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# Effect of Y<sub>2</sub>O<sub>3</sub> addition on the microstructure and mechanical properties of an Al<sub>1.8</sub>CoCrCu<sub>0.5</sub>FeNi BCC HEA

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# ABSTRACT

The present study investigated the influence of  $Y_2O_3$  addition by mechanical alloying (MA) on the microstructure evolution of a BCC High Entropy Alloy (HEA). The characterisation and mechanical properties of the alloy were explored using X-ray diffraction, SEM, EBSD, and nano-indentation. The sintered  $Al_{1.8}CoCrCu_{0.5}FeNi$  HEA shows a microstructure formed by an ordered BCC phase (Al-rich) and a second disordered BCC (Cr-rich), while a minor FCC (Cu-rich) appears. These BCC phases show a wide morphology evolution from cuboidal and wave-like structures to irregular shapes. The minor FCC phase also adopts several morphologies as the MA is performed. The introduction of oxide reinforcements and microstructure refinement through mechanical alloying yields a change in phase quantification and grain structure. In accordance with the hardness and elastic modulus values from ordered/disordered BCC phases, the disordered BCC shows higher values than the ordered one. The grain size reduction as well as the solid solution strengthening from the microstructure evolution consequence of the MA are shown to be the main contributors to the increase in hardness and elastic modulus in the consolidated samples.

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# 1. Introduction

Since the pioneer works on multiprincipal alloys in 2004 [1], many studies have been devoted to High Entropy Alloys (HEAs). This category of multiprincipal alloys differs from the traditional alloy design, which is related to one or two main elements [1]. In these alloys the presence of five or more principal elements in equi- or near-equimolar ratio, leads to an increase in the configurational entropy ( $\Delta$ Sconfig) [2]. The properties related to the composition of their phases, crystalline structure and microstructure have been widely studied and classified. HEA are found mainly within facecentred cubic (FCC), body-centred cubic (BCC) or FCC+BCC families, and in a few cases, hexagonal closest packed (HCP) [3,4]. They can show a sole crystalline structure, FCC or BCC, for instance [5,6]. However, they are rarely single phase, as their large number of components tend to stabilize several phases with different compositions, but with a similar lattice parameter [7]. In these alloys, the high mixing entropy values between element pairs hinder the formation of brittle intermetallic compounds [8]. Although HEA can have some interesting functional properties,

HEA have been mainly explored as a structural material. Several authors have reported high values of hardness [9–13], yielding strengths [14,15], large ductility [16,17], fatigue resistance [18,19] and fracture toughness [20,21]. Other properties such as wear resistance [22], electrochemical resistance [23] and oxidation behaviour [24] have also been studied. In the case of oxidation resistances, some approaches are based on the formation of stable and dense oxide layers on the surface, usually Al/Cr/Si-O, which could protect the material from further oxidation.

Refractory HEAs (RHEAs) appeared after initial studies focused on HEAs formed by 3d transition metals, addressed for high temperature applications. They are composed of refractory elements (mainly Mo, V, Zr, Ta, W and Nb), which usually show single-phase BCC structures [25]. Their strength and hardness values are above FCC HEAs with high temperature strength and creep resistance. These alloys show great potential to compete with Ni-base superalloys. However, they have very high densities (12–13 g/cm<sup>-3</sup>) with poor elongation leading to fracture at room temperature [26–28], which is critical for industrial service.

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The search is thus for a suitable high temperature material without refractory elements that maintains strength and hardness values at moderately high temperatures. At the same time, the composition should be optimised accordingly. Several criteria have been evaluated for the prediction HEA, which include parameters like entropy/enthalpy of mixing, differences in atomic radius, lattice distortion, electronegativity or the valence electron concentration (VEC) [29–35]. However, it was concluded that these terms individually cannot produce robust predictions in some cases. This is due to the complexity and strong interaction within alloy elements [36].

One of the approaches followed for predictions is the Hume-Rothery rules considering the number of itinerant electrons per atom (e/a) [37–39]. Using this method, three domains were identified which correlate HEA compositions, structure, and hardness with e/a values and the average radius of elements. The outcome of this method has already resulted in the design of BCC HEAs, such as some variations of the alloy system AlCoCrCuFeNi [40,41]. Moreover, their microstructures and properties have already been either produced from ingot metallurgy (IM) [9,16,22] or powder metallurgy (PM) techniques [23,42]. Taking into account the potential of the Al-CoCrCuFeNi family to develop BCC structures, this work is focused on the composition Al<sub>1.8</sub>CoCrCu<sub>0.5</sub>FeNi. This composition was previously designed using the e/a domains from [37–39], showing a BCC structure [43,44].

