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Original Article

A new insight into the temperature induced molecular aggregations in tris(8-hydroxyquinoline) metals



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ABSTRACT

Annealing of tris(8-hydroxyquinoline) gallium (GaQ_3) film at various temperatures in dry N_2 atmosphere has shown the existences of four different phases of molecular aggregations before the burning out of the film at about $310^\circ C$. The first three phases, up to $235^\circ C$, are amorphous molecular aggregations, while the fourth one at $255^\circ C$ is a crystalline structure, very likely α -polymorph. The photoluminescence (PL) intensity was increased to about five times greater than that of the pristine film at $235^\circ C$, while the PL peak was blue shifted consistently. Although a small contribution of Rayleigh scattering cannot be excluded at high temperatures when crystallites appear, the PL blueshift was mainly attributed to the nanosstructured molecular aggregations followed by enhanced PL intensity. These new findings can be a common characteristic of organometallic complexes at varied annealing temperatures. The presented results open a new route of fabricating highly emissive thin films of amorphous nanostructure, which are specifically important for organic light emitting diode (OLED) based displays.

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1. Introduction

Over the past decade, research progress in organic materials has led to great achievements in the development of nano-photonics and nano-electronic devices [1,2]. This is

mainly due to the excellent chemical and physical properties offered by these kinds of materials including easy molecular designability, optical and electronics tunability [3–8]. One of the most practical achievements is related to the technological advancement of organic light emitting diode (OLED) thanks to the unique emissive behavior of organometal-

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lic complexes used as emitting layer of the devices [9–12]. The organometallic complexes offer important optical and electronic properties arising from the interplay between the inorganic materials and the organic ligands. The combination of metals with organic molecules seems to offer possible solutions to many problems in metal extraction and purification [13], in formulation of heat-resistant polymers [14], enhanced photocatalytic performance [15], asymmetric supercapacitors [16–18], surface coatings and chemical synthesis [19–21]. Tris(8-hydroxyquinoline) metals, Mq_3 ($M = Ga$ and Al), are the most widely studied organometallics for their interesting optoelectronics properties [22–24], photonic response selectivity and thermal stability [25,26]. The use of Alq_3 and Gaq_3 has also been extended for the application of organic solar cells (OSCs) as buffer/hole blocking layer between the active layer and aluminum electrode [27], and as a dopant molecular acceptor in the active layer of solution-processed OSCs [28,29]. It has been found in these studies that the contribution of Gaq_3 and Alq_3 is to improve the overall performance, reproducibility and stability of OSCs.

It is obvious that intrinsic and extrinsic characteristics of organic materials are playing a vital role in the performance selectivity of organic electronics devices. In these contexts, various approaches have been reported to modify the nanoscale properties of organic materials for desired applications [30–32]. Although these changes are responsible to the enhancement of the device's performance, there is yet a lack of complete understanding and reasoning for such improvements. That is because the operation of electronic devices is highly dependent on the nanoscale behavior, size and architecture of their active organic components [33–37], which is not an easy task to be fully addressed due to both instrumental limitations and sophisticated phenomena at the low dimensional scales.

The morphology of organic films are highly affected by the nature of molecular aggregations due to weak interactions between adjacent molecules, such as $\pi-\pi$ interaction, Van der Waals forces and hydrogen bonding [38]. The resultant forces define the packing pattern of the molecules and hence the shaping of their final periodicity in terms of structural distribution [39,40]. Hence, efforts on modifying/tuning molecular aggregations are of great importance for desired applications, especially when different nanostructures are in concern [41–45]. Nanostructured sizes and shapes are capable of providing enhanced charge carriers mobility in a specific direction, which is quite interesting for the application of single-crystalline based devices [46–48]. Xu et al. produced crystalline Alq_3 sub-microwires have by a one-step anti-solvent (the solvent with poor solubility to the solute) diffusion method combined solvent-evaporation-induced self-assembly without surfactant assistant [49], whereas Boulet et al. used a combination of evaporative and antisolvent crystallization to generate vertically-oriented hexagonal prism arrays of Alq_3 , and vertical half-disks and sharp-edged rectangular prisms of Znq_2 [50]. Besides, Gu et al. reported that through a suitable combination of solvent and anti-solvent with controllable surface tension difference, the droplets can automatically cracked into the micro-droplets,

which provides an aggregation force directing toward the center of the droplet to drive the low-dimensional building blocks to form the special aggregations during the self-assembly process [51]. The concept of antisolvent approach was also realized Yin et al. to achieve a sequential epitaxial growth to synthesize dual-color-emitting organic heterostructures with 9,10-bis(phenylethynyl) anthracene (BPEA) microwire trunks and Alq_3 microstructure branches [52].

