

**SOME PRELIMINARY STUDIES ON COMPOSITIONS OF LIGHTWEIGHT STRUCTURAL SHAPES BY FOAM METHODS**

**By**  
**Ronald G. Hardy**

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By

RONALD G. HARDY

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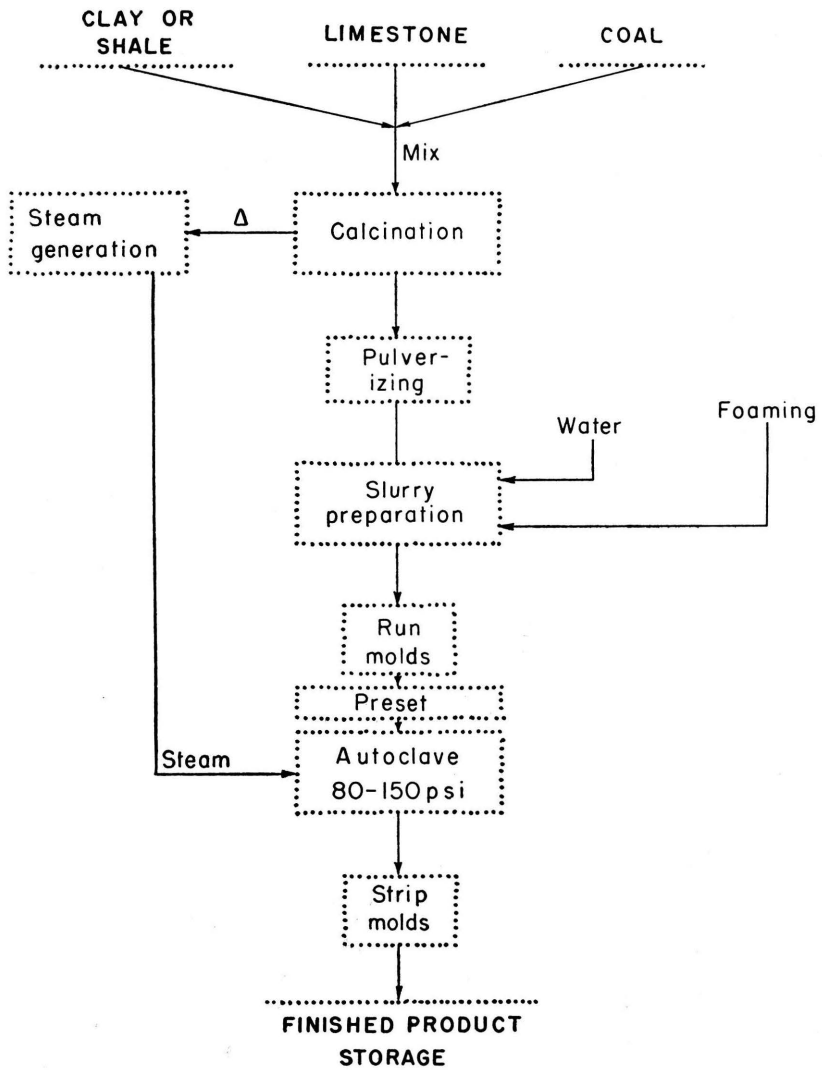


FIG. 1.—Diagram illustrating the steps necessary to produce cellulated shapes from raw materials.

## ABSTRACT

Cellular products made from foamed materials exhibiting pozzolanic action offer the building industry an attractive solution to problems involving stable, lightweight, durable, attractive, and fireproof structural shapes. A wide variety of Kansas raw materials is available to a manufacturer of such products. Three methods of introducing a foam have been generally used: whipping a foam in a suitable slurry, generating gases in the mix, and mixing a slurry and a pre-formed foam. Subsequent autoclaving is usually employed where structural strengths are desired in the products.

Production and materials costs are in line with other comparable constructional materials presently available.

