

Effect of soil acidification on regrowth of orchardgrass (*Dactylis glomerata*) under application of grazing cattle dung, cattle manure compost, and chemical fertilizer

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| 著者 | Hidetoshi Kakihara, Shin-ichiro Ogura |
| journal or publication title | Grassland Science |
| year | 2022-03-22 |
| URL | http://hdl.handle.net/10097/00137166 |

doi: 10.1111/grs.12361

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5 Hidetoshi Kakihara and Shin-ichiro Ogura

6

7 Graduate school of Agricultural Science, Tohoku University, Osaki, Miyagi, Japan

8

9

10 Correspondence

11 Hidetoshi Kakihara, Graduate School of Agricultural Science, Tohoku University, Osaki,

12 Miyagi, Japan 989-6711, Japan.

13 Email: kakihara@tohoku.ac.jp

14 Tel & fax: +81 (0)229-84-7382

15

16 Abstract

17

18 Orchardgrass (*Dactylis glomerata*) persists poorly in acidic soils. Not many studies
19 have looked into the effects of fertilizers in improving orchardgrass persistence within
20 acidic soils. We conducted experiments on 64 individual potted orchardgrass plants,
21 which were defoliated to 5 cm and assigned to one of the following four treatment
22 groups: unfertilized control (CNT), chemical fertilizer (CHE), grazing cattle dung
23 (DNG), and cattle manure compost (CMP). Half the pots in each treatment received
24 aluminum sulfate solution to induce further soil acidification (Al-add), while the others
25 received water (no-Al). On days 20 and 47, after defoliation, soil properties and dry
26 weights of aboveground biomass (AGB) (separated into leaves and stubble) and roots of
27 four pots in each treatment group were measured. Al-add induced soil acidification in all
28 fertilizers across the experiment ($p < 0.05$). On day 20, AGB and leaves in CHE was
29 increased by acidification ($p < 0.05$), which was not observed in other fertilizer
30 treatments ($p > 0.1$). Stubble growth increased following acidification in all fertilizer
31 treatments ($p < 0.05$). Acidification did not increase AGB on day 47; no effect was seen
32 on root growth at either day 20 or 47 ($p > 0.1$). On day 20, soil concentrations of
33 inorganic nitrogen (IN), nitrate-nitrogen, and ammonium-nitrogen ($\text{NH}_4\text{-N}$) were
34 significantly elevated in Al-add pots ($p < 0.05$). The increment was greater in CHE and
35 CMP than in CNT and DNG on day 20, with a similar trend being observed for IN and
36 $\text{NH}_4\text{-N}$ concentrations at day 47. IN and $\text{NH}_4\text{-N}$ concentrations in DNG with no-Al
37 increased over the regrowth period. These results indicate that orchardgrass regrowth in
38 acidic soils can be improved by fertilizer addition, depending on fertilizer type. The
39 increased concentration of soil IN, induced by soil acidification, is likely to be one of

40 the factors encouraging growth. This increase of regrowth may favor the persistence of
41 orchardgrass in strongly acidic soils.

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43

44 **KEYWORDS**

45 acidic soil, forage regrowth, inorganic nitrogen, pasture management, soil nutrient

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49 Orchardgrass (*Dactylis glomerata* L.) is a widely used temperate forage grass. In
50 pasture-based farming, the persistence of forage grass is one of the most important
51 factors for sustainable production (Tozer et al., 2011). However, this grass species often
52 declines within a couple of years of planting (Scott et al., 2000), with particularly low
53 persistence under acidic soil conditions (Hojito et al., 1987; Poozesh et al., 2010).

54 Acidic soil affects plant growth by altering the availability of soil nutrients and
55 releasing aluminum ions harmful to plants. It induces the accumulation of inorganic
56 nitrogen (Aciego Pietri & Brookes, 2008; Kemmitt et al., 2005) and limits the
57 availability of phosphorus and exchangeable cations to plants (Follett & Wilkinson,
58 1985; Hojito et al., 1987; Naidu et al., 1990). The characteristics of soil nutrients in
59 pastures also differ under different pasture management regimes, such as hay cutting
60 and livestock grazing (Bryan et al., 2019; Franzluebbbers & Stuedemann, 2010), which
61 can have a compounded impact on the effect of acidic soil on pasture plants' growth.

62 We have previously shown that the dominance response of orchardgrass to soil
63 acidity was different between hay cutting and cattle grazing. Orchardgrass sown in hay
64 meadows survived better than other plant species and became the dominant species
65 more frequently under strongly acidic soil conditions compared to mildly acidic soil. We
66 did not observe any significant effect of soil acidity under cattle grazing conditions
67 (Kakihara & Ogura, 2021). This may be due to the difference in pasture management
68 regimes, such as fertilizer application. In general, chemical fertilizer and manure
69 compost is used in hay meadows, whereas pastures utilize dung and urine which are
70 directly deposited onto the field by animals. Chemical fertilizers, compost, and cattle

71 manure affect the properties of pasture soils, which in turn changes the growth and
72 species composition of plants in the pasture, even though these fertilizers may contain
73 the same amount of nutrients (Min et al., 2003; Mauchamp et al., 2016).

