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HEAT THANSFER COEFFICIENTS

OF

NON-NETTONIAN SLURRIES

IN

LAWINAR FLOW

À THESIS

SUBMITTED TO THE FACULIY

OF

THE DEPARTMENT OF CHELFCAL ENGINEERING

\mathbf{OF}

NEWARK COLLEGE OF ENGINEERING

by

ROBERT F. ROTH, B.S.Ch.E.

and

DAVID B. SMANSON, B.S. Ch.E.

IN PARTIAL FULFILLMENT OF THE REQUEREMENTS FOR THE DEGREE

OF MASTER OF SCHENCE IN CHECTCAL ENGINEERING.

HEMARK, NEW JERSEY

SEPTEMBER, 1961

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FOR

DEPARTMENT OF CHEMICAL ENGINEERING

MEWARK COLLEGE OF INGINEERING

BY

FACULTY COMMITTEE

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NEWARK, NEW JERSEY

SEPTEMBER, 1961

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ABSTRACT

A total of 51 heat transfer runs were made. In the first 25 runs, slurries of half micron and finer kaolin at 3.0, 5.6, 9.6 and 12.9% solids were used. Slurries of two microns and finer kaolin at 10.1, 13.7, 20.h and 23.1% solids were used in the remaining 26 runs.

The Reynolds Numbers ranged from 0.9 to 2,020 for the 51 runs.

The data was correlated by the following equation:

$$\frac{h_f}{c_b v_b r_b} \qquad \frac{c_b \mu_b}{r_b} \qquad = 0.70 \qquad \frac{D v_b r_b}{\mu_b} \qquad -2/3$$

for an L/p ratio of 68.7.

This is essentially the accepted correlation for Newtonian fluids in laminar flow. The viscosity used for the slurries investigated was the apparent bulk viscosity obtained from measurements with a Fann V-O viscometer which gave shear rate versus viscosity data. The viscosity used was that at the same shear rate present in the heat exchanger.

INTRODUCTION

Since non-Newtonian slurries are becoming increasingly important systems in the chemical industries heat transfer data for such fluids are a growing necessity. Because most of the work done to date has been for non-Newtonian fluids in turbulent flow, this thesis deals with heat transfer coefficients in laminar flow.

One purpose of this research was to determine heat transfer coefficients for kaolin-water slurries in laminar flow. The clay industry is currently faced with great demands for products which will serve many new and diversified uses. These new demands mean modification of existing products by methods and processes which until now were unrelated to the industry. Consequently, a great lack of technological information on clay and clay-water systems exists. Heat transfer coefficients are engineering tools which will almost certainly be in demand in the very near future.

Another purpose of this research was to determine a correlation for heat transfer coefficients of non-Newtonian slurries in laminar flow and compare it with those correlations already established for Newtonian fluids. If the Newtonian correlations apply to non-Newtonian fluids or apply under certain circumstances, much difficult, complicated, and expensive measuring would be eliminated. Metzner, Vaughn, and Houghton have presented some data which suggests this is the case for solutions of Carbopol in water and carboxymethylcellulose in water. It was thought that data on suspensions as well as on solutions would be required before any general conclusions could be considered.

THFORY

In recent years heat transfer data have frequently been gathered and correlated for non-Newtonian slurries flowing in the turbulent regime. In many instances the approach has been to compare these data with the classical correlations for Newtonian fluids. By changing coefficients and exponents and adding new dimensionless groups, correlations have been arrived at which are similar in form to those for Newtonian fluids. However, little work has been done on non-Newtonian slurries in the laminar regime. It is, therefore, a purpose of this thesis to determine the applicability of correlations for Newtonian slurries in laminar flow.

Non-Newtonian fluids are defined as all fluids which do not exhibit a direct proportionality between shear stress and shear rate. Shear rate is the velocity gradient established between two planes within the fluid a distance (dr) apart when one of the planes moves with a velocity of (du) with respect to the other. The shear stress is the force required to move the plane at velocity (du) divided by the area of the plane. Mirich (5) in his three volumes on Rheology has compiled practical and theoretical aspects of many non-Newtonian fluids and processes involving non-Newtonian fluids of widespread importance.

Host rheologists (17) (6) (20) agree on a method of classification of non-Newtonian fluids which includes three major categories:

a. Time independent non-dewtonian fluids,

b. Time dependent non-Newtonian fluids,

c. Biscoelastic fluids.

Time independent non-Newtonian fluids are those for which the shear stress remains constant at a given shear rate regardless of the length of time for which the shear is applied. The shear stress of time dependent non-Newtonian fluids varies with time at a given shear rate until it too becomes constant. Some rheologists (4) place both these types of fluids in the same category, one being merely an extension of the other into the realm of measurable time. Viscoelastic fluids are those possessing viscose as well as elastic properties.

Several types of fluids are members of the time independent group. The most obvious are Newtonian fluids which can be considered special cases among the range of non-Newtonian fluids.

The simplest type of non-bewtonian fluids, in theory at any rate, is the Bingham plastic. Although several materials have been classified as Bingham plastics Otto and Setzner (14) as well as Both and Bich (18) have demonstrated that most of them deviate from the direct proportionality between shear stress and shear rate when measurements are taken over wide ranges of shear rate. Such evidence questions the reality of any true Bingham plastics if shear rate range were extended even further.

