

Global Journal of Human-Social Science: B Geography, Geo-Sciences, Environmental Science & Disaster Management

Volume 22 Issue 4 Version 1.0 Year 2022

Type: Double Blind Peer Reviewed International Research Journal

Publisher: Global Journals

Online ISSN: 2249-460x & Print ISSN: 0975-587X

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GJHSS-B Classification: DDC Code: 338.479172981 LCC Code: G155.C35



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Lichen Species as Bio-Accumulator of Some Halogens on Mount Cameroon Volcano, West Africa

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Abstract- Lichens diversities are informative indicators for assessing impacts of air pollution, climate change, environmental health, volcanic activities, habitat heterogeneity and continuity. Lacking roots, vascular tissues, stomata and waxy cuticle, they absorb and accumulate airborne nutrients/pollutants from the atmosphere over their entire surface. Halogens, especially fluorides are released into the atmosphere in large amounts by volcanic eruptions and their pollutants levels in lichens can be determined quantitatively by chemical analysis of species. The objective of this study was to examine the lichens potentials on Mount Cameroon (MC) as bio-monitors and bio-accumulators for some halogens (F, Cl, Br) levels. To achieve this objective, 34 lichen species were analysed using selective ion electrode. The species were collected from eight sampling sites on two flanks of MC at elevations ranging from 3-2178 m above sea level. The significance of the difference between means was tested using ANOVA test to compare the concentration of Halogens according to Elevation, Post Hoc Multiple Comparisons Altitudes, Independent Samples t-Test to compare the concentration of Halogens according to substrate types and Box – plot to confirm the test.

Student t-test for the comparism of the halogens with regards to the substrate (Table 4) shows no significance difference in the means of the all the halogens but a slight difference (p= 0.048 at 0.05 levels) in the variance of Cl. The box-plot (fig 3) also confirms the slight difference in Cl. Leptogium gelatinosum was identified as the highest sequesters, followed by Heterodermia obculata and least was Parmotrem atinctorum. Fluorine was the most dominant halogen with concentrations ranging from 0-188 μ g/g with mean values of 78 ± 49 and bromine the least. The concentrations were higher for specimens located close to the coastal areas in the downwind direction compared to those found further inland.

Keywords: lichens, mount cameroon, *leptogium* gelatinosum, fluorine and halogens.

Introduction I.

olcanic regions have always attracted many people worldwide because of the high fertility of their soils (Diana et al., 2019). However, human proximity to volcanoes can lead to several health problems as consequence of the chronic exposure to

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the materials released from the volcanic activity. An element often found in elevated concentrations in volcanic regions is fluorine. Although fluoride is recognized to have a beneficial effect on the rate of occurrence of dental caries when ingested in small amounts, its excessive intake results in a widespread but preventable pathological disease called fluorosis (Dey and Giri, 2015) While skeletal fluorosis, the most severe form of fluorosis, requires a chronic exposure to high concentrations of fluoride in water (4-8 mg/L), dental fluorosis occurs after shorter periods of exposure to fluoride in lower concentrations (1.5-2.0 mg/L). In some volcanic regions, where exposure to elevated amounts of fluoride is persistent, biomonitoring programs are fundamental (Garcia and Borgnino, 2015).

In the present world pollution scenario, a comprehensive knowledge of pollutants and their adverse effects on the ecosystem are required for selection of a workable monitoring and conservation technique (Munzi et al., 2012; Ahmad et al., 2007). The increasing awareness of the potential hazards and impact of air pollution on the health of human populations, forest decline, climate change and loss of agricultural productivity, for example, has been a cause of increasing public concern throughout the world (Smodis, 2007). This has highlighted the need for continuous monitoring of the levels of pollutants in the environment (Garty et al., 2002). Environmental monitoring approaches that are cheap, can be used anywhere, and respond to many kinds of airborne pollutants are needed to fingerprint the pollutant sources and their dispersion pattern (Larsen et al., 2007). A comprehensive approach to reduce the impacts of pollution and climate change, an approach that decreases emissions across all sectors and enhances the adaptive capacity of all nations with economic reflections is needed (Pinhoa et al., 2012).

Lichens emerge as the key answer to this monitoring problem and are the flora of choice for monitoring studies (Notcutt and Davis, 1989). They obtain their nutrients directly from the atmosphere and their chemical composition therefore holds the promise of becoming a natural or 'green' technique for monitoring the health of the environment around passively degassing volcanoes and industries. Bioaccumulation in lichen thalli has been used as a major tool for assessing air quality in volcanic and industrial areas (Bennett, 2006). They are extremely valuable in environmental monitoring since they exist worldwide and

are sensitive to many different kinds of pollutants (Brodeková et al., 2006). They are slow growing; do not shed parts and are perennial pioneer plants commonly described as sentinel organisms (Loppi et al., 2002). They are good bio-accumulators of heavy metals and trace elements and can be transplanted where they do not occur in nature (Llop et al., 2012).

Lichens are mutual symbiosis between fungi with an algal and/or a cyanobacterial partner (Morris and Purvis, 2007). The success in lichenization is attributed to a genetic combination resulting from metabolic biomolecules and influenced by environmental factors (Jatinder et al., 2012). This process has created unique characteristics in lichens such as the unique anatomical (absence of roots, stomata, vascular tissues and cuticle) and physiological (poikilohydry and absorbance of nutrient from general thallus surface from the atmosphere. These peculiarities, allow lichens to grow in all sorts of terrestrial habitats comprising 8% of vegetation.

