caramel-like and toffee-like flavors in dairy products (Patton, 1955), likely enhanced by the spray drying process. The strong association of the buttery/toffee attribute with the TMR WMP may be the result of more Maillard or oxidative browning, which was subsequently masked by the production of LO products over storage.



**Figure 1.** (A) Principal component analysis (PCA) biplot, representing the correlation structure of the variables (sensory attributes identified by full descriptive analysis and volatile compounds identified by headspace solid-phase microextraction gas chromatography mass spectrometry) and the relationship between the whole milk powder samples derived from grass (GRS) from times (T) 0–6 mo. (B) PCA biplot analysis of the variables and the milk powder samples derived from perennial ryegrass and white clover (CLV) from T0–T6. (C) PCA biplot analysis of the variables and the milk powder samples derived from TMR from T0–T6. Color gradient: low = white, mid = blue, high = red; midpoint set at 1.0. Flav = flavor; AE = aftereffect. Dim = dimension.

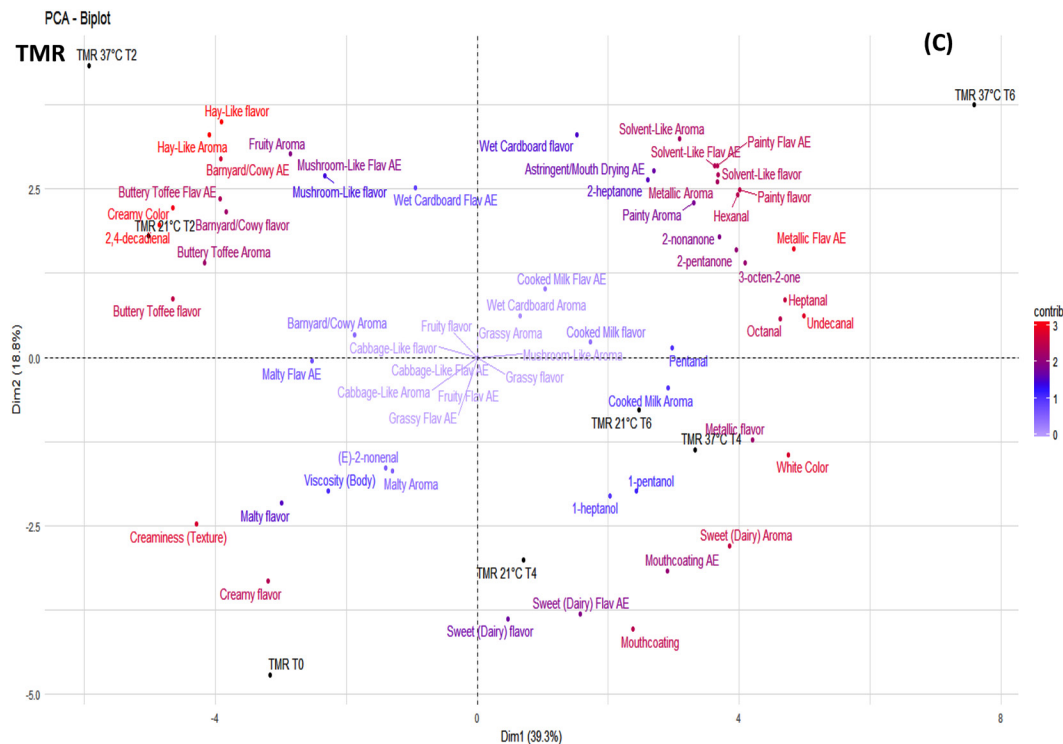


Pearson correlation relationships indicated that the sensory attributes hay-like aroma and flavor, grassy aroma and flavor, and barnyard/cow aroma and flavor can be grouped together, and therefore an increase in one attribute indicates that all increase. However, this could also indicate that all 3 attributes may have had similar identification criteria by the sensory panel. Additionally, buttery/toffee aroma and flavor were consistently correlated with dairy sweet attributes. Another group of attributes that were correlated were painty, solvent-like, and metallic aromas, flavors, and aftereffects. Specific individual volatile aldehydes, ketones, and alcohols are good indicators of how their overall chemical classes behave during the LO process. Therefore, an increase of one compound within a chemical class can indicate that others within that class derived from the same process, such as LO, will also increase. The increased concentrations of the alcohol compounds 1-heptanol and 1-pentanol observed in GRS and CLV WMP were very likely directly associated with the levels of precursors heptanal and pentanal, respectively (Kilcawley et al., 2018).