Another type of time independent non-Newtonian fluids is dilatant fluids. With these fluids the ratio of shear stress to shear rate increases with increasing shear rate. When the flow rate of these materials is increased, their resistance to flow becomes greater. Consequently dilatant materials are seldom used in process equipment.

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The most important type of time independent non-Newtonian fluids is the pseudoplastic fluids. These fluids are the opposite fluids in that the ratio of shear stress to shear rate decreases with increasing shear rate. The logarithmic plot of shear stress and shear rate is often a straight line over moderate ranges of shear rate. Therefore, Ostwald (13) proposed the following approximate relationship which was elaborated upon by Beiner (17):

 $r = k \dot{\sigma}^{h}$ Equation (1)

where (k) and (n) are constants for the particular fluid, and (n) lies between 0 and 1. Since the ratio of shear stress to shear rate is defined as apparent viscosity,

 $H = k \dot{\sigma}^{\prime}$

Therefore, the logarithmic plot of apparent viscosity and shear rate should be a straight line with a negative slope from 0 to -1.

Hooney (11) in his classic paper on explicit formulas for slip and fluidity defines fluidity as the ratio of shear rate to shear stress. Fore simply it is the reciprocal of apparent viscosity. Hooney further defines the mean fluidity of the fluid in capillary viscometers, and, it may be assumed, in pipelines which there is laminar flow, as $(h_{\rm ex}/\pi a^3)$ divided by the shear stress at the wall of the tube. In this relationship (E_x) is the efflux due to fluidity in units of volume per time and (a) is the radius of the tube in units of distance. $(h_{\rm ex}/\pi a^3)$ can easily be converted to engineering terms of (8V/D) where (V) is the bulk velocity and (D) the tube diameter. Although Equation (2) is an approximate equation in that (n) is not a true constant, for pseudoplastic materials (n) can only vary between 0 and 1. Consequently, there can only be one apparent viscosity for a given shear rate. Therefore, the mean fluidity, or inversely the mean apparent viscosity, of a fluid flowing through a tube in laminar flow can be found by any device which will give an apparent viscosity for a shear rate numerically equal to (8V/D).

The theoretical relation for heat transfer based on parabolic distribution of mass velocity and assuming uniform radial conduction of heat is given by the analytical equation (8).

$$\frac{h_{a}D}{K_{b}} = \frac{2wC_{pb}}{\pi K_{b}L} \frac{1-8\Psi(n_{l})}{1+8\Psi(n_{l})}$$
 Equation (3)

In this equation

$$\Psi(\mathbf{n}) = 0.10238 \frac{-14.627n}{+0.01220} + 0.01220 \frac{-89.22n}{+0.00237} + \dots$$
 Squation (14)

in which (n) represents the group mKbb/lwCob.

Equation (3) is extremely unwieldy and for values of wCpb/KbL greater than 10 it is closely approximated by the empirical expression

$$\frac{h_{a}D}{K_{b}} = 1.75 \left(\frac{wC_{b}}{K_{b}L}\right)^{1/3} = 1.62 \left(\frac{hwC_{b}}{\pi_{b}L}\right)^{1/3}$$
 Legation (5)

Fluids become non-Newtonian when their viscosity varies with shear rate or in the case of pipelines with mass velocity. It is conceivable that the property of viscosity may influence heat transfer to non-Newtonian fluids. Equation (5) contains no viscosity term. However, when the right side of the equation is multiplied by $(DH_b/DH_b)^{1/3}$ it may be written as

$$\frac{h_{a}D}{K_{b}} = 1.62 \left[\left(\frac{C_{b}\mu}{K} \right)_{b} \left(\frac{D}{L} \right) \left(\frac{DR_{b}V}{\mu_{b}} \right) \right]^{1/3}$$
 Squation (6)

For convenience this equation may be expressed in terms of the Keynold's number by dividing both sides of the equation by $(c_p \varkappa/\kappa)_b^{-1/3} (0 e_b \nu/\kappa_b)$ and rearranging. The resulting equation is

Any heat transfer data can be plotted with the Deynold's number as the abscissa and the groups $(h/C_{pb}V\rho_b)(C_p H/K)_b^{2/3}$ as the ordinate and compared with Equation (7).

LITERATURE SEARCH

A search was made of the available literature to determine progress already made by other investigators. It was noted that heat transfer to non-Newtonian fluids has only been studied for the past decade or two. However, non-Newtonian fluids as such have been under consideration since the 1920's.

Several methods of evaluating viscosities for heat transfer data have been used. Miller (10), Salamone and Newman (19), and Binder and Pollara (1) placed a pipe of dimensions identical to their heat exchanger in series with the exchanger. Pressure taps were installed in both ends of the pipe. The fluid leaving the heat exchanger was cooled to the average temperature of the fluid in the exchanger before entering the pipe. A curve relating the Reynolds number with the friction factor was made for this apparatus by calibrating it with a Newtonian fluid. When the non-Newtonian fluid was introduced to the system the pressure drop across the pipe was measured. This permitted the friction factor to be calculated and the corresponding Leynolds number to be determined. The only unknown in the Reynolds number was the viscosity which could then be easily calculated. This system has the advantage of measuring the fluid at exactly the same conditions of flow and temperature as exist in the heat exchanger. It also involves simple measurements and quick calculations. However, it is not too applicable to laminar flow. With flow rates low enough to insure laminar flow and pine diameters large enough to insure a substantial heat transfer area the pressure drop across the viscometer section of the equipment is often too small to be measured.