In terms of material processing, most HEAs are developed by IM, using either induction or arc melting to mix the components that are usually based on 3d transition metals. As an alternative route, PM is also employed using fully prealloyed powders and further consolidation by either conventional and non-conventional sintering such as Spark Plasma Sintering (SPS) technique or Cold Isostatic Pressing (CIP) [45]. PM also allows several strengthening methods to enhance the alloys' performance. One of these approaches is the particle reinforced HEAs. Several studies have shown results in this direction. Typically, the reinforcements include using SiC particles or oxides such as Al<sub>2</sub>O<sub>3</sub> or even Y<sub>2</sub>O<sub>3</sub> of different sizes and quantities [46–48]. Therefore, it is of interest to evaluate the strengthening of the Al<sub>18</sub>CoCrCu<sub>0.5</sub>FeNi while maintaining its BCC structure. Different processing PM routes have been evaluated for the AlCoCrCuFeNi family depending on the starting powder: i) mechanical alloying (MA) of elemental powders with further sintering by techniques like Spark Plasma Sintering (SPS) in [42] ii) mixture of elemental powders combined with vacuum sintering in [44] and iii) prealloyed powder with further sintering by fast-sintering techniques like SPS or Electric Resistance Sintering (ERS) in [43]. Following these procedures, the alloys showed a variety of FCC and BCC structures depending on their compositions. When these alloys exhibit BCC structures, they have low density, reduced cost and a medium to high hardness (slightly below refractory HEAs) [43].

In summary, the processing method combined with the compositions selected determines the final structure and phases of the alloy. In the case of  $Al_{1.8}CoCrCu_{0.5}FeNi$  alloy, it has been shown to form BCC structures. However, it is always composed of two BCC phases, disordered BCC and ordered BCC (usually called B2 in the literature) [43,44]. These two phases appear coupled in several morphologies in the literature, such as spherical, cuboidal nanoprecipitates or even weave-like [49]. They have been found to be related to the processing method and to the lattice misfit of the alloy, which is closely related to Al in non-refractory BCC HEAS [50,51].

The mechanical properties are derived from these two BCCs and their morphology, which is dependent on the processing route. Therefore, one objective of the present study is to evaluate the microstructure evolution of  $Al_{1.8}CoCrCu_{0.5}FeNi$  alloy by powder metallurgy while aiming to particle reinforce it with  $Y_2O_3$  oxides and MA. This composition was originally designed from the e/a approach and domains proposed by Calvo-Dahlborg et al. [18] to obtain a BCC

HEA. The influence of particle reinforcement in the alloy was studied, evaluating its phase evolution and crystalline structures. The possible strengthening mechanisms have been discussed based on their microstructural and mechanical properties characterisation.

# 2. Materials and methods

In this study, Al<sub>1.8</sub>CoCrCu<sub>0.5</sub>FeNi has been considered and processed by PM. The HEA prealloyed powder was obtained from pure elements in a lab scale gas atomizer equipped with an induction furnace (Atomising Systems Limited, Sheffield, UK). For the development of the reinforced material, the atomised powder was mechanically alloyed with 0.25 wt% Y<sub>2</sub>O<sub>3</sub> particles (from TJ Technologies & Materials Inc.) of 7  $\mu$ m mean size using high energy milling (planetary mono mill Pulverisette 6, Fritsch, Idar-Oberstein, Germany).

The MA process was composed of two alternative paths: 1) dry MA under a highly pure Ar atmosphere plus short wet MA cycle and 2) wet MA with isopropyl alcohol ( $(CH_3)_2CHOH$ ) as a Process Control Agent (PCA). Both milling routes were made at 300 rpm with an effective milling time of 30.5 h and 45 h respectively. The ball to powder ratio (b:p) was set to 10:1 with 10 mm diameter stainless steel balls.

In the first case of dry plus wet milling, powder particles tend to agglomerate on balls and container walls making them difficult to work with. Therefore, a further 30 min wet milling step in addition to the 30 h dry milling was performed using 200 ml of ethanol as PCA (300 rpm) to disperse the powder particles. Finally, the resulting wet powders were filtered and dried afterwards. This methodology of dry plus wet milling was also successfully applied in HEAs by Moravcik et al. [52]. These wet/dry MA powders were chosen for later consolidation instead of dry only MA powders.

Independently of the state of the initial powder (mechanical alloyed vs plain prealloyed HEA powders), all of them were consolidated via spark plasma sintering (Dr Sinter, SPS-1050CE from SPS Syntex). The consolidation was made in a 30 mm diameter graphite die previously covered with boron nitride to hinder carbon contamination inwards to produce 30 mm samples of 5 mm height. The sintering started from room temperature up to 1100 °C with a heating rate (H.R.) of 200 °C/min. The dwell time at 1100 °C was set to 1 min to avoid further Cu segregation and overall growth of the microstructure. The temperature in all stages was measured with a K type thermocouple placed inside a 6 mm hole in the middle section of the die, close to the powder. The pressure was kept to a steady 50 MPa during the sintering cycle with pulsed current densities around 1.2 kA/cm<sup>2</sup>. Table 1 summarises the parameters for the MA and sintering process.

