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# Al<sub>2</sub>O<sub>3</sub> surface passivation characterized on hydrophobic and hydrophilic *c*-Si by a combination of QSSPC, CV, XPS and FTIR

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#### Abstract

In this work, the influence of the *c*-Si surface finishing (hydrophobic/hydrophilic) prior to the deposition of the  $Al_2O_3$  passivation layer on the passivation quality is investigated. The samples are characterized by a combination of Quasi-Steady-State-PhotoConductance (QSSPC) Capacity-Conductance (CV), X-ray Photoelectron Spectroscopy (XPS) and Fourier Transformed InfraRed (FTIR) measurements. Furthermore, FTIR measurements are used to determine the thickness of interfacial SiO<sub>x</sub> layer.

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# 1. Introduction

Thin (<30 nm) Al<sub>2</sub>O<sub>3</sub> layers has been proven to provide an excellent level of surface passivation of *n*and *p*-type *c*-Si due to its unique property of a high negative fixed charge density ( $Q_f$ ) and a low interface defect density ( $D_{it}$ ) [1]. In the field-effective passivation mechanism, interfacial SiO<sub>x</sub> layers between the bulk Al<sub>2</sub>O<sub>3</sub> and the *c*-Si play an important role [2-4]. The thickness and the quality of the SiO<sub>x</sub> interfacial layer is influenced by the *c*-Si surface state prior to deposition and the post-deposition anneal [1-6].

In this work,  $Al_2O_3$  films were deposited on *c*-Si with oxidized or H-terminated surfaces. Prior and after post-deposition anneal, effective minority carrier lifetimes were measured using QSSPC. For  $Al_2O_3$  layers on Czochralski (CZ) grown silicon, the fixed charge density and the interface defect density were determined using Capacitance-Voltage (C-V) measurements.  $Al_2O_3/SiO_x$  structures layers were characterized using non-destructive Fourier Transform Infrared Spectroscopy (FTIR) and X-ray

Photoelectron Spectroscopy (XPS). Furthermore, it will be shown that the thickness of the interfacial  $SiO_x$  layer can be calculated using non-destructive FTIR measurements.

# 2. Experimental

Al<sub>2</sub>O<sub>3</sub> layers (30 nm) were deposited using thermal Atomic Layer Deposition (ALD) (Al(CH<sub>3</sub>)<sub>3</sub> + H<sub>2</sub>O) at a temperature ( $T_{dep}$ ) of 200 °C, on both sides of 4" mirror polished *p*-type Float-Zone (FZ) *c*-Si with a resistivity of 2  $\Omega$ ·cm and a thickness of 200  $\mu$ m. Table 1 shows the cleaning process prior to Al<sub>2</sub>O<sub>3</sub> deposition by ALD.

Table 1. Schematic overview of c-Si cleaning process to achieve the hydrophobic (left) or hydrophilic (right) c-Si surface state

| Hydrophobic |                               |        | Hydrophilic |                                  |         |
|-------------|-------------------------------|--------|-------------|----------------------------------|---------|
|             | Solution                      | Time   |             | Solution                         | Time    |
| 1           | $H_2O + HF (2\%) + HCl (5\%)$ | 5 min. | 1.          | $H_2O + HF (2\%) + HCl (5\%)$    | 5 min.  |
|             |                               |        | 2.          | $NH_4OH + H_2O_2 + H_2O$ (1:1:5) | 10 min. |

The samples received a post-deposition thermal anneal in N<sub>2</sub> environment for 20 min. at a temperature of 300, 500 or 800 °C. FTIR spectra were constructed from 400 to 4000 cm<sup>-1</sup> by a Brucker TENSOR FT-IR spectrometer. Furthermore, the effective carrier lifetimes in the *c*-Si sample ( $\tau_{eff}$ ) were measured by QSSPC measurement using a Sinton instrument lifetime tester in the generalized mode at an injection level of  $\Delta n = 1 \times 10^{14}$  (low) or at  $\Delta n = 1 \times 10^{15}$  (high) cm<sup>-3</sup>. Prior and after post-deposition anneal (20 min. 300 °C), the thickness of the interfacial SiO<sub>x</sub> layer between a 3 nm thick Al<sub>2</sub>O<sub>3</sub> layer and the *c*-Si surface was measured using a Thermo Fish Theta 300 XPS spectrometer at an analyzing angle of 28°.

