

Hydroxy-aromatic compound, process for the preparation thereof, and use of the compound

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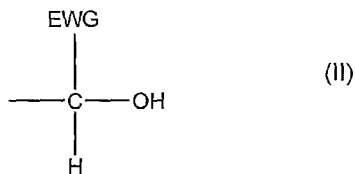
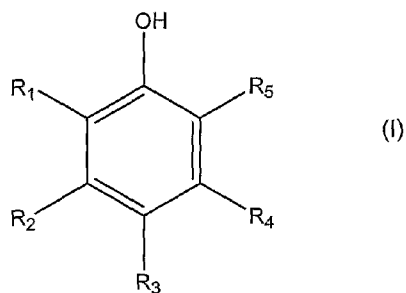
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(54) Title: HYDROXY-AROMATIC COMPOUND, PROCESS FOR THE PREPARATION THEREOF, AND USE OF THE COMPOUND



(57) Abstract: The invention relates to a hydroxy-aromatic compound of formula (I): Formula (I) wherein: at least one of the set consisting of R₁, R₃, and R₅ is a group of formula (II); any remaining one or two of the set consisting of R₁, R₃, and R₅ being H, OH, a C₁-C₁₂ alkyl group or an oligomeric or polymeric system; R₂ and R₄ are H, OH, a C₁-C₁₂ alkyl group or an oligomeric or polymeric system; formula (II) is the following group: formula (II) wherein EWG is an electron-withdrawing group.



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HYDROXY-AROMATIC COMPOUND, PROCESS FOR THE PREPARATION
THEREOF, AND USE OF THE COMPOUND

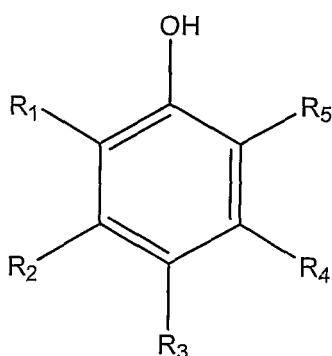
The invention relates to a hydroxy-aromatic compound.

5 Hydroxy-aromatic compounds as such are known, and are defined as compounds having an aromatic ring with at least one -OH group attached directly to it. An example of such a compound is phenol. Another example of such a compound is the known adduct of phenol and formaldehyde; this hydroxy-aromatic compound is used in the preparation of phenol-formaldehyde resins. These resins are known from
10 for example A. Knop, L.A. Pilato, Phenolic Resins, Springer Verlag Berlin 1990. These resins have many known uses, such as for example the use of these resins in adhesives for the preparation of particle boards.

A disadvantage of the known hydroxy-aromatic compounds, and in particular of their formaldehyde adducts, is that their use is associated with health risks,
15 relating to the emission of formaldehyde during resin preparation, resin curing and in end products.

It is the objective of the present invention to reduce or even eliminate the said disadvantage while still providing a compound suitable for the preparation of hydroxy-aromatic resins.

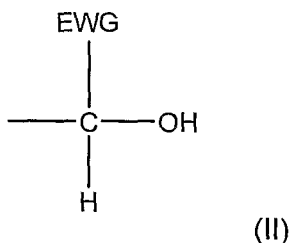
20 The objective is achieved in that the hydroxy-aromatic compound is a compound of formula (I)



wherein:

- 25
- at least one of the set consisting of R_1 , R_3 , and R_5 is a group of formula (II); any remaining one or two of the set consisting of R_1 , R_3 , and R_5 being H, OH, a C_1 - C_{12} alkyl group or an oligomeric or polymeric system;

- R_2 and R_4 are H, OH, a C_1 - C_{12} alkyl group, or an oligomeric or polymeric system;
- formula (II) is the following group:



wherein EWG is an electron-withdrawing group.

5 The advantage of the compound according to the invention is that hydroxy-aromatic-based resins can be made that suffer less, or even not at all, from the health risks associated with the use of formaldehyde. Thus, resins prepared with the compound according to the present invention are in particular suitable for use in many applications such as adhesives, coatings, laminates, and shaped articles.

10 As is known in hydroxy-aromatic chemistry, the positions on the aromatic ring adjacent to and opposite the hydroxy group (i.e., ortho and para) have a different reactivity than the remaining two meta-positions. In formula (I), therefore, the groups R_1 , R_3 , and R_5 should be regarded within a similar context and are herein referred to as a set.

15 In the compound according to the invention, at least one of the groups in the set consisting of R_1 , R_3 , and R_5 is given by formula (II); the other one or two groups in the said set - in case not all three of the said set is given by formula (II) - is/are H, OH, or C_1 - C_{12} alkyl group, preferably H, OH, a C_1 - C_9 alkyl group, or an oligomeric or polymeric system. If there are two groups not according to formula (II)

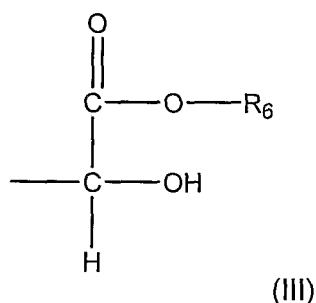
20 then they may be the same or may be different. The oligomeric or polymeric system may be a hydroxy-aromatic resin, either of the resol or of the novolac type, preferably of the novolac type; or it may be a different type of thermosetting or thermoplastic system. Some examples of how the set according to the invention can be shaped are: R_1 is a group according to formula (II), R_3 is H, and R_5 is H; R_1 is a group according to

25 formula (II), R_3 is H, and R_5 is CH_3 ; R_1 is H, R_3 is a group according to formula (II), and R_5 is H; R_1 and R_3 are a group according to formula (II), R_5 is H; R_1 , R_3 , and R_5 are all a group according to formula (II).

30 In the compound according to the invention, R_2 and R_4 are H, OH, a C_1 - C_{12} alkyl group, or an oligomeric or polymeric system; preferably R_2 and R_4 are H, OH or a C_1 - C_9 alkyl group. R_2 and R_4 may be the same or may be different. Some

preferred embodiments of R_2 and R_4 are: R_2 is OH and R_4 is H; R_2 is CH_3 and R_4 is H; R_2 is CH_3 and R_4 is CH_3 ; R_2 is H and R_4 is C_4H_9 . R_1 and R_2 may be part of a multicyclic compound; the same holds mutatis mutandis for R_2 and R_3 , R_3 and R_4 , or R_4 and R_5 .

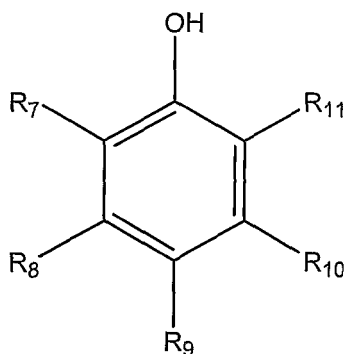
The group according to formula (II) is an integral part of the
 5 compound according to the invention; it is either R_1 , R_3 , or R_5 in formula (I), or two of those, or all three. In formula (II), EWG is an electron-withdrawing group. EWG's are as such known to the skilled person. Examples of an EWG are acid-, ester-, cyano-, di-alkylacetal-, aldehyde-, substituted phenyl-, or trihalomethyl groups. Hydrogen is not an EWG. In a preferred embodiment, the group of formula (II) is a group according to
 10 formula (III):



wherein R_6 is a C_1 - C_{12} alkyl group, aryl group, aralkyl group or cycloalkyl group. Preferably R_6 is a C_1 - C_{12} alkyl group; examples hereof are methyl, ethyl, propyl, butyl, pentyl, hexyl; more preferably, R_6 is a methyl group or an ethyl group.