In recent years, investigations on photoluminescence (PL) intensity of Alq_3 and its structural related consequences have become the subject of a lively debate, whereby different explanations for PL blueshift were proposed [53–55]. On one hand, the PL blueshift has been ascribed to the roleplay between meridional and facial isomers [56,57], and on the other hand to the intermolecular interactions of the meridional isomer alone [55]. PL blueshift in Alq_3 was wrongly attributed to the Rayleigh scattering phenomenon [58,59], and this explanation appeared rather vague in Gaq_3 , where it has been noticed that the energy gap of non-crystalline Gaq_3 film is red shifted, while its PL is blue shifted [26]. Hence, it was concluded that intermolecular interactions and molecular rearrangements are responsible for the observed blueshift [26,53]. Baldacchini et al. [60] claimed the existence of four different components of molecular aggregation, because the PL of Alq_3 films decays in air with four different amplitudes and constant times, which has led to formulize the Four Components Model (FCM). In an extended approach, the stretched FCM model, was able to fit the PL of Alq_3 molecules during a decay of six years in Air, and an interpretation of the physical phenomena has been given by introducing a 0th-order kinetics [53].

It is commonly held that structure and morphology of Mq_3 films play a crucial role in defining their PL behavior, hence any systematic study addressing correlation between PL and molecular aggregations is of special importance. Moreover, PL quenching and degradation processes have been often ascribed to the formation of crystallites, migration of ionic and molecular species, and chemical reactions, but none of them have been quantified in a simple and reproducible way [61–63]. We have previously showed that meridional α -polymorph can be grown within the amorphous phase of Gaq_3 films with increasing temperature, and hence contributing to a PL blueshift [64]. However, the weakly onset of a Gaq_3 polymorphism at high temperatures has introduced a dubious debate, because of the opposite shifting of PL and energy gap.

In the current research work, an exclusive attention is devoted to any possible correlation between molecular aggregations and PL position or PL intensity in Gaq_3 films. For this purpose, morphological growth states and PL profile of Gaq_3 films have been specifically investigated at temperatures from 85 to 255 °C in dry N_2 gas. The contribution of this study is to understand how molecular aggregations, either to be polymorphism or polyamorphism, contribute to the PL blueshift and luminescence variation in Gaq_3 , and in Mq_3 films. The remainder of the paper is organized as follows: the methodology of the film formation and temperature induced morphological modifications are presented in section 2, followed by analysis and discussion of the obtained results in section 3. Finally, the main conclusions are given in section 4.

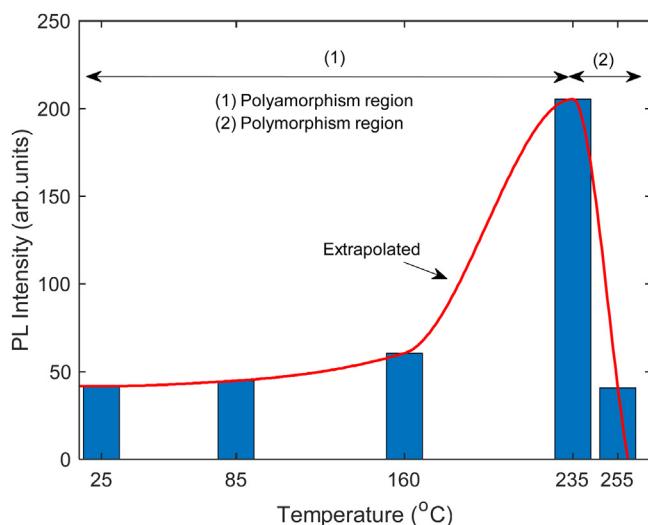


Fig. 1 – PL intensity of Gaq_3 film with increased annealing temperature.

2. Materials and methods

Tris(8-hydroxyquinolinate) gallium (Gaq_3), $\text{Ga}(\text{C}_9\text{H}_6\text{NO})_3$, was purchased from Sigma-Aldrich in powder form and used as received without further purification. Precleaned quartz substrates were used at room temperature (RT) to deposit Gaq_3 films onto them via a homemade thermal evaporator under a pressure of about 10^{-4} mbar. The quartz substrates were cleaned with Decon® Neutracon foam for 15 min in ultrasonic bath. Later on, they were rinsed and sonicated with acetone, ethanol and distilled water sequentially for 10 min. Lastly, the quartz substrates were dried thoroughly by nitrogen gas. Thickness of the deposited Gaq_3 films was measured by a KLA Tensor P-6 surface profilometer. The films were post annealed within a barrel furnace under dry nitrogen gas for 10 min. The annealing temperatures were set at 85, 160, 235, and 255 °C. The PL spectra were measured with a LS50B PERKIN ELMER luminescence spectrometer in the wavelength range from 200 to 800 nm at the excitation wavelength of 396 nm. X-ray diffractometer (Bruker AXS) was applied to record the XRD patterns using Cu K_α radiation of wavelength $\lambda = 1.5406 \text{ \AA}$ as a source. Field emission scanning electronic microscopy, FESEM (Quanta 200 F), was utilized to capture the morphology of the films before and after annealing processes. The photo absorption properties of the films were assessed by a Jasco V-570 UV-VIS-NIR spectrophotometer.