According to Lawrey (1986), lichens produce a wide array of more than 1000 unique secondary metabolites (depsides, depsidones, β -orcinoldibenzyl esters, and xanthones, usnic acid and pulvinic acid derivatives, for example) as adaptations for life in marginal habitats. These secondary metabolites assist to maintain the lichen symbiotic association and compete with organisms sharing the same niche (Culberson and Culberson, 2001). Another characteristic stress-resistance mechanism is the accumulation of melanin and oxalate crystals in their thallus, which provide a crystal layer on the thallial surface making lichens tolerant to extreme environments and good bioaccumulators of atmospheric substances (Hess et al., 2008). Most lichens are tolerant to high concentrations of atmospheric pollutants well beyond levels necessary for their physiological requirement by sequestrating and accumulating varied oxalate crystals (Garty et al., 2002; Bjerke et al., 2002; Chen et al., 2000). The aggregates of these oxalate crystals disintegrate and provide a crystal layer on the thallial surface making lichens good bioaccumulators. Since lichens do not shed parts (Walker et al., 2003; Monge-Nájera et al., 2002), and bioaccumulate pollutants safely in their thalli over time, pollutants levels can be professionally determined quantitatively by chemical analysis of species and qualitatively by observing species diversity, abundance and distribution (Jovans, 2008). With their indiscriminate ability to absorb and bio-accumulate both nutrients/ pollutants from the atmosphere, elevated concentrations of certain elements in lichens are a sure sign of atmospheric deposition (van Herk, 1999).

Lichens can be used as bio-monitors of pollutants by quantifying the amount of trace element(s) accumulated within them over time (Srivastava et al., 2015, Bargagli, 2016). They have been used to assess deposition and air quality in hundreds of studies worldwide (Donahue, 2018). Ayrault et al., 2007, have shown a relationship between the quantities of pollutants in the environment and those concentrated in lichen thallus. A variety of elements and chemical compounds affecting lichen growth and distribution are found in the atmosphere (Bajpai et al. 2011). Pollutants, including sulfur dioxide (SO₂), nitrogen dioxide (NO₂) and fluoride (F) compounds, remain in the same chemical form after they are emitted into the atmosphere and are easily absorbed by lichens. Gases like chlorine and fluorine, leads to the injury of fundamental metabolic processes, which arise by acidifying the water and the substrates, resulting in the loss of most sensitive lichen species (Brodeková et al., 2006). Many lichens are sensitive to fluorine pollutant as it can concentrate in hydrated lichens to more than 200 times ambient concentrations (Notcutt and Davis, 1989). Fluoride are highly toxic to lichens, and elevated levels of fluoride are correlated with chlorophyll breakdown, reduced ATP concentration, reduced photosynthesis and disappearance of species (Stefano and Luisa, 2006). In general, obvious damage to lichens begins at levels of 50-70 ppm.

In most parts of Europe like Germany for example, (Hauck, 2005), lichen transplants from pristine to polluted areas are carried out to bioatmospheric pollutants. accumulate The lichen Hypogimnia physodes was used to bio-accumulate radionuclide Uranium (Golubev et al., 2006) and rare earth elements in Czech Republic (Jitka et al., 2010). The fruticose lichen Stereocaulon vesuvianum, growing on the slopes of Mount Vesuvius in South Italy, was used as a bio-monitor of ¹³⁴Cs, ¹³⁷Cs, ¹⁰³Ru and ¹⁰⁶Ru derived from the April 26, 1986 Chernobyl nuclear reactor disaster (Environment Canada, 2003). Grasso et al. (1999) found that lichen composition reflects the contribution of the volcanic activity in Mount Etna and Vulcano Island. They noted that, distribution of the degassing elements (arsenic (As), antimony (Sb), Br, and lead.

Volcanoes emit a variety of gases both between and during eruptions, including H₂O, CO₂, SO₂, HCl, NH₃, H₂S, HF and a few other minor constituents (Cronin and Sharp, 2002). These gases interact rapidly with the ash particles of a volcanic plume and especially atmospheric water to form acidic aerosols. These aerosols given off during and after volcanic eruptions caused problems on a number of occasions when it has accumulated in low lying areas. Exposure to excessive amounts of fluoride may cause adverse health effects for humans and animals (Conti et al., 2016) The plume dispersed by winds after a volcanic eruption contains volcanic ash that may also be a source of fluoride at levels that are potentially toxic. Fluorides are released into the environment naturally through the weathering and dissolution of minerals, the emissions from volcanoes and from marine aerosols (WHO, 2002) fluoride emissions from volcanoes and the natural occurrence of excessive amounts of fluoride in drinking water have affected the health of humans and livestock for centuries, if not millennia. Although sometimes of anthropogenic origin, high levels of fluorine are generally related to natural sources. Volcanic emissions of fluorine take the form of either sluggish permanent release from quiescent volcanoes (passive degassing) or rarer but more impacting discharges during short-lived volcanic eruptions (Schwandner et al., 2004 Linhares et al., 2017). It has been estimated that passive degassing, like that existing at Mt. Etna (Italy) and Masaya (Nicaragua) volcanoes, accounts for about 90% of the volcanic fluorine release. The influence of these emissions on the surrounding environment and in particular on vegetation has been investigated by several authors (Nelson and Wheeler, 2016).

