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Bioaccumulation and biochemical patterns of *Ruditapes philippinarum* clams: Responses to seasonality and low contamination levels

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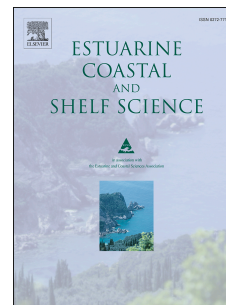
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1 Bioaccumulation and biochemical patterns of *Ruditapes philippinarum* clams:  
2 responses to seasonality and low contamination levels.

3  
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27

28 **ABSTRACT**

29 Shellfish farming and shellfish harvesting has been practiced for a long time in the Ria de Aveiro  
30 coastal lagoon (Portugal). Among commercial bivalves, Manila clam *Ruditapes philippinarum*  
31 represents one of the most important species inhabiting the Ria de Aveiro. Introduced in Portugal in  
32 1984, naturalised *R. philippinarum* clam populations have been subjected to several pressures that  
33 may threaten this resource sustainable management: illegal fishing, harvesting in chemically polluted  
34 sites with impacts on human health, lack of control in terms of productivity with the risk of a  
35 progressive decline of the biomass. On behalf of the ASARISAFE project (with the title Safety and  
36 sustainable management of valuable clam product in Portugal and China) the environmental quality of  
37 Manila clam harvesting sites was evaluated, focusing on inorganic pollution, health status of clams in  
38 terms of bioaccumulation as well as biochemical performance. Seasonal sampling campaigns were  
39 conducted in six *R. philippinarum* harvesting areas evaluating inorganic pollution levels, in clam's  
40 tissues, sediment and water. Clams biochemical performance in terms of metabolism, energy  
41 reserves and oxidative stress was also assessed. The results obtained showed that mercury and  
42 arsenic (As) were the elements with the highest BAF (Bioaccumulation factor) values, but  
43 contamination levels in tissues and sediments varied among sampling areas and seasonal  
44 campaigns. The amount of clams consumed per week to exceed Provisional Tolerable Week Intake  
45 (PTWI, kg) was the lowest for As, revealing that less 0.05 kg of clams was enough to exceed PTWI.  
46 However, the results obtained further demonstrated that the clam's biochemical performance was not  
47 responding to tissues contamination levels but were closely related to seasons, with distinct metabolic  
48 capacity and oxidative stress levels among distinct sampling periods during the year.

49

50 **Keywords:** Clams; metal contamination; safety consumption; seasons; biochemical performance.

51

## 52 1. INTRODUCTION

53

54 Coastal ecosystems, including lagoons and estuaries, are complex systems with high primary  
55 production (McLusky, 1999). They have a role of paramount importance in providing several  
56 ecosystem services, often associated to the sustenance of vast biological resources (Lillebø et al.,  
57 2015). However, these ecosystems are often negatively impacted by natural shifts (Govender et al.,  
58 2011) and anthropogenic activity (Langston et al., 2010), including climate change related factors and  
59 pollution. Classical environmental monitoring and ecological health status evaluation through water  
60 and sediment chemical analysis associated to the evaluation of biological effects upon inhabiting biota  
61 are approaches commonly used (WFD, 2000/60/EC) in order to assess negative impacts derived from  
62 chemical exposure before it becomes relevant in superior levels of the biological organization (Picado  
63 et al., 2007). Environmental monitoring has been based on the effects induced in benthic organisms,  
64 by the evaluation of alterations at the community level (benthic community parameters), and more  
65 recently, on individual and cellular levels (physiological and biochemical markers), mainly to assess  
66 the impacts of pollutants but, more recently, to investigate alterations derived from climate change,  
67 especially related with extreme weather events. Cellular alterations are widely described in literature  
68 as a response to natural and anthropogenic stressors (Magalhães et al., 2018, Munari et al, 2018,  
69 Gonçalves et al., 2017, Velez et al., 2016a, Carregosa et al., 2014, Harley et al. 2006). In particular,  
70 cellular biomarkers have been used to assess the negative impacts of metals and metalloids  
71 (Coppola et al., 2018), temperature (Keller et al., 2004), salinity (Freitas et al., 2015, Moreira et al.,  
72 2016) and pH (Velez et al., 2016b). Within benthic macrofauna assemblages, clam species are  
73 identified as important bioindicators due to their high abundance and filter-feeding habits and socio-  
74 economic relevance (reviewed in Bebianno et al., 2004).

75 The Manila clam (*Ruditapes philippinarum*) is a native species from the Indo-Pacific region,  
76 introduced in Europe at the beginning of the 1970s for culture purposes (Flassch and Leborgne, 1992,  
77 Jensen et al., 2004), becoming a highly exploited resource (Pranovi et al., 2006, Dang et al., 2010).  
78 This species is commonly exploited in a wide variety of aquatic systems due to its fast adaption to  
79 new environmental scenarios, fast growth and high commercial value (Usero et al., 1997). More  
80 recently, *R. philippinarum* (Adams and Reeve, 1850) was introduced in Portugal, being currently one  
81 of the most widely used bivalve species to assess environmental quality (Costa et al., 2013, Martín-

82 Diaz et al., 2007, Shin et al., 2002). As an example, Costa et al. (2013) performed histopathological  
83 assays in *R. philippinarum* specimens, aiming to assess the environmental quality of the Portuguese  
84 south coast. Studies conducted by Moschino et al. (2012) also demonstrated the capacity of Manila  
85 clam as a bioindicator species, revealing the clam's responses to pollutants concentrations.  
86 Nevertheless, under environmental conditions when ecosystem pollution levels are low it is often  
87 difficult to determine whether effects are due to pollutants or natural environmental shifts closely  
88 linked with the organism's life cycle (Sheehan and Power, 1999, Hook et al., 2014), which can  
89 seriously compromise the interpretation of monitoring data. Thus, it is important to understand how  
90 the natural variations associated with seasonal changes such as salinity and temperature may impact  
91 the inhabiting fauna life cycle and, consequently, can alter the organism's responses to pollutants.

92 Therefore, the general aim of the present study was to evaluate the capacity of *R.*  
93 *philippinarum* as bioindicator species in a low contaminated coastal system along four distinct  
94 seasons, testing the hypothesis that pollution levels may hide the effects induced by seasons on the  
95 clam's natural biochemical performance. For this, the biochemical performance of *R. philippinarum*  
96 specimens, collected from six different areas along the Ria de Aveiro (Portugal), characterized by  
97 different metal(oid)s concentrations, was assessed during four seasons (spring, summer, fall, winter).  
98 The risk for human health derived from clam's consumption was also evaluated.

## 99 2. METHODOLOGY

### 100 2.1. SITE DESCRIPTION

101 The present study was conducted at the Ria de Aveiro (Figure 1), a shallow, vertically  
102 homogeneous, coastal lagoon located on the northwest coast of Portugal. This aquatic system is 45  
103 km long and 10 km wide, comprising a total surface area of 83 km<sup>2</sup> at high tide, with 17 km<sup>2</sup> of  
104 intertidal flats emerging at low tide (Dias et al., 2000). In addition, this aquatic system is characterized  
105 by narrow channels and by large areas of mud flats and salt marshes (Picado et al., 2009).

106 Sampling was conducted in six different areas selected along the lagoon: Torreira (T - 40°45'  
107 43.0" N, 8°41' 56.7"W), Sporting (S - 40°40' 15 .2" N, 8°38' 45.9"W), São Jacinto (SJ - 40°42' 24.1"  
108 N, 8° 41' 50.6"W), Ílhavo (I - 40°36' 59.3" N, 8 ° 40' 51.3"W), Murtosa (M - 40° 43' 25.8" N, 8° 3 9'  
109 33.8"W) and Cale do Ouro (CO - 40°42' 02.9" N, 8 ° 41' 09.3"W) (Figure 1).

110

### 111 2.2. SAMPLING PROCEDURE

112 In each studied area, three sampling sites were selected and seasonally sampled (Winter,  
113 Spring, Summer and Autumn), from late 2017 to early 2019.

