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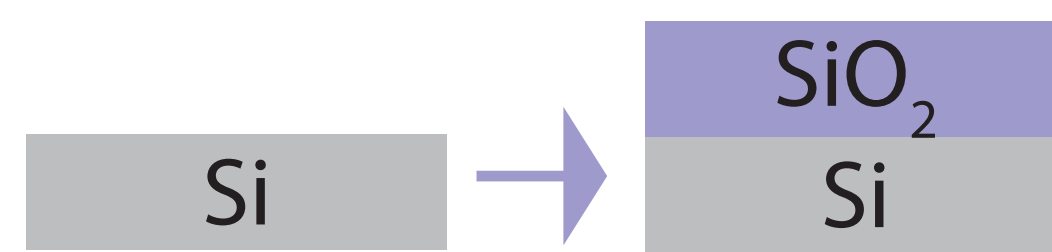
INTRODUCTION

Chemical vapour deposition (CVD) is known as the most promising route for industrially applicable wafer scale graphene synthesis. The CVD process mainly relies on the decomposition of a gaseous carbon source on a metal catalyst at high temperatures. Due to the e.g. inhomogeneous out-diffusion of carbon and metal groove formation, uniform graphene synthesis is still challenging. A new promising catalyst for uniform mono and multi-layer graphene (MLG) synthesis with high temperature stability is Mo₂C, having noble metal like catalytic properties and low cost[1,2].

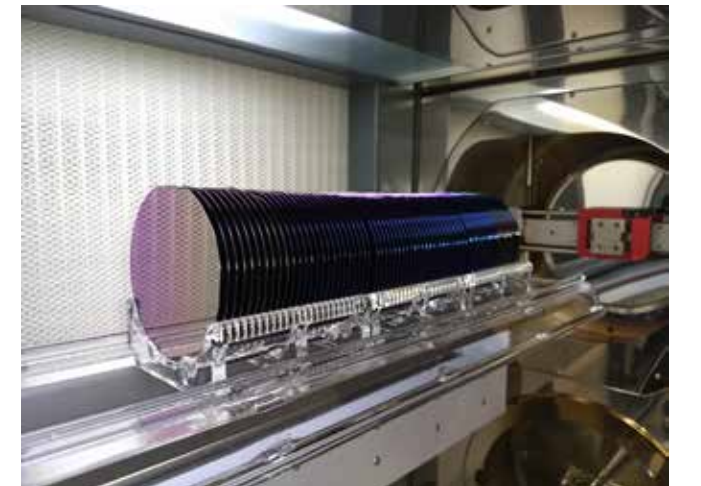
Unlike traditional graphene catalyst materials, Mo₂C is not directly deposited but is formed by a rapid transformation of a Mo layer directly after exposure to CH₄ at high temperature. The properties of the initial Mo layer, such as purity and density, are critical for the formed Mo₂C structure and also the subsequent graphene nucleation. Previously, the number of graphene layers showed to be different for Mo foils and thin films[2] but the influence of the Mo properties is not investigated. In this work, the relation is studied between graphene growth and the properties of the as deposited Mo layers before CVD, in particular the effect of Mo oxygen content and density.

