

Supplementary information

of the article

Synthesis, Structural Analysis, and Thermal and Spectroscopic Studies of Methylmalonate-Containing Zinc(II) complexes

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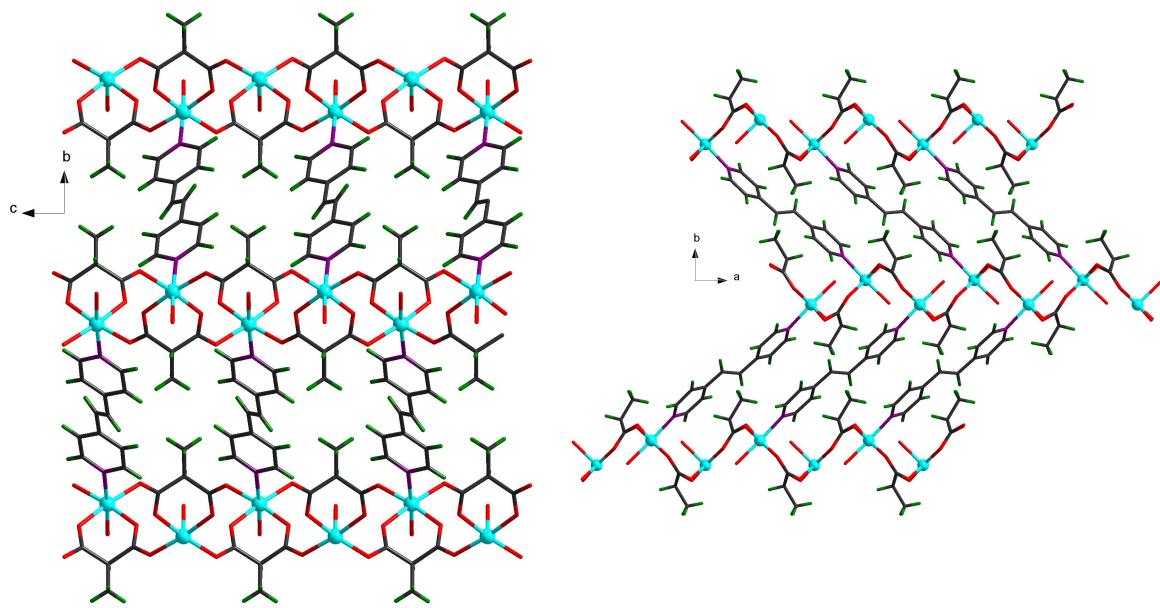


Figure S1. A view of the crystal packing of **3** along the *a* (left) and *c* axes (right).

