

An integrated lignin biorefinery: Scaling-up lignin depolymerization technology for biofuels and chemicals

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An integrated lignin biorefinery: Scaling-up lignin depolymerization technology for biofuels and chemicals

Kouris P.D.^{1*}, Oevering H.², Boot M.D.¹ and Hensen E.J.M.¹

¹ Eindhoven University of Technology, 5600 MB Eindhoven, The Netherlands

² DSM Ahead Innovative Synthesis, Royal DSM, Brightlands Chemelot Campus, 6161BB Geleen, The Netherlands

*Corresponding author: p.d.kouris@tue.nl



Technische Universiteit

University of Technology

Inorganic

Materials

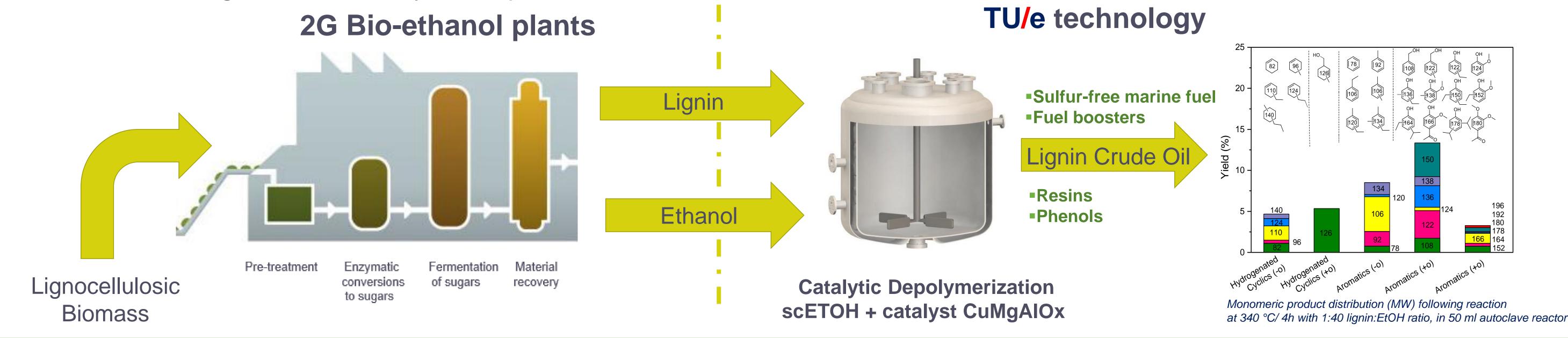
Chemistry

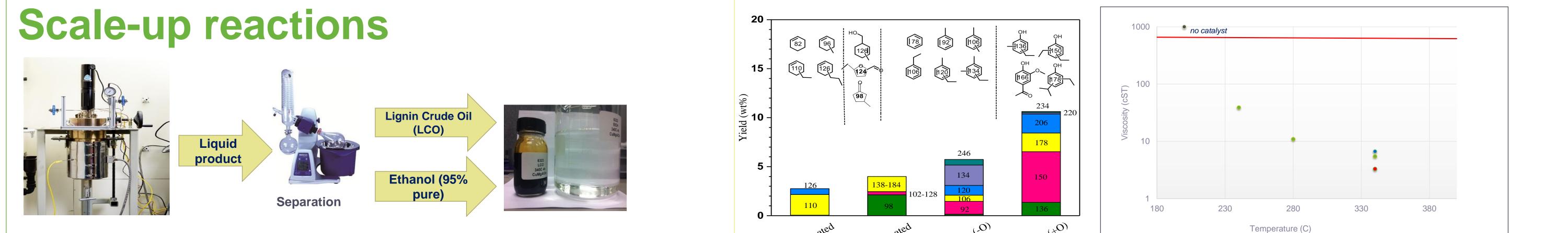
Where innovation starts

Biobased materials

Lignin RICHES (<u>Resins Chemicals Fuels</u>)

Lignin is one of the major components of lignocellulosic biomass, constituting 15-30 % of the weight and approximately 40 % of the energy content depending on the source. Currently the lignin produced in 2G bio-ethanol plants is mainly used for on-side energy production. At Eindhoven University of Technology a method was explored to depolymerize lignin in super critical ethanol with cheap non-noble catalysts to produce a mixture of monomeric aromatics. The product might be applied directly as a bio marine fuel, or as a source for chemical building blocks (Resins), octane boosters or biofuels when blended with gasolines. The primary goal for pilot activities is to produce Lignin Crude Oil (LCO) from lignin with a viscosity spec < 800 cSt at 40 °C, on a ton scale and to collect information for designing a demo plant with the aim of having an economically viable process.





