

Novel transport phenomena at complex oxide interfaces

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Novel electronic and magnetic phases are being observed at interfaces between insulating, non-magnetic oxide compounds, with the most notable example being the interface between SrTiO₃ and LaAlO₃. The basic properties of these interfaces will be discussed, as well as prospects for applications and possible future developments.

Introduction

Interface conduction and the exemplary LaAlO₃/SrTiO₃ two-dimensional electron gas/liquid

It is in use as artificial diamond, but for many scientists, it has the appeal of a complete jewelry shop: Strontium-titanate, SrTiO₃, the perovskite crystal structure that looks deceptively simple, but which harbors tremendously rich physics, especially when electronically doped. In 1969, Marvin Cohen, the co-discoverer of superconductivity in oxygen-reduced SrTiO₃,¹ said, “If SrTiO₃ had magnetic properties, a complete study of this material would require a thorough knowledge of all of solid state physics.”² In their 1987 Physics Nobel lecture, J. Georg Bednorz and K. Alex Müller mention SrTiO₃ together with LaAlO₃ as key materials under study, leading to the discovery of high-temperature superconductivity in cuprate perovskites.³ Notably, these two electrically insulating materials marked the beginning, in 2004, of a new chapter in the research on oxides, now focusing on their interfaces. Akira Ohtomo and Harold Y. Hwang showed that the contact area between a LaAlO₃ film and a SrTiO₃ substrate, of which the top-most atomic layer was chemically prepared to be TiO₂, is conducting.⁴ Meanwhile, many additional remarkable properties have been observed for the LaAlO₃/SrTiO₃ interface, and these even include magnetic effects. In addition, this field of research has expanded to include many other oxide interface combinations.

These developments raise the questions: What makes these oxide interfaces so special, and what could they have further in store? Before turning to these questions, it is useful to note that among scientists working in the field of oxide electronics, a distinction is made between “simple” and “complex” oxides. The former includes such materials as SiO₂, TiO₂, and ZnO, which have many interesting and useful properties—not really simple, by the way. Especially noteworthy in relation to interface conductance is research on two-dimensional electron gases (2DEGs) at the interface between ZnO and MgZnO, in which electron mobilities up to 700,000 cm²/Vs at temperatures below 1K have been reached.⁵ The observation of the fractional quantum Hall effect in these systems, requiring ultra-low defect densities, exemplifies the excellent quality of these interfaces. In many respects, the ZnO/MgZnO interfaces can be viewed as an oxide analogue to semiconductor 2DEGs such as GaAs/AlGaAs with potentially very long spin coherence, of interest for spintronic (quantum) devices.⁵

The designation of “complex” oxides is used for materials comprising—next to the oxygen anion—two or more cationic elements. Canonical examples of these are the perovskite oxides with the generic crystal unit cell composition ABO₃, in which A is typically an element from the alkaline earth or rare-earth metal groups, and B is a (post-)transition metal element (Ti, V, Mn, Fe, Cu, Ru, Ir, Al, Bi). In their stoichiometric “parent compound” phase, many of the perovskites are insulating,

either in the form of standard band insulators, as is the case for SrTiO₃ and LaAlO₃, or in the form of Mott insulators, such as LaTiO₃. The “Mottness” is a specialty of materials in which charges at the Fermi energy are loosely bound to a sublattice of ions, in principle able to hop between sites. For the case of half filling, meaning that all available states are occupied by one electron with either spin up or spin down, a conductance deadlock occurs, and the Mott-insulating state arises. Magnetic and electron-lattice interactions can give rise to various forms of spin and orbital ordering in this insulating state.

For the Mott insulators, *p* and *n* doping are both possible by shifting the chemical potential away from the half filling state using chemical substitution or electrostatic gating. This doping breaks the conductance deadlock, giving rise to an insulator-to-metal transition. For the band insulators, *p* doping is often counteracted by the creation of oxygen vacancies, making it difficult to create *p*-type oxide conductors. Electron doping, however, can be accomplished quite straightforwardly. For example, in the previously mentioned SrTiO₃, replacing part of the Ti⁴⁺ ions with Nb⁵⁺ or creating oxygen deficiency by annealing in a vacuum provides mobile electrons that fill a Ti 3*d*-derived conduction band.⁶ The conducting states of complex oxides are very tantalizing, resulting from the strong and local interplay between the charge, spin, orbital, and lattice degrees of freedom. Many special phenomena occur, such as steep metal-to-insulator transitions, non-Fermi liquid behavior, (high-*T*_c) superconductivity, and colossal magnetoresistance.⁶

Interfacing with other materials presents an alternative way of donating charge to the stoichiometric parent compound. The creation of a confining potential at interfaces, resulting, for example, from work function differences or from Fermi-level pinning by surface states, is a well-known phenomenon. These confining potentials become filled with electrons, provided by, for instance, diffusion from the adjacent materials when these are conductors or Mott insulators, by cation impurities or oxygen vacancies, by photo-excitation, or by electric field gating. These electrons can be delocalized along the interface, giving rise to a 2DEG.