3. Results and discussion

The strength of PL emission of Gaq_3 films at increased annealing temperatures is reported in Fig. 1; an extrapolated red line joins the peak values smoothly in order to better observe the trend of the changes. The presence of a unique PL peak, which reaches a maximum of five times higher than that of the pristine film, at temperature of 235 °C, could be a consequence of the structural and thickness variations of the films. Indeed, it has been reported that PL intensity depends on the isomeric

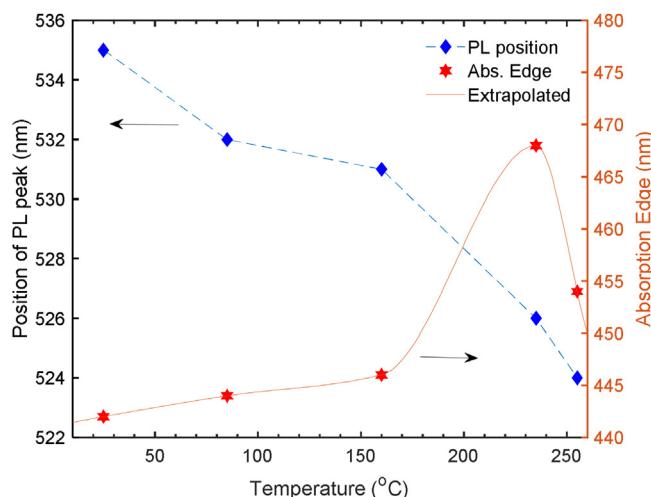


Fig. 2 – Wavelength position of PL peak and absorption edge of Gaq_3 film with increased annealing temperature.

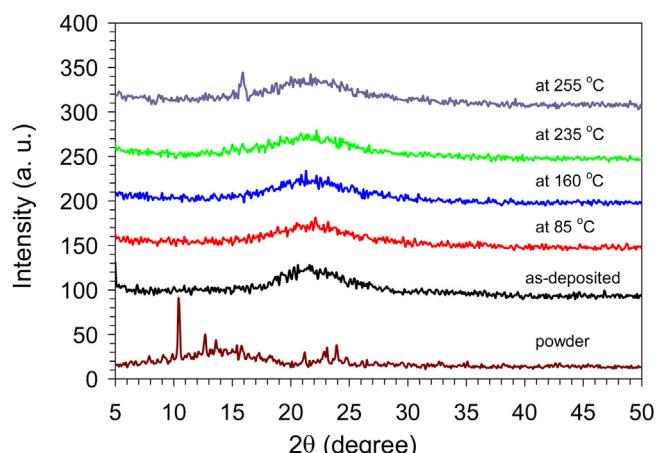


Fig. 3 – XRD patterns for the pristine and annealed Gaq_3 films under N_2 gas.

type and structure of the film [25,54]. It was observed that PL intensity is increased with the film thickness [65], and it is quenched with increasing crystallinity [26,66].

In our films, the impact of thickness is ruled out for two reasons. First, a notable shrink in the energy gap of the films, due to enhanced overlap of compact molecules and denser $\pi-\pi$ stacking [58,65], was found by increasing temperature [26]. Hence, an opposite trend of decreasing PL intensity, due to denser film, should have been observed along the entire temperature range, which is not the case (see Fig. 1). Second, the absorption edge of the energy gap and the wavelength position of the PL peak was seen to be changed in opposite manner when annealing temperature was increased, as shown in Fig. 2. Therefore, the consistent increase of PL emission in Fig. 1, and noteworthy blueshift of the PL peak in Fig. 2, can be very likely attributed to the morphological changes or molecular aggregations of the Gaq_3 .

We noticed that α -polymorph can be produced from the amorphous phase of meridianal Gaq_3 upon rising temperature, leading to a quenched luminescence and PL blueshift [64].

