

Oxygen diffusion characterization in ultrathin films by low energy ion scattering

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Abstract

Experimental determination of oxygen diffusion kinetic parameters for ultrathin films is of interest for applications as electronic devices and protective layers of extreme ultraviolet optics. Techniques usually applied to derive these data (XPS, SIMS) often lack details regarding surface chemistry and/or resolution. In this work, the oxygen diffusion in ultrathin films is analysed through LEIS, a technique that provides the possibility of correlating the selective analysis of the outermost atomic layer with high-resolution static depth profiling.

Low Energy Ion Scattering (LEIS)



Sputter depth profile – Peak analysis

- LEIS peaks: outermost surface characterization
- Sensitive to scattered ions



Concentration of species with accuracy at each sputter step LEIS Sputter DP – determine change in oxygen concentration in *depth*

Static depth profile – Background signal analysis

Oxide growth through Sputter and Static depth profile









✓ Metal and oxide: differences in electronic stopping powers \checkmark Thicker oxide: higher ΔE between surface peak and interface LEIS Static DP – Change oxide thickness with oxygen exposure *time* in a non-destructive way

Methodology

- Rapid formation
- Saturation thickness with time
- Inverse logarithmic growth





Depth (nm)

 \checkmark Adsorbed oxygen layer + Metal = Potential:

 $V_M = e^{-1}(\Phi_{Me} - \Phi_{o-})$

✓ Generating a field:

 $E_b = -\frac{V_M}{d_{ox}(t)}$

Driving force of oxide growth

✓ Effective while potential drop in unit cell is comparable to diffusion activation energy

Conclusions and next steps

- ✓ LEIS Static DP: Fast and non-destructive way to determine oxidation of ultrathin films
- Static + Sputter DP: complete picture of thin-film oxidation
- Thin films (Ru, Zr, Mo) oxidation at RT: Cabrera-Mott theory

- 3 keV He⁺ \rightarrow Ru, Zr and Mo single layers (15 nm)
- Atomic O: RT, 10⁻⁴ mbar



Mechanisms and kinetics of oxide growth of thin films

Oxidation

 $d_{ox}(t)$

 \checkmark Apply evaluation to different systems (varying T, materials)

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Literature

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