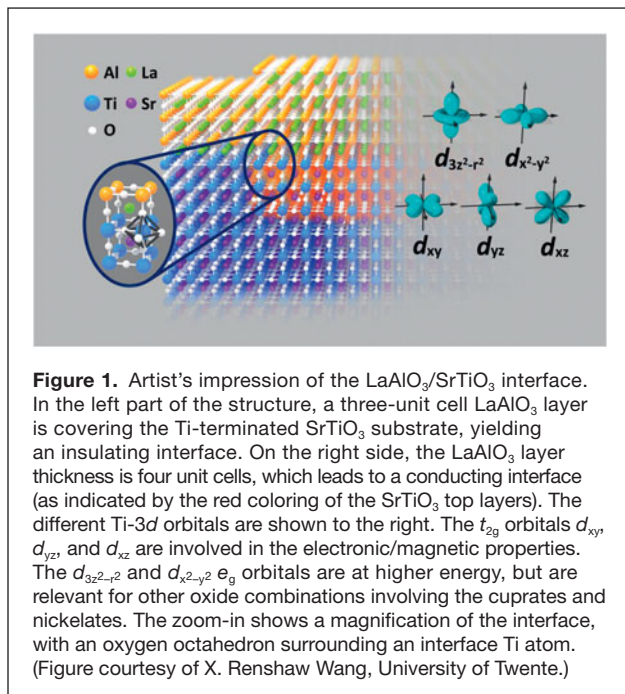
An additional mechanism for charge transfer can be at play in particular oxide materials combinations, for which a discontinuity in the internal polarization state of the material arises at the interface. This has been suggested as a key mechanism in the case of SrTiO₃/LaAlO₃.⁷ In these respective materials, the Sr²⁺O²⁻ and Ti⁴⁺(O²⁻)₂ layers are in first approximation charge neutral, whereas the adjacent La³⁺O²⁻ and Al³⁺(O²⁻)₂ layers present a polar system. When growing a thin film of LaAlO₃ on a SrTiO₃ substrate chemically prepared to have only one type of atomic layer at the surface, such as TiO₂ which selects the film to start with a LaO layer, an electrostatic potential build-up occurs inside the LaAlO₃. Before this potential build-up can reach unphysically high values with increasing LaAlO₃ layer thickness, denoted as the “polar catastrophe,” compensating mechanisms such as structural reconstructions or charge transfer from the LaAlO₃ surface to the SrTiO₃/LaAlO₃ interface will

set in. The latter is referred to as “electronic reconstruction,” a term first coined in the context of polar discontinuities/polar surfaces for the case of K₃C₆₀.⁸ For the SrTiO₃/LaAlO₃ system, the charge transfer is expected to occur when the LaAlO₃ thickness exceeds three unit cells; in that case, the internal potential build-up is large enough to allow electrons from the valence band of the LaAlO₃ to reach the conduction band of the SrTiO₃.⁹ An interface projected sheet carrier density of maximally 3 × 10¹⁴ charges/cm² is then expected, equivalent to one electron per two unit cells.⁷

There are strong arguments that such electronic reconstruction is indeed at work in the LaAlO₃/SrTiO₃ system. For example, the conductance only occurs when the LaAlO₃ film exceeds a critical thickness of four unit cells,¹⁰ which is exactly the theoretically expected value. Also, when diluting the LaAlO₃ with SrTiO₃, the potential build-up is expected to be reduced, and in that case, the threshold thickness has been found to scale inversely with the LaAlO₃ fraction.¹¹ Attempts to measure the internal potential by photoelectron spectroscopy, however, have given much smaller values than expected.^{12,13} Also, samples grown under very different conditions, for example, different oxygen pressures during film growth by pulsed laser deposition, give rise to very different carrier densities and transport characteristics, but still tend to show the 3–4 unit cell transition as a very robust feature. This may suggest that for this transition, other mechanisms are of importance as well, of which the pinning of interface charges by LaAlO₃ surface potential fluctuations is one possibility.¹⁴

Many experiments are currently reported that provide further pieces to the puzzle of the mechanisms for interface conductivity in LaAlO₃/SrTiO₃ and its relation to the atomic structure. A complete review of this is beyond the scope of this article—an interesting recent discussion is provided in Reference 15—but some noteworthy aspects to consider are that the atomic structure at the interface is not sharp and abrupt (as in **Figure 1**), but rather interdiffusion and structural reconstructions take place as well. Various interface-sensitive x-ray diffraction studies have revealed insights into this,^{16,17} which is of relevance when comparing results from density functional theory calculations on idealized interfaces with experimental observations. Further, a strong dependence of the transport properties on the La/Al ratio appears to exist in the LaAlO₃. With La-rich LaAlO₃, reduced or no conductance occurs at the interface.^{18,19} Finally, the interface conductivity is not limited to the case of LaAlO₃ on (001)-oriented SrTiO₃ but is also observed unexpectedly at LaAlO₃/SrTiO₃ (110)-interfaces.²⁰ An interesting feature of the latter system is that it is very anisotropic.

