**RESEARCH ARTICLE** 



# Source apportionment of PM<sub>10</sub> and health risk assessment related in a narrow tropical valley. Study case: Metropolitan area of Aburrá Valley (Colombia)

Carlos Ramos-Contreras<sup>1,2</sup> · María Piñeiro-Iglesias<sup>1</sup> · Estefanía Concha-Graña<sup>1</sup> · Joel Sánchez-Piñero<sup>1</sup> · Jorge Moreda-Piñeiro<sup>1</sup> · Amaya Franco-Uría<sup>3</sup> · Purificación López-Mahía<sup>1</sup> · Francisco Molina-Pérez<sup>2</sup> · Soledad Muniategui-Lorenzo<sup>1</sup>

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### Abstract

This study investigates spatio-temporal variations of  $PM_{10}$  mass concentrations and associated metal(oid)s,  $\delta^{13}C$  carbon isotope ratios, polycyclic aromatic hydrocarbons (PAHs), total organic carbon (TOC) and equivalent black carbon (eBC) concentrations over a half year period (from March 2017 to October 2017) in two residential areas of Medellín (MED-1 and MED-2) and Itagüí municipality (ITA-1 and ITA-2) at a tropical narrow valley (Aburrá Valley, Colombia), where few data are available. A total of 104 samples were analysed by using validated analytical methodologies, providing valuable data for PM<sub>10</sub> chemical characterisation. Metal(oid)s concentrations were measured by inductively coupled plasma mass spectrometry (ICP-MS) after acid digestion, and PAHs concentrations were measured by Gas Chromatography-Mass Spectrometry (GC-MS) after Pressurised Hot Water Extraction (PHWE) and Membrane Assisted Solvent Extraction (MASE). Mean PM<sub>10</sub> mass concentration ranged from 37.0 µg m<sup>-3</sup> to 45.7 µg m<sup>-3</sup> in ITA-2 and MED-2 sites, respectively. Al, Ca, Mg and Na (from 6249 ng m<sup>-3</sup> for Mg at MED-1 site to 10,506 ng m<sup>-3</sup> for Ca at MED-2 site) were the major elements in PM<sub>10</sub> samples, whilst As, Be, Bi, Co, Cs, Li, Ni, Sb, Se, Tl and V were found at trace levels (<5.4 ng m<sup>-3</sup>). Benzo[g,h,i] perylene (BghiP), benzo[b+j]fluoranthene (BbjF) and indene(1,2,3-c,d)pyrene (IcdP) were the most profuse PAHs in PM<sub>10</sub> samples, with average concentrations of 0.82-0.86, 0.60-0.78 and 0.47-0.58 ng m<sup>-3</sup>, respectively. Results observed in the four sampling sites showed a similar dispersion pattern of pollutants, with temporal fluctuations which seems to be associated to the meteorology of the valley. A PM source apportionment study were carried out by using the positive matrix factorization (PMF) model, pointing to re-suspended dust, combustion processes, quarry activity and secondary aerosols as PM<sub>10</sub> sources in the study area. Among them, combustion was the major  $PM_{10}$  contribution (accounting from 32.1 to 32.9% in ITA-1 and ITA-2, respectively), followed by secondary aerosols (accounting for 13.2% and 23.3% ITA-1 and MED-1, respectively). Finally, a moderate carcinogenic risk was observed for PM10-bound PAHs exposure via inhalation, whereas significant carcinogenic risk was estimated for carcinogenic metal(oid)s exposure in the area during the sampling period.

**Keywords** Particulate matter  $\cdot$  Metal(oid)s  $\cdot \delta^{13}C$  carbon isotope ratios  $\cdot$  Polycyclic aromatic hydrocarbons  $\cdot$  Source apportionment  $\cdot$  Health risk assessment

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Jorge Moreda-Piñeiro jorge.moreda@udc.es

<sup>1</sup> Department of Chemistry, Faculty of Sciences, Grupo Química Analítica Aplicada (QANAP), University Institute of Research in Environmental Studies (IUMA), University of A Coruña, Campus de A Coruña, S/N. 15071, A Coruña, Spain

<sup>2</sup> Grupo de Investigación en Gestión y Modelación Ambiental (GAIA), Escuela Ambiental, Facultad de Ingeniería, Universidad de Antioquia UdeA, Calle 70 No. 52-21, Medellín, Colombia

<sup>3</sup> Dept. of Chemical Engineering, School of Engineering, University of Santiago de Compostela, 15782 Santiago de Compostela, Spain

#### Introduction

Several adverse health effects have been associated with atmospheric particulate matter (PM) exposure by epidemiological studies decades ago (Manisalidis et al. 2020; Pope and Dockery 2006), resulting in increased mortality and morbidity rates mainly due to respiratory and cardiovascular diseases, including lung cancer (Chen and Hoek 2020; Mueller et al. 2021; Wang et al. 2020). Due to the spatial-temporal variability of atmospheric particles, PM may encompass many associated pollutants (both being constituents of particles or being adsorbed on their surfaces) which are potential contributors to PM adverse health effects as they can be potentially absorbed to bloodstream after inhalation (Arias-Pérez et al. 2020; Lu et al. 2015; Mousavi et al. 2017).

As posing a great threat to human health, air quality policies should entail PM source apportionment studies to identify possible PM sources and their chemical profile since they would lead to a better assessment of PM-associated health risks. Considering that, it would allow the development and implementation of effective mitigation policies and strategies to protect human health from PM pollution (Hopke 2008). Measurement of PM elemental and organic composition together with PM properties such as elemental and organic carbon would be of great interest as they can be used as source tracers for PM apportionment studies (Hsu et al. 2016; Liu et al. 2020). In addition, study of parameters such as stable carbon isotope ratio  $({}^{13}C/{}^{12}C$  expressed as a  $\delta^{13}$ C) in PM samples can support apportionment studies to identify potential anthropogenic aerosol sources (road traffic or industrial emissions) in typical urban environments (Buczyńska et al. 2013; Kunwar et al. 2018, 2016; Morera-Gómez et al. 2021; Widory 2006).

Exposure to atmospheric pollutants and their health impacts will be influenced by meteorological conditions and topography of the areas studied. On this basis, PM exposure could potentially be increased in settlements located in narrow valleys where local conditions do not allow an adequate diffusion such as Aburrá Valley (Medellín Metropolitan Area, Antioquia, Colombia), located in a mountainous area in which some municipalities of Medellín Metropolitan Area are settled. Hence, the study of atmospheric pollutants transport and transformation require comprehensive knowledge of related meteorological phenomena (such as wind speed and direction, temperature, precipitation and solar radiation). Epidemiological studies conducted in the Metropolitan Area of Aburrá Valley (ANVA) show that the burden of disease attributable to PM exposure represented about 9.2% of total deaths during 2011, whereas around 72% of the mortality due to air pollution in Medellín was associated with Aburrá Valley area (AMVA 2017a). According to the ANVA's air quality network, vehicular exhaust emissions were the major source of PM in the area, whilst local studies concerning mountain meteorology shown that daytime winds rise during dry days along the valley, entering by the northeast branch (Adarve and Molina 1984). Additionally, significant thermo-dependent diffusion processes at ground level have been reported to cause important variations of atmospheric boundary layer's (ABL) height in the area, ranging between 200 and 1800 m (Herrera-Mejía and Hoyos 2019). As ABL height is influenced by atmospheric stability conditions, dispersion of pollutants could be favoured and hindered under unstable and stable meteorological conditions, respectively (Correa et al. 2009; Rendón et al. 2020). It is in the ultimate case when formation of thermal inversion layer prolongs, increasing the exposure to atmospheric pollutants. Mostly, thermal inversion and high PM concentration episodes take place between 17:00 and 10:00 during March-April and October, when the meteorological conditions are of low-height cloudiness are presented (AMVA 2017a).