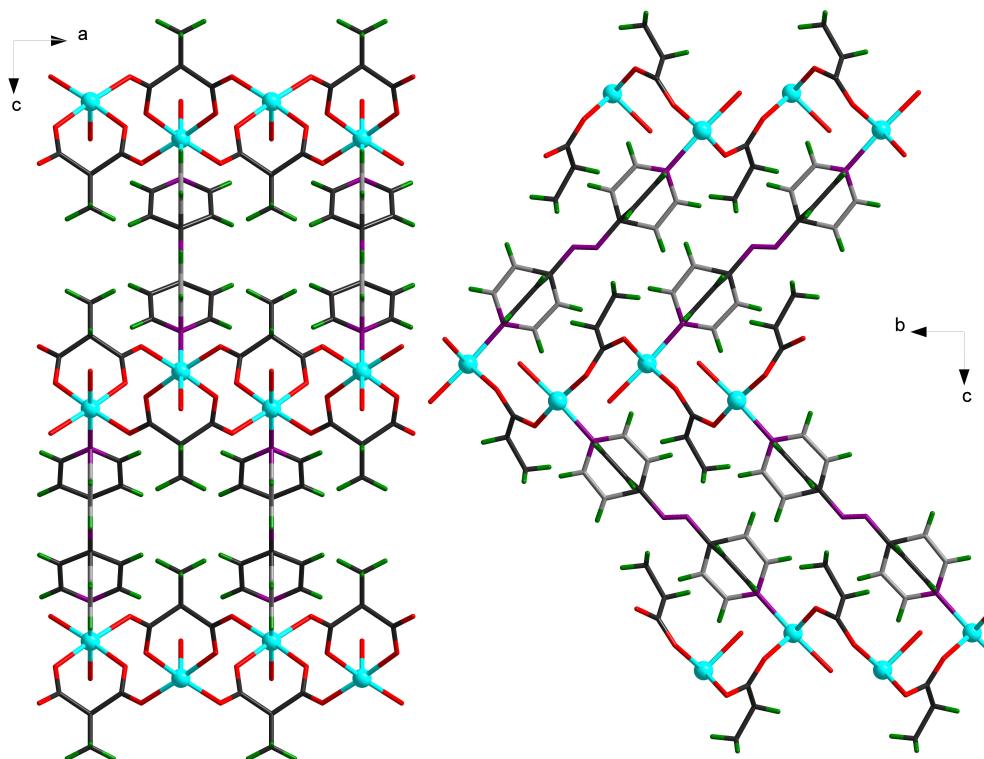


Figure S2. A view of the crystal packing of **5** along the *b* (left) and *a* axes (right). The two possible positions for the pyridyl rings of the azpy ligands have been also depicted.

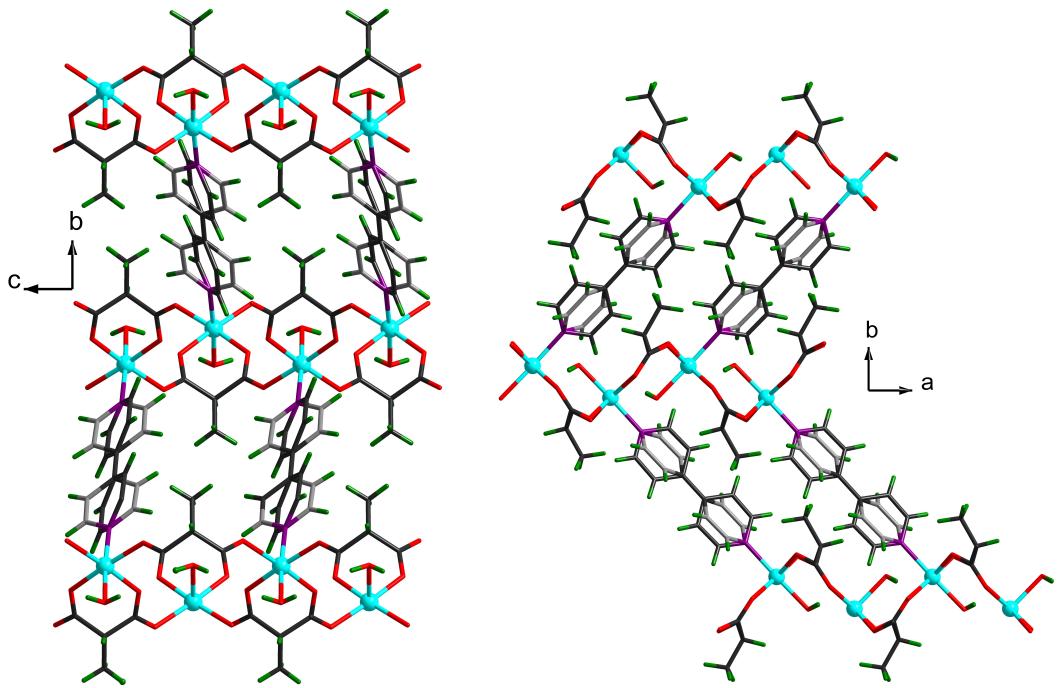


Figure S3. A view of the crystal packing of **2** along the *a* (left) and *c* axes (right). The two possible positions for the pyridyl rings of the 4,4'-bpy ligands have been also depicted.

