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COMMUNICATION

Synergistically activated dual-locked fluorescent probes to monitor H₂S-induced DNA damage

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Naphthalimide-based fluorescent probes (NANO-N₃ and NAN6-N₃) were developed with dual locked fluorescence. Significantly, the use of different linkers enabled the probes to adopt different conformations, resulting in changes in the energy dissipation of the excited states, respectively. Where, $\geq 1.9 \times 10^{-2}$ mM of H₂S and $\geq 2.2 \times 10^{-2}$ $\mu\text{g}/\text{mL}$ of DNA could unlock a highly sensitive off-on fluorescence response through synergistic changes of the molecular structure and conformation. As such the probes could monitor DNA damage induced by the overexpression of H₂S, and were able to evaluate the degree apoptosis of living cells mediated by H₂S-induced mtDNA or nDNA damage.

As a typical endogenous gaseous transmitter, H₂S can regulate the redox equilibrium during oxidative stress.¹ In addition, H₂S participates in DNA damage and repair, and it has a vital role in the DNA damage response (DDR).² When H₂S is overexpressed in cells, it affects the cellular antioxidant system, where autooxidation and trace metal-mediated Fenton reaction results in abundant reactive oxygen species (ROS),^{3,4} which leads to intracellular redox imbalance and DNA damage. When compared with nuclear DNA (nDNA), mitochondrial DNA (mtDNA) is more susceptible to H₂S, resulting in irreversible and severe damage,⁵ because it is located in ROS-producing regions and lacks histone protection.^{6,7} H₂S-induced DNA damage results in mutagenic or carcinogenic effects, and seriously accelerates the progression of malignant diseases.^{5,6} For example, promoting the breaking of DNA strands and the development of cancer.⁷ Therefore, it is essential to devise a suitable tool for monitoring H₂S-induced DNA damage, which is not only beneficial to the study of DNA repair and damage pathways induced by H₂S, but also of great significance for the diagnosis and treatment of DNA-related diseases.

Fluorescent probes are typically composed of a fluorophore, linker, and recognition group. They have been used as effective tools for the visual monitoring of H₂S⁸⁻¹¹ and DNA¹²⁻¹⁵ in living cells. For example, an off-on probe with symmetric structure and dual

recognition sites has been developed for H₂S that exhibits a low detection limit (2.0×10^{-5} mM) by using -N₃ as the recognition group in a bis-naphthalimide.⁸ While a 7-nitro-1,2,3-benzoxadiazolid has been used as the recognition group to construct a H₂S-specific fluorescent probe, which could detect H₂S down to 2.5×10^{-3} mM in cells.¹⁰ In addition, we have used thread like probes based on dicyanoisoflurone as the fluorophore to specifically bind mtDNA at 7.1×10^{-2} $\mu\text{g}/\text{mL}$ generating a highly sensitive fluorescence response due to the inhibition of intramolecular torsion.¹² These previous systems exhibit excellent detection and monitoring ability based on different kinds of fluorescence signal changes. To date, most research has focused on structural modification of the fluorophore and recognition groups and has ignored the importance of the linker. However, a series of G4-DNA specific fluorescent probes were synthesized containing linkers with varying lengths (- (CH₂)₃-, - (CH₂)₂O(CH₂)₂-, -CH₂(CH₂OCH₂)₂CH₂-). Under the synergistic action of the fluorophore and the linker, the probes exist in a folded form, which quenches the fluorescence. After binding with G4-DNA, the conformation of the probes were altered and the fluorescence signal was turned on.¹³ Based on the intramolecular dimerization-caused quenching (DCQ) strategy, a series of DCQ probes were constructed using different linkers (lysine, PEG8) and two squaraine units, the probes are quenched in aqueous media. Then when bound to the receptor, molecular conformational changes result in fluorescence turn on.¹⁶ In addition, the triphenylamine and imidazole recognition unit were linked by a short linker (ie. single bond), causing intramolecular torsion, resulting in fluorescence quenching. Thus, when bound to mtDNA G4, the intramolecular torsion was restricted, and the fluorescence was turned on.¹⁷ Inspired by these systems we developed a fluorophore, linker and recognition group system, where the fluorescence is quenched by dual-locking, and can only be turned on by the dual-activation by H₂S reaction and DNA binding. As such, our system enables the detection of H₂S-induced DNA damage.