114 From each sampling site, eighteen specimens of *R. philippinarum* with similar size were  
115 collected (length: 4.6±0.64; width: 3.7±0.26). The whole soft tissue of three individuals was used for  
116 elements quantification, while the other fifteen individuals were used for biochemical analyses.

117 Additionally, at each sampling site, pH, salinity, dissolved oxygen (DO), redox potential (Eh)  
118 and temperature were measured in the sediment-water interface using a handheld multiparametric  
119 probe. Sediment samples were collected for sediment grain-size analysis, total organic matter (TOM)  
120 determination and elements (chromium, Cr; nickel, Ni; copper, Cu; lead, Pb; cadmium, Cd; mercury,  
121 Hg; arsenic, As) quantification.

122 Organisms and sediment samples for TOM determination and elements quantification were  
123 stored and transported on ice (approx. 0 °C) to the laboratory and afterwards preserved at -20 °C  
124 until analyses.

125

### 126 2.3. LABORATORY PROCEDURES

#### 127 2.3.1. ENVIRONMENTAL PARAMETERS

128 Sediment grain-size was carried out following the procedure described by Quintino et al.  
129 (1989). Silt and clay fraction (fine particles, with diameter  $< 0.063$  mm) were wet sieved, whereas  
130 sand ( $0.063 - 2.000$  mm) and gravel (particles with diameter  $> 2.000$  mm) fractions were dry sieved  
131 through a tower of sieves spaced at  $1 \text{ phi } (\Phi)$  ( $\Phi = -\log_2$  the particle diameter (mm)). Data were used  
132 to calculate the median grain-size value, P50, expressed in  $\Phi$  units. Median grain-size and the  
133 percentage of fines content were used to classify the sediment, according to the Wentworth scale:  
134 very fine sand (median from 3 to 4  $\Phi$ ); fine sand (2–3  $\Phi$ ); medium sand (1–2  $\Phi$ ); coarse sand (0–1  $\Phi$ );  
135 very coarse sand (–1 to 0  $\Phi$ ). All sediment grain-size fractions were expressed as a percentage of the  
136 whole sediment dry weight (DW). The final classification adopted the description 'clean', 'silty' or 'very  
137 silty' when the silt and clay fraction ranged from 0% to 5%, from 5% to 25% and from 25% to 50% of  
138 the total sediment DW, respectively (Doeglas, 1968). Samples with more than 50% fines content were  
139 classified as mud.

140 Total organic matter content (TOM) was determined according to Byers et al. (1978) as loss  
141 on ignition at  $450 \text{ }^\circ\text{C}$  (with minimal risk of volatilizing inorganic carbon) during 5 h.

142

### 143 2.3.2. ELEMENTS DETERMINATION

144 The concentration of mercury in water was determined by cold vapor atomic fluorescence  
145 spectroscopy (CV-AFS), with a PSA Millennium Merlin 10.036 analytical instrument, equipped with a  
146 detector PSA model 10.003. Stannous chloride (2% in 10% HCl) was used as reductant, and six  
147 standard solutions of Hg ranging between 2.5 and 60 ng/L, prepared by dilution of a commercial stock  
148 solution ( $\text{Hg}(\text{NO}_3)_2$ ,  $1000 \pm 2$  mg/L) in  $\text{HNO}_3$  (2% v/v), were used to obtain the calibration curve. The  
149 limit of quantification of the method was assumed as the lowest calibration standard, and a relative  
150 standard deviation among replicates  $< 5\%$  was considered.

151 In sediment and organisms, mercury was directly quantified in freeze-dried samples (2-20  
152 mg) by thermal decomposition atomic absorption spectrometry with gold amalgamation (LECO model  
153 AMA-254), as described by Costley et al. (2000). Detection and quantification limits were 0.01 ng Hg  
154 and 0.03 ng Hg, respectively. Each sample was analysed at least in triplicate with an acceptable  
155 relative standard deviation among replicates  $< 10\%$ . Blanks were run between sample analyses, and  
156 Certified Reference Materials TORT-2 (Lobster hepatopancreas;  $0.27 \pm 0.06$  mg/kg of total Hg) and

157 MESS-3 (Marine Sediment,  $0.091 \pm 0.009$  mg/kg of total Hg) were analysed several times daily. All  
158 percentages of recovery were within the range of 90-110%.

159 The concentrations of Cu, As, Cd, Pb, Ni and Cr in water were measured by Inductively  
160 Coupled Plasma-Mass Spectrometry (ICP-MS) at the Central Analysis Laboratory of the University of  
161 Aveiro. All samples were previously diluted 15x in  $\text{HNO}_3$  (2 %, v/v) to avoid interferences (due to the  
162 matrix) associated with salinity. The limits of quantification in water samples for the studied elements  
163 were assumed as the lowest calibration standards: 2  $\mu\text{g/L}$  (As, Cu), 1  $\mu\text{g/L}$  (Ni), 0.5  $\mu\text{g/L}$  (Cr), 0.2  $\mu\text{g/L}$   
164 (Pb) and 0.1  $\mu\text{g/L}$  (Cd).

165 In sediments and clams, the concentrations of Cu, As, Cd, Pb, Ni and Cr were also analysed  
166 by ICP-MS, after microwave assisted acid digestion, using a microwave system CEM MARS 5, model  
167 240/50. For quantification in sediments, 200 mg of homogenized air-dried sample was digested with 3  
168 mL of  $\text{HNO}_3$  (69%) and 6 mL of HF (40%) in Teflon vessel during 5 min with a ramping heating until  
169 175 °C, followed by 5 min at constant temperature of 175 °C. The samples were then evaporated  
170 near to dryness at 175 °C in a plate heater, followed by re-dissolution with 1.5 mL HCl (1:1 V/V) and 1  
171 mL  $\text{HNO}_3$  (69%), and finally transferred into 25 mL polyethylene flasks with the volume made up with  
172 ultrapure water. For quantification of the clam's soft tissues, 200 mg (previously frozen-dried) was  
173 transferred to Teflon bombs with 1 mL  $\text{HNO}_3$  65% (v/v) (Suprapur, Merck), 2 mL  $\text{H}_2\text{O}_2$  and 1 mL milli-  
174 Q  $\text{H}_2\text{O}$ . Samples were left 15 min in the microwave with increasing temperature up to 180 °C, which  
175 was maintained for 3 min. After cooling, samples were collected in polyethylene flasks, made up to a  
176 final volume of 25 mL with ultrapure water and stored at room temperature until quantification. The  
177 quality control was assured by running procedural blanks (reaction vessels without sample) and  
178 certified reference materials TORT-2 (for clams) and MESS-3 (for sediments) in parallel with samples.  
179 All blanks were below the quantification limit and the element recoveries in reference materials were  
180 always within the acceptable range of 80 to 120%.

181

### 182 2.3.3. BIOCHEMICAL PARAMETERS

183 After sampling, the clams were frozen, pulverized individually with liquid nitrogen and divided  
184 in 0.3 g fresh weight (FW) aliquots. Biochemical analyses were repeated in duplicate for each sample  
185 and biomarker. Extractions were performed using a 1:2 (w/v) proportion of specific buffers such as  
186 20% (w/v) trichloroacetic acid (TCA) buffer to perform lipid peroxidation (LPO). Reduced (GSH) and



187 oxidized (GSSG) glutathione parameters were carried out using a KPE buffer with 0.1% (v/v) Triton X-  
188 100 and 0.6% (w/v) sulfosalicylic acid. Potassium phosphate (50 mM, pH=7), 1mM EDTA, 1% (v/v)  
189 Triton X-100, 1mM DTT was used to perform superoxide dismutase (SOD), catalase (CAT),  
190 glutathione peroxidase (GPx), S-glutathione transferase (GST's), protein (PROT), glycogen (GLY)  
191 and Acetylcholinesterase (ATChI-ChE) tests. To assess electron transport system (ETS) activity,  
192 samples were extracted using a 0.1 M Tris-HCl pH 8.5 with 15% (w/v) PVP, 153  $\mu$ M magnesium  
193 sulfate ( $\text{MgSO}_4$ ) and 0.2% (v/v) Triton X-100 buffer.

