Supplementary Materials

Hydrogels from a Self-Assembling Tripeptide and Carbon Nanotubes (CNTs): Comparison between Single-Walled and Double-Walled CNTs

Petr Rozhin [1], Slavko Kralj [2,3], Brigitte Soula [4], Silvia Marchesan [1,*] and Emmanuel Flahaut [4,*]

[1] Department of Chemical and Pharmaceutical Sciences, University of Trieste, 34127 Trieste, Italy
[2] Department for Materials Synthesis, Jožef Stefan Institute, Jamova 39, 1000 Ljubljana, Slovenia
[3] Department of Pharmaceutical Technology, Faculty of Pharmacy, University of Ljubljana, Aškerčeva 7, 1000 Ljubljana, Slovenia
[4] Centre Interuniversitaire de Recherche et d'Ingénierie des Matériaux, Université Paul Sabatier, UMR CNRS N°5085, 31062 Toulouse, France
* Correspondence: smarchesan@units.it (S.M.); emmanuel.flahaut@univ-tlse3.fr (E.F.)

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1. Vis-NIR spectra of SWCNTs

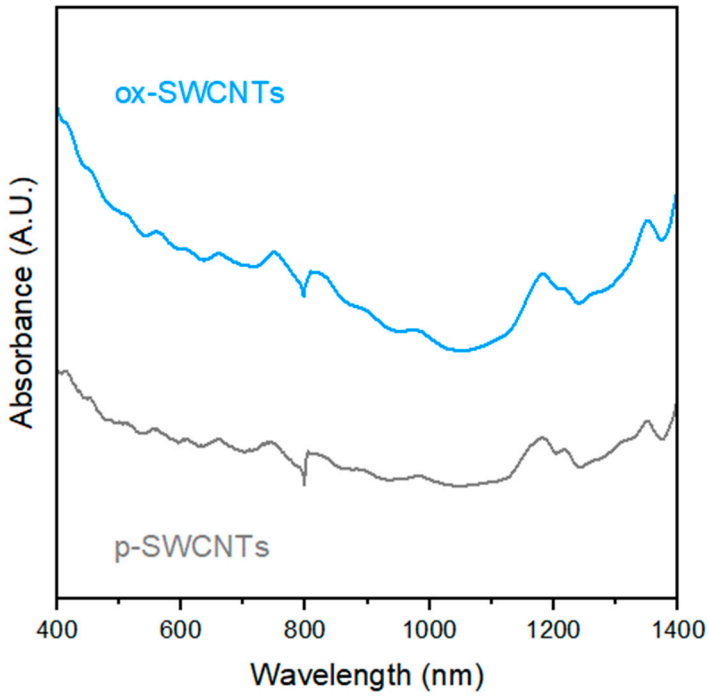

Figure S1. Vis-NIR absorbance spectra of pristine SWCNTs (p-SWCNTs, grey) and oxidized SWCNTs (ox-SWCNTs, light blue). No significant difference was found between the two simples.

