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The Mg-Sb System

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Comments and Addenda

Addenda

The Al-Sb System

In Vol. 5, No. 5, page 462, Table 1, the space group for AlSb should be $F\overline{4}3m$, and the space group for (Sb) should be $R\overline{3}m$. In Table 2, the second entry in the fourth column and the first entry in the fifth column should be 50.0.

The Au-Ba System

Please add the following information to the evaluation in Vol. 7, No. 5, pages 336 and 337.

Bruzzone [70Bru] studied Au_2Ba by X-ray analysis. The compound was prepared from Au 99.99% and Ba 99.6% by fusion in an argon atmosphere at 1200 °C followed by slow cooling (6 days) to the temperature 600 °C. Au_2Ba has a hexagonal structure of the AlB_2 type with parameters a = 0.4804 and c = 0.4119 nm.

Cited Reference

70Bru: G. Bruzzone, "Some Intermetallic MX₂ Compounds Formed by Ca,Sr and Ba," Atti. Acad. Naz. Lincei, Cl. Sci. Fis. Mat. Natur., 48 235-241 (1970) in Italian. (Crys Structure; Experimental)

The Au-Fe System

In Vol. 5, No. 6, page 592, Table 1, the approximate composition range for (α Fe) is 98.6 to 100 at.% Fe.

The Au-Sn System

In Vol. 5, No. 5, page 492, Table 1, the space group for (αSn) should be Fd3m. The following Table 3 replaces Table 3 on page 494.

The Cu-Pb System

In Vol. 5, No. 5, page 506, Table 2, the composition at the eutectic is given in at.% Cu.

The Mg-Sb System

In Vol. 5, No. 6, page 580, Table 1, the space group for βMg_3Sb_2 should be Ia3, and the space group for αMg_3Sb_2 should be P3m1.

The Mg-Si System

In Vol. 5, No. 6, page 589, Eq 7, for the free energy of fusion of Mg_2Si , the coefficient of the $T \ln T$ term should have been multiplied by 3 for conversion from $J/g \cdot$ atom to J/mol (where the mol was defined as the formula unit, Mg_2Si). The correct coefficient is 194.61.

Equation 7 was not used in subsequent calculations.

Table 3 Special Points of the Assessed Au-Sn System

Reaction	Compositions of the respective phases, at.% Sn	e ————	Temperature, °C	Reaction type
L ⇄ (Au)	0		1064.43(a)	Freezing point
$L + (Au) \rightleftharpoons \zeta \dots \sim 23$	6.8	?	483	Peritectic
$(Au) + \zeta \rightleftharpoons \beta$?	?	9.1	?	Peritectoid
$\beta \rightleftharpoons (Au) + \zeta \dots 9.1$?	?	$\sim \! 250$	Eutectoid
$\zeta \rightleftharpoons (Au) + \zeta'$?	?	16.7	?	Eutectoid
$L \rightleftharpoons \zeta + \delta$	17.5	50	278	Eutectic
$\zeta \rightleftarrows \zeta'$	16.7		195	Congruent
$\ddot{\zeta} \rightleftarrows \ddot{\zeta}' + \delta$	16.7	50	190	Eutectoid
$L \rightleftharpoons \delta$	50		419.3	Congruent
$L + \delta \rightleftarrows \varepsilon$	50.5	66.7	309	Peritectic
$L + \varepsilon \rightleftharpoons \eta$	66.7	80	252	Peritectic
$L \rightleftharpoons \eta + (\beta Sn)$	80	99.8	217	Eutectic
$\eta \rightleftharpoons \varepsilon + (\beta Sn)$	66.7	$\sim \! 100$?	Eutectoid
ε , (βSn) , (αSn)	~100	~100	\sim 13.05	?
$L \rightleftharpoons (\beta Sn)$	100		231.9681(a)	Freezing point
$(\beta Sn) \rightleftarrows (\alpha Sn)$	100		13.05	Allotropic
(a) From [Melt].				