Several investigators (2) (25) (10) (12) have used laboratory viscometers for the evaluation of the viscosities of the fluids in their heat transfer experiments. Of the types used the most prevalent is the capillary variety. One of the more elaborate capillary viscometers is the one designed by Orr and DallaValle (12). It was especially designed for susnensions of solids in liquids. It consists of a constantly agitated suspension reservoir in an oil bath. A capillary tube leads from the bottom of the reservoir to a receiving vessel upon which varying degrees of vacuum can be applied. The heating oil is caused to flow downward through an annulus surrounding most of the length of the capillary and upward through a return line to the main oil bath. Since the fluid is agitated up until the time it enters the capillary settling is minimized. This apparatus has the disadvantage of being limited to a maximum pressure differential of 14.7 psi. If this is not enough to cover the desired range of shear rates the diameter of the capillary must be changed. Changing the capillary tube is a cumbersome and time consuming task. A further disadvantage is the fact the fluid reservoir is open to the atmosphere. Gost heat transfer experiments employ steam as the heat giving medium. Further, most suspensions of interest are those of solids in water. If the viscosity of the fluid in question at the temperature of the heat exchanger wall were attempted to be measured in this equipment, rapid vaporization or boiling would occur in the fluid reservoir thus changing the percent solids. Boiling would also occur in the capillary causing three phase flow. Therefore, the Sieder-Tate (21) correction factor of $(\mathcal{H}/\mathcal{H}_W)^{\circ.14}$ cannot be applied to heat transfer data obtained by experiments in which this method of viscosity evaluation is employed. This may not be serious

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for often this correction lies within experimental error.

Metzner et al. (9) employed a capillary viscometer in which the fluid reservoir consisted of a closed vessel to which a pressure could be supplied. This eliminates the disadvantages of the Orr-DallaValle viscometer but provides no means of preventing settling of suspensions.

Among the many rotational viscometers a particularly suitable one for relatively low shear rates is the Fann V-G. 1t consists of a bob suspended by a torsion spring around which a cylinder rotates. The cylinder speed and consequently the shear rate can be switched instantaneously to any one of six different settings. The shear rate range from the lowest to the highest setting varies a hundred fold. Beadings are almost instantaneous and are given directly in units of shear stress. The fluid reservoir in this apparatus, as indeed in all rotational viscometers, is open to the atmosphere. Therefore, the Sieder-Tate correction factor must again be neglected in experiments employing this type of viscometer. Lirich (5), Wilkinson (24), Green (6), and others describe many other types of viscometers which might be useful to a particular system.

Metzner et al. (9) presented experimental data on heat transfer coefficients of non-Mewtonian fluids in laminar flow and compared them with the following theoretical relationships which they had derived:

$$\frac{hD}{K} = 1.75 \delta^{1/3} \left(\frac{wC_p}{KL}\right)^{1/3}$$
Equation (8)
$$\frac{hD}{K} = 1.75 \Delta^{1/3} \left(\frac{wC_p}{KL}\right)^{1/3}$$
Equation (9)

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The choice between Equation (8) and Equation (9) depends on the range of Graetz numbers and flow behavior indices used. (The flow behavior index is approximately equal to (n) in Equation (1). The expression for (δ) and (Δ) were derived analytically, and the approach was based on the theoretical work of Pigford (16). It should be mentioned that, although the data agroed reasonably well with Equations (8) and (9), the agreement was no better than with the theoretical relationship for heat transfer coefficients of Newtonian fluids,

$$\frac{hD}{K} = 1.75 \left(\frac{wC_p}{KL}\right)^{1/3}$$
 (quation (5)

This raises some doubt as to the justification of including the (δ) or (Δ) factor particularly in work of an engineering nature. Netzner et al. brought the data into closer agreement with the theoretical curves by including the Sieder-Tate correction factor. However, this caused most of the data points to fall below the theoretical curve. This was explained by the fact that the input heat flux was used to determine the heat transfer coefficients. It may also be true that the Sieder-Tate correction factor does not apply exactly to non-Newtonian fluids whose viscosities are affected by not only the difference in temperature but also the difference in shear rate encountered at the tube wall.

Martinelli et al. (7) give the following analytical relationship for heating fluids flowing vertically upward and cooling fluids flowing vertically downward in laminar flow:

$$\frac{hD}{K} = 1.75 F_1 \int_{KL}^{3} \frac{wC_D}{KL} + 0.0722 \left[\frac{D}{L} H_{Gr}, 10 P_r \right]_{W}^{n} F_2$$

Equation (10)

The F factors and the quantity 0.0722 $[1/6, 1]_{6}$, $[1N_6]_{W}^n$ are included for natural convection. The material closest to the well of the tube has a different temperature than the bulk of the fluid and consequently has a different density. This difference in density creates a buoyant force which increases the velocity of the fluid at the well. However, Scadars (5) suggests that with fluids of high kinematic viscosities flowing in laminar flow through pipes of small diameter and having moderate temperature potentials the effect of natural convection becomes almost perhipible command to that of forced convection. Shen the effect of natural convection is neglected equation (10) reduces to equation (7).

Thermal conductivities of suspensions have been of considerable interest for several years. Targef (22) reasoned that the problem is exactly the same as that for electrical conductivity. He therefore wrote the thermal analog for Carmell's equation for electrical conductivities as

where F_b , F_l and F_p are the thermal conductivities of the suspension, liquid and solid particles respectively and X_v is the volume fraction of colids. Be checked this equation with previously reported experimental data and found good agreement. Orr and allavalle (12) compared the equation with other experimental data and found excellent agreement. Orr and ballevalle

also measured the thermal conductivity of various accordions. Their method of measurement was unique in that they added to the suspension to be measured two percent by weight agar gel in order to prevent settling during the tests. The data obtained by this method agreed with Equation (11) to within three percent.