194

#### 195 *2.3.3.1. Indicators of cellular damage and redox balance*

196 LPO was measured by quantifying malondialdehyde (MDA) according to the method  
197 described by Ohkawa et al. (1979) and the respective modifications referred in Carregosa et al.  
198 (2014). Absorbance was read at 535 nm ( $\epsilon = 156 \text{ mM}^{-1} \text{ cm}^{-1}$ ). LPO levels were calculated using  
199 Lambert-Beer Law and expressed in nmol of MDA formed per g of FW.

200 GSH and GSSG glutathione contents were determined according to Rahman et al. (2006)  
201 using GSSG as standards. Absorbance of GSH and GSSG was read at 412 nm. GSH and GSSG  
202 were expressed in  $\mu$ mol per g of FW. Reduced to oxidised glutathione ratio (GSH/GSSG) was  
203 calculated dividing GSH content by 2 x the amount of GSSG (adimensional).

204

#### 205 *2.3.3.2. Enzymatic defences*

206 SOD activity was determined based on Beauchamp and Fridovich (1971) method. SOD  
207 standards (0.25 – 60 U/ml) were used to perform calibration curve and SOD activity was measured  
208 spectrophotometrically at 560 nm. Activity was expressed in units of enzyme (U) per g of FW. One U  
209 corresponds to the conversion of 1  $\mu$ mol per min.

210 CAT activity was quantified according to Johansson and Borg (1988). The assay was carried  
211 out using formaldehyde standards and the absorbance was measured at 540 nm. The results were  
212 expressed in U per g of FW. One U is defined as the amount of enzyme that caused the formation of  
213 1.0 nmol of formaldehyde, per min.

214 Activity of GPx was quantified following Paglia and Valentine (1967). The absorbance was  
215 measured at 340 nm and determined using  $\epsilon = 6.22 \text{ mmol}^{-1} \text{ cm}^{-1}$  and the results were expressed as U

216 per g of FW. One unit of enzyme (U) represents the number of enzymes that caused the formation of  
217 1.0  $\mu\text{mol}$  nicotinamide adenine dinucleotide phosphate (NADPH) per min.

218 GSTs was determined following Habig et al. (1974). The absorbance was determined at 340  
219 nm using an extinction coefficient of  $9.6 \text{ mM}^{-1} \text{ cm}^{-1}$ , expressed in U per g of FW. One unit of enzyme  
220 (U) corresponds to the amount of enzyme that caused the formation of 1  $\mu\text{mol}$  of dinitrophenyl  
221 thioether per min.

222

#### 223 2.3.3.3. *Metabolic capacity and energy reserves*

224 ETS activity was measured based on King and Packard (1975) and modifications performed  
225 by De Coen and Janssen (1997) and the absorbance was read at 490 nm using  $\epsilon = 15,900 \text{ M}^{-1} \text{ cm}^{-1}$ ,  
226 expressed in nmol/min per g of FW.

227 PROT content was determined according to Robinson and Hogden (1940) and was carried  
228 out using bovine serum albumin (BSA) standards. The absorbance was read at 540 nm. Results were  
229 expressed in  $\text{mg}\mu$  of PROT per g FW.

230 Following the procedure described by Dubois et al (1956), GLY was quantified by the phenol–  
231 sulfuric acid method using glucose standards. The absorbance was measured at 492 nm. Results  
232 were expressed in mg of GLY per g of FW.

233

#### 234 2.3.3.4. *Neurotoxicity*

235 Acetylthiocholine iodide (ATChI, 5 mM) substrates were used for the determination of  
236 Acetylcholinesterase (ATChI-ChE) following the methods of Ellman et al. (1961) and modifications by  
237 Mennillo et al. (2017). The absorbance was measured at 412 nm and determined using  $\epsilon = 13600$   
238  $\text{nmol}^{-1} \text{ cm}^{-1}$ . The results were measured in nmol per min per g of FW and express the formation of the  
239 dianion of 5-thio-2-nitrobenzoic acid (TNB) per unit time (minute).

240

241

## 242 2.4. DATA ANALYSIS

243 Bioaccumulation factor (BAF) was determined dividing the total concentration of a given  
244 element in the organism tissue (DW) by the concentration of that element in the sediment (DW)  
245 (McGeer et al., 2003). The data matrix with the BAF per site and season [BAF x sampling area x

246 season] was normalised and the Euclidean similarity calculated between sampling areas. A Principal  
247 Coordinates Ordination analysis (PCO) was used to visualize differences among areas. The abiotic  
248 data highly correlated ( $r > 0.75$ ) were represented as superimposed vectors in the graph.

249 Data on sediment characteristics (contamination and physico-chemical properties) and  
250 species contamination were submitted to hypothesis testing using permutation multivariate analysis of  
251 variance with the PERMANOVA + add-on in PRIMER v6 (Anderson et al., 2008). The pseudo-F  
252 values in the PERMANOVA main tests were evaluated in terms of significance. When the main test  
253 revealed statistical significant differences ( $p \leq 0.05$ ), pairwise comparisons were performed. The t-  
254 statistic in the pair-wise comparisons was evaluated in terms of significance among different  
255 conditions. The main null hypotheses tested were: a) considering clams bioaccumulation, for each  
256 area, no significant differences existed among different seasons; b) considering clams  
257 bioaccumulation, for each season, no significant differences existed among different areas.

258 The data matrix including biomarkers and environmental data per site was normalised and the  
259 Euclidean distance calculated among centroids (i.e. the mean position of all the points representing a  
260 given sampling site for each one of the 4 seasons). Afterwards, the Euclidean similarity matrix was  
261 analysed using the PERMANOVA + add-on in PRIMER-E v.6 (Anderson et al., 2008) following  
262 unrestricted permutation of the raw data (9999 permutations) and the calculation of type III sums of  
263 squares. The main null hypotheses were: 1) considering clams biochemical responses, for each area,  
264 no significant differences existed among different seasons; 2) considering clams biochemical  
265 responses, for each season, no significant differences existed among different areas.

266 Afterwards, the matrix containing biomarkers and metal(oids) concentrations per sampling  
267 area and season was used to perform another Principal Coordinates Ordination (PCO) analysis. In  
268 the PCO graph, the variables presenting a correlation higher than 75 % with samples ordination were  
269 represented as superimposed vectors.

270

## 271 3. RESULTS

### 272 3.1 ENVIRONMENTAL PARAMETERS

273 In the present study, the obtained results showed that salinity and water temperature were  
274 higher in the Summer compared to the coldest seasons, Winter and Autumn, which presented the  
275 lowest values respectively. In warmer seasons, area T presented the highest temperature values,  
276 while areas S and CO were the coldest ones. Nonetheless, the lowest water temperature value of this  
277 study was recorded during Autumn in area M. Regarding salinity, the highest value was obtained in  
278 area T, during summer sampling, but CO presented the highest values in the remaining seasons.  
279 Area S showed the lowest values of salinity. Additionally, the highest values of pH and DO were also  
280 registered in Summer with areas M and T. On the other hand, area S, on average, presented the  
281 lowest pH values and area CO the lowest DO. Eh values were higher during Autumn in area M, while  
282 the lowest values were in Summer in area CO. Nevertheless, both highest and lowest single values  
283 were obtained in Autumn in areas T and CO, respectively (Table 1 mean values per season, Table 1  
284 Supplementary material for full data).

285 Concerning sediment data, Summer displayed higher median grain-size and percentage of fines.  
286 On the other hand, sediment mean grain-size was lower in Winter, whereas the lowest fines content  
287 was found in Spring. Comparing sampling areas, higher grain-size values and fines percentage were  
288 found in area CO and the lowest values observed in area SJ. Nonetheless, the majority of the  
289 sediments were classified as fine to very fine sand, as the areas presented  $\Phi$  values between 2 and  
290 4, with general proportions of fines greater than 10 %. Although values were similar most seasons,  
291 the percentage of organic matter content (TOM) was higher during Winter and lower during Spring.  
292 Area S presented the highest TOM, while area SJ showed the lowest content of organic matter  
293 (Table 1, Table 1 Supplementary material).