2. Raman spectra of SWCNTs

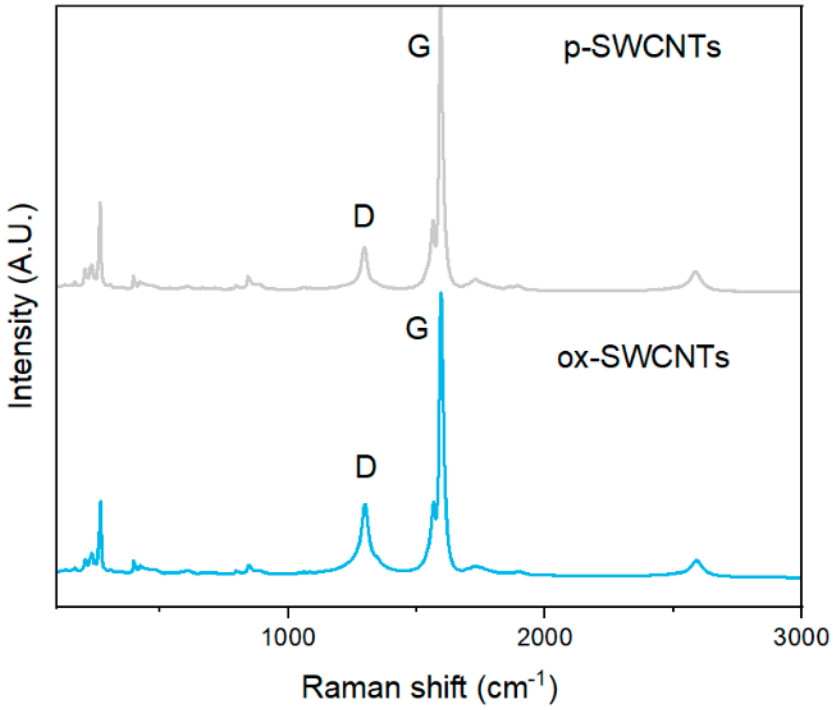

Figure S2. Raman spectra of pristine (p-) SWCNTs (grey) and oxidized (ox-) SWCNTs (light blue).

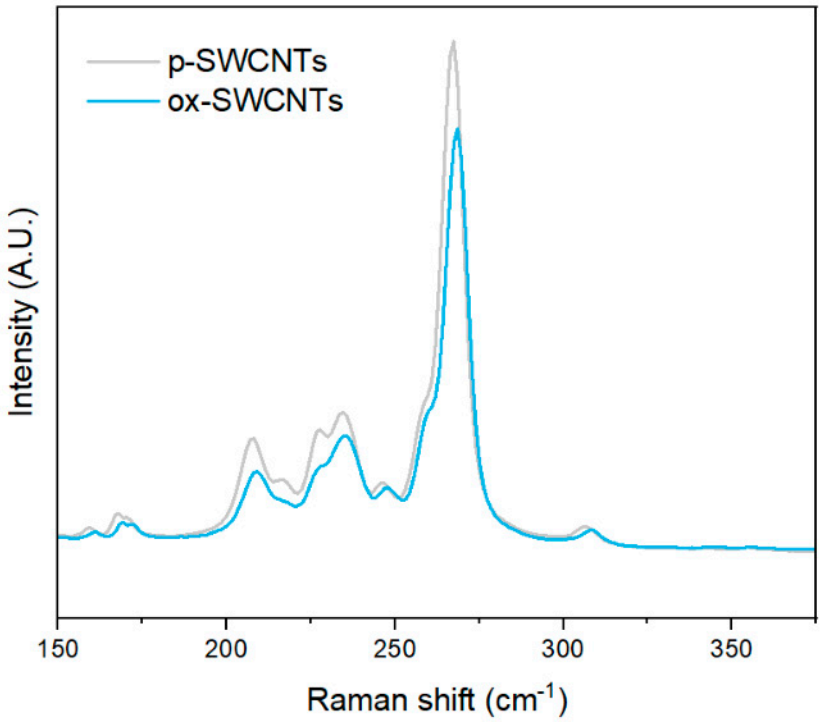

Figure S3. RBM region of Raman spectra of pristine (p-) SWCNTs (grey) and oxidized (ox-) SWCNTs (light blue).

Table S1. SWCNT diameters (d) calculated from the RBM frequencies (ω), using the equation $\omega_{RBM} = A/d_t + B$, where A is associated with the vibrational force constant of the sp^2 C–C bond, and B is related to environmental effects. For typical SWCNT bundles, A = 234 cm^{-1}·nm and B = 10 cm^{-1} (see S. Costa et al., Materials Science-Poland 2008, 26, 433). The average calculated SWCNT diameter was 1.0 ± 0.1 for both pristine and oxidized SWCNTs.

p-SWCNTs		**ox-SWCNTs**	
ω (cm^{-1})	***d* (nm)**	**ω (cm^{-1})**	***d* (nm)**
208	1.2	208	1.2
217, 228	1.1	226	1.1
235, 247	1.0	235, 248	1.0
267	0.9	269	0.9

