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Citation for published version (APA): Gorbunov, A. V., Putzeys, T., Urbanavičiute, I., Janssen, R. A. J., Wübbenhorst, M., Sijbesma, R. P., & Kemerink, M. (2016). True ferroelectric switching in thin films of trialkylbenzene-1,3,5-tricarboxamide (BTA). Physical Chemistry Chemical Physics, 18(34), 23663-23672. https://doi.org/10.1039/c6cp03835b

DOI: 10.1039/c6cp03835b

Document status and date:

Published: 01/01/2016

Document Version:

Publisher's PDF, also known as Version of Record (includes final page, issue and volume numbers)

Please check the document version of this publication:

• A submitted manuscript is the version of the article upon submission and before peer-review. There can be important differences between the submitted version and the official published version of record. People interested in the research are advised to contact the author for the final version of the publication, or visit the DOI to the publisher's website.

• The final author version and the galley proof are versions of the publication after peer review.

• The final published version features the final layout of the paper including the volume, issue and page numbers.

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Cite this: Phys. Chem. Chem. Phys., 2016, 18, 23663

True ferroelectric switching in thin films of trialkylbenzene-1,3,5-tricarboxamide (BTA)

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We have investigated the ferroelectric polarization switching properties of trialkylbenzene-1,3,5tricarboxamide (BTA), which is a model system for a large class of novel organic ferroelectric materials. In the solid state BTAs form a liquid crystalline columnar hexagonal phase that provides long range order that was previously shown to give rise to hysteretic dipolar switching. In this work the nature of the polar switching process is investigated by a combination of dielectric relaxation spectroscopy, depth-resolved pyroelectric response measurements, and classical frequency- and time-dependent electrical switching. We show that BTAs, when brought in a homeotropically aligned hexagonal liquid crystalline phase, are truly ferroelectric. Analysis of the transient switching behavior suggests that the ferroelectric switching is limited by a highly dispersive nucleation process, giving rise to a wide distribution of switching times.

Received 2nd June 2016, Accepted 3rd August 2016

DOI: 10.1039/c6cp03835b

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1. Introduction

In recent years there has been a significant interest in organic molecular materials that show hysteretic dipolar switching behavior.¹⁻⁹ Even though true ferroelectricity was not always demonstrated, we shall refer to these behaviors as 'ferroelectric' in fact demonstrating true ferroelectricity for a model compound is a major topic of this work, as will be discussed in detail below. Compared to inorganic ferroelectrics both the interest in and the technological relevance of organic ferroelectrics is still modest, possibly with the exception of the polymeric ferroelectric polyvinylidene difluoride (PVDF) and its copolymers with trifluoroethylene (TrFE).^{10,11} Apart from historic reasons, the often modest characteristic values for the remnant polarization Pr and the coercive field E_c of organic ferroelectrics may partially explain the low interest. In addition, lacking knowledge about basic properties makes it hard to judge the potential scientific and technological relevance of organic molecular ferroelectrics. However, compared to inorganic ferroelectrics, organic materials

offer a number of significant key advantages such as the absence of costly, difficult epitaxial growth and the absence of toxic and/ or rare elements. In addition, organic molecular ferroelectrics possess great mechanical flexibility, they allow for easy solutionbased processing on various substrates and, importantly, they provide nearly infinite freedom to tune the material properties through chemical modification. Furthermore, materials that are similar to those studied in this manuscript have recently been suggested to show interesting photovoltaic properties.^{12,13}

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The design freedom offered by organic chemistry is reflected by the different moieties that are responsible for the dipolar response in reported organic molecular ferroelectrics. A major class of organic ferroelectrics is based on molecular dipoles introduced by amide groups.4-6,8 Apart from introducing a molecular dipole of \sim 3 D per amide group that can be fieldreversible,^{5,6,8} the amide groups also contribute to the molecular stacking and the associated long-range order through the formation of hydrogen bonds.4,14,15 In previous work we demonstrated the dipolar switching of an amide-based model compound, trialkylbenzene-benzene-1,3,5-tricarboxamide (BTA, see inset to Fig. 1(d) for chemical structure).^{5,6} Although hysteresis and saturation in the polarization vs. field (P-E) loops were observed, the poor polarization retention, in the order of seconds to minutes, at the temperatures needed for dipolar switching, casted doubt on the true polarized nature of the materials' ground state.6

Here, we present a detailed study of the ferroelectric properties of thin-film capacitors with the BTA model compound as an active layer. We show that BTAs, when brought in a homeotropically aligned hexagonal liquid crystalline phase, are truly ferroelectric.

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 $[\]dagger$ Electronic supplementary information (ESI) available. See DOI: 10.1039/ c6cp03835b

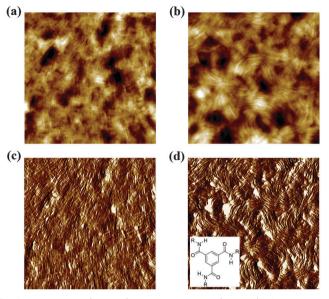


Fig. 1 Topography (a and b) and phase images (c and d) of spin coated films of BTA-C10 (a and c) and BTA-C18 (b and d) obtained by tapping mode AFM. Inset: BTA structure, R = $C_{10}H_{21}$ for BTA-C10 and R = $C_{18}H_{37}$ for BTA-C18. Scan size 10 \times 10 μ m², vertical scales are 78 nm (a), 155 nm (b), 38° (c), 47° (d).

To this end, we combine dielectric relaxation spectroscopy (DRS), depth-resolved pyroelectric probing (LIMM), and quasi-static and transient polarization switching experiments and study two BTA compounds with different alkyl chain lengths (BTA-C10 and BTA-C18). The switching properties of the compounds seem largely dominated by disorder.

2. Results and discussion

In this study we employed devices in which the BTA compounds were sandwiched between gold or indium tin oxide (ITO) electrodes. The details of the device fabrication can be found in the Experimental section. After deposition from solution both BTA-C10 and BTA-C18 form achiral stacks that are stabilized by N–H–O hydrogen bonding between successive amide groups and π -stacking of the conjugated benzene core.⁵ After deposition by spin coating, the individual BTA stacks form bundles that lie in the plane of the metal electrode. This is demonstrated in Fig. 1, which presents topography and phase images obtained by tapping mode atomic force microscopy (TM-AFM), showing the in-plane aligned bundles after spin coating on Au/glass substrates.