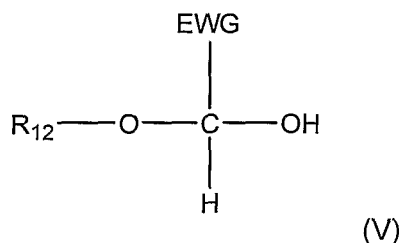
15 In a preferred embodiment of the hydroxy-aromatic compound according to the invention, at least one of the set consisting of R_1 , R_3 , and R_5 is H. This has the advantage that the hydroxy-aromatic compound is better suitable for the preparation of the oligomeric or polymeric structures typical for hydroxy-aromatic resins. In another preferred embodiment, two of the set consisting of R_1 , R_3 , and R_5 are
 20 H. This has the advantage that such a hydroxy-aromatic compound can be used to create three-dimensional networks, an ability often desired in hydroxy-aromatic resins. The same ability of the hydroxy-aromatic compound to create three-dimensional networks is present in those embodiments where all of R_1 , R_3 , and R_5 are either H or a group according to formula (II).

25 The invention further relates to a process for the preparation of the hydroxy-aromatic compound as described above. The process according to the invention comprises a reaction step wherein a hydroxy-aromatic compound of formula (IV) is brought into contact with a compound according to formula (V), optionally in the presence of a catalyst, whereby formula (IV) is:



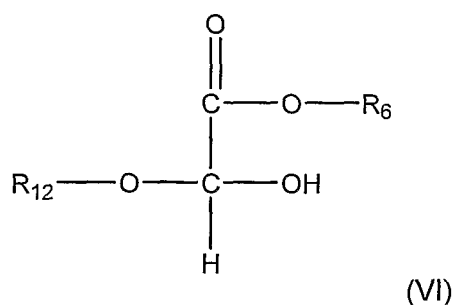
wherein R_7 , R_8 , R_9 , R_{10} and R_{11} are H, OH, a C_1 - C_{12} alkyl group or an oligomeric or polymeric system, whereby at least one and preferably two or even three of the set consisting of R_7 , R_9 , and R_{11} is or are H;

5 and formula (V) is:



wherein EWH is an electron-withdrawing group and wherein R_{12} is H, a C_1 - C_{12} alkyl group, aryl group, aralkyl group or cycloalkyl group.

10 Preferably, the compound according to formula (V) is an alcohol hemiacetal according to formula (VI):



wherein R_6 is a C_1 - C_{12} alkyl group, aryl group, aralkyl group or cycloalkyl group and wherein R_{12} is H, a C_1 - C_{12} alkyl group, aryl group, aralkyl group or cycloalkyl group. Preferably R_6 and R_{12} are C_1 - C_{12} alkyl groups. Examples thereof are methyl, ethyl, 15 propyl, butyl, pentyl, hexyl, and heptyl. R_6 and R_{12} are in particular a methyl group or an ethyl group.

The process according to the invention comprises a reaction step.

The purpose of the reaction step is to let a hydroxy-aromatic compound of formula (IV) react with a compound of formula (V). Thus these compounds must be brought together. The compound according to either formula (IV) or formula (V) may be one single compound or a mixture of two or more compounds falling within the scope of the formulas as defined above. Examples of preferred compounds according to formula (IV) are phenol, (2, 3, or 4-)cresol, resorcinol, (2, 3, or 4-)tert-butylphenol, (2, 3, or 4-)nonylphenol, (2,3- 2,4- 2,5- 2,6- or 3,4-)dimethylphenol, (2, 3, or 4-)ethylphenol, bisphenol A, bishenol F, and hydrochinon. Examples of compounds according to formula (V), in particular of the preferred alkanol hemiacetals according to formula (VI), are methylglyoxylate methanol hemiacetal (GMHATM, DSM Fine Chemicals, Linz); ethylglyoxylate ethanol hemiacetal (GEHATM, DSM Fine Chemicals, Linz); ethylglyoxylate methanol hemiacetal; butylglyoxylate butanol hemiacetal; butylglyoxylate methanol hemiacetal; butylglyoxylate ethanol hemiacetal; isopropylglyoxylate isopropanol hemiacetal; propylglyoxylate propanol hemiacetal; cyclohexylglyoxylate methanol hemiacetal and 2-ethylhexylglyoxylate methanol hemiacetal. Further examples of compounds according to formula V are glyoxylic acid hydrate, methylglyoxylate hydrate and ethylglyoxylate hydrate.

It may be beneficial to execute the reaction step according to the invention in a solvent or dispersant. As solvents, those compounds are suitable in which the reactants dissolve sufficiently to let the reaction take place. Examples of such solvents are water and various organic solvents. Depending on the specific compound or compounds of formula (IV) and (V), it may well be possible to use one or more of the reactants as solvent; in such a case, it can be possible to forego on the use of a solvent that is essentially a non-reactant and to execute the reaction step in bulk. In particular, many of the compounds according to formula (V) and in particular according to formula (VI) are a liquid at temperatures between 10°C and 100°C and can act as dispersant/solvent as well as reactant.

Although the reaction step may proceed spontaneously once the respective compounds have been brought together, it may be useful to bring the compounds together in the presence of a catalyst in order to accelerate the reaction. As catalyst, preferably an acid or a base is used; in particular, a Lewis or a Brønsted type of acid is preferred - such as for example sulphuric acid - whereby the pH is reduced to between 0 and 5, preferably to between 1 and 4, in particular to between 2 and 3. Suitable examples of acid catalysts are sulphuric acid, nitric acid, hydrochloric acid, phosphoric acid, boric acid, tetrafluoroboric acid, paratoluene sulphonic acid,

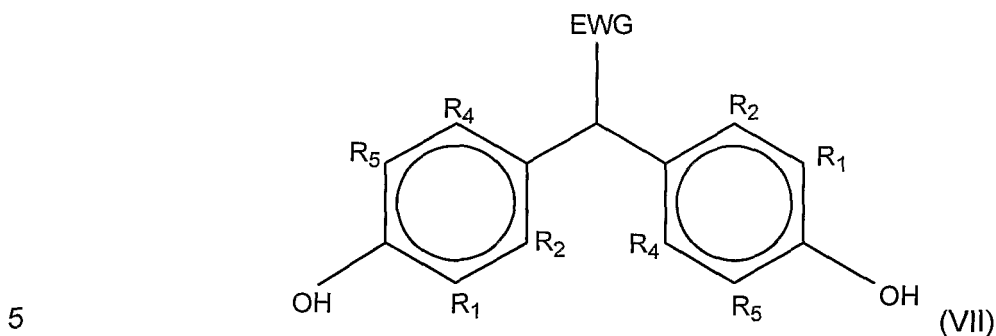
formic acid, ammonium sulphate, ammonium chloride, ammonium nitrate. Suitable examples of basic catalysts are ammonia, trimethyl amine, triethyl amine, DABCO (diazabicyclo-octane), DBU (diazabicyclo-undecene), DMAP (4-dimethylaminopyridine), sodium hydroxide, potassium hydroxide.

