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1 Nanostructured Au(111)/Oxide Epitaxial Heterostructures with 2 Tailoring Plasmonic Response by a One-Step Strategy

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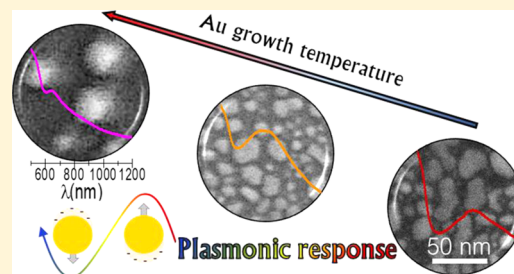
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11 Supporting Information

12 **ABSTRACT:** In this work, we present a strategy for developing epitaxial
13 incommensurate nanostructured Au/oxide heterostructures with tunable
14 plasmonic response. Previously, high-quality single-phase and single-
15 oriented α -Fe₂O₃(0001) thin films were achieved, which have been used as
16 a template for noble-metal epitaxial deposition. The complex systems have
17 been grown by pulsed laser deposition on two different types of oxide
18 substrates: α -Al₂O₃(0001) and SrTiO₃(111). A one-step procedure has
19 been achieved tailoring the isolated character and the morphological
20 features of Au nanostructures through the substrate temperature during Au
21 growth, without altering the structural characteristics of the hematite layer
22 that is identified as a single iron oxide phase. The epitaxial character and the lattice coupling of Au/oxide bilayers are mediated
23 through the sort of oxide substrate. Single-oriented Au(111) islands are disposed with a rotation of 30° between their
24 crystallographic axes and those of α -Fe₂O₃(0001). The Au(111) and SrTiO₃(111) lattices are collinear, while a rotation of 30°
25 happens with respect to the α -Al₂O₃(0001) lattice. The crystallographic domain size and crystalline order of the hematite
26 structure and the Au nanostructured layer are dependent on the substrate type and the Au growth temperature, respectively.
27 Besides, the functional character of the complex systems has been tested. The localized surface plasmons related to Au
28 nanostructures are excited and controlled through the fabrication parameters, tuning the optical resonance with the degree of
29 Au nanostructuring.



1. INTRODUCTION

30 Hybrid nanomaterials receive a great attention from both
31 fundamental and applied points of view due mainly to coupling
32 effects and enhanced properties over the single compo-
33 nents.^{1–6} In this aspect, these materials in which one of the
34 constituents is based on noble metals exhibiting surface
35 plasmon resonance (SPR) response suppose an excellent
36 system for active applications as the heterogeneous cataly-
37 sis.^{1,2,5,7,8} For example, noble nanoparticles (NPs) supported
38 on iron oxides are highly attractive to improve the catalytic
39 activity.^{1,2,9} Besides, the surface of epitaxial iron oxide films has
40 been shown as an appealing framework to support the noble-
41 metal nanostructures (NSs) exhibiting a large rate of
42 oxidation.^{10–12} Specifically, α -Fe₂O₃ (hematite) is one of the
43 iron oxides with a high potential for numerous applications due
44 to its richness, nontoxic nature, and high chemical stability.
45 This iron oxide polymorph has already been proposed as an
46 active oxide for CO oxidation at low temperature and for water
47 splitting processes.^{13,14} In addition, a great photochemical
48 reactivity for the Ag reduction has been proven in the specific

configuration of the epitaxial α -Fe₂O₃ film prepared on
SrTiO₃(111).¹⁵

Au NSs have been extensively studied in a wide range of
investigation areas such as catalysis, magnetism, or plas-
monics.^{16–19} Specially, integrated or supported Au NPs are
quite acclaimed due to their active response presenting SPR to
functionalize materials.^{1–3,9,20–22} The final response of noble-
metal NSs is dependent on morphological and structural
parameters such as the geometry, the crystallinity, or the
surrounding media among others, favoring to control the final
properties with the growth conditions and so tailoring the
implications in each field.

An interesting method to change in a controlled way the
morphological features of noble metallic NSs on oxide surfaces
is the well-known dewetting process.^{23,24} Dewetting of metallic
thin films can be achieved in several ways. The most employed

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65 method to achieve a nanostructured character from an initial
66 continuous film is by means of a *posteriori* thermal treatment at
67 a defined temperature.^{19,25} Other methodologies to induce the
68 dewetting mechanism can be by the irradiation with an
69 electron or ion beam on the desired region^{26,27} or by using a
70 prestructured template on which the film is submitted to the
71 dewetting process.²⁸ Other method much less widespread but
72 advantageous consists in the dewetting process during the film
73 growth retaining the heterostructure up to a certain temper-
74 ature and so avoiding a subsequent annealing process after the
75 film deposition.^{10,29} In this way, the control of the resulting
76 morphology of NSs is also possible tuning accurately the
77 growth parameters. During the growth process in which the
78 substrate is maintained at a specific temperature, the mobility
79 of the metallic atoms on the surface is really large enhancing
80 the superficial diffusion process and achieving a stage of three-
81 dimensional (3D) islands. In this sense, the nanostructuration
82 can be tailored with the substrate temperature.

83 In the present work, epitaxial Au(111) NPs are formed on
84 oxide heterostructures (α -Fe₂O₃ layer/oxide substrate) by a
85 one-step strategy in which the control of the morphological
86 and structural parameters is performed through the substrate
87 temperature during the Au growth by subsequent deposition,
88 i.e., without breaking the vacuum in the growth chamber and
89 without a *posteriori* post-treatment of samples. The effect of the
90 underlying substrate is evaluated by using two different oxide
91 substrates, which induces different macroscopic behavior on
92 the oxide heterostructure surface. Besides, we discuss the
93 ability of the growth process to achieve the Au nano-
94 structuration without alteration of the iron oxide properties
95 and tuning the plasmonic activity of complex system.