After deposition of the Au top contact the BTA stacks are homeotropically aligned with the surface normal, *i.e.* out-ofplane, by the application of an electric field. Both the alignment and the subsequent investigation of the ferroelectric properties are performed at elevated temperatures where the materials are in a columnar hexagonally packed liquid crystalline (Col_{hex}) phase.⁶ Fig. S1 (ESI[†]) shows the differential scanning calorimetry (DSC) traces of both compounds, indicating that the Col_{hex} phase of BTA-C10 and BTA-C18 lies between 12–200 °C

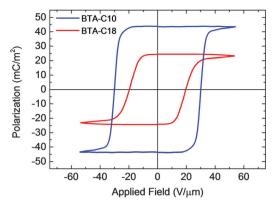


Fig. 2 Ferroelectric polarization versus applied field for BTA-C10 (blue) and BTA-C18 (red) under quasi-static conditions (1 Hz, 100 $^{\circ}$ C).

and 69–190 $^{\circ}$ C, respectively. Homeotropic out-of-plane alignment was checked by polarization optical microscopic (POM) as shown in Fig. S2 (ESI†). It should be stressed that no macroscopic polarization could be measured on non-aligned thin film devices, indicating that an out-of-plane macrodipole can only be formed along the stacking axis of the material. Supramolecular organization is essential for the dipolar response.

2.1 Establishing ferroelectric behavior

Fig. 2 shows the saturated polarization vs. field (*P*–*E*) curves of BTA-C10 and BTA-C18 as measured quasi-statically by the double wave method (DWM) that is described in detail in ref. 16 and 17; see also the Experimental section. Both materials show well-developed saturation behavior, with $P_r = 44 \text{ mC m}^{-2}$ and $E_c = 30 \text{ V } \mu \text{m}^{-1}$ for BTA-C10 and $P_r = 25 \text{ mC m}^{-2}$ and $E_c = 20 \text{ V } \mu \text{m}^{-1}$ for BTA-C18. Compared to our previous work, ref. 5 and 6, the shape of the curves is substantially closer to ideal and the values for the remnant polarization are a factor ~ 1.5 (BTA-C18) to ~ 2.5 (BTA-C10) higher. The coercive field values are essentially unchanged. We attribute the differences to improved alignment procedures and a further development of the DWM. Non-saturated inner loops for both materials can be found in Fig. S6 of the ESI.†

The difference in $P_{\rm r}$ between BTA-C10 and BTA-C18 can largely be attributed to the larger number of amide dipoles per unit area in the former. The lattice constant of the hexagonal columnar LC phase is 2.00 nm and 2.55 nm, respectively.⁶ Assuming a constant dipole moment per amide group leads to a geometric difference in dipole density of $(2.55/2.00)^2 \approx 1.62$, which is close to the $P_{\rm r}$ -ratio of 44/25 ≈ 1.76 .

The measured values of P_r allow us to make an updated estimate for the molecular dipole μ of a single BTA unit.⁶ This yields $\mu = 16$ D for BTA-C10 and $\mu = 15$ D for BTA-C18. These numbers are on the higher end of the range of theoretical estimates for isolated stacks of BTAs that yield $\mu = 6-14$ D.^{14,18-20}

2.1.1. Dielectric relaxations around the paraelectric/ferroelectric phase transition. As discussed in a previous paper,⁶ BTA-C10 and BTA-C18 show a couple of molecular relaxation processes that are specific to the isotropic high temperature phase (T > 200 °C) and the hexagonal columnar phase.

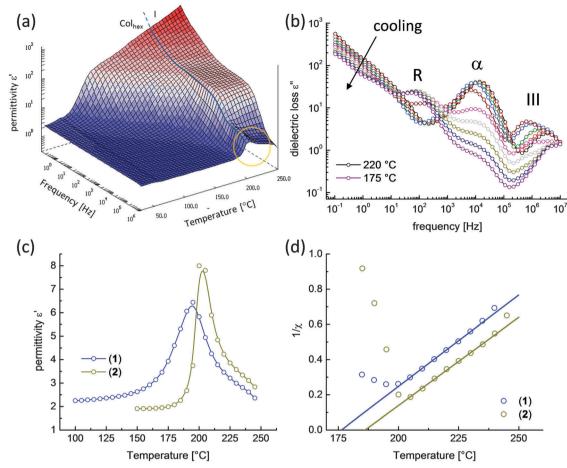


Fig. 3 (a) Dielectric permittivity $\varepsilon'(f,T)$ for BTA-C18 measured upon cooling from 250 °C to 25 °C. The orange circle marks the pre-transitional region characterized by Curie–Weiss behavior. (b) Evolution of the dielectric loss $\varepsilon''(f)$ for BTA-C18 while passing the paraelectric to ferroelectric phase transition upon cooling (same experiment as shown in (a)). The temperature independent, high-frequency peak around 5 MHz is an artefact originating from the RC time constant of the ITO sample cell. (c) Temperature dependence $\varepsilon'(T)$ at f = 220 kHz for two samples of BTA-C18. Sample (1) displays a broadened phase transition compared to (2). (d) Same data plotted in Curie–Weiss representation according to eqn (1).

An overview of the dielectric permittivity $\varepsilon'(f,T)$ measured on BTA-C18 between ITO electrodes in a parallel plate geometry is given in Fig. 3(a), showing the isotropic to hexagonal columnar liquid crystal phase transition as an abrupt change of the permittivity spectra around $T_{I \to Colhex} \approx 200$ °C. For a closer inspection of the molecular relaxation processes, the evolution of the dielectric loss spectra $\varepsilon''(f)$ around the phase transition is displayed in Fig. 3(b); the full 3D spectrum of $\varepsilon''(f,T)$ is shown in the ESI,† as Fig. S4. Comparison of dielectric relaxation spectroscopy (DRS) spectra of BTA-C10 and BTA-C18 showed only minor, quantitative differences, hence we will focus the following on BTA-C18.