5 The temperature in the reaction step of present process can vary within wide limits, and preferably lies between 10°C and 100°C. More preferably the process is carried out at between 40°C and 90°C. The pressure in the present process preferably is between 0.005 MPa and 1.0 MPa, preferably between 0.02 MPa and 0.2 MPa; most preferably, the pressure is atmospheric.

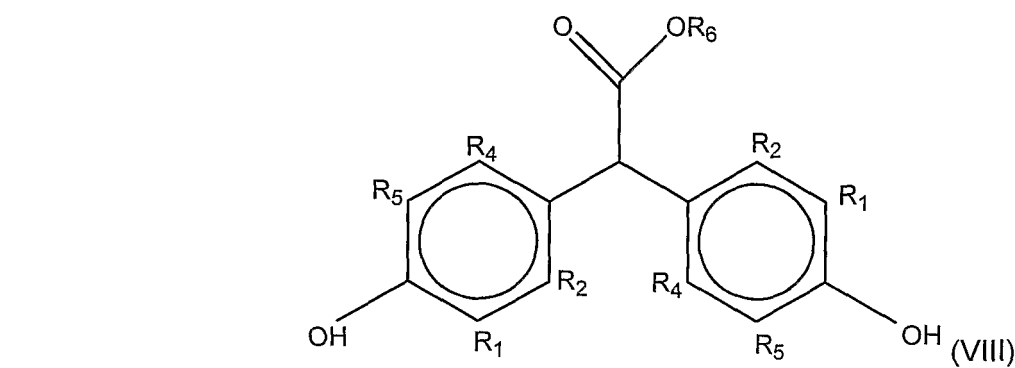
10 As consequence of the reaction step, a compound according to formula (I) is formed; additionally, the compound $R_{12}OH$ is released as by-product. It may be desirable to isolate the compound according to formula (I); this may be achieved through techniques that are as such known, such as for example a combination of pH change, solvent exchange, evaporation and/or precipitation. If the
15 compound according to formula (I) is not isolated, it may still be desirable to remove $R_{12}OH$; this may be achieved through techniques that are as such known, such as for example distillation. It may, however, also be acceptable or even desirable to let $R_{12}OH$ remain in the presence of the compound according to formula (I).

 In the process for the preparation of the hydroxy-aromatic compound
20 according to the invention, the molar ratio between the EWG-containing compound according to formula (V) (E) and the hydroxy-aromatic compound according to formula (IV) (H), herein referred to as E/H ratio, may vary between wide limits. Preferably, the E/H ratio lies between about 0.1 and about 10, more preferably between about 0.5 and about 3. If the E/H ratio is about 0.5 or lower, the resulting hydroxy-aromatic compound
25 according to the invention can be a mixture having a significant amount of a compound according to formula (I) in which one of the set consisting of R_1 , R_3 , and R_5 is a group of formula (II). If the E/H ratio is about 3 or higher, the resulting hydroxy-aromatic compound according to the invention can be a mixture having a significant amount of a compound according to formula (I) in which all three of the set consisting of R_1 , R_3 , and
30 R_5 are a group of formula (II). If the E/H ratio is about 1 or 2, the resulting hydroxy-aromatic compound according to the invention can be a mixture in which compounds according to formula (I) in which one, two or all three of the set consisting of R_1 , R_3 , and R_5 are a group of formula (II) are all clearly represented.

When executing the reaction step as described above, it was found that a further reaction can also be made to take place, namely the formation of a compound according to formula VII:



In case the EWG is according to formula (VI), the compound according to (VII) will be as in formula (VIII):



15 It was found that when executing the reaction step according to the invention, many hydroxy-aromatic compounds have a preference to first react on the para location of the aromatic moiety before doing so on the ortho location; hence the creation of compounds according to formula (VII) or (VIII). The present invention therefore also relates to compounds of formula (VII), in particular of formula (VIII), most preferably with R₁, R₂, R₄ and R₅ being all H and R₆ being methyl.

The compounds according to formula (VII) and (VIII) can typically be made by prolonged execution of the reaction step as described above for the preparation of compounds according to formula (I), whereby the E/H molar ratio preferably lies between 0.3 and 0.7, more preferably between 0.4 and 0.6.

20 The invention further relates to a process for the preparation of a hydroxy-aromatic resin. Such processes are as such known and comprise condensation reactions between a hydroxy-aromatic compound and a compound such

as an aldehyde, and typically also subsequent condensation reactions; an example of such a process is the process for preparation of a phenol-formaldehyde resin. In the process according to the invention, a compound according to formula (I) is used in the (subsequent) condensation reactions. The (subsequent) condensation reactions may
5 be executed in the same fashion and under similar conditions as described above for the preparation of the compound according to formula (I), (VII) and (VIII), although typically for a - further - prolonged period of time. The compound falling within the scope of formula (V) and in particular formula (VI) may be - aside from the hydroxy-aromatic compound according to formula (I) and/or the already formed oligomeric or
10 polymeric structures - the sole other compound participating in the condensation reactions in the resin; it may also be possible to use other compounds such as aldehydes like formaldehyde or furfural ($C_5H_4O_2$) in combination with the compound according to formula (V). Preferably, however, at least 5 or 10 mol.% of the compounds participating in the condensation reactions with a hydroxy-aromatic moiety in the resin
15 are one or more compounds according to formula (V); more preferably, this is at least 20 or 30%; in particular, this is at least 40 or 50%; with strong preference, at least 60 or 70 mol.% of the compounds reacting with a hydroxy-aromatic moiety in the resin are one or more compounds according to formula (V); most preferably, this is at least 80 or 90% or even essentially 100%.

20 The resin comprises hydroxy-aromatic moieties (H) derived from hydroxy-aromatic compounds used as starting materials. The resin also comprises EWG-derived moieties and possibly aldehyde-derived moieties, together referred to as A. The resin thus has a molar A/H ratio. The molar A/H ratio in the resin preferably lies between 0.5 and 3, more preferably between 0.75 and 2. If the molar A/H ratio lies
25 above 1, resol-type of resins can be formed whereby reactive 'A'-derived hydroxy groups are available. If the molar A/H ratio lies below 1, novolac-type of resins can be formed, in which essentially all 'A'-derived hydroxyl functionality has reacted away to form C-C and C-O ether bonds.

30 Resulting from the process as described above, the invention also relates to hydroxy-aromatic resins thus obtainable.

According to an embodiment of the invention, a hydroxy-aromatic resin can be prepared directly from raw materials comprising a compound according to formula (IV) as hydroxy-aromatic compound, and a compound according to formula (V). The conditions for achieving this are similar to those given above for the process or
35 preparing the compound according to formula (I), and can be established by the skilled

person via simple routine experimentation and using also his knowledge of the preparation of phenol-formaldehyde resins.