## INTRODUCTION

### PURPOSE OF INVESTIGATION

The concept of using pozzolanic materials, substantially set with cellular structures, to form lightweight structural shapes is by no means new, as a review of literature (Anderegg, 1936, 1948; Anon., 1951, 1953; Dilnot, 1952) and patents (U.S. Patent 1,932,971, Oct. 31, 1933; U.S. Patent 2,442,519, June 1, 1948; British Patent 522,271, June 26, 1940; Canadian Patent 370,989, Jan. 3, 1938) reveals. The use of such materials has been developed rather extensively in some European countries, notably Sweden (Anderegg, 1947) and England. Recently definite plans have been evolved to produce structural shapes in the United States.

Basically, structural shapes made from cementlike materials that develop setting and bonding due to pozzolanic action and wherein lightness of weight is achieved by entrapping a multitude of cells offer certain attractive features. In addition to lightness of weight, structural strengths can be obtained by autoclaving, shapes may be reinforced with steel, their thermal conductivities are very low, the physical appearance is attractive, and a wide variety of materials can be employed.

Most of the raw material suitable for use in foamed concrete is abundantly available in Kansas. These include Portland cement, pulverized lightweight aggregates and stack dust from lightweight aggregate plants (Plummer and Hladik, 1951), fly ash, volcanic ash (Carey and others, 1952), limestone (Runnels, 1951), coal mine refuse, and fine silica sand (Nixon, Runnels, and Kulstad, 1950; Rose, 1950). This investigation was undertaken with the purpose of determining the relative merits of various foaming

materials and foaming techniques in mixes containing raw materials available in Kansas and from other sources.

#### METHODS OF INCORPORATING CELL STRUCTURE

In general there are three basic methods for incorporating a cell structure in a slurry of cementitious materials: (1) the slurry with or without additives can be suitably stirred or whipped in order to entrain air bubbles, (2) certain materials can be added to the slurry which generate gases causing bubble formation, and (3) a foam may be pre-generated and then mixed with the slurry (Abbot, 1952). All these methods have been tried by experimenters with varying degrees of success. In European practice much material has been foamed by the addition to slurries of a special aluminum powder which generates gas, producing a multitude of small bubbles and causing the mass to swell. Recently with the introduction of certain hydrolized proteins it is possible to produce very stable foams that lend themselves admirably to the method of whipping a cell structure in a cementitious slurry or pre-generating a foam and subsequently mixing with a slurry. The latter process has been used with considerable success in England; it has the distinct advantage of being easily controlled so that the final product characteristics can be determined in advance and corrections made as the process is in operation. Furthermore, the foamed slurry is pumpable.

#### CLASSIFICATION AND PROPERTIES OF AUTOCLAVED PRODUCTS

Generally speaking the autoclaved products from foamed slurries can be divided into three classes:

(1) *Products with a bulk density of 40 to 55 pounds per cubic foot and strengths of 700 to 1,500 psi.* This type of product has good structural qualities, high resistance to passage of water and heat transfer, and fire resistance. It is excellent for load-bearing walls.

(2) *Products with a bulk density of 30 to 40 pounds per cubic foot and strengths of 500 to 1,200 psi.* Such products are suited for use in nonload-bearing partition walls and roof slabs because of their heat resistance and good heat and acoustical qualities.

(3) *Products with a bulk density of 15 to 30 pounds per cubic foot.* The load-bearing ability of these densities is low (200 to 500

psi), but they afford good insulating qualities for door cores, sandwich wall construction, and refrigerated rooms.

The thermal conductivities (K factor) of these products range from 1.1 for the 40 to 45 pound per cubic foot material to 0.56 for the 20 pound per cubic foot material. Where low moisture penetration and good freezing resistance are desired, tests indicate these can be achieved by proper cell formation.

Proper autoclaving of structural shapes (Anon., 1953a) substantially reduces the reversible expansion and shrinkage due to wetting and drying as compared to unautoclaved products. This reduction can amount to one-half to two-thirds of the dimensional changes encountered in unautoclaved products.