### Volatile Analysis

The highest concentrations of the aldehyde compounds hexanal, pentanal, and heptanal were observed in GRS and CLV WMP stored at 37°C at T4 and T6 (>1,500 mg/kg; Figure 3B and 3D). These compounds were consistently higher in GRS and CLV WMP compared with TMR WMP. Aldehydes ( $\geq 8$  carbons) and branched-chain aldehydes were best correlated with CLV WMP. The ketone compounds 2-nonanone, 2-heptanone, 2-pentanone, and 3-octen-2-one were consistently higher in GRS and TMR WMP samples stored at 37°C throughout the study, whereas the alcohol compounds 1-heptanol and 1-pentanol were highest in GRS and CLV WMP stored at 37°C. One-way ANOVA analysis with post hoc Tukey's test showed significant ( $P < 0.001$ ) increases in the concentrations of hexanal, heptanal, 2-nonanone, 2-heptanone, 2-pentanone, 3-octen-2-one, and 1-pentanol in all the WMP from T0 to T6.

Increases in the concentrations of hexanal, pentanal, heptanal, octanal, 2,4-decadienal, undecanal, and

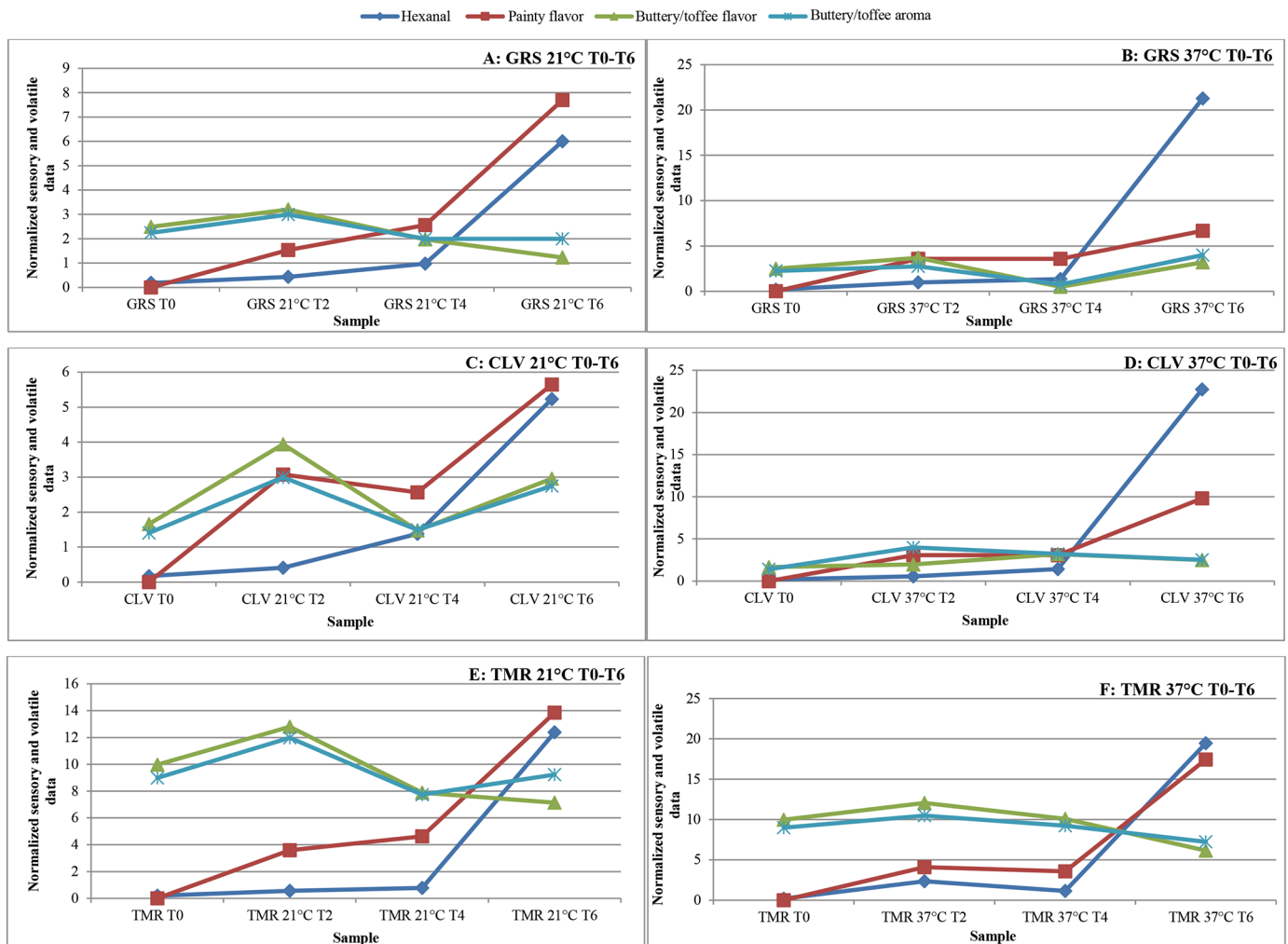


**Figure 1 (Continued).** (A) Principal component analysis (PCA) biplot, representing the correlation structure of the variables (sensory attributes identified by full descriptive analysis and volatile compounds identified by headspace solid-phase microextraction gas chromatography mass spectrometry) and the relationship between the whole