Little or no bio-accumulation and monitoring work have been carried out on the active Mt. Cameroon. Mount Cameroon volcano with a return period of 20 years (Suh et al., 2003), has been the most frequent erupted volcano in West Africa, with eight eruptions in the last 100 years (1909, 1922, 1925, 1954, 1959, 1982, 1999 and 2000). It constantly releases various constituents into the environment during active eruptions and even in quiescent degassing periods (Suh et al.,

2008). These researchers reported that, Mt. Cameroon basanites melt inclusions has shown high levels of carbon dioxide with a concentration of 967 μ g/g, sulphur 2400 μ g/g, chlorine 1270 μ g/g and fluorine 1530 μ g/g. In spite of these findings, there is little knowledge of lichens toxic levels and remediation of the high levels of halides release from this volcano.

The objective of this paper was to determine the concentration levels of some halogens and identify potential lichens species that can be used as appropriate bio-accumulators of halide toxicity for Mt. Cameroon degassing volcano.

Materials and Methods II.

a) Description of the Study Area

i. Location

The study area (Fig.1) is the active MC volcano located in the coastal belt of the Gulf of Guinea, South West Region of Cameroon. It lies between Latitudes 3°57' to 4°27'N and Longitudes 8°58' to 9°24'E (Suh et al., 2003). It is the highest peak in West Africa, is of volcanic origin and rises from the Atlantic Ocean to a height of 4100 m. It covers a surface area of about 1750 km² (DeLancey and Mark, 2000).

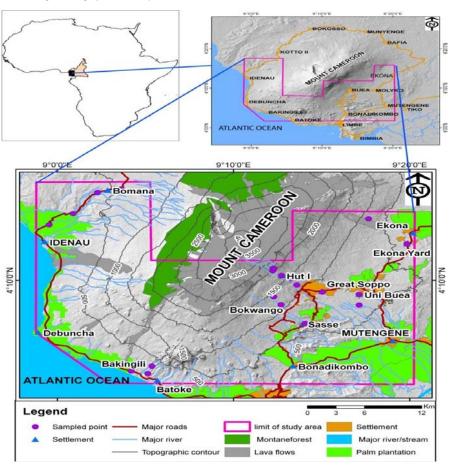


Fig. 1: Topographic features of Mt. Cameroon and location of sampling sites mentioned in the text.

b) Sample Sites

The survey sites were divided into Northern and Southern contrasting flanks following wind direction and ash fall trends from various eruptions. The northern flank was called the leeward and southern flank was called the windward. Out of the eight sites selected on the two flanks, four were on the leeward flank (Lower Buea, Upper Buea, IRAD-Ekona and Ekona-Mbenge) and four from the windward flank (Batoke, Bakingili, Idenau and Bomana). Lower Buea on the leeward flank comprises of University of Buea campus, Great Soppo, and Sasse. Some species were collected from control area of Mamfe (5.7512° N – 9.3146° E) about 270 km from Mt. Cameroon.

The survey was also done based on altitudinal levels which ranged between 3 to 2178 m above sea level (Table 1). The altitudinal levels were divided into three (low, mid and high) ranges.

Table	1:	Altitudinal	ranges	of	samp	oling	sites.

Altitude	Range(m)	Sites
Low	3 -499	IRAD-Ekona Batoke Bakingili Idenau
Mid	500-1000	Ekona-Mbenge Lower Buea Bomana
High	>1000	Upper Buea

Sample Selection

Thirty-four macro lichens (Foliose and Fruticose) species from six families, eight genera were collected from 8 sites around Mt. Cameroon. From each sites, different sampling points were surveyed given a total of about 12 sampling points in the study. These species were collected from trees and rocks (Table S2). All the collecting points were Georeferenced with a High Sensitive ErexGlobal Positioning System (GPS).

The samples were selected based on the criteria shown on Table 2. The selected species were common to most sites and represent various altitudinal levels on the two flanks of the edifice.

Table 2: Lichen species selected for halogens (fluorine (F), bromine (Br), and chlorine (Cl) analysis.

S/N	Criteria (species abundance at sites, elevation, flanks, morphology)	Species
1.	Foliose Species common to all sampling sites	Leptogium gelatinosum,
2.	Species of mid elevation	Heterodermia obscurata
		Heterodermia jabonica
3.	Species found on the same sampling (Leeward) site but differ in substrate	Parmotrema tinctorum
	(tree/rock)	
4.	Species restricted to the mid and high altitudes	Flavoparmelia caperata
5.	Site-specific species (These are species found only in particular sampling	Canoparmelia concrescens
	points and not seen in any other area)	Cladonia sp, Sticta stenroos
		Usnea dasypoga,
		Usnea florida, Usnea articulate

Sample preparation and analytical procedure

In the Life Sciences laboratory, University of Buea, the lichen species to be analysed for their halogens (F, Br and Cl) levels, were sorted and curated to remove adhering bark, mosses, other lichen species, soil particles, etc. Following Lorenzini et al., 2006, no washing procedure was done, to avoid the leaching of soluble matter from tissues. The species were put in labeled envelops and oven-dried to constant weight in a Gallenkemp Hot box oven fan size 3 at 60 °C for 48 hours. The different species were put in small zip locked bags and labeled with chemical codes. Samples were chemically analysed by selective ion electrode method, at the department of geology, university of Botswana. A 0.5 g split of each sample was digested with hydrogen fluoride (HF), then aqua regia and the aliquots analysed. The detection limits ranged from 0.01 to 0.02 μ g/g. Replicate analyses were performed on selected samples and data quality was excellent with standard deviation values less than 1%. Standards were run between samples and quality control of the analyses was ensured by inserting blanks into the analytical run after 6 samples. Prior to statistical analysis of the geochemical parameters, the data set was regrouped based on the lichen species. The entire data were then log-transformed to normalise skewed distributions. The significance of the difference between means was tested using ANOVA test to compare the concentration of Halogens according to Elevation, Post Hoc Multiple Comparisons Altitudes, Independent Samples t-Test to compare the concentration of Halogens according to substrate types and Box – plot to confirm the test.