The fluorophore and DNA specific binding unit was a bis-naphthalimide.¹⁸ Two diamines with different lengths (0-carbon, 6-carbon) were introduced into the bis-naphthalimide structure as linkers. Thus, when the probes bind with DNA, the molecular conformations are changed which removed the fluorescence quenching caused by intramolecular torsion (**NANO-N₃**) or DCQ (**NAN6-N₃**). This represents a lock, which is specifically opened by DNA binding. Moreover, a fluorescence quenching group -N₃, as a specific recognition group for H₂S, was introduced into both ends of the bis-naphthalimide structure. This further reduces the energy of radiative transitions, and acts as a lock for the fluorescence of the probe, which can be selectively opened using H₂S. The design of these two locks acts as double insurance for quenching of the fluorescence and results in a highly sensitive off-on fluorescence

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response when both locks are opened using the two keys H₂S and DNA, facilitating the detection of H₂S-induced DNA damage. The specific synthetic route (Fig. S1) and structural characterization of the two probes (NANO-N₃, NAN6-N₃) are given in the Supporting Information.

Firstly, the recognition ability of NANO-N₃ and NAN6-N₃ (6.7 μM) for H₂S or/and DNA were verified. NANO-N₃ and NAN6-N₃ generated spectral changes, controlled by changes in the conformation and structure. As shown in Fig. 1 and S2, NANO-N₃ and NAN6-N₃ were non-fluorescent in the free state. While ND-N₃ (2-carbon) was used as a model probe whose initial fluorescence was negligible (Fig. S3).⁸ The different linkers, result in different conformations in the free state. NANO-N₃ (0-carbon) with shorter linker exists in the torsional form (NANO-N₃(T), Fig. 2), while NAN6-N₃ (6-carbon) exists in the folded form (NAN6-N₃(F), Fig. 2), which greatly reduces the energy of radiative transitions, *i.e.* fluorescence quenching (first lock).^{16,17,19} When H₂S was added, the absorption of NANO-N₃ at 375 nm decreased, and a new absorption signal appears at 495 nm which gradually increases with increasing H₂S (Fig. S2a). The same result was obtained for the absorption spectra of NAN6-N₃ (Fig. S2b). In addition, the new peaks (m/z[M+Na]: 445.1064) and (m/z[M+Na]: 515.3523) appeared in the HRMS spectra with the addition of H₂S, respectively. It confirms that H₂S causes a change of structure from NANO-N₃ (m/z[M+3H]: 477.1438) or NAN6-N₃ (m/z[M+Na+2H]: 569.3312) to NANO-NH₂ or NAN6-NH₂ (Fig. S17). However, the fluorescence intensities of these molecules (NANO-NH₂ and NAN6-NH₂) were only slightly larger than those of NANO-N₃ and NAN6-N₃ (1.1 fold Fig. S2c and S2d), indicating that they are still quenched. This is because NANO-NH₂ and NAN6-NH₂ are still in a torsional conformation (NANO-NH₂(T), Fig. S4) or folded conformation (NAN6-NH₂(F), Fig. S4), even though their molecular structures have changed (Fig. S4). Yet, for the same process, the fluorescence of ND-N₃ was significantly enhanced with increasing H₂S. Significantly, there was no observed fluorescence enhancement for ND-N₃ with DNA (Fig. S3).⁸ This is because there is no conformational change for ND-N₃ only a structural change after reaction with H₂S.