74 The effect of fertilizer type on the regrowth of orchardgrass, when grown in hay
75 meadows, needs to be studied to determine its mechanism of persistence in strongly
76 acidic soil. Therefore, in the present study, we investigated the effect of fertilizer type
77 on the concentration of soil nutrients and the regrowth of orchardgrass under different
78 soil acidification levels by performing a pot experiment.

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80

81 2 | MATERIALS AND METHODS

82

83 2.1 | Study site and pot experiment

84 The study was conducted at the Kawatabi Field Science Center (FSC), Graduate School
85 of Agricultural Science, Tohoku University (Osaki, Miyagi, Japan; 38.75°N, 140.76°E,
86 200–250 m above sea level). The mean annual air temperature at the study site in 2019
87 was 11.1°C, with a maximum temperature of 35.8°C in August and a minimum
88 temperature of –7.8°C in January (Japan Meteorological Agency, 2020).

89 To obtain individual orchardgrass plants, seeds (cv. Akimidori II) were sown in a
90 rectangular plastic planter on November 2, 2018, and 64 seedlings were transplanted
91 into 64 pots (*i.e.*, one plant per plastic pot with \varnothing 12.5 cm and 15 cm depth) filled with
92 sieved soil (<6 mm) on June 10, 2019. The potting soil was obtained from a hay
93 meadow in the FSC. The soil type was classified as Haplic non-allophanic Andosol
94 according to the Soil Classification System of Japan (The fifth committee for soil

95 classification and nomenclature, 2017), Pachic Melanudand according to Soil
96 Taxonomy (Soil Survey Staff, 2014), or Aluandic Andosol in The World Reference Base
97 classification (IUSS Working Group WRB, 2015).

98 The chemical composition of the potting soil is described in Table 1. The
99 orchardgrass plants were grown outdoors for three months, then defoliated at the height
100 of 5 cm on September 12, 2019, to impose the same cutting condition on all individuals.
101 After defoliation, the pots were divided into four fertilizer treatments groups: 1)
102 unfertilized control (CNT), 2) chemical fertilizer of N, P₂O₅, K₂O at 34, 28, and 28
103 kg/ha as urea, double superphosphate, and potassium chloride, respectively (CHE), 3)
104 10 g of grazing cattle dung (DNG), and 4) 2.39 g of cattle manure compost (CMP)
105 (Table 1). The grazing cattle dung was collected from pastures dominated by
106 orchardgrass, tall fescue (*Festuca arundinacea*), redtop (*Agrostis alba*), and sweet
107 vernalgrass (*Anthoxanthum odoratum*), grazed by Japanese Black cows. For one month,
108 the cattle manure compost was produced by mixing cattle manure and sawdust in a
109 machine, 1–2 times a week. The dung and compost were kept refrigerated at 8°C for
110 four days until application. The amount applied was determined based on the nutrient
111 content in 10 g of cattle dung, with total nitrogen (TN) for compost and TN, phosphorus,
112 and potassium for chemical fertilizer. The amount of applied cattle dung was equivalent
113 to that of 150 days of grazing by 3.3 cattle/ha, which is the conventional grazing
114 management in the FCS. To half of the pots in each fertilizer treatment (8 pots per
115 treatment), 1 g of aluminum sulfate 14–18 hydrate (Al₂(SO₄)₃·14–18H₂O) was added
116 per pot as a water solution to induce further soil acidification (Al-add), whereas the
117 other half received water only (no-Al). The fertilizers and aluminum sulfate were
118 top-dressed immediately after the defoliation. The pots were placed outdoor and

119 irrigated every 2–3 days to maintain soil moisture throughout the experiment. Excess
120 water was drained out of the pots.

121

122 2.2 | Measurement

123 The measurements and samplings were carried out 20 and 47 days after the defoliation.

124 The number of tillers was recorded, then aboveground biomass (AGB) was harvested at

125 0 cm height from the soil. For each pot, the AGB was divided into leaf samples (≥ 5 cm

126 from the soil) and stubble samples (0–5 cm from the soil). Both the leaf and stubble

127 samples were force air-dried for 48 h at 70°C, and then their dry weights were obtained.

128 Roots were collected by gentle washing in water, then air-dried and weighed using the

129 same protocol as for AGB.