The powders and sintered samples were analysed by XRD, SEM and EBSD (transversal sections for sintered samples). The equipment used for microstructure characterisation was a field emission scanning electron microscope (FEI Teneo FEG-SEM, Hillsboro, OR, USA) equipped with an electron back-scatter diffraction (EBSD) detector and an energy dispersive spectrometer (EDS). The chemical composition and grain size/morphology via backscattered electron diffraction (EBSD) was analysed on sintered samples as well as the precursor powders. Due to the high plastic deformation sustained by the powder particles during milling for reinforced-HEA, large lattice distortion developed and no Kikuchi patterns can be observed in EBSD for that powder. EBSD acquisitions were performed using a step size of 90 nm with a tolerance angle of 2° to determine grain distribution and orientation.

To evaluate the milling process, the crystallite size and microstrain values (calculated by Scherrer method [53] on the peak (110)) were analysed at different time stages of milling by XRD, performed in a Siemens D5000 diffractometer (Siemens, Munich, Germany) with a Cu cathode ray tube ( $\lambda = 1.5406$ , 1.5444 Å). As mentioned by E. Reverte, C. Keller, M. Calvo-Dahlborg et al.

#### Table 1

Summary of processing conditions for powders and SPS consolidated material.

Nomenclat	ure				
Feedstock	Gas Atomised	Al <sub>1.8</sub> CoCrCu <sub>0.5</sub> FeNi		-	
Milling Conditions	Wet Milled	Al <sub>1.8</sub> CoCrCu <sub>0.5</sub> FeNi + 0.25Y <sub>2</sub> O <sub>3</sub>	300 rpm   milling	time 45 h   b:p 10:1	
	Dry/wet Milled		300 rpm   milling t	ime 30.5 h   b:p 10:1	
Consolidated Materials	Plain HEA	Sintering T (°C) 1100	Dwell 1min	H.R. 200°C/min	
	MA-HEA				



(a)

(b)



Fig. 1. (a) SE-SEM image of Wet milled powder for 45 h; (b) SE-SEM image of Dry/Wet milled powder for 30 h; (c) Microstrain and Crystallite size evolution versus milling time and (d) XRD patterns of all the specimens at the selected milling times.

Moravcik et al. [52], powder milling using a process control agent can lead to contamination from interstitial elements like Oxygen/ Nitrogen or Carbon/Sulphur. To this aim, powder contamination was quantified by solid-state infrared (IR) and thermal conductivity (TC) detectors using a LECO TC-500 and LECO TC CS-200 (both of them from LECO, St Joseph, MI, USA), respectively.

To assess the mechanical properties of the alloys, the hardness tests were performed with two different types of equipment: One for micro-indentations in a ZHV $\mu$  hardness tester (Indentec Hardness Testing Machines Limited, Hillsboro, West Midlands, UK) with a peak load of 100 mN; and the second for nano-indentations at a peak load of 500  $\mu$ N (Hysitron TS 77 Select, Bruker, Massachusetts, USA). The

nano-indentations were performed on  $11 \times 11 \mu m$  squares with 0.5  $\mu m$  of separation between indentations and at a speed of 60  $\mu N/nm$ .

# 3. Results and discussion

# 3.1. Powder consolidation and characterisation: Effect of MA and particle reinforcement on HEA microstructure

The initial atomised powder was milled adding particles of  $Y_2O_3$  to achieve a nominal  $Al_{1.8}CoCrCu_{0.5}FeNi + 0.25$  wt%  $Y_2O_3$  to develop the reinforced alloy. In the high energy milling stage, plastic deformation occurs from the collision between the milling media, steel balls and

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#### Table 2

Chemical compositions of the atomised powder and dry/wet milled powder.

Tag	Al	Fe	Cr	Со	Ni	Cu	Y <sub>2</sub> O <sub>3</sub>
HEA (at%)a	25 ± 3	20 ± 1	14.8 ± 0.7	16 ± 1	15.7 ± 0.3	7.4 ± 0.4	-
Nominal HEA (at%)	28.5	15.8	15.8	15.8	15.8	7.9	-
MA-HEA (wt%)a	$12.2 \pm 0.2$	$24.1 \pm 0.4$	15.2 ± 0.3	19.3 ± 0.2	19 ± 0.2	$10.2 \pm 0.2$	-
Nominal MA-HEA (wt%)	15.8	18.2	16.9	19.2	19.1	10.3	0.25

Measured via Energy-dispersive X-ray spectroscopy (EDS)

Table 3	Tal	bl	e	3
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Oxygen and nitrogen chemical analysis of the MA powders.