RESULTS III.

ANOVA test on the variation in the concentration of halogens across the different elevation revealed that there was a significant difference (p = 0.022) for F and Br and p= 0.030 for Cl at 95% confidence level (Table 2). The box-plots (fig 2) also revealed that Cl concentration are high at lower elevations.

ANOVA test to compare the concentration of Halogens according to Elevation

	Sum of Squares	Df	Mean Square	F	Sig.
Between Groups	17554.069	2	8777.035	4.321	.022
Fluorine Within Groups	62966.666	31	2031.183		
Total	80520.735	33			
Between Groups	7630.711	2	3815.355	3.926	.030
Chlorine Within Groups	30129.760	31	971.928		
Total	37760.471	33			
Between Groups	2814.601	2	1407.300	4.322	.022
Bromine Within Groups	10093.429	31	325.594		
Total	12908.029	33			

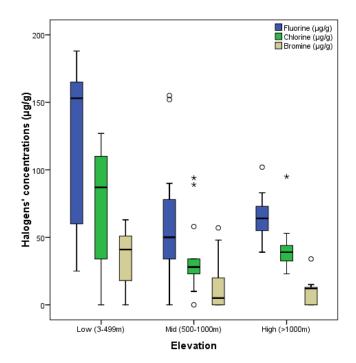


Fig. 3: Box – plot of Halogens concentrations in the different elevations

The Post Hoc Multiple Comparison test (Table 3) shows a significance difference at 0.05 levels of the halogens at low - mid altitude, F (p=0.030), Cl (p=0.027) and no significance difference of Br (p=0.053) at this altitude.

Table 3: Post Hoc Multiple Comparisons

Tukey HSD

			Mean Difference (I-J)	Std.		95% Confidence Interval		
	Dependent Va	riable		Error	Sig.	Lower Bound	Upper Bound	
Fluorine	Low (3-499m)	Mid (500-1000m)	51.849 [*]	19.255	.030	4.46	99.24	
		High (>1000m)	51.051*	20.257	.044	1.19	100.91	
	Mid (500-1000m)	Low (3-499m)	-51.849 [*]	19.255	.030	-99.24	-4.46	
		High (>1000m)	799	18.159	.999	-45.49	43.89	
	High (>1000m)	Low (3-499m)	-51.051 [*]	20.257	.044	-100.91	-1.19	
		Mid (500-1000m)	.799	18.159	.999	-43.89	45.49	
Chlorine	Low (3-499m)	Mid (500-1000m)	36.357 [*]	13.320	.027	3.57	69.14	
		High (>1000m)	29.364	14.012	.107	-5.12	63.85	
	Mid (500-1000m)	Low (3-499m)	-36.357*	13.320	.027	-69.14	-3.57	
		High (>1000m)	-6.994	12.561	.844	-37.91	23.92	
	High (>1000m)	Low (3-499m)	-29.364	14.012	.107	-63.85	5.12	
		Mid (500-1000m)	6.994	12.561	.844	-23.92	37.91	
Bromine	Low (3-499m)	Mid (500-1000m)	18.762	7.709	.053	21	37.74	
		High (>1000m)	22.333*	8.110	.026	2.37	42.29	
	Mid (500-1000m)	Low (3-499m)	-18.762	7.709	.053	-37.74	.21	
		High (>1000m)	3.571	7.270	.876	-14.32	21.46	
	High (>1000m)	Low (3-499m)	-22.333 [*]	8.110	.026	-42.29	-2.37	
		Mid (500-1000m)	-3.571	7.270	.876	-21.46	14.32	

^{*.} The mean difference is significant at the 0.05 level.

Student t-test for the comparism of the halogens with regards to the substrate (Table 4) shows no significance difference in the means of the all the halogens but a slight difference (p= 0.048 at 0.05 levels) in the variance of Cl. The box-plot (fig 3) also confirms the slight difference in Cl.