PROCESS FLOW AND RESULTS

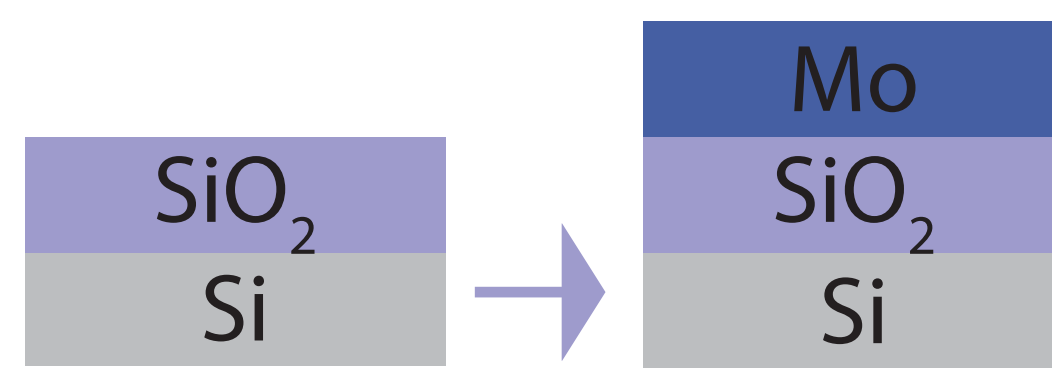
Dry oxidation of Si wafers



SiO₂ (300 nm) serves as a diffusion barrier in the CVD process



Sputtering of Mo layer



DEPOSITION SYSTEM 1

Sputter system for deposition of films with sub-nm accuracy, with possibility of ion polishing during deposition and substrate bias