	Atomise	d powder	Wet mill	ed powder	Dry+Wet	milled powder
	%Oxygen	%Nitrogen	%Oxygen	%Nitrogen	%Oxygen	%Nitrogen
Total: % Y <sub>2</sub> O <sub>2</sub> :	0.052	0.015	0.148 0.053	0.002	0.246 0.053	0.016
% Ex.O: % Ex.O*:			0.095 0.043		0.193 0.141	

Ex.O = Total O content -  $\frac{48}{226}$  Y<sub>2</sub>O<sub>3</sub> content (wt%) Ex.O<sup>\*</sup> = Total O content -  $\frac{48}{226}$  Y<sub>2</sub>O<sub>3</sub> content (wt%) - O content (wt%) in atomised powder

powder. The severe transfer of mechanical energy to the powder causes its fracture and cold-welding, conditioning the microstructure and properties of the final material. Regarding the particle size and according to some previous studies in the literature, powder fracture and cold-welding effects are balanced during the progression of milling, [54,55]. However, when wet milling was performed in this study, powder fracture prevailed over cold-welding even for long periods, resulting in an elongated and flake-like powder (see Fig. 1-a). In contrast, when the milling is performed under dry conditions, coldwelding predominates, sticking the powder, vial and balls together, resulting in low efficiency. However, if both routes are mixed, first a dry stage followed by a second wet milling, the powder reaches an irregular shape more suitable for consolidation in SPS (see Fig. 1-b). This was the powder selected for subsequent sintering.

Related to the solid solution formation, atom diffusion does not significantly affect the inherent arbitrary positions of atoms due to their sluggish diffusion. However, previous studies have shown that copper tends to segregate [44]. As the  $\Delta H_{mix}$  of unlike pairs become the dominant factor over  $\Delta S_{mix}$  it promotes the segregation [43], which may also occur during the milling process. Plastic strain and residual stress generated in the powder should relax due to heat in the SPS consolidation resulting in broader copper segregation than in milling. Moreover, other studies have shown that minor copper segregation is observed even in gas atomised powders [43,44].

In Fig. 1-c, the microstrain and the size of coherently scattering domains evolution as a function of milling time are represented. These measurements from different milling strategies (i.e., dry or wet/dry) were obtained from XRD patterns in Fig. 1-d. Once a critical level reaches a plateau, both parameters stabilize in a steady state even with increasing milling times. It is possible to observe that crystalline domains are rapidly stable after 8 h, with an average size of 8.7 nm, while the microstrain reaches a stable value of around 1.05% close to two hours later. When milling time is increased, there is a decrease in the size of the crystallites, indicated by the peak width enlargement from XRD (Fig. 1-d).

The microstrain achieved during the high energy milling (≈1.05%) will determine the consolidated grain size distribution [56]. Furthermore, the balance between dislocation densities due to cold work in MA and thermal recrystallization during SPS consolidation will also establish the minimum average grain size [56].

The nominal and chemical composition of the atomized HEA together with the Dry+Wet milled powder is shown in Table 2. They



Fig. 2. XRD patterns and comparison between HEA powders (atomised and milled) and consolidated samples (HEA and MA-HEA).

show overall similarities between nominal and experimental data. Foreseeing the formation of undesired oxides and nitrides due to potential contamination arising from the atomisation process, as well as the addition of a process control agent, the interstitial content in the alloy was measured. In the case of other contamination like Fe or Cr from the vials and balls during milling, they could be considered non-relevant contamination as the compositions usually vary from the nominal values.

Aside from interstitial contamination, the literature regarding ODS steels has shown the importance of evaluating and controlling the excess oxygen contents (Ex.O). This parameter is obtained by subtracting the oxygen content from the  $Y_2O_3$  compound from the total O in the alloy. Moreover, a second excess oxygen parameter (Ex.O\*) has been calculated which also considers the oxygen content measured in atomised HEA powders. These values summarise the enrichment of oxygen in the milled powder (see Table 3).



Fig. 3. BSE-SEM images of Plain HEA.

Table 4							
Chemical	compositions	(at%) of	f phases	detected	on	Plain	HEA.

Phase	Al	Cr	Fe	Со	Ni	Cu
Cr-Fe-rich phase "Disordered BCC"	22.3	16.9	22.4	15.5	14.6	8.1
Al-Ni-rich phase "Ordered BCC"	26.7	10.6	19.1	17	18.4	7.9
Cu-rich phase "FCC"	14.2	10.5	11.2	6.6	5.7	51.8

Measured via Energy-dispersive X-ray spectroscopy (EDS)

Indeed, the Ex.O produced in the alloys can drastically improve the microstructure and the high-temperature performance of alloys. Raja Rao et al. [48] studied the strengthening of an AlCoCrFeNi HEA with different oxide weight percentages (ranging from 0 to 3 wt%  $Y_2O_3$ ). This study showed that  $Y_2O_3$  offers a beneficial effect on the mechanical properties of HEA. However, Ohtsuka et al. [57] concluded that merely increasing the particle addition cannot improve properties such as high-temperature strength. These ideas focus on optimising the right proportion of Ex.O available in the HEA instead of adding large quantities of particle oxides. According to Ohtsuka et al. [57,58], a minimum amount of Ex.O required in their ODS steels is about 0.1 wt%, a value reached in the dry/wet specimens [59].