294

### 295 3.2. ELEMENTS CONCENTRATION

296 Comparing seasons, Summer presented the highest elements concentrations in the water  
297 column for As, Ni and Cr compared to the 3 other seasons that for the majority of the elements  
298 presented concentrations lower than the LOQ (Table 2 mean values per season, Table 2  
299 Supplementary material for full data). In terms of sampling areas, area CO was, in general, the area  
300 that presenting the highest levels of elements' concentration regardless the season. Nonetheless,

301 area M and area SJ presented the highest elements concentration in Autumn and Winter, respectively  
302 (Table 2 Supplementary material).

303 Considering elements concentrations in sediments, Summer and Autumn were the seasons  
304 that for all areas presented higher values, with Cr being the element with higher concentration for the  
305 majority of the sampling areas (mean values equal to 26.2  $\mu\text{g}/\text{g dw}$  and 27.9  $\mu\text{g}/\text{g dw}$  for Summer  
306 and Autumn, respectively) (Table 2). Comparing areas, sediments with higher elements  
307 concentrations were found mainly in area CO in Summer and Autumn, while higher elements  
308 concentrations were observed in area S in spring and area SJ in winter (Table 2 Supplementary  
309 material).

310 The highest element concentration for clam's tissues were recorded during Winter season.  
311 Elements concentration in clams showed higher As concentration compared to the remaining  
312 elements (Winter mean: 72.3  $\mu\text{g}/\text{g dw}$ ) (Table 2). Among areas, the highest concentrations were  
313 observed in area I (148  $\mu\text{g}/\text{g dw}$ ), where As was the element with the highest concentration (Table 2  
314 Supplementary material).

315 Winter was the season that recorded higher BAF values namely for As (13.5), which was the  
316 element more bioaccumulated in comparison with the remaining metal(oids) (Table 2). Area I  
317 presented the highest BAF levels for most of the elements (Table 2 Supplementary material).

318 The Principal Coordinates Ordination (PCO) regarding BAF values demonstrated that the axis  
319 1 explained approximately 49 % of the total variation, separating most of the area M samplings  
320 (except Winter sampling) and the Summer sampling of area I, on the positive side of the axis, from  
321 areas T, S and SJ regardless the season, on the negative side. Nonetheless, no abiotic factor showed  
322 a strong correlation with this axis. On the other hand, the axis 2, that described 21 % of total variation,  
323 divided areas M and T, on the positive side of the axis, from area I except summer sapling, on the  
324 negative side. Salinity presented a positive correlation with this axis (Figure 2).

325

326

### 327 **3.3. DIETARY RISK ASSESSMENT**

328 The concentrations of most of the elements quantified in clam's tissues were below the EFSA  
329 (European Food Safety Authority), USFDA (U.S. Food and Drug Administration) and FSANZ (Food  
330 Standards Australia New Zealand) maximum levels (Table 3 Supplementary material), except for As,

331 which exceeded safety limits. Overall, for the sampling sites under this study, As was the element of  
332 most concern in terms of human health. The obtained data showed that As values ranged between  
333 0.033 kg and 0.351 kg to exceed PTWI ( $1.05 \text{ mg } 70 \text{ kg}^{-1} \text{ week}^{-1}$ ) (Table 3). When comparing sampling  
334 areas, results showed higher human health risks for areas S and I, regardless the season (Table 3).

335

336

### 337 **3.4. BIOCHEMICAL PARAMETERS**

338 When analysing each sampling period (season) independently some patterns are  
339 highlighted: i) summer was the season that presented the highest values for LPO and GSH, while  
340 presenting the lowest values for ETS, GSTs and SOD. Autumn registered the highest values for SOD  
341 and CAT with the lowest values for AChE. Spring showed the highest values of ETS, GLY, GSTs and  
342 GSSG, while presenting the lowest values for PROT and GSH/GSSG. Winter was the season with  
343 higher levels of PROT, GPx, and AChE, while the lowest for GLY and CAT; ii) regardless the season,  
344 it was possible to identify a particular area for each set of biomarkers with the highest values for LPO,  
345 ETS and GLY in area T; PROT, GSTs and SOD in area S; area M for GPx and CAT; area I for AChE,  
346 GSH, GSSG and GSH/GSSG (Table 3).

347

348 The PCO axis 1 explained 40.1% of the total variation of data separating Winter (in the  
349 negative side) from the remaining seasons (in the positive side). Axis 2 described 24.9% of the total  
350 variation, separating Summer (in the negative side) from Spring (in the positive side). The results  
351 obtained clearly demonstrated that sampling areas grouped together according to season, with  
352 different areas from the same season clustering together. GPx, AChE and PROT presented a strong  
353 correlation ( $r > 0.75$ ) with PCO axis 1 negative side, with higher values associated to clams collected  
354 in Winter, regardless of the sampling area. On the other hand, CAT, GSSG and Ni content presented  
355 high correlation ( $r > 0.75$ ) with PCO axis 1 positive side, with higher values associated with clams  
356 collected in Autumn. ETS, GSTs and GLY showed a strong correlation ( $r > 0.75$ ) with the positive side  
357 of PCO axis 2, with higher values associated to clams collected during Spring in all the areas. LPO  
358 and GSH content presented a strong negative correlation with PCO axis 2, with higher values  
359 associated with clams from all areas collected during Summer (Figure 3).

## 4. DISCUSSION

### 4.1. ELEMENTS DETERMINATION

In the present study the results obtained showed higher water contamination during Summer, with area CO being the one with the highest values. The seasonal effect upon element concentration in water column was highlighted in this study suggesting higher metal(loids) water levels led to an additional concern about the elements' bioavailability, particularly during Summer. Generally, higher metal(loids) concentrations in water are related to environmental parameters such as temperature and DO as a result of greater dissolution of metals (Waldichuk, 1985).

Overall, the results obtained showed that the sampling areas represent low polluted to uncontaminated areas, with the concentration of elements in the sediments similar to values found in unpolluted areas (Chiesa et al., 2018; Velez et al., 2015; Freitas et al., 2012). In the sediments, higher concentrations of elements were obtained during Summer and Autumn and the most polluted area for these sampling periods was CO. Sediment becomes an important sink for metals that originally contaminate the water. Changes in the physicochemical parameters of water alter the bioavailability of the metals (Simpson and Batley 2003). The complex processes which influence the metal concentrations in the sediment are mainly pH, temperature, salinity, dissolved oxygen and organic matter content (Simpson et al. 2003), resulting in complex chemical reactivity and interactions between the solid and the solution phases of the metals (Guieu and Martin 2002, Peng et al. 2009). As example, Gati et al. (2016) assessed the sediment contamination with metals in the Danube Delta and how environmental shifts during one year can impact elements contamination. The authors showed that the deposition process was more intense at higher pH and temperature conditions.

Despite sediments showed to be an important source of metals, Bat et al. (2013) showed that sediment can reduce metals toxicity to mussels. Concerning clams' elements concentrations, the results obtained showed that Winter was the season with higher metal(loids) concentration where the most polluted area was I. The element of main concern was As. Velez et al. (2015) also assessed metals and As contamination on native and invasive clams from several areas of Ria de Aveiro (Portugal). The authors recognized this ecosystem as a low contaminated despite the concern associated to high concentrations of As. Despite higher metal(loids) concentrations in the water column during hotter seasons, clam's tissues showed higher accumulation during colder periods. This might be related with the low pH verified in the most affected sites (S and I). pH affects both solubility

390 of metal hydroxide minerals and adsorption-desorption processes. The solubility of metal hydroxide  
391 minerals increases along with the acidification leading to more dissolved metals that might become  
392 available for incorporation in biological processes. Ionic metal species are also commonly the most  
393 toxic forms to aquatic organisms (Salomons, 1995) which may explain different biochemical  
394 responses for different seasons. Riba et al. (2003) studied the effect of both pH and salinity upon  
395 water and sediment interactions by assessing biological effects on *R. philippinarum* organisms. The  
396 authors concluded that at low values of both variables (pH=6.5 and S=10), the biological effects were  
397 the highest, and it was related with free ion occurrence. Thus, this hypothesis might help to establish  
398 a pattern that varies not only with the temperature but also with other physico-chemical parameters.