130 Two types of soil samples were collected from each pot: soil at a depth of 0–5 cm

131 (100 g) was used to measure soil pH and exchangeable acidity (y_1), and that at a depth

132 of 0–15 cm (500 g) was used to analyze the concentration of soil nutrients. These soil

133 samples were dried at room temperature, sieved (using a 2 mm mesh) to remove gravel,

134 then stored for the subsequent analysis. Soil pH was measured using 15 g of dried soil

135 by dipping the glass electrode of a pH meter into a 1:2.5 soil to water suspension. The

136 y_1 value was measured using 10 g of dried soil by titrating a 1 M potassium chloride

137 (KCl) extract (soil:solution = 1:2.5) with 0.1 M sodium hydroxide (NaOH) to the

138 endpoint of phenolphthalein (Mori & Shimada, 1970). The value of y_1 is closely related

139 to the amount of KCl extractable Al (Saigusa et al., 1980), so y_1 was used as an

140 indicator of exchangeable Al. TN and total carbon (TC) were measured by dry

141 combustion. Nitrate-nitrogen ($\text{NO}_3\text{-N}$) and ammonium-nitrogen ($\text{NH}_4\text{-N}$) were

142 evaluated by the hydrazine sulfate reduction and indo-phenol methods, respectively.

143 Inorganic nitrogen (IN) was estimated as total amount of NO₃-N and NH₄-N. Available
144 phosphorus (P₂O₅) was determined by the Bray-2 method. Potassium oxide (K₂O),
145 calcium oxide (CaO), and magnesium oxide (MgO) were quantified via atomic
146 absorption spectrometry after Schollenberger extraction.

147

148 2.3 | Data analysis

149 Before performing data analysis, one pot data in CMP with no-Al was excluded as it
150 showed extremely sparse roots due to a large number of beetle larvae in it. No larvae or
151 traces were observed in all the other pots.

152 Dry matter yield per tiller was considered to be the amount of regrowth of
153 orchardgrass as the number of tillers per pot differed among the pots (4–15 tillers per
154 pot), which might influence the total dry matter yield per pot.

155 Two-way analysis of variance (ANOVA) for each sampling day (days 20 and 47)
156 was performed to examine the effect of fertilizer and acidification treatment on the soil
157 acidity, dry matter weights of the AGB, leaves, stubble, and roots of a plant. If
158 significant interactions were detected, post hoc analysis for each treatment was
159 performed by pairwise multiple comparisons with Holm's method. The soil pH, y₁, and
160 weight of each part of the plant were used as response variables, and the fertilizer and
161 acidification treatment were used as fixed effects. Thus, the model is given by:

$$162 \quad Y = \beta_0 + \beta_1 \text{ Fertilizer} + \beta_2 \text{ Acidification} + \beta_3 \text{ Interaction}$$

163 where Y is either soil pH, y₁, or weights of each part of the plant. In the equation of the
164 models, 'Fertilizer', 'Acidification', and 'Interaction' represent the fertilizer treatment,
165 acidification treatment, and the interaction between 'Fertilizer' and 'Acidification',
166 respectively. β_0 is the intercept, β_1 , β_2 , and β_3 represent coefficients of the exploratory

167 variables and their interaction.

168 All analyses were performed at a significance level of 0.05 using the ‘lm’, ‘anova’,
169 and ‘interaction’ functions of the ‘base’ package, ‘emmeans’ functions of the ‘emmeans’
170 package (version 1.4.5), and the ‘glht’ function of the ‘multcomp’ package (version
171 1.4.12) in R version 3.6.1 (R Core Team, 2019).

172

173 3 | RESULTS

174

175 3.1 | Soil pH and exchangeable acidity

176 Soil acidification was induced significantly by the addition of aluminum sulfate in all
177 fertilizer treatments ($p < 0.05$, Table 2). The pH values were lower in Al-add
178 (5.07–5.16) than in no-Al (5.61–5.63) on day 20. A similar trend was observed at day 47,
179 with the pH of Al-add pots being lower (5.27–5.40) than that of no-Al pots (5.65–5.76).
180 The y_1 values were significantly higher in Al-add (9.02–11.34) than no-Al (5.98–8.69)
181 pots across the experiment.

182

183 3.2 | Regrowth of orchardgrass

184 The effect of fertilizer and acidification on the regrowth of orchardgrass varied across
185 regrowth days (Table 3). On day 20, the dry weight of AGB and leaves increased by
186 acidification in the CHE group ($p < 0.05$) but not in the other fertilizer treatments ($p >$
187 0.1, Table 3 and Figure 1). The dry weight of stubble increased by acidification in all
188 fertilizer types ($p < 0.05$). On day 47, the acidification did not increase the dry weight of
189 AGB regrowth ($p > 0.1$). Root regrowth was not affected by the acidification treatment

190 on both sampling dates.