Table 4: Independent Samples t-Test to compare the concentration of Halogens according to substrate types

		Levene's Test for Equality of Variances		t-test for Equality of Means							
		F	Sig.	Т	Df	Sig. (2- tailed)	Mean Difference	Std. Error Difference	Interval	Confidence rval of the fference	
									Lower	Upper	
Fluorine	Equal variances assumed	3.568	.068	623	32	.538	-16.533	26.541	-70.595	37.528	
	Equal variances not assumed			435	3.324	.690	-16.533	38.030	-131.152	98.086	
Chlorine	Equal variances assumed	4.234	.048	-1.169	32	.251	-20.933	17.907	-57.408	15.541	
	Equal variances not assumed			760	3.266	.498	-20.933	27.552	-104.716	62.849	
Bromine	Equal variances assumed	2.837	.102	761	32	.452	-8.067	10.595	-29.648	13.515	
	Equal variances not assumed			541	3.340	.623	-8.067	14.914	-52.907	36.774	

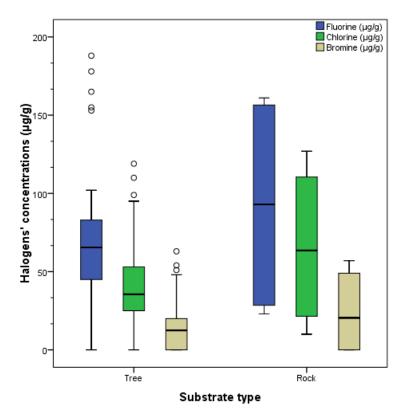


Fig. 3: Box – plot of Halogens concentrations in the substrate types

Table S2 revealed that, of the eight lichen genera used for analyses, Leptogium was the most abundant, widely distributed and had very high concentrations of halogens compared to all the other species. Leptogium gelatinosum accumulating ability differs between sites. Leptogium gelatinosum samples from northern leeward flank (MC 02 from Upper Buea, MC 04 from Bomana and MC 10 and MC 11 from Ekona-Mbenge) show low accumulation of halogens over samples collected from the western windward flank Sasse (MC01 and MC03), Idenau (MC 05 and MC06), Bakingili (MC07) and Batoke (MC08 and MC09).

Although, Leptogium gelatinosum recorded highest concentrations, other species like Heterodermia obscurata, Cladonia sp, Usnea articulate in BmtUM recorded high concentrations. Lichen species also showed differing accumulating potentials in substrates (Parmotrema tinctorum (MC27) a corticolous sample collected in Ekona-Mbenge accumulated more than the saxicolous specimen (MC26) collected in the same point).

All the eight lichen genera showed very high concentrations of F and low levels of Br. F concentration is highest in Leptogium (188 µg/g) and lowest in Parmotrema (0 μg/gto 23 μg/g). The geochemical analysis of the lichens indicated that F was the most dominant halogen (Fig. S1) with concentrations that range from 25-188 (mean = $78.00 \pm 48.68 \, \mu g/g$). The range for CI concentration was 10-127 μg/g (mean =47.23 \pm 33.37 μ g/g), while that for Br, ranges from 0-63 μ g/g (mean= 18 \pm 19.58).

Fig.S2 shows that, there was highest F- bioaccumulation in Leptogium gelatinosum species in Idenau which ranges from (153 to188 μg/g), Batoke (165 μg/g), Bakingili (178 μg/g), Sasse (152 to 155 μg/g) and low in Ekona-Mbenge (34 µg/g).

CL bio-accumulation in the Leptogium gelatinosum species in Idenau ranges from (110 to119 μg/g) was highest and lowest in the Leptogium gelatinosum of Bomana (23 µg/g), Ekona (33 µg/g) and Hut 1(28 μg/g) (Fig. S3).

Fig. S4, revealed no Br bio-accumulation in the Leptogium gelatinosum species in Ekona (0 μg/g) and Bomana (0 μ g/g) but higher in Idenau (63 μ g/g).

DISCUSSION

Lichens from Mt. Cameroon demonstrated significant compositional variation between species as observed on the multi-element distribution patterns even for those growing in the same area. However, specimens from the same species tend to have similar element concentration patterns. This could be explained by the fact that lichens species selectively accumulated some elements. Similarly, Rani et al. (2011) found out that, the estimated nine heavy metals in lichen samples from 12 different sites of Dehradun city by periodic

monitoring and spotted Zn, Ni, Cd and Cr were higher in lichens, collected from road side while maximum quantity of Fe, Cu and Al were reported in lichens collected from central sites of the city while the lowest amounts of all the metals were reported in sites farther from city.

The species Leptogium gelatinosum with very high concentrations of the halogens in all the sampling sites and even the control area of Mamfe 270 km from Mt. Cameroon has a higher tendency to sequestrate these elements than all the other species recorded in the study. This is in accordance with the study about the suitability of the fruticose lichens Evernia prunastri, Cetraria islandica and Ramalina farinacea collected from oak trees in a remote area located in the Chianti Region (Tuscany,central Italy), as transplants for biomonitoring trace element, showed that E. prunastri has to be preferred for its higher accumulating capacity (Cercasov et al., 2002). Different lichens species in different environments have different sequestration potentials, for example, Ramalina fastigiata has been used as a bio indicator of the impact of a coal mine in Portugal Jóźwiak, (2012).

The very high concentrations of F in lichens in this study reflects the study of Suh et al. (2008) who measured halogen content in melt inclusions at this volcano and their results indicated an average value of 1530 μg/g F and 1270 μg/g Cl. They recorded unusually high F concentrations when compared to glass inclusions from Etna, Hawaii and Piton de la Fournaise. However, the CI concentrations from Mt. Cameroon were midway between the high values measured for Etna and the low values for Hawaii and Piton de la Fournaise. These exceptionally high values relative to those recorded in this study which maybe an indication that these halogens are an important component of the volatile budget at Mt. Cameroon. These researchers reported that, the concentrations of F in olivine hosted glass inclusion from Mt. Cameroon are the highest known F concentrations for basaltic glass inclusions in the world.