399 The concentrations of As, Cd and Hg presented higher BAF values, with higher  
400 concentrations in the organisms (BAF>1) than in sediments, while for the remaining elements the  
401 concentrations were higher in the sediments than in the organisms (BAF<1). The toxicity of an  
402 element is not only dependent on the total amount accumulated but on its' partition as well. Elements  
403 in solution interfere with macromolecules with metabolically important functions, such as enzymes,  
404 transporters or DNA and therefore are more toxic than insoluble elements (Valko et al., 2005;  
405 Pytharopoulou et al., 2008; Zhang et al.,2010). Elements such as Cd and As, that are accumulated in  
406 higher proportions in the soluble fraction, are potentially more toxic than the others. These results are  
407 in agreement with the study conducted by Freitas et al. (2012) that performed an environmental study  
408 for *R. philippinarum* in the same ecosystem.

409

#### 410 **4.2. DIETARY RISK ASSESSMENT**

411 Clams are one of the main shellfish resources in the world, with 3.5 million tonnes produced  
412 in 2010 (FAO, 2011). The results presented in this work show that the risk of dietary exposure to  
413 inorganic elements from clam's consumption occurs predominantly in Spring time. However, for all the  
414 seasons, data showed a distinct hazard associated to As, especially during Spring at I area. In this  
415 area, data showed that an adult (70 kg) is in health risk danger by consuming 0.033 kg of clams, in  
416 one week. When comparing element concentrations of clams with previous studies also from an area  
417 in Ria de Aveiro with similar element concentrations (Figueira and Freitas et al., 2013), it is possible to  
418 observe the same pattern for the same elements. However, the metal(loids) concentrations increased  
419 when comparing the obtained Spring results with this study conducted by Figueira and Freitas et al.



420 (2013) (carried out in March) which may suggest a decrease of the ecosystem quality in the past  
421 years. When comparing with other systems worldwide, consumers of clams from this coastal system  
422 have a similar or lower risk of exceeding the PTWI for Cd, As, Pb, and Hg (Hamza-Chaffai et al.,  
423 2000, Kucuksezgin et al., 2010, I et al., 2012). In agreement with the study conducted by Figueira and  
424 Freitas (2013) the results obtained also evidence that, even at low-contamination areas, the maximum  
425 levels for some elements can easily be achieved, prohibiting marketing and preventing clams culture  
426 for commercial purposes in many areas. However, it is well known that other ecosystems (more  
427 polluted) have higher hazard standards such as China, in which the Environmental Quality Standards  
428 (EQSs) for Cd, Cu, and Zn are 0.35, 3.02, 51.4 µg/g dw in clams, respectively (Lu et al., 2019). Liu et  
429 al. (2017) conducted an environmental assessment along Laizhou Bay, China where Cd (53.19 mg/kg  
430 DW) and Hg (9.18 mg/kg DW) were the metals with higher tissues' concentrations. In comparison with  
431 the present study it is important to highlight that all the results are within the health risk levels and  
432 clams from Ria de Aveiro constitute a low source of metal(loids) through diet. Moreover, the hazard  
433 character of each element is well marked regardless the contamination level of the ecosystem, once  
434 that Liu et al. (2017) also concluded that As was the element of higher human health concern by  
435 hazard quotient assessment.

436 It is of paramount interest to carry out more environmental assessment and quality  
437 monitoring to understand how the potentially hazardous elements will impact human health. Despite  
438 the fact that this general concern is already well established, the dietary risk assessment is yet  
439 neglected favouring bioaccumulation, water and sediment concentration evaluations.

440

#### 441 **4.3. BIOCHEMICAL PARAMETERS**

442 The present study reveals biochemical responses of *R. philippinarum* when subjected to  
443 environmental stressors, analysing the relationship between pollutants levels in clams' tissues and  
444 biochemical responses. When exposed to stress factors, organisms may be subjected to oxidative  
445 damage resulting from increased concentration of Reactive Oxygen Species (ROS). Lipid  
446 peroxidation (LPO) and protein oxidation are well known effects of cellular oxidative damage,  
447 provoked by peroxidation of membrane lipids and proteins, respectively. In order to eliminate ROS  
448 and prevent cellular damages, bivalves may be capable of activating their antioxidant defences,  
449 namely by increasing their antioxidant enzymes.

450 In the present study the abiotic factors had an important role once the cellular damage  
451 increased along temperature, salinity and DO. Moreover, during colder seasons organisms were able  
452 to activate defence mechanisms in order to successfully prevent oxidative stress. Also higher  
453 detoxification rates seemed to respond to lower pH and Eh. In view to this, higher metabolic rates  
454 were verified which may explain the increase of the element's bioaccumulation. The importance and  
455 impact not only of environmental factors but also pollutants upon cellular fitness is already described  
456 in literature (Andrade et al., 2019; Coppola 2017, 2018; De Marchi et al., 2017; Velez et al., 2016a,b,c;  
457 Carregosa et al., 2014; Verlecar et al., 2007; Heise et al., 2003).

458 The results obtained further demonstrated that clams biochemical performance was closely  
459 related with seasons and not with areas elements concentrations, highlighting that pollution levels in  
460 all the studied areas were low and seasons induced higher impacts than pollutants on clams'  
461 biochemical responses. The present findings are in accordance with to a previously study by Guo et  
462 al. (2017), who detected the impact of contaminants in Qingdao coastal area of China in scallop  
463 *Chlamys farreri* during the year. In particular, the results obtained showed that during Summer clams  
464 presented higher LPO levels, which may indicate cellular damage, while it was demonstrated that  
465 higher neurotoxicity was observed during Winter. Higher LPO levels and GSH/GSSG ratio during  
466 Summer may result from ROS overproduction due to thermal stress but also due to increased  
467 mitochondrial respiratory activity as well observed in the Spring period. To prevent the accumulation  
468 of these molecules organisms produce and/or activate antioxidant enzymes. However, during  
469 summer, these defence mechanisms did not show well marked responses to detoxify ROS caused by  
470 thermal stress, leading to the occurrence of cellular damage (Solan and Whiteley, 2016). Also, Velez  
471 et al. (2017) verified higher LPO levels when exposed *R. decussatus* and *R. philippinarum* to warming  
472 conditions (21 °C) for 28 days. Furthermore, during Autumn the highest SOD and CAT activities were  
473 observed. These results may indicate that temperatures from 17 to 19 °C influence the antioxidant  
474 enzymes activations to prevent the cellular damage. Moreover, during Winter (lower temperatures),  
475 the GPx, GRed as well SOD and CAT activities showed to be able to prevent the membranes  
476 integrity, reduce the GSH/GSSG ratio and AChE showed a well-marked increase. Also De Marchi et  
477 al. (2017, 2018) studied the impact of pH and salinity combined with multi-walled carbon nanotubes  
478 and did not observe significant differences for AChE activities when exposed *R. philippinarum*  
479 organisms to low pH and salinity alone. Thus, the present study highlights the importance of

480 temperature and elements interactions. The water temperature rise is well documented in literature as  
481 an important environmental stressor alone or combined with other pollutants upon bivalves (Andrade  
482 et al., 2019; Coppola 2017, 2018; Verlecar et al., 2007; Heise et al.,2003), causing several effects  
483 assessed through different biomarkers, namely upon clams (Dubousquet et al.,2016; Anacleto et al.,  
484 2014; Abele et al., 2002). However, it is of paramount interest to assess and control environmental  
485 conditions in order to understand how several pollutants mixed in the water column interact among  
486 each other and with the inhabitant organisms. This purpose will allow adequate laws establishment  
487 and ensure the consumer's safety in terms of health risk.

488

## 489 **5. CONCLUSIONS**

490 It is of major importance to highlight that element partitioning is nowadays neglected by  
491 shellfish marketing. The impact that this has on the elements' bioavailability is very important to the  
492 consumers once that depends on the digestive capacity of each person (Rainbow and Smith, 2010;  
493 Metian et al., 2009). Bivalve species that have higher proportion of elements in solution generally  
494 constitutes higher risk to consumers than species that accumulate insoluble forms of metal(loids). The  
495 present study further highlights the importance of identifying the potential interfering factors and their  
496 impacts on the biomarker signals observed in wild populations. Biomarkers can thus, be significantly  
497 affected not only by anthropogenic or natural stressors but also by the combined action of both.  
498 Moreover, the optimal season for carrying out biomarker field studies or regular monitoring is of  
499 utmost relevance and should be investigated prior to including biomarkers in monitoring programs.