Although the increasing concern about inorganic pollutants and carbonaceous content (including organic pollutants such as polycyclic aromatic hydrocarbons (PAHs)) determination in PM, there are few studies focused on polluted and densely populated regions settled in tropical narrow valleys, characterised by weather conditions that hinder atmospheric pollutants' dilution (Mueller et al. 2019; Zalakeviciute et al. 2020). Moreover, no reports regarding spatio-temporal variations of trace metal(oid)s and PAHs concentrations in Aburrá Valley have been found in literature, whilst PM source apportionment studies are scarce in Latin America, especially in high altitude cities (with lowdensity air) (Zalakeviciute et al. 2020). The present work aims to assess the chemical composition (encompassing elements and metal(oid)s, PAHs and organic content) of PM<sub>10</sub> in several sites of Aburrá Valley during 2017, providing novel contribution to the field due to the lack of studies in the area. Also, spatio-temporal variations in the area and  $PM_{10}$  sources will be explored, whilst carcinogenic human health risks will be assessed by following the United States Environmental Protection Agency's (USEPA) guidelines.

### Material and methods

### Study area description

AMVA is a densely populated region (3.220 inhabitants per square kilometre) with eminently urban characteristics (DANE 2018), where Medellín is the main city (1500 masl). The combination of diverse anthropogenic emission sources (household, industrial and vehicular emissions) with tropical



Fig.1 Location of sampling sites at Aburrá Valley South Zone. Source: Google Earth®. Windrose corresponds to average value of speed and direction wind during the study

climate and topographical peculiarities, make ANVA one of the most polluted sites in Colombia. In addition, air masses movement of the Intertropical Convergence Zone causes a bimodal cycle of precipitation in the area throughout the year, with a first rainy season (from March to May) and a second one (from September to November) (CICE 2017). In the present study,  $PM_{10}$  samples were collected in four sampling stations (belonging to the ANVA Air Monitoring Network) at the south-central area of AMVA, comprising Medellín (MED-1 and MED-2) and Itagüí municipalities (ITA-1 and ITA-2). Further details concerning location of sampling sites are given in Fig. 1 and STable 1 (Supporting Information, SI). Also, meteorological data (wind speed, solar radiation and rainfall) were measured at the station of the air quality network near MED-1. Moreover, data from continuous  $PM_{10}$  and  $PM_{2.5}$  measurements were acquired from the Air Monitoring Network.

# Sampling and PM<sub>10</sub> mass determination

 $PM_{10}$  samples were collected on quartz-fiber filters (203×254 mm, Whatman) for 24 h at 66.6 m<sup>3</sup> h<sup>-1</sup> according to US reference method (US Government 1991) by using Graseby-Andersen GBM2360 high-volume samplers (Graseby-Andersen Inc., Smyrna, GA). Quartz-fiber filters were pre-heated at 450 °C for 4 h (for organics

contamination removal) before using. After sampling and filter conditioning (by keeping a temperature and humidity of 25 °C and 50%, respectively, for 24 h),  $PM_{10}$  mass was gravimetrically determined. Afterwards, filters were stored at -20 °C until further analysis. Field blanks (blank filters placed inside the samplers without PM collection) were also collected along with daily samples and analysed following the same procedure. A total of 104 samples were collected simultaneously from the four sampling sites between March and October 2017.

#### Analytical methods for chemical characterisation

#### Metal(oid)s extraction and quantification

Four circular pieces of quartz filters (diameter of 16 mm) were taken from each  $PM_{10}$  filter using a steel puncher (Selecta, Barcelona, Spain) and subjected to acid digestion (Piñeiro-Iglesias et al. 2003). In brief, filter portions of each sample were transferred to a polytetrafluoroethylene (PTFE) digestion bomb with 2.5 ml of nitric acid (Baker®, Phillipsburg, PA, USA) and 5 ml of concentrated hydrochloric acid (Baker®) and heated at 90 °C for 12 h. After addition of 2.5 ml of perchloric acid (Baker®) and 1 ml of nitric acid, the mixture was driven to dryness. Finally, residue was reconstituted by adding 2.5 ml of nitric acid and taken to 25 ml using ultrapure water.

Quantification of the isotopes <sup>27</sup>Al, <sup>75</sup>As: <sup>137</sup>Ba, <sup>9</sup>Be, <sup>209</sup>Bi,<sup>44</sup>Ca, <sup>111</sup>Cd, <sup>59</sup>Co, <sup>52</sup>Cr, <sup>133</sup>Cs <sup>63</sup>Cu, <sup>56</sup>Fe, <sup>39</sup>K, <sup>7</sup>Li, <sup>24</sup>Mg, <sup>55</sup>Mn, <sup>95</sup>Mo, <sup>23</sup>Na, <sup>60</sup>Ni, <sup>29</sup>Si, <sup>31</sup>P, <sup>208</sup>Pb, <sup>121</sup>Sb, <sup>78</sup>Se, <sup>118</sup>Sn, <sup>87</sup>Sr, <sup>205</sup>Tl, <sup>51</sup> V and <sup>66</sup>Zn was performed by inductively coupled plasma mass spectrometry (ICP-MS) (Thermo Finnigan X Series, Waltham, MA, USA) in the peak jump mode under the following instrumental conditions: radio frequency (RF) power 1350 W, nebuliser gas flow 0.8 L min<sup>-1</sup>, auxiliary gas flow 0.9 L min<sup>-1</sup>, and plasma gas flow 15.0 L min<sup>-1</sup>. Calibration graphs were constructed with aqueous standard metal(oid)s solutions (with 2.0 M nitric acid) covering a concentration range of 0 to 2000 µg L<sup>-1</sup> (STable 2). Also, <sup>45</sup>Sc, <sup>72</sup>Ge, <sup>89</sup>Y, <sup>103</sup>Rh and <sup>115</sup>In were used as internal standards. At least one procedural blank (field blanks subjected to the same digestion procedure as PM<sub>10</sub> samples) was analysed in each extraction batch. The limits of quantification (LOQs) (mean blank  $\pm 10$  standard deviation (SD) criterion) were estimated by analysing 11 procedure blanks (STable 2), being in the range of 0.05 (Tl) and 930 ng m<sup>-3</sup> (Al). In addition, trueness of the method was assessed by analysing the SRM 1649a urban particulate matter reference material (National Institute of Standards and Technology, Gaithersburg, MD, USA) in triplicates. Concentrations found are in good agreement with the certified values (STable 2) after statistical evaluation by applying a t test at 95% confidence level for two degrees of freedom. A

statistical summary of metal(oid)s concentrations found in  $PM_{10}$  samples (N = 104) are shown in STable 3.