The invention moreover relates to the use of the hydroxy-aromatic aldehyde resin according to the invention for the preparation of coatings or shaped
5 articles such as wood-based panels like particle boards and laminates, or mineral wool such as stone wool or glass wool. To this end, the resins may be used by methods and under conditions similar to those known per se from the use of known hydroxy-aromatic aldehyde resins like phenol-formaldehyde resins. A catalyst and other additives may be added to the resin before the resin is used for processing in its final application.

10 Examples of customary additives are mould release agents, antistatic agents, adhesion promoters, plasticizers, colour enhancing agents, flame retardants, fillers, flow promoters, colorants, diluents, polymerization initiators, UV-stabilizers and heat stabilizers. Examples of fillers are glass fibres, mica, carbon fibres, metal fibres, clay, aramide fibres and strong polyethylene fibres.

15 The resin according to the invention may be used as such; however, it is also possible to subject the resin to a modification step; this is a reaction step designed to alter or enhance its functionality in a specific way. An example of an altered functionality is the solubility of the resin in water. An example of an enhanced functionality is the addition of a reactive group. An example of a modification step is to
20 bring the resin in contact with compounds that react with the -OH groups; an example of such a compound is epichlorohydrin. Another example of a modification step is to bring the resin in contact with compounds that react with the -OR₆ groups; an example of such a compound is water; the hydrolysis of the -OR₆ group into a -COOH group increases the solubility of the resin in water. Also, the modification step may be
25 achieved through a transesterification reaction between the -OR₆ groups and suitable compounds such as amines.

The present invention can be carried out in a variety of ways, as illustrated by the following non-limiting illustrative embodiments.

In a first illustrative embodiment, phenol and formaldehyde are used
30 as raw materials to prepare - in a known fashion - a novolac-type of resin at a pH of 2. The phenol-formaldehyde novolac resin is then brought together with methylglyoxylate methanol hemiacetal (GMHA); this is done in an aqueous environment, at a temperature of about 80°C and with an acidic catalyst at a pH of 2. The reaction is continued for 60 - 120 minutes, methanol being released and a resin according to the
35 invention is formed. The resin according to the invention is then either used as

adhesive in the production of a particle board or in the preparation of mineral wool, preferably with an acidic catalyst such as sulphuric acid added to the resin. For the production of particle board, dosing, press factors and other circumstances are normal (eg a temperature of about 140°C). The resulting particle board will have a greatly
5 reduced emission of formaldehyde compared to a particle board prepared by using only a standard phenol-formaldehyde resin. For the production of mineral wool such as stone wool, the resin is mixed with water so that a mixture having about 95-98 wt.% water content is formed, optionally aided by adding a base such as a metal hydroxide or amine or ammonia. This mixture is then sprayed on stone wool as it is just formed
10 and has a temperature of about 1900°C. The mixture acts as coolant, and at the same time the resin is partially cured thereby substantially forming the end product; this all takes place in about 1 to 3 seconds. In an after-treatment with hot air of about 140 - 200°C a fully cured product is obtained.

In a second illustrative embodiment, phenol and GMHA are brought
15 together in an aqueous solution. The molar ratio between phenol and GMHA is 2:3. At an elevated temperature (about 80°C) and at pH of 2 or lower, a resol-type of resin will be formed in about 60 minutes. Subsequently, the temperature is raised to about 150°C and the mixture stirred at that temperature for about 3 hours to yield a stable resin. Afterwards, the resin is brought ready for use by adding an additional acid
20 catalyst such as sulphuric acid. The resin is then used to prepare stone wool in the same fashion as in the previous illustrative embodiment.

In a third illustrative embodiment, cresol and GMHA are brought together in an aqueous solution. The molar ratio between cresol and GMHA is 4:3. Under conditions of elevated temperature (90°C) and reduced pH (2), a novolac-type of
25 resin will be formed. This resin can be converted into a water-born coating by adding a hardener such as adipic acid dihydrazide (ADH). If ADH is used as hardener, the coating will cure quickly (within 30 minutes) and at room temperature.

In a fourth illustrative embodiment, cresol and GMHA are brought together in an aqueous solution. The molar ratio between cresol and GMHA is 4:3.
30 Under conditions of elevated temperature (90°C) and reduced pH (2), a novolac-type of resin will be formed. The ester functionalities of the resin are then optionally saponified by raising the pH to 10. Subsequently, a reduction of pH to about 2 and evaporation of the water will yield a acid-functional solid resin. This resin is then mixed with an epoxy resin such as Epikote™ 828 (a liquid epoxy resin produced from bisphenol A and
35 epichlorohydrin). The resulting mixture is then moulded and pressed to cure in the form

of a shaped article.

The present invention is illustrated with the following example, without being limited thereto, aided with Figures 1 and 2.

In the Figures, Figure 1 and 2 represent the results of NMR
5 measurements as referred to in Example 1.

Example 1

Under stirring, 313.7 gram of methylglyoxylate methanol hemiacetal
10 (GMHA; 90 wt.% purity, 3 mol) and 190.1 gram of phenol (100% purity, 2 mol) were added to a 1 liter vessel at 23°C and atmospheric pressure. The vessel was brought under nitrogen atmosphere to prevent phenolic-type discolorations as much as possible. The solid phenol dissolved in the liquid GMHA. Under stirring 5 ml of sulfuric acid of 85 wt.% was added drop-wise to the reaction mixture. The clear reaction
15 mixture was heated to 100°C, and kept at 100°C for 3 hours. During this reaction time, the viscosity of the resin (measured at room temperature) increased from 1 Pa.s. to 330 Pa.s. The reaction was then ended by cooling down the resin to room temperature. An NMR scan of the resin so obtained - dissolved in DMSO - is given in Figure 1. Attention is drawn to the presence of peaks 3 and 6, evidencing a strong presence of
20 para-only substituted phenol moieties, indicative of presence and/or resin-forming from compounds according to formula (VIII).

With the resin so obtained, two further tests were carried out: a reactivity test and a curing/solubility test.

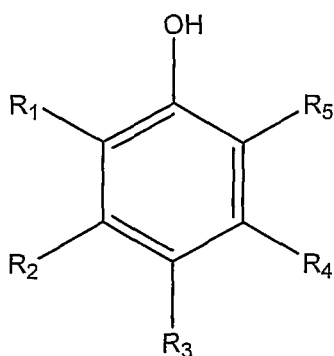
In the reactivity test, a small amount of resin was put into a vessel
25 and heated to 130°C. No catalysts or other compounds were added. A stirring rod was stuck into the resin, and with it the resin was continuously stirred until this was no longer possible due to the resin having reacted to such a stage that it became a lump / gel; this moment occurred within 10 minutes. The now partially cured resin was dissolved - with difficulty - in DMSO and an NMR spectrum was determined, see Figure
30 2. In comparison with Figure 1, peaks 3 and 6 have diminished or even disappeared. This means that para-only substituted phenol moieties were virtually absent; all these phenol moieties are still there but have also reacted on at least one ortho location.