A special aspect of the oxide 2DEGs, which because of their electronic correlations and rich physics are also called two-dimensional electron liquids,²¹ is the steepness of the confining potential. The ionic nature of these crystals, with their possibility for ferro- and para-electric responses to electric fields, as well as the polarizability of the oxygen atoms gives rise to very strong electric field screening. Typically, the interface



2DEG in the oxides extends only a few unit cells or nanometers into the abutting crystals.²²

Advanced materials tailoring and orbital engineering

Typical electron mobility values of about 100–2000 cm²/Vs are found at low temperatures for common LaAlO₃/SrTiO₃ interfaces, depending on the details of the fabrication. At room temperature, electron-phonon interactions set the mobility to about 7 cm²/Vs.²³

Using optimized procedures for the growth and annealing of samples, involving lower than usual growth temperatures (650°C instead of 800°C), low temperature mobility values of up to 6600 cm²/Vs were obtained.²⁴ By optimizing the surface of LaAlO₃ using chemical and electronic techniques, mobilities exceeding 20,000 cm²/Vs were achieved, with concomitant reduction of the interface charge carrier density.²⁵ Recently, Huijben et al. showed that by incorporating a SrCuO₂ capping layer on top of LaAlO₃, the mobility could be greatly enhanced, up to 50,000 cm²/Vs at *T* = 4.2 K (see **Figure 2**). With this capping layer, a low and practically temperature-independent carrier density was found, which signals an efficient elimination of oxygen vacancies as charge donors and impurity potentials.²⁶

At these high mobilities, Shubnikov-de Haas quantum oscillations in the resistance at high magnetic fields are resolved,²⁷ but interestingly, so far no clear-cut quantum Hall effect has been reported for such interfaces. The reason for this is not yet clear, but it may be related to the multiband/multiorbital nature of the conductivity.

To further elucidate this multiband/multiorbital character,²⁸ we again consider bulk crystalline SrTiO₃. In this insulating

compound, the Ti⁴⁺ ions are in a 3*d*⁰ electron configuration, and when doped with electrons, the conduction band formed by its 3*d* orbitals is partially filled. The crystal field splits the 3*d* manifold into the lower lying *t*_{2g} levels (*d*_{xy}, *d*_{xz}, and *d*_{yz}) and the *e*_g levels (*d*_{x²-y² and *d*_{3z²-r²), which are about 3 eV further up in energy. These different orbital states are depicted in Figure 1. At the interface, additional level splitting occurs due to the symmetry breaking and to possible deformations and rotations of the oxygen octahedra surrounding the Ti. The *d*_{xy} orbitals, oriented parallel to the interface, form the lowest lying band and have the lowest effective mass.²⁹ The heavier, more anisotropic *d*_{xz} and *d*_{yz} states have their band edge only slightly (in the 10 meV range) above that of the *d*_{xy} band, and thus also become easily populated. The 2D-confinement furthermore leads to a discretization of the electron momenta in the direction perpendicular to the interface.}}

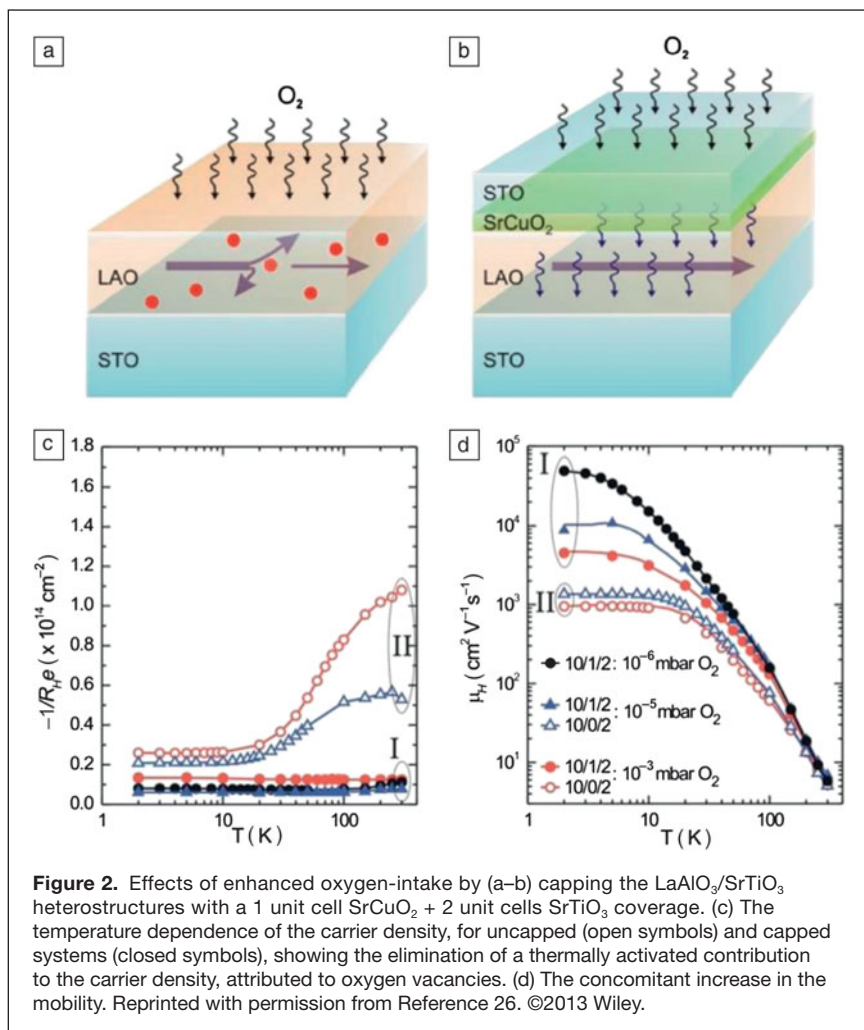
Finally, the carrier density as well as the *d*_{xy}/*d*_{yz}/*d*_{xz} level splitting vary as a function of depth from the interface, and with this the orbital character of the mobile electrons also varies.^{23,30} The result is a complex, carrier density-dependent multiband/multiorbital system, in which the different bands are expected to exhibit different properties in terms of mobility and magnetic character. For the samples in Figure 2, up to five different high mobility bands have been seen in Shubnikov-de Haas oscillations, with varying effective masses in the 0.9–3 *m*_e range.²⁷