### Stable isotopic $\delta^{13}$ C quantification

The isotope ratios of  $\delta^{13}$ C in the PM<sub>10</sub> samples were determined by mass spectrometer of isotopic proportions (Delta V Advantage, Thermo Fisher Scientific, Whaltman, MA, USA) calibrated with certified reference materials (NBS-22, IAEA-CH- 6 and USGS 24) from International Atomic Energy Agency-IAEA (Vienna, Austria), coupled to an elemental sample analyser (Flash EA1112 HT, Thermo Fischer Scientific, Bremen, Germany). Procedures and details for  $\delta^{13}$ C quantification are shown in SI.

# Equivalent black carbon, total organic carbon and elemental analysis

Equivalent Black Carbon (eBC) were determined by using an optical transmissometer Model OT-21 (Magee Scientific, California, USA), whereas a ThermoQuest Flash EA 1112 (ThermoQuest, Rodano, Italy) elemental analyser were used for the analysis of H, C, S and N and Total Organic Carbon (TOC) (after sample acidification) according to a previous study (Fernández-Amado et al. 2018). Further details concerning eBC quantitation procedure are given in SI, whilst statistical summary of eBC, TOC and elemental (H, C, S and N) concentrations in PM<sub>10</sub> samples are shown in STable 3.

# Polycyclic aromatic hydrocarbon extraction and quantification

PAHs were extracted from PM<sub>10</sub> samples extracted by Pressurised Hot Water Extraction (PHWE) (using an ASE 200 accelerated extraction solvent system, Dionex, Sunnyvale, CA, USA) with water: methanol (3:1) as extracting solvent (Ramos-Contreras et al. 2019). Six circular pieces (16 mm diameter) of each sample were placed in extraction cells with cellulose filters at both ends and spiked with 150 µL of deuterated-labelled PAHs surrogate solution mix (naphthalene-d<sub>8</sub>, acenaphthylene-d<sub>8</sub>, phenanthrene-d<sub>10</sub>, fluoranthene-d<sub>10</sub>, pyrene-d<sub>10</sub>, chrysene-d<sub>12</sub>, benzo[e]pyrene-d<sub>12</sub> and benzo[g,h,i]perylene d-12 in hexane, 200  $\mu$ g L<sup>-1</sup>). A single extraction cycle was performed at 200 °C and 2000 psi, with a static time of 5 min. Once finished, extracts were driven to 40 ml with water: methanol (3:1). Subsequently, 15 mL aliquots were pre-concentrated and cleaned up by Membrane Assisted Solvent Extraction (MASE) using a Gerstel device (Mülheim, Germany) consisting of a 20-mL glass vial and a membrane insert made of dense polypropylene (4-cm long with a wall thickness of 0.03 mm and an internal diameter of 6 mm). Membranes were filled with 500  $\mu L$  of internal standard solution (antracene-d\_{10} and dibenzo[a,h]anthracene- $d_{14}$  in hexane, 100 µg L<sup>-1</sup>) and the vial was sealed with a metallic crimp cap provided with PTFE septa. MASE devices were orbitally shaken (730 rpm) and incubated (30 °C) during 90 min by using a Combi PAL autosampler (CTC-Analytics, Zwingen, Switzerland) tool.

Subsequently, PAHs in hexane extracts were quantitated by a Thermo-Finnigan Trace GC chromatograph (Waltham, MA, USA) equipped with the GC PAL autosampler (CTC-Analytics, Zwingen, Switzerland), Programmed Temperature Vaporizing (PTV) injector and coupled to an ion trap mass spectrometer (Polaris Q), using the Xcalibur software as data processor. A PTV injector provided with a glass wool packed PTV Silcosteel® liner with 2 mm of inner diameter (Thermo Finnigan, Thermo Electron Corporation, Waltham, USA) was used, setting a sample injection volume of 25 µL and a PTV programme starting at 55 °C and heated at 3 °C s<sup>-1</sup> until 300 °C (held for 20 min). The separation was performed with a DB-XLB column (60 m×0.25 mm, 0.25 µm film thickness) (J& W Scientific, Folsom, CA, USA), whilst GC oven temperature started at 50 °C (3 min), increased by 4 °C min<sup>-1</sup> to 325 °C, and held for 20 min. The mass spectrometer [electron impact (EI); 70 eV] operated in tandem mass spectrometry detection mode. Transfer line and ion source temperatures were set at 300 °C and 270 °C, respectively. Helium (99.9999%) was used as the collision gas at the ion trap chamber, and as the carrier gas, under a constant flow rate of 1 mL min<sup>-1</sup>.

A total of 23 PAHs were analysed, comprising naphthalene (NAP), methylnaphthalene (Me-NAP), acenaphthene (ACE), acenaphthylene (ACY), fluorene (FLU), methyl fluorene (Me-FLU), dibenzothiophene (DBT), phenanthrene (PHE), anthracene (ANT), methyl anthracene (Me-ANT), fluoranthene (FLT), pyrene (PYR), retene (RET), benzo[a] anthracene (BaA), triphenylene (TPY), chrysene (CHR), benzo[b+j] fluoranthene (BbjF), benzo[k]fluoranthene (BkF), benzo[e] pyrene (BeP), benzo[a]pyrene (BaP), dibenzo[a,h] anthracene (DahA), indene(1,2,3-c,d)pyrene (IcdP) and bezo[g,h,i] perylene (BghiP). External calibration graphs were carried out in a concentration range of 0 to 300  $\mu$ g L<sup>-1</sup> for linearity check, being correlation coefficients ( $\mathbb{R}^2$ ) between 0.9965 to 0.9994 for all the PAHs. The limits of quantification (LOQs) calculated as mean blank + 10 SD criterion (N = 11 procedure blanks) were 0.075, 0.039, 0.005, 0.003, 0.003, 0.021, 0.051, 0.011, 0.015, 0.004, 0.003, 0.003, 0.002, 0.001, 0.001, 0.001, 0.006, 0006, 0.006, 0.002, 0.002, 0.010 and  $0.01 \text{ ng m}^{-3}$ , for NAP, Me-NAP, ACY, ACE, FL, Me-FL, DBT, PHE, ANT, Me-ANT, FLT, PYR, RET, BaA, TPY, CHR, BbjF, BkF, BaP, BeP, DahA, IcdP and BghiP, respectively; being low enough to perform PAHs quantification in studied PM<sub>10</sub> samples. The inter-day precision and trueness of the analytical procedure were estimated by analysing the SRM 1649b Urban Dust (National Institute of Standards and Technology, Gaithersburg, MD, USA) within different days (N=8). All PAHs demonstrated good inter-day precision, obtaining relative standard deviations (RSDs) of 4.9 to 22.3% for Me-NAP and ACY, respectively; whilst analytical recoveries obtained from SRM 1649b analysis were satisfactory (65 to 117%). A statistical summary of PAHs concentrations found in PM<sub>10</sub> samples (N=104) are shown in STable 4.

### Source apportionment

Positive Matrix Factorization (PMF) has been used by many researchers to recognise and characterise the major  $PM_{10}$  sources. Therefore, a PM apportionment study to estimate the possible contribution of different PM sources in ANVA area was performed by using PMF software provided by USEPA (EPA PMF 5.0 software) (Paatero and Hopke 2003; USEPA 2014; Norris et al. 2014).

# Human health risk assessment of PM<sub>10</sub>-bound metal(oid)s and PAHs

The cancer risk of  $PM_{10}$ -bound metal(oid)s and PAHs was evaluated according to the USEPA's human health risk assessment models, being further described in the SI (USEPA 2009).