500 In conclusion, the present study emphasizes that benthic communities may provide more  
501 reliable information relatively to environmental fluctuations. Biomarkers can be used as  
502 complementary tools, however special attention is needed to choose appropriate bioindicator species,  
503 season as well as suitable battery of markers depending on nature of possible contaminants. Thus,  
504 this may lead to an increased ability to discriminate natural effects from others making biomarkers  
505 reliable in risk assessment studies.

506

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## Figure captions

Figure 1. Sampling areas: T - Torreira; M - Murtoza; SJ - São Jacinto; CO - Cale do Ouro; S - Sporting; I - Ílhavo.

Figure 2. Centroid ordination diagram (PCO) based on water and sediment physico-chemical parameters and values for the bioaccumulation factor (BAF), measured for all the sampling areas along four seasons. Pearson correlation vectors are superimposed as supplementary variables ( $r > 0.7$ ).

Figure 3. Centroid ordination diagram (PCO) based on biochemical data and clams metal(loids) concentrations, measured for all the sampling areas along four seasons. Pearson correlation vectors are superimposed as supplementary variables ( $r > 0.8$ ).

Table 1. Environmental characterization for each season (mean values  $\pm$ standard deviation), in terms of water parameters (salinity, temperature (TEMP. / °C), pH, dissolved oxygen (DO /  $\mu\text{g/L}$ ) and redox potential (Eh / mV)) and sediment parameters (median value ( $\Phi$ ), fines (%), total organic matter content (TOM / %)).

SEASON	TEMP	pH	SALINITY	DO	Eh	$\Phi$	FINES	TOM
SUMMER	20.0 $\pm$ 1.8	8.1 $\pm$ 0.3	35.1 $\pm$ 0.8	9.3 $\pm$ 3.4	120 $\pm$ 27	2.7 $\pm$ 0.9	38.3 $\pm$ 29.8	4.54 $\pm$ 2.30
AUTUMN	10.1 $\pm$ 0.7	8.0 $\pm$ 0.1	34.4 $\pm$ 0.9	7.5 $\pm$ 0.3	187 $\pm$ 72	2.6 $\pm$ 0.7	23.1 $\pm$ 13.7	4.25 $\pm$ 1.76
SPRING	19.1 $\pm$ 1.4	7.8 $\pm$ 0.1	25.3 $\pm$ 4.7	7.7 $\pm$ 1.0	148 $\pm$ 53	2.4 $\pm$ 0.4	15.7 $\pm$ 9.3	4.15 $\pm$ 1.93
WINTER	11.5 $\pm$ 1.2	7.8 $\pm$ 0.2	23.9 $\pm$ 7.4	8.7 $\pm$ 0.2	141 $\pm$ 10	2.4 $\pm$ 0.3	27.0 $\pm$ 18.4	4.63 $\pm$ 2.80

Table 2. Elements concentration (Cu, As, Cd, Pb, Hg, Ni and Cr) in Water ( $\mu\text{g/L}$ ), sediments ( $\mu\text{g/g}$ , dry weight) and clams' tissue ( $\mu\text{g/g}$ , dry weight) and the bioaccumulation factor (BAF: ratio between element concentration in the tissue and in the sediment) for each season (mean value  $\pm$  standard deviation (SD)). Quantification limit (QL) in water samples per element in  $\mu\text{g/L}$  (Cu, 2; As, 2; Cd, 0.1; Pb, 0.2; Hg, 0.0025; Ni, 1; Cr, 0.5). Values below this limit represented as <QL.

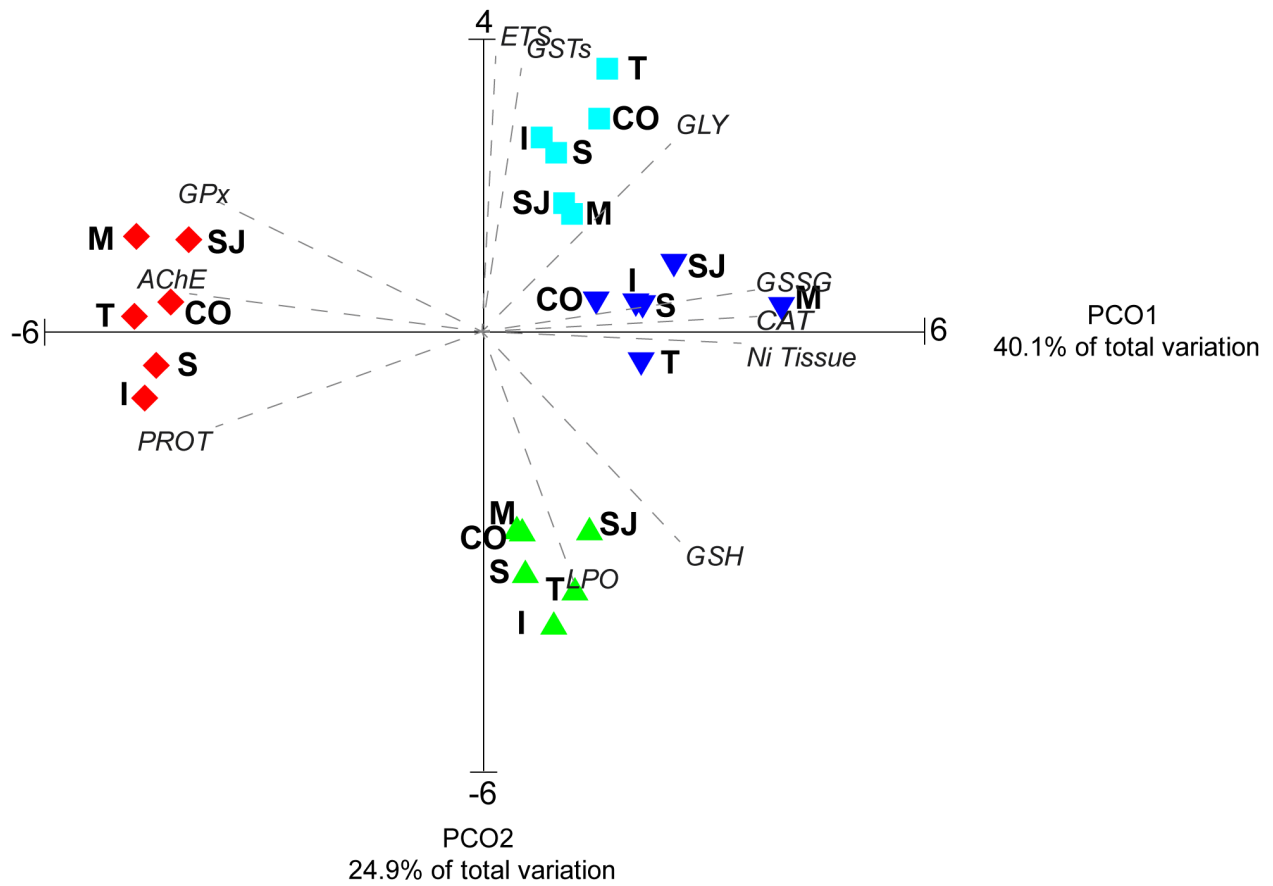
Season	Cu				As				Cd				Pb				Hg				Ni				Cr			
	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF	Wat.	Sed.	Tis.	BAF
SUMMER	40.7 $\pm 9.7$	11.7 $\pm 6.6$	6.3 $\pm 0.9$	0.9 $\pm 0.7$	37.8 $\pm 6.2$	11.9 $\pm 4.8$	50.0 $\pm 30.9$	6.2 $\pm 7.2$	<QL	0.3 $\pm 0.1$	0.6 $\pm 0.1$	3.3 $\pm 2.0$	<QL	25.2 $\pm 10.7$	3.0 $\pm 1.5$	0.16 $\pm 0.11$	18.2 $\pm 18.8$	0.12 $\pm 0.11$	0.2 $\pm 0.1$	3.5 $\pm 2.2$	25.7 $\pm 2.5$	11.3 $\pm 6.5$	6.38 $\pm 1.3$	1.1 $\pm 1.1$	14.0 $\pm 4.0$	26.2 $\pm 15.2$	2.6 $\pm 0.3$	0.2 $\pm 0.1$
AUTUMN	45.8 $\pm 3.7$	12.2 $\pm 6.3$	6.1 $\pm 1.1$	0.8 $\pm 0.5$	<QL	13.0 $\pm 4.3$	39.4 $\pm 27.6$	4.4 $\pm 2.4$	<QL	0.3 $\pm 0.1$	0.8 $\pm 0.1$	3.4 $\pm 1.8$	4.1 $\pm 0.3$	25.5 $\pm 6.7$	1.3 $\pm 0.4$	0.06 $\pm 0.01$	30.2 $\pm 18.4$	0.11 $\pm 0.09$	0.2 $\pm 0.1$	8.7 $\pm 15.5$	29.3 $\pm 0.0$	12.5 $\pm 5.8$	10.0 $\pm 2.8$	1.27 $\pm 1.0$	8.3 $\pm 0.7$	27.9 $\pm 12.9$	8.7 $\pm 8.1$	0.1 $\pm 0.2$
SPRING	46.0 $\pm 0.0$	7.7 $\pm 4.2$	6.3 $\pm 1.0$	2.2 $\pm 3.1$	<QL	6.8 $\pm 3.1$	53.6 $\pm 54.0$	10.6 $\pm 10.6$	<QL	0.3 $\pm 0.1$	0.4 $\pm 0.1$	2.1 $\pm 1.9$	3.8 $\pm 0.3$	24.1 $\pm 8.8$	1.2 $\pm 0.6$	0.06 $\pm 0.04$	9.9 $\pm 6.5$	0.10 $\pm 0.07$	0.2 $\pm 0.1$	6.9 $\pm 11.1$	<QL	7.80 $\pm 4.2$	6.49 $\pm 1.3$	1.45 $\pm 1.1$	<QL	19.9 $\pm 9.2$	3.4 $\pm 1.1$	0.1 $\pm 0.1$
WINTER	<QL	9.4 $\pm 4.0$	7.6 $\pm 0.8$	1.1 $\pm 0.9$	<QL	9.4 $\pm 4.0$	72.3 $\pm 46.3$	13.5 $\pm 18.8$	<QL		0.7 $\pm 0.2$		<QL		0.5 $\pm 0.1$		53.8 $\pm 55.0$	0.07 $\pm 0.04$	0.3 $\pm 0.1$	5.3 $\pm 1.5$	<QL	14.0 $\pm 7.6$	2.98 $\pm 0.8$	0.32 $\pm 0.3$	<QL	27.6 $\pm 13.8$	0.9 $\pm 0.5$	0.1 $\pm 0.1$