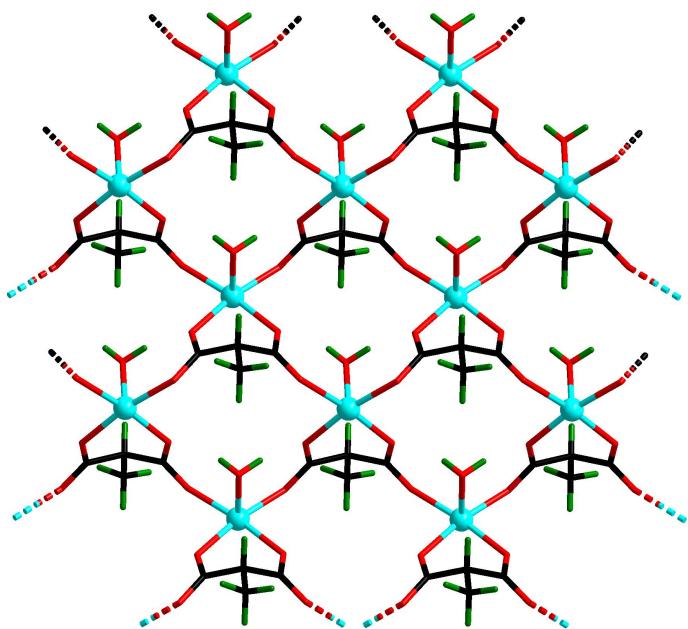


Figure S4. A perspective view of a fragment of the carboxylate-bridged square grid of the zinc atoms in the isostructural compounds **2-5**, which is extended in the *ac* (**2-3**) and *ab* (**4-5**) planes.

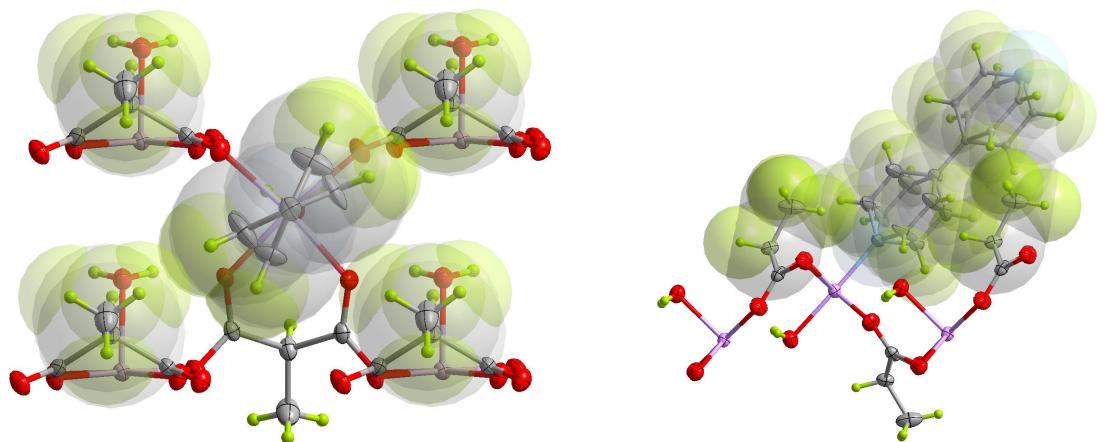


Figure S5. A view of the delocalization of the pyridyl groups of the 4,4'-bpy in **2**. The space filling mode is superimposed to the ortep view to show how the methyl groups preclude the free rotation of the pyridyl, and therefore it has to vibrate around these two main positions.

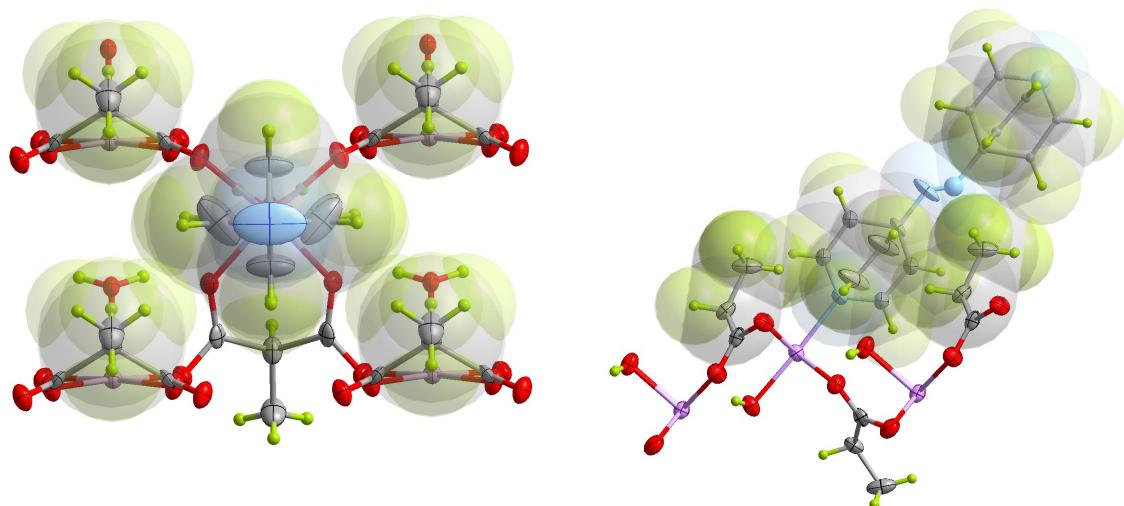


Figure S6. A view of the delocalization of the pyridyl groups of the azpy in **5**. The space filling mode is superimposed to the ortep view to show how the methyl groups allow the free rotation of the pyridyl.