### **Results and discussion**

#### PM<sub>10</sub> mass concentration

The average  $PM_{10}$  concentration in the study area was 41.6 µg m<sup>-3</sup> (STable 3), ranging from 37.0 µg m<sup>-3</sup> (ITA-2) to 45.7 µg m<sup>-3</sup> (MED-2) (Table 1, Fig. 2A). Although  $PM_{10}$  mean concentration is below the Colombian annual limit value (50 µg m<sup>-3</sup>) (RC- MADS 2017); it is quite above the limit set by the World Health Organization guidelines (15 µg m<sup>-3</sup>) (WHO 2021). Compared to mean values reported in Bogotá (most populated city in Colombia), average concentration found in the present study is lower than mean concentrations reported for traffic (53 µg m<sup>-3</sup>) or industrial sites (110 µg m<sup>-3</sup>), whereas it is quite similar to mean concentration found in residential areas (41.4 µg m<sup>-3</sup>) (Vargas et al. 2012).

As can be seen from SFigure 1, mean daily fluctuation of  $PM_{10}$  concentration shows two peaks at 08:00–10:00 and 18:00–20:00, which seems to be associated with the peak working hours when anthropogenic activity is expected to be higher (mainly vehicular traffic sources). However, a thorough study considering the climate should be considered as it may affect  $PM_{10}$  levels significantly (Herrera-Mejía and Hoyos 2019; Roldán-Henao et al. 2020). Additionally,  $PM_{10}$ levels observed for the sampling sites (MED-1, MED-2, ITA-1 and ITA-2) showed no statistically significant differences (p > 0.05) among them (Fig. 2B). Regarding temporal

**Table 1** Mean, minimum (Min) and maximum (Max) values of  $PM_{10}$  mass (µg m<sup>-3</sup>) and elements, equivalent black carbon (eBC) and total organic carbon (TOC) concentrations (ng m<sup>-3</sup>) found in each sampling site

|                       | MED-1 (N=27) |        |        | MED-2 ( | (N=25) |        | ITA-1 ( <i>N</i> =27) |        |        | ITA-2 $(N=25)$ |        |        |
|-----------------------|--------------|--------|--------|---------|--------|--------|-----------------------|--------|--------|----------------|--------|--------|
|                       | Mean         | Min    | Max    | Mean    | Min    | Max    | Mean                  | Min    | Max    | Mean           | Min    | Max    |
| PM <sub>10</sub> mass | 40.1         | 16.5   | 88.7   | 45.7    | 21.0   | 75.3   | 43.7                  | 22.3   | 76.0   | 37.0           | 18.1   | 61.8   |
| Al                    | 8575         | <930   | 51,709 | 9313    | < 930  | 51,292 | 8385                  | < 930  | 52,922 | 6712           | <930   | 45,580 |
| As                    | 1.3          | 0.31   | 3.4    | 1.6     | 0.34   | 5.3    | 2.2                   | 1.0    | 4.5    | 1.6            | 0.24   | 4.1    |
| Ba                    | 103          | <7.3   | 578    | 115     | <7.3   | 606    | 95.8                  | 11.7   | 540    | 76.6           | 9.1    | 485    |
| Be                    | 0.19         | < 0.09 | 0.98   | 0.23    | < 0.09 | 1.4    | 0.23                  | < 0.09 | 1.3    | 0.19           | < 0.09 | 1.2    |
| Bi                    | 1.5          | < 0.15 | 5.0    | 1.8     | < 0.15 | 7.4    | 1.6                   | < 0.15 | 7.4    | 1.8            | < 0.15 | 7.6    |
| Ca                    | 8580         | < 820  | 41,249 | 10,506  | < 820  | 84,893 | 10,048                | < 820  | 83,873 | 9051           | < 820  | 74,990 |
| Cd                    | 0.47         | < 0.17 | 1.7    | 0.66    | < 0.17 | 1.6    | 0.75                  | 0.21   | 1.4    | 0.80           | < 0.17 | 2.0    |
| Co                    | 0.73         | < 0.19 | 2.0    | 0.87    | < 0.19 | 2.7    | 1.0                   | 0.22   | 2.8    | 0.79           | < 0.19 | 2.6    |
| Cr                    | 16.9         | < 6.8  | 94.0   | 19.1    | < 6.8  | 85.4   | 16.7                  | < 6.8  | 83.9   | 16.2           | < 6.8  | 73.3   |
| Cs                    | 0.38         | < 0.62 | 1.4    | 0.36    | < 0.62 | 1.5    | 0.34                  | < 0.62 | 1.5    | 0.30           | < 0.62 | 1.3    |
| Cu                    | 25.6         | < 9.6  | 72.1   | 73.9    | < 9.6  | 183    | 43.6                  | 21.3   | 89.5   | 20.9           | < 9.6  | 60.4   |
| Fe                    | 680          | 212    | 1518   | 899     | 206    | 1756   | 616                   | 214    | 1173   | 653            | 224    | 2834   |
| Κ                     | 492          | < 96.0 | 1207   | 992     | < 96.0 | 12,357 | 962                   | < 96.0 | 12,185 | 960            | < 96.0 | 10,894 |
| Li                    | 4.0          | < 5.9  | 11.0   | 4.3     | < 5.9  | 17.3   | 4.2                   | < 5.9  | 17.3   | 3.8            | < 5.9  | 14.5   |
| Mg                    | 6249         | < 203  | 33,189 | 7038    | < 203  | 46,032 | 6368                  | <203   | 44,147 | 5585           | < 203  | 39,890 |
| Mn                    | 14.7         | 3.6    | 36.4   | 19.7    | < 3.0  | 65.5   | 17.0                  | 4.2    | 45.3   | 14.7           | 3.9    | 35.0   |
| Мо                    | 12.3         | <17.2  | 37.1   | 12.1    | <17.2  | 40.5   | 9.6                   | <17.2  | 30.3   | 12.5           | <17.2  | 38.9   |
| Na                    | 8066         | < 547  | 14,759 | 7419    | < 547  | 14,006 | 7045                  | < 547  | 13,523 | 7438           | < 547  | 13,072 |
| Ni                    | 1.7          | < 1.5  | 6.7    | 1.9     | <1.5   | 10.1   | 2.2                   | <1.5   | 6.8    | 1.9            | <1.5   | 7.0    |
| Р                     | 341          | <76.2  | 1552   | 310     | <76.2  | 1158   | 319                   | <76.2  | 1398   | 329            | <76.2  | 1467   |
| Pb                    | 17.0         | 5.6    | 41.8   | 16.9    | 4.5    | 60.2   | 13.9                  | 3.7    | 68.1   | 17.8           | 4.6    | 65.8   |
| Sb                    | 5.6          | 0.71   | 19.3   | 6.2     | 0.88   | 13.1   | 5.2                   | 1.3    | 13.7   | 4.9            | 1.8    | 10.8   |
| Se                    | 1.9          | < 3.7  | 4.7    | 2.4     | < 3.7  | 6.1    | 2.7                   | 3.7    | 6.1    | 2.1            | 3.7    | 4.5    |
| Sn                    | 8.3          | < 3.1  | 35.5   | 13.2    | < 3.1  | 34.9   | 11.0                  | < 3.1  | 34.9   | 10.6           | < 3.1  | 28.5   |
| Sr                    | 23.9         | < 2.1  | 140    | 25.6    | < 2.1  | 143    | 24.0                  | < 2.1  | 136    | 20.2           | < 2.1  | 133    |
| Tl                    | 0.07         | < 0.05 | 0.27   | 0.11    | < 0.05 | 0.22   | 0.16                  | < 0.05 | 0.56   | 0.25           | < 0.05 | 0.68   |
| V                     | 2.1          | 0.36   | 5.6    | 2.3     | 0.47   | 5.4    | 3.3                   | 0.96   | 7.7    | 2.1            | 0.54   | 4.3    |
| Zn                    | 86.0         | < 20.8 | 299    | 129     | < 20.8 | 426    | 140                   | 27.1   | 422    | 211            | 34.4   | 805    |
| С                     | 8597         | 3310   | 16,300 | 10,250  | 3280   | 16,420 | 8762                  | 4050   | 14,060 | 7743           | 3970   | 13,560 |
| Н                     | 1408         | 410    | 12,260 | 1559    | 270    | 15,090 | 1388                  | 410    | 11,830 | 1405           | 100    | 14,730 |
| Ν                     | 617          | 310    | 1090   | 696     | 270    | 1450   | 696                   | 340    | 1390   | 593            | 310    | 1150   |
| S                     | 737          | < 40.0 | 3100   | 535     | <40.0  | 1480   | 917                   | <40.0  | 3270   | 950            | <40.0  | 3670   |
| eBC                   | 3235         | 1206   | 5447   | 3746    | 1177   | 5245   | 3168                  | 1573   | 4720   | 2831           | 1306   | 4605   |
| TOC                   | 8925         | 3390   | 15,670 | 10,630  | 3220   | 16,020 | 9177                  | 4110   | 15,250 | 8180           | 4070   | 14,220 |