3. TEM micrographs of SWCNTs

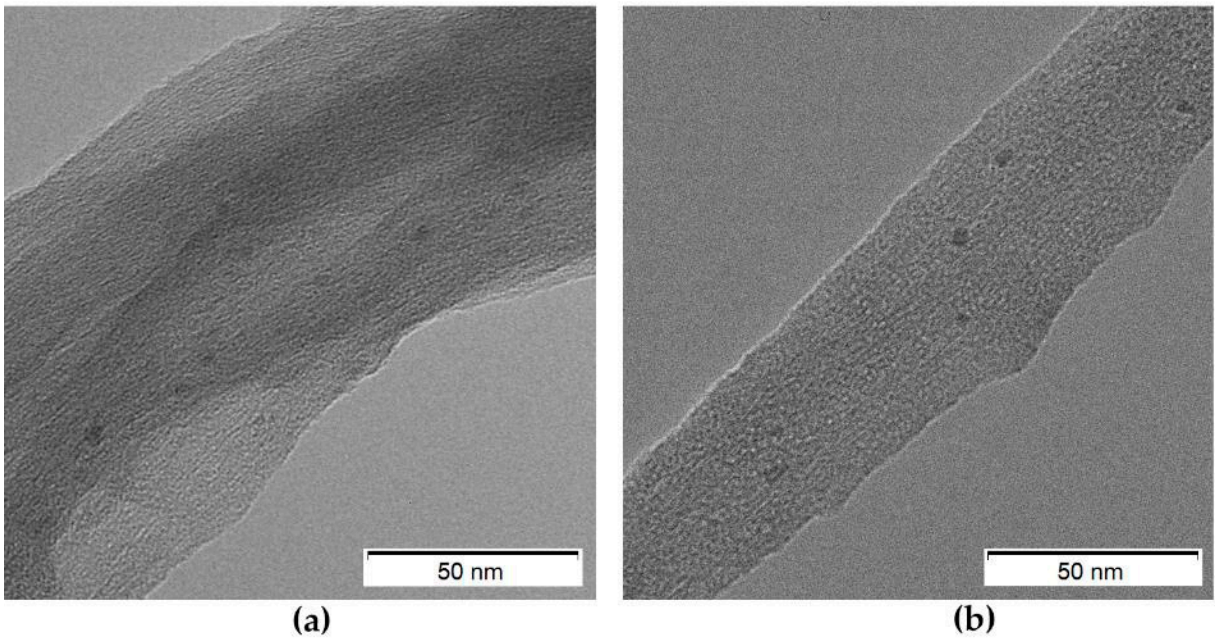

Figure S4. TEM micrographs of **(a)** pristine SWCNTs and **(b)** oxidized SWCNTs.