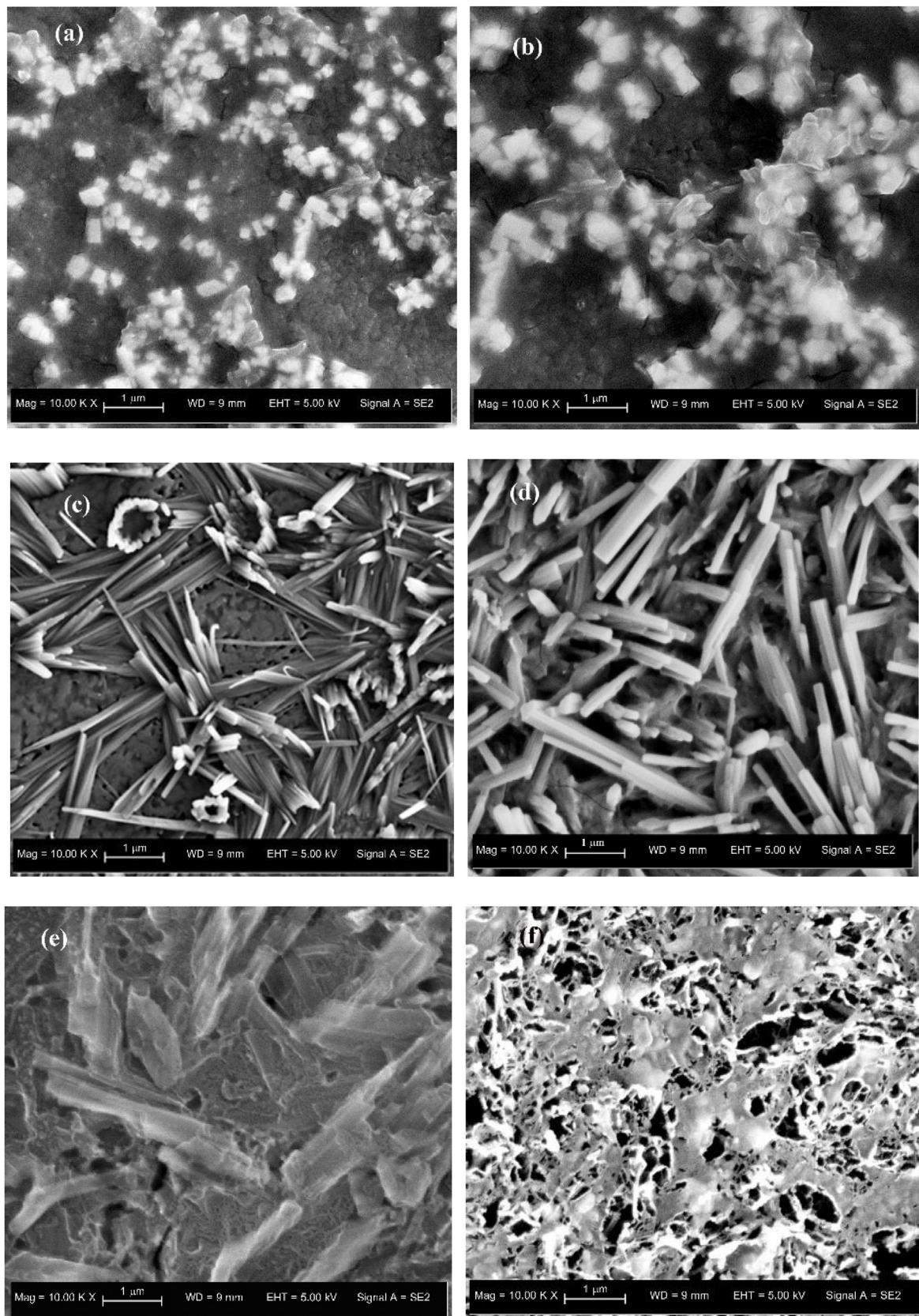


Fig. 4 – Molecular aggregation images of Gaq₃ films at RT (a), 85 °C (b), 160 °C (c), 235 °C (d), 255 °C (e) and 310 °C (f), captured by FESEM technique.

Meanwhile, it was observed from the XRD measurements (see Fig. 3) that this polymorphism occurs weakly at 255 °C, that is beyond the glass transition temperature (T_g) [26]. Because of these findings and that Gaq₃ films are yet in the amorphous phase, the steady rise of PL emission and its continuous blueshift, from room temperature (RT) to 235 °C, require a new interpretation for the contribution of amorphous states (we call it here polyamorphism) to these optical changes. Therefore, the attribution of the PL blueshift to Rayleigh scattering alone [58,59] should be considered as untenable explanation.

PL blueshift has been already ascribed to the intermolecular interactions of the meridional isomer [55], whereas molecular rearrangements can play a vital role in this shift [26]. Although morphological changes of Alq₃ at different temperatures have not yet accurately measured, Baldacchini et al. [53] reported the existence of four different components of molecular aggregations, claiming that the PL of Alq₃ films decays in air following four different pathways. To clarify the previous lacking of structural and morphological states, we have specifically taken into consideration the deep surface images of Gaq₃ films at different temperatures, as shown in Fig. 4. Upon a close look at the surface of the Gaq₃ films, it is clear that temperature has produced pronounced modifications in the molecular aggregations. In particular, four different phases of molecular aggregation/rearrangement have been observed. In the first phase, from RT to about 160 °C, Fig. 4(a) and (b), there exist bunches of molecular aggregation, while at 160 °C, Fig. 4(c), the second phase of molecular aggregation is observed in the form of a grown grass. Increasing the temperature to about 235 °C led to the formation of a third phase in the form of nanorod aggregation, Fig. 4(d), while the laminated form of molecular aggregation at 255 °C can be considered as the fourth phase that was resulted from degradation of the nanorods due to hard heating, as shown in Fig. 4(e). Noteworthy, at the high enough temperature of about 310 °C, the film burned out with clear dark spots left on the surface of the substrate, as shown in Fig. 4(f).

Because of the structural changes in the films, the crystallites were formed only at 255 °C. Therefore, the first three phases of the molecular aggregations are amorphous packings, while crystalline structures are assembled in the fourth phase, very likely α -polymorph. In conclusion, polyamorphism seems to be the main factor affecting luminescence enhancement and its blueshift in the organometallic complexes, such as Alq₃, Gaq₃, and Inq₃. In the present case, meridional Gaq₃ films under flowing N₂ gas at atmospheric pressure are stable in different amorphous phases from RT to about 235 °C. It is worth noting that the onset of meridional α -polymorph at 255 °C is a further evidence of the difficulties, if not impossibility, to activate thermally the isomeric transition from meridional to facial in Gaq₃ films, which is in agreement with the results obtained in Alq₃ films deposited by thermal evaporation [55].