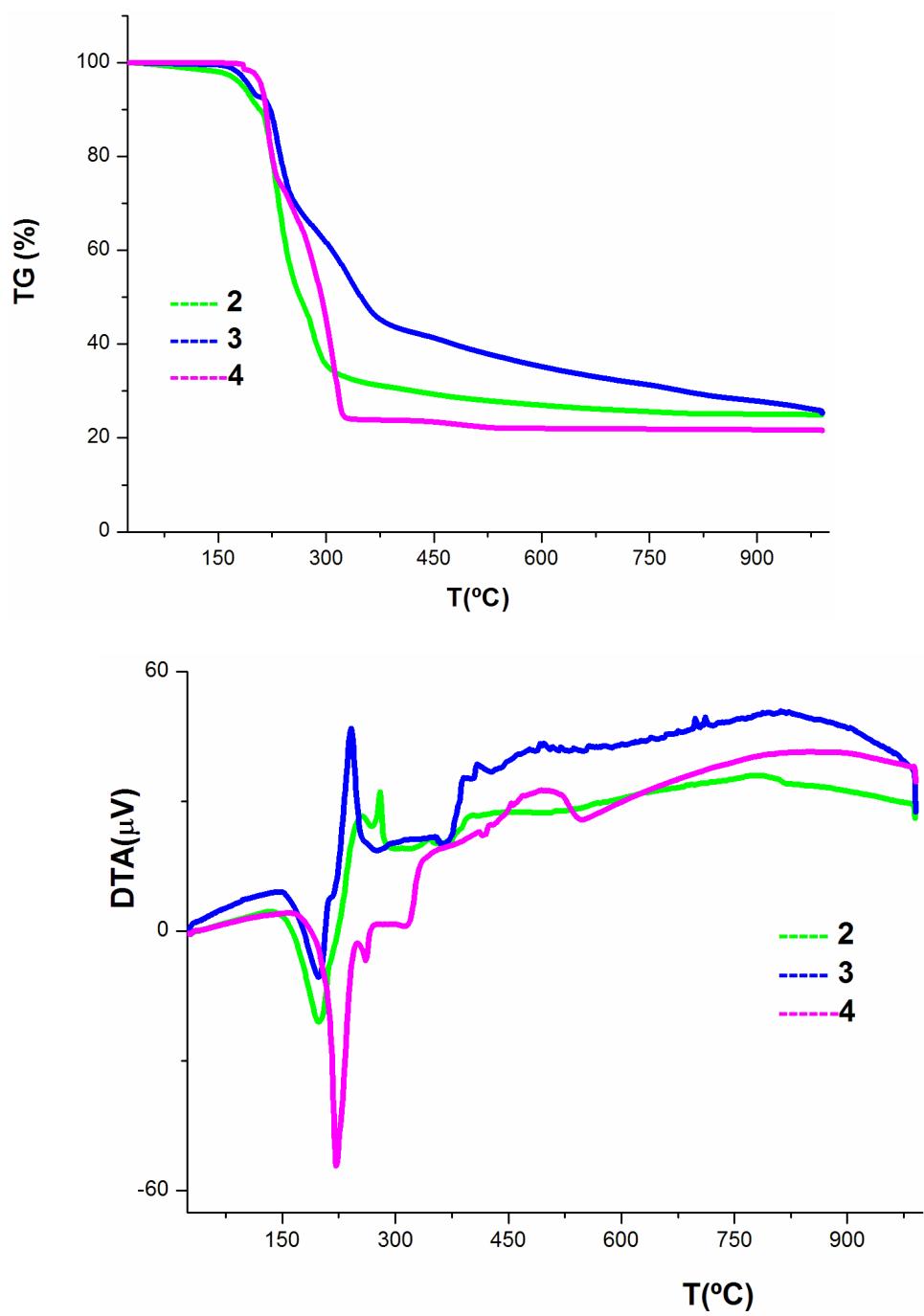


Figure S7. TG/DTG-DTA curves of **2-4** complexes. TG = mass loss (percent); DTA = ΔT (μ V) and DTG = percent per minute (\downarrow endo and \uparrow exo).