4. TEM micrographs of composite gels

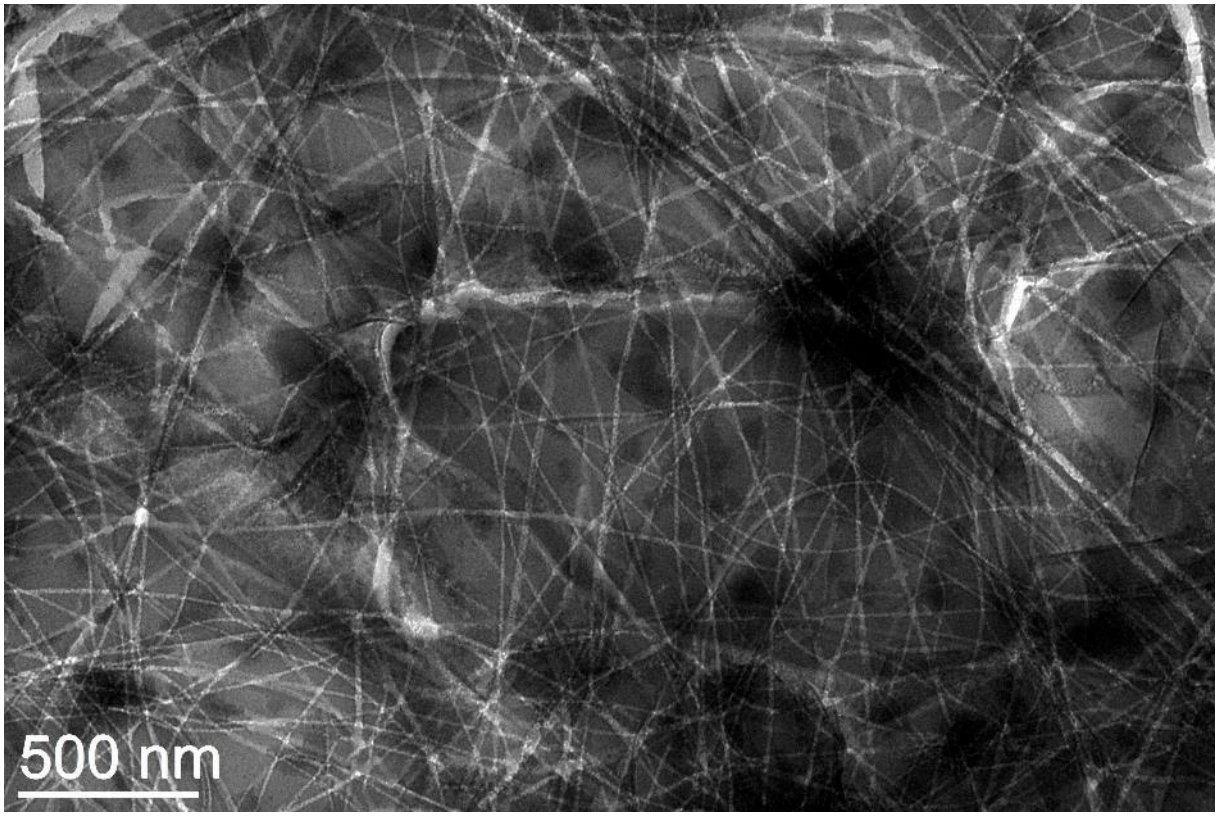

Figure S5. TEM micrograph of composite gel with 0.1 mg/mL ox-DWCNTs.