DEPOSITION SYSTEM 2

Single wafer sputter coater for deposition of metallic layers



Low sputter pressure

High sputter pressure

High deposition energy

Low deposition energy

Density
from X-Ray Reflectometry

Roughness
from Atomic Force Microscopy

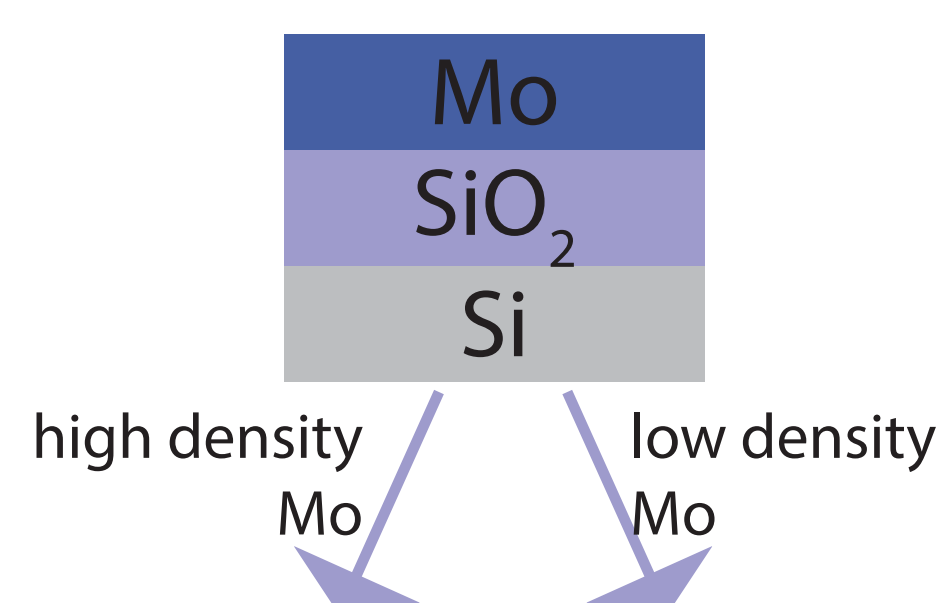
High
about 10.2 g/cm³

Low
about 8.8 g/cm³

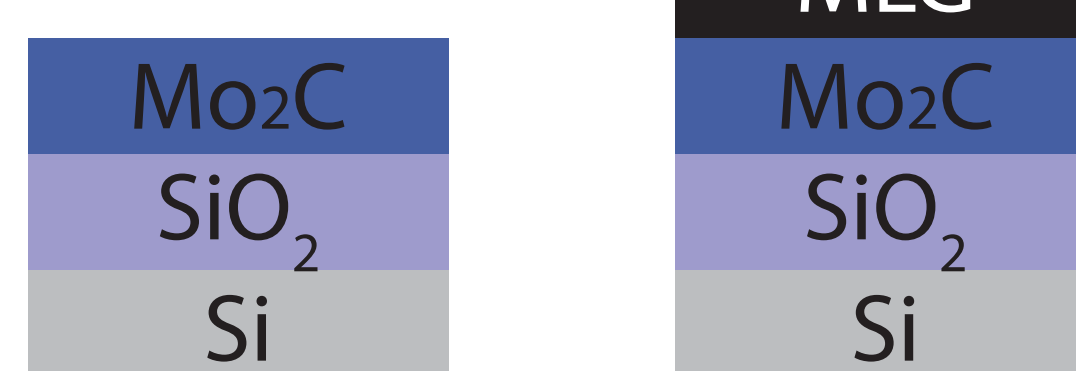
Very low