191

192 3.3 | Concentration of soil nutrients

193 Both fertilizer and acidification treatment affected the mineral concentration of the soil
194 (Table 4). On day 20, IN, NO₃-N, and NH₄-N concentrations were significantly
195 increased by adding aluminum sulfate in all fertilizer types ($p < 0.05$, Table 4 and
196 Figure 2). The increment was greater in CHE and CMP than CNT and DNG on day 20.
197 A similar trend was observed on day 47, with IN and NH₄-N concentration in CHE and
198 CMP being higher in Al-add compared with no-Al (Figure 2). In addition, IN and
199 NH₄-N concentrations in DNG with no-Al increased over the regrowth period.

200 The concentration of MgO was higher in DNG and CMP than other fertilizer
201 treatments and was increased by the addition of ammonium sulfate in all the fertilizer
202 treatments on day 20. However, on day 47, the addition of ammonium sulfate decreased
203 MgO concentration in all the fertilizer treatments ($p < 0.05$, Table 4). Concentration of
204 CaO in Al-add treatment significantly rose on day 20 ($p < 0.05$) and tended to increase
205 on day 47 ($p = 0.07$). Acidification treatment did not significantly influence P₂O₅ and
206 K₂O contents ($p > 0.1$).

207

208 4 | DISCUSSION

209 Orchardgrass persists poorly in acidic soils, and little research has investigated the
210 effect of fertilization on the regrowth as well as the persistence of this species under
211 acidic soil conditions. In the present study, we investigated the effect of fertilizer type
212 on the concentration of soil nutrients and the regrowth of orchardgrass under different
213 soil acidification levels.

214 Top-dressed application of aluminum sulfate to potting soil induced further soil
215 acidification. This acidification increased the dry weight of AGB, which consisted of
216 leaves and stubble, in the CHE treatment on day 20 (Table 3 and Figure 1), indicating
217 that acidic soil encourages regrowth of orchardgrass with chemical fertilization at least
218 in the early stages after fertilization. This result suggests that chemical fertilizers, the
219 major type of fertilizer applied in hay meadows, increase the survivability of
220 orchardgrass in strongly acidic soil under cutting conditions. The increase of dry weight
221 of stubble and leaves by soil acidification is notable from the perspective of the
222 persistence of orchardgrass. Stubble is an important part of the orchardgrass regrowth
223 because carbohydrate reserves, primarily stored in the lower regions of stems, support
224 initial regrowth after defoliation (White, 1973).

225 The encouraging orchardgrass regrowth observed in the CHE with acidification
226 treatment is probably caused by high IN in the acidified potting soil. As shown in Table
227 4 and Figure 2, the soil content of IN, NO₃-N, and NH₄-N significantly increased on day
228 20 following soil acidification. Fertilized nitrogen is generally lost through
229 denitrification, volatilization of ammonia, and leaching of NO₃-N from grassland soil;
230 however, all of these forms of losses are decreased in acidic soil conditions (Aciego
231 Pietri & Brookes, 2008; Müller et al., 1980). Water-soluble nutrients such as NO₃-N are
232 leached out by excess water drained out of the potting soil. On the contrary, NH₄-N,
233 which was not subject to leaching, is accumulated by nitrification inhibition in acidic
234 soil (Aciego Pietri & Brookes, 2008; Kemmitt et al., 2005). Furthermore, orchardgrass
235 absorbs and utilizes ammonium ions as well as nitrate ions (MacLeod & Carson, 1965).
236 Therefore, our result suggests that soil acidification provides more available nitrogen for
237 orchardgrass. A good supply of nitrogen is an important factor affecting orchardgrass

238 leaf growth and its persistence (Van Santen & Sleper, 1996). Orchardgrass is a
239 responsive species to nitrogen fertilizer and becomes very competitive to other species
240 when nutrients are available (Jung & Baker, 1985). For this reason, the increase of soil
241 IN may increase the survivability of orchardgrass under strongly acidic soil conditions.

242 Despite significantly higher soil IN (Figure 2), AGB and leaves of orchardgrass did
243 not increase in the CNT, DNG, and CMP treatments (Figure 1). As far as measured in
244 the present study, no possible factor other than IN was found to be associated with a
245 difference in the growth of orchardgrass between acidification treatments. These results
246 suggest that the increase of soil IN in these treatments until day 20 is insufficient to
247 promote orchardgrass regrowth. Saarijarvi & Virkajarvi (2009) showed that chemical
248 fertilizer increased soil IN substantially within a couple of days, whereas soil IN
249 underneath cattle dung pats was highest on days 21 and 49. Application of organic
250 manure, such as cattle manure compost, usually results in temporal immobilization of
251 IN in the soil (Whitehead, 1995). Our result that IN and $\text{NH}_4\text{-N}$ concentration in the
252 DNG with no-Al increased over the regrowth period also suggests that a high
253 concentration of IN is not maintained during the regrowth period.