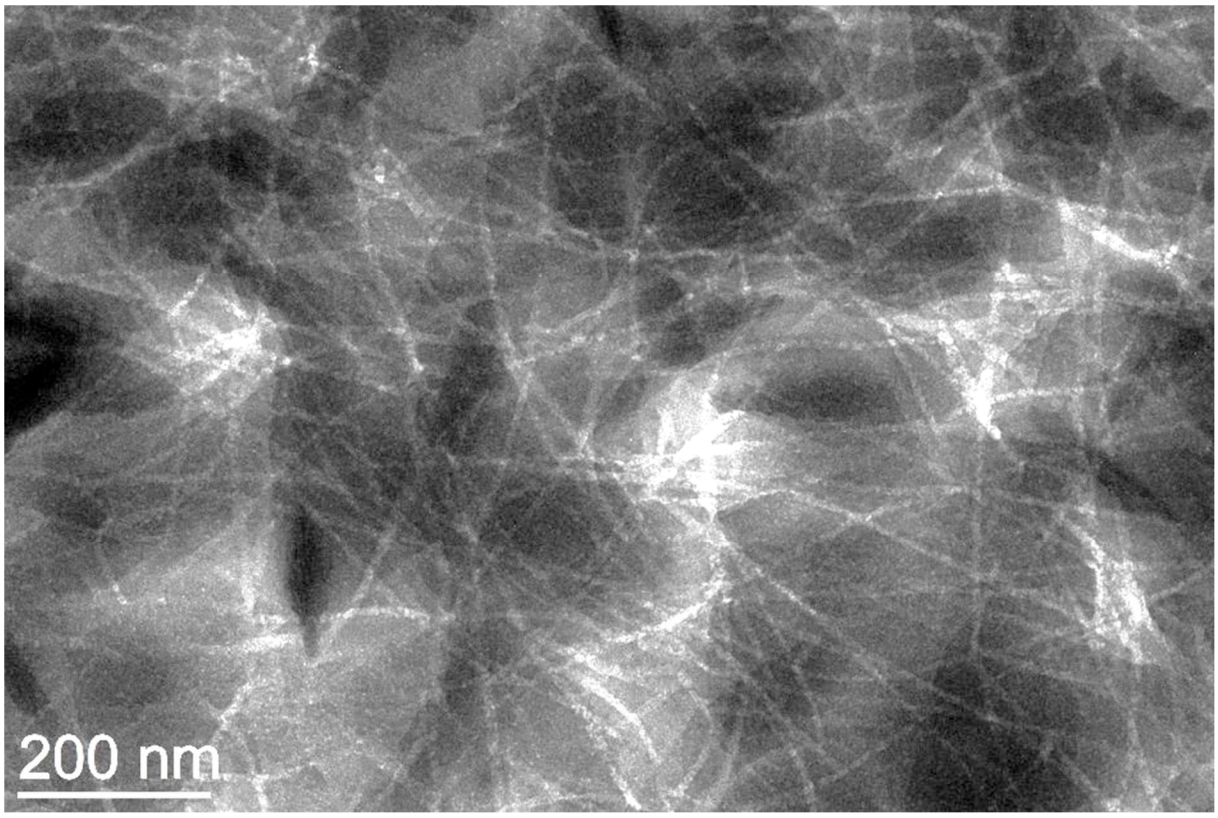

Figure S6. TEM micrograph of composite gel with 1.0 mg/mL ox-DWCNTs.

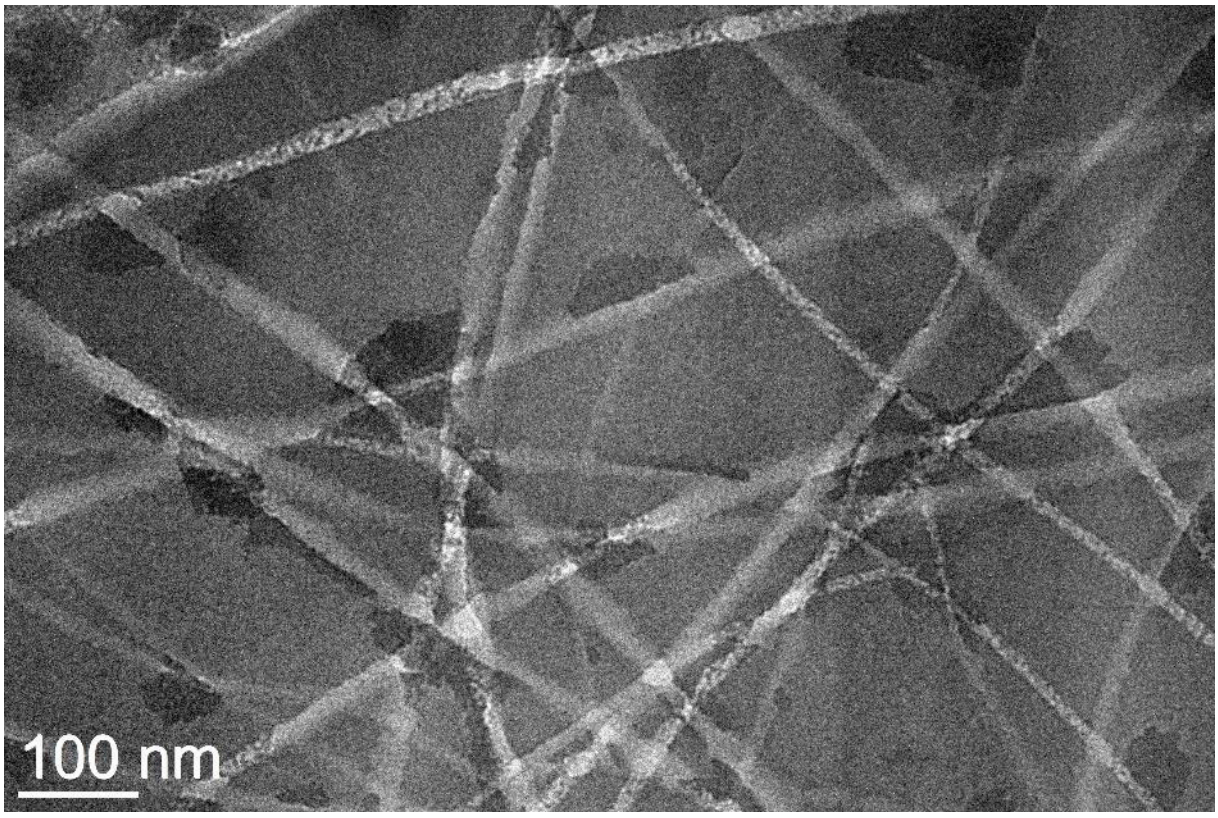

Figure S7. TEM micrograph of composite gel with 0.1 mg/mL ox-SWCNTs.