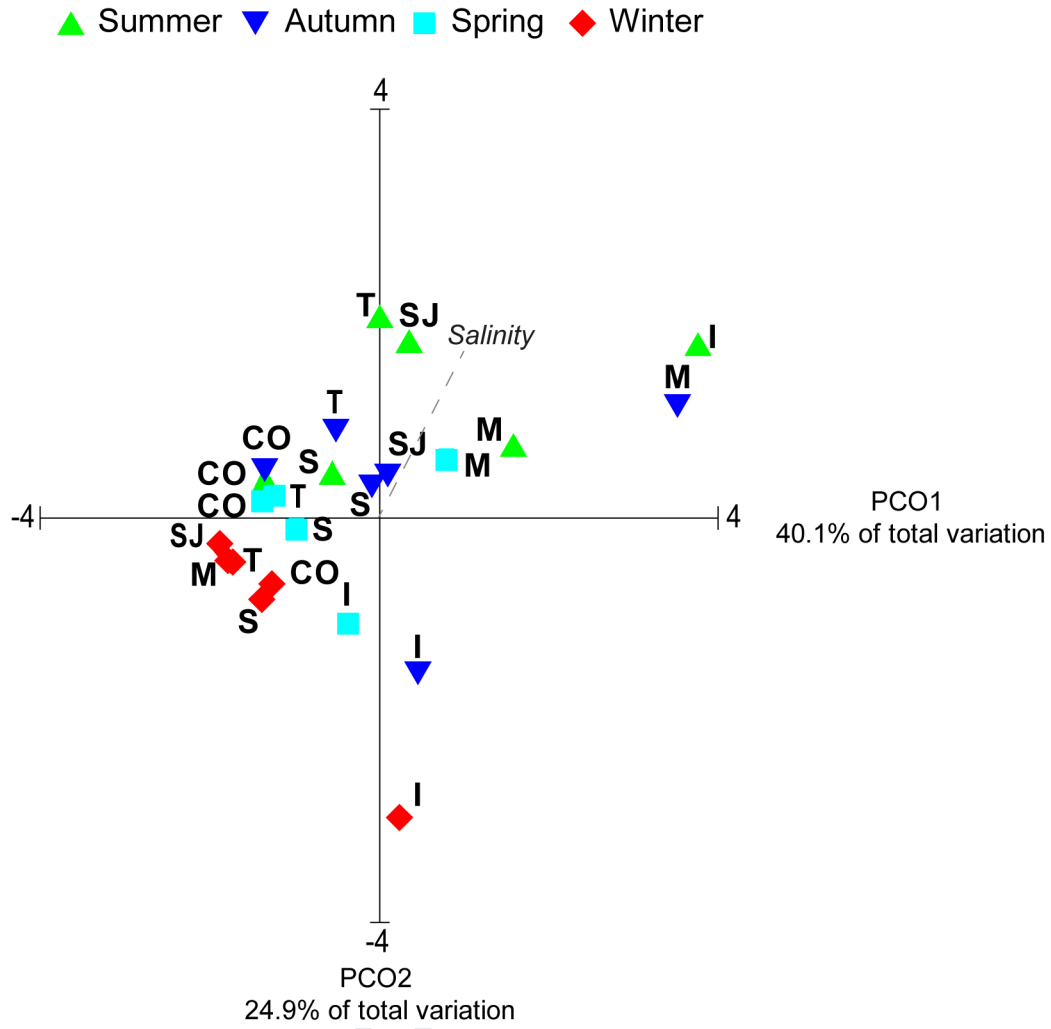


Table 3. Biochemical parameters results for each sampling site (T, Torreira; S, Sporting; SJ, São Jacinto; I, Ílhavo; M, Murtosa; CO, Cale do Ouro) and season (Summer, Autumn, Spring and Winter). The highest values per season are with **bold** and highest levels per year are marked with a \*. Results expressed in the following values: LPO in nmol MDA/g FW; ETS in nmol/ min/g FW; PROT and GLY in mg/ g FW; SOD, CAT, GPx and GRed in U/g FW; GSH and GSSG in  $\mu\text{mol/g FW}$ ; AChE in  $\mu\text{mol/ min/ g FW}$  (FW: fresh weight).