Tsao (23) recently published an analytical approach to thermal conductivities of two phase materials. His final equation, however, is quite cumbersome and requires some experimental measurements. The value of this approach seems questionable in the light of the results of Tareef and Orr and DallaValle.

Description Of Apparatus

Figure I is a diagram of the apparatus. The details for each piece of equipment referred to on the diagram by a number are as follows:

1. Twenty gallon slurry storage tank.

2. Submerged type $3/8^n$ vane pump with neoprene impeller driven by a 1/4 HP motor at 1.725 RPM.

3. Experimental Heat Exchanger - Double annulus heat exchanger consisting of a 7' long, 1" schedule h0 red brass pipe inside a 6' long, 2", schedule h0 steel pipe and both inside a 6' long, 3", schedule h0 steel pipe. The inner annulus between the brass pipe and the 2" steel pipe was sealed by a packing box at each end. The boxes were made from 1 1/2" steel pipe couplings, 1 1/h" x 1 1/2" bushings and asbestos graphite packing. The outer annulus between the 2" and 3" steel pipes was sealed by welding the ends of the pipes together. A 3/h" steel pipe coupling served as the steam inlet to both jackets. A 1/2" steel pipe coupling was welded to both the 2" steel pipe and the 3" steel pipe on the other end of the heat exchanger to serve as condensate outlets from each jacket.

Six iron-constantan thermocouples (#24 wire), placed a foot apart and displaced clockwise 60° , were soldered halfway into the wall of the brass pipe to measure the temperature along the entire length of the pipe. The thermocouple wires, insulated with teflon, were brought out to the ends of the brass pipe (three at each end) by cementing them into a $3/16^{\circ}$ wide x 0.133" deep x 2 1/2! long machine groove with litharge and glycerin. The heat exchanger was well insulated to minimize heat losses.

h. Steam Calorimeter - Throttling type steam calorimeter to measure the quality of the steam going into the heat exchanger. An aspirator with a water cooled condenser maintained a partial vacuum on the low pressure side of the calorimeter.
5. Coolers - An annulus-type cooler consisting of a 9' long,
1" schedule h0 stainless steel pipe inside a 8' long, 1 1/2" schedule h0 steel pipe. The jacket was sealed by welding the ends of the outer pipe to the inner pipe. A 1/2" steel pipe coupling was welded at each end of the 1 1/2" pipe to serve as connections to the jacket.

6. Scale - Scale to weigh condensate from heat exchanger.
7. Two 1/2" bucket type steam traps.

8. Condensate Gooler - An annulus type cooler consisting of a 24" long, 1/2" schedule 40 steel pipe inside a 22" long, 1" schedule 40 steel pipe. The jacket was sealed by welding the ends of the outer pipe to the inner pipe. A 1/2" steel pipe coupling was welded at each end of the outer pipe to serve as connections to the jacket.

9. Thermometer - 0° to 100° C with 0.1° C subdivisions.

10. Slurry Mixing Valve.

11. Valve - Valve to control flow rate of slurry through apparatus.

The heat exchanger and the coolers were mounted vertically around and clamped to a 2" support pipe which was held firm by being fastened to the floor and the ceiling. The slurry flowed upward through the heat exchanger.

The thermocouples were connected to a multipoint switch which was connected to a potentiometer.

A Fann rotational viscometer was used to measure the viscosities of the slurries. The viscometer consisted of a rotating cylinder around a bob supported by a torsion wire. The cylinder radius was 93/128 in. The bob had a radius of 87/128 in. and a height of 1.5 in. The torsion wire was connected mechanically to the instrument dial which read directly in pounds per 100 square feet. The cylinder rotated at speeds of 3, 6, 100, 200, 300 and 600 RFC and could be changed instantaneously while the instrument was running. Readings at all six speeds could be taken in less than a minute. Because of the speed of reading plus the very fine particles of the slurry, there was little or no settling during viscosity measurement.

Figures 2a and 2b are photographs of the apparatus.

DIAGRAM OF APPARATUS



FIGURE 1

12 . 14



Figure 2a View of apparatus showing heat exchanger, coolers, potentiometer, multipoint thermocouple switch, and slurry storage tank with submerged circulating pump.



Figure 2b View of apparatus showing heat exchanger, coolers, steam calorimeter, condensate cooler, and slurry storage tank with submerged circulating pump.

Experimental Procedure

About a ten gallon batch of kaolin slurry was prepared by mixing the dry powder with water. The freshly prepared batch was made with the highest percentage of solids that was to be used. As experiments were made with the apparatus, the slurry was diluted with water to lower the solids content.

The apparatus was put into operation for collecting heat transfer data by starting the slurry circulating pump, turning on the tap water to all the coolers, turning on the steam and adjusting the steam pressure to about 10 to 15 psig, turning on the steam throttling calorimeter, connecting the potentiometer to the thermocouple switch, immersing the reference thermocouple junction into a dewar flask containing melting ice, and removing air from the slurry by opening the air vent valves.