Tests indicate that it is possible to produce unautoclaved material in the 15 to 20 pounds per cubic foot range with strengths of 75 to 150 psi and K factors of 0.50 to 0.60, which is entirely adequate for many uses.

## TESTS

Some of the materials that can be employed for foaming include Portland cement, hydrated lime, pulverized sand or flint, crushed and calcined pozzolanic shales, crushed cinders, fly ash, coal mine refuse, volcanic ash, lightweight aggregates, and pulverized chat. Tests indicate which materials alone, or in combination, will give satisfactory results.

Obviously, most if not all these materials are quite readily available in Kansas. Several preliminary tests performed on some combinations of these materials are described in the following discussions. Inasmuch as this investigation was somewhat exploratory, the majority of the results reported are qualitative rather than quantitative and as such are intended to point up lines for future detailed work.

### TESTS WITH FOAMING AGENTS IN THE MIX

*Series 1.*—For the first series of tests mixtures of limestone and coal washing refuse were calcined, ball milled until all passed a 60-mesh screen, made into a neat cement paste, and cast in small cylinders. The limestone had been crushed until all was less than one-fourth inch; the waste was used as received. Table 1 shows

TABLE 1.—*Preliminary limestone and refuse mixtures calcined for use in series 1 tests*

Mix no.	Limestone, percent	Coal refuse, percent
1	80	20
2	60	40
3	40	60
4	20	80
5	50	50
6	0	100
7	30	70

mixtures of refuse and limestone tried. Each mix was calcined at 1700° F., 1900° F., and 2100° F.

From observation of the apparent hardness of cast cylinders, mix 2 was most pozzolanic. The range of temperatures chosen gave no appreciable differences in the degree of set. The 50-50 mix (mix 5) was selected for more detailed tests.

Mix 5 was calcined at 1900° F., ground, made into a neat paste, and autoclaved at 15 psi. After initial set, it was a hard dense mass which showed no signs of cracking.

In another test mix 5 was pulverized in a ball mill using 2,225 grams of mix, 3 grams of Santomerse No. 1 (foaming agent), and 1,000 cc water. This slurry, in which a good cell structure of very fine cells had developed, was then run into a mold. This mass when autoclaved at 15 psi had very little strength. Autoclaving at 110 psi for 8 hours, however, resulted in quite good strengths.

Mix 5 used in the proportions of 2,225 grams of mix, 45 grams of Portland cement, 225 grams of minus 28-mesh fired clay aggregate, and 1,000 cc water, when autoclaved at 100 psi for 8 hours, developed fair strength.

It was apparent from this series of tests that suitable mixtures and techniques could be combined with the net result that a strong, lightweight, inert material could be produced. The foaming agent Santomerse would have performed better with the addition of a stabilizing agent, although it did fairly well alone. Higher and longer application of autoclaving pressures also undoubtedly would have helped to produce better strengths. The amount of fuel that still remains in the coal washing waste would be of considerable advantage where it might be desirable to calcine mixes on sintering machines—in many cases little or no additional fuel would be required.



TABLE 2.—*Preliminary mixtures of limestone, loess, and shale sintered for use in series 2 tests\**

Mix no.	Limestone, percent	Loess, percent	Shale, percent
1	40	40	20
2	50	30	20
3	40	20	40
4	50	20	30

\* 10 percent by weight of semi-anthracite coal was added before sintering.

*Series 2.*—In the course of some experiments with a batch-type sintering grate the mixtures listed in Table 2 were sintered, using 10 percent by weight of a semi-anthracite coal. After sintering, these materials were ball milled until 100 percent passed an 80-mesh screen. Using a high-speed stirrer, the ball-milled material was thoroughly mixed with water to a moderately thick slurry (approx. 125 cc water per 100 grams dry material) after which 0.5 percent aluminum powder was added and stirred thoroughly; then the slurry was run into molds. After setting 2 or 3 minutes the generated gas caused the mass to swell from 1½ to 2 times its original volume. After presetting overnight the samples were autoclaved at 100 psi for 6 hours. All these mixes produced fairly sound, moderately strong cellulated products. The cell shape was definitely directional.