While a complicating factor, the special orbital aspects of the complex oxide 2DEGs also allow for a new paradigm for materials tailoring: “interface orbital engineering.” This has been introduced on theoretical grounds by Hansmann et al., who suggested that the orbital state of high-*T*_c superconductors, with its characteristic partially filled 3*d*_{x²-y²}-like band, can be mimicked by the LaNiO₃-LaAlO₃ interface.³¹ In its bulk metallic state, LaNiO₃ has one electron per unit cell in the two-fold degenerate *e*_g-orbital states. When sandwiched between LaAlO₃, the *d*_{3z²-r² levels are shifted to higher energy, and the partially filled *d*_{x²-y² band is predicted to form the only conduction band, a situation similar to the cuprates. X-ray reflectivity studies indeed show a shift in orbital occupation,^{32,33} although smaller than predicted, and a Mott insulator state is observed by DC transport measurements in ultrathin LaNiO₃/LaAlO₃ superlattices.³⁴ So far, high-*T*_c superconductivity has not been observed, but the concept of orbital engineering at interfaces is still in its infancy and may hold surprises for the future.}}

It is also noteworthy that hole-doped conductivity has been realized in LaVO₃ when sandwiched between LaAlO₃.³⁵ Together with *n*-doped interfaces such as LaAlO₃/SrTiO₃ these two-dimensional hole gases (2DHG) provide an interesting basis for novel devices, such as *p*-*n* junctions and closely coupled 2DEG/2DHG systems. The latter are of interest for thermoelectrics and for the exploration of exciton Bose-Einstein condensation in correlated electron systems.³⁶

Superconductivity and magnetism

The rich phenomenology of the complex oxide interfaces becomes especially apparent from the fact that under certain



conditions, superconductivity³⁷ as well as magnetic effects³⁸ are observed. For LaAlO₃/SrTiO₃ interfaces, superconductivity occurs below about 300 mK, which is similar to values found in bulk electron-doped SrTiO₃. It has been verified that this superconducting phase has a true 2D character though, and electric field effect experiments, using voltages applied to a metallic gate-electrode on the backside of the substrate, have mapped out the dependence of the critical temperature on the carrier concentration.³⁹

An outstanding question is whether the observed superconductivity has additional special features, such as an unconventional order parameter symmetry or unusual finite momenta of the Cooper pairs,⁴⁰ the electron pairs that establish the superconducting state. Also, the multi-orbital character of the conductance at the interfaces is again of interest here: do the d_{xy} and the d_{xz} , d_{yz} electrons both contribute to the superconductivity? Are the Cooper pairs arising from predominant inter-band or intraband pairing,⁴¹ and is there a special role for the transition from single- to multiband occupation with increased carrier density?⁴²

Such questions become particularly tantalizing when considering the reported coexistence of superconductivity and

magnetism. Magnetic effects were initially observed in the form of strong magnetoresistance and temperature dependence of the resistance that is characteristic for Kondo scattering, in which a cloud of mobile electrons surrounds localized magnetic moments.³⁸ Later studies have shown the coexistence of different ferromagnetic, dia-, and paramagnetic phases.⁴³ These magnetic effects are found, in particular, for LaAlO₃/SrTiO₃ interfaces grown by pulsed laser deposition at comparatively high oxygen pressures ($>10^{-3}$ mbar), in which a different ground state appears at the superconducting interfaces, which are typically grown at around 10^{-4} mbar.

Also the thickness of the LaAlO₃ film plays a role; magnetic effects are more prominently observed for LaAlO₃ film thicknesses exceeding about 10 unit cells. In several subsequent studies, magnetic effects and superconductivity have been reported to occur on the same sample.^{44–46} While investigations are still ongoing to prove whether this coexistence is truly an intrinsic effect or whether this is due to sample inhomogeneity or phase separation effects, an interesting scenario emerges that superconductivity could be carried by one type of the carrier, for example the heavier d_{xz} and d_{yz} electrons,⁴⁷ whereas the magnetism is provided by the d_{xy} electrons, possibly mediated by interactions with localized magnetic moments.⁴⁸ Also other suggestions have been made, for example, that the magnetism arises in the d_{xz} and d_{yz} bands related

to their anisotropic nature.⁴⁹ Obviously, a complete understanding on the mechanisms for the interface superconductivity and magnetism, and their interplay, is still in development.