milk powder samples derived from grass (GRS) from times (T) 0–6 mo. (B) PCA biplot analysis of the variables and the milk powder samples derived from perennial ryegrass and white clover (CLV) from T0–T6. (C) PCA biplot analysis of the variables and the milk powder samples derived from TMR from T0–T6. Color gradient: low = white, mid = blue, high = red; midpoint set at 1.0. Flav = flavor; AE = aftereffect. Dim = dimension.

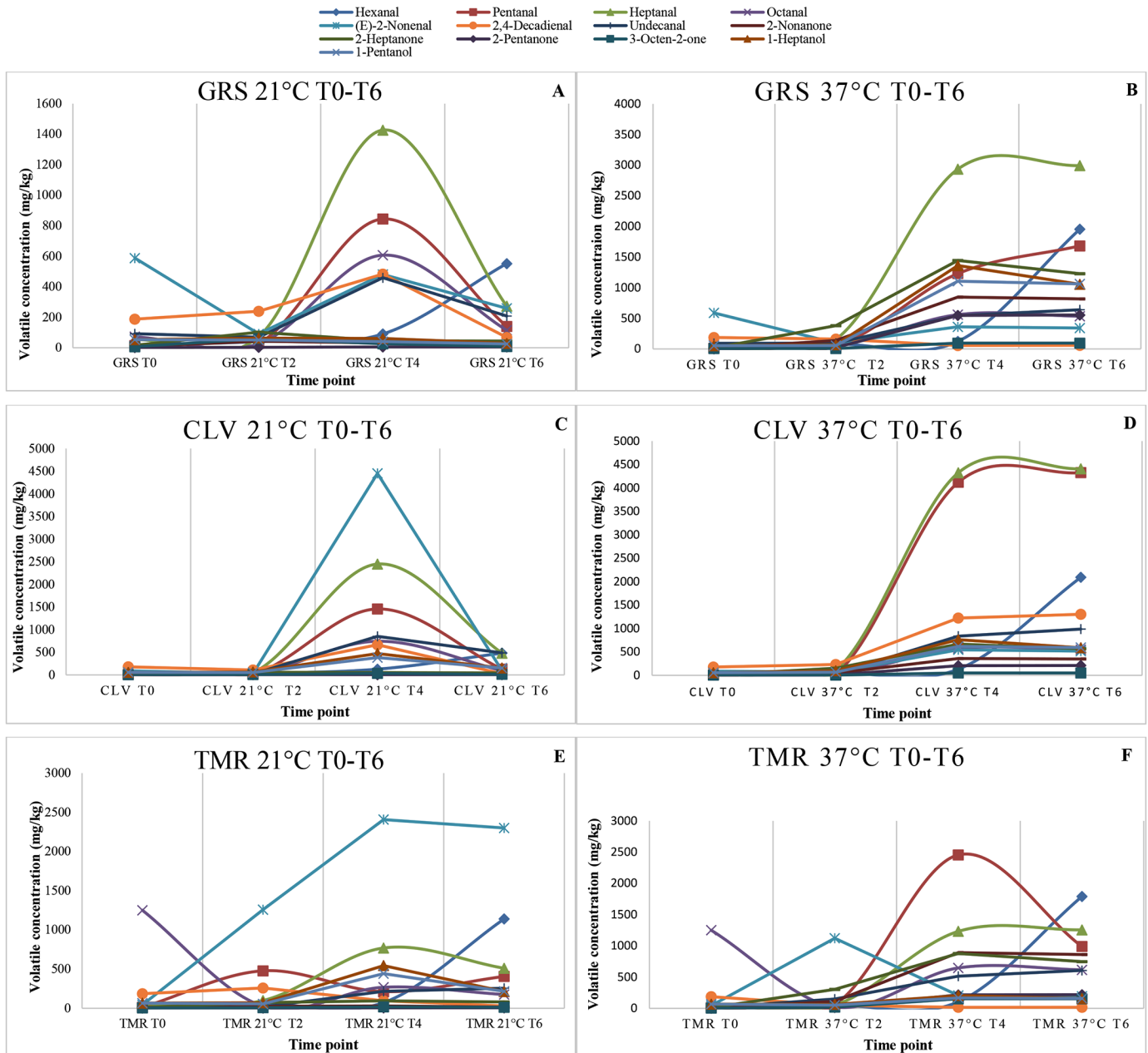
2-pentanone were observed in the GRS WMP stored at 21°C from T0 to T4. After T4, hexanal was the only compound that continued to increase; all the other compounds either remained at the same concentration or decreased slightly (Figure 3A and Supplemental Table S4A, <http://hdl.handle.net/11019/2424>), a trend that has been observed previously in WMP after 10 mo of storage (Lloyd et al., 2009b) and in whey protein concentrate after 4 mo of storage (Javidipour and Qian, 2008). The decrease in certain volatile compounds after they have peaked has previously been attributed to their degradation or catabolism (Neilson et al., 2006; Wright et al., 2009). Other studies have speculated that proteins, flavonoids, and some enzymes such as superoxide dismutase can inhibit LO by antioxidant activity (Eriksson 1982). The same study stated that LO inhibitors induced by thermal processing, such as Maillard