Furthermore, Al<sub>2</sub>O<sub>3</sub> films with a thickness of 5, 10, 15 or 20 nm were synthesized by an equivalent deposition process as described above, on both sides of 4" Saw Damage Removed (SDR) *p*-type Czochralski (CZ) *c*-Si with a resistivity of 1.5  $\Omega$ ·cm and a thickness of 150 ± 10 µm with hydrophilic or hydrophobic surfaces. The Al<sub>2</sub>O<sub>3</sub> layers were removed from the rear by HF vapor, prior to the deposition of 2 µm thick Al layer by e-beam evaporation. Pt dots electrodes with an area of  $(2.0 \pm 0.5) \times 10^{-3}$  cm<sup>2</sup> were sputtered on the front, after the activation of the passivation layer by a thermal anneal as described above. Capacity-Voltage and Conductance-Voltage measurements were recorded using HP 4156 precision LCR-meter at low and high (1 - 100 kHz) frequency. All measurements were carried out at room temperature.

30 nm thick  $Al_2O_3$  films were grown using the thermal ALD process as above, on both sides of 4" mirror polished *p*-type FZ *c*-Si with a resistivity of 2  $\Omega$ ·cm and a thickness of 200 µm capped by a thermal grown SiO<sub>x</sub> layer with varying layer thickness. The SiO<sub>x</sub> layer thickness is varied by etching 100 nm thick, high quality, thermal grown (1050 °C) SiO<sub>x</sub> layers using a buffer HF solution. The thickness was determined by Spectroscopic Ellipsometry (SE) (SENTECH SE400adv-PV) prior to  $Al_2O_3$  deposition. FTIR spectra from 400 to 4000 cm<sup>-1</sup> were constructed by a Brucker TENSOR FT-IR spectrometer.

#### 3. Result/Discussion

#### 3.1. c-Si surface finishing

Effective minority carrier lifetimes ( $\Delta n = 1 \times 10^{15} \text{ cm}^{-3}$ ) in H-terminated and oxidized *c*-Si samples passivated by Al<sub>2</sub>O<sub>3</sub> are shown in Fig. 1(a) as a function of the post-deposition annealing temperature. Furthermore, the ratio between  $\tau_{eff}$  at low ( $\Delta n = 1 \times 10^{14} \text{ cm}^{-3}$ ) injection level and high ( $\Delta n = 1 \times 10^{15} \text{ cm}^{-3}$ ) injection level is calculated, see Fig. 1(b).

Fig. 2(a) displays the fixed charge density and the interface defect density of *c*-Si samples passivated by an Al<sub>2</sub>O<sub>3</sub> layer with a thickness of 10 nm as a function of the annealing temperature. Fig. 2(b) shows the thickness of the Al<sub>2</sub>O<sub>3</sub> and interfacial SiO<sub>x</sub> layers prior and after a thermal anneal at 300 °C.



Fig. 1. Passivation properties of ALD Al<sub>2</sub>O<sub>3</sub> deposited on *c*-Si with H-terminated or oxidized surface after an anneal at a temperature displayed on the x-axis (N<sub>2</sub>, 20 min.) (a) Effective carrier lifetime at  $\Delta n = 1 \times 10^{15}$  cm<sup>-3</sup> (b) ratio between effective carrier lifetime at low ( $\Delta n = 1 \times 10^{14}$  cm<sup>-3</sup>) injection level and high ( $\Delta n = 1 \times 10^{15}$  cm<sup>-3</sup>) injection level. The black line is a guide for the eye

The  $Al_2O_3$  layer and the Si/ $Al_2O_3$  interface is furthermore characterized using FTIR. The absorption and differential spectrum of as-deposited  $Al_2O_3$  layers is shown in Fig. 3(a) and Fig. 3(b) displays the absorption and differential spectrum of  $Al_2O_3$  after a post-deposition anneal at 800 °C.

#### 3.2. As-deposited

For as-deposited layers, lower surface recombination velocities are observed for Al<sub>2</sub>O<sub>3</sub> deposited on *c*-Si with H-terminated surface. This difference is caused by a higher interface defect density for Al<sub>2</sub>O<sub>3</sub> deposited on *c*-Si with an oxidized surface, as shown in Fig. 2(a). Fig. 2(b) shows the presence of 1.0 nm thick interfacial SiO<sub>x</sub> layer, grown by the wet-chemical oxidation step. The low-quality oxide layer on the *c*-Si surfaces provides a poor chemical passivation, resulting in surface recombination velocity of  $270 \pm 20 \text{ cm} \cdot \text{s}^{-1}$ . A surface recombination velocity of  $11 \pm 2 \text{ cm} \cdot \text{s}^{-1}$  has been observed in *c*-Si samples with hydrophobic surface prior Al<sub>2</sub>O<sub>3</sub> deposition due to the low  $D_{ir}$ . A positive fixed charge density ( $Q_f = (8 \pm 2) \times 10^{11} \text{ cm}^{-2}$ ) is reported for as-deposited layers. The