Nanostructures and transistors

The sharp phase transitions in the complex oxide 2DEGs, such as the insulator-to-metal transition between 3–4 unit cells in the case of LaAlO₃/SrTiO₃,¹⁰ provide a tantalizing basis for applications in future nanoscale electronic circuitry. Levy and co-workers from the University of Pittsburgh have provided an alluring demonstration for this in a series of experiments^{50–52} (see **Figure 3**). Starting with insulating samples of three unit cells of LaAlO₃ on SrTiO₃, they induced conductivity in the LaAlO₃/SrTiO₃ interface underneath the voltage-biased tip of a conducting atomic force microscopy, traversing over the LaAlO₃ surface. This way, (super-)conducting lines only a few nanometers wide were written and subsequently erased, and small devices such as side-gated field-effect transistors were realized. The understanding at this moment is that the tip voltage bias leads to selective removal of OH⁻ groups at the LaAlO₃ surface, occurring as a result of ionization of water at this surface, leaving behind excess positive charges

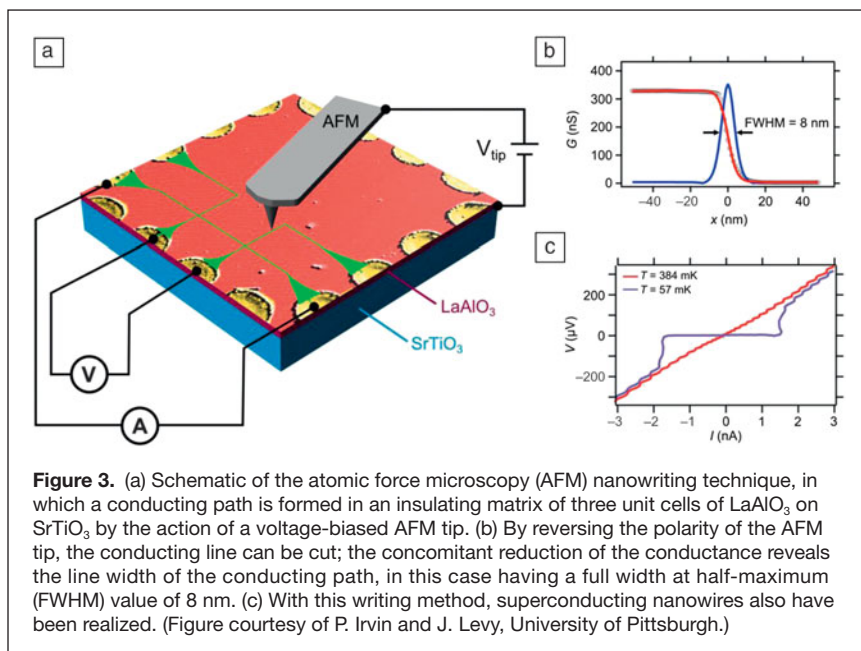


Figure 3. (a) Schematic of the atomic force microscopy (AFM) nanowriting technique, in which a conducting path is formed in an insulating matrix of three unit cells of LaAlO₃ on SrTiO₃ by the action of a voltage-biased AFM tip. (b) By reversing the polarity of the AFM tip, the conducting line can be cut; the concomitant reduction of the conductance reveals the line width of the conducting path, in this case having a full width at half-maximum (FWHM) value of 8 nm. (c) With this writing method, superconducting nanowires also have been realized. (Figure courtesy of P. Irvin and J. Levy, University of Pittsburgh.)

in the form of H⁺-ions.⁵³ These positive ions provide a local field-effect gating, enough to drive the insulator-to-metal transition. While this “water cycle” is not directly suited for widespread technology, these experiments beautifully demonstrate the possibility of realizing nanoscale electronic device structures and the sharpness of the potential profiles that can be created in these oxides.

The development of field-effect transistors with metallic gates is also advancing rapidly. Initial studies have been performed using gates at the backsides of the SrTiO₃ substrates, showing the possibility of switching superconductivity on and off.³⁹ More recently, metal oxide interface field-effect transistors (MOIFETs) have also been demonstrated in top-gated geometries,^{55–58} enabling local modulation of the (super-) conductivity with modest gate voltages (see **Figure 4**). A range of experiments is now within reach, reminiscent of the quantum/nano research done on GaAs/AlGaAs, such as single-electron manipulation, with the additional attraction of the rich physics of the oxides—displayed, for example, in its superconductivity—and the versatility in materials combinations.

Tunable spin-orbit coupling and topological insulators

Another special feature of the complex oxide 2DEGs is the large and tunable Rashba spin-orbit coupling, which is of great interest for spintronics. Rashba spin-orbit coupling arises when an electron experiences an electric field perpendicular to its direction of motion, which in the case of the oxide 2DEGs implies a field perpendicular to the interface.^{59,60} The structural inversion

symmetry breaking at the interface already provides a strong field and concomitant spin-orbit coupling, which can be further enhanced and manipulated by applying external gate voltages. One of the prospects arising from this is the realization of a functional spin field-effect transistor,⁶¹ in which spin-precession is controlled by an external gate. In this, the combination of top- and backgating may be particularly powerful.

Finally, oxide interfaces may also play a role in the rapidly developing field of topological insulators. Predictions for 3D-topological insulator behavior in iridium-based pyrochlores such as Y₂Ir₂O₇,^{62,63} and the bismuthate perovskite YBiO₃,⁶⁴ have been made, which are awaiting experimental confirmation. Particularly fruitful in this respect will be the large bandgap and bulk resistivity of the oxides. When these predictions will be confirmed experimentally, the complex oxides will prove themselves all the more as true gemstones of physics.