reaction products and native protein hydrolysates, may also be factors in inhibiting the LO mechanism. In addition to flavonoids,  $\beta$ -carotene has the potential to protect dairy products against oxidation (Havemose et al., 2006). As previously mentioned,  $\beta$ -carotene is known to be higher in dairy products produced from pasture and is responsible for their greater yellow color compared with those derived from TMR systems. Strecker-type degradation of amino acids has also been shown to be produced by some LO products (Hidalgo and Zamora, 2004), which may indicate the consumption of some LO end products in these reactions. A decrease in water activity within WMP during storage may also inhibit LO reaction rates.

In GRS WMP stored at 37°C, 4 aldehydes, namely hexanal, pentanal, heptanal, and undecanal, continued to increase from T4 to T6. The other 8 volatile



**Figure 2.** Graphs illustrating the increase in painty flavor and hexanal and the decrease in buttery/toffee flavor and aroma in whole milk powders derived from (A) grass (GRS) stored at 21°C and (B) GRS 37°C; (C) clover (CLV) 21°C and (D) CLV 37°C; and (E) TMR 21°C and (F) TMR 37°C during 6 mo of storage (T0–T6). Normalized data were used for the purpose of these graphs.



**Figure 3.** Graphs illustrating the concentrations (mg/kg) of the 13 selected volatile compounds in (A) grass (GRS) whole milk powder (WMP) stored at 21°C during 6 mo of storage (T0–T6); (B) clover (CLV) WMP 21°C; (C) TMR WMP 21°C; (D) GRS WMP 37°C; (E) CLV WMP 37°C; and (F) TMR WMP 37°C.

compounds reached their highest concentrations at T4 and decreased by T6, except for 2,4-decadienal and (E)-2-nonenal, which were highest at T0 (Figure 3B and Supplemental Table S4B, <http://hdl.handle.net/11019/2424>).