When higher density Gaq₃ nanorods was formed, larger PL intensity was observed on the emission spectra, as in Fig. 1. This result is similar to that obtained for α -polymorph Alq₃ nanowires grown under Ar gas at appropriate pressure [67], except that in our films complete amorphous nanorods were formed. However, crystalline Gaq₃ in the form of one-

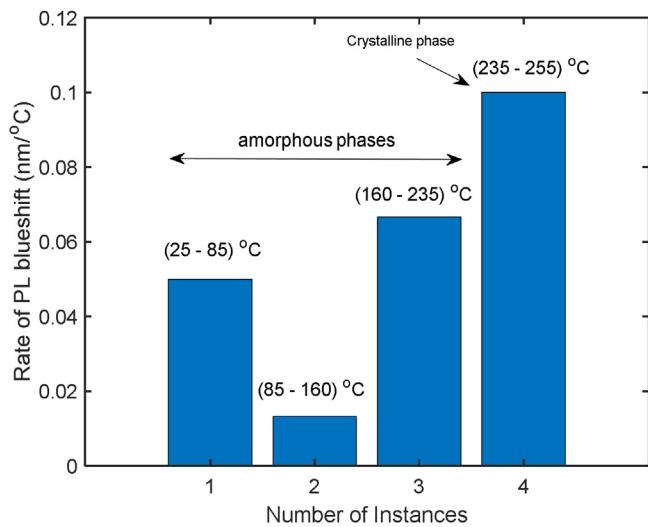


Fig. 5 – The rate of PL blueshift of Gaq₃ films with temperature.

dimensional nanostructure and nanosphere, fabricated by thermal evaporation under cold trap in the presence of pressurized He or Ar gas, showed similar PL intensity [68]. These differences in the formation of various nanostructures and their optical properties might be ascribed to the effects of gases and pressures utilized during the grown process [25,69]. Anyway, it seems well ascertained that PL intensity of amorphous nanostructures is always higher than that of their crystalline counterparts. Indeed, this feature could be a consequence of excited charge carriers in crystalline regions moving at longer distances before their recombination or relapsing, leading to an increased non-radiative probability.

Now, considering the rate of PL blueshift with temperature, as shown in Fig. 5, it is observed that the crystalline α -polymorph produced the highest PL blueshift, followed by the third amorphous phase in which the nanostructures possess the highest density. It is concluded that the nanostructured amorphous phases (polyamorphism) is highly contributed to the PL blueshift, which adds on the importance of such molecular aggregations on PL intensity and its blueshift in these organometallic complexes. Nevertheless, a small contribution may also derive by the Rayleigh scattering in the crystalline phase, although such phenomenon was excluded in Alq₃ [49]. More detailed information can be obtained on isomeric transitions and their phases, only by performing new systematic experiments at more annealing temperatures and gases.

4. Conclusions

A careful study on Gaq₃, in particular its PL and blueshift at different annealing temperatures under dry N₂ gas was performed. It has been observed at 235 °C that the PL intensity increased five times with respect to the pristine film, and at the same time, a PL blueshift is emerging in a consistent way as temperature increases. Moreover, four different phases of molecular aggregations have been observed by means of field emission microscopy. Of them, the first three are polyamorph

bunches, while the fourth one appears at 255 °C, and very likely is a crystalline α -polymorph. These crystallites produced the highest blueshift, soon followed by the third amorphous phase that possesses the highest density. However, x-rays diffraction has shown that the amount of crystalline aggregations is very low in the annealed films at 255 °C, so that their contribution to the Rayleigh scattering is scarcely significant, if any. On the contrary, a continuous blueshift is produced by the three amorphous phases from RT to 235 °C, which are composed of meridional isomers and are the bulk of films. It can be concluded that nanostructured molecular aggregation is the main reason affecting PL and blueshift in particular and the optical properties of organometallic complexes in general.

Conflict of interest

The authors declare no conflicts of interest.

Data availability

The raw/processed data required to reproduce these findings are presented in the manuscript.