fluctuations, maximum  $PM_{10}$  levels were reported during March (Fig. 2B), which might be resulted from high cloudiness and thermal inversion episodes that mostly occur in the transition to the rainy season (SFigure 2, in SI).

# Elemental composition of PM<sub>10</sub> and source apportionment

Elements concentrations of  $PM_{10}$  collected at the four sampling sites (MED-1, MED-2, ITA-1, ITA-2), together with

eBC and TOC concentrations, are summarised in Table 1. Significant positive correlations were observed between  $PM_{10}$  mass and eBC, TOC and C concentrations (r>85%), which could point to a common source. On this background, carbon content accounted for 32.1% and 32.9% (for ITA-1 and ITA-2, respectively) of the  $PM_{10}$  material in all studied areas (Table 2). Similar results were reported in Kennedy site (Bogotá, Colombia), where the  $PM_{10}$  carbon content was associated to contribution of both medium and small industries and road traffic sources (Vargas et al. 2012). Fig. 2 Spatial (A) and temporal (B) variation of  $PM_{10}$  mass (µg m<sup>-3</sup>)



Among the elements, the highest concentrations were observed for Al, Ca, Mg and Na, which could be associated to a common PM crustal origin (high correlation observed for them (r > 86%, SFigure 3), being mostly derived from soil and dust resuspension (Khodeir et al. 2012; Zalakeviciute et al. 2020), whilst metals such as Li, Be and Cs were not quantitated in most samples (STable 3, in SI). Moreover, concentrations of carcinogenic metal(oid)s in  $PM_{10}$  ranged between 0.24–5.3 ng m<sup>-3</sup> (As), <0.17-2.0 ng m<sup>-3</sup> (Cd), <0.19-2.8 ng m<sup>-3</sup>(Co), <1.5-10.1 ng m<sup>-3</sup>(Ni) and 3.7–68.1 ng m<sup>-3</sup> (Pb) (Table 1), being all their averages below the annual values set by the Colombian government (5.0, 180 and 500 ng m<sup>-3</sup> for Cd, Ni and Pb, respectively) (RC-MADS 2017). Despite scarce studies concerning PM chemical characterisation in tropical sites have been reported, metal(oid)s levels found is this research were compared to studies conducted in other tropical areas. Results obtained were lower than those recently reported at Quito city (Ecuador) by Zalakeviciute et al. (Zalakeviciute et al. 2020) and similar to those reported in Bogotá city (Colombia) (Ramírez et al. 2018; Vargas et al. 2012).

As commented above, PMF model 5.0 was applied to estimate possible PM sources in the area. Exploratory tests were performed to set a factor number that would allow an acceptable confidence level (lower Q/Qexp) and identifying the different sources to be separated. For each trial, 200 runs were performed considering a randomly generated seed value. As is illustrated by SFigure 4, stabilisation of the residuals is mostly achieved by setting 6 factors, which were associated to different sources basing on PM elemental proportion (Reff et al. 2007). Results obtained are described in Table 2, whilst source profiles are shown in SFigure 5 (A-D), in SI. The fitness between PMF model and PM10 data was mainly successful ( $r^2$  values between 0.93 and 0.97), whilst those elements which did not show a good fit ( $r^2 < 0.5$ ) were excluded from the model. Elements were categorised according to their signal to noise (S/N) ratio (Brown et al. 2015; Paatero and Hopke 2003), considering strong variables those elements that showed a S/N ratio greater than or equal to five, whilst those elements whose S/N ratio was under detection limit was sorted as weak variables (STable 6, in SI).

**Table 2**  $PM_{10}$  sources identifiedfor each site, along with their

associated main constituents

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| Site  | Identified factors                                     | Contribution |                            |  |  |
|-------|--|--------------|----------------------------|--|--|
|       |  | (%)          | Main constituents          |  |  |
| MED-1 | 1. Re-suspended powder                                 | 7.5          | Na, Mg, Al, Ca, Sr, Ba, Bi |  |  |
|       | 2. Recovery of batteries and other nonferrous smelting | 9.9          | Pb                         |  |  |
|       | 3. Combustion  | 32.5         | H, C, TOC, eBC             |  |  |
|       | 4. Quarries (rocks and clays)                          | 21.1         | Cr, V, Mn, Fe, Co, Sb, K   |  |  |
|       | 5. Secondary aerosols/ions                             | 23.3         | S, N                       |  |  |
|       | 6. Tire wear   | 12.7         | Zn, As, Cd                 |  |  |
| MED-2 | 1. Re-suspended powder                                 | 6.8          | Na, Mg, Al, Ca, Sr, Ba     |  |  |
|       | 2. Quarries (rocks and clays)                          | 19.8         | Cr, Fe, Mn, Co             |  |  |
|       | 3. Combustion, mobile sources                          | 32.8         | eBC, C, TOC, N             |  |  |
|       | 4. Secondary aerosols/sulphates                        | 14.6         | S                          |  |  |
|       | 5. Recovery of batteries and other no ferrous smelting | 10.9         | Pb, Cu                     |  |  |
|       | 6. Tire wear   | 15.0         | Ni, Zn, As, Sn             |  |  |
| ITA-1 | 1. Recovery of batteries and other no ferrous smelting | 7.8          | H, Pb                      |  |  |
|       | 2. Tire wear   | 9.6          | Ni, Zn                     |  |  |
|       | 3. Re-suspended powder                                 | 7.4          | Na, Mg, Al, Sr, Ba, Bi     |  |  |
|       | 4. Secondary aerosols/sulphates                        | 13.2         | S                          |  |  |
|       | 5. Combustion, mobile sources                          | 32.1         | eBC, TOC, C, N             |  |  |
|       | 6. Quarries (rocks and clays)                          | 29.9         | Cr, V, Fe, Mn, Co, Cd      |  |  |
| ITA-2 | 1. Re-suspended powder                                 | 2.2          | Na, Mg, Al, Ca, Sr, Ba     |  |  |
|       | 2. Recovery of batteries and other no ferrous smelting | 16.6         | Pb, Ni                     |  |  |
|       | 3. Combustion  | 32.9         | eBC, TOC, C, N             |  |  |
|       | 4. Secondary aerosols/sulphates                        | 14.8         | S                          |  |  |
|       | 5. Tire wear   | 2.5          | Zn, Tl                     |  |  |
|       | 6. Quarries (rocks and clays)                          | 31.1         | Cr, V, Mn, Fe, Co, Cd, Sn  |  |  |
|       | 6. Quarries (rocks and clays)                          | 31.1         | Cr, V, Mn, Fe, Co, Cd, S   |  |  |