Several mixtures containing volcanic ash, hydrated lime and/or high early cement were cellulated with aluminum powder. In general these mixes, while developing a cell structure, were inclined to be too soft and friable, even when autoclaved, to show promise as a structural material. One such mix consisting of 20 percent volcanic ash, 30 percent pulverized flint, 20 percent high early strength cement, and 30 percent hydrated lime, when cellulated with 0.5 percent aluminum powder and autoclaved at 100 psi for 6 hours, had a compressive strength of only approximately 100 psi. It should be noted that the cellular structure of this mix was very coarse.

In general, the mixes in which aluminum powder was used to produce a cellular structure were characterized by large irregular cells, directional in shape. Also it was noted that in order to use aluminum powder successfully, very close control of the various operations would be required.

TABLE 3.—*Slurry batches used in tests with pre-generated foam*

Mix no.	High early Portland cement	Volcanic ash	Pulverized silica	Hydrated lime
1	40	30	30	—
2	50	—	50	—
3	60	10	30	—
4	60	—	40	—
5	80	—	20	—
6	—	—	50	50

## TEST USING THE PRE-FOAMING TECHNIQUE

*Series 3.*—A series of mixes were investigated in connection with a pre-foaming technique. This method requires a generated foam and a cementitious slurry prepared separately. Then these two are mixed and the entire mix run into molds. In general this was a very satisfactory method for producing a cellulated mass.

Two foaming ingredients were tried, Armac 12-D, a fatty amine acetate (90 percent dodecylamine acetate) and Mearl-crete P, which is of the hydrolyzed protein type. In each case the foam was generated with a high-speed stirrer.

In the test employing Armac 12-D the foam-producing material had the following composition:

Armac 12-D	— 5 parts
Water	—100 parts
Plaster of Paris	— 30 parts

This gave a foam that was stable enough for later mixing.

TABLE 4.—*Properties of foamed material prepared by adding varying amounts of Armac 12-D foam to mix 1 shown in Table 3*

Mix no.	Ratio, foam to slurry	Oven-dry weight, pounds per cubic foot	Compressive load, pounds per square inch*
0	0	81.5	4,280
1	1-100	69.0	1,510
2	2-100	61.1	3,430
3	3-100	60.0	2,100
4	4-100	52.5	500
5	5-100	52.0	400
6	6-100	45.0	250

\* One specimen, approximately 2 by 2 by 2 inch cubes.

TABLE 5.—Properties of shapes made from mixes 2-6 listed in Table 3 after autoclaving

Mix no.	Oven-dry weight, pounds per cubic foot	Compressive load, pounds per square inch*
2	33.3	216
3	36.5	486
4	39.0	421
5	41.8	710
6	28.9	347

\* One specimen, a 2 by 2 by 2 inch cube.

Using a slurry batch that consisted of mix 1 shown in Table 3 prepared with 55 percent water, based on total weight of the dry material, increasing amounts of foam were added. Properties of the product after autoclaving for 6 hours at 100 psi are shown in Table 4.

The same foam composition was tried with several other combinations of Portland cement, volcanic ash and/or pulverized silica as well as with a hydrated lime and silica combination (Table 3, mixes 2-6). A thick slurry was made from each by adding water. To each 100 grams of dry materials in these mixes 135 grams of the foam composition was added. In this case the foam composition consisted of 5 grams Armac 12-D, 100 grams water, and 30 grams plaster of Paris. The combination of materials was poured into forms; after the initial set the shapes were removed from the forms and autoclaved at 100 psi for 6 hours. The properties of shapes made from the various mixes are given in Table 5.