Well above $T_{I \rightarrow Colhex}$, a single relaxation process located around f = 10 kHz dominates the relaxation spectrum, which was assigned to the structural α -relaxation of the isotropic melt, see ref. 6. By approaching the phase transition, a second, faster relaxation mode (III) emerges that coexists with the α -process and gains intensity upon further cooling.

When passing $T_{I \rightarrow Colhex}$ and entering the hexagonal columnar phase, the relaxation pattern changes drastically, see Fig. 3(b). As expected, the α -relaxation, being due to columnar macro-dipole

fluctuations in randomly oriented, short columns that are present in the isotropic phase,²¹ almost vanishes and is basically replaced by the slower R-relaxation according to the labeling by Fitié *et al.*⁶ Ideally, the α -process should be gone in a perfect ferroelectric order. The residual intensity in the ferroelectric hexagonal columnar phase is thus indicative for the existence of short (disordered) stacks, probably near interfaces.

The slow R-process is attributed to macro-dipole fluctuations of the long columnar macro-dipoles.²¹ The same macro-dipoles are the molecular origin for the ferroelectric depolarization. Its intensity is however not directly linked to the polarization itself. Instead, since R-relaxation is the manifestation of spontaneous fluctuations of macro-dipoles within the columnar hexagonal order, its intensity should be a measure for the number of active columnar defects (up/down), a quantity that is related to the temporal stability of the remnant polarization – in ref. 6 R-relaxation could indeed be directly linked to the polarization loss.

Interestingly, the third relaxation (III) also disappears upon the phase transition, a behavior that classifies process (III) as a transient phenomenon that solely occurs just above $T_{I \rightarrow Colhex}$. We hypothesize that this fast relaxation is related to pre-transitional effects in the isotropic phase and is likely caused by individual disks, dimers or other, weakly correlated entities, the concentration of which is highest just before entering the ferroelectric phase (upon cooling). In terms of a ferroelectric/paraelectric phase transition, these appear to give rise to the typical temperature dependence of the dielectric susceptibility $\chi = \varepsilon - 1$ according to the Curie–Weiss-law

$$\chi = C/(T - T_{\rm C}),\tag{1}$$

where *C* and $T_{\rm C}$ represent the Curie constant and the Curie temperature, respectively. Further analysis of the temperature dependence of the R-, α - and III-relaxation processes along the lines outlined in ref. 22 and 23 can be found in the ESI,[†] as Fig. S5.

Revealing Curie-Weiss behavior for BTA-C18 (or BTA-C10) would make for very strong evidence towards unambiguously confirming paraelectric and ferroelectric behavior for the BTAs. Revisiting Fig. 3(a) and inspecting the marked high-frequency region just above $T_{I \rightarrow Colhex}$ shows a progressive rise of the permittivity towards the phase transition, strongly suggesting Curie-Weiss characteristics. To investigate this behavior more quantitatively, we have plotted the temperature dependence of the permittivity $\varepsilon'(T)$ at f = 220 kHz in Fig. 3(c) for two samples of BTA-C18 with different alignment history. Though both curves reveal the phase transition as a peak in $\varepsilon'(T)$, a particular broadening for sample (1) is discernible as compared to the better (AC at 3 V_{RMS} instead of DC, at slightly higher T) aligned sample (2). The differences between the two samples show the critical importance of morphology, where the desired out-of-plane alignment is promoted by the field annealing described in the Experimental section.

By replotting the data in the form $1/\chi(T)$ we can check for the validity of the Curie–Weiss behavior. The result for both samples of BTA-C18 is given in Fig. 3(d) and clearly reveals a linear trend confirming true paraelectric behavior in the high-temperature phase. Extrapolation of $1/\chi(T)$ to zero yields the Curie-temperatures of both samples, which are generally somewhat lower than $T_{I\rightarrow Colhex}$; here the difference amounts to 15 K for sample (2) and 25 K for the poorly aligned sample (1). These numbers are in excellent agreement with the findings in ref. 21, where strong hydrogen bonding in large, aligned columnar aggregates was found to change to weak (but finite) hydrogen bonding in small, isotropic aggregates some 15 °C below $T_{I\rightarrow Colhex}$ as probed by DSC.

2.1.2. Pyroelectric depth profiling and hysteresis of BTA-C10 and BTA-C18. By using high-frequency photothermal waves in the range of 25 kHz to 25 MHz, and measuring the resulting pyroelectric response caused by the local change in temperature, a depth profile of the local polarization in metal–insulator–metal thin film devices can be obtained after an appropriate mathematical deconvolution.²⁴ Analysis of the depth profiles obtained with this laser intensity modulation method (LIMM) with variation in DC bias field reveals a polar switching of both BTA-C10 and BTA-C18 during reversal of the external field. A reconstruction of the *P*–*E* loop identifies a hysteresis during polarization switching, indicating ferroelectric behavior as shown in Fig. 4.

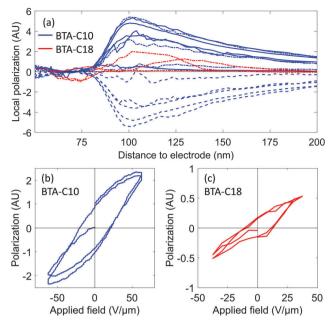


Fig. 4 LIMM experiments on BTA-C10 (blue lines) and BTA-C18 (red lines). (a) Pyroelectric depth profiles after various poling histories: solid: $0 \rightarrow 75 \rightarrow 0 \vee \mu m^{-1}$; dash-dotted: $0 \rightarrow 75 \vee \mu m^{-1}$; dashed: $0 \rightarrow -75 \rightarrow 0 \vee \mu m^{-1}$. (b and c) Polarization hysteresis as obtained by LIMM. The external field frequency is $\approx 100 \mu$ Hz.