96 Based on a previous work,³⁰ we use high-quality single-phase
97 and single-oriented α -Fe₂O₃(0001) thin films grown on
98 SrTiO₃ (STO)(111) and α -Al₂O₃ (AO)(0001) oxide sub-
99 strates as a bare template for the Au NPs deposition. The
100 identification of the lattice coupling mechanism and the
101 morphological and structural characteristics of the hematite
102 layers as a function of lattice parameter and crystallographic
103 orientation of substrate were studied in such a work.³⁰ We
104 concluded that high-quality epitaxial hematite layers with
105 bulklike structure and free of oxygen vacancies defects were
106 obtained on both substrates. However, although the hematite
107 layers behaved identical on both oxide substrates in terms of
108 microscopic roughness and interface abruptness, the macro-
109 scopic surface and interface quality of the hematite layers were
110 different as revealed by the presence and absence of Kiessig
111 fringes in the X-ray reflectivity (XRR), specular and off-
112 specular X-ray diffraction (XRD) patterns. Such a difference
113 may induce important modifications on the nanostructuration
114 process of the Au. Hence, the achieved α -Fe₂O₃(0001)/
115 STO(111) and α -Fe₂O₃(0001)/AO(0001) systems become
116 ideal candidates as templates for the study of the formation
117 and morphological tuning of Au NPs on oxide hetero-
118 structures.

2. EXPERIMENTAL METHODS

119 Epitaxial Au NPs were grown by pulsed laser deposition
120 (PLD) on α -Fe₂O₃(0001)/AO(0001) and α -Fe₂O₃(0001)/
121 STO(111) oxide heterostructures. The growth was performed
122 by using an ultraviolet (UV) pulsed laser source (355 nm) of
123 high power (1 W) to obtain the plasma and a base pressure of
124 10⁻⁹ mbar. Previously, continuous and homogeneous iron
125 oxide (hematite) films were deposited by PLD under an O₂

atmosphere (P_{O₂} of 10⁻⁴ mbar) with the AO(0001) and
STO(111) substrates at 400 °C using an α -Fe₂O₃ target,
following the same procedure as that in our previous work.³⁰
Subsequently, nanostructured Au layers were grown on
hematite layers from a Au target keeping the oxygen pressure
P_{O₂} about 10⁻⁴ mbar at three different substrate temperatures:
250, 500, and 750 °C. Deposition dose (mass thickness) was
fixed for iron oxide and Au, respectively. During the deposition
process, the crystalline character of both layers was confirmed
by in situ reflection high-energy electron diffraction obtaining
intense patterns from the initial stages of deposition, as shown
in the Supporting Information (SI). Table 1 shows the set of
samples selected for this work.

Table 1. Representative Nanostructured Au/ α -Fe₂O₃/oxide Substrate Systems Selected for this Work, in Which the Type of Substrate and the Substrate Temperature for the Au Growth are Varied

sample	substrate	Au growth temperature (°C)
AO_250	AO(0001)	250
AO_500	AO(0001)	500
AO_750	AO(0001)	750
STO_250	STO(111)	250
STO_500	STO(111)	500
STO_750	STO(111)	750

The morphology of the Au NPs on oxide heterostructures
was studied by scanning electron microscopy (SEM) with an
S-4700 Hitachi instrument at 20 kV, and the morphological
features were examined by using ImageJ software. Grazing
incidence X-ray diffraction (GIXRD), XRR, and X-ray
absorption spectroscopy (XAS) were carried out at the
Spanish CRG synchrotron beamline BM25-SpLine at The
ESRF—The European Synchrotron, Grenoble (France).
GIXRD and XRR measurements were carried out in a high-
precision six-circle diffractometer in vertical geometry³¹ at
room temperature (RT) using a photon wavelength of 0.826 Å
($h\nu = 15$ keV). XAS measurements were performed at the Fe
K-edge (7112 eV) and the Au L₃-edge (11919 eV) at RT in
fluorescence detection mode using a 13-element Si(Li)
detector located 90° with respect to the incoming X-ray
beam. Spectra of reference standards (Au foil, Fe foil, FeO,
Fe₃O₄, α -Fe₂O₃, and γ -Fe₂O₃) were collected for comparison
in transmission detection mode with two ionization chambers
as detectors. XAS data were analyzed using the Demeter
package.³² Optical absorption measurements were collected in
transmission mode with an UV–vis–NIR Shimadzu 3100
double-beam spectrophotometer equipped with an integrating
sphere. Spectra were recorded at RT from 400 to 1400 nm
with a spectral resolution of 0.5 nm.

3. RESULTS AND DISCUSSION

3.1. Morphological and Structural Characterization: Au Nanostructuration. The growth of noble-metal thin films
on oxide surfaces is associated with a behavior in which the
film deposition proceeds through the nucleation of islands that
coalesce resulting in a continuous film. Specifically, 3D islands
can be presented directly as a result of the deposition under
certain film thickness and specific conditions, being possible to
control the final structures.^{33,34} Upon annealing, the dewetting
mechanism of the metallic film is activated to relieve the