The apparatus was allowed to run for a while at a high slurry flow rate (Reynolds number about 2,000) to let the temperatures of the slurry going into and coming out of the heat exchanger become constant. At this point, the apparatus was ready for taking data. The data that was taken is as follows: barometric pressure, temperature of the slurry going into and coming out of the heat exchanger, the six readings in millivolts of the thermocouples cemented into the brass pipe wall of the heat exchanger, temperatures and vacuum of the throttling calorimeter, heat exchanger condensate rate by taking a two, three, or four minute weight increment, steam inlet pressure to the heat exchanger, and rate of slurry flowing through the apparatus by taking a one-half, one, or two minute weight increment. After the data for a run was taken, the slurry flow rate was reduced by about 15 to 20% by adjusting the two slurry rate control valves. Again the apparates was allowed to run for a while, letting the temperatures of the slurry going into and coming out of the heat exchanger become constant. At this point, data for another run was taken. This procedure was followed for about six or seven runs, then the concentration of solids in the slurry was reduced by removing about 15 to 30% of the slurry from the batch and replacing it with water. After the slurry was thoroughly mixed, the procedure, outlined above, was repeated.

The apparatus was shut down by first turning off the steam, turning off the water to the coolers, stopping the slurry circulating pumps, and disconnecting the potentiometer.

A sample was taken from each batch of slurry. After each sample was checked for percent solids, viscosity runs were made with a Fann viscometer. Viscometer speeds of 3, 6, 100, 200, 300 and 600 rpm were made on the slurry at the average temperature the slurry was at in the heat exchanger. The shear stress at each of these viscometer speeds was recorded.

A total of 51 runs were made with the apparatus. A 1/2 micron and finer kaolin was used in the first 25 runs. The remaining 26 runs were made with a 2 micron and finer kaolin.

Table I is a tabulation of the original data. Figure 3 is a plot of slurry density versus percent kaolin in water. Figure 4 is a plot of degrees fahrenheit versus millivolts for iron-constantan thermocouple.

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Figure 5 is a particle size distribution curve of 1/2 micron and finer kaolin. Figure 6 is a particle size distribution curve of 2 micron and finer kaolin.

TABLE I

TABULATION OF ORIGINAL DATA

· · · · ·	Run No. 1	Run No. 2	Run No. 3	Run No. 4	Run No. 5	Run No. 6	Run No. 7	Run No. 8	Run No. 9	Run Nc
Fluid Inlet Temp. N.V.	1.48	1.50	1.16	1.17	1.17	1.47	1.43	1.19	1.39	1.18
Ex. Wall Temp. #2 N.V.	6.02	5.97	6.00	° 6.01	6.09	5.75	5.78	5.93	5.92	5 . 96
Ex. Wall Temp. #3 M.V.	6.35	6.19	6.30	6.35	6.39	5.39	5.85	6.12	6.13	6.13
Ex. Wall Temp. #4 N.V.	6.30	6.18	6.15	6.10	6.30	593	5.94	6.11	6.10	6.15
Ex. Wall Temp. #5 N.V.	6.03	5.92	6.02	6.10	6.10	i. T	5.72	5.78	5.85	5.89
Ex. Wall Temp. #6 M.V.	6.30	6.20	6.28	6.35	6.31	5.93	5.93	6.10	6.11	6.15
Ex. Wall Temp. #7 H.V.	6.37	6.23	6.32	6.35	6.37	5.98	5.93	6.14	6.16	6.15
Fluid Outlet Temp. #8 N.V.	1.90	1.90	2.00	2.14	2.32	3.12	2.64	1,62	1.81	2.03
Fluid Mass Rate	11 3/8 1b/30 se	c 9 1/4 1b/30 sec	e 7 3/16 1b/30 sec	5 1b/30 sec	3 9/16 1b/30 sec	2 1 1/8 15/30 sec	3 5/16 1b/min 1	2 5/16 1b/30 sec	10 11/16 16/30	sec 8 1/8 1b/
Steam Inlet Pressure, PSIG	14.0	11.6	12.7	1.3.7	13.8	7.7	7.3	9.7	10.5	10.4
Steam Mass Rate	530 grm/L min	351 grm/3min	327 grm/3 min	308 grm/3 min	293 grm/3 min	229grm/l min	259 grm/h min	477 grm/3 min	437 grm/3 min	1 :1:01; grm/3
Orifice Upstream Temp., 'F	279	275	214	246	246	232	232	238	239	238
Orifice Downstream Temp., "F	180	175	177	178	177	168	1.66	1. 71	171	1.70
Orifice Downstream	15.1	17.1	16.4	15.8	15.7	19.0	19.7	18.3	time to	18.2
Solids	12.9	12.9	12.9	12.9	12.9	12.9	12.9	9.6	9.6	9.6
Barometric Pressure, In. Hg.	29.65	29.65	29.65	29.65	29.65	29.65	29.65	, 29.65	29.65	29.69
Avg. Temp. OF	90.5	90.8	91.6	94.3	90.L	115.4	102.2	2. 28	89.1	92.6
16/100 ft ² @ 3 RPM	13.5	13.5	13.5	13.2	13.5	12.5	13.5	7.5	6.7	6.7
16/100 ft ² @ 6 RPM	17.0	17.0	17.0	17.2	17.0	16.5	18.0	8.5	7.5	7.5
16/100 rt2 @ 100 RPM	26.7	26.7	26.7	27.5	26.7	27.2	27.6		11.5	11.5
16/100 ft ² @ 200 RPM	29.8	29.8	29.8	30.0	29.8	29.7	30.0	and a w	13.2	13.2
16/100 ft ² @ 300 RPM	31.5	31.5	31.5	31.5	Jan a ju	30.5	32.5	16.2	14.7	
16/100 ft2 @ 600 RFM	36.5	36.5	36.5	36.2	36.5	35.2	36.0	19.8	18.5	18.5