Altitude contributed to the halogens concentration variations as intra-species variations consistently yield high concentrations in samples collected from the downwind SW flank of the volcano. These localities lie in the path of wind bearing volcanic gas plumes from Mt. Cameroon and therefore pin their higher halogen content to passive degassing. Aiuppa et al. (2004) reported that, during explosive activity huge quantities of fluorine are deposited with ashes around the volcano up to distances of hundreds of km. Fluorine is present as an adsorbed outer layer on the tephra particles which adsorption occurs by condensation of fluoride onto the tephra particles in the plume above the volcano as it cools. The smaller tephra particles have a larger surface area, so carry more absorbed fluoride than the larger particles". The smaller particles are likely to be carried further from the volcanic source, and so their greater fluorine-carrying capacity extends the zone of potential fluorine poisoning considerably, even to regions where only a 1 mm thick deposit forms. It is advisable to sample and analyse the tephra or vegetation to identify hazardous regions.

This suggests that gases emitted from the volcano are blown south-westwards and are eventually deposited close to the coast resulting in higher halogen content in the lichen species from these areas. More so, the inputs from the sea and from agricultural farms might have increased the high levels in this coastal areas. Studies by Ndlovu et al. (2019), on moss and lichen biomonitoring of atmospheric pollution in the Western Cape Province (South Africa) observed halogens to have elevated concentrations for samples collected from areas with close proximity to the ocean. That is, for both moss and lichen samples at areas closer to the ocean had higher halogen concentrations. Their results also confirmed elevated concentration levels for halogens (Cl, I, Br) in areas closest to the ocean. However, since fluorides are released into the atmosphere in large amounts by volcanic eruptions (Bajpai et al., 2011) and fluoride (F) compounds, remain in the same chemical form after they are emitted into the atmosphere, the very high levels of F bioaccumulation in this study, might have come from the degassing volcanic winds of Mt. Cameroon. Also, since other sampling points inland shows high concentration of F, degassing winds and ash deposition should have a greater influence.

In this study, even though Leptogium gelatinosum was the highest bioaccmulator, Stictastenroos and Heretodemia obscurata are also good accumulators, while all the Usnea species are poor accumulators. According to Brodo et al. (2001) fruticose lichens like Usnea are very sensitive to air pollutants than foliose lichens and occur only in very pure environments. Out of different growth forms of lichens, foliose lichens are prior to metal accumulation followed by crustose and squamulose lichens (Kar et al., 2014) and least by fruticose lichens (Shukla et al., 2014). However, lichens from the Usnea species have been used to evaluate heavy metal deposition patterns in the Antarctic (Poblet et al., 2011). Certain epiphytic lichens have been particularly gained attention for their bioaccumulation potential like Hypogymnia physodes for bioaccumulation of trace elements and Pvxine cocoes for bioaccumulation of metals (Bajpai et al., 2012; Daimari et al., 2020).

The differing accumulating potentials in substrates of lichen species in this study, (example, Parmotrema tinctorum (MC27) a corticolous sample collected in Ekona-Mbenge accumulated more than the saxicolous specimen (MC26) collected in the same point). Contrary, the findings of (Chettri et al., 1997), who used the lichen species Neophuscelia pulla and Xanthoparmelia taractica to study the bioaccumulation of heavy metals in abandoned copper mines in Greece, where there was a significant correlation between the copper content in the soil(saxicolose) and that of the tree(corticolous) lichen thalli. However, it is for this reason that most studies use epiphytic macro lichens as bio-monitors for air pollutants (Loppi and Pirintsos, 2003). For example, Käffer et al. (2011) also reported corticolous lichens as environmental indicators in urban areas in southern Brazil. Furthermore, the no to slight significant difference in means of halogens concentrations with regards to substrates in this study is in accordance with the study of Bajpai et al. (2011) in Mandav city in central India illustrated that although most of the metals were absent, or present in insignificant amount in substrates, yet the thallus of lichens had significantly higher concentration of metals such as Cd, Cr, Ni and Zn. Thus it is apparent that the accumulated metals were air borne.

All the eight lichen genera showed very high concentrations of F and low levels of Br. Weinstein et al. (1998) reported that, Br and I emissions are not usually of environmental importance and there is virtually no scientific literature on either element. The gas CI is potentially very hazardous but it is very rare to be released in sufficient quantities to pose a risk (Temple et al., 1998). Chlorine concentrations of 0.4 - 2.5 μg/g range cause severe symptoms like upper surface bleaching, epinasty (distorted growth), chlorosis (yellowing) and leaf drop to plants (Temple and Krause, 1998). The CI/F ratio in the specimens' ranges from 0.29 to 0.94, which is lower than those measured in lichen specimens at Mt. Etna which ranges from 0.51 to 1.46 (Notcutt and Davies, 1989). According to Delmelle et al. (1997) changes in the CI/F ratio may reflect different physico-chemical behaviour of the gases entering the atmosphere. However, Halmer et al. (2002) reported that in areas without nearby emission sources, the mean concentrations of fluoride in ambient air are generally less than $0.1\mu g/g$. This was observed from the control area (Mamfe 270 Km) with lower concentrations as compared to those from Mt. Cameroon. Even near emission sources, the levels of airborne fluoride usually do not exceed 2-3 μ g/g and in most soils, fluoride is present at concentrations ranging from 20 to 1000 μ g/g. This figure can reach several thousand $\mu g/g$ in mineral soils with natural phosphate or fluoride deposits. Therefore, the atmospheric halogens load at Mt. Cameroon is significantly high and lichens can be potential monitors of this volcanic gas flux.