		BIOCHEMICAL PARAMETERS											
	SITE	LPO	ETS	PROT	GLY	GPx	GSTs	SOD	CAT	GSH	GSSG	GSH/GSSG	AChE
SUMMER	T	<b>27.0*</b>	9.15	<b>25.2</b>	2.14	$4.59 \times 10^{-3}$	$7.66 \times 10^{-3}$	$7.86 \times 10^{-2}$	<b>19.9</b>	$5.94 \times 10^{-2}$	$5.56 \times 10^{-2}$	1.13	<b><math>5.53 \times 10^{-1}</math></b>
	S	24.2	11.5	24.0	1.91	$8.41 \times 10^{-3}$	$4.54 \times 10^{-3}$	$8.13 \times 10^{-2}$	14.6	$6.06 \times 10^{-2}$	$5.61 \times 10^{-2}$	1.10	$4.63 \times 10^{-1}$
	SJ	15.2	7.84	24.5	<b>5.31</b>	$1.85 \times 10^{-2}$	<b><math>8.02 \times 10^{-3}</math></b>	$8.70 \times 10^{-2}$	21.4	$6.24 \times 10^{-2}$	$6.10 \times 10^{-2}$	1.06	$6.01 \times 10^{-1}$
	I	25.4	<b>12.5</b>	22.4	1.59	<b><math>1.86 \times 10^{-2}</math></b>	$7.42 \times 10^{-3}$	<b><math>9.84 \times 10^{-2}</math></b>	21.6	<b><math>1.03 \times 10^{-1}</math>*</b>	$5.38 \times 10^{-2}$	<b>2.25</b>	$4.36 \times 10^{-1}$
	M	18.7	8.95	21.4	1.59	$6.15 \times 10^{-3}$	$7.62 \times 10^{-3}$	$7.97 \times 10^{-2}$	17.5	$6.94 \times 10^{-2}$	<b><math>6.74 \times 10^{-2}</math></b>	1.05	$5.37 \times 10^{-1}$
	CO	19.5	10.7	22.4	1.59	$8.93 \times 10^{-3}$	$5.01 \times 10^{-3}$	$7.74 \times 10^{-2}$	9.57	$7.10 \times 10^{-2}$	$6.02 \times 10^{-2}$	1.20	$4.87 \times 10^{-1}$
AUTUMN	T	<b>16.3</b>	55.0	19.2	8.14	$1.02 \times 10^{-2}$	$9.84 \times 10^{-2}$	1.44	15.8	$4.07 \times 10^{-2}$	$5.48 \times 10^{-2}$	<b><math>7.65 \times 10^{-1}</math></b>	$2.38 \times 10^{-1}$
	S	12.8	55.1	<b>25.1</b>	<b>11.3</b>	$6.69 \times 10^{-3}$	<b><math>2.12 \times 10^{-1}</math></b>	<b>3.69*</b>	26.7	$3.81 \times 10^{-2}$	$5.40 \times 10^{-2}$	$7.10 \times 10^{-1}$	<b><math>3.44 \times 10^{-1}</math></b>
	SJ	9.14	49.1	15.1	9.14	$9.29 \times 10^{-3}$	$1.19 \times 10^{-1}$	1.84	22.6	$3.62 \times 10^{-2}$	<b><math>6.60 \times 10^{-2}</math></b>	$5.63 \times 10^{-1}$	$2.21 \times 10^{-1}$
	I	13.8	54.7	19.1	7.88	$5.95 \times 10^{-2}$	$1.52 \times 10^{-1}$	2.52	21.9	$3.99 \times 10^{-2}$	$5.38 \times 10^{-2}$	$7.50 \times 10^{-1}$	$3.38 \times 10^{-1}$
	M	14.5	<b>70.0</b>	15.0	6.02	<b><math>1.17 \times 10^{-2}</math></b>	$1.61 \times 10^{-1}$	3.29	<b>31.8*</b>	<b><math>4.08 \times 10^{-2}</math></b>	$6.37 \times 10^{-2}$	$7.01 \times 10^{-1}$	$2.85 \times 10^{-1}$
	CO	8.62	38.7	16.7	3.83	$9.93 \times 10^{-3}$	$1.36 \times 10^{-1}$	3.43	29.4	$3.65 \times 10^{-2}$	$5.49 \times 10^{-2}$	$6.74 \times 10^{-1}$	$2.43 \times 10^{-1}$
SPRING	T	7.55	<b>95.7*</b>	14.9	<b>15.9*</b>	$8.68 \times 10^{-2}$	$5.81 \times 10^{-1}$	$4.12 \times 10^{-1}$	19.6	$2.52 \times 10^{-2}$	$9.38 \times 10^{-2}$	$1.41 \times 10^{-1}$	$7.54 \times 10^{-1}$
	S	8.07	82.1	10.4	3.89	$8.10 \times 10^{-2}$	<b><math>7.18 \times 10^{-1}</math>*</b>	$3.98 \times 10^{-1}$	21.1	$2.39 \times 10^{-2}$	$7.83 \times 10^{-2}$	$1.57 \times 10^{-1}$	$4.36 \times 10^{-1}$
	SJ	<b>14.7</b>	71.4	17.4	7.07	$9.18 \times 10^{-2}$	$4.26 \times 10^{-1}$	$1.34 \times 10^{-1}$	17.7	$2.21 \times 10^{-2}$	$8.58 \times 10^{-2}$	$1.44 \times 10^{-1}$	$5.60 \times 10^{-1}$
	I	10.4	95.3	16.4	7.94	$6.61 \times 10^{-2}$	$6.65 \times 10^{-1}$	$3.80 \times 10^{-1}$	<b>25.9</b>	<b><math>2.68 \times 10^{-2}</math></b>	<b><math>1.08 \times 10^{-1}</math></b>	$1.27 \times 10^{-1}$	<b>1.07</b>
	M	14.2	82.7	<b>20.9</b>	6.44	<b><math>1.14 \times 10^{-1}</math></b>	$4.42 \times 10^{-1}$	<b>1.91</b>	17.8	$2.28 \times 10^{-2}$	$6.73 \times 10^{-2}$	$1.55 \times 10^{-1}$	$4.02 \times 10^{-1}$
	CO	12.3	87.3	16.1	11.0	$8.92 \times 10^{-2}$	$5.88 \times 10^{-1}$	1.43	16.3	$2.47 \times 10^{-2}$	$7.81 \times 10^{-2}$	<b><math>1.69 \times 10^{-1}</math></b>	$7.22 \times 10^{-1}$
WINTER	T	9.15	<b>70.4</b>	38.1	1.77	$2.70 \times 10^{-1}$	$6.22 \times 10^{-2}$	$1.91 \times 10^{-1}$	7.53	$9.56 \times 10^{-3}$	$4.09 \times 10^{-3}$	1.40	4.39
	S	11.5	63.8	<b>38.9*</b>	1.69	$2.48 \times 10^{-1}$	<b><math>7.99 \times 10^{-2}</math></b>	$1.92 \times 10^{-1}$	7.15	$1.06 \times 10^{-2}$	$7.74 \times 10^{-3}$	<b>2.30*</b>	2.60
	SJ	7.84	27.2	33.1	1.87	$2.67 \times 10^{-1}$	$7.86 \times 10^{-2}$	$1.91 \times 10^{-1}$	4.81	$3.50 \times 10^{-3}$	<b><math>1.39 \times 10^{-2}</math>*</b>	$1.94 \times 10^{-1}$	2.95
	I	<b>12.5</b>	35.1	37.2	1.70	$2.65 \times 10^{-1}$	$7.57 \times 10^{-2}$	$1.91 \times 10^{-1}$	<b>7.78</b>	<b><math>2.15 \times 10^{-2}</math></b>	$8.92 \times 10^{-3}$	1.43	4.20
	M	8.95	51.2	36.0	1.55	<b><math>3.14 \times 10^{-1}</math>*</b>	$7.20 \times 10^{-2}$	$1.92 \times 10^{-1}$	6.70	$9.17 \times 10^{-3}$	$6.52 \times 10^{-3}$	$7.68 \times 10^{-1}$	<b>4.60*</b>
	CO	10.7	38.3	34.4	<b>2.63</b>	$2.68 \times 10^{-1}$	$6.24 \times 10^{-2}$	<b><math>1.93 \times 10^{-1}</math></b>	4.74	$7.47 \times 10^{-3}$	$9.90 \times 10^{-3}$	$3.85 \times 10^{-1}$	3.61

▲ Summer ▼ Autumn ■ Spring ◆ Winter





- Seasonal changes overlaps pollution levels effects on clam's biochemical machinery.
- Cu, Cr and As are the elements with higher concentrations in sediments, water and tissues, respectively.
- As is the element of most concern in terms of human health, with values as low as 0.05 mg to exceed PTWI
- Higher LPO levels and GSH/GSSG ratio during Summer and Spring;
- Membranes' integrity prevention during Winter and Autumn due to antioxidant defences.

### **Conflict of Interest**

The Authors whose names are listed immediately below certify that they have NO affiliations with or involvement in any organization or entity with any financial interest (such as honoraria; educational grants; participation in speakers' bureaus; membership, employment, consultancies, stock ownership, or other equity interest; and expert testimony or patent-licensing arrangements), or non-financial interest (such as personal or professional relationships, affiliations, knowledge or beliefs) in the subject matter or materials discussed in this manuscript.

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**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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