254 The Al-add treatment, regardless of fertilizer type, temporarily increased CaO and
255 MgO on day 20 (Figure 3), which was inconsistent with previous studies that
256 demonstrated decreases in these cations with soil acidification (Bojórquez-Quintal et al.,
257 2017; Martins et al., 2014). In this study, top-dressed aluminum sulfate induced an
258 increase of $\text{NH}_4\text{-N}$. According to Di & Cameron (2004), the loss of cations from the soil
259 is mainly due to $\text{NO}_3\text{-N}$ leaching and is reduced by nitrification inhibitors. Therefore,
260 nitrification inhibition by soil acidification may have temporarily suppressed cation
261 leaching from the soil during the experimental period.

262 As previous studies show that application of cattle dung ameliorates acid soil
263 (During et al., 1973; During & Weeda, 1973), it was anticipated that reduction of soil
264 acidification by grazing cattle dung would be an effective factor in varying orchardgrass
265 regrowth. However, soil acidification was observed regardless of fertilizer type in our
266 study, indicating that soil amendment effects of grazing cattle dung are very restricted
267 underneath or very close to thick dung pats (Yoshitake et al., 2014), and does not
268 explain why regrowth of orchardgrass is altered under different soil acidification levels.

269

270

271 5 | CONCLUSIONS

272 The regrowth of orchardgrass was increased by soil acidification with the application of
273 chemical fertilizer, possibly due to the high concentration of inorganic nitrogen in the
274 acidified soil. Although soil acidification also increased the concentration of inorganic
275 nitrogen in the soil that applied cattle manure compost and grazing cattle dung, these
276 fertilizers did not substantially increase the regrowth of orchardgrass under strongly
277 acidic soil conditions. The different regrowth rates of orchardgrass depending on the
278 type of fertilizer applied suggests that the fertilizer application regime is one of the
279 factors contributing to the persistence of orchardgrass in strongly acidic soil under
280 cutting conditions.

281

282 ACKNOWLEDGEMENTS

283 This research was supported by the Project of Integrated Compost Science.

284

285

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TABLE 1 Chemical composition of potting soil, chemical fertilizer (CHE), grazing cattle dung (DNG), and cattle manure compost (CMP) at the beginning of the experiment

| Component | Potting soil (per kg dry soil) | CHE (per pot) | DNG (per pot) | CMP (per pot) |
|------------------------------------|-----------------------------------|------------------|------------------|------------------|
| Moisture (%) | - | - | 87.5 | 46.5 |
| TC (mg) | - | - | 595 | 486 |
| TN (mg) | 5200 | 42.0 | 33.5 | 31.2 |
| NO ₃ -N (mg) | 6.1 | - | 0.01 | 2.01 |
| NH ₄ -N (mg) | 5.9 | - | 0.09 | 0.18 |
| P ₂ O ₅ (mg) | 682 | 34.0 | 33.3 | 29.1 |
| K ₂ O (mg) | 383 | 34.0 | 14.4 | 49.6 |
| CaO (mg) | 1080 | - | 19.9 | 23.7 |
| MgO (mg) | 151 | - | 19.2 | 15.8 |
| C/N ratio | - | - | 17.77 | 15.57 |
| pH | 5.20 | 3.77 | 8.04 | 7.35 |

TABLE 2 Soil pH and exchangeable acidity (y_1) in each treatment on Day 20 and Day 47

| Measurement item | CNT† | | CHE | | DNG | | CMP | | RMSE | Effect of treatment | | |
|------------------|--------|--------|-------|--------|-------|--------|-------|--------|------|---------------------|---------------|-------------|
| | no-Al‡ | Al-add | no-Al | Al-add | no-Al | Al-add | no-Al | Al-add | | Fertilizer | Acidification | Interaction |
| Day 20 | | | | | | | | | | | | |
| pH | 5.61 | 5.16 | 5.61 | 5.07 | 5.63 | 5.15 | 5.63 | 5.09 | 0.09 | NS | p < 0.05 | NS |
| y_1 | 7.69 | 9.55 | 8.69 | 11.34 | 8.07 | 9.42 | 7.56 | 9.89 | 1.14 | NS | p < 0.05 | NS |
| Day 47 | | | | | | | | | | | | |
| pH | 5.76 | 5.31 | 5.65 | 5.27 | 5.75 | 5.40 | 5.73 | 5.36 | 0.11 | NS | p < 0.05 | NS |
| y_1 | 6.32 | 9.66 | 7.49 | 10.63 | 5.98 | 9.16 | 6.66 | 9.02 | 1.16 | NS | p < 0.05 | NS |

Results are presented as estimated marginal means with root mean square error (RMSE).

† CNT, unfertilized control; CHE, chemical fertilizer; DNG, grazing cattle dung; CMP, cattle manure compost.