The contribution to combustion processes source was similar in all sampling sites (around 32%), being mainly attributed to vehicle traffic sources due to the high eBC, C and TOC contents. Then, all sites are categorised could be defined as urban, with a certain traffic and industrial activity influence. On this basis, the organic PM fraction could be attributed to smaller particles such as PM2 5, corresponding to approximately 58% of PM10 (SFigures 1 and 6, in SI), which agrees with studies conducted at urban industrial areas (Khodeir et al. 2012; Spandana et al. 2021; Sugimoto et al. 2016). Also, some metal(oid)s are frequently associated to anthropogenic PM sources such as road dust tracers (Al, Mn, K and Sr), fuel oil combustion (V) and burning waste or abrasive wear of tires (Zn) (Fauser et al. 2002); construction activities (Cr) (Watson and Chow 2015); and smelting of non-ferrous material, battery recycling and waste incineration (Pb) (Landis et al. 2017). As the use of Pb as an antiknock agent in gasoline has been banned in Colombia since 1991, Pb may be released to atmosphere as a result of local anthropogenic sources. Since Pb concentrations are not correlated with other elements, the highest concentrations of Pb observed in ITA-1 and ITA-2 sites (13.9 ng m<sup>-3</sup> and 17.8 ng m<sup>-3</sup>, respectively) could be associated with winds from the south and southwest

 $(1.4-1.6 \text{ m s}^{-1})$ , where industrial activity is significant. The great contribution of Co, Cr, Fe, K, Mn, Sb and V (linked to extraction of materials and stone) in ITA-1 and ITA-2 sites could be due to the proximity to quarries in operation located at municipality of Itagüí (Table 2). Sources attributed to secondary aerosols offered a less contribution (13.2 to 23.3%), with respect to sources associated with combustion processes (32.1 to 32.9%) (Table 2). $\delta^{13}$ C values were between - 25.0 and - 27.0% (STable 7, in SI), which support the predominant traffic emission in all the sites (Cao et al. 2011; Widory 2006). Results observed were similar to those reported at places such as Mexico City ( $\delta^{13}$ C: -26.3 to -24.3%) (López-Veneroni 2009), Paris ( $\delta^{13}$ C: -26.75to -25.75% ) and Tuscany ( $\delta^{13}C$ : -26.5 to -25.5% ) (Grassi et al. 2007). As is shown by Fig. 3,  $\delta^{13}$ C values found seem to be related to gasoline and diesel emissions rather than carbon emissions since values in commercial diesel samples were reported to be between -33.3 to -25.8% (Muhammad et al. 2015). In addition, results observed are in agreement with reports regarding consumption of diesel (about 110 mill gallons) and gasoline (about 190 mill gallons) during 2016 in AMVA (AMVA 2017b).

Stationary sources associated to the use of coal are not predominant in AMVA, accounting for 19% of  $PM_{2.5}$ 



**Fig. 3** Fuel source association for each sampling site, basing on  $\delta^{13}$ C (‰) analysis. (regular fuel), (idesel (Cao et al. 2011; Widory 2006)

emissions. In this regard,  $PM_{10}$  emissions from stationary sources were mainly associated with three industrial sectors: textile, beverage, food and tobacco and chemical (SIGAIRE 2022).

### PAHs analysis in PM<sub>10</sub>

PAHs concentrations (mean, median, minimum, maximum and SD) in PM<sub>10</sub> samples in each sites during the whole period are given in Table 3. The Kruskal–Wallis analysis showed no statistically significant differences (p < 0.05) between PAHs concentrations found in each sampling site (Fig. 4). High molecular weight PAHs (5 and 6 condensed rings) were predominant in all samples collected (Table 3 and SFigure 7, in SI), being BghiP, BbjF and IcdP the compounds that showed the highest concentration values (between 0.093–1.6 ng m<sup>-3</sup>, 0.069–2.7 ng m<sup>-3</sup> and <0.001–1.6 ng m<sup>-3</sup> for BghiP, BbjF and IcdP, respectively) (Table 3).

The sum of PAH concentrations ( $\Sigma$ PAHs) considering all sites ranged from 0.90 to 12.8 ng m<sup>-3</sup>, whilst mean  $\Sigma$ PAHs found in each site ranged between 4.1 (ITA-2) to 4.8 (MED-2)

**Table 3** Mean, minimum (Min) and maximum (Max) values of individual  $PM_{10}$ -bound PAHs, PAHs summation ( $\Sigma PAHs$ ) and equivalent BaP (BaP<sub>eq</sub>) concentrations (ng m<sup>-3</sup>) found in each sampling site