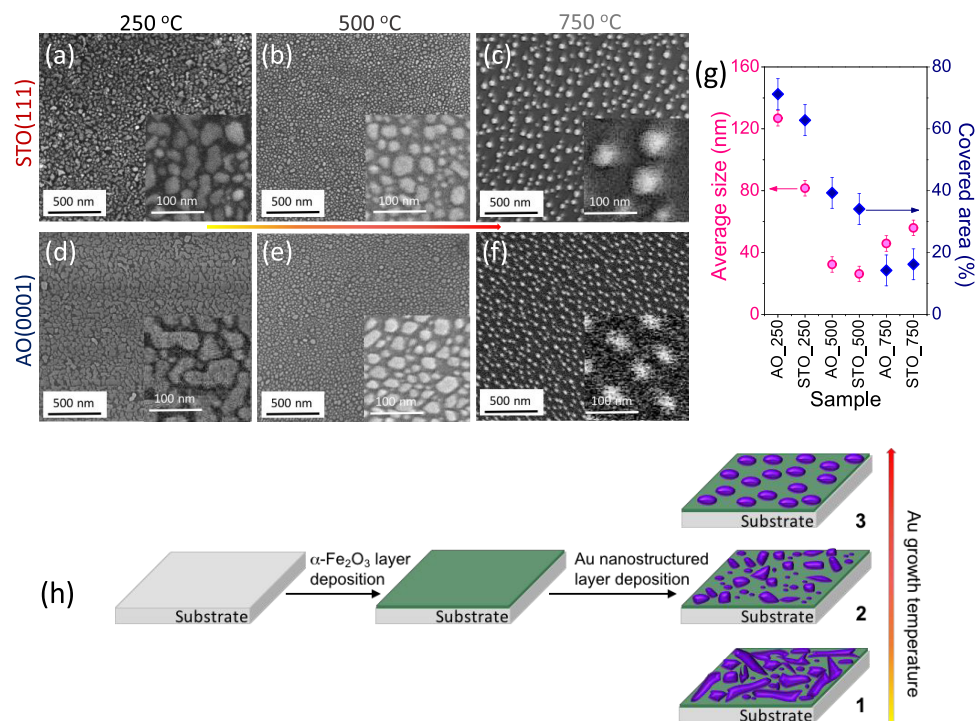


Figure 1. (a–f) SEM images and (g) average size and covered area obtained from SEM images for Au NPs prepared on α -Fe₂O₃(0001)/AO(0001) and α -Fe₂O₃(0001)/STO(111) oxide heterostructures controlling the growth temperature (250, 500, and 750 °C) of the nanostructured Au layer. (h) Representative scheme of growth for the heterostructure system varying the Au growth temperature (1 → 3).

172 mechanisms of surface and interfacial stress, giving rise to
 173 morphological modifications that have been extensively
 174 explored.^{19,35–37} In this process, the appearance of holes and
 175 their subsequent growth is promoted by the surface diffusion,
 176 increasing several orders of magnitude in the presence of
 177 oxygen, until the holes percolate leading to the formation of
 178 islands.^{3,23,38,39} In this work, a different procedure is performed
 179 in which epitaxial Au nanostructured layers are prepared on
 180 hematite surface from high-quality epitaxial thin films (see ref
 181 30) at a certain substrate temperature in just one step.

182 **Figure 1** shows the SEM images for Au NPs prepared on α -
 183 Fe₂O₃(0001)/AO(0001) and α -Fe₂O₃(0001)/STO(111)
 184 oxide heterostructures varying the substrate temperature in
 185 the growth process of the top Au layer. A nanostructured top
 186 layer is obtained irrespective of the substrate and its
 187 temperature selected for the Au growth. However, a clear
 188 evolution of the nanostructured layer features is attained as a
 189 function of the substrate temperature: the larger the temper-
 190 ature during the Au growth, the larger is the nanostructuring
 191 of layer attaining more rounded and isolated Au NPs. An
 192 analysis of the morphological parameters gives values of the
 193 covered surface area from 70 to 15%, diminishing as the Au
 194 growth temperature increases (see **Figure 1g**). With respect to
 195 the average size of Au NSs, we identify a decrease from around
 196 100 to 30 nm when the Au growth temperature passes from
 197 250 to 500 °C, while for the sample prepared at 750 °C, a
 198 slight size increase to around 50 nm is obtained. Specifically, at
 199 the highest temperatures, we find more homogeneous Au
 200 islands that may be an indication of a more pronounced
 201 Oswald's ripening. Besides, seemingly, the morphological
 202 effects related to the used substrate and therefore to the
 203 characteristics of hematite layers in the final Au islands and
 204 their distribution are slight or not presented. Results expected
 205 taking into account the slight differences in the values of grain

average height and rms roughness of the bare α -Fe₂O₃ films
 grown on AO(0001) and STO(111) substrates.³⁰

Hence, the nanostructuring process of Au layer on α -
 Fe₂O₃/oxide substrate heterostructures is renowned and the
 proposed one-step growth procedure provides qualitatively
 similar morphological results to those in a post-annealing
 process of a continuous metallic film. A representative scheme
 of the growth mechanism for nanostructured Au/ α -
 Fe₂O₃(0001)/oxide substrate system depending on the
 substrate temperature for the Au growth is represented in
Figure 1h. During the Au growth, the Au atoms diffuse as soon
 as they arrive on the hematite surface because the binding
 energy between the Au atoms is larger than that between the
 hematite and Au atoms resulting in Au NSs. Such a
 nanostructuring is more stable than a continuous thin film.
 As depicted, the superficial diffusion processes are increased
 with the substrate temperature during the Au growth,
 generating a larger Au nanostructuring and islands with
 certain morphological characteristics (1 → 3).

High-resolution GIXRD and XRR techniques were
 employed to analyze the substrate effect and the Au growth
 temperature on the crystallographic character of the system.
 The layer thickness was calculated from low-angle XRR
 measurements obtaining 27 ± 2 nm for the hematite layers and
 6 ± 1 nm for the Au layers grown at lower temperature (see
SI). Representative high-angle XRR measurements for the Au/
 α -Fe₂O₃/oxide substrate heterostructures are shown in **Figure**
2a together with spectra for single α -Fe₂O₃ thin films grown on
 AO(0001) and STO(111) substrates, for comparison. The
 deposition of Au on top of the hematite layers results in a wide
 peak for low Au growth temperature masking the diffraction
 peak related to the epitaxial α -Fe₂O₃ layer, especially for the
 layer grown on STO(111). Increasing the growth temperature
 of noble metal, the Au peak becomes narrower and the