	Run No. 8	Run No. 9	Run No. 10	Ran No. 11	Run No. 12	Run Nos. 13	Run No. 14	Bun No. 15	nun No. 16	Run No. 17	Run No. 18	Run No. 19
	1.19	1.39	1.18	1.18	1.58	1.50	1.18	1.13	1.16	1.55	1.58	1.58
	593	5.92	5.96	5.93	5.83	5.89	5.82	5.91	5.90	5.95	5.85	5.99
	6.12	6.13	6.13	6.15	6.00	6.12	6.04	6.09	6.11	6.10	5.98	6.20
	6.11	6.10	6.15	6.14	6.01	6.08	6.00	6.10	6.08	6.11	6.00	6.18
	5.78	5.85	5.89	5.89	5.65	5.90	5.82		5.80	5.78	5.70	6.00
	6.10	6.11	6.15	6.13	6.00	6.08	6.00	6.11	6.10	6.11	5.98	6.16
	6.14	6.16	6.15	6.18	6.00	6.08	6.00	6.11	6.10	6.11	6.10	6.20
	1.62	1.81	2.03	2.19	2.1.	1.10	3.38	1.62	2.02	2.33	2.61	4.30
ln 1:	2 5/16 1b/30 sec	10 11/16 15/30 s	c 8 1/8 1b/30 sec	6 1/8 1b/30 sec	7 3/8 1b/min	1 3/4 1b/min	2 1/2 16/min	12 1/2 1b/30 sec	10 1/h 1b/30 sec	7 1b/30 sec	1/8 1b/30 sec	2 1/8 1b/65 sec
	9.7	10.5	10.4	10.6	8.2	93	9.1	10.0	10.6	10.2	8.2	10.8
ti.n	477 grm/3 min	137 grm/3 min	LOL grm/3 min	368 grm/3 min	313 grm/3 min	209 grm/3 min	21,8 grm/3 min	622 grm/3 min	566 grm/3 min	513 grm/3 min	394 grm/3 min	246 grm/3 min
	238	239	238	210	236	236	235	238	239	239	234	210
	1.71	171	1.70	1.71	169	169	169	1.69	170	1.70	165	. 170
	18.8		18.2	18.1	18.8	18.8	18.9	18.8	18.5	18.5	20.0	18.5
	9.6	9.6	9.6	9.6	9.6	9.6	9.6	5.6	5.6	5.6	5.6	5.6
	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65
	82.1	89.1	92.6	95.L	100.0	128.0	119.2	82,5	93.4	100.2	104.0	132.7
	7.5	6.7	6.7	6.7	7.0	7.2	6.8	1.5	2.0	2.0	2.0	
	8.5	7.5	7.5	7.5	8.0	6.5	0,8	2.0	2.2	2.0	2.0	2 + 2 2 + 2
	11.8	11.5	11.5	11.5	11.5	I.L. a 5	11,2	3*5	3.5	3.5	3.5	3.5
	13.7	13.2	13.2	13.2	13.2	32.5	12.8	5.0	4.8	1.7	1:.7	4.2
	16.2	14.7	14.7	14.7		13.5	13.2	5.5	5.5	5.5	5.5	1.8
	19.8	18.5	18.5	18.5	17.5	16.5	16.5	6.0	7.5	7.2	7.2	6.0