A positive fixed charge density  $(Q_f = (8 \pm 2) \times 10^{11} \text{ cm}^2)$  is reported for as-deposited layers. The presence of positive charges is also indicated by the low  $\tau_{low}/\tau_{high}$  ratio. Martin *et al.* hypothesized that an inversion layer in the subsurface region, present due to the absence of negative fixed charges, is responsible for the reduction of lifetime at low injection level [7].

Absorption peaks related to Al-O and O-H bonds are present in the FTIR spectrum. In contrast with  $Al_2O_3$  layers synthesized by plasma-assisted ALD using  $Al(CH_3)_3$  and an  $O_2$  plasma, no C-H absorption peak is observed in the noise signal [1,8]. The  $H_2O$  precursor is less volatile than O radicals from the oxygen plasma; therefore, less C contamination is incorporated. The differential spectrum demonstrates by an absorption peak (1060 cm<sup>-1</sup>) which originates from the TO mode of thermally grown amorphous  $SiO_x$  the difference between both  $Al_2O_3/Si$  interfaces [9]. The absorption peak indicates the presence of a interfacial  $SiO_x$  layer between the  $Al_2O_3$  layer and the *c*-Si with hydrophilic surface finishing. No significant difference is visible for the absorption peak associated with Si-H bonds.



Fig. 2. (a)  $Q_f$  and  $D_{tt}$  of 10 nm as-deposited, low ( $\leq$  500 °C) or high (> 500 °C) temperature annealed Al<sub>2</sub>O<sub>3</sub> deposited on the hydrophilic or hydrophobic *c*-Si surface states. (b) Al<sub>2</sub>O<sub>3</sub> and SiO<sub>x</sub> thickness of as-deposited and annealed (N<sub>2</sub>, 20 min. at 300 °C) *c*-Si samples with hydrophilic or hydrophobic surface

#### 3.3. Post-deposition anneal

An excellent level of passivation is reported for both surface finishing after a post-deposition anneal at  $T_{ann} = 300$  °C;  $S_{eff}$  decreases to  $7.5 \pm 0.5$  cm·s<sup>-1</sup> and  $22 \pm 2$  cm·s<sup>-1</sup> for Al<sub>2</sub>O<sub>3</sub> deposited on *c*-Si in the hydrophobic or hydrophilic surface state respectively. Fixed negative charges are formed in the passivation layer ( $Q_f = -(1.0 \pm 0.7) \times 10^{12}$  cm<sup>-2</sup> for  $T_{ann} = 300$  °C) and the interface defect density of *c*-Si samples with a hydrophilic surface state decreases while the  $D_{it}$  of passivated *c*-Si samples with hydrophobic and hydrophilic surface finishing respectively after a thermal anneal at  $T_{ann} = 300$  °C). Also the ratio  $\tau_{low}/\tau_{high}$  of both surface finishes indicates that no subsurface inversion layer is present after anneal due to the fixed negative charges in the passivation layer.

Fig 2(b) shows the formation of an interfacial SiO<sub>x</sub> layer between the Al<sub>2</sub>O<sub>3</sub> and the hydrophobic *c*-Si surfaces after a thermal anneal at 300 °C while the thickness of the SiO<sub>x</sub> interfacial layer between Al<sub>2</sub>O<sub>3</sub> and the hydrophilic *c*-Si surfaces remains constant. The formation of a high quality (confirmed by a low  $D_{it}$ ) interfacial SiO<sub>x</sub> layer for both *c*-Si surface finishes is also depicted by an absorption peak at 1060 cm<sup>-1</sup> in the FTIR spectrum.

For  $T_{ann} = 500$  °C, the fixed charge density ( $Q_f = -(2.4 \pm 0.4) \times 10^{12}$  cm<sup>-2</sup>) increases compared with layers annealed at 300 °C, the interface defect density increases slightly ( $D_{it} = (10 \pm 1) \times 10^{10}$  and ( $7 \pm 1$ )  $\times 10^{10}$  cm<sup>-2</sup> for hydrophobic and hydrophilic surface finishing respectively, leading to an increase of the effective surface recombination velocity. No difference in surface passivation is observed between both surface finishes. By inspecting the differential spectrum, an interfacial SiO<sub>x</sub> layer of equivalent thickness is confirmed by the absence of an absorption peak at 1060 cm<sup>-1</sup>.