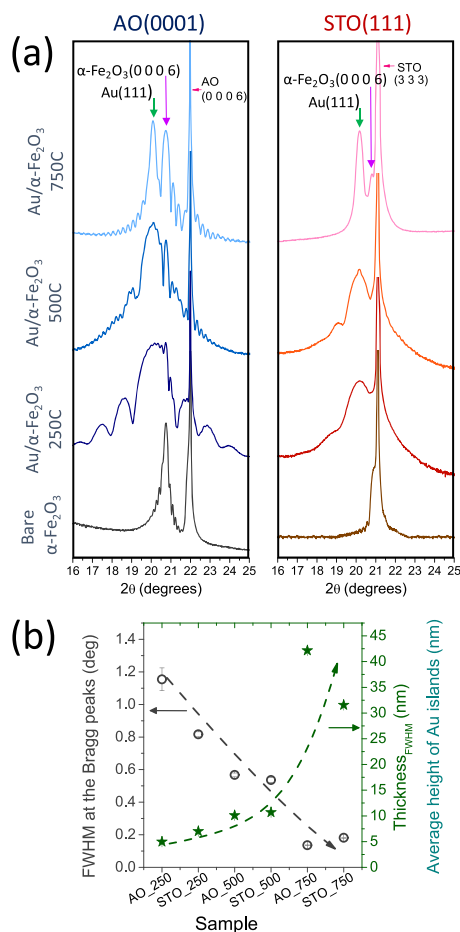


Figure 2. (a) High-angle θ - 2θ scans for Au NPs prepared on α - $\text{Fe}_2\text{O}_3(0001)/\text{AO}(0001)$ and α - $\text{Fe}_2\text{O}_3(0001)/\text{STO}(111)$ oxide heterostructures varying the substrate temperature (250, 500, and 750 °C) for the growth of the Au nanostructured layers. GIXRD measurements for bare α - Fe_2O_3 thin films are also presented. Bragg peaks related to substrate, Au, and α - Fe_2O_3 are identified. A clear modification of the diffraction signal associated with the Au is observed due to its morphological features. (b) FWHM at the Au Bragg peaks and average height of Au islands obtained from the analysis of Bragg peaks of high-angle θ - 2θ measurements.

240 diffraction peak related to hematite is more clearly
241 distinguished on both types of substrates. The high-angle
242 XRR spectra also reveal the presence of single-phase epitaxial
243 α - Fe_2O_3 with (0001) orientation, as we identified for similar α -
244 Fe_2O_3 systems in ref³⁰ and epitaxial Au with an orientation
245 (111) irrespective of the chosen substrate. Besides, it should be
246 noted that Kiessig fringes around thin-film Bragg peaks can be
247 observed, more notably for sample grown on AO(0001) as we
248 have recognized in single hematite layers grown on AO(0001)
249 substrate.³⁰ Here, other fringes periodicity related to the
250 nanostructured Au layer is recognized, indicating the
251 occurrence of high-quality and smooth surfaces and abrupt
252 hematite-substrate and Au-hematite interfaces. An in-depth
253 analysis from high-angle XRR data helps to complete the
254 morphological results provided by SEM, offering additional
255 information about the thickness of Au nanostructured layer
256 and of the average height of Au islands. From a careful analysis
257 of the full width at half-maximum (FWHM) at the Au
258 diffraction peaks, we have estimated the average height of Au
259 islands to vary from around 5 to 40 nm, which increases as the

substrate temperature increases during the Au growth, as
presented in Figure 2b.

The morphological and structural features of the final Au
islands can be attributed to diverse factors such as the
processing conditions,^{19,25,40-42} and the evolution of the
nanostructuration can be ascribed to a convolution of several
effects: the difference in the thermal expansion coefficient
between the Au ($14.2 \times 10^{-6} \text{ K}^{-1}$ at 20 °C) and the hematite
film ($8 \times 10^{-6} \text{ K}^{-1}$ at 20 °C),⁴³ the activation energy under the
specific growth conditions in oxygen atmosphere, diffusion
effects, Ostwald's ripening, and/or possible sublimation
processes, of which none of them can be discarded.

These results indicate that the most significant morpho-
logical parameters such as the average size and height of Au
islands and the hematite covered surface are dependent on the
substrate temperature for the Au growth and not significantly
on the type of oxide substrate employed in this work. However,
the epitaxial quality and abruptness of the interfaces keep
influenced by the selected substrate, which are enhanced for
samples prepared on AO(0001) substrates.

Out-of-plane lattice parameters have been calculated from
the measured peak position with respect to the substrate
obtaining $c_{\text{H/AO}} = 13.72$ (6) Å- c_{AO} and $c_{\text{H/STO}} = 13.72$ (6) Å-
 $2c_{\text{STO}}$ for hematite layers grown on AO(0001) and STO(111),
respectively, and $c_{\text{Au/H/AO}} = 7.05$ (3) Å- $1/2 c_{\text{H/AO}}$ and
 $c_{\text{Au/H/STO}} = 7.05$ (4) Å- $1/2 c_{\text{H/STO}}$ for Au layers prepared on α -
 $\text{Fe}_2\text{O}_3(0001)/\text{AO}(0001)$ and α - $\text{Fe}_2\text{O}_3(0001)/\text{STO}(111)$, res-
pectively, as resumed in Table 2. These calculated lattice
parameters are consistent with those values obtained from the
analysis of nonspecular out-of-plane scans (not shown) and the