Conclusion

In addition to the many special bulk properties of complex oxides, their interfaces present exciting additional platforms for the creation of novel electronic and magnetic phases. With enhanced control at the materials science of these oxides, for example with respect to substrate preparation and thin-film growth, new materials combinations can now be realized with atomic precision. It is to be expected that further discoveries will arise from such interfaces, of which higher temperature superconductivity would be a most tantalizing prospect. Because of their versatility and stability, the fact that they can be fabricated using relatively conventional thin-film growth techniques, and the possibility to controllably create field-effect transistors and functional nanostructures, complex oxides and their interfaces present serious candidate materials systems for future nano-electronic device technologies.

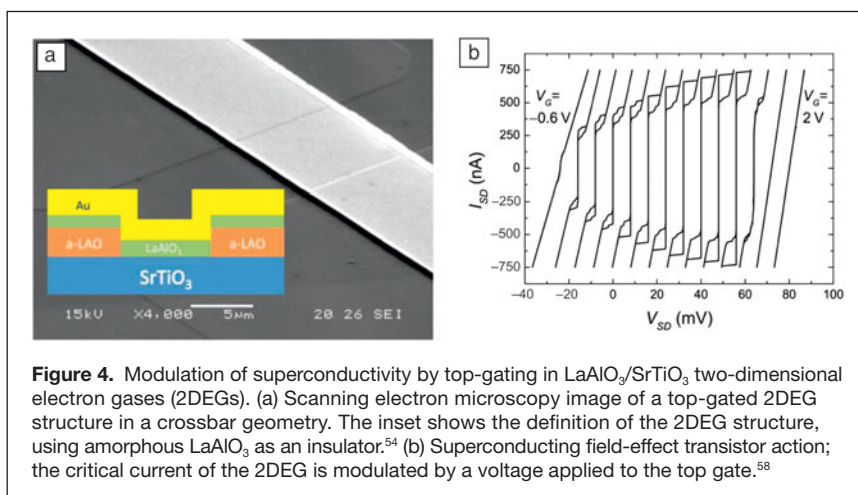


Figure 4. Modulation of superconductivity by top-gating in LaAlO₃/SrTiO₃ two-dimensional electron gases (2DEGs). (a) Scanning electron microscopy image of a top-gated 2DEG structure in a crossbar geometry. The inset shows the definition of the 2DEG structure, using amorphous LaAlO₃ as an insulator.⁵⁴ (b) Superconducting field-effect transistor action; the critical current of the 2DEG is modulated by a voltage applied to the top gate.⁵⁸