In CLV WMP stored at 21°C, similar trends to those of the GRS WMP stored at 21°C were observed. Concentrations of hexanal, pentanal, heptanal, octanal,

(E)-2-nonenal, 2,4-decadienal, undecanal, 2-nonanone, 3-octen-2-one, 1-heptanol, and 1-pentanol increased from T0 to T4 and decreased thereafter, apart from hexanal, which continued to increase to T6 (Figure 3C and Supplemental Table S4C, <http://hdl.handle.net/11019/2424>). For CLV WMP stored at 37°C, the aldehydes hexanal, pentanal, and heptanal reached much higher concentrations than in CLV WMP stored at

21°C. Similar to GRS WMP stored at 37°C, hexanal, pentanal, heptanal, and 2,4-decadienal continued to increase after T4 in the CLV WMP stored at 37°C. The other 9 volatile compounds remained the same or decreased by T6 [octanal, (E)-2-nonenal, undecanal, 2-nonanone, 2-heptanone, 2-pentanone, 3-octen-2-one, 1-heptanol, and 1-pentanol; Figure 3D and Supplemental Table S4D, <http://hdl.handle.net/11019/2424>].

In TMR WMP, (E)-2-nonenal was higher in samples stored at 21°C than 37°C; it increased gradually from T0 to T4 and then decreased by T6. In TMR WMP stored at 21°C (E)-2-nonenal was the most abundant compound at a concentration of 2,407 mg/kg at T4. In the 37°C samples it reached its highest concentration at T2 (1,122 mg/kg) but decreased thereafter. For TMR WMP stored at 21°C, hexanal, undecanal, and 2-pentanone increased gradually from T0 to T6; the remaining 10 compounds either decreased [heptanal, (E)-2-nonenal, 2-nonanone, 2-heptanone, 3-octen-2-one, 1-heptanol, and 1-pentanol] from T4 to T6 or fluctuated throughout storage (pentanal, octanal, and 2,4-decadienal; Figure 3E and Supplemental Table S4E, <http://hdl.handle.net/11019/2424>). In TMR WMP stored at 37°C hexanal, heptanal, undecanal, and 2-pentanone increased from T0 to T6. The remaining 8 volatile compounds either decreased (pentanal, 2-heptanone, 1-heptanol, and 1-pentanol) from T4 to T6, fluctuated throughout storage (octanal), plateaued (3-octen-2-one), or reached their highest concentration early in storage and began to decrease thereafter [2,4-decadienal and (E)-2-nonenal; Figure 3F and Supplemental Table S4F, <http://hdl.handle.net/11019/2424>]. As previously mentioned, compounds that begin to decrease in storage are likely degraded or catabolized, in many cases to other volatile compounds. Thus, for most volatile LO compounds in this study, the rate of formation was exceeded by degradation after 4-mo storage, presumably due to a lack of available FA substrate or other factors inhibiting the primary LO process. Maximum levels of some compounds were reached more rapidly in the WMP stored at 37°C due to the increased temperature.

### Correlations Between Volatile Components and Sensory Attributes

Pearson correlation analysis was carried out on the sensory and volatile data using SPSS. Numerous positive correlations were evident between volatile components and sensory attributes, but only strong correlations  $\geq 0.7$  are reported here. Hay-like aroma was significantly ( $P < 0.001$ ) correlated with barnyard/cow aroma (0.794), grassy aroma (0.734), cooked milk flavor (0.777), grassy flavor (0.956), hay-like flavor

(0.930), and barnyard/cow flavor (0.792). Buttery/toffee aroma was correlated with dairy sweet aroma (0.809), dairy sweet flavor (0.969), and buttery/toffee flavor (0.969). Painty aroma was correlated with painty flavor (0.791), painty flavor aftereffect (0.812), and metallic flavor aftereffect (0.754). Metallic aroma was correlated with increases in solvent-like flavor (0.828). Creamy flavor was correlated with viscosity (0.897), creaminess (0.870), and astringency (−0.776). Painty flavor was correlated with solvent-like flavor (0.716) and increasing concentrations of hexanal (0.793). Painty flavor aftereffect was also correlated with hexanal (0.749). Increases in concentrations of pentanal were correlated with increases in heptanal (0.911), 2,4-decadienal (0.750), and undecanal (0.843). Increases in heptanal were also correlated with increases in undecanal (0.784), 1-heptanol (0.784), and 1-pentanol (0.794). 2-Nonanone was significantly correlated with increases in levels of 2-heptanone (0.924), 2-pentanone (0.874), and 3-octen-2-one (0.923). As observed with heptanal, 2-heptanone and 2-pentanone were correlated with increases in the alcohol compounds 1-heptanol (0.750 and 0.858, respectively) and 1-pentanol (0.756 and 0.880, respectively). 2-Heptanone was also correlated with 3-octen-2-one (0.743), and 1-heptanol was correlated with 1-pentanol (0.991).