all the second												
Niela Augstei	Run No. 16	Hun No, 17	Run No. 18	Run No. 19	Run No. 20	Run No. 21	Run No. 22	Run No. 23	Run No. 24	Run No. 25	Run No. 26	Run No. 27
	1.46	1.55	1.58	1.58	1.56	1.20	1.41	1,51	1.61	1.60	*35.4 °C	34.7 %
a particular for the second	5.90	5.95	5.85	5.99	5.79	5.80	5.95	5.89	5.98	5.99	5.90	6.05
and the second second	6.11	6.10	5.98	6.20	5.90	6.09	6.12	6.10	6.10	6.18	6,10	6.20
and a state of the	6.08	6.11	6.00	6.18	5.92	6.06	6.10	6.08	6.10	6.17	6.09	6.19
S. C.	5.80	5.78	5.70	6.00	5.67	5.70	5.79	5+77	5.81	5.95	5.90	6.00
and the second second	6.10	6.11	5.98	6.16	5.91	6,07	6.10	6.09	6.10	6.15	6.10	6.18
And and a second second	6.10	6.11	6.10	6.20	5.96	6.07	6.10	6.09	6.10	6.20	6,10	6.21
	2.02	2.33	2.61	4.30	3.28	100	2.34	2.63	3.87	4.70	15.5 °c	52.0 °C
4 10	1/4 1b/30 sec	7 1b/30 sec	4 1/8 1b/30 se	c 2 1/8 1b/65 se	e 1: 7/16 15/min 1	13 5/16 15/30 sec	8 5/16 1b/30 sec	s 7 1/4 1b/30 sec	4 13/16 1b/min	3 15/min	11 1/2 1b/30 sec	: 6 1/16 16/min
and a state of the second	10.6	10.2	8.2	10.8	7.5	10,6	11.0	10.2	10.1	11.1	13.8	12.6
rt 5	66 grm/3 min	513 grm/3 min	394 gra/3 min	21,6 grm/3 min	354 grm/3 min	563 grm/2 min	199 grm/2 min	186 grm/2 min	358 grm/2 min	310 grm/2 min	519 grm/3 min	266 gra/2.5 min
	239	239	234	270	232	239	240	240	855	21,0	243	241
n fin an	170	1.70	165	170	165	170	170	170	166	170	127	126
All and the first states	18.5 11	18.5	20.0	18.5	19.9	10.7	18.5	10.7	19.0	10.1	25.5	25.4
a ni ko errik dita girik	5.6	5.6	5.6	5.6	5.6	3.0	3.0	3.0	3.0	3.0	23.5	23.1
and the second second	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.65	29.54	29.54
and the second second second	93.4	100.2	104.0	1.32.7	114.9	86.9	98.9	102.6	125.7	139.6	104.9	110.3
	2.0	2.0	2,0	1.5	and the second se	0.2	0.2	0.2	0.2	0.2	13.5	13.5
an and an and an an	2.2	2.0	2.0	2.2	2.2	0.5	0.5	0.5	0.5	0.5		17.7
and and the second second	3.5	3.5	3.5	3.5	3.5	the second	1.2	1.2	1.2	1.2	30.1	30,1
A Comment of the same	1.8	4.7	1:.7	1.2	- 5	2.0	1.07	Sandy Sandy Sandy	1.7	a stand	34.0	34.0
and a far far and	5.5	5.5	5.5	1.8	5.0	2.5	2.2	2.2	2.0	2.0	36.9	36.9
nin personance	7.5	7.2	7.2	6.0	6.5	3.8	3.5	3.5	3.0	3.0	ha.9	11.9

* 0° 100°C Thermometer used instead of thermocouple - Runs 26 through Runs 51

TABLE I (cont.) TABULATION OF ORIGINAL DATA

	Fun No. 28	Run No. 29	Run No. 30	Run No. 31	Run No. 32	Run No. 33	Rum No. 34	Run No. 35	Rum No. 36	Run [
Fluid Inlet Temp. N.V.	33.3 °c	32.6 °C	32.9 °C	32.6 %	37.0 %	37.0 °C	36.3 °C	35.1 °c	34.8 00	3/11
Er. Wall Temp. #2 M.V.	6.10	5.83	5.75	5.77	5,81	5.13	5.72	5.83	5.88	5.1
Ex. Wall Temp. #3 M.V.	6.29	5.98	5.90	5.91	6.02	6.00	5.95	6.08	6.10	5.5
Ex. Wall Temp. #4 H.V.	5.25	é.a	5 • F \$	5.90	6.00	5.90	5.90	6.00	6.00	5.1
Ex. Wall Temp. #5 K.V.	6.07	5.77	5.70	5.69	5.80	5.80	5.80	5.00	5.21	5.'
Ex. Wall Temp. #6 M.V.	6.22	6.02	.5.95	5.92	6.00	5.91	5.90	6.00	6.00	5.t
Ex. Wall Temp. #7/M.V.	6.29	6.02	5.95	5.90	5.95	5.89	5.90	6.00	6.00	S.L
Fluid Ontlet Temp. #8 M.V.	60.8 °c	16.5 °C	h4.2 °C	10.0 oc	17.0 °C	49.3 °C	70°0°	52.2 00	57.1 %	
Fluid Mass Bate	10 15/2 min	13 7/6 16/min	9 3/8 1b/30 sec	9 5/8 16/min	ll 7/8 lb/30 sec	8 1/h 1b/30 sec	12 1/2 16/min	9 1/2 1b/min	10 1/2 16/2 min	12 16/30
Steam Inlet Pressure, PSIG	14.0	10.1	9.0	8.5	9.2	6.5	8.0	9.0	9.5	7.6
Stean Mass Hate	21,5 gra/2.5 min	269 gm/2 min	301 gm/2 min	228 gra/2 min	31:2 grm/2 min	273 grm/2 min	215 grm/2 min	222 grm/2 mir	183 grm/2 min	324 Em
Crifice Upstream Temp., °F	239	852	232	231	236	234	233	236	237	230
Crifice Downstream Temp., ^O F	157		1.52	152	152	150	118.5	150	151	1.45
Orifice Downstream	25.5	25.6	22.05	23.0	23.2	23.6	23.9	23.6	23.5	23.
5 Solids	22.1	23.1	23.1	23.1	20.4	20.4	20.4	50 ° f	20.4	- 20.
Barometric Pressure, In. Hg.	29.54	29.5h	29.54	29 . 54	29.60	29.60	29.60	29.60	29,60	29,
Avg. Tesp., °F	11.6.6	103.1	101.13	101.3	107.6	109.6	108.8	209.9	114.6	103.
16/100 rt ² @ 3 RPM	9.0	13.0		13.5	10.0	10.0	10.0	10.0	10.0	10.
16/100 ft ² @ 6 RPM	17.5	13.0	1.7.1	27.7	12.0	12.0	12.0	12.0	12.0	12.
16/100 ft ² @ 100 RPA	29.0	30.0	30.1	30.1		19.4	19.4	19.4	19.2	20.
16/100 18 ² @ 200 EFM	33.0	33.5	34.0	34.0	22.1	22.1	22.1	22.1	22.0	22.
16/100 ft2 @ 300 RFM	35.5	36.2	36.9	36.9	24.I	24.1	24.1	24.1	27 F	
16/100 ct ² © 600 RPM	10.2	112.00	Ja.9	41.9	28.0	28.0	28.0	28.0	27.5	28.