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References

- J.F. Schooley, W.R. Hosler, M.L. Cohen, *Phys. Rev. Lett.* **12**, 474 (1964).
- M.L. Cohen, in *Superconductivity, Volume 1*, R. Parks, Ed. (Marcel Dekker, New York, 1969).
- J.G. Bednorz, K.A. Müller, *Rev. Mod. Phys.* **60**, 585 (1988).
- A. Ohtomo, H.Y. Hwang, *Nature* **441**, 7089 (2006).
- Y. Kozuka, S. Teraoka, J. Falson, A. Oiwa, A. Tsukazaki, S. Tarucha, M. Kawasaki, *Phys. Rev. B* **87**, 205411 (2013).
- M. Imada, A. Fujimori, Y. Tokura, *Rev. Mod. Phys.* **70**, 1039 (1998).
- N. Nakagawa, H.Y. Hwang, D.A. Muller, *Nat. Mater.* **5**, 204 (2006).
- R. Hesper, L.H. Tjeng, A. Heeres, G.A. Sawatzky, *Phys. Rev. B* **62**, 16046 (2000).
- R. Pentcheva, W.E. Pickett, *Phys. Rev. Lett.* **102**, 107602 (2009).
- S. Thiel, G. Hammerl, A. Schmehl, C.W. Schneider, J. Mannhart, *Science* **313**, 1942 (2006).
- M.L. Reinle-Schmitt, C. Cancellieri, D. Li, D. Fontaine, M. Medarde, E. Pomjakushina, C.W. Schneider, S. Gariglio, P. Ghosez, J.-M. Triscone, P.R. Willmott, *Nat. Commun.* **3**, 932 (2012).
- M. Sing, G. Berner, K. Goss, A. Müller, A. Ruff, A. Wetscherek, S. Thiel, J. Mannhart, S.A. Pauli, C.W. Schneider, P.R. Willmott, M. Gorgoi, F. Schafers, R. Claessen, *Phys. Rev. Lett.* **102**, 176805 (2009).
- E. Slooten, Z.C. Zhong, H.J.A. Molegraaf, P.D. Eerkes, S. de Jong, F. Massee, E. van Heumen, M.K. Kruize, S. Wenderich, J.E. Kleibeuker, M. Gorgoi, H. Hilgenkamp, A. Brinkman, M. Huijben, G. Rijnders, D.H.A. Blank, G. Koster, P.J. Kelly, M.S. Golden, *Phys. Rev. B* **87**, 085128 (2013).
- N.C. Bristowe, P.B. Littlewood, E. Artacho, *Phys. Rev. B* **83**, 205405 (2011).
- Z.Q. Liu, C.J. Li, W.M. Lü, X.H. Huang, S.W. Zeng, X.P. Qi, L.S. Huang, A. Annadi, J.S. Chen, J.M.D. Coey, T. Venkatesan, Ariando, *Phys. Rev. X* **3**, 21010 (2013).
- P.R. Willmott, S.A. Pauli, R. Herger, C.M. Schlepütz, D. Martocchia, B.D. Patterson, B. Delley, R. Clarke, D. Kumah, C. Cionca, Y. Yacobi, *Phys. Rev. Lett.* **99**, 15502 (2007).
- V. Vonk, J. Huijben, D. Kukuruznyak, A. Stierle, H. Hilgenkamp, A. Brinkman, S. Harkema, *Phys. Rev. B* **85**, 045401 (2012).
- H.K. Sato, C. Bell, Y. Hikita, H.Y. Hwang, *Appl. Phys. Lett.* **102**, 251602 (2013).
- M.P. Warusawithana, C. Richter, J.A. Mundy, P. Roy, J. Ludwig, S. Paetel, T. Heeg, A.A. Pawlicki, L.F. Kourkoutis, M. Zheng, M. Lee, B. Mulcahy, W. Zander, Y. Zhu, J. Schubert, J.N. Eckstein, D.A. Muller, C. Stephen Hellberg, J. Mannhart, D.G. Schlom, *Nat. Commun.* **4**, 2351 (2013).
- A. Annadi, Q. Zhang, X. Renshaw Wang, N. Tuzla, K. Gopinadhan, W.M. Lu, A.R. Barman, Z.Q. Liu, A. Srinivastava, S. Saha, Y.L. Zhao, S.W. Zeng, S. Maekawa, H. Hilgenkamp, T. Venkatesan, Ariando, *Nat. Commun.* **4**, 1838 (2013).
- M. Breitschaft, V. Tinkl, N. Pavlenko, S. Paetel, C. Richter, J.R. Kirtley, Y.C. Liao, G. Hammerl, V. Eyert, T. Kopp, J. Mannhart, *Phys. Rev. B* **81**, 153414 (2010).
- M. Basletic, J.L. Maurice, C. Carretero, G. Herranz, O. Copie, M. Bibes, E. Jacquest, K. Bouzehouane, S. Fusil, A. Barthelemy, *Nat. Mater.* **7**, 621 (2008).
- J. Mannhart, D.G. Schlom, *Science* **327**, 1607 (2010).
- A.D. Caviglia, S. Gariglio, C. Cancellieri, B. Sacépé, A. Fête, N. Reyren, M. Gabay, A.F. Morpurgo, J.-M. Triscone, *Phys. Rev. Lett.* **105**, 236802 (2010).
- Y. Xie, C. Bell, Y. Hikita, S. Harashima, H.Y. Hwang, *Adv. Mater.* **213** (2013), doi:10.1002/adma.201301798.
- M. Huijben, G. Koster, M.K. Kruize, S. Wenderich, J. Verbeeck, S. Bals, E. Slooten, B. Shi, H.J.A. Molegraaf, J.E. Kleibeuker, S. van Aert, J.B. Goedkoop, A. Brinkman, D.H.A. Blank, M.S. Golden, G. van Tendeloo, H. Hilgenkamp, G. Rijnders, *Adv. Funct. Mater.* (2013), doi:10.1002/adfm.201203355.
- A. McCollam, S. Wenderich, M.K. Kruize, V.K. Guduru, H.J.A. Molegraaf, M. Huijben, G. Koster, D.H.A. Blank, G. Rijnders, A. Brinkman, H. Hilgenkamp, U. Zeitler, J.C. Maan, *Condens. Matter* (2012), available at <http://arxiv.org/abs/1207.7003v1>.
- Y. Tokura, N. Nagaosa, *Science* **288**, 462 (2000).
- M. Salluzzo, J.C. Cezar, N.B. Brookes, V. Bisogni, G.M. De Luca, C. Richter, S. Thiel, J. Mannhart, M. Huijben, A. Brinkman, G. Rijnders, G. Ghiringhelli, *Phys. Rev. Lett.* **102**, 166804 (2009).