|        | MED-1 |         |       | MED-2 | D-2 ITA-1 |       |       | ]       |       | ITA-2 | ITA-2   |       |
|--------|-------|---------|-------|-------|-----------|-------|-------|---------|-------|-------|---------|-------|
|        | Mean  | Mín     | Max   | Mean  | Min       | Max   | Mean  | Min     | Max   | Mean  | Min     | Max   |
| NAP    | 0.17  | < 0.075 | 0.36  | 0.29  | < 0.075   | 2.0   | 0.20  | < 0.075 | 1.0   | 0.18  | < 0.075 | 1.0   |
| Me-NAP | 0.22  | < 0.039 | 0.68  | 0.43  | 0.082     | 3.4   | 0.26  | 0.088   | 1.6   | 0.68  | 0.040   | 7.5   |
| ACE    | 0.035 | < 0.003 | 0.10  | 0.052 | 0.013     | 0.34  | 0.026 | < 0.003 | 0.072 | 0.019 | < 0.003 | 0.042 |
| ACY    | 0.011 | < 0.005 | 0.076 | 0.032 | < 0.005   | 0.30  | 0.017 | < 0.005 | 0.083 | 0.015 | < 0.005 | 0.093 |
| FLU    | 0.037 | 0.010   | 0.12  | 0.083 | 0.006     | 0.69  | 0.055 | 0.012   | 0.16  | 0.045 | < 0.003 | 0.19  |
| Me-FLU | 0.024 | < 0.021 | 0.22  | 0.17  | < 0.021   | 1.2   | 0.051 | < 0.021 | 0.48  | 0.098 | < 0.021 | 0.52  |
| PHE    | 0.17  | 0.050   | 0.44  | 0.23  | 0.089     | 0.61  | 0.17  | 0.074   | 0.34  | 0.12  | 0.019   | 0.27  |
| ANT    | 0.051 | < 0.015 | 0.14  | 0.075 | < 0.015   | 0.38  | 0.056 | < 0.015 | 0.14  | 0.036 | < 0.015 | 0.078 |
| Me-ANT | 0.016 | < 0.004 | 0.082 | 0.026 | < 0.004   | 0.13  | 0.018 | 0.005   | 0.065 | 0.011 | < 0.004 | 0.045 |
| FLT    | 0.14  | 0.019   | 0.41  | 0.14  | 0.071     | 0.22  | 0.14  | 0.025   | 0.35  | 0.097 | 0.022   | 0.24  |
| PYR    | 0.22  | 0.027   | 0.59  | 0.20  | 0.098     | 0.33  | 0.19  | 0.054   | 0.45  | 0.13  | 0.036   | 0.33  |
| RET    | 0.013 | < 0.002 | 0.14  | 0.005 | < 0.002   | 0.072 | 0.041 | < 0.002 | 0.29  | 0.005 | < 0.002 | 0.072 |
| BaA    | 0.098 | < 0.001 | 0.37  | 0.078 | < 0.001   | 0.22  | 0.077 | < 0.001 | 0.25  | 0.028 | < 0.001 | 0.13  |
| TPY    | 0.10  | 0.033   | 0.24  | 0.086 | < 0.001   | 0.18  | 0.099 | 0.015   | 0.33  | 0.064 | < 0.001 | 0.14  |
| CHR    | 0.22  | 0.021   | 0.58  | 0.18  | 0.033     | 0.35  | 0.19  | 0.057   | 0.52  | 0.13  | 0.034   | 0.27  |
| BbjF   | 0.85  | 0.069   | 2.7   | 0.74  | 0.18      | 1.3   | 0.78  | 0.20    | 2.3   | 0.60  | 0.13    | 1.3   |
| BkF    | 0.21  | 0.016   | 0.74  | 0.17  | 0.058     | 0.27  | 0.22  | 0.051   | 0.85  | 0.16  | 0.040   | 0.31  |
| BeP    | 0.18  | 0.026   | 0.66  | 0.14  | 0.041     | 0.25  | 0.17  | 0.037   | 0.76  | 0.094 | 0.011   | 0.20  |
| BaP    | 0.32  | 0.040   | 0.95  | 0.27  | 0.082     | 0.46  | 0.28  | 0.082   | 0.78  | 0.25  | 0.067   | 0.54  |
| DahA   | 0.059 | < 0.002 | 0.33  | 0.016 | < 0.002   | 0.064 | 0.094 | < 0.002 | 0.52  | 0.013 | < 0.002 | 0.15  |
| IcdP   | 0.49  | < 0.010 | 1.2   | 0.58  | < 0.010   | 1.6   | 0.54  | < 0.010 | 1.3   | 0.47  | < 0.010 | 0.93  |
| BghiP  | 0.86  | 0.093   | 1.6   | 0.82  | 0.26      | 1.4   | 0.84  | 0.46    | 1.5   | 0.82  | 0.15    | 1.5   |
| ΣPAHs  | 4.5   | 0.900   | 12.4  | 4.8   | 2.5       | 10.4  | 4.5   | 2.0     | 10.4  | 4.1   | 1.4     | 12.8  |
| BaPeq  | 0.90  | 0.15    | 4.9   | 0.76  | 0.29      | 3.3   | 1.2   | 0.15    | 6.1   | 0.64  | 0.10    | 3.8   |



Fig. 4 Spatial variation of PAHs levels (ng  $m^{-3}$ )

ng m<sup>-3</sup> (Table 3). Previous studies in Colombia reported values among 1.49–8.55 ng m<sup>-3</sup> ( $\Sigma_{16}$ PAHs, from an exploratory study in AMVA) (Mueller et al. 2019). Also,  $\Sigma$ PAHs found in the present study were higher than those reported in European cities (Callén et al. 2011; Oliveira et al. 2017), but lower than those observed in China (Yin and Xu 2018).

Concerning PAHs contribution, low ring-number (2-3 rings) PAHs ( $\Sigma_{2-3rings}$ PAHs: NAP, Me-NAP, ACE, ACY, RET, FLU, Me-FLU, PHE, ANT and Me-ANT) accounted for 16–30% of  $\Sigma$ PAHs probably due to their volatility (being mainly part of atmospheric gaseous phase), whereas some PAHs such as DBT and other volatile PAHs (Me-FLU, ACY and RET) were found in concentrations < LOQs in most of PM<sub>10</sub> samples (STable 4, in SI). Middle ring number (4 rings) PAHs ( $\Sigma_{4rings}$ PAHs: FLT, PYR, BaA, TPY and CHR) accounted for 10-15%, whilst high ring-number molecules (S5-6ringsPAH), BeP, BbjF, BkF, BaP, IcdP, DBahA and BghiP were predominant in all sampling sites (accounting for 59-67%). Several studies associated emissions of low molecular weight PAHs ( $\leq 4$  rings) to diesel and heavy vehicles; being release of high molecular weight PAHs (5 and 6 rings) linked to emissions from light vehicles or gasoline engines, and considered the PAHs fraction that triggers the most adverse human health effects (Hwang et al. 2003; Liu et al. 2015). Also, the high levels found for 5-6 rings PAHs would support the predominant road traffic source observed in sampling sites.

Carcinogenic PAHs (BbF, CHR, IcdP, BaP, BkF, BaA, DahA) concentrations ( $\Sigma_c$ PAH) ranged between 1.8 ng m<sup>-3</sup> (ITA-2) and 1.2 ng m<sup>-3</sup> (ITA-1), representing 27–44% of the total PAHs levels. Although statistically significant spatial variations of  $\Sigma_c$ PAH concentrations were not found, monthly differences were observed; observing a low  $\Sigma_c$ PAH concentration during July (1.0 ng m<sup>-3</sup>), which could be due to an increasing dispersion of pollutants by local and mesoscale meteorological events during this period, as well as due to a minor vehicular traffic during mid-year holidays. Significant correlation (r > 70%) was observed between some heavy PAHs (BghiP, BbjF, BkF, CHR) and eBC levels, which

might be attributed to a strong adsorption affinity of PAHs in carbon particles (eBC) (Guo et al. 2021).

Analysis of molecular diagnostic ratios suggested that around 70% of the emissions of PM10-bound PAHs would be associated with combustion of liquid fuel (0.4 < FLT)(FLT + PYR < 0.5) and around 9% would be linked to biomass or diesel combustion (IcdP/IcdP + BghiP = 0.37)(Fig. 5A-B); being barely 1% associated to wood burning (RET/RET + CHR > 0.8) (Fig. 4C) and observing no association to coal combustion source (BaP/BghiP>0.9) (Fig. 5B). Finally, 92% of the emissions of PM<sub>10</sub>-bound PAHs would be linked to pyrogenic processes (ANT/ANT + PHE > 0.1)(Fig. 5D) (Park et al. 2011; Tobiszewski and Namieśnik 2012) (Fig. 5A–D). These results are in accordance with  $^{13}C$ data and the AMVA's emissions inventory for 2015 (AMVA 2017b), reporting that 79.8% of the circulating fleet used gasoline, 15.9% diesel and 3.8% natural gas; and being Medellin the municipality with the highest diesel fuel consumption (over 60 mill gallons/year in 2016) (AMVA 2017b). Therefore, mobile sources were estimated to be about 70% of the contribution to fine particulate matter (AMVA 2017b).