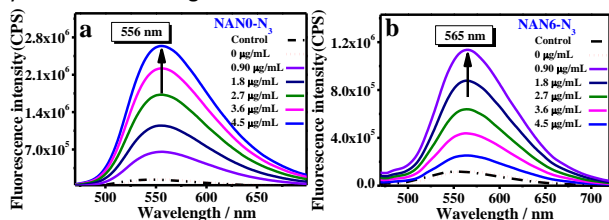


Fig. 1. The response of NANO-N₃ and NAN6-N₃ (6.7 μM) towards DNA and H₂S in tris-HCl (0.05 M, pH = 7.4, 37 °C). a and b, The emission spectra of NANO-N₃ (a, λ_{ex} = 425 nm) and NAN6-N₃ (b, λ_{ex} = 410 nm) towards DNA (0–4.5 μg/mL) in the presence of Na₂S (3.3 mM). Black dotted line (Control): probe (6.7 μM); Red dotted line (0 μg/mL): probe (6.7 μM) and 3.3 mM Na₂S. Na₂S solution (0.1 M) was used as H₂S releasing agent. Commercial smDNA was used for this experiment.

Furthermore, the absorption spectra of NANO-N₃ (Fig. S2e) and NAN6-N₃ (Fig. S2f) exhibit no change when DNA was added. That is, the addition of DNA has no effect on the molecular structure of NANO-N₃ and NAN6-N₃ (Fig. S5). The molecular docking results of NANO-N₃ and NAN6-N₃ with DNA (Fig. S5) indicated that NANO-N₃ (binding energy = -70.06 kcal/mol) and NAN6-N₃ (binding energy = -88.18 kcal/mol) are bound to DNA through minor groove interactions, resulting in conformation changes and transformation into NANO-N₃(I) and NAN6-N₃(I) (Fig. S5), respectively. As such, the torsion of NANO-N₃(I) was restricted and stacked NAN6-N₃(I) was unfolded (Fig. S5). These changes decreased the energy of the non-radiative transition, however, the sensitive fluorescence turn on was still inhibited by the -N₃ (*i.e.* the second lock) of NANO-N₃(I) and NAN6-

N₃(I). However, with an increase of DNA, the fluorescence of NANO-N₃ (Fig. S2g) and NAN6-N₃ (Fig. S2h) exhibited a weak enhancement. These results indicated that the probes could respond to H₂S or DNA, but only generated a weak fluorescence response due to a single key (recognition factor) being used with these dual locked systems.

NAN6-N₃ exhibited a good linear response for the targets (Fig. 1 and Fig. S6). And a highly sensitive fluorescence output could be activated with ≥ 2.7×10⁻² mM H₂S and ≥ 8.2×10⁻² μg/mL DNA coexisting in the test system. In addition, NANO-N₃ could be activated to generate a highly sensitive fluorescence signal in the presence ≥ 1.9×10⁻² mM H₂S and ≥ 2.2×10⁻² μg/mL DNA (Fig. 1 and S6). Significantly, the lowest concentrations of H₂S that could activate the probes were higher than the normal expression range for H₂S in cells, however higher levels associated with over expression could activate the probes.^{20,21} These spectral results are because probes (NANO-N₃(T) and NAN6-N₃(F)) underwent conformational changes (NANO-N₃(I) and NAN6-N₃(I)) after binding to DNA, which reduced the energy dissipation in the excited state. In addition, H₂S reduces the -N₃ to -NH₂ (NANO-NH₂(T) and NAN6-NH₂(F)), and the intramolecular electron cloud density increased. Therefore, under the synergistic action of conformational and structural changes (NANO-NH₂(I), NAN6-NH₂(I)), a highly sensitive fluorescence signal was generated (Fig. 2). That is, NANO-N₃ and NAN6-N₃ could be activated by H₂S and DNA in solution. And importantly, this response could be used for the detection of DNA damage induced through the overexpression of H₂S in cells.