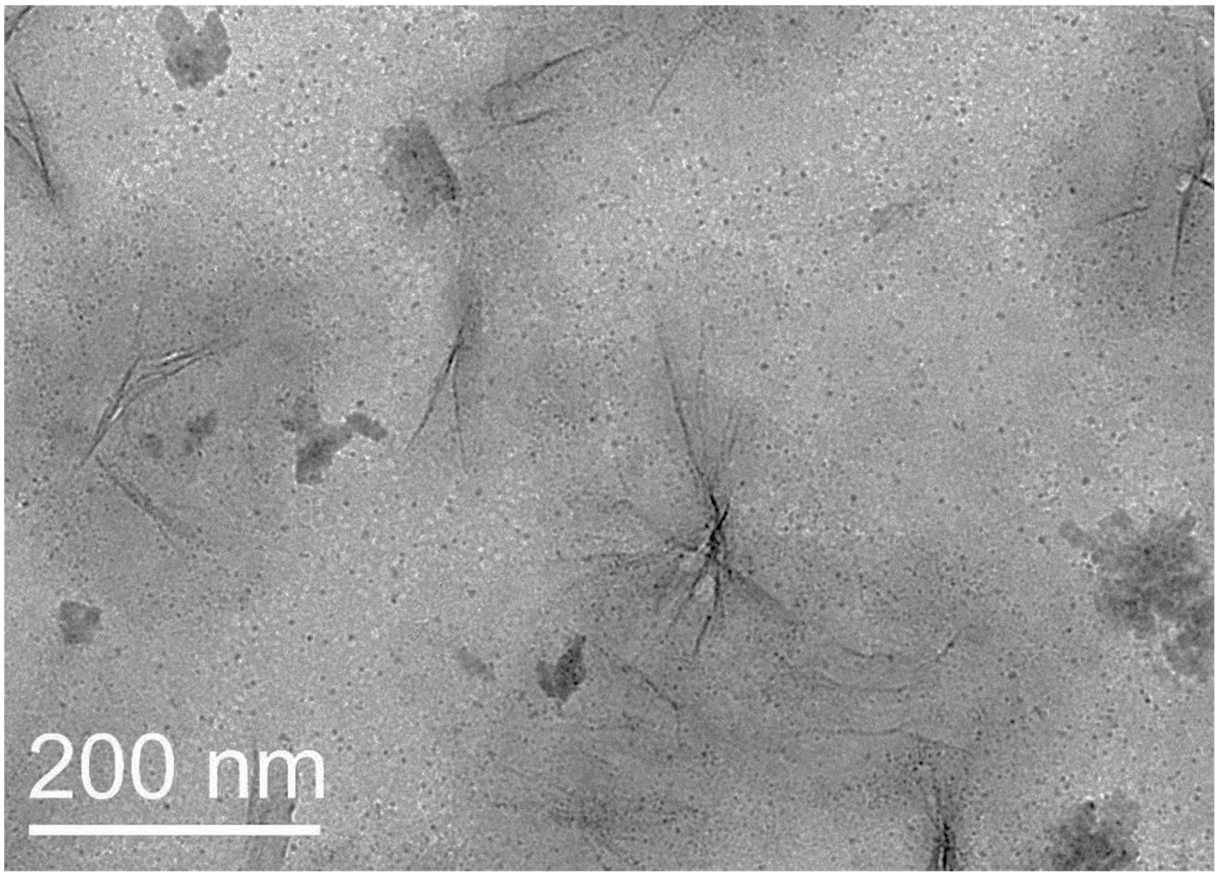

Figure S8. TEM micrograph of composite gel with 1.0 mg/mL ox-SWCNTs.