In the solubility test, the resin was brought to 200°C and kept at that temperature for 2 hours. No catalysts or other compounds were added. After 2 hours,
35 the resin had cured into a hard, glassy material - this is indicative of significant to full

curing, further confirmed by the fact that it was no longer possible to dissolve the material in DMSO. After cooling down the cured resin was crushed to small particles. Then, a mixture was made consisting of 5 grams of the crushed resin and 95 grams of water. This mixture was heated to 80°C and kept at that temperature for 3 hours. After
5 cooling down, the cured-resin particles were filtered from the mixture, dried at 120°C and weighed. 96 wt.% or 4.80 grams of the resin was recovered.

CLAIMS

1. Hydroxy-aromatic compound of formula (I):



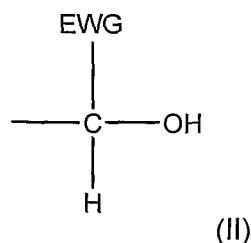
5

(I)

wherein:

- at least one of the set consisting of R₁, R₃, and R₅ is a group of formula (II); any remaining one or two of the set consisting of R₁, R₃, and R₅ being H, OH, a C₁-C₁₂ alkyl group or an oligomeric or polymeric system;
- R₂ and R₄ are H, OH, a C₁-C₁₂ alkyl group or an oligomeric or polymeric system;
- formula (II) is the following group:

10

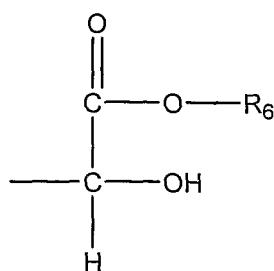


(II)

wherein EWG is an electron-withdrawing group.

15 2.

Hydroxy-aromatic compound according to claim 1, wherein formula (II) is a group according to formula (III):

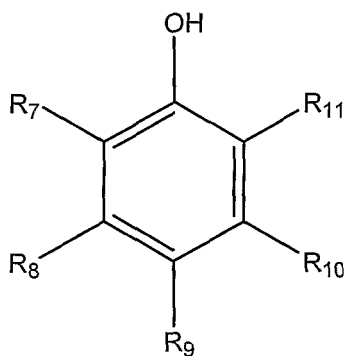


(III)

wherein R₆ is a C₁-C₁₂ alkyl group, aryl group, aralkyl group or cycloalkyl

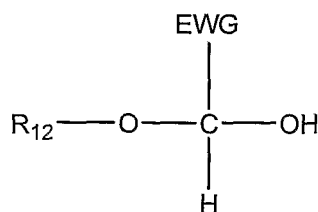
group.

3. Hydroxy-aromatic compound according to claim 1 or 2, wherein one or two of the set consisting of R_1 , R_3 , and R_5 is/are a group of formula (II) and wherein at least one of the said remaining one or two of the set consisting of R_1 , R_3 , and R_5 is H.
4. Process for the preparation of a hydroxy-aromatic compound according any one of claims 1-3, comprising a reaction step wherein a hydroxy-aromatic compound of formula (IV) is brought into contact with a compound according to formula (V), optionally in the presence of a catalyst, whereby formula (IV) is:



(IV)

wherein R_7 , R_8 , R_9 , R_{10} and R_{11} are H, OH, C_1 - C_{12} alkyl group, or an oligomeric or polymeric system, whereby at least one of the set consisting of R_7 , R_9 , and R_{11} is H;

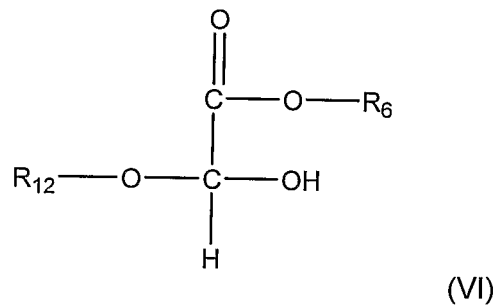


and formula (V) is:

(V)

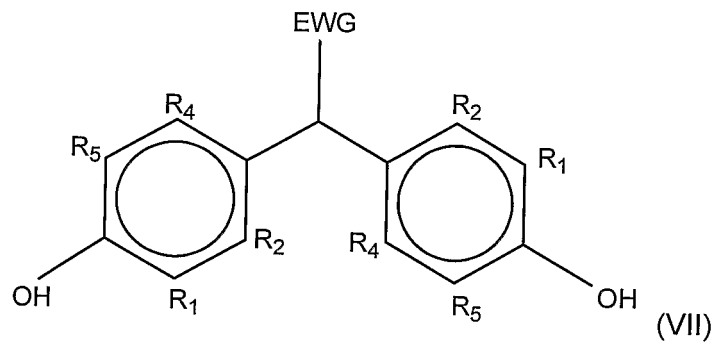
wherein R_{12} is a C_1 - C_{12} alkyl group, aryl group, aralkyl group or cycloalkyl group.

5. Process according to claim 4, wherein the compound according to formula (V) is an alkanol hemiacetal according to formula (VI):



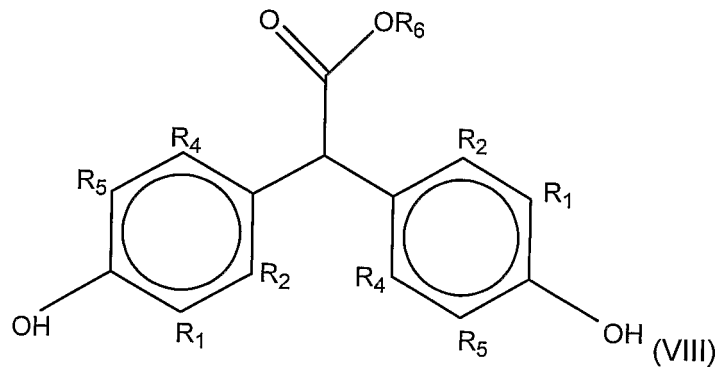
wherein R_6 and R_{12} are as defined above.

6. Hydroxy-aromatic compound of formula (VII)



- 5 wherein R_1 , R_2 , R_4 , R_5 and EWG are as defined above.

7. Hydroxy-aromatic compound of formula (VIII)



wherein R_1 , R_2 , R_4 , R_5 and R_6 are as defined above

8. Process according to claim 4 or 5, wherein an acid is used as catalyst.
- 10 9. Process for the preparation of a hydroxy-aromatic resin, characterized in that a compound according to any one of claims 1, 2, 3, 6 or 7 is used as a raw material.
10. Process for the preparation of a hydroxy-aromatic resin, characterized in that the raw materials comprise a compound according to formula (IV) as hydroxy-aromatic compound, and a compound according to formula (V).
- 15 11. Hydroxy-aromatic resin, obtainable by the process of claim 9 or 10.
12. Use of the hydroxy-aromatic resin according to claim 11 in coatings.

13. Use of the hydroxy-aromatic resin according to claim 11 in the manufacture of shaped articles, in particular wood-based panels or mineral wool.