- P. Delugas, A. Filipetti, V. Fiorentini, I. Bilc, D. Fontaine, P. Ghosez, *Phys. Rev. Lett.* **106**, 166807 (2011).
- P. Hansmann, X.P. Yang, A. Toschi, G. Khaliulin, O.K. Andersen, K. Held, *Phys. Rev. Lett.* **103**, 016401 (2009).
- E. Benckiser, M.W. Haverkort, S. Brück, E. Goering, S. Macke, A. Frañó, X. Yang, O.K. Andersen, G. Cristiani, H.-U. Habermeier, A.V. Boris, I. Zegkinoglou, P. Wochner, H.-J. Kim, V. Hinkov, B. Keimer, *Nat. Mater.* **10**, 189 (2011).
- A.V. Boris, Y. Matiks, E. Benckiser, A. Frañó, P. Popovich, V. Hinkov, P. Wochner, M. Castro-Colin, E. Detemple, V.K. Malik, C. Bernhard, T. Prokscha, A. Suter, Z. Salzman, E. Morenzoni, G. Cristiani, H.-U. Habermeier, B. Keimer, *Science* **332**, 6032 (2011).
- J.A. Liu, S. Okamoto, M. van Veenendaal, M. Kareev, B. Gray, P. Ryan, J.W. Freeland, J. Chakhalian, *Phys. Rev. B* **83**, 161102 (2011).
- T. Higuchi, Y. Hotta, T. Susaki, A. Fujimori, H.Y. Hwang, *Phys. Rev. B* **79**, 075415 (2009).
- L. Rademaker, J. van den Brink, H. Hilgenkamp, J. Zaanen, *Phys. Rev. B* **88**, 121101 (2013).
- N. Reyren, S. Thiel, A.D. Caviglia, L. Fitting Kourkoutis, G. Hammerl, C. Richter, C.W. Schneider, T. Kopp, A.-S. Rüetschi, D. Jaccard, M. Gabay, D.A. Muller, J.-M. Triscone, J. Mannhart, *Science* **317**, 1196 (2007).
- A. Brinkman, M. Huijben, M. van Zalk, J. Huijben, U. Zeitler, J.C. Maan, W.G. van der Wiel, G. Rijnders, D.H.A. Blank, H. Hilgenkamp, *Nat. Mater.* **6**, 493 (2007).
- A.D. Caviglia, S. Gariglio, N. Reyren, D. Jaccard, T. Schneider, M. Gabay, S. Thiel, G. Hammerl, J. Mannhart, J.-M. Triscone, *Nature* **456**, 624 (2008).
- K. Michaeli, A.C. Potter, P.A. Lee, *Phys. Rev. Lett.* **108**, 117003 (2012).
- R.M. Fernandes, J.T. Haraldson, P. Wölfle, A.V. Balatsky, *Phys. Rev. B* **87**, 014510 (2013).
- A. Joshua, S. Pecker, J. Ruhman, E. Altman, S. Ilani, *Nat. Commun.* **3**, 1129 (2012).
- Ariando, X. Wang, G. Baskaran, Z.Q. Liu, J. Huijben, J.B. Yi, A. Annadi, A.R. Barman, A. Rusydi, S. Dhar, Y.P. Feng, J. Ding, H. Hilgenkamp, T. Venkatesan, *Nat. Commun.* **2**, 188 (2011).
- D.A. Dikin, M. Mehta, C.W. Bark, C.M. Folkman, C.B. Eom, V. Chandrasekhar, *Phys. Rev. Lett.* **107**, 056802 (2011).
- J.A. Bert, B. Kalisky, C. Bell, M. Kim, Y. Hikita, H.Y. Hwang, K.A. Moler, *Nat. Phys.* **7**, 767 (2011).
- L. Li, C. Richter, J. Mannhart, R.C. Ashoori, *Nat. Phys.* **7**, 762 (2011).
- Y. Nakamura, Y. Yanase, *J. Phys. Soc. Jpn.* **82**, 083705 (2013).
- M. Salluzzo, S. Gariglio, D. Stornaiuolo, V. Sessi, S. Rusponi, C. Piamonteze, G.M. De Luca, M. Minola, D. Marré, A. Gadaleta, H. Brune, F. Nolting, N.B. Brookes, G. Ghiringhelli, *Phys. Rev. Lett.* **111**, 087204 (2013).
- G. Chen, L. Balents, *Phys. Rev. Lett.* **110**, 206401 (2013).
- C. Cen, S. Thiel, G. Hammerl, C.W. Schneider, K.E. Andersen, C.S. Hellberg, J. Mannhart, J. Levy, *Nat. Mater.* **7**, 298 (2008).
- C. Cen, S. Thiel, J. Mannhart, J. Levy, *Science* **323**, 1026 (2009).
- Y. Ma, M. Huang, S. Ryu, C.W. Bark, C.-B. Eom, P. Irvin, J. Levy, *Nano Lett.* **13**, 2884 (2013).
- F. Bi, D.F. Bogorin, C. Cen, C.W. Bark, J.-W. Park, C.-B. Eom, J. Levy, *Appl. Phys. Lett.* **97**, 173110 (2010).
- C.W. Schneider, S. Thiel, G. Hammerl, C. Richter, J. Mannhart, *Appl. Phys. Lett.* **89**, 122101 (2006).
- B. Förg, C. Richter, J. Mannhart, *Appl. Phys. Lett.* **100**, 053506 (2012).
- M. Boucherit, O.F. Shoron, T.A. Cain, C.A. Jackson, S. Stemmer, S. Rajan, *Appl. Phys. Lett.* **102**, 242909 (2013).
- M. Hosoda, Y. Hikita, H.Y. Hwang, C. Bell, *Appl. Phys. Lett.* **103**, 103507 (2013).
- P.D. Eerkes, W.G. van der Wiel, H. Hilgenkamp, *Appl. Phys. Lett.* (2013), in press.
- A.D. Caviglia, M. Gabay, S. Gariglio, N. Reyren, C. Cancellieri, J.-M. Triscone, *Phys. Rev. Lett.* **104**, 126803 (2010).
- M. Ben Shalom, M. Sachs, D. Rakhmievitch, A. Palevski, Y. Dagan, *Phys. Rev. Lett.* **104**, 126802 (2010).
- S. Datta, B. Das, *Appl. Phys. Lett.* **56**, 665 (1990).
- D. Pesin, L. Balents, *Nat. Phys.* **6**, 376 (2010).
- B.-J. Yang, Y.B. Kim, *Phys. Rev. B* **82**, 085111 (2010).
- H. Jin, S.H. Rhim, J. Im, A. Freeman, *Sci. Rep.* **3**, 1651 (2013). □

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