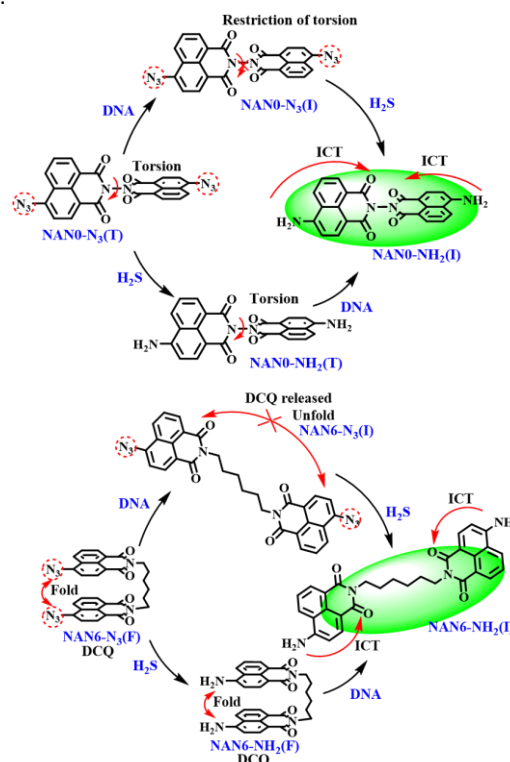


Fig. 2. The response mechanism of NANO-N₃ and NAN6-N₃ to H₂S and DNA.

Fig. S7–S10 verified that NANO-N₃ and NAN6-N₃ exhibited good biocompatibility, which was conducive to the specific monitoring of H₂S-induced DNA damage in cells. To verify the specific binding ability of NANO-N₃ and NAN6-N₃ for DNA in cells, HepG2 cells were pretreated with different concentrations of DNA digesting enzymes (0, 50.0 and 100.0 U/mL) for 1.0 h. These results (Fig. S11) indicated that all the probes (6.0 μM) exhibited fluorescence in the green channel (535–575 nm). In addition, as the concentration of DNA digestive enzymes increased, DNA damage was enhanced and the fluorescence intensity of NANO-N₃ and NAN6-N₃ decreased significantly. That

is, **NANO-N₃** and **NAN6-N₃** could only bind with intact DNA in cells, while DNA binding altered the molecular conformation, which reduced the fluorescence quenching caused by intramolecular torsion (**NANO-N₃**) or DCQ (**NAN6-N₃**), and resulted in the fluorescence being partially turned on. These results are consistent with those from the solution-based experiments (Fig. S2). In addition, when HepG2 cells were pretreated with 1.0 mM PAG (DL-Propargylglycine, an inhibitor of cystathionine γ -lyase during endogenous H₂S synthesis) for 1.0 h,¹⁰ their fluorescence intensity remained constant (Fig. S12). That is, normal levels of H₂S expressed in live cells cannot reduce -N₃ (one of the locks) to -NH₂, and the fluorescence remains in a semi-quenched state. In contrast, the fluorescence of **ND-N₃** did not change after the addition of DNA digestive enzyme, and decreased significantly with treatment by PAG, indicating that **ND-N₃** exhibits no response to DNA and its fluorescence signal was only influenced by H₂S.

In addition, the stained areas of living cells using these probes were significantly different. The commercial mitochondrial dye (Mito-Tracker Deep Red) and nuclear dye (Hoechst 33258) were used as standard localization dyes in HepG2 cells to investigate the localization, respectively. As shown in Fig. S13, the Pearson's correlation coefficient (Rr) for the staining region of **NAN6-N₃** (6.0 μ M), Hoechst 33258 (10.0 μ M) and Mito-Tracker Deep Red (5.0 μ M) were 0.37, 0.96, respectively, which indicated that **NAN6-N₃** could accumulate in the DNA of the mitochondria, *i.e.* mtDNA. Furthermore, the region stained by **NANO-N₃** overlapped with that for Hoechst 33258 (10.0 μ M, Rr = 0.91), suggesting that **NANO-N₃** could enter the nucleus and specifically detect nDNA. That is, the different length of the linkers could regulate the flexibility, biological activity and lipophilicity of the probes, thus realizing the accumulation in different subcellular regions of the cell. These imaging results confirmed that **NANO-N₃** and **NAN6-N₃** could achieve specific binding with nDNA and mtDNA, respectively.