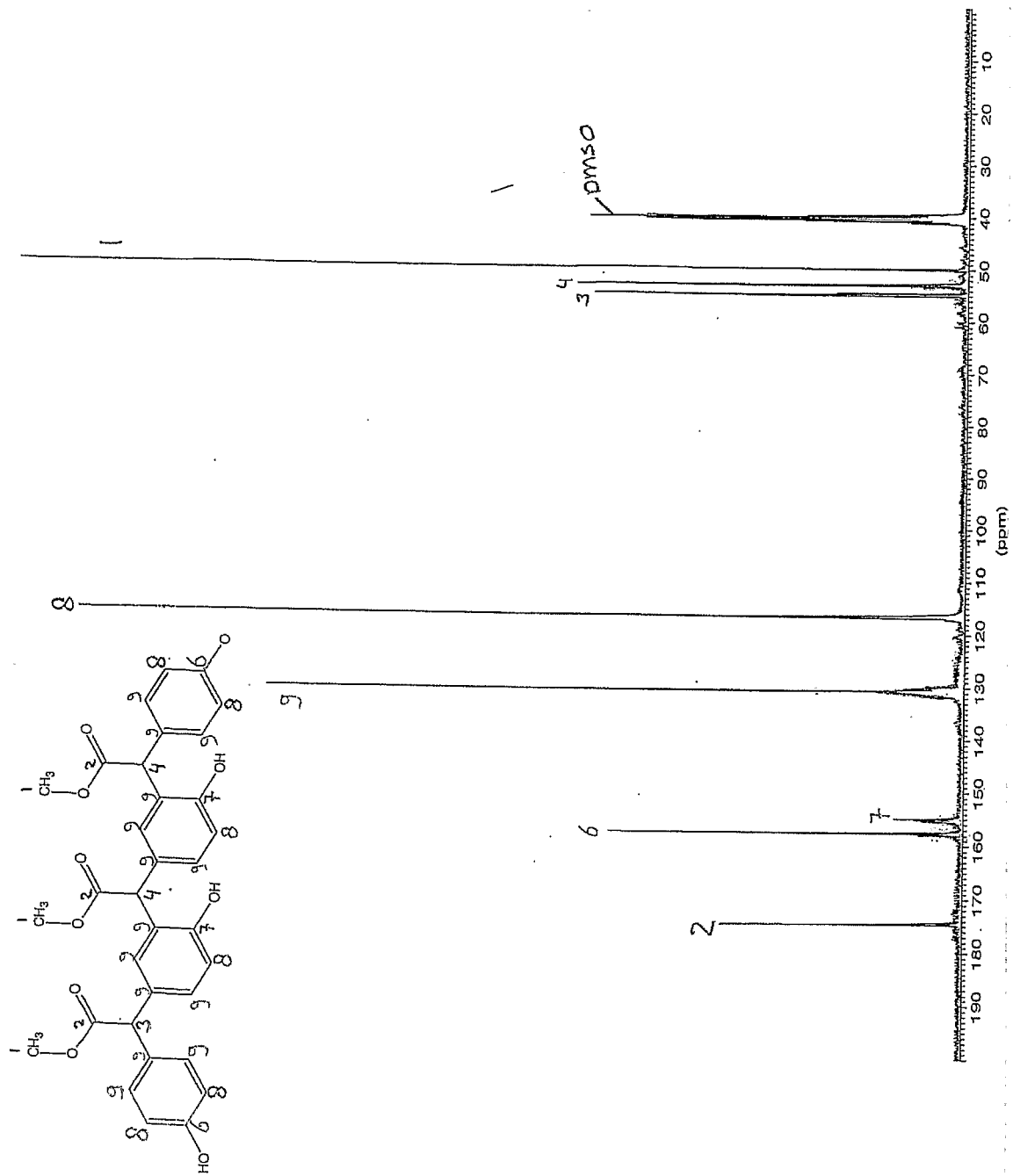


Fig. 1

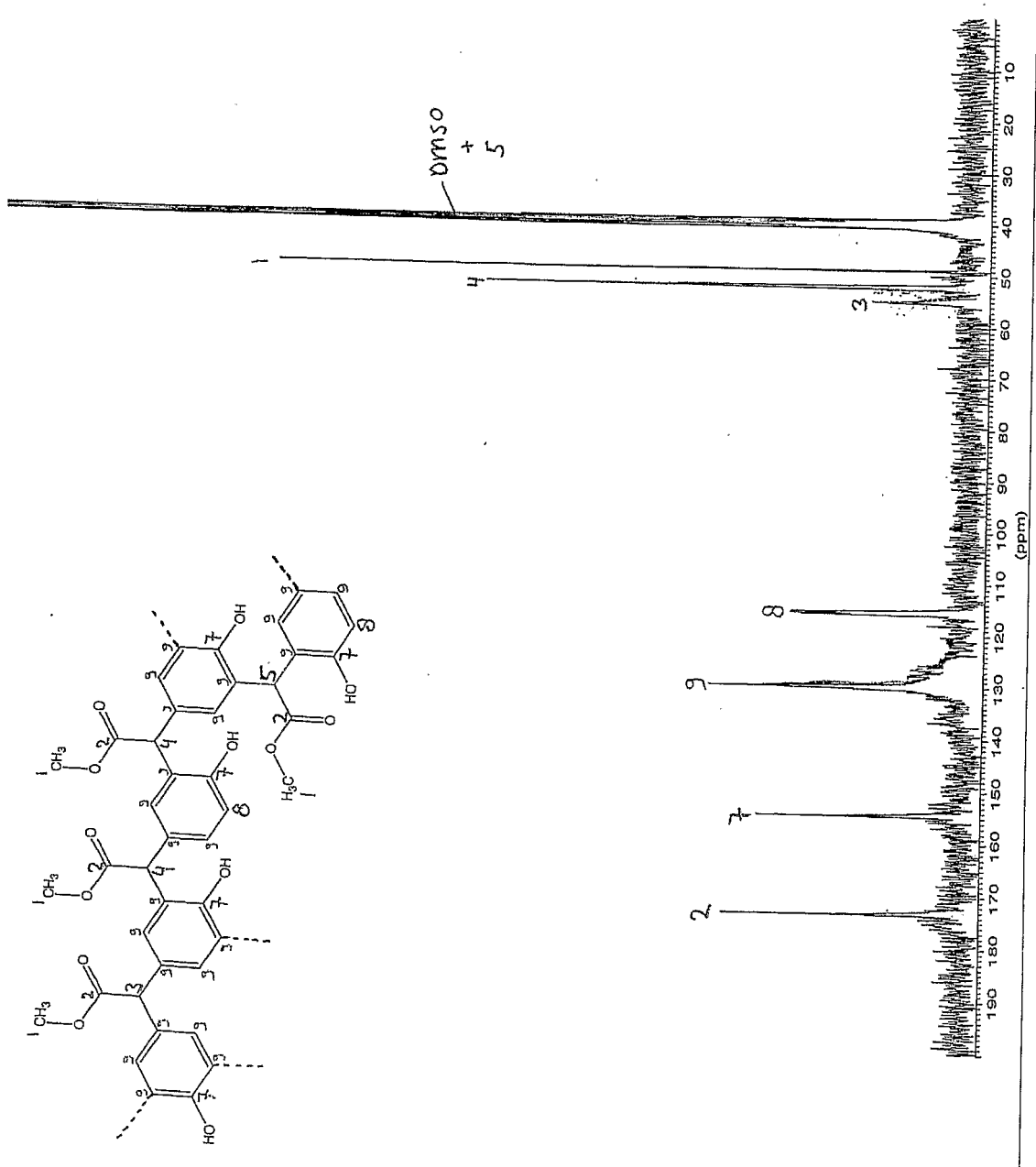


Fig. 2

INTERNATIONAL SEARCH REPORT

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A. CLASSIFICATION OF SUBJECT MATTER
 C08G14/04 C08G14/12 C07C69/732

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
 C08G C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
 EPO-Internal, PAJ, WPI Data, CHEM ABS Data, BEILSTEIN Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP 05 197151 A (JAPAN SYNTHETIC RUBBER CO LTD) 6 August 1993 (1993-08-06) paragraphs '0083! - '0102!; claim 1 -----	6,9-13
X	FR 1 579 608 A (E. I. DU PONT DE NEMOURS AND COMPANY) 29 August 1969 (1969-08-29) example D -----	6
X	FR 4 823 M (M. JEAN BAILENGER) 13 February 1967 (1967-02-13) the whole document -----	6
X	FR 6 857 M (M. JEAN BAILENGER) 8 April 1969 (1969-04-08) the whole document -----	6
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Further documents are listed in the continuation of Box C. See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E" earlier document but published on or after the international filing date	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
"O" document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search 15 February 2006	Date of mailing of the international search report 03/03/2006
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Cooper, S
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INTERNATIONAL SEARCH REPORT

International application No
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C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP 01 203343 A (SUMITOMO CHEM CO LTD) 16 August 1989 (1989-08-16) Compound (a) on p.288 -----	6
X	US 4 465 765 A (LEPPARD ET AL) 14 August 1984 (1984-08-14) Compounds 16-23 in column 7; claim 1 -----	6
X	US 4 888 444 A (RAVICHANDRAN ET AL) 19 December 1989 (1989-12-19) claim 1; example II -----	6
X	US 4 007 282 A (MAUZ ET AL) 8 February 1977 (1977-02-08) Compounds 1-5,8 mentioned in column 6; compounds 9-12,32-35 mentioned in column 7;claim 1; examples 3-5,20 -----	6,7
X	US 4 229 382 A (MAYER ET AL) 21 October 1980 (1980-10-21) claim 1; examples 6,18-21 -----	6
X	DE 30 29 026 A1 (HOECHST AG) 18 February 1982 (1982-02-18) claims 1-6; examples 1,4-7 -----	6,9-13
X	JP 04 274243 A (JAPAN SYNTHETIC RUBBER CO LTD) 30 September 1992 (1992-09-30) Compounds 6-1 to 6-28; claim 1 -----	6
X	PATENT ABSTRACTS OF JAPAN vol. 018, no. 511 (C-1253), 27 September 1994 (1994-09-27) & JP 06 172257 A (NIPPON KAYAKU CO LTD), 21 June 1994 (1994-06-21) abstract -----	6
X	JP 08 211600 A (SUMITOMO CHEM CO LTD) 20 August 1996 (1996-08-20) paragraphs '0014!', '0015!', '0036!', '0038!; claim 1 -----	6,7
X	US 4 178 466 A (GARDNER, RUSSELL A ET AL) 11 December 1979 (1979-12-11) claim 1; examples 1,2,4 -----	6
X	US 4 837 293 A (SILVIS ET AL) 6 June 1989 (1989-06-06) column 5, line 48 - column 5, line 65; claim 4; example 1 -----	6