Table 2. Domain Size and Lattice Parameters Obtained from the GIXRD Characterization for Au NPs Prepared on α - $\text{Fe}_2\text{O}_3(0001)/\text{AO}(0001)$ and α - $\text{Fe}_2\text{O}_3(0001)/\text{STO}(111)$ Oxide Heterostructures Varying the Substrate Temperature (250, 500, and 750 °C) during the Growth of the Au Nanostructured Layer

samples	domain size (Å)	out-plane lattice parameter (Å)	in-plane lattice parameter (Å)
Bare α - Fe_2O_3 on AO(0001)	400 (5)	13.72 (6)	5.08 (4)
AO_250 Au	150 (5)	7.05 (3)	2.90 (2)
α - Fe_2O_3	370 (18)	13.72 (3)	5.08 (1)
AO_500 Au	170 (5)	7.05 (3)	2.90 (3)
α - Fe_2O_3	380 (12)	13.70 (6)	5.08 (1)
AO_750 Au	200 (6)	7.05 (2)	2.90 (1)
α - Fe_2O_3	350 (16)	13.69 (2)	5.08 (1)
Bare α - Fe_2O_3 on STO(111)	250 (4)	13.72 (6)	5.08 (4)
STO_250 Au	170 (5)	7.04 (2)	2.89 (2)
α - Fe_2O_3	250 (1)	13.72 (3)	5.07 (1)
STO_500 Au	190 (10)	7.05 (4)	2.90 (1)
α - Fe_2O_3	250 (11)	13.72 (3)	5.07 (1)
STO_750 Au	230 (9)	7.05 (3)	2.90 (1)
α - Fe_2O_3	250 (9)	13.70 (6)	5.08 (2)

^aValues obtained for single α - Fe_2O_3 films, reported in ref 30, are included for comparison.

290 values estimated in single α -Fe₂O₃ layers, as previously
291 reported.³⁰

292 Figure 3 shows representative reciprocal space maps (RSM)
293 for Au NPs grown on α -Fe₂O₃(0001)/AO(0001) and α -

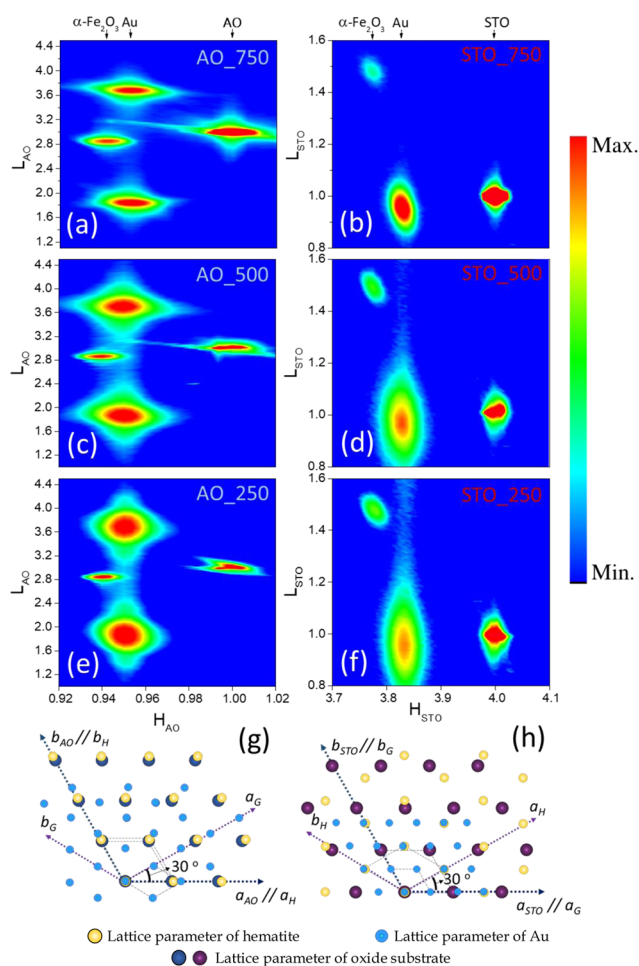


Figure 3. (a–f) LH reciprocal space maps (RSM) for Au NPs grown on α -Fe₂O₃(0001)/AO(0001) and α -Fe₂O₃(0001)/STO(111) oxide heterostructures varying the growth substrate temperature (250, 500, and 750 °C) of the Au nanostructured layer. From the in-plane diffraction measurements, we identify a noncoincidence of Au, α -Fe₂O₃, and substrates peaks, evidencing an incommensurate epitaxial growth. The color scale corresponds to the signal intensity. (g, h) Representative scheme in real space of the lattice coupling for the Au/ α -Fe₂O₃ system on both substrates. a_x and b_x represent the in-plane lattice parameters of x : S (substrate), H (hematite), and G (gold).

294 Fe₂O₃(0001)/STO(111) oxide heterostructures. Both Au and
295 hematite layers grow incommensurate as is envisaged, based on
296 the noncoincidence of the in-plane diffraction maxima from
297 both type of layers and the corresponding substrate. An
298 incommensurate growth was already established for single α -
299 Fe₂O₃ fabricated on both oxide substrates.³⁰ Here, some
300 features related to the nanostructuring of Au layer can be
301 noted on RSM measurements. The surface signal revealed for
302 Au diffraction peaks, which is characterized by the diffuse
303 signal present between Bragg peaks, is reduced as the Au
304 growth temperature increases. This result is related to a more
305 isolated character of Au islands, promoting a larger roughness
306 of the Au layer surfaces. Besides, differences in the shape of Au
307 Bragg peaks are avoided. Narrower peaks in the out-of-plane

direction of reciprocal space are obtained for the bilayers
grown on both substrates as the substrate temperature
increases in the Au growth, which is related to higher Au
islands in excellent accordance with the high-angle XRR
results.