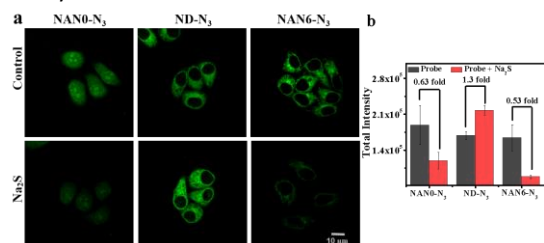


Fig. 3. Imaging of H₂S-induced DNA damage in HepG2 cells. a, The imaging of H₂S-induced DNA damage in HepG2 cells. HepG2 cells were pretreated with Na₂S to establish H₂S-induced DNA damage cell model. Stained with **NANO-N₃**, **ND-N₃** and **NAN6-N₃** (6.0 μ M), respectively. Excitation wavelength = 405 nm, scan range = 535–575 nm (green channel). Incubation time: 30.0 min. b, Histogram of the green channel in (a). Fold: intensity final / intensity at start.

Pretreatment of cells with exogenous Na₂S (H₂S-donor) can cause significant DNA damage in cells as shown by gel electrophoresis and comet assays, etc.^{2,4,7} Thus, HepG2 and CHO cell models of H₂S-induced DNA damage could be obtained. As shown in Fig. 3 and S14, the fluorescence intensity of **NANO-N₃** and **NAN6-N₃** in the green channel were significantly lower than that of the control group. This is due to excessive H₂S resulting in DNA damage, which results in release of the probes from the minor groove of DNA, and enables their return to a twisted or folded form (**NANO-NH₂(T)**, **NAN6-NH₂(F)**), and decreases the fluorescence intensity due to a reduction of the excited state energy. In addition, the fluorescence reduction for **NAN6-N₃** was greater than that of **NANO-N₃** (Fig. 3 and S14). This was because mtDNA is more vulnerable to damage by H₂S than nDNA^{5–7,12} In contrast, the fluorescence intensity of **ND-N₃** was enhanced, since it only interacts with H₂S. These imaging results

verified that **NANO-N₃** and **NAN6-N₃** were suitable for the specific and highly sensitive imaging of nDNA and mtDNA damage induced by H₂S, respectively.

mtDNA can be used as a sensitive marker to distinguish healthy cells from apoptotic cells.¹² Thus, **NAN6-N₃** was anticipated to be a powerful tool for evaluating the degree of apoptosis. Apoptosis of HepG2 cells was induced at different temperatures, and Propidium Iodide (PI) was used as a standardized indicator for the degree of apoptosis. As shown in Fig. 4a–c, the green channel (535–575 nm) exhibited a strong fluorescence, while the red channel (600–630 nm) displayed no fluorescence in the control group (*i.e.* healthy cells). With the development of apoptosis, the fluorescence of the green channel decreased significantly, and that of red channel increased. In addition, the same results were obtained in CHO cells (Fig. S15). Confirming that, the fluorescence intensity of the green channel decreased significantly with an increase of apoptosis. This is because the damage of mtDNA increases with an increase in the degree of apoptosis. When mtDNA damage increases the amount of binding by **NAN6-N₃** to the minor groove in the non-folded form decreases, and its fluorescence signal intensity decreases. Therefore, **NAN6-N₃** can be used as a highly sensitive tool to help evaluate the degree of cell apoptosis based on the highly sensitive response to H₂S-induced mtDNA damage.