The first ~ 80 nm of the BTA-C10 profiles appears to be an inactive or "dead" layer, yielding a nearly non-detectable polarization switching, see panel (a) of Fig. 4. Also for BTA-C18 a dead layer is observed, but its thickness is hard to determine. The presence of dead layers can be attributed to two effects. Due to the spin coating of the thin film, the columnar bundles at the electrode interface are oriented in-plane, *i.e.* parallel to the electrodes. It is assumed that dipolar switching occurs in the axial orientation of the columns, hence these first layers of flat columns do not contribute to any dipolar switching. A second effect is a screening of the external field, somehow related to the Debye-screening length due to (real or virtual) free charge concentrations near the electrodes, similar to the dead layer found in thin film P(VDF-TrFE).²⁵ The screening of the external field in the first ~ 80 nanometers implies the molecules do not experience the field required for macroscopic polarization.

At a distance further than 110 nm from the electrode, the local polarization appears to decrease. This artefact is inherent to LIMM deconvolution techniques on thin (<1000 nm) organic films. As the thermal diffusion depth increases (*i.e.* when the thermal waves can reach deeper into the film), the spatial resolution decreases. In addition, in thin films, the effusivity of the substrate will play a major role in the thermal amplitude at greater depths, reducing the thermal amplitude, resulting in a lower observed local polarization. Correction for this effect requires accurate knowledge of the thermal parameters of the sample, which is beyond the scope of the present work.

From the LIMM measurements on BTA-C10 and BTA-C18, it appears that the local polarization and susceptibility are

inversely proportional to the size of the residual group (C10 or C18). BTA-C10 has a larger susceptibility and a higher coercive field compared to BTA-C18. While the residual group is necessary to enable dipolar switching, an increase in size seems to diminish its amplitude, as the associated molecular relaxation hinders the switching or screens the ferroelectric entities from the external field.

The hysteresis curves obtained *via* pyroelectric means differ in shape from those using the DWM technique in Fig. 2. This can be in part explained by the difference in external field frequency. The double wave method alternates the sign of the field at a rate around once per second, the pyroelectric setup only changes the sign every 2 hours: to increase the signal-tonoise ratio, several integration, correlation and averaging techniques are used. The longer timescale may invoke structural changes not visible in the DWM experiment and, more importantly, will allow polarization decay at lower fields. These effects introduce a linear slope in the *P*-*E* hysteresis curve, which is not present at higher switching frequencies, *cf.* Fig. 2.

2.2 Kinetic properties

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Having established the true ferroelectric nature of the dipolar switching we shall now address its kinetic properties. To this end we performed step response measurements, in which the transient current response to a polarization-reversing voltage step is measured, as shown in Fig. 5(a and b). As expected for a ferroelectric material,^{26,27} the current transient shifts to shorter times with increasing field.

The switching current is proportional to $d(\Delta P(t))/dt$, with $\Delta P(t)$ the time-dependent change in polarization. In the classical Kolmogorov–Avrami–Ishibashi (KAI) theory that accounts for space-filling domain growth after nucleation, $\Delta P(t)$ is given by^{28,29}

$$\frac{\Delta P(t)}{2P_{\rm S}} = 1 - \exp\left(-\left(\frac{t}{\tau}\right)^d\right) \tag{2}$$

with *d* a parameter that equals the effective dimensionality of the switching process or one plus the effective dimensionality, depending on whether the nucleation occurs only at t = 0 or at a constant rate, and τ a characteristic time.³⁰ Fitting eqn (2) to the data in Fig. 5 gives $d \approx 2$ –2.7, but the predicted peak shape only follows the experiment over a fraction of the measurement range as shown by the thin dotted lines in Fig. 5(c and d). The failure of eqn (2) to describe the experimental data sits mainly in the longer time range, where the growth is substantially slower in the experiments than predicted by the KAI theory. This suggests the experimental

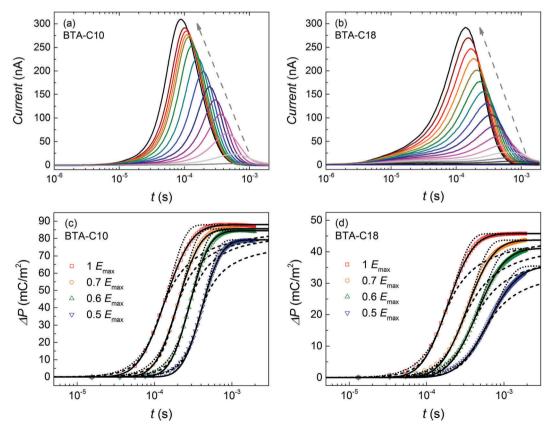


Fig. 5 Transient switching current after stepping the electric field at t = 0 measured on (a) BTA-C10 and (b) BTA-C18 at 100 °C. The dashed arrow indicates the peak shift with increasing electric field strength from 19.5 V μ m⁻¹ to 50 V μ m⁻¹ (BTA-C10) and 19.5 V μ m⁻¹ to 70.5 V μ m⁻¹ (BTA-C18). The integrated transient switching current, representing the time-dependence of the actual switched part at 1, 0.7, 0.6 and 0.5 of the maximal probed electric field (50 V μ m⁻¹ and 70.5 V μ m⁻¹ for BTA-C10 and BTA-C18, respectively) is given in panels (c) for BTA-C10 and (d) for BTA-C18. Thin black lines are fits to the data using KAI (eqn (2), dotted line) and NLS (eqn (4)) with log-normal (eqn (3), solid line) and log-Lorentzian (dashed line) distribution of the switching times.

response is dominated by a distribution of relaxation times.^{31–33} This is the main idea of the nucleation-limited-switching (NLS) theory.³⁴

In ref. 32 a Lorentzian distribution of logarithmic switching times was suggested for similar experiments performed on polycrystalline Pb(Zr,Ti)O₃ films. For the present experiments, we found this distribution to be inadequate, as shown by the dashed lines. However, we noticed that a simple, phenomenological lognormal distribution of switching times (eqn (3)), integrated into the NLS model (eqn (4)) gives for both materials and all fields an accurate description of the region around and beyond the maximum of the current distribution, as shown by the thin solid lines in Fig. 5(c and d). It is noticeable that the experimental polarization curves saturate at different values (used for fitting), which most likely is caused by the disorder in the active layer, which is also expressed as the broadening of the P-E loops. Apparently, due to a distribution in (local) coercive fields we cannot achieve the full switching at lower fields.¹⁷ In eqn (3) w is the width of the distribution and τ is its central value.

$$F(t) = \frac{1}{w\sqrt{2\pi}} \exp\left(-\frac{(\log(t) - \log(\tau))^2}{2w^2}\right)$$
(3)

$$\Delta P(t) = 2P_{\rm S} \int_{-\infty}^{+\infty} \left[1 - \exp\left(-\left(\frac{t}{\tau}\right)^d\right) \right] F(t) \mathrm{d}t \tag{4}$$

We found that the fitting parameter d in eqn (4) – the dimensionality of the domain growth – equals 1, which would correspond to 1D stripe-like domains which is not unlikely in discotic columnar liquid crystalline systems.