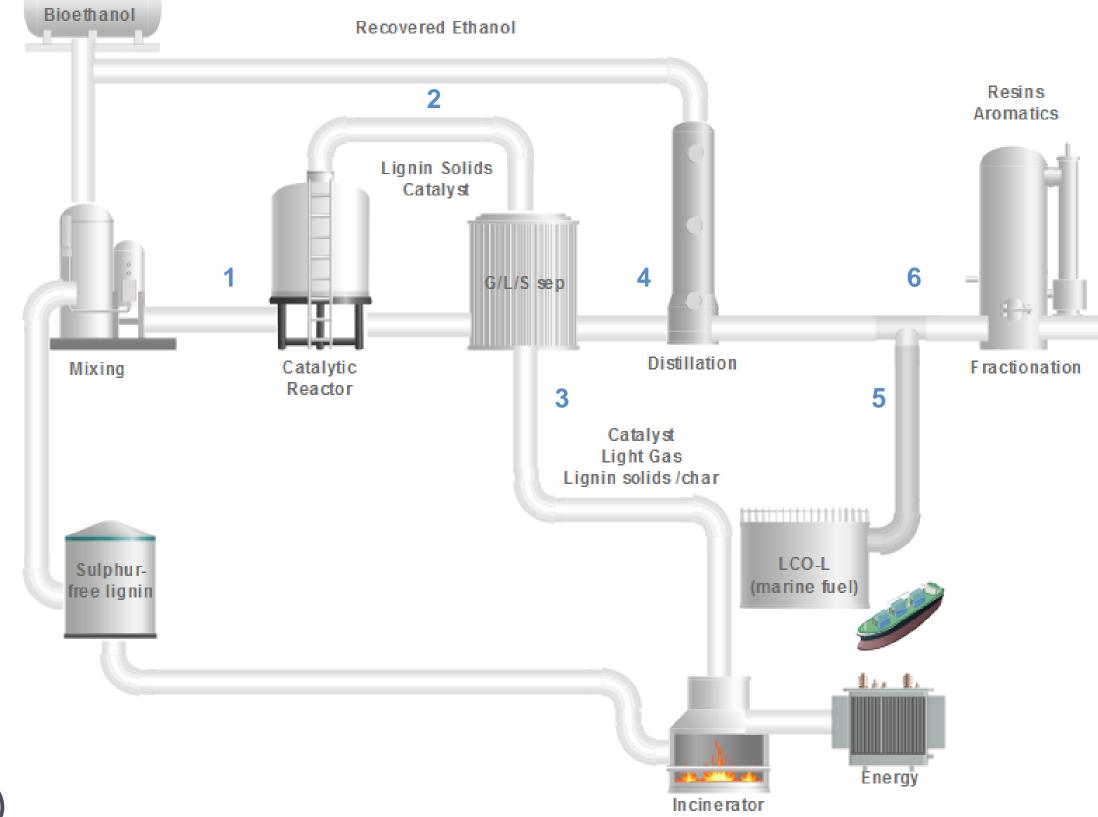
Temp [°C]	Lignin:EtOH ratio (w/v)	Monomer Yield (wt%)	LCO viscosity @40°C (cST)	Ethanol conversion (wt%)	
200 (no cat)	1:40	1	>1000	4	
240	1:40	1	39	12	
280	1:40	3	11	21	
340	1:40	11	5,5	53	
340	1:30	14	3,3	46	
340	1:20	25	6,7	47	

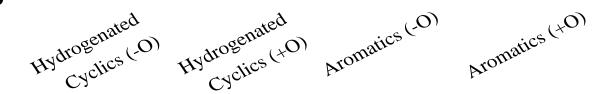
Conditions: Lignin (20-50 g), CuMgAIOx (10-25g), Ethanol (800-1200 ml), Lignin: EtOH ratio : 1:40 -1:10 4L Autoclave reactor, t=4h



Process steps

- 1. Feed lignin / ethanol to catalytic reactor
- 2. Re-use catalyst / unconverted lignin
- 3. Burn catalyst / gas products / char for energy production
- 4. Liquid stream for separation





Monomeric product distribution (MW) following reaction at 340 °C with 1:20 ratio, in a 4L reactor

Kinematic viscosity of LCO over different reaction temperatures

(red line: maximum viscosity limit)

Elemental analysis						GHV	Viscosity
	Oxygen(%)	Carbon(%)	Hydrogen(%)	Nitrogen(%)	Sulphur(%)	GJ/tn	cST
LCO 1:20	15,5	73,4	10,3	0,8	0	38,1	6,7

- High monomer yield is important for the LCO fractionation process (resins and phenols)
- > Gross Heating Value (GHV) and viscosity specifications are crucial for marine fuel application
- Ethanol conversion products are contributing to decreasing the LCO viscosity to very low levels

Design Challenges

- Process complexity
- Batch vs continuous
- Operating window
- Lignin / catalyst loading
- Ethanol conversion
- Reactor design
- Catalyst regeneration

process

- 5. Lignin Crude Oil production directly for low viscosity marine fuel (LCO-L)
- 6. Downstream process of LCO for resins application
- 7. More concentrated LCO for marine fuel application (LCO-M)

Separation steps Ethanol concentration in LCO Ethanol losses in the process

Approach

LCO-M marine fuel)

- Experimental data (input for process design)
- Mass & energy balances for all process streams
- Perform techno-economical study
- Optimize the most feasible process routes



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