The key factor influencing the perception of LO compounds is their odor thresholds (the lowest concentration of a compound perceivable by the human nose), which vary considerably and also depend upon the matrix effect (the binding of compounds to components of the sample affecting their release). The highest concentrations of the aldehydes hexanal, pentanal, heptanal were observed in GRS and CLV WMP stored at 37°C at T4 and T6 ( $>1,500$  mg/kg; Figure 2B and 2D). These compounds were consistently higher in GRS and CLV WMP compared with TMR WMP. Conversely, painty attributes, which are commonly associated with increases in hexanal were correlated more with TMR WMP toward the latter stages of storage. Although the results of this study show correlations between painty attributes and concentrations of hexanal, it is unlikely that one single compound is responsible for specific attributes when numerous other odor-active compounds are present (Kobayashi and Nishimura, 2014). However, increases in the volatile compounds evaluated in this study are good indicators of LO state, but perhaps other approaches such as olfactometry are required to absolutely associate the concentration of specific volatiles with specific aroma descriptors. Overall, aldehydes ( $\geq 8$  carbons) and branched-chain aldehydes were most correlated with CLV WMP. The ketone compounds 2-nonanone, 2-heptanone, 2-pentanone, and 3-octen-

2-one were consistently higher in GRS and TMR WMP samples stored at 37°C throughout the study, whereas the alcohol compounds 1-heptanol and 1-pentanol were highest in GRS and CLV stored at 37°C. Concentrations of the FA (C4:0–C12:0) were higher overall in GRS and CLV WMP compared with TMR WMP (Table 1), which may have influenced LO susceptibility, particularly in relation to aldehyde formation. The concentrations of oleic acid were not significantly different between the WMP samples, but it is thought to be a major precursor for hexanal formation. Linoleic acid was significantly ( $P < 0.05$ ) higher in TMR WMP and is also a precursor of hexanal, in addition to pentanal and 3-octen-2-one. Linoleic acid likely influenced the higher levels of 3-octen-2-one in TMR WMP, but because both hexanal and pentanal were higher in pasture-derived WMP, other sources or factors must be influencing their formation, such as the presence of natural pro- and antioxidants.  $\alpha$ -Linolenic acid (C18:3 n3) has been shown to produce hexanal by Tawfik et al. (2017) and was significantly higher ( $P < 0.05$ ) in pasture-derived WMP (CLV > GRS). Eicosanoic acid (C20:0) and heneicosanoic acid (C21:0),  $\alpha$ -linolenic acid (CLA; c10t12), and CLA (c9t11) were also significantly ( $P < 0.05$ ) different based on diet and likely also affected aldehyde formation. In addition, CLA has been shown to oxidize more rapidly than linoleic acid, supporting evidence that the conjugated double bond is more susceptible to oxidation than a nonconjugated double bond, thus facilitating volatile compound formation and release (Moon et al., 2008).

## CONCLUSIONS

The main finding from this study was that the bovine feeding system pasture (GRS and CLV) versus nonpasture (TMR) significantly affected the TFA, FFA, volatile profile, and sensory attributes of WMP. Pasture-derived WMP (GRS and CLV) were best correlated with creamy color, dairy sweet aroma, and hay-like attributes, whereas nonpasture-derived WMP (TMR) was best correlated with white color and buttery/toffee and painty attributes. Buttery/toffee attributes were found to be more closely correlated with TMR WMP. Increases in many of the volatiles studied were evident during storage at both 21°C and 37°C, with some compounds peaking at T4 and then plateauing or decreasing slightly by T6, likely due to degradation exceeding formation. Pasture-derived WMP (GRS and CLV) were found to be more susceptible to LO from a volatile perspective, particularly in relation to aldehyde development, possibly due to increased concentrations of CLA and  $\alpha$ -linolenic acid. Pleasant attributes, pos-

sibly associated with Maillard reaction products were perceivable in the WMP at the beginning of the study but became masked by LO compounds with off-flavors by T4. Correlations were made between concentrations of hexanal and painty attributes, but it is unlikely that a single compound was responsible for these attributes. Regardless of this, however, the recommended shelf life for WMP once opened was <4 mo from a sensory perspective.

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