In general, lognormal distributions result from processes that depend exponentially on a (normally distributed) random variable. Such processes are abundant and include thermal activation over and tunneling through a random barrier, which may be of particular relevance for the specific case of a disordered ferroelectric.

The above analysis suggests a situation where at short times the switching is limited by a KAI-like process of nucleation and subsequent growth. At longer times, including those where most of the switching occurs, the dispersion in characteristic switching times dominates the transient. Next the process that governs this dispersive region will be further investigated.

Fig. 6 shows the position of the current maxima in Fig. 5 as a measure of the switching speed *vs.* the reciprocal applied field. At all temperatures, we find that the classical Merz law for nucleation-limited switching,^{26,27}

$$t_{\rm S} = t_0 \cdot \exp\left(\frac{E_0}{E}\right),\tag{5}$$

only gives a reasonable description at low fields. The black dash-dotted lines give an example of typical fits. In eqn (5), t_0 and E_0 are a typical (minimum) switching time and a characteristic electric field, respectively. It should be noted that eqn (5) is often phenomenologically combined with KAI theory, eqn (2), to get a phenomenological expression for the field

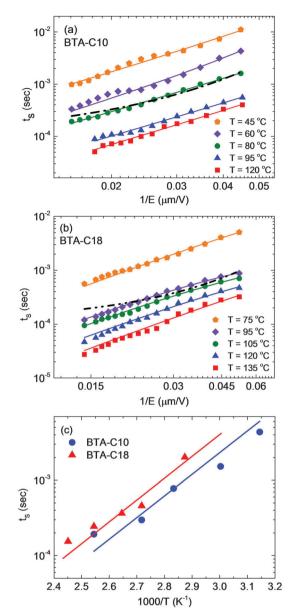


Fig. 6 (a and b) Step response switching time from Fig. 5 vs. electrical field for BTA-C10 (a) and BTA-C18 (b) plotted at different temperatures at double logarithmic scale. Black dash-dotted curves are fits to the Merz nucleation model (eqn (5), $t_0 \sim 10^{-12}$ s), the solid lines are fits to the hybrid model eqn (6). (c) Switching time vs. reciprocal temperature at an applied field of 33 V μ m⁻¹. The solid lines correspond to activation energies of 0.57 eV = 55 kJ mol⁻¹ (BTA-C10) and 0.58 eV = 56 kJ mol⁻¹ (BTA-C18).

dependence of the characteristic time τ . In a number of works variations on the Merz expression eqn (5) of the shape

$$t_{\rm S} = A \cdot E^{-n} \cdot \exp\left(\frac{E_0}{E}\right) \tag{6}$$

have been proposed.^{35–37} Fitting eqn (6) with fixed n = 2 to the data in Fig. 6 gives a satisfactory description over the full field and temperature range, as shown by the solid lines.

It has been proposed that eqn (6) with n = 2 can be due to a switching process that is rate-limited by Fowler–Nordheim charge injection.³⁵ In view of the presence of a 'dead' interfacial

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layer of dielectric nature in our devices (Fig. 4), it is in principle not implausible that a tunneling charge injection mechanism is relevant to these experiments. However, the mechanism proposed in ref. 35 gives rise to a continuously decaying current transient, as expected for a switching current that is determined by essentially a contact resistance. This is in stark contrast to the peaked shape observed in Fig. 6. Alternatively, assuming that a Fowler–Nordheim-type injection current limits the onset of the switching process, *e.g.* because the applied field initially drops over one or two interfacial layers, could possibly explain the observed peaked current transients that follow the field dependence of eqn (6). We will come back to (and rule out) this scenario in the discussion of Fig. 7 below.

Similar behavior as in Fig. 6 has been observed for the polymeric ferroelectric PVDF by Ishii *et al.*,^{38,39} who interpreted their results in the frameworks laid out by Hayashi³⁶ and Fatuzzo.³⁷ Hayashi has shown that at high fields the size of the critical 2D nucleus can become smaller than the lattice constant of the system, giving rise to a transition to 1D nucleation and a nucleation rate that depends on the field as a power law.³⁶ For the parameters investigated in ref. 36 an exponent $n \approx 1.54$ is found, with the notion that *n* depends on the critical size of the 1D nucleus and the domain boundary energies. In view of the quasi-1D columnar morphology of the aligned BTA thin films, a field-induced transition to 1D nucleation seems intuitively plausible.

Fatuzzo rationalizes a transition from a purely exponential field dependence of the switching time (eqn (5)) to the shape eqn (6) with n = 4/3 in terms of a transition from nucleation limited to apparent sideways domain wall motion to true sideways domain wall motion.³⁷ As was acknowledged by Fatuzzo, it is difficult to justify the assumption of apparent to true sideways domain growth. For the present case a cooperative effect may be expected for the switching of neighboring columns, but this might as well lead to apparent as to real sideways growth of domains. We finally notice that the predicted exponent differs significantly from the best fit to our data, making this model less favored.