### Health risk assessment

Carcinogenic risks of  $1.0 \times 10^{-6}$  and  $1.0 \times 10^{-4}$  were set by USEPA to define human carcinogenic health risk thresholds (Davie-Martin et al. 2017; Sah et al. 2017; USEPA 2019): values higher than  $1.0 \times 10^{-4}$  would suggest significant carcinogenic adverse risks, whilst carcinogenic risks under  $1.0 \times 10^{-6}$  would suggest no significant carcinogenic risk.

# Life-time cancer risk for inhalation exposure to metal(oid)s

Estimated average life-time cancer risks (LCRs) were between  $2.8 \times 10^{-6}$  (MED-1) to  $4.2 \times 10^{-6}$  (ITA-1) (STable 8, in SI), being As and Co the metal(oid)s that mainly contributed to LCRs (SFigure 8A, in SI), suggesting that the risk would be associated mainly to industrial and mining activities rather than combustion processes. As can be seen from the table, LCRs exceeded the limit of  $1.0 \times 10^{-6}$  at several sites. Furthermore, no significant spatial variations were observed for LCRs (p > 0.05); however, LCRs estimated in June and July were lower (SFigure 8A-B, in SI). As commented above, this could be explained by the wind speed and temperature variation during those months, favouring pollutants dispersion in the area (SFigure 9-10, in SI). Maximum average wind speed values occur during the month of July, conditions that favour dilution of the pollutants emitted (SFigure 9, in SI). Also, the low average temperatures during March and October slow down the atmospheric dynamics causing a longer exposure to atmospheric pollutants (SFigure 10, in SI).



**Fig. 5** Cross-plots for selected PAHs diagnostic ratios, considering each sampling site: **A** FLT/(FLT + PYR) ratio vs. IcdP/(IcdP + BghiP) ratio, **B** FLT/(FLT + PYR) ratio vs. BaP/BghiP ratio, **C** Date vs. RET/

# Life-time cancer risk (LCR) for inhalation exposure to PAHs

PAHs concentrations were used to estimate the BaP equivalent concentrations (BaPeq), using the toxicity factors defined in STable 5 (in SI). The  $BaP_{eq}$  calculated ranged between 0.10 and 6.1 ng m<sup>-3</sup> (Table 3). Several exceedances of the BaP<sub>eq</sub>  $(BaP_{eq} \text{ concentrations} > 1.0 \text{ ng m}^{-3} \text{ set by current Colombian}$ legislation) were observed during the sampling period at all sampling sites (RC-MADS 2017) and the European Commission Guideline for Air Quality (EU 2004). Average LCRs for each sampling site were  $9.9 \times 10^{-7}$ ,  $8.4 \times 10^{-7}$ ,  $1.4 \times 10^{-6}$  and  $7.0 \times 10^{-7}$  at MED-1, MED-2, ITA-1 and ITA-2, respectively, whilst an average LCR value of  $9.8 \times 10^{-7}$  (Table 4) was estimated considering all the sites, suggesting that approximately 1 out of every 1,000,000 people could eventually develop cancer due to inhalation PM10-associated PAHs exposure. Although some exceedances of the acceptable risk limit of  $1.0 \times 10^{-6}$  (Table 4) were observed at several sites, the average risk estimated during the sampling period for all sites was smaller than the upper risk limit of  $1.0 \times 10^{-4}$  set by USEPA (Davie-Martin et al. 2017; USEPA 2019). Also, no statistically significant differences of LCRs between each site was



(RET+CHR) ratio, and **D** Date vs. ANT/(ANT+PHE) ratio (Park et al. 2011; Tobiszewski and Namieśnik 2012)

Table 4Lifetime cancer risks (LCRs) estimated for  $PM_{10}$ -boundPAHs exposure via inhalation

| Site  | n   | Mean                  | RSD% | Min                   | Max                   |
|-------|-----|-----------------------|------|-----------------------|-----------------------|
| MED-1 | 27  | $9.9 \times 10^{-07}$ | 1.1  | $1.6 \times 10^{-07}$ | $5.4 \times 10^{-06}$ |
| MED-2 | 25  | $8.4 \times 10^{-07}$ | 0.82 | $3.2 \times 10^{-07}$ | $3.7 \times 10^{-06}$ |
| ITA-1 | 27  | $1.4 \times 10^{-06}$ | 1.4  | $1.6 \times 10^{-07}$ | $6.7 \times 10^{-06}$ |
| ITA-2 | 25  | $7.0 \times 10^{-07}$ | 1.2  | $1.1 \times 10^{-07}$ | $4.2 \times 10^{-06}$ |
| Total | 104 | $9.8 \times 10^{-07}$ | 1.3  | $1.1 \times 10^{-07}$ | $6.7 \times 10^{-06}$ |

observed (p < 0.05). Nevertheless, high risks were estimated during March and May, which could be attributed to an unfavoured dispersion of pollutants due to meteorological events.

# Conclusions

In the present study, chemical composition (comprising elements, PAHs, eBC, TOC and  $\delta^{13}$ C carbon isotope ratios) were assessed in a total of 104 PM<sub>10</sub> samples collected from four sites located at a tropical narrow valley, where pollutants dispersion could be hindered by several meteorological events. Spatio-temporal variability in the area was studied, founding non-significant differences between the concentrations of pollutants within the different sampling sites, which would point that the atmospheric dynamics in the Aburrá Valley, at least in the southern zone, behaves like an atmospheric airshed. Concerning temporal fluctuations, pollutions peaks might be associated with rainy periods, observing the highest concentrations in March-May and lower values in July. Intraday or hourly studies would be recommended to evaluate the effect of valley shape and the influence of daily heating and cooling over metal(oid) s and PAHs concentrations. Besides, PMF results pointed that combustion and mining activities (quarries) were the main PM<sub>10</sub> source in the studied area. Mean concentration of toxic metal(oid)s (with potential risk even at low concentrations) such as As, Cd, Co, Ni, Pb and Sb were found below the values set by the Colombian regulation, whereas average BaP concentration  $(0.25-0.32 \text{ ng m}^{-3})$  did not exceed the limit value set by directive 2004/107/EC. Among PAHs, BbjF and BghiP were the most profuse in PM<sub>10</sub> samples, following by IcdP, BaP, BkF, Chry PYR and Me-NAP. Furthermore, 5-ring and 6-ring PAHs accounted for 59-67% of PAHs content, whilst 27-44% of total PAHs concentration was attributed to carcinogenic PAHs. Diagnostic ratios suggested a pyrogenic origin for PM<sub>10</sub>-associated PAHs in all sampling sites. Carcinogenic risks estimated for PM<sub>10</sub>-bound PAHs exposure via inhalation could be considered as moderate, whereas significant inhalation carcinogenic risk was observed for carcinogenic metal(oid)s exposure in the Aburrá valley during the sampling period.

On the basis of the results obtained, the present study would provide useful data to achieve a better understanding of  $PM_{10}$  exposure in valleys with limited atmospheric diffusion, where scarce information is available, as well as its health impact on people living in such areas. Also, further studies in the area would be of great interest so as to support suitable policies to decrease population's exposure.

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**Data availability** The datasets generated during and/or analysed during the current study are not publicly available but are available from the corresponding author on reasonable request.

#### Declarations

Ethics approval and consent to participate Not applicable.

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Competing interests The authors declare no competing interests.

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