Mearl-crete P, a dark-brown liquid, is a complete foaming agent requiring no further additions. The mixes listed in Table 6 were recommended by the manufacturer of Mearl-crete P. The

TABLE 6.—Raw material mixes recommended by the manufacturer of Mearl-crete P, a foaming agent

Mix no.	Water, cc	High early strength Portland cement, grams	Pulverized silica, grams	Ground Kaw River —35 mesh sand, grams	Mearl-crete P foaming agent, cc
1	151	218	—	—	2.4
2	150	130	86	—	2.5
3	100	130	—	86	2.5
4	80	109	—	109	2.5
5	80	86	—	130	2.5

TABLE 7.—*Properties of shapes made from mixes listed in Table 6 after autoclaving*

Mix in Table 6	Oven-dry weight, pounds per cubic foot	Compressive load, pounds per square inch
1	41.9	805
2	40.5	967
3	55.0	637
4	45.0	342
5	33.6	258

water and dry ingredients were mixed thoroughly; then the foaming agent was added and the entire mix stirred vigorously for another 3 minutes. The resulting slurry was run into molds, allowed to preset, and the shapes autoclaved 6 hours at 100 psi. Properties of the processed shapes are given in Table 7. Mixes 1 and 2 were excellent, 3 was only fair, and 4 and 5 were poor from a working viewpoint. Mixes 4 and 5 exhibited bad slumping characteristics before preset.

#### CONCLUSIONS

Tests of foamed concrete in which the foaming agent was included in the mix indicate that good results can be obtained using an agent like Santomerse. Concrete foamed with aluminum powder, however, was weak and the cell structure was not good, indicating that very close control of the process would be necessary.

In general, tests in which the pre-foaming technique was used yielded better results than those in which the foaming agent was included in the batch. The superiority of the pre-foaming technique was evident in the better and more controllable cell structure, controllable densities, and greater ease in the preparation of batches and the molding of shapes.

Although the primary aim of this investigation was the testing of various foaming techniques, the tests demonstrated that a number of materials readily available in Kansas, such as Portland cement, pulverized lightweight aggregate, fly ash, volcanic ash, limestone, coal mine refuse, and fine silica sand, are suitable for use in foamed concrete.

**ECONOMIC CONSIDERATIONS**

The role of foamed cementitious materials in conjunction with precast shapes and autoclaving has been demonstrated as being a very useful one; European practice has demonstrated the practicability of such a combination. Such precast shapes include structural and insulating blocks; floor, wall, and roof slabs; beams; and lintels. It is to be noted that reinforcing steel can be incorporated where necessary.

Foamed autoclaved cementitious materials offer the building industry a solution to many of its problems. Several advantages that can be realized from the use of these materials are (1) the raw materials are widely available; (2) processing is relatively simple; (3) the products from such materials exhibit lightness in weight, adequate strength, high thermal insulation, fireproofness, high resistance to moisture penetration and frost action, and extremely low wetting and drying movement, and they can be cut, sawed, drilled, screwed, or nailed; (4) they are a good plastering material; and (5) they are highly inert to normal agencies.

Reported data show that the costs of the raw material prepared for foam additions falls in the range of 7 to 12 cents per cubic foot of product, weighing 40 to 45 pounds per cubic foot. Foaming agent will add another 2 to 5 cents. Direct labor will add another 12 to 15 cents per cubic foot of product. It has been reported that a highly mechanized plant is capable of producing a unit 4 by 8 by 16 inches for a total cost, including overhead, of approximately 6.5 cents which is equivalent to a cost of 21.9 cents per cubic foot. The equivalent cost per cubic foot of 8 by 8 by 16 inch blocks, within the range of 14 to 20 cents per block, is shown in Table 8.

**TABLE 8.—Cost per cubic foot of aggregate at various prices per block**

8 by 8 by 16 inch block cost, cents	Equivalent cost per cubic foot, cents
14	23.6
15	25.2
16	27.0
17	28.6
18	30.4
19	32.0
20	33.8

In the field of roof slabs and planks, foamed autoclaved cementitious silicates offer an ideal basic material. The market price for good roofing materials is in general higher than for structural blocks which would indicate that these materials might have their greatest initial usefulness in this field.