‡ no-Al, only water was added; Al-add, aluminum sulfate hydrate was added as a water solution to induce further soil acidification.

TABLE 3 Growth of orchardgrass in each treatment on Day 20 and Day 47

| Growth parameters | CNT† | | CHE | | DNG | | CMP | | RMSE | Effect of treatment | | |
|-------------------|--------|--------|-------|--------|-------|--------|-------|--------|------|---------------------|---------------|-------------|
| | no-Al‡ | Al-add | no-Al | Al-add | no-Al | Al-add | no-Al | Al-add | | Fertilization | Acidification | Interaction |
| Day 20 | | | | | | | | | | | | |
| AGB § | 0.09 | 0.13 | 0.11 | 0.28 | 0.11 | 0.11 | 0.09 | 0.15 | 0.05 | p < 0.05 | p < 0.05 | p < 0.05 |
| Leaves | 0.02 | 0.02 | 0.03 | 0.08 | 0.02 | 0.02 | 0.02 | 0.04 | 0.01 | p < 0.05 | p < 0.05 | p < 0.05 |
| Stubble | 0.07 | 0.11 | 0.08 | 0.20 | 0.08 | 0.09 | 0.06 | 0.12 | 0.04 | NS | p < 0.05 | NS |
| Roots | 0.16 | 0.17 | 0.17 | 0.25 | 0.17 | 0.16 | 0.18 | 0.18 | 0.05 | NS | NS | NS |
| Day 47 | | | | | | | | | | | | |
| AGB | 0.18 | 0.17 | 0.18 | 0.22 | 0.13 | 0.14 | 0.12 | 0.11 | 0.04 | p < 0.05 | NS | NS |
| Leaves | 0.05 | 0.04 | 0.06 | 0.07 | 0.03 | 0.03 | 0.02 | 0.03 | 0.01 | p < 0.05 | NS | NS |
| Stubble | 0.13 | 0.13 | 0.12 | 0.16 | 0.10 | 0.11 | 0.09 | 0.08 | 0.03 | p < 0.05 | NS | NS |
| Roots | 0.25 | 0.23 | 0.24 | 0.30 | 0.19 | 0.19 | 0.15 | 0.19 | 0.06 | p < 0.05 | NS | NS |

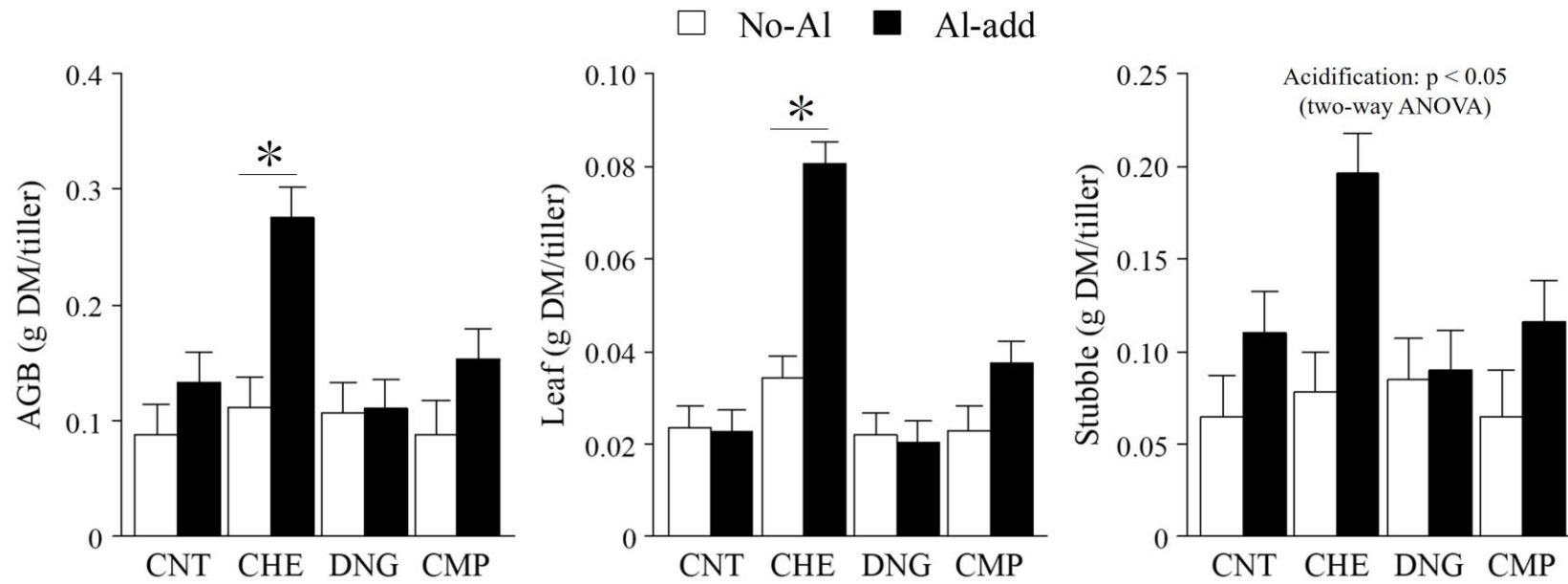
Results are presented as estimated marginal means with root mean square errors (RMSE). † CNT, unfertilized control; CHE, chemical fertilizer; DNG, grazing cattle dung; CMP, cattle manure compost. ‡ no-Al, only water was added; Al-add, aluminum sulfate hydrate was added as a water solution to induce further soil acidification. § AGB, Aboveground biomass.