Aluminium has a FCC crystalline structure. However it is found to induce the formation of BCC structures in HEA. This leads to higher hardness and yield strength [9,49,51,60,61] than FCC alloys. Hence, Al-Transition metals HEAs, such as the one used in this study, generally exhibit BCC structures as Al concentration increases [62]. They are usually composed of two phases, an ordered BCC enriched in Al-Ni and a disordered BCC enriched in Cr-Fe. The ordered BCC precipitates are arranged in spherical, cuboidal nano-precipitates and even weave-like morphologies within the BCC disordered matrix, according to the literature [49,50]. Recent studies on these three microstructures have shown that their formation is affected by the lattice misfit, which is affected by Al content. Moreover, the anisotropy difference of Young's modulus between the two BCC phases has also been proven to affect their morphologies. Both BCCs have shown exchangeable arrangement as either precipitate or matrix. It is of interest that cuboidal and coherent precipitation in the matrix exhibits much higher thermal stability than the weave-like morphology that results in an embrittlement of the alloy [49,50].

Fig. 2 shows that in XRD diffractograms from powder and consolidated samples the BCC structure is predominant. However, a more detailed analysis reveals that the microstructure consists of two BCCs, ordered and disordered, with the combined intensity of both indistinguishable peaks. This feature is in agreement with the literature but further analysis with advanced techniques is needed to correctly distinguish both phases [49]. Nevertheless, SEM analysis of the consolidated HEA using backscattered electrons (see Fig. 3) clearly illustrates these two BCC phases. Both phases adopt different morphologies, ranging from cuboidal-like nano-precipitates to weave-like morphologies (marked in a red circle in Fig. 3-c). This arrangement can be attributed to lattice misfit between ordered and disordered BCCs [49]. The compositional analysis is shown in Table 4. The cuboidal precipitates appear enriched in Cr-Fe (disordered BCC) while the matrix is enriched in Al-Ni (ordered BCC). Moreover, a third minor phase enriched in Cu (FCC) was also observed in the microstructure. Finally, Fig. 4 shows a chemical map of a HEA sample with a Cr carbide. These minor contaminations have appeared multiple times in different processing routes in the literature [43,44].

Fig. 5-a and 5-b show the microstructure of MA-HEA, obtained after the consolidation of the dry/wet milled powder. The most noticeable difference compared to the previous samples is the white phase dispersed along the microstructure. These precipitates are a Cu-rich phase whose composition can be seen in Table 5. Moreover, the oxides introduced in the MA are not visible. This suggests that they are mixed within the microstructure. The influence of these oxides should not be relevant to either BCC formations, either cuboidal or weave-like. In Ni-based superalloys, some oxides formers such as Ti or Al favour the formation of  $\gamma$ ' phase [63]. In contrast, Y or Y<sub>2</sub>O<sub>3</sub> do not interfere with BCC phase formations in HEA.



Fig. 4. Chemical compositions detail of carbide in HEA sample.



Fig. 5. BSE-SEM images of MA-HEA sample.

Table 5	
Chemical compositions (at%) of phases detected on MA-HEA samples.	

Phase	Al	Cr	Fe	Со	Ni	Cu
Cr-Fe-rich phase "Disordered BCC"	20.9	18.7	24.1	15.1	13.9	7.1
Al-Ni-rich phase "Ordered BCC"	25.4	12	20	16.9	18.3	7.2
Cu-rich phase "FCC"	17.0	12.5	15.3	8.8	7.9	38.3

Measured via Energy-dispersive X-ray spectroscopy (EDS)

From an image analysis using ImageJ software and colour contrast, this Cu-rich phase occupies about 4–5% of the microstructure. Fig. 5-c shows the presence of two phases randomly distributed within the microstructure. The first phase is visible as a dark matrix, while the second phase appears as cuboidal precipitates similar to those found previously in plain HEA (circled in red). However, these cuboidal morphologies only appear in some regions. This structure may be the consequence of an uneven recrystallization of less deformed particles from the milling stage. As a result, the alloys developed a heterogeneous microstructure.

The microstructures of both processed HEAs are different with complex BCC phases and cuboidal and weave-like morphologies. A closer observation of the two microstructures is represented in Fig. 6 with an image quantification of a reduced section. From image analysis using ImageJ software, the more refined microstructure of the MA HEA sample has decreased the amount of disordered BCC developed in the ordered-BCC matrix from 50% to 30%.

To analyse crystallographic aspects of the HEA evolution through production processes, EBSD observations were carried out on atomised powder, plain HEA and MA-HEA samples (Figs. 7-a, 7-b and 7-c). In the



BCC distribution

Fig. 6. SEM based BCCs distribution quantification in (a) HEA and (b) MA-HEA microstructure.

case of the milled powder, it could not be observed due to the severe crystallographic strain and plastic deformation from the process.