At  $T_{ann} > 500$  °C, surface recombination velocities increase due to blistering formation and an increase of the interface defect density (see Fig. 2(a)) [10]. Si-H bonds break at higher annealing temperature and H is effused from the *c*-Si interface resulting in dangling bonds, and therefore, an increase of the  $D_{it}$  [11]. The use of forming gas (H<sub>2</sub>) during the annealing would probably prevent the increase of  $D_{it}$ . Furthermore, fixed charges are removed from the *c*-Si surface due to blistering of the passivation layer. The negative fixed charge density is not affected by the high (> 500 °C) annealing temperature, leading to the conclusion that the level of passivation is reduced due to blistering formation and an increase of the interface defect density.



Fig. 3. Infrared absorption spectra of 30 nm  $Al_2O_3$  deposited on both sides of *c*-Si with hydrophobic or hydrophilic surfaces. The differential absorption spectrum is displayed to indicate the difference between the passivation layers. (a) The spectrum of as-deposited  $Al_2O_3$  layers and (b) after a post-deposition anneal (N<sub>2</sub>, 20 min. at 500 °C). The spectra are off-set for clarity

#### 3.4. Interfacial SiO<sub>x</sub> thickness

As described in Section 3.1 above, the interfacial  $SiO_x$  layer causes an absorption peak at 1060 cm<sup>-1</sup> in the FTIR spectrum. For non-destructive determination of the thickness of the interfacial  $SiO_x$  layer after annealing, FTIR spectra of  $Al_2O_3/SiO_x/c$ -Si structures with 30 nm thick  $Al_2O_3$  and varying  $SiO_x$  thickness are constructed. Fig. 4 shows the absorbance at 1060 cm<sup>-1</sup> as a function of the  $SiO_x$  thickness.



Fig. 4. Absorbance at 1060 cm<sup>-1</sup> of the  $Al_2O_3/SiO_x/c$ -Si samples a function of the  $SiO_x$  thickness. The  $SiO_x$  thickness has been measured by SE prior to the deposition of the 30 nm thick  $Al_2O_3$  layer by ALD

As displayed in Fig. 4, the absorbance at 1060 cm<sup>-1</sup> depends linearly on the thickness of the  $SiO_x$  layer. This can be understood, since, FTIR spectra apply to Lambert-Beer's law:

$$-ln[I/I_0] = \alpha l \tag{1}$$

With *I* the transmittance intensity of the sample,  $I_0$  the absorbance of the reference sample (bare *c*-Si), the  $\alpha$  absorbance coefficient and *l* the transmittance length. Eq. (1) shows that the logarithm of the absorption divided by the absorption of the reference sample depends linearly on the transmittance length. Therefore, Fig. 4 can be used to determine the thickness of the SiO<sub>x</sub> layer after a thermal treatment, under the assumption that the quality of the SiO<sub>x</sub> interlayer is equal to the quality of the thermally grown SiO<sub>x</sub> layer. The linear fit shows that  $\alpha = (6.5 \pm 0.1) \times 10^{-3} \text{ nm}^{-1}$ , leading to a SiO<sub>x</sub> thickness of 2.0 ± 0.2 and 1.7

 $\pm$  0.2 nm for the hydrophilic and hydrophobic *c*-Si finishing respectively after a post-deposition anneal at 500 °C, thus, no difference interfacial SiO<sub>x</sub> layer is observed between both *c*-Si surface finishes after a thermal anneal at 500 °C.

# 4. Conclusion

The influence of the *c*-Si surface finishing on the characteristics of  $Al_2O_3$  passivation layers has been investigated by a combination of FTIR, CV, XPS and QSSPC measurements. It has been proven that FTIR is powerful, non-destructive technique to characterize the thickness of the interfacial SiO<sub>x</sub> layer. The interface defect density is the dominating parameter for the surface passivation quality. The hydrophobic *c*-Si surface finishing is preferred because of the low  $D_{it}$  for as-deposited layers; however, an equivalent level of surface passivation has been observed after a thermal treatment. For both thicknesses, a high quality SiO<sub>x</sub> interfacial layer is formed during a thermal treatment.

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