While the application of this type of foamed material has not been studied extensively for this particular use, it has indications of being a good infill component of prefabricated walls.

The investment in a complete efficient plant to produce a finished product from raw material for an average operation would probably represent \$2,000,000 to \$3,000,000. If a manufacturer did not wish to produce a cementitious material but preferred only to make products, a satisfactory plant would require \$400,000 to \$700,000 for installing an efficient, economical operation. For the latter type of operation the autoclaves represent 60 to 80 percent of the total cost.

#### REFERENCES

- ABBOTT, J. A. (1952) How low-density solids are made: *Chemical Engineering*, vol. 59, no. 7, pp. 212-215, 401-404.
- ANDEREGG, F. O. (1936) It's Microporite: *Rock Products*, vol. 39, no. 5, pp. 48-51.
- \_\_\_\_\_ (1947) Ytong, a Swedish lightweight building unit: *Rock Products*, vol. 50, no. 8, p. 110.
- \_\_\_\_\_ (1948) Lightweight concretes; gas, foam and lime concretes: *Rock Products*, vol. 51, no. 11, p. 91.
- ANONYMOUS (1951) Cellular concrete: *Architectural Record*, vol. 110, no. 2, p. 159.
- \_\_\_\_\_ (1953) Cellular concrete: *Pit and Quarry*, vol. 46, no. 1, pp. 259-260.
- \_\_\_\_\_ (1953a) High pressure curing: *Rock Products*, vol. 56, no. 6, pp. 198-199.
- CAREY, J. S., AND OTHERS (1952) Kansas volcanic ash resources: *Kansas Geol. Survey, Bull.* 96, part 1, pp. 1-68.
- DILNOT, S. (1952) Cellular calcium silicate products: *Rock Products*, vol. 55, no. 10, pp. 110-116.
- NIXON, E. K., RUNNELS, R. T., AND KULSTAD, R. O. (1950) The Cheyenne sandstone of Barber, Comanche, and Kiowa Counties, Kansas, as raw material for glass manufacture: *Kansas Geol. Survey, Bull.* 86, pt. 3, pp. 41-84.
- PLUMMER, NORMAN, AND HLADIK, W. B. (1951) Lightweight aggregate from Kansas clays and shales: *Kansas Geol. Survey, Bull.* 91, pp. 1-100.
- ROSE, K. E. (1950) Silica sand from south-central Kansas for foundry use: *Kansas Geol. Survey, Bull.* 86, pt. 4, pp. 85-104.
- RUNNELS, R. T. (1951) Some high-calcium limestones in Kansas: *Kansas Geol. Survey, Bull.* 90, pt. 5, pp. 77-104.

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- PART 2. A SPINY APTYCHUS FROM THE CRETACEOUS OF KANSAS, by Alfred G. Fischer and Robert O. Fay, pp. 77-92, fig. 1, pls. 1-2, June 15, 1953.
- PART 3. COMPOSITION OF SOME URANIUM-BEARING PHOSPHATE NODULES FROM KANSAS SHALES, by Russell T. Runnels, John A. Schleicher, and H. S. Van Nortwick, pp. 93-104, figs. 1-3, October 1, 1953.
- PART 4. EXPERIMENTAL SEPARATION OF IRON-BEARING MINERALS FROM CERTAIN KANSAS CLAYS, by Frank W. Bowdish, pp. 105-116, October 15, 1953.
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- PART 7. IMPROVEMENT OF SOME KANSAS CLAYS THROUGH THE CONTROL OF pH AND SOLUBLE SULFATES, by Norman Plummer and William B. Hladik, pp. 145-173, figs. 1-8, October 30, 1953.

### BULLETIN 109

#### 1954 REPORTS OF STUDIES

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