For the atomised HEA powder (Fig. 7-a), the grain orientation map shows grains with smaller sizes than the average particle ones (D50 =  $35.6 \,\mu$ m), suggesting polycrystalline powder particles. Intragranular misorientations are also observed in this image showing the existence of sub-grains and Low Angle Grain Boundaries (LAGBs). These boundaries were considered within rotation angles changes of 5° to 15°. This feature is confirmed by the Kernel Average Misorientation map (KAM) combined with LAGBs (green boundaries) that shows internal sub-grain structures restricted by LAGB. This observation implies that the starting HEA powder has a relatively high value of residual stress developed from the gas atomisation process.

Focusing on sintered HEA samples, the grain orientation maps show well-defined grain structures based on the BCC phase for both sintered samples without intragranular misorientations suggesting a full recrystallization process during sintering. This feature is also confirmed by the KAM maps where large KAM values are always associated with grain boundaries. Some FCC phase (less than 1% in volume fraction) is also observed for these samples. The pole figures do not prompt any preferential grain orientations (maximal density poles of about 2).

A broad grain size distribution is observed focusing on the average grain size, which is  $3.7 \pm 2.8 \,\mu\text{m}$  for the plain HEA versus  $0.47 \pm 0.43 \,\mu\text{m}$  for the MA-HEA. This reveals reduced grain growth during the sintering process, probably linked to the lower scattering domain size produced from the milling stage. This has also been observed for stainless steels produced by similar PM routes [64].

# 3.2. Mechanical properties and strengthening mechanism discussion

The mechanical properties in alloys depend on multiple factors and different phenomena such as grain size, order, solid solution and dislocation density. To study the influence of different strengthening mechanisms on HEAs, Ganji et al. [65] and Sriharitha et al. [66] evaluated some alloys from the Al<sub>x</sub>CoCrCuFeNi family. They observed that small variations in the composition changed the microstructure completely, yielding to a dual phase FCC and an ordered BCC (referred to as B2 in this study). Compared to the alloy  $Al_{1.8}$ CoCrCu<sub>0.5</sub>FeNi, the influence on the mechanical properties of dissimilar phases like FCC and BCC will be lower.

The microhardness values of Plain HEA and MA-HEA are 579 ± 37 HV<sub>0.01</sub> (5.67 GPa) and 614 ± 29 HV<sub>0.01</sub> (6.02 GPa), from at least 10 indentations respectively. The MA-HEA sample shows an increase in hardness that could be related to different contributions. First of all, the Hall-Petch strengthening or grain size dependent strengthening " $\sigma_{gss}$ " (Eq. 1).

$$\sigma_{\rm gss} \approx k d^{-1/2} \tag{1}$$

This parameter depends on an empirical coefficient "k". According to different publications on HEAs, several "k" values have been used, varying from 0.677 MPa m<sup>1/2</sup> [67,68] to 0.27–0.33 MPa m<sup>1/2</sup> [65,66]. However, the most important factor is the grain size "d" which changes following a reciprocal function ( $y = 1/x^{1/2}$ , with "y" being the grain size strengthening and "x" the grain size value). Therefore,  $\sigma_{gss}$  rapidly increases with the reduction of the mean grain size. The MA-HEA shows a smaller average grain size of about 400 nm in diameter, compared to the 3 µm from the standard HEA. Therefore, it can be assumed that the MA-HEA sample underwent strengthening produced by grain size refinement of about 36% following Eq. (1).

Another strengthening mechanism is dislocation hardening or "Taylor hardening", which develops from the crossing of dislocations during deformation. In the case of HEAs, Edalati et al. [69] performed a plastic deformation study where the deformation mechanisms of FCC and BCC HEAs were evaluated. FCC phases with low stacking fault energy form nano-twins and stacking faults that expand the lattice structure. However, the BCC's main mechanism is based on dislocation accumulations over strain while the lattice is contracted. The present alloy is mainly formed from BCC structures and therefore dislocation accumulations should be considered. The Taylor strengthening is represented as follows (Eq. 2) [65].

$$\sigma_{TH} \approx M \alpha G b \rho^{1/2}$$

(2)



Fig. 7. Inverse pole figures (IPF), KAM + High/Low angle grain boundaries (HAGBs/LAGBs) and Pole figures (PF) via EBSD for (a) atomised powder (b) Plain HEA sample and (c) MA-HEA sample.

The strengthening is composed of M as the Taylor factor that varies from the type of structure,  $\alpha$  is a correction factor dependent on each material, G is the shear modulus, b is the Burgers vector and  $\rho$  is the dislocation density, which can be assumed as in Eq. (3) from [65].

$$\rho \approx \frac{2\sqrt{3}}{d} \frac{\mathcal{E}}{b} \tag{3}$$

In this equation,  $\varepsilon$  is considered the microstrain (calculated by Scherrer method [53] as previous analysis in particles characterisation), taking into account the difference in grain size and microstrain values calculated from HEA and MA-HEA samples. The Taylor strengthening of MA-HEA triples the one approximated from HEA samples. The Kernel average misorientation maps in Fig. 7 confirm this with some regions where dislocation density is high. Therefore, this results in an increase of hardness values from the Taylor hardening component in the MA-HEA.