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REFERENCES

- [1] McAlpine MC, Friedman RS, Jin S, Lin K-h, Wang WU, Lieber CM. High-performance nanowire electronics and photonics on glass and plastic substrates. *Nano Lett* 2003;3:1531–5.
- [2] Sekitani T, Nakajima H, Maeda H, Fukushima T, Aida T, Hata K, et al. Stretchable active-matrix organic light-emitting diode display using printable elastic conductors. *Nat Mater* 2009;8:494.
- [3] Muhammad FF, Sulaiman K. Tuning the optical band gap of DH6T by Alq3 dopant. *Sains Malays* 2011;40:17–20.
- [4] Muhammad FF, Yahya MY, Aziz F, Rasheed MA, Sulaiman K. Tuning the extinction coefficient, refractive index, dielectric constant and optical conductivity of Gaq3 films for the application of OLED displays technology. *J Mater Sci Mater Electron* 2017;28:14777–86.
- [5] Kim M-J, An TK, Kim S-O, Cha H, Kim HN, Xiofeng T, et al. Molecular design and ordering effects of alkoxy aromatic donor in a DPP copolymer on OTFTs and OPVs. *Mater Chem Phys* 2015;153:63–71.
- [6] Omidvar A. Electronic structure tuning and band gap opening of nitrogen and boron doped holey graphene flake: the role of single/dual doping. *Mater Chem Phys* 2017;202:258–65.
- [7] Lim LW, Teh CH, Daik R, Sari NM, Teridi MAM, Muhammad FF, et al. Synthesis and characterization of 2, 2'-bithiophene end-capped dihexyloxy phenylene pentamer and its application in a solution-processed organic ultraviolet photodetector. *RSC Adv* 2016;6:61848–59.
- [8] Gahungu G, Zhang J. "CH"/N substituted mer-Gaq3 and mer-Alq3 derivatives: an effective approach for the tuning of emitting color. *J Phys Chem B* 2005;109:17762–7.
- [9] Chen D, Su S-J. White organic light-emitting diodes based on organometallic phosphors. In: Organometallics and related molecules for energy conversion. Springer; 2015. p. 285–311.
- [10] Dodabalapur A. Organic light emitting diodes. *Solid State Commun* 1997;102:259–67.
- [11] Thompson ME, You Y, Shoustikov A, Sibley S, Burrows PE, Forrest SR. OLEDs doped with phosphorescent compounds. Google Patents; 2001.
- [12] Singh D, Nishal V, Bhagwan S, Saini RK, Singh I. Electroluminescent materials: metal complexes of 8-hydroxyquinoline—a review. *Mater Des* 2018;156:215–28.
- [13] Efome JE, Rana D, Matsuura T, Lan CQ. Effects of operating parameters and coexisting ions on the efficiency of heavy metal ions removal by nano-fibrous metal-organic framework membrane filtration process. *Sci Total Environ* 2019;674:355–62.
- [14] Zhang J, Zhang J, Imler GH, Parrish DA, Shreeve JM. Sodium and potassium 3, 5-dinitro-4-hypyrazolate: three-dimensional metal-organic frameworks as promising super-heat-resistant explosives. *Acs Appl Energy Mater* 2019;2:7628–34.
- [15] Zhang GC, Zhong J, Xu M, Yang Y, Li Y, Fang Z, et al. Ternary BiVO4/NiS/Au nanocomposites with efficient charge separations for enhanced visible light photocatalytic performance. *Chem Eng J* 2019;375:122093.
- [16] Feng M, Gu J, Zhang GC, Xu M, Yu Y, Liu X, et al. Homogeneous nickel bicarbonate nanocrystals as electrode materials for high-performance asymmetric supercapacitors. *J Solid State Chem* 2020;282:121084.
- [17] Chu Y, Zhong J, Fang Z, Yang Y, Qi J, Yu S, et al. Ni (HCO3) 2 nanosheet/nickel tetraphosphate (Ni (P4O11)) nanowire composite as a high-performance electrode material for asymmetric supercapacitors. *Nanotechnology* 2019;31:015401.
- [18] Wang Z, Gu J, Li S, Zhang GC, Zhong J, Fan X, et al. One-step polyoxometalates-assisted synthesis of manganese dioxide for asymmetric supercapacitors with enhanced cycling lifespan. *ACS Sustain Chem Eng* 2018;7:258–64.
- [19] Begum S, Hassan Z, Bräse S, Wöll C, Tsotsalas M. Metal-organic framework-templated biomaterials: recent progress in synthesis, functionalization, and applications. *Acc Chem Res* 2019;52:1598–610.
- [20] Zhang L, Liu H, Shi W, Cheng P. Synthesis strategies and potential applications of metal-organic frameworks for electrode materials for rechargeable lithium ion batteries. *Coord Chem Rev* 2019;388:293–309.
- [21] Zhang W, Wang S, Yang F, Yang Z, Wei H, Yang Y, et al. Synthesis of catalytically active bimetallic nanoparticles within solution-processable metal-organic-framework scaffolds. *CrystEngComm* 2019;21:3954–60.
- [22] Muhammad FF, Hapip AIA, Sulaiman K. Study of optoelectronic energy bands and molecular energy levels of tris (8-hydroxyquinolate) gallium and aluminum organometallic materials from their spectroscopic and electrochemical analysis. *J Organomet Chem* 2010;695:2526–31.
- [23] Muhammad FF, Sulaiman K. Utilizing a simple and reliable method to investigate the optical functions of small molecular organic films—Alq3 and Gaq3 as examples. *Measurement* 2011;44:1468–74.
- [24] Muhammad FF, Yahya MY, Ketuly KA, Muhammad AJ, Sulaiman K. A study on the spectroscopic, energy band, and optoelectronic properties of α , ω -dihexylsexithiophene/tris (8-hydroxyquinolate) gallium blends; DH6T/Gaq3