From the RSM measurements, the coupling behavior
between the Au lattice, the hematite lattice, and the
corresponding substrate lattice is achieved. In a previous
study,³⁰ we determined an incommensurate coupling in the
 α -Fe₂O₃(0001)/STO(111) and the α -Fe₂O₃(0001)/AO(0001)
systems, in which in-plane crystallographic axes of hematite
lattice remain parallel to those of the AO(0001) substrate
while a rotation of 30° occurs between hematite and
STO(111) substrate lattices. In the present case, a similar
behavior is obtained, concluding that the lattice coupling of the
 α -Fe₂O₃ on both substrates remains unaltered after the Au
growth. With respect to the Au nanostructured layer grown on
hematite one, its in-plane crystallographic axes are rotated 30°
with respect to the hematite ones irrespectively of the
substrate. Thus, a rotation of 30° is found between the Au
and AO(0001) lattices, whereas in-plane crystallographic axes
of Au lattice are collinear with those of the STO(111)
substrate. Representative lattice models in real space of growth
for Au/ α -Fe₂O₃/oxide substrate systems are depicted in
Figure 3g,h. The lattice orientation relationships between the layers
and the substrates are revealed to be: Au/ α -Fe₂O₃/AO(0001):
(111) [110]_{Au} || (0001) [100]_H || (0001) [100]_{AO} and Au/ α -
Fe₂O₃/STO(111): (111) [110]_{Au} || (0001) [100]_H || (111)
[110]_{STO}.

The obtained in-plane real space lattice parameters of the
Au/ α -Fe₂O₃ heterostructures based on the positions at several
diffraction peaks in reciprocal space are $a_H = b_H \sim 5.08$ (4) Å
for hematite and $a_{Au} = b_{Au} \sim 2.90$ (2) Å for Au lattice, as Table
2 presents. It should be noted that epitaxial hematite thin films
grown on both substrates present the same lattice parameters
before and after the deposition of Au layers, suggesting the
stability of hematite lattice during the growth process of Au.
For all layers, the obtained lattice parameters match, within the
error, those of the bulk structure, revealing a stress-free
character.

The in-plane domain size of the grown films was also
calculated to be around 350 and 250 Å for the epitaxial
hematite in the complex system deposited on AO(0001) and
STO(111) substrate, respectively. Similar values were reported
for bare hematite layers grown on the same substrates³⁰
without the *posteriori* evaporation of Au islands, reaffirming the
nonalteration of hematite with the growth process of Au layer.
As explained in ref,³⁰ the larger domain size of the
 α -Fe₂O₃(0001)/AO(0001) film is related to the more favorable
growth based on the isostructural character between
compounds. With respect to the nanostructured Au thin
films, the in-plane domain size is estimated to be between 150
and 230 Å, following a trend with the Au growth conditions:
higher the Au growth temperature, larger the domain sizes of
epitaxial Au. Table 2 shows lattice parameters and the average
values of the domains size obtained for each epitaxial layer on
both oxide substrates employed.

Although from a structural point of view the hematite layers
remain unaltered after Au deposition, important modifications
on the iron and oxygen stoichiometry could happen as a way to
favor the incommensurate Au- α -Fe₂O₃ coupling. To discharge
possible iron oxide phase changes induced during the Au
nanostructured adlayer growth, a series of XAS measurements

371 have been carried out. From X-ray absorption near-edge
 372 structure (XANES) measurements achieved at the Fe K-edge,
 373 we corroborate the nature of hematite layers as single iron
 374 oxide phase for all samples after the deposition of Au islands.
 375 Indistinctly of oxide substrate selected and substrate temper-
 376 ature for the Au growth, the XANES spectra features follow
 377 those obtained for bare hematite layers³⁰ and for powder
 378 hematite reference, as Figure 4a shows. Fitting the XANES

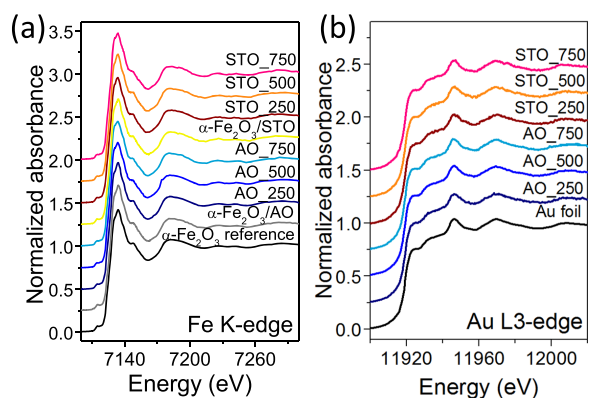


Figure 4. XANES spectra at the (a) Fe K-edge energy and (b) Au L₃-edge energy of Au(111)/ α -Fe₂O₃(0001)/AO(0001) and Au(111)/ α -Fe₂O₃(0001)/STO(111) systems varying the substrate temperature (250, 500, and 750 °C) for the growth of the Au nanostructured layer. XANES spectra of α -Fe₂O₃ powder reference, bare α -Fe₂O₃ films (α -Fe₂O₃/AO(0001) and α -Fe₂O₃/STO(111)),³⁰ and Au foil reference are presented for comparison in each case.

379 spectra with a linear combination of different iron oxide
 380 reference compounds (FeO, Fe₃O₄, α -Fe₂O₃, and γ -Fe₂O₃
 381 references)⁴⁴ confirms a valence 3+ for Fe corresponding to
 382 a 100% stoichiometric hematite phase for all Au(111)/ α -
 383 Fe₂O₃(0001)/AO(0001) and Au(111)/ α -Fe₂O₃(0001)/STO-
 384 (111) heterostructures, independently of the Au growth
 385 temperature reached in this work.

386 For the Au islands layers, the XANES profile follows that of
 387 Au foil and no differences are noted irrespective of the
 388 substrate and/or the Au growth temperature, as shown in
 389 Figure 4b. Similar results are obtained from the X-ray
 390 photoelectron spectroscopy (XPS) measurements (see the
 391 SI) on the Au 4f core level. The spin-orbit splitting of 3.7 eV
 392 and the intensity relationship between both peaks (4f_{5/2} to
 393 4f_{7/2}) corresponds to that of pure Au being identical for all
 394 samples. Moreover, we detect signal corresponding to the α -
 395 Fe₂O₃ surface on samples with the Au prepared at higher
 396 temperature corroborating the more isolated character of Au
 397 NSs. The combination of XANES and XPS demonstrates the
 398 absence of mixed Fe–Au phases present on the surface and
 399 buried interfaces.