V. Conclusion

Leptogium gelatinosum and Heretodemia obscurataare good accumulators, while Usnea species poor accumulators and therefore can be used for pollution bio-monitoring programs in Cameroon. The Leptogium gelatinosum species is therefore a suitable species for monitoring passive degassing at Mt. Cameroon. Also, considering that, lichens of the windward flank of MC accumulated more elemental contents than those from the leeward flanks, shows that wind direction and ash fall contribute largely to pollutant load in lichen species in the windward flanks of mount Cameroon reflecting volcanic degassing as the source. This chemical analysis serves as a baseline data for future studies.

Supplementary Information

The online version of this article offers supplementary material (https://doi.org/xxxxx).

ACKNOWLEDGMENTS

We are grateful for the support by the University of Botswana Gaborone (UBG) where the chemical analyses were performed. We thank Professor Suh Tening Aaron of the University of Buea, who is part of the interdisciplinary research framework under the theme "Understanding the environment of Mount Cameroon". We thank Dr Smith B. Babiaka of the Department of Chemistry, University of Buea, for proof reading of the manuscript.

Conflict of Interests

The authors declare that there is no conflict of interest whatsoever

This study did not receive any funding.

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Supplementary Material

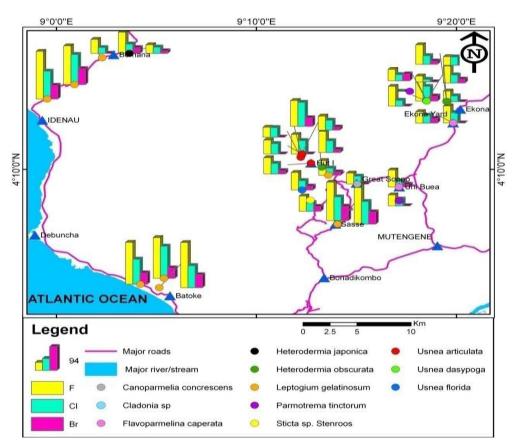


Figure S1: The different Halide concentrations in lichen species studied at the various sampling sites.

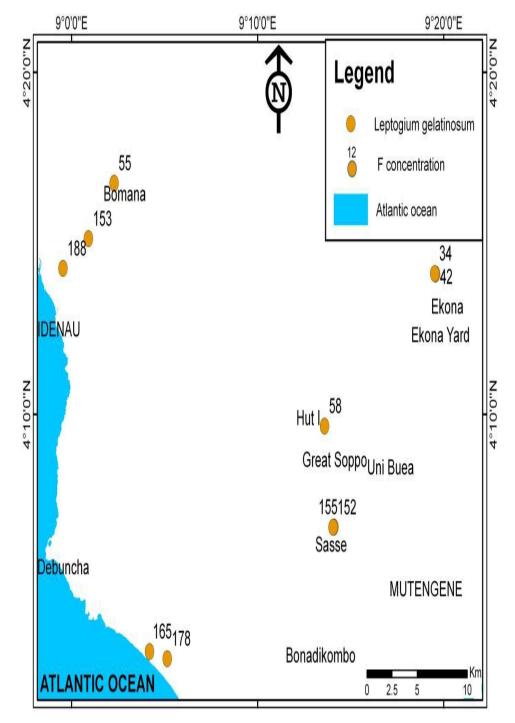


Figure S2: F bio-accumulation in Leptogium gelatinosum from different sampling sites on MC.

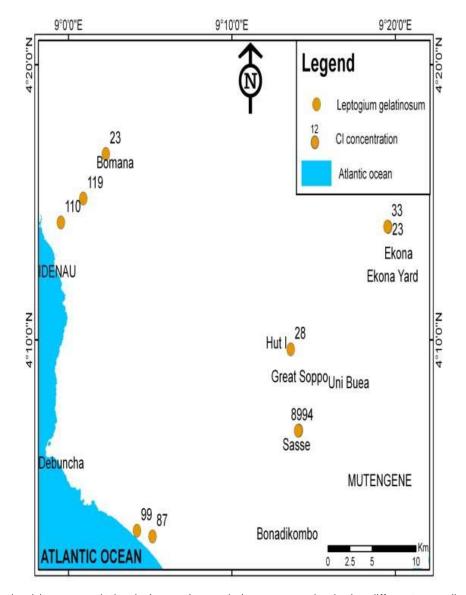


Figure S3: Chlorine bio-accumulation in Leptogium gelatinosum species in the different sampling sites on MC.

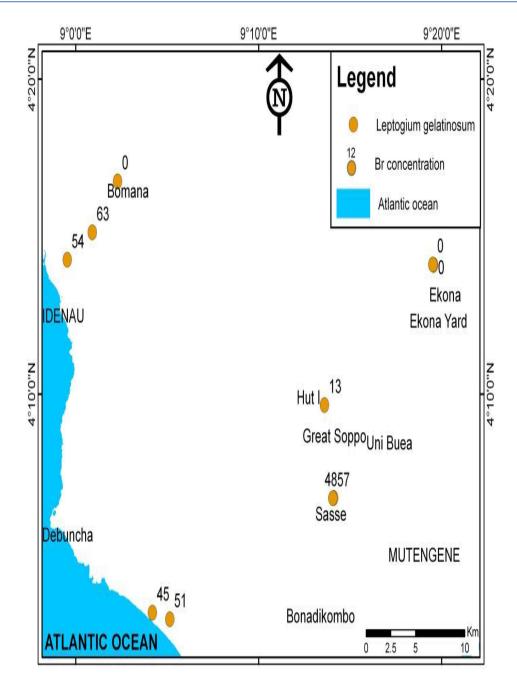


Figure S4: Br bio-accumulation in Leptogium gelatinosum at the different sampling sites of MC

LIST OF TABLES

Table S1: Lichen species code, sites and names used for chemical analysis.