- composite system. *Spectrochim Acta A Mol Biomol Spectrosc* 2016;169:144–51.
- [25] Hernández I, Gillin W, Somerton M. Spectroscopic study of Mq_3 ($M = Al, Ga, In$, $q = 8$ -hydroxyquinolinate) at high pressure. *J Lumin* 2009;129:1835–9.
- [26] Muhammad FF, Sulaiman K. Effects of thermal annealing on the optical, spectroscopic, and structural properties of tris (8-hydroxyquinolinate) gallium films grown on quartz substrates. *Mater Chem Phys* 2011;129:1152–8.
- [27] Vivo P, Jukola J, Ojala M, Chukharev V, Lemmetyinen H. Influence of Alq_3/Au cathode on stability and efficiency of a layered organic solar cell in air. *Sol Energy Mater Sol Cells* 2008;92:1416–20.
- [28] Muhammad FF, Yahya MY, Sulaiman K. Improving the performance of solution-processed organic solar cells by incorporating small molecule acceptors into a ternary bulk heterojunction based on DH6T: Mq_3 : PCBM ($M = Ga, Al$). *Mater Chem Phys* 2017;188:86–94.
- [29] Muhammad FF. Impedance spectroscopy analysis of DH6T: PCBM bulk heterojunction incorporating Gaq_3 : experiment and model. *J Mater Sci Mater Electron* 2016;27:637–44.
- [30] Kim JS, Park JH, Lee JH, Jo J, Kim D-Y, Cho K. Control of the electrode work function and active layer morphology via surface modification of indium tin oxide for high efficiency organic photovoltaics. *Appl Phys Lett* 2007;91:112111.
- [31] Virkar AA, Mannsfeld S, Bao Z, Stigelin N. Organic semiconductor growth and morphology considerations for organic thin-film transistors. *Adv Mater* 2010;22:3857–75.
- [32] Kango S, Kalia S, Celli A, Njuguna J, Habibi Y, Kumar R. Surface modification of inorganic nanoparticles for development of organic-inorganic nanocomposites—a review. *Prog Polym Sci* 2013;38:1232–61.
- [33] Ma W, Yang C, Gong X, Lee K, Heeger AJ. Thermally stable, efficient polymer solar cells with nanoscale control of the interpenetrating network morphology. *Adv Funct Mater* 2005;15:1617–22.
- [34] Oosterhout SD, Wienk MM, Van Bavel SS, Thiedmann R, Koster LJA, Gilot J, et al. The effect of three-dimensional morphology on the efficiency of hybrid polymer solar cells. *Nat Mater* 2009;8:818.
- [35] Yang SY, Shin K, Park CE. The effect of gate-dielectric surface energy on pentacene morphology and organic field-effect transistor characteristics. *Adv Funct Mater* 2005;15:1806–14.
- [36] Arnold MS, Avouris P, Pan ZW, Wang ZL. Field-effect transistors based on single semiconducting oxide nanobelts. *J Phys Chem B* 2003;107:659–63.
- [37] Muhammad FF, Sulaiman K. Photovoltaic performance of organic solar cells based on DH6T/PCBM thin film active layers. *Thin Solid Films* 2011;519:5230–3.
- [38] Yao W, Han G, Huang F, Chu M, Peng Q, Hu F, et al. “H”-like organic nanowire heterojunctions constructed from cooperative molecular assembly for photonic applications. *Adv Sci* 2015;2:1500130.
- [39] Yan Y, Zhao YS. Organic nanophotonics: from controllable assembly of functional molecules to low-dimensional materials with desired photonic properties. *Chem Soc Rev* 2014;43:4325–40.
- [40] Yao W, Zhao YS. Tailoring the self-assembled structures and photonic properties of organic nanomaterials. *Nanoscale* 2014;6:3467–73.
- [41] Nguyen T-Q, Martel R, Bushey M, Avouris P, Carlsen A, Nuckolls C, et al. Self-assembly of 1-D organic semiconductor nanostructures. *J Chem Soc Faraday Trans* 2007;9:1515–32.
- [42] Burda C, Chen X, Narayanan R, El-Sayed MA. Chemistry and properties of nanocrystals of different shapes. *Chem Rev* 2005;105:1025–102.
- [43] Li R, Hu W, Liu Y, Zhu D. Micro-and nanocrystals of organic semiconductors. *Acc Chem Res* 2010;43:529–40.
- [44] Huang L, Liao Q, Shi Q, Fu H, Ma J, Yao J. Rubrene micro-crystals from solution routes: their crystallography, morphology and optical properties. *J Mater Chem* 2010;20:159–66.
- [45] Xia Y, Yang P, Sun Y, Wu Y, Mayers B, Gates B, et al. One-dimensional nanostructures: synthesis, characterization, and applications. *Adv Mater* 2003;15:353–89.
- [46] Fan C, Zoombelt AP, Jiang H, Fu W, Wu J, Yuan W, et al. Solution-grown organic single-crystalline p-n junctions with ambipolar charge transport. *Adv Mater* 2013;25:5762–6.
- [47] Garcia-Frutos EM. Small organic single-crystalline one-dimensional micro-and nanostructures for miniaturized devices. *J Mater Chem C* 2013;1:3633–45.
- [48] Tang Q, Li H, Liu Y, Hu W. High-performance air-stable n-type transistors with an asymmetrical device configuration based on organic single-crystalline submicrometer/nanometer ribbons. *J Am Chem Soc* 2006;128:14634–9.
- [49] Xu G, Tang Y-B, Tsang C-H, Zapien J-A, Lee C-S, Wong N-B. Facile solution synthesis without surfactant assistant for ultra long Alq_3 sub-microwires and their enhanced field emission and waveguide properties. *J Mater Chem* 2010;20:3006–10.
- [50] Boulet J, Mohammadpour A, Shankar K. Insights into the solution crystallization of oriented Alq_3 and Znq_2 microprisms and nanorods. *J Nanosci Nanotechnol* 2015;15:6680–9.
- [51] Gu J, Yin B, Fu S, Feng M, Zhang Z, Dong H, et al. Surface tension driven aggregation of organic nanowires via lab in a droplet. *Nanoscale* 2018;10:11006–12.
- [52] Yin B, Gu J, Feng M, Zhang GC, Zhang Z, Zhong J, et al. Epitaxial growth of dual-color-emitting organic heterostructures via binary solvent synergism driven sequential crystallization. *Nanoscale* 2019;11:7111–6.
- [53] Baldacchini G, Chiacchiarella P, Montereali R, Pode R, Vincenti M. Singular photoluminescence behavior of Alq_3 films at very long decay time. *J Lumin* 2018;193:106–13.
- [54] Baldacchini G, Chiacchiarella P, Pode R, Vincenti M, Wang Q-M. Phase transitions in thermally annealed films of Alq_3 . *Low Temp Phys* 2012;38:786–91.
- [55] Baldacchini G, Chiacchiarella P, Reisfeld R, Zigansky E. The origin of luminescence blueshifts in Alq_3 composites. *J Lumin* 2009;129:1849–52.
- [56] Levichkova M, Assa J, Fröb H, Leo K. Blue luminescent isolated $Al q 3$ molecules in a solid-state matrix. *Appl Phys Lett* 2006;88:201912.
- [57] Amati M, Lelj F. Luminescent compounds f ac-and m er-aluminum tris (quinolin-8-olate). A pure and hybrid density functional theory and time-dependent density functional theory investigation of their electronic and spectroscopic properties. *J Phys Chem A* 2003;107:2560–9.
- [58] Higginson KA, Zhang X-M, Papadimitrakopoulos F. Thermal and morphological effects on the hydrolytic stability of aluminum tris (8-hydroxyquinoline)(Alq_3). *Chem Mater* 1998;10:1017–20.
- [59] Auzel F, Baldacchini G, Baldacchini T, Chiacchiarella P, Pode RB. Rayleigh scattering and luminescence blue shift in tris (8-hydroxyquinoline) aluminum films. *J Lumin* 2006;119:111–5.
- [60] Baldacchini G, Baldacchini T, Chiacchiarella P, Pace A, Pode RB. Photoluminescence and morphology of Alq_3 films and four-components model. *J Electrochem Soc* 2007;154:J217–25.
- [61] Aziz H, Popovic ZD. Degradation phenomena in small-molecule organic light-emitting devices. *Chem Mater* 2004;16:4522–32.
- [62] Djurišić A, Lau T, Lam L, Chan W. Influence of atmospheric exposure of tris (8-hydroxyquinoline) aluminum (Alq_3): a