400 The short-range ordering of cations around the Fe and Au
 401 and the neighbor bond lengths in nanostructured Au/ α -
 402 Fe₂O₃/oxide substrate systems were analyzed by extended X-
 403 ray absorption fine structure (EXAFS) technique. Figure 5a
 404 displays the modulus of the Fourier transform (FT) of the
 405 EXAFS signal at the Au L₃-edge. The FT is performed in the k^2
 406 $\chi(k)$ weighted EXAFS signal between 2.8 and 8.0 Å⁻¹.
 407 Experimental EXAFS results are fitted in R-space in the
 408 range 1.6–3.3 Å (first neighbors) using the FEFFIT code.⁴⁵
 409 The fitting was performed fixing the shift at the edge energy E_0 ,
 410 which was previously calculated from Au foil. Therefore, the

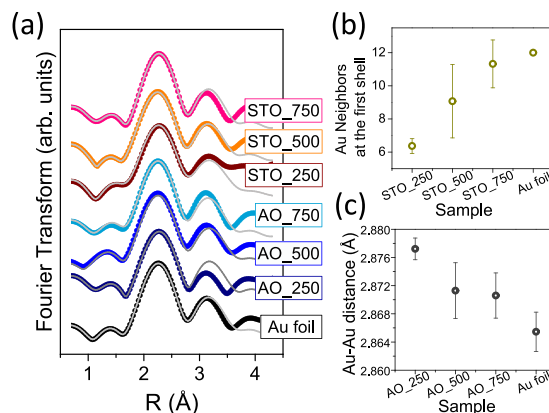


Figure 5. (a) FT modulus of the EXAFS signal (lines with symbols) and best-fitting simulations (continuous lines) at the Au L₃-edge of all samples and Au foil reference, (b) first neighbors Au–Au number for Au(111)/ α -Fe₂O₃(0001)/STO(111) system, and (c) Au–Au distance at the first shell number for Au(111)/ α -Fe₂O₃(0001)/AO(0001) system obtained from experimental EXAFS results measured at RT for heterostructures prepared varying the substrate temperature (250, 500, and 750 °C) for the growth of the Au nanostructured layer. Results of Au–Au first shell number and distance are representative of the two substrate employed and compared to the values obtained for the Au foil reference.

amplitude reduction factor S_0^2 , the interatomic distance R , and
 the Debye–Waller (DW) factors σ^2 for the first shell are used
 as free parameters for the fitting.

With respect to the DW factors (not shown) obtained for
 the samples analyzed at the Au L₃-edge, these are of same
 order for all samples and any tendency is observed as a
 function of the chosen substrate and/or the substrate
 temperature during the Au growth. However, a variation of
 the coordination number (calculated from the amplitude factor
 S_0^2) and the first neighbors distance is found, as Figure 5b,c
 shows representatively for the heterostructures prepared on
 STO(111) and AO(0001) substrate, respectively. An increase
 of the Au neighbors number and a shortening of the Au–Au
 bond length at the first coordination shell are obtained as the
 substrate temperature is increased. Lower Au-neighbor number
 and larger Au–Au bond lengths are obtained for all samples
 compared to bulk Au. These EXAFS results agree with XRD
 measurements presented in the SI (Figure S3), indicating a
 larger crystalline order of the Au layer grown at higher
 temperatures.

For the Au/ α -Fe₂O₃/oxide substrate systems at the Fe K-
 edge, a similar EXAFS study was carried out (not shown),
 revealing the absence of an additional short order with respect
 to single hematite films grown on AO(0001) and STO(111)
 substrate,³⁰ in accordance with XRD and XANES results.

Optical Characterization: Surface Plasmon Activity.

The plasmonic response of Au NPs grown on α -Fe₂O₃(0001)/
 AO(0001) and α -Fe₂O₃(0001)/STO(111) oxide heterostruc-
 tures was examined by the extinction spectra, and the results
 are displayed in Figure 6, representing a convolution of
 absorption and scattering processes. All systems present an
 extinction maximum associated with the localized surface
 plasmon resonance (LSPR) of Au islands arranged on the α -
 Fe₂O₃ surface. This LSPR band is located between 600 and
 900 nm, varying its position depending on the morphological
 features of the NSs based on the different substrate
 temperature during the Au growth process.

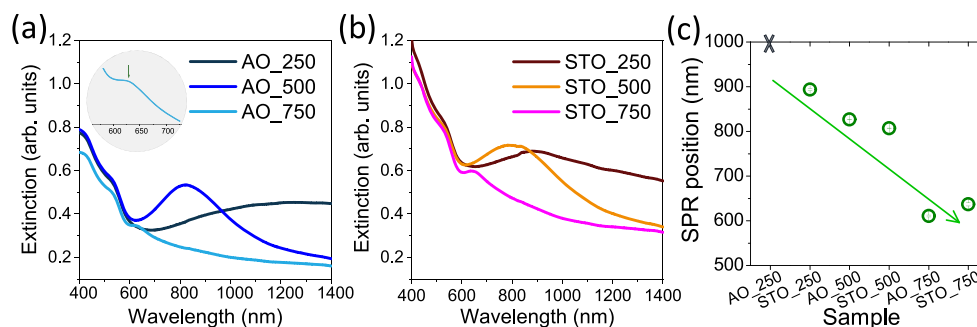


Figure 6. Extinction spectra for Au(111) islands prepared on (a) α -Fe₂O₃(0001)/AO(0001) and (b) α -Fe₂O₃(0001)/STO(111) oxide heterostructures as a function of the Au growth temperature (250, 500, and 750 °C). (c) Position at the LSPR maxima for the several samples prepared in this work. Sample AO_250 is marked with a cross in (c) indicating the absence of an extinction maximum. The inset in (a) shows a zoom of the LSPR region for the sample AO_750 for a better identification.