Regarding other strengthening contributors, there are two more plausible factors which are order strengthening ( $\sigma_0$ ) and solid solution strengthening ( $\sigma_{ss}$ ). In the first case, the order contribution stands for the lattice frictional stress offered by the lattice in the dislocation motion. The two-phase interaction (ordered/disordered BCC) and lattice complexity contribute to increasing the frictional stress produced by the lattice for mobile dislocations, and hence the  $\sigma_0$  contribution. At present, no frictional stress data can be found in the literature due to the novelty of HEAs. Some authors have calculated estimations by the rule of mixtures [65,66]. This suggests that this component depends on the composition and not on the processing method.

Other contributions to evaluate would be the solid solution strengthening mechanism and the precipitation hardening produced from the oxide introduced. The solid solution strengthening involves both BCC phases, as they are essentially solid solutions with



Fig. 8. Processing route for the heat-treated SPS HEA.

#### Table 6

Hardness and reduced modulus measured via nano-indentation in the different HEA specimens in this study.

Material		H (GPa)	E <sub>r</sub> (GPa)
Powders	Plain HEA Atomized Powder	6.3 ± 1.5	41 ± 12
Consolidated Samples	Dry/Wet Milled Powder	$9.6 \pm 1.5$	74 ± 13
consonuated samples	MA-HEA HEA + HT	$9 \pm 2$ 9.7 ± 0.8 8.4 ± 1.1	201 ± 13 207 ± 13

dissimilar element compositions. As dislocations progress elastically through the lattice, they interact with the distortions created by the substitutional solutes. This interaction in cubic structures distorts the crystal both geometrically (atom size mismatch) and chemically (electrical and bonding variations). However, these norms are not applicable to HEAs and the concept of solute/solvent does not exist for these materials. Some studies have attempted to extend conventional models of solute strengthening to HEA [70–72]. An example of this is the model based on an average effective medium by atomistic simulations, which has successfully been able to predict



Fig. 9. Hardness by nano-indentation from the current study compared to various literature HEAs, Al<sub>x</sub>CoCrFeNi [61], AlCoCrCuFeNi [65] and Al<sub>x</sub>CoCrCuFeNi [40].

the strength of FCC HEAs with experimental results [73]. For BCC HEAs, the most accepted deformation mechanism is kink-pair nucleation for dislocations motion, although it is still unknown if the solid solution promotes or hinders the dislocation motion. Baruffi et al. [74]. studied two recent theories based on the wavering behaviour of dislocations in random mediums. They suggest that in highly concentrated alloys, the strength is dominated by edge dislocations instead of screw dislocations. This behaviour is key to understanding the high temperature properties of BCC HEAs. Moreover, the shear modulus of the alloy, the elastic modulus mismatch and the atomic size mismatch are the dominant factors affecting the strength of the alloy [74–76].

The precipitation strengthening also can play a leading role in hardening the alloy. Raja Rao et al. [48] showed that even small additions of 1 wt% of  $Y_2O_3$  can increase the hardness of a mechanical alloyed HEA by 34%. These precipitates could not be seen by regular SEM observations. Therefore, this suggests that these oxides are dispersed in submicron regions in the microstructure.

To analyse the hardness influence of both BCC phases in solid solutions, nano-indentation tests were performed on the microstructure of the alloys at a peak load of  $500 \,\mu$ N. It is important to note that cuboidal precipitates are usually coherent in these types of alloys and their strengthening contributions should not be very relevant [49].

It was not possible to clearly discern the hardness of each individual phase due to the small size of the microstructure. For this purpose, the plain HEA was heat treated at 1100°C for 3 h to increase the grain and phase sizes developed in the microstructure. The nano-hardness tests were performed in the transversal section of the sample after mirror polishing and cleaning the surface for reliable data acquisition. An illustration of the heat treatment performed is shown in Fig. 8, with an SEM micrograph as an example of the obtained microstructure. Both phases are clearly visible, and they were analysed compositionally and structurally to be the same two ordered/disordered BCC phases and with a similar composition as the non-heat-treated HEA. The brighter phases correspond to the FCC Cu-rich phase also observed without heat treatment.

The average hardness and reduced modulus values extracted from the nano-indentation in all HEA samples are summarised in Table 6. The reduced elastic modulus act as a correction between the intender tip and sample elastic strains to accurately measure hardness values. The lowest hardness value corresponds to the atomised powder, and it greatly increases after the milling process, as observed from the Dry/Wet milled powder. Once both powders are sintered, hardness and elastic modulus also increase due to the consolidation process and strengthening mechanisms. On the contrary, the hardness of the heat-treated sample is lower than the consolidated HEA sample, as expected due to stress relaxation, solid state diffusion, and grain size growth. It is important to note that the low elastic modulus obtained in the powder particles is expected to be lower in spherical particles [77]. In addition, the measurements may suffer from substrate interference.



Fig. 10. Contour mapping from the nano-indentation performed on the heat-treated HEA sample (a) Hardness (b) Reduced Elastic Modulus and (c) BSE-SEM image after indentations.