Summarizing the discussion above, the data in Fig. 6 alone do not allow a clear conclusion regarding the physical process(es) limiting the switching kinetics. Turning to the activation energies extracted from the data in Fig. 6(c), 55 kJ mol⁻¹ for BTA-C10 and 56 kJ mol⁻¹ for BTA-C18, we notice that these numbers are a factor 2–3 lower than was previously found for the polarization-reversing R-relaxation.⁶ R-relaxation being a collective reversal of the amide bonds in a ferroelectric domain containing several BTA units as discussed above. This notion is consistent with the switching mechanisms discussed in the context of the data in Fig. 5 and 6 that all imply that bulk polarization reversal is not rate-limiting for polarization reversal driven by an electric field exceeding the coercive field.

In the DWM one measures either the negative or – in our case – the positive part of the hysteresis loop. In Fig. 7, the negative part has been added by a simple mirroring as no significant difference with negative switching has been observed. The bump that is visible in the P(E) plot of the Fig. 7(a) only

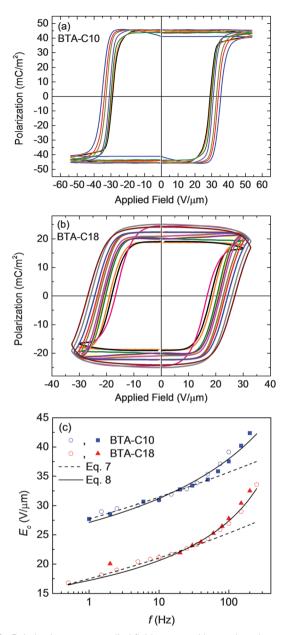


Fig. 7 Polarization *versus* applied field measured in quasi-static mode on Au/BTA-C10/Au (a) and Au/BTA-C18/Au (b) capacitors at different frequencies, taken at 100 °C. (c) Corresponding coercive field *versus* switching frequency for BTA-C10 (blue circles and squares) and BTA-C18 (red pentagons and triangles). Open and closed symbols are taken in quasi-static and dynamic mode, respectively. The hysteresis loops corresponding to the dynamic mode are given in the ESI,† Fig. S7. The dashed lines are fits to the Ishibashi–Orihara power law (eqn (7)), the solid lines are fits to the Du–Chen model (eqn (8)).

appears at high frequencies due to the interaction between preparation and measurement signals and the increasing influence of the displacement part resulting from the inertia of the ferroelectric processes.

In Fig. 7 we investigate the frequency dependence of the (apparent) coercive field for both BTA-C10 and BTA-C18. As is common for ferroelectric materials E_c shifts to higher values with increasing frequency f due to slower switching domains

starting to lag behind the driving signal. In the KAI framework, such shifts are commonly interpreted using the Ishibashi– Orihara (KAI) power law,

$$E_{\rm c} \propto f^{d/\alpha}$$
 (7)

with *d* the effective dimensionality of domain growth and α an empirical parameter that depends on the waveform. For sinusoidal and triangular waves one has $\alpha \approx 6$.^{40,41} The best fit to the data is obtained for $d/\alpha \approx 0.06$ for BTA-C10 and 0.08 for BTA-C18, see the thin dashed lines in Fig. 7. Clearly, eqn (7) fails to capture the full frequency dependence. Moreover, even though the values found for d/α are in the range reported for other ferroelectric materials,⁴¹ they are inconsistent with $d \approx 2-3$ (Fig. 6) and $\alpha \approx 6$. In view of this and the analysis of Fig. 5 and 6, we can rule out domain growth as the rate limiting process for our amide-based organic ferroelectrics.

Interestingly, assuming a Fowler–Nordheim tunnelinglimited onset of the switching process as discussed at Fig. 6 above, leads to a switching rate and hence a characteristic frequency that is proportional to the injected current $f \propto j_{\rm FN} \propto E^2 \exp(-E_0/E)$ with E_0 a parameter with dimension V m⁻¹ that depends on the injection barrier. Fitting this to the data in Fig. 7 leads to a functional shape that is indistinguishable from the KAI model eqn (7) (not shown). Hence, a rate-limiting Fowler–Nordheim process is not consistent with the experimental data.

Alternatively, the Du–Chen model assumes switching is limited by the nucleation process which in turn is described as the random, thermally driven formation of a nucleus exceeding the critical size.^{42–44} This gives rise to an apparent coercive field that depends on the frequency as

$$E_{\rm c} = \frac{B'}{\sqrt{k \cdot T \cdot \ln(f_0/f)}} \tag{8}$$

with $B' = \frac{2 \cdot \sqrt{\pi} \cdot \gamma_{\rm B} \cdot \sqrt{\gamma}}{\Delta P}$ a parameter that depends on the domain wall energy γ , the binding energy between structural defects and the wall $\gamma_{\rm B}$, the polarization change due to passing of the domain wall *P* and that here will be used as empirical fitting parameter. Furthermore, f_0 is the limiting domain switching frequency, $k_{\rm B}$ the Boltzmann constant and *T* the absolute temperature. Applying this model to the data in Fig. 7(c) (thin solid lines) gives an excellent match over the full frequency range. Fitting parameters for BTA-C10 are $f_0 = 12$ kHz for both quasi-static and dynamic modes. For BTA-C18 we find $f_0 = 2$ kHz. In combination with the interpretation of Fig. 5 and 6 above a consistent physical picture arises by assuming that the switching kinetics are dominated by a nucleation process with a roughly log-normal distribution of waiting times. At higher fields these nuclei are most likely quasi-1D.

3. Conclusions

The dipolar properties of trialkylbenzene-1,3,5-tricarboxamides (BTA) were investigated by a combination of techniques. True ferroelectric behavior was demonstrated using dielectric relaxation

spectroscopy, in which a ferroelectric–paraelectric phase transition at the Curie temperature was observed, with strict Curie–Weiss behavior at the paraelectric side. Depth-resolved pyroelectric measurements revealed a thin non-pyroelectric interfacial layer, separating the contacts from the switching ferroelectric bulk. The kinetic switching behavior of BTA-bases metal–insulator– metal diodes indicates that the switching kinetics are dominated by domain nucleation with a highly dispersive log-normal distribution of switching times. At higher fields these nuclei are most likely quasi-1D, consistent with the 1D columnar morphology of aligned BTA films. Jointly, these results univocally show that BTAs are true ferroelectrics, albeit of a disordered type.