- photoluminescence and absorption study. *Appl Phys A* 2004;78:375–80.
- [63] Xu M, Xu J. Nanoscale study on origins of the bright clusters in/on moisture-exposed tris (8-hydroxyquinoline) aluminum thin films. *Synth Met* 2004;145:177–82.
- [64] Muhammad FF, Sulaiman K. Optical and morphological modifications in post-thermally treated tris (8-hydroxyquinoline) gallium films deposited on quartz substrates. *Mater Chem Phys* 2014;148:473–7.
- [65] Koay J, Sharif KAM, Rahman SA. Influence of film thickness on the structural, electrical and photoluminescence properties of vacuum deposited Alq (3) thin films on c-silicon substrate. *Thin Solid Films* 2009;517:5298–300.
- [66] Singh R, Kumar J, Singh RK, Kaur A, Sood K, Rastogi R. Effect of thermal annealing on surface morphology and physical properties of poly (3-octylthiophene) films. *Polymer* 2005;46:9126–32.
- [67] Cho C-P, Yu C-Y, Perng T-P. Growth of AlQ₃ nanowires directly from amorphous thin film and nanoparticles. *Nanotechnology* 2006;17:5506.
- [68] Yu Y-W, Cho C-P, Perng T-P. Crystalline Gaq 3 nanostructures: preparation, thermal property and spectroscopy characterization. *Nanoscale Res Lett* 2009;4:820.
- [69] Akkuzina A, Kozlova N, Avetisov R, Avetisov IK. The homogeneity range of crystalline tris (8-hydroxyquinoline) gallium. Springer; 2018. p. 85–8. Doklady Chemistry.