Chemical Code	Site	Species name	Chemical Code	Site	Species name
MC 01	BmtSas	Leptogium gelatinosum	MC 18	BmtBok	Sticta stenroos
MC 02	Bmt UM	Leptogium gelatinosum	MC 19	BmtUM	Cladonia sp
MC 03	BmtSas	Leptogium gelatinosum	MC 20	BmtUM	Usnea articulata
MC 04	Boma	Leptogium gelatinosum	MC 21	BmtUM	Usnea articulata
MC 05	lden	Leptogium gelatinosum	MC 22	BmtUM	Usnea articulata
MC 06	lden	Leptogium gelatinosum	MC 23	BmtUM	Usnea florida
MC 07	Bakin	Leptogium gelatinosum	MC 24	BmtSop	Canoparmelia concrescens
MC 08	Batok	Leptogium gelatinosum	MC 25	EkLav	Parmotrema tinctorum
MC09	Batok	Leptogium gelatinosum	MC 26	EkLav	Parmotrema tinctorum

MC 10 MC 11	EkLav EkLav	Leptogium gelatinosum Leptogium gelatinosum	MC27 MC 28	EkLav Eklrad	Parmotrema tinctorum Parmotrema tinctorum
MC 12	Mamfe	Leptogium gelatinosum	MC 29	EkLav	Flavoparmelia caperata
MC 13	BmtUM	Parmotrema tinctorum	MC 30	Eklrad	Flavoparmelia caperata
MC 14	BmtUB	Parmotrema tinctorum	MC 31	EkLav	Heterodermia obscurata
MC 15	BmtUM	Flavoparmelia caperata	MC 32	EkLav	Heterodermia obscurata
MC 16	BmtUB	Flavoparmelia caperata	MC 33	EkLav	Usnea dasypoga
MC 17	BmtUM	Heterodermia obscurata	MC34	Boman	Heterodermia japonica

Table S2: Concentration of halogens analyzed for in lichen samples from MC(μg/g).

Chemical Code	Site	Species name	Substrate type	F	CI	Br
MC 01	BmtSas	Leptogium gelatinosum	Tree	155	89	48
MC 02	Bmt UM	Leptogium gelatinosum	Tree	58	28	13
MC 03	BmtSas	Leptogium gelatinosum	Rock	152	94	57
MC 04	Boma	Leptogium gelatinosum	Tree	55	23	0
MC 05	lden	Leptogium gelatinosum	Tree	188	110	54
MC 06	lden	Leptogium gelatinosum	Tree	153	119	63
MC 07	Bakin	Leptogium gelatinosum	Tree	165	99	45
MC 08	Batok	Leptogium gelatinosum	Tree	178	87	51
MC09	Batok	Leptogium gelatinosum	Rock	161	127	41
MC 10	EkLav	Leptogium gelatinosum	Tree	42	23	0
MC 11	EkLav	Leptogium gelatinosum	Rock	34	33	0
MC 12	Mamfe	Leptogium gelatinosum	Tree	25	0	0
MC 13	BmtUM	Parmotrema tinctorum	Tree	55	39	12
MC 14	BmtUB	Parmotrema tinctorum	Tree	43	27	0
MC 15	BmtUM	Flavoparmelina caperata	Tree	55	41	0
MC 16	BmtUM	Flavoparmelina caperata	Tree	67	23	11
MC 17	BmtUM	Heterodermia obscurata	Tree	83	46	12
MC 18	BmtBok	Stictastenroos	Tree	60	34	19
MC 19	BmtUM	Cladonia sp	Tree	102	95	34
MC 20	BmtUM	Usnea articulate	Tree	64	42	13
MC 21	BmtUM	Usnea articulate	Tree	39	38	0
MC 22	BmtUM	Usnea articulate	Tree	77	53	0
MC 23	BmtUM	Usne aflorida	Tree	69	37	15
MC 24	BmtSop	Canoparmelia concrescens	Tree	43	31	0
MC 25	EkLav	Parmotrema tinctorum	Tree	0	0	0
MC 26	EkLav	Parmotrema tinctorum	Rock	23	10	0
MC27	EkLav	Parmotrema tinctorum	Tree	67	32	15
MC 28	Eklrad	Parmotrema tinctorum	Tree	45	31	18
MC 29	EkLav	Flavoparmelina caperata	Tree	78	23	0
MC 30	Eklrad	Flavoparmelina caperata	Tree	67	41	0
MC 31	EkLav	Heterodermia obscurata	Tree	90	58	20
MC 32	EkLav	Heterodermia obscurata	Tree	77	34	13
MC 33	EkLav	Usnea dasypoga	Tree	45	24	27
MC34	Boma	Heterodermia japonica	Tree	34	25	10
Mean				78	47	18
Min				0	0	0
Max				188	127	63
SD				49	33	20