R_q = 0.3 nm

Low
R_q = 1.0 nm

CVD progress



AFTER CVD progress

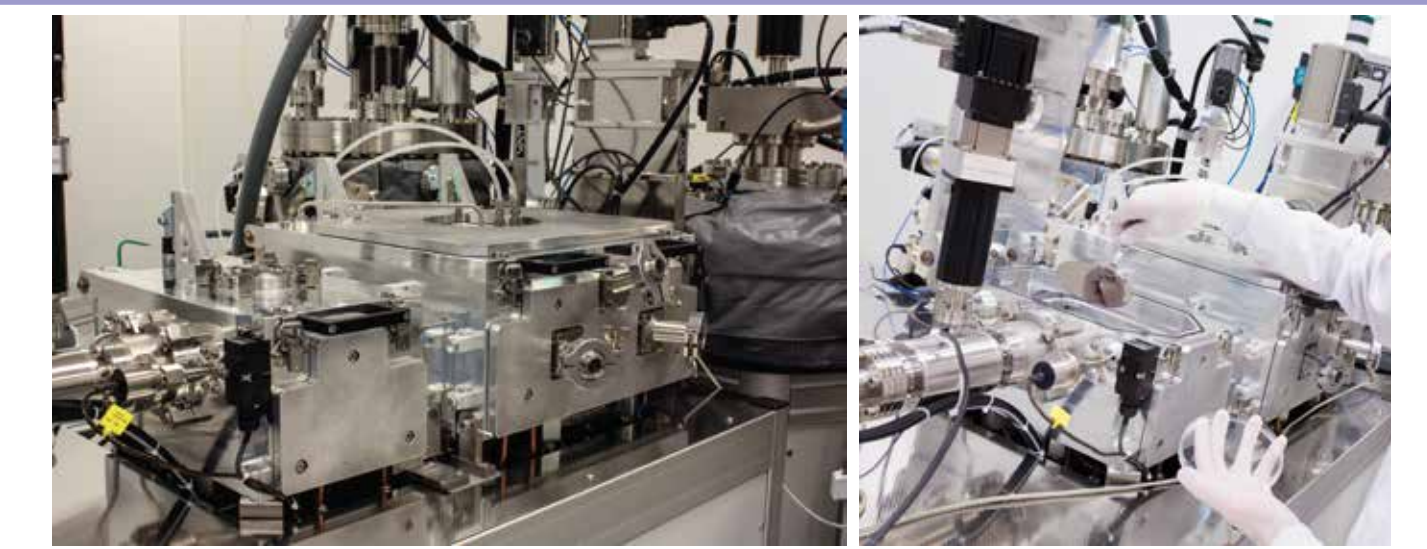


Graphene growth
from Raman spectroscopy

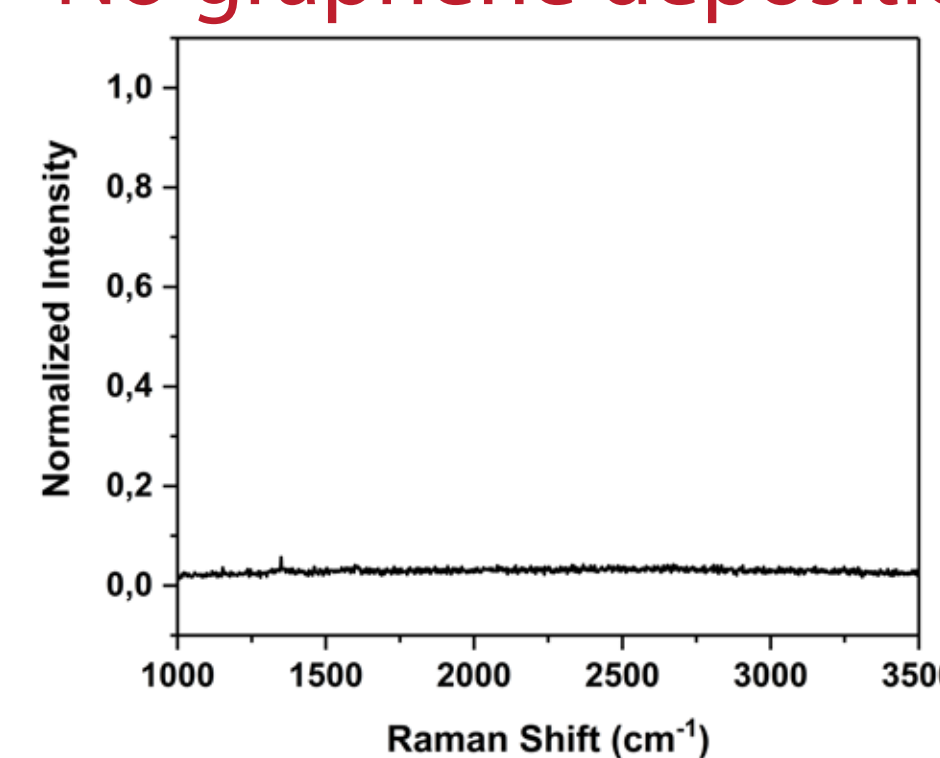
Roughness
from Atomic Force Microscopy

Defects
Observed by Scanning Electron Microscopy

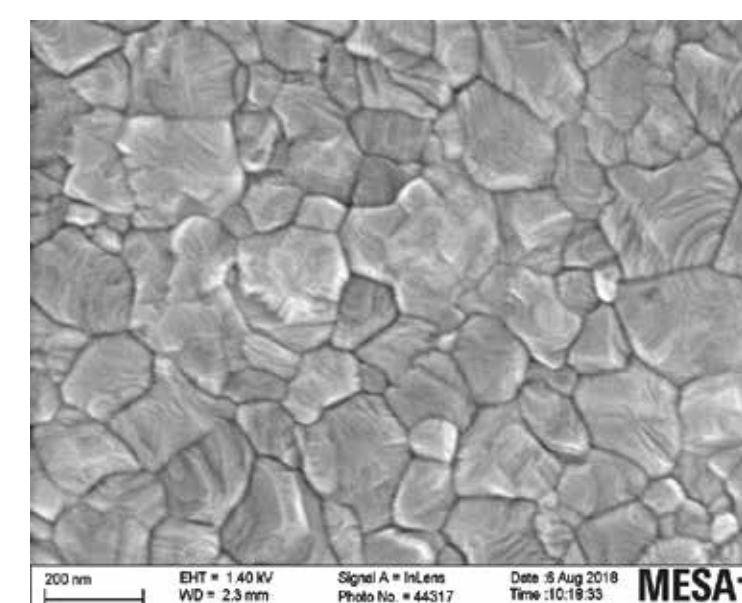
Cold wall reactor chamber
at 1000C using a feedstock of CH₄, H₂ and Ar