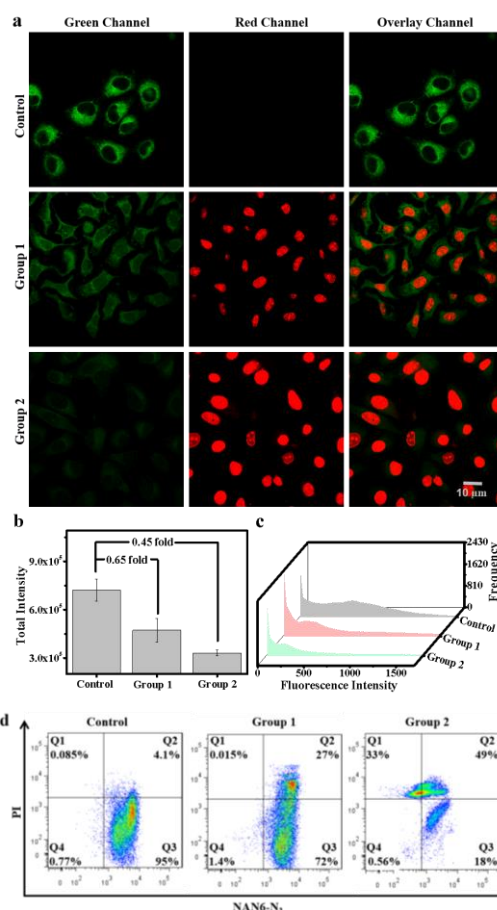


Fig. 4. HepG2 cells staining with **NAN6-N₃** (6.0 μ M) and PI (10.0 μ M) during cell apoptosis. a, Cell images during cell apoptosis. b, Histogram of the green channel in (a). c, Data for the green channel in (a). d, FCM analyses during cell apoptosis. Control group: untreated cells, healthy cell model. Experimental group 1: the cells were incubated at 56 °C for 4.0 h, early apoptotic cell model. Experimental group 2: the cells were incubated at 65 °C for 4.0 h, late apoptotic cell model. Q1: death. Q2: late apoptosis. Q3: early apoptosis. Q4: healthy cells. **NAN6-N₃**: excitation wavelength = 405 nm, scan range = 535–575 nm (green channel); PI: excitation wavelength = 559

nm, scan range = 600–630 nm (red channel). Incubation time: 30.0 min. Fold: intensity final / intensity at start.

The above results were further verified by flow cytometric (FCM) analysis and high throughput data analysis (Fig. 4d). In the control group, 95% of cells in early apoptosis were illuminated by **NAN6-N₃** and exhibited a bright green fluorescence signal (Fig. S16). As the degree of apoptosis gradually increased, the number of cells that were illuminated by **NAN6-N₃** decreased significantly (experimental group 1, 72%; experimental group 2, 18%). In addition, the number of cells illuminated by both **NAN6-N₃** and PI increased significantly during apoptosis (control group, 4.1%; experimental group 1, 27%; experimental group 2, 49%), and the green fluorescence signal of **NAN6-N₃** was significantly reduced (Fig. S16). This result is consistent with the above cell imaging results. As such, **NAN6-N₃** could evaluate the degree of cell apoptosis through the specific and highly sensitive monitoring of H₂S-induced mtDNA damage in cells.

In summary, two dual-locked probes (**NANO-N₃** and **NAN6-N₃**) were developed for the specific monitoring of H₂S-induced DNA damage in living cells. Spectral experiments and theoretical calculations confirmed that **NANO-N₃** and **NAN6-N₃** could strongly bind to DNA through minor groove interactions and cause molecular conformational changes, which inhibited the non-radiative transition energy. Meanwhile, the molecular structure was changed by H₂S (*i.e.* –N₃ to –NH₂), which increased the density of the intramolecular electron cloud. In this work, $\geq 1.9 \times 10^{-2}$ mM of H₂S and $\geq 2.2 \times 10^{-2}$ μ g/mL of DNA as dual-keys could activate a highly sensitive fluorescence signal through synergistic changes of the structure and conformation. **NANO-N₃** and **NAN6-N₃** with different amphiphilicity could specifically monitor H₂S-induced nDNA and mtDNA damage. In addition, **NAN6-N₃** could in real-time evaluate apoptosis by monitoring H₂S-induced mtDNA damage. Therefore, **NANO-N₃** and **NAN6-N₃** could be used as favorable tools for investigating the pathway of DNA repair and damage caused by H₂S.

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Conflicts of interest

There are no conflicts to declare.

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