4. Experimental section

Thin film metal-insulator-metal (MIM) capacitor devices were prepared as follows. Bottom contacts from gold (70 nm) were deposited by thermal evaporation under vacuum conditions $(\sim 3 \times 10^{-7}$ Torr) through a shadow mask on chemically cleaned glass substrates. Prior to Au deposition 5 nm Cr was deposited to promote adhesion. The liquid crystalline BTA-C10 and BTA-C18 materials were spin-coated at 500 rpm from solutions of 40 mg dry BTA-C10 and 35 mg dry BTA-C18 dissolved by 1 ml of chloroform. Obtained thin films were annealed at 70-80 °C for 10-15 min for complete solvent evaporation. Afterwards, top contacts from gold (70 nm) were evaporated through a 90°-rotated shadow mask as described above, giving rise to a cross-bar diode at the intersection of the contacts with a device area between 0.25 and 0.5 mm². Typical film thicknesses were 600-700 nm for BTA-C10 and 350-400 nm for BTA-C18, as measured by a Dektak XT profilometer.

Homeotropic (perpendicular) alignment of columnar liquid crystalline BTAs in LC cells and thin-film capacitor devices was achieved by an external electric field above the coercive field. For MIM capacitors alignment was performed at elevated temperatures of 100–110 °C; for LC cells 150–170 °C was used. The electrical alignment consisted of 30–60 minutes application of a 0.5 Hz triangular wave with an external field amplitude $\approx 30 \text{ V } \mu \text{m}^{-1}$. The positive and negative parts of the triangular wave had equal amplitude but durations that differed by a factor 2. This alignment technique is more robust in the sense of avoiding device short-circuiting than application of DC field for a shorter time. After achieving the homeotropically aligned state devices were cooled down to the desired temperatures for ferroelectric switching measurements.

Homeotropically aligned thin-film capacitor devices were electrically characterized in the dark inside a Janis probe station. Alignment and switching signal waveforms were applied by an Agilent 33120a arbitrary waveform generator and amplified by a Falco WMA-300 or TReK Model PZD350A high voltage amplifier. The actual circuit current was measured by a Keithley 6485 picoammeter, visualized and saved *via* an Agilent DSO7104A oscilloscope. In dynamic mode, displacement-field (D-E) hysteresis loops were obtained *via* direct integration of the measured positive and negative current responses under a

triangular periodic switching signal. Quasi-static measurements were performed using the double-wave method (DWM) in which polarization-field (P–E) hysteresis loops were obtained by integration of the pure switching-current which was calculated as the difference between the full switching and background responses. The DWM is described in detail in ref. 16 and 17. For the stepresponse switching experiments a similar background current subtraction scheme as in the DWM was used.

Dielectric relaxation spectra $\varepsilon^*(\omega) = \varepsilon'(\omega) - i\varepsilon''(\omega)$ were obtained using a high precision dielectric analyser (ALPHA analyser, Novocontrol Technologies) in the frequency range of 0.1 Hz to 10⁷ Hz in combination with a Novocontrol Quatro temperature system providing control of the sample temperature with high stability of 0.05 K. BTA-filled Linkam ITO-cells were used as samples in a parallel plate geometry. After measurement, the high frequency parasitic cell peak was partially corrected using a lumped circuit approach.

Local polarization depth profiles were obtained with a stateof-the-art LIMM setup (Laser Intensity Modulation Method). In its most basic form, the LIMM setup consists of a frequency generator (Tektronix AFG3021), which modulates the power of a laser diode (658 nm, 10 mW, Lisa Laser, HL25/M2) onto a thin sheet of sample material onto which electrodes are vacuum deposited, *i.e.* a MIM device is measured. These electrodes are connected to a high bandwidth current amplifier (Femto Messtechnik, HCA-5M-500K) and a Lock-in amplifier (SR 844) to isolate the pyroelectric current and correlate it to its real and imaginary part. An external bias field is applied by connecting the voltage output of a Keithley 237 Source-Measure unit to the sample electrodes with a Tektronix 2220 Voltage Probe. A 100 nF ceramic capacitor was used to block the DC contribution from the bias field to the current amplifier. The whole setup is automated via a MatLab script.

Surface topographies of spin-coated BTA films after annealing were measured by atomic force microscopy (AFM) using a Veeco MultiMode in tapping-mode at room temperature. Measurements have been done using Si AFM cantilevers (Nanosensors PPP, spring constant $k \approx 42$ N m⁻¹, apex radius <10 nm).

Acknowledgements

The work of A. V. G. was supported by the NWO Nano program. I. U. acknowledges funding by Vetenskapsrådet.

References

- K. Kishikawa, S. Nakahara, Y. Nishikawa, S. Kohmoto and M. Yamamoto, *J. Am. Chem. Soc.*, 2005, **127**, 2565.
- 2 S. Horiuchi and Y. Tokura, Nat. Mater., 2008, 7, 357.
- 3 S. Horiuchi, Y. Tokunaga, G. Giovannetti, S. Picozzi, H. Itoh, R. Shimano, R. Kumai and Y. Tokura, *Nature*, 2010, 463, 789.
- 4 D. Miyajima, F. Araoka, H. Takezoe, J. Kim, K. Kato, M. Takata and T. Aida, *Science*, 2012, 336, 209.
- 5 C. F. C. Fitié, W. S. C. Roelofs, M. Kemerink and R. P. Sijbesma, *J. Am. Chem. Soc.*, 2010, **132**, 6892.