No graphene deposition

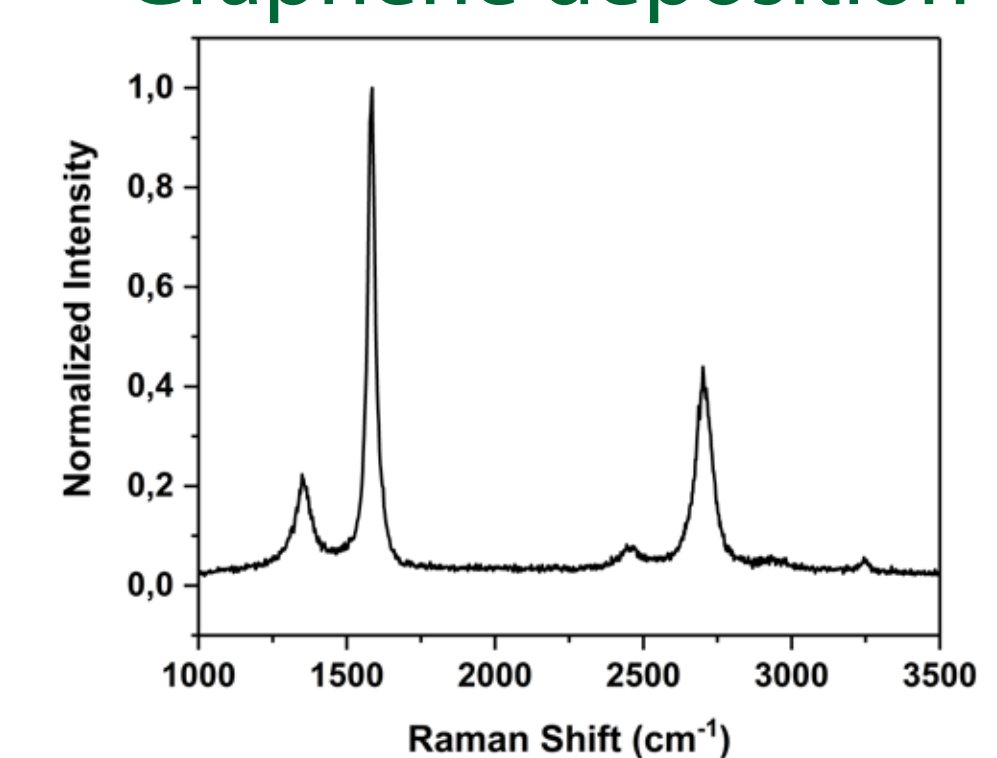


High
R_q = 4 nm

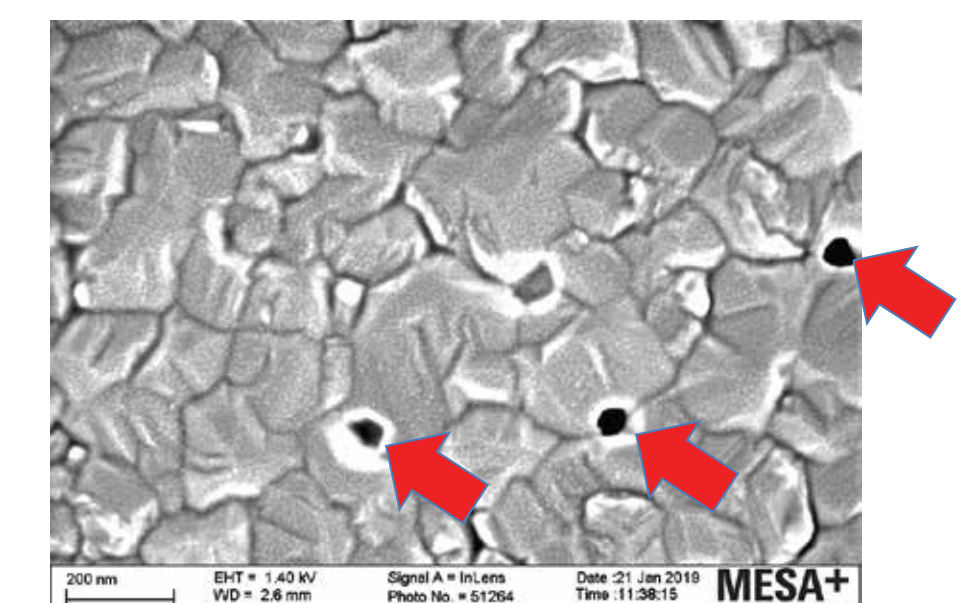


No pinholes visible

Graphene deposition



Very high
R_q = 6 nm



Pinholes visible

CONCLUSION

Results obtained from samples deposited under conditions varying the oxygen content and density show that samples with low density generally lead to graphene growth, in contrast to samples with high density. Additionally, the presence of excess oxygen may play a role in inhibiting graphene growth at higher oxygen levels. The importance of density is tentatively explained by the formation of defects in the low density layers, serving as nucleation points for graphene growth. This suggestion is supported by SEM images, which show that when starting from low density Mo layers, the resulting structure is much more open with increased surface area to volume ratio, resulting in much higher catalytic activity.

These results show that the structure of the initial Mo layer has a profound effect on the graphene growth process, and as such should always be considered in any study of graphene growth on such, and likely similar, catalysts.

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REFERENCES

- [1] Zou, Zhiyu, et al. Nano letters 14.7 (2014): 3832-3839
- [2] Grachova, Yelena, et al. Procedia Engineering 87 (2014): 1501-1504