TABLE 4 Soil nutrients in each treatment on Day 20 and Day 47

| Soil nutrients (per kg dry soil) | CNT† | | CHE | | DNG | | CMP | | RMSE | Effect of treatment | | |
|-------------------------------------|--------|--------|-------|--------|-------|--------|-------|--------|------|---------------------|---------------|-------------|
| | no-Al‡ | Al-add | no-Al | Al-add | no-Al | Al-add | no-Al | Al-add | | Fertilizer | Acidification | Interaction |
| Day 20 | | | | | | | | | | | | |
| TC (g) | 75.1 | 76.7 | 76.3 | 76.9 | 76.4 | 79.3 | 79.1 | 76.7 | 3.8 | NS | NS | NS |
| TN (g) | 5.0 | 5.2 | 5.1 | 5.2 | 5.2 | 5.4 | 5.4 | 5.2 | 0.3 | NS | NS | NS |
| IN (mg) | 10.9 | 15.8 | 13.8 | 28.7 | 12.1 | 19.2 | 15.1 | 29.8 | 2.1 | p < 0.05 | p < 0.05 | p < 0.05 |
| NO ₃ -N (mg) | 3.9 | 5.1 | 3.9 | 7.3 | 4.5 | 5.9 | 4.4 | 7.1 | 1.1 | p = 0.07 | p < 0.05 | NS |
| NH ₄ -N (mg) | 7.1 | 10.7 | 9.9 | 21.5 | 7.7 | 13.4 | 10.6 | 22.8 | 1.9 | p < 0.05 | p < 0.05 | p < 0.05 |
| P ₂ O ₅ (mg) | 795 | 795 | 777 | 803 | 793 | 820 | 807 | 799 | 46 | NS | NS | NS |
| K ₂ O (mg) | 225 | 244 | 221 | 201 | 244 | 246 | 293 | 286 | 19 | p < 0.05 | NS | NS |
| CaO (mg) | 1052 | 1287 | 1216 | 1355 | 1119 | 1381 | 1193 | 1431 | 72 | p < 0.05 | p < 0.05 | NS |
| MgO (mg) | 147 | 172 | 148 | 166 | 161 | 183 | 160 | 182 | 11 | p < 0.05 | p < 0.05 | NS |
| Day 47 | | | | | | | | | | | | |
| TC (g) | 75.6 | 74.8 | 74.8 | 76.9 | 75.1 | 74.7 | 72.4 | 75.8 | 3.6 | NS | NS | NS |
| TN (g) | 5.1 | 5.0 | 5.0 | 5.1 | 5.1 | 5.0 | 4.8 | 5.1 | 0.2 | NS | NS | NS |
| IN (mg) | 12.3 | 7.8 | 8.7 | 16.9 | 16.9 | 12.6 | 12.0 | 17.5 | 2.7 | p < 0.05 | NS | p < 0.05 |
| NO ₃ -N (mg) | 1.2 | 1.4 | 1.2 | 0.8 | 1.3 | 1.1 | 1.0 | 1.4 | 0.4 | NS | NS | p = 0.08 |
| NH ₄ -N (mg) | 11.2 | 6.5 | 7.5 | 16.2 | 15.6 | 11.5 | 11.1 | 16.1 | 2.8 | p < 0.05 | NS | p < 0.05 |
| P ₂ O ₅ (mg) | 768 | 761 | 742 | 731 | 731 | 758 | 724 | 773 | 27 | NS | NS | NS |
| K ₂ O (mg) | 228 | 216 | 197 | 196 | 235 | 215 | 249 | 258 | 15 | p < 0.05 | NS | NS |
| CaO (mg) | 1099 | 1128 | 1080 | 1157 | 1085 | 1113 | 1089 | 1148 | 71 | NS | p = 0.07 | NS |
| MgO (mg) | 157 | 129 | 143 | 110 | 161 | 132 | 146 | 135 | 15 | p = 0.07 | p < 0.05 | NS |