- 6 C. F. C. Fitié, W. S. C. Roelofs, P. C. M. M. Magusin, M. Wübbenhorst, M. Kemerink and R. P. Sijbesma, *J. Phys. Chem. B*, 2012, **116**, 3928.
- 7 A. S. Tayi, A. K. Shveyd, A. C.-H. Sue, J. M. Szarko, B. S. Rolczynski, D. Cao, T. J. Kennedy, A. A. Sarjeant, C. L. Stern, W. F. Paxton, W. Wu, S. K. Dey, A. C. Fahrenbach, J. R. Guest, H. Mohseni, L. X. Chen, K. L. Wang, J. F. Stoddart and S. I. Stupp, *Nature*, 2012, **488**, 485.
- 8 H. Anetai, Y. Wada, T. Takeda, N. Hoshino, S. Yamamoto, M. Mitsuishi, T. Takenobu and T. Akutagawa, *J. Phys. Chem. Lett.*, 2015, 6, 1813.
- 9 J. Guilleme, E. Cavero, T. Sierra, J. Ortega, C. L. Folcia, J. Etxebarria, T. Torres and D. González-Rodríguez, *Adv. Mater.*, 2015, 27, 4280.
- 10 M. Li, H. J. Wondergem, M.-J. Spijkman, K. Asadi, I. Katsouras, P. W. M. Blom and D. M. de Leeuw, *Nat. Mater.*, 2013, **12**, 433.
- I. Katsouras, K. Asadi, M. Li, T. B. van Driel, K. S. Kjær, D. Zhao, T. Lenz, Y. Gu, P. W. M. Blom, D. Damjanovic, M. M. Nielsen and D. M. de Leeuw, *Nat. Mater.*, 2016, 15, 78.
- 12 A. L. Sobolewski, Phys. Chem. Chem. Phys., 2015, 17, 20580.
- 13 M. Wawrzyniak-Adamczewska and M. Wierzbowska, J. Phys. Chem. C, 2016, **120**, 7748.
- 14 A. Rochefort, É. Bayard and S. Hadj-Messaoud, *Adv. Mater.*, 2007, **19**, 1992.
- 15 A. S. Tayi, A. Kaeser, M. Matsumoto, T. Aida and S. I. Stupp, *Nat. Chem.*, 2015, 7, 281.
- 16 M. Fukunaga and Y. Noda, J. Phys. Soc. Jpn., 2008, 77, 64706.
- 17 V. Khikhlovskyi, A. V. Gorbunov, A. J. J. M. van Breemen, R. A. J. Janssen, G. H. Gelinck and M. Kemerink, *Org. Electron.*, 2013, 14, 3399.
- 18 T. Shikata, Y. Kuruma, A. Sakamoto and K. Hanabusa, J. Phys. Chem. B, 2008, 112, 16393.
- 19 P. J. M. Stals, J. C. Everts, R. de Bruijn, I. A. W. Filot, M. M. J. Smulders, R. Martín-Rapún, E. A. Pidko, T. F. A. de Greef, A. R. A. Palmans and E. W. Meijer, *Chem. – Eur. J.*, 2010, **16**, 810.
- 20 R. Q. Albuquerque, A. Timme, R. Kress, J. Senker and H.-W. Schmidt, *Chem. Eur. J.*, 2013, **19**, 1647.
- 21 A. Timme, R. Kress, R. Q. Albuquerque and H.-W. Schmidt, *Chem. – Eur. J.*, 2012, **18**, 8329.
- 22 M. Wübbenhorst and J. van Turnhout, J. Non-Cryst. Solids, 2002, **305**, 40.
- 23 M. Serwadczak, M. Wübbenhorst and S. Kucharski, J. Non-Cryst. Solids, 2007, 353, 4303.
- 24 S. B. Lang and D. K. Das-Gupta, J. Appl. Phys., 1986, 59, 2151.
- 25 T. Putzeys and M. Wübbenhorst, *Phys. Chem. Chem. Phys.*, 2015, **17**, 7767.
- 26 W. J. Merz, Phys. Rev., 1954, 95, 690.
- 27 W. J. Merz, J. Appl. Phys., 1956, 27, 938.
- 28 H. Orihara, S. Hashimoto and Y. Ishibashi, J. Phys. Soc. Jpn., 1994, 63, 1031.
- 29 Y. Ishibashi and H. Orihara, Integr. Ferroelectr., 1995, 9, 57.
- 30 X. J. Lou, J. Phys.: Condens. Matter, 2009, 21, 12207.
- 31 O. Lohse, M. Grossmann, U. Boettger, D. Bolten and R. Waser, J. Appl. Phys., 2001, 89, 2332.

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- 32 J. Y. Jo, H. S. Han, J.-G. Yoon, T. K. Song, S.-H. Kim and T. W. Noh, Phys. Rev. Lett., 2007, 99, 267602.
- 33 Y. A. Genenko, S. Zhukov, S. V. Yampolskii, J. Schütrumpf, R. Dittmer, W. Jo, H. Kungl, M. J. Hoffmann and H. von Seggern, Adv. Funct. Mater., 2012, 22, 2058.
- 34 A. K. Tagantsev, I. Stolichnov, N. Setter, J. S. Cross and M. Tsukada, Phys. Rev. B: Condens. Matter Mater. Phys., 2002, 66, 214109.
- 35 A.-Q. Jiang, H. J. Lee, C. S. Hwang and T.-A. Tang, Phys. Rev. B: Condens. Matter Mater. Phys., 2009, 80, 24119.
- 36 M. Hayashi, J. Phys. Soc. Jpn., 1972, 33, 616.
- 37 E. Fatuzzo, Phys. Rev., 1962, 127, 1999.

- 38 H. Ishii, T. Nakajima, Y. Takahashi and T. Furukawa, Appl. Phys. Express, 2011, 4, 31501.
- 39 H. Ishii, T. Nakajima, T. Furukawa and S. Okamura, Jpn. J. Appl. Phys., 2013, 52, 41603.
- 40 S. Hashimoto, H. Orihara and Y. Ishibashi, J. Phys. Soc. Jpn., 1994, 63, 1601.
- 41 J. F. Scott, Integr. Ferroelectr., 1996, 12, 71.
- 42 X. Du, I.-W. Chen, Symposium U Ferroelectric Thin Film VI, MRS Online Proceedings Library Archive, 1997, vol. 493.
- 43 Takaakitsurumi, S.-M. Num, Y.-B. Kil and S. Wada, Ferroelectrics, 2001, 259, 43.
- 44 S.-M. Nam, Y.-B. Kil, S. Wada and T. Tsurumi, Jpn. J. Appl. Phys., 2003, 42, L1519.

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