X	US 3 639 447 A (GEORGE WRIGHT TAYLOR ET AL) 1 February 1972 (1972-02-01) claim 1; examples 1,2 -----	6
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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 00/55123 A (VANTICO AG) 21 September 2000 (2000-09-21) Formula (A) on p.4, preparations of compounds 11-14,18,20 starting on p.14;examples 1,2	6
X	US 4 358 624 A (MARK ET AL) 9 November 1982 (1982-11-09) claim 11; examples 8,9	6
X	US 3 450 746 A (CHRISTIAN STAPFER) 17 June 1969 (1969-06-17) claims 1-6; examples 1-4	6
X	US 4 365 098 A (MARK ET AL) 21 December 1982 (1982-12-21) the whole document	6
X	YUEFA GONG ET AL: "Regioselective Substitution of Phenols with Trifluoroacetaldehyde Ethyl Hemiacetal" BULL. CHEM. SOC. JPN., vol. 74, 2001, pages 377-383, XP009059286 Compound 14	1,3,4,6, 8
X	PATENT ABSTRACTS OF JAPAN vol. 012, no. 467 (C-550), 7 December 1988 (1988-12-07) & JP 63 188647 A (NIPPON SYNTHETIC CHEM IND CO LTD:THE), 4 August 1988 (1988-08-04) abstract	6
X	DATABASE CAPLUS 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; TAKASHIMA, MASANOBU ET AL: "Recording material" XP002362757 retrieved from STN Database accession no. 1987:587584 abstract & JP 62 060686 A2 (FUJI PHOTO FILM CO., LTD., JAPAN) 17 March 1987 (1987-03-17)	6
X	DATABASE CAPLUS 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; IWAKURA, KEN ET AL: "Selective esterification of alcoholic hydroxy group-containing phenols" XP002362758 retrieved from STN Database accession no. 1993:124206 abstract	6

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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>& JP 04 257543 A2 (FUJI PHOTO FILM CO., LTD., JAPAN) 11 September 1992 (1992-09-11)</p> <p>-----</p> <p>DATABASE CAPLUS 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; ITONAGA, YASUHIRO ET AL: "Preparation of tetrakisphenols from phenols or benzenediols via bisphenol monoaldehydes" XP002362759 retrieved from STN Database accession no. 2004:823198 abstract</p> <p>& JP 2004 277358 A2 (ASAHI ORGANIC CHEMICALS INDUSTRY CO., LTD., JAPAN) 7 October 2004 (2004-10-07)</p> <p>-----</p>	6
X	<p>DATABASE CAPLUS 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; BRZOZOWSKI, ZBIGNIEW ET AL: "Chlorobis(phenols)" XP002362760 retrieved from STN Database accession no. 1979:103611 abstract</p> <p>& PL 86 401 P (POLITECHNIKA WARSZAWSKA, POL.) 31 May 1976 (1976-05-31)</p> <p>-----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362761 retrieved from XFIRE Database accession no. 5577300, 5530989 abstract</p> <p>& CAN. J. CHEM., vol. 64, 1986, pages 932-935,</p> <p>-----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362762 retrieved from XFIRE Database accession no. 2699259, 2655540, 2696376 abstract</p> <p>& J. ORG. CHEM., vol. 26, 1961, pages 1420-1422,</p> <p>-----</p> <p style="text-align: center;">-/--</p>	6,7