448 Samples prepared at 500 and 750 °C of Au growth
 449 temperature show average size of Au NPs around 30 and 50
 450 nm, respectively, as shown in Figure 1g, which exhibits the
 451 extinction spectra with an LSPR signal that is practically due to
 452 absorption effects. However, for the Au nanostructures
 453 prepared at 250 °C, the lateral average size is larger than 80
 454 nm, with an LSPR shift toward larger wavelengths and a
 455 broader and asymmetric peak that is attributed to larger
 456 scattering effects, which has to be considered for several
 457 applications as, for example, for biological applications or for
 458 cell imaging and photochemical studies.^{46,47}

459 Independently of the oxide substrate, the LSPR related to
 460 the Au NPs for all complex heterostructures exhibits a blue
 461 shift and a narrowing of band as Au substrate temperature
 462 increases, as can be clearly seen in Figure 6, which can be
 463 attributed to variations in the shape and interparticle distance
 464 of Au islands.^{48,49} Moreover, in all instances, an additional shift
 465 of the LSPR bands is shown toward larger wavelength with
 466 respect to single Au islands. These changes in the position of
 467 the resonance band are due to the hematite layer rounding
 468 partially the Au islands, prompting an increase of the dielectric
 469 constant of their surrounding medium and with that the
 470 wavelength shift of plasmon resonance.^{49,50}

471 For the case of samples with the Au layer prepared at high
 472 temperature (750 °C), the LSPR band is more similar to that
 473 expected for isolated Au NPs with a large distance between
 474 particles, a more rounded character, and a more homogeneous
 475 particle distribution. For the samples where the Au NPs are
 476 prepared at a substrate temperature of 500 °C, we find a
 477 shorter interparticle distance and a wider distribution of Au
 478 particle size, shifting the LSPR position toward larger
 479 wavelengths and widening its LSPR band.^{49,50} This effect is
 480 accentuated for the Au/ α -Fe₂O₃/STO(111) system grown at 250
 481 °C with a more continuous Au layer. For the specific case of
 482 Au/ α -Fe₂O₃/AO(0001) prepared at a temperature of 250 °C
 483 during the Au growth, we identify a plateau extended up to the
 484 near-infrared region in the extinction spectrum, instead of a
 485 well-defined maximum characteristic of localized surface
 486 plasmons (LSPs). This feature has already been reported for
 487 single nanostructured layers in the proximity of a continuous
 488 film and explained through the interaction between LSPs and
 489 extended SPs (expected in metallic thin films under specific
 490 conditions),^{19,25,51} raising the plasmonic applications of these
 491 complex systems.

492 On the other hand, intensity variations of the LSPR are
 493 noted depending on the substrate temperature in the growth of

494 Au islands, which might associate with parameters such as the
 495 Au islands size and/or the average dielectric constant in each
 496 system prepared.⁴⁸ Even, a sublimation process cannot be
 497 discarded at higher temperatures (750 °C), as commented
 498 above, diminishing markedly the intensity of the resonance
 499 band.

500 Finally, it should be noted that for all optical spectra of Au/
 501 α -Fe₂O₃/oxide substrate heterostructures, the LSPR bands of
 502 Au NPs appear superimposed on the optical signal related to
 503 the electronic transitions of α -Fe₂O₃ layer.^{52,53} However, in the
 504 wavelength range where SPs of Au islands are excited, the
 505 absorption signal related to iron oxide is low and can be
 506 considered negligible.

4. CONCLUSIONS

507 We have shown the way to grow epitaxial incommensurate
 508 nanostructured Au(111) on α -Fe₂O₃(0001)/ α -Al₂O₃(0001)
 509 and α -Fe₂O₃(0001)/SrTiO₃(111) oxide heterostructures. The
 510 Au nanostructured layer is obtained by a just one-step process
 511 controlling the substrate temperature during the Au growth,
 512 being the parameter that allows to tune the morphological and
 513 structural features of the Au nanostructures. After the Au
 514 growth, the stability of single-phase and single-oriented α -
 515 Fe₂O₃(0001) layers is confirmed, finding the same structural
 516 characteristics that prior to the deposition of Au nanoparticles.
 517 The coupling of the Au(111) islands lattice with the α -
 518 Fe₂O₃(0001) lattice is mediated by a rotation of 30° between
 519 their crystallographic axes. The Au(111) and SrTiO₃(111)
 520 lattices are hence collinear, while a rotation of 30° happens
 521 respect to the α -Al₂O₃(0001) lattice. The increase of the Au
 522 growth temperature leads to a larger crystallographic domain
 523 size, a larger crystalline order, a larger interparticle distance,
 524 and a more rounded character for Au islands. In addition,
 525 complex bilayers exhibit plasmonic activity. Controlling the
 526 substrate temperature in the Au growth, a tailorable response
 527 of the localized surface plasmons related to Au nanoparticles is
 528 mediated through the morphological characteristics of Au
 529 islands and the proximity effect of the hematite layer. This
 530 processing route allows obtaining ensembles of plasmonic
 531 islands on oxide surface with an epitaxial character, allowing
 532 the combination of multiple functions of these hybrid
 533 nanomaterials and their increasing highly value for several
 534 applications.

535 ■ ASSOCIATED CONTENT

536 ⑤ Supporting Information

537 The Supporting Information is available free of charge on the
538 ACS Publications website at DOI: 10.1021/acs.jpcc.9b04768.

539 Reflection high-energy electron diffraction patterns
540 during the deposition of Au/ α -Fe₂O₃ heterostructures
541 prepared on different substrates; thin-film thickness
542 calculation from low-angle X-ray reflectivity measure-
543 ments; crystallinity analysis from X-ray diffraction
544 measurements as a function of the substrate employed
545 and the evaporation temperature for the Au growth; and
546 X-ray photoelectron spectroscopy measurements on the
547 Au 4f core level of Au(111)/ α -Fe₂O₃(0001)/oxide
548 substrate systems (PDF)

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555 All authors have given approval to the final version of the
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557 Notes

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