The Serbian Society for Ceramic Materials

Institute for Multidisciplinary Research, University of Belgrade

Institute of Physics, University of Belgrade

Center of Excellence for the Synthesis, Processing and Characterization of Materials for use in Extreme Conditions "CEXTREME LAB" - Institute of Nuclear Sciences "Vinča", University of Belgrade

Faculty of Mechanical Engineering, University of Belgrade

# PROGRAMME and the BOOK of ABSTRACTS

4CSCS-2017

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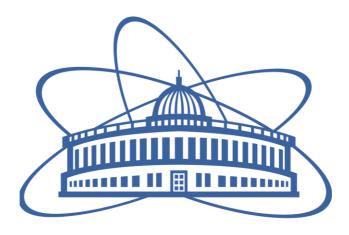
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#### **SPECIAL THANKS TO**



JOINT INSTITUTE FOR NUCLEAR RESEARCH

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#### INFLUENCE OF SYNTHESIS CONDITIONS ON MORPHOLOGICAL FEATURES OF SBA-15

<u>Maja Kokunešoski</u><sup>1</sup>, Aleksandra Šaponjić<sup>1</sup>, Zvezdana Baščarević<sup>2</sup>, Zlatko Rakočević<sup>1</sup>, Đorđe Šaponjić<sup>1</sup>, Branko Matović<sup>1</sup>, Biljana Babić<sup>1</sup>

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Mesoporous SBA-15 materials, of high specific surface area, (~700 m² g⁻¹), were synthesized by using block copolymer Pluronic P123 as a template and tetraethoxysilane as a silica source. The obtained materials were characterized by X-ray diffraction, scanning electron microscopy (SEM), energy dispersive X-ray (EDS) analysis, and atomic force microscopy (AFM). It was found that small modifications of synthesis conditions influenced morphological features of the synthesized SBA-15 samples. The SEM analysis has shown that the SBA-15 synthesized at a lower temperature and a longer time of reaction (80 °C, 48 h) provided elongated rod-shaped grains about 1μm long. The other sample synthesized at a higher temperature and a shorter time of reaction (100 °C, 24 h) has rounded grains and grains of regular spherical shape with diameters up to 2 μm. The EDS analysis confirmed that the particles of both synthesized samples were of SiO<sub>2</sub> content. In addition, the AFM analysis has shown different surface morphologies of the materials synthesized under various conditions.

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### NANOSCALE-TO-BULK: SIZE-DEPENDENT CRYSTALLINITY OF CERAMIC OXIDES

#### Stefan Bromley<sup>1,2</sup>

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Naturally abundant and inherently a reactive electronegative element, oxygen chemically bonds with almost all other atoms to produce a corresponding oxide. Numerous ceramic oxide materials have had a massive technological impact in fields as diverse as catalysis, abrasives, absorbents, photovoltaics, and electronics. Key to this success is the wide range of chemical and physical properties exhibited