Fig. 11. Phase deconvolution of hardness and reduced modulus values extracted from nano-indentations maps and showing the results of the properties for the individual phases.

Fig. 9 shows a plot of the hardness versus e/a values of different HEAs from the literature. The majority of literature values come from the FCC+BCC domain, while the composition of this study is somewhere between domains II (FCC+BCC) and III (BCC) [38]. A single study [40] has obtained values above 10 GPa with an Al<sub>3</sub>CoCrCuFeNi composition, heavily BCC stabilized due to the high Al content, higher than the values obtained for Al<sub>1.8</sub>CoCrCu<sub>0.5</sub>FeNi in the present work. Even though each study used different test parameters, which introduces a scattering in the hardness data, Fig. 9 shows a general trend where the hardness of BCC HEAs with large e/a values (Domain III) is higher than that of mixed FCC+BCC HEAs from Domain II. Considering that e/a values correlate to the proportion of FCC and BCC phases, our HEAs are expected to present hardness values around 8-9 GPa, consistent with the measurements obtained in Table 6 and validating the e/a approach for the hardness as well as the structure prediction. Moreover, the graphs also illustrate the hardness changes produced in the Al<sub>1.8</sub>CoCrCu<sub>0.5</sub>FeNi alloy through different processing methods.

Contour mappings of nano-indentations were performed on the surface of the HT sample to obtain individual phase measurements. The results are shown in Fig. 10. A backscattered SEM micrograph of the same region, post-indentations, is also included to identify each measurement and phase. From the hardness contour map, various phases can be distinguished, the most clearly visible one being the blue region in the middle, which when compared to the SEM image is attributed to the softer Cu-rich phase. This softer FCC phase shows a pile-up effect in the surface surrounding the nano-indentation pit. This phenomenon has been observed by Muthupandi et al. [78] in an AlCoCrFeNi with ordered/disordered BCC phases and a FCC phase similar to those observed in the present study. The pile-up behaviour occurs due to the absence of sufficient dislocations shifting through the slip direction, parallel to the surface, h. Additionally, the activation energy for the dislocation slip is lower in FCC structures compared to BCC lattices, making pile-up more likely to happen in the former, especially at low loads.

To complete the mechanical characterisation of the individual phases, a Gauss deconvolution of the hardness and reduced modulus distribution from the data in the contour maps was also analysed and is displayed in Fig. 11. Both BCC structures show higher reduced elastic modulus and hardness than the FCC phase. In the case of the hardness, the two BCC phases vary between 8 and 10 GPa, with the ordered phase being softer than the disordered BCC. However, in the case of the reduced modulus, the difference is larger, between 200 and 225 GPa, with the disordered BCC being the highest one.

# 4. Conclusions

The influence on hardness and elastic modulus in an  $Al_{1.8}$ CoCrCu<sub>0.5</sub>FeNi HEA obtained by different processing methods was studied. It was found that the crystalline structure in the different processes does not change. However, the phase morphology evolved from cuboidal precipitates to weave-like phases.

In all the processing routes, the HEA is composed of various phases: ordered BCC (Al-rich), disordered BCC (Cr-rich) and a minor FCC (Cu-rich). Following the MA process produces a change in the morphology and refinement of grain structure, as well as the distribution of the FCC phase. The grain size decreased, while the original arrangement of both ordered/disordered BCC phases changed, along with an increase of the FCC phase up to about 5 vol%.

The different samples have been characterised mechanically by nano-indentation. The results obtained are considered in a multiscale frame, in relation to the production methods as well as the properties of the constituent phases. The results show that the MA method increased the hardness and young modulus compared to the same composition of the HEA obtained by gas atomization and subsequent SPS. The mechanical properties of the constituent phases were studied. The ordered/disordered BCC phases show slight differences in hardness values but larger differences in the elastic modulus. Several strengthening mechanisms were evaluated and discussed from the reported literature, to observe their influence on the mechanical properties from the unique combination of a BCC-HEA and the MA approach. The grain size reduction, the increased dislocation density and the precipitation hardening have been shown to be the main contributors to the increase in hardness and elastic modulus. The consequences of this study open up new possibilities in the future development of the HEA that show ordered/ disordered BCCs structures.

# **CRediT** authorship contribution statement

Eduardo Reverte: Formal analysis, Investigation, Methodology, Validation, Writing – original draft. Clement Keller: Methodology, Investigation, Validation, Writing – review & editing. Monique Calvo-Dahlborg: Investigation, Validation, Writing – review & editing. German Alcalá: Methodology, Investigation, Validation, Writing – review & editing. Monica Campos: Methodology, Investigation, Supervision, Validation, Writing – review & editing, Funding acquisition. Juan Cornide: Conceptualization, Formal analysis, Funding acquisition, Investigation, Methodology, Supervision, Validation, Writing – review & editing.

# **Data Availability**

No data was used for the research described in the article.

# **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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