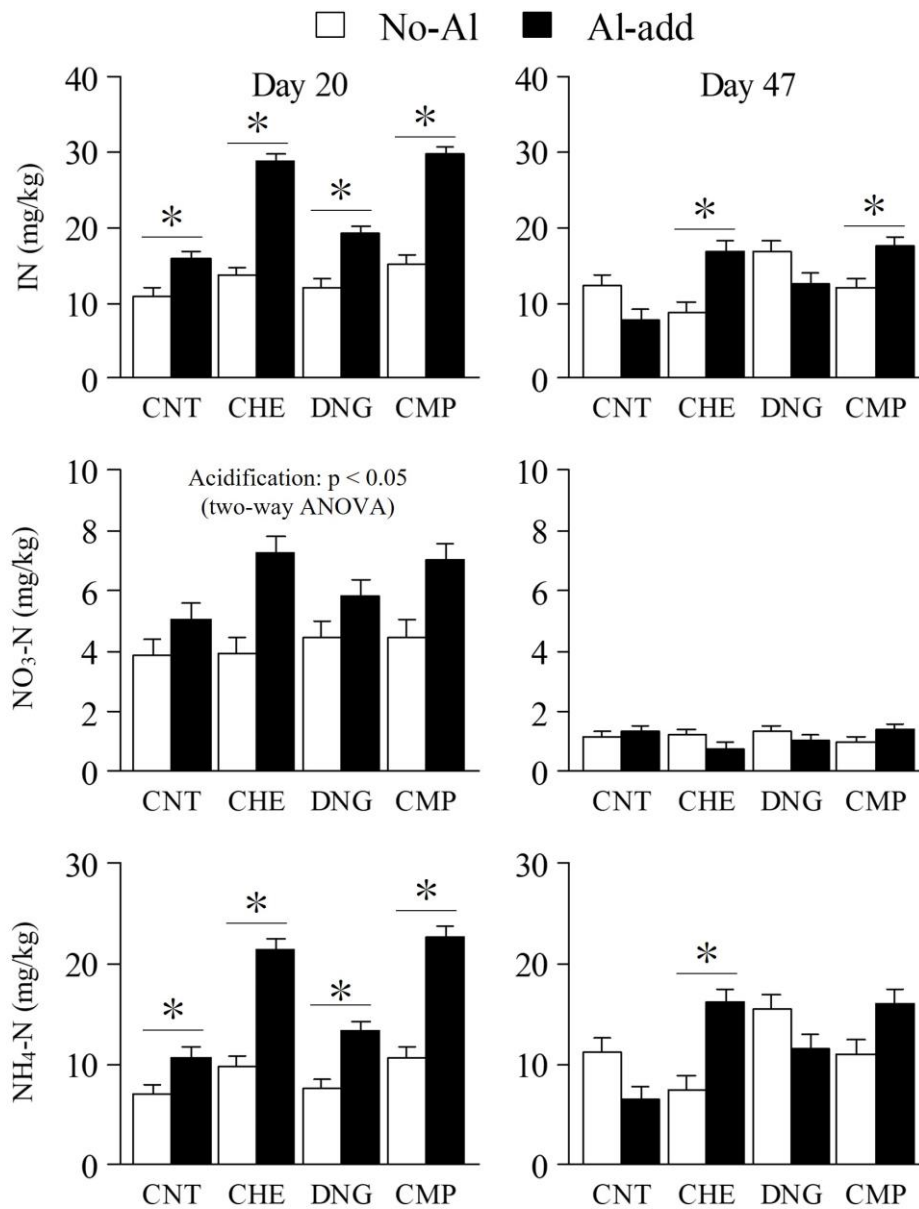
Results are presented as estimated marginal mean with root mean square error (RMSE). † CNT, unfertilized control; CHE, chemical fertilizer; DNG, grazing cattle dung; CMP, cattle manure compost. ‡ no-Al, only water was added; Al-add, aluminum sulfate hydrate was added as a water solution to induce further soil acidification.

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395

396 **FIGURE 1** Dry weight of aboveground biomass (AGB), leaves, and stubble of orchardgrass in each treatment (estimated marginal mean +
397 standard error) on day 20 after fertilization. Asterisks (*) indicate a significant difference between no-Al and Al-add within each fertilizer
398 treatment (two-way ANOVA and pairwise multiple comparisons with Holm's method, $p < 0.05$).



399

400 **FIGURE 2** Inorganic nitrogen (IN), nitrate-nitrogen (NO₃-N), and ammonium-nitrogen
 401 (NH₄-N) in dry soil of each treatment (estimated marginal mean + standard error) on
 402 day 20 (left) and day 47 (right) after fertilization. Asterisks (*) indicate a significant
 403 difference between no-Al and Al-add within each fertilizer treatment (two-way ANOVA
 404 and pairwise multiple comparisons with Holm's method, $p < 0.05$).