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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362763 retrieved from XFIRE Database accession no. 2703830 abstract & J. ORG. CHEM., vol. 27, 1962, pages 3092-3095, -----</p>	6,7
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362764 retrieved from XFIRE Database accession no. 7500084 abstract & RUSS. J. ORG. CHEM., vol. 31, no. 1.1, 1995, pages 61-63, -----</p>	6,7
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362765 retrieved from XFIRE Database accession no. 8949041 abstract & COLLECT. CZECH. CHEM. COMMUN., vol. 66, no. 8, 2001, pages 1250-1256, -----</p>	6,7
X	<p>GB 1 574 944 A (CIBA-GEIGY AG) 10 September 1980 (1980-09-10) claim 1; example 1 -----</p>	6,7
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362766 retrieved from XFIRE Database accession no. 3533432, 3453030, 3440221, 3433570, 3420661 3405277, 3408533, 3496042 abstract & J. CHEM. SOC., 1928, page 2226, -----</p> <p style="text-align: center;">-/--</p>	6

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C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DATABASE BEILSTEIN 'Online! Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362767 retrieved from XFIRE Database accession no. 8801049, 8799606, 8797479, 8788892, 8786565 3397089 abstract & ORG. PREP. PROCED. INT., vol. 33, no. 1, 2001, pages 95-100, -----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362768 retrieved from XFIRE Database accession no. 3397465, 3424175, 3816566 abstract & ARCH. PHARM., vol. 292, 1959, pages 690-695, -----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362769 retrieved from XFIRE Database accession no. 2710289 abstract & HELV. CHIM. ACTA, vol. 43, 1960, pages 1086-1112, -----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362770 retrieved from XFIRE Database accession no. 9575084 abstract & CHEM. PHARM. BULL., vol. 51, no. 11, 2003, pages 1325-1327, -----</p>	6
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362771 retrieved from XFIRE Database accession no. 3476206, 3493531, 3503234 abstract</p>	6

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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>& J. AM. CHEM. SOC., vol. 45, 1923, page 1062, -----</p> <p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362772 retrieved from XFIRE Database accession no. 4211353, 4211344 abstract</p>	6
X	<p>& J. AM. CHEM. SOC., vol. 112, no. 22, 1990, pages 7861-7868, -----</p> <p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362773 retrieved from XFIRE Database accession no. 3450820, 3451642 abstract</p>	6
X	<p>& J. CHEM. SOC., 1955, pages 3914-3918, -----</p> <p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362774 retrieved from XFIRE Database accession no. 2180369 abstract</p>	6
X	<p>& BULL. ACAD. SCI. USSR DIV. CHEM. SCI., 1968, page 863, -----</p> <p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002362775 retrieved from XFIRE Database accession no. 2606237 abstract</p>	6
X	<p>& J. ORG. CHEM., vol. 22, 1957, pages 1439-1442, -----</p> <p>US 5 674 482 A (REGAN ET AL) 7 October 1997 (1997-10-07) claims 1,4,23,31,32; examples 14,15 -----</p> <p style="text-align: center;">-/--</p>	1,3,9-11

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International application No
PCT/NL2005/000826

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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	TINAPP P: "Herstellung von Mandelaldehyden durch katalytische Hydrierung von Mandelsäurenitrilen mit Raney-Nickel in Gegenwart von Schwefelsäure" CHEMISCHE BERICHTE, VERLAG CHEMIE GMBH. WEINHEIM, DE, vol. 104, no. 7, 1971, pages 2266-2272, XP002240442 ISSN: 0009-2940 Compounds 1e and 3e in the scheme on p.2267; last full paragraph on p.2268 - table 4 inclusive.	1,3,9-11
X	EP 1 125 908 A (CLARIANT) 22 August 2001 (2001-08-22) paragraphs '0001! - '0016!; claims 1-15; examples 1-4,6,7,10	1,3,9-11
X	WO 2004/026807 A (KISSEI PHARMACEUTICAL CO., LTD; TANAKA, NOBUYUKI; TAMAI, TETSURO; MUKA) 1 April 2004 (2004-04-01) claim 3; example 1	1,3
P,X	-& EP 1 535 897 A (KISSEI PHARMACEUTICAL CO., LTD) 1 June 2005 (2005-06-01)	1,3
X	YUEFA GONG ET AL: "Facile Substitution of N,N-Dimethylanilines and Phenols with Trifluoroacetaldehyde Ethyl Hemiacetal" SYNLETT, THIEME INTERNATIONAL, STUTTGART, DE, no. 9, 1999, pages 1403-1404, XP002360732 ISSN: 0936-5214 the whole document	1,3,4,8
X	CASIRAGHI G ET AL: "Selectivity in Reactions between Metal Phenolates and Trichloroacetaldehyde; A Mild Synthesis of 2-(2,2,2-Trichloro-1-hydroxyethyl)-phenols" SYNTHESIS, GEORG THIEME VERLAG, STUTTGART, DE, 1979, pages 824-825, XP002360733 ISSN: 0039-7881 the whole document	1,3
X	AT 141 159 B (DR. HERMANN HAAKH UND DR. ING. ALBERT SMOLA IN WIEN) 25 March 1935 (1935-03-25) example 2	1,3
X	DE 193 634 C (FARBWERKE VORM. MEISTER LUCIUS & BRÜNING IN HÖCHST A. M.) 8 December 1906 (1906-12-08) page 2, lines 17-35	1,3
	-/--	

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C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360790 retrieved from XFIRE Database accession no. 5033436, 5029530 abstract & TETRAHEDRON, vol. 39, no. 13, 1983, pages 2147-2150, -----	1,3
X	DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360791 retrieved from XFIRE Database accession no. 2114830, 2108470, 2109370, 2109441 abstract & BULL. SOC. CHIM. FR., 1979, pages 583-591, -----	1,3
X	DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360792 retrieved from XFIRE Database accession no. 9196198, 9196197 abstract & CHEM. COMMUN., vol. 17, 2002, pages 1914-1915, -----	1,3
X	DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360793 retrieved from XFIRE Database accession no. 9208211, 2104039 abstract & EUR. J. ORG. CHEM., vol. 12, 2002, pages 1984-1988, -----	1,3
X	DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360794 retrieved from XFIRE Database accession no. 3242862 abstract & LIEBIGS ANN. CHEM., vol. 313, 1900, page 87, -----	1,3

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International application No

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C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360795 retrieved from XFIRE Database accession no. 2086319 abstract & J. CHEM. SOC., 1959, pages 257-262, -----</p>	1,3
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360796 retrieved from XFIRE Database accession no. 2092210 abstract & CAN. J. BIOCHEM., vol. 53, 1975, page 920, -----</p>	1,3
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360797 retrieved from XFIRE Database accession no. 4746216, 4675177 abstract & CAN. J. CHEM., vol. 69, no. 5, 1991, pages 772-778, -----</p>	1,3
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360798 retrieved from XFIRE Database accession no. 3135094 abstract & HOPPE-SEYLER'S Z. PHYSIOL. CHEM., vol. 52, 1907, page 392, -----</p>	1,3
X	<p>DATABASE BEILSTEIN Beilstein Institut zur Förderung der Chemischen Wissenschaft, Frankfurt am Main, DE; XP002360799 retrieved from XFIRE Database accession no. 4297381, 4297382, 4298426, 4298427, 4301572 4301573 abstract & TETRAHEDRON ASYMMETRY, vol. 1, no. 12, 1990, pages 861-864, -----</p>	1-3
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International application No

PCT/NL2005/000826

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Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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