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(cont.)

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								경험 이는 이번 것이라. 영영 영어 생각				See 1
m No., 34	Run Ho. 35	Run No. 36	Run 37	Run No. 38	Run No. 39	Run No. 10	Run No. 11	Run No. 12	. Run No. 13	Run No. 14	Run No. 15	Bo
36.3 °C	35.4 °C	34.8 %	34.8 °c	35.9 °c	33.4 °C	35.5 °c	39.4 °c	10.1 °c	0.0°C	37.8 %	31.0 °c	3
5.72	5.83	5.88	5.65	5.70	5.89	5.83	5.81	5.96	5.88	5.60	5.77	
5.95	6.08	6.10	5.90	5.90	. 6.16	6.00	5.98	6,17	6.17	5.84	5.94	
5.90	6.00	6.00	5.83	5.88	6.08	6.00	6.00	6.14	6.07	5.75	5.96	
5.80	5.88	5.91	5.72	5.71.	5.92	5.81	5.88	6.00	6.02	5.75	5.74	
5.90	6.00	6.00	5.81	5.36	6.07	5.93	5.93	80.0	6.09	5 . 75	5.88	
5.90	6.00	6.00	5.85	5.90	6.10	6.01	6.02	6.20	6.20	5.85	6.00	
9.0 00	51.2 %	57.1 %	ևև.կ °շ	46.5 °C	111.5°C	50.0 °C	51.7 °C	62.3 °C	65.5 °C	52.4 °C	12.1 %	
1/2 16/min	9 1/2 1b/min	10 1/2 16/2 min	12 1b/30 sec	9 7/8 15/30 sec	12 1/1 10/30 sec	2 8 3/8 15/30 se	c 10 1/4 16/30 s	sec 9 lb/min	9 5/8 16/ 1 1/2 mi	n 13 3/8 16/min	12 1/2 1b/30 see	10
8.0	9.0	9.5	7.0		Inter of T	9.1	9.5	11.5	11.5	6.6	8.6	1
5 grm/2 min	222 grm/2 min	183 grm/2 min	321. grm/2 min	299 grm/2 min	209 grm/min	185 grm/min	196 grm/min	166 gra/min	135 Erm/min	299 grm/2 min	446 grm/2 min	1,51
233	236	237	230	232	5110	236	236	217	57ि	230	235	
148.5	150	151	149	148	158		139	1/15	140	1.35	95.5	
23.9 11	23.6	23.5	23.8	24.0	55.0	25.4	25.5	25.2	25.5	26.2	29.4	
20.1	20.4	20.1	20.1	20.1	· · · · · · · · · · · · · · · · · · ·	13.7	13.7	13.7	13.7		10.1	
29.60	29,60	29.60	29,60	29.60	29.93	27.93	29.93	29.93	29.93	29.93	29.98	
108.8	109.9	111.6	103.3	106.1	102.2	109.0	114.1	125.6	128.0	113.1	98.0	
10.0	0.01	10.0	10.0	10.0	and the second sec	1.	4.0	3.8	3.8	<u>1</u> ,.0	1.3	
12.0	12.0	12.0	125	12.3		4.1	4.3	4.2	2.2	and the second sec	4.8	
19.4	19.4	19.2	20.0	19.7	6.2	6.1	6,0	5.2	5.2	6.0	3 a š	
22.3	22.1	22.0	22.3	22.2	in the	7.3	7.2	6.9	6.9	762	and the second se	
24.1	24.1	23.5	24.5	21.3	88	8.3	8.1	7.9	7.9	8.1	19	3
28.0	28.0	27.5	28.7	28.3	10.9	10.3	10.1	9.3	9.3	LO.L	6.8	
			4 mile 2007 SBE 2	da \$54		- 6ar						
in No. 42	Run No. 13	Run No. 44	Run No. 15	Run No. 46	<u>Run No. 17</u>	Run No. 18	Run No. 19	Run No. 50	Run No. 51			
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0.1 00	7° 0.01	37.8 °C	31.0 °c	33.2 °0	34.5 °C	35.9 °c	34.7 °C	34.5 °C	33.7 °C			
5.96	5.88	5.60	5.77	5.70	5. 73	5.72	5.71	5.76	5.13			
5.17	6.17	5.8L	5.94	5.96	6.05	6.03	5.88	5.98	5.95			
,14	6.07	5.75	5.96	5.87	5.89	5.88	5.90	5.95	5.94			
i.00	6.02	5.75	3 . Ma	5.83	5.85	5.85	5.70	5.81	5.74			
5.08	6.09	5.75	5.88	5.90	6.00	6.00	5.00	5.90	5.89			
6.20	6.20 /	5.85	6.00	6.00	6.11	6.08	5.94	6.05	5.99			
2.3 00	65.5 °C	52.4 °C	42.4 °C	46.2 °C	19.7 °c	52.9 %	unsh oc	50.4 °c	43.5 °G			
15/min	9 5/8 16/ 1 1/2 mi	.n 13 3/8 16/min	12 1/2 10/30 sec	10 1b/30 sec	7 1b/30 sec	12 lb/min	11 3/8 1b/30 sec	7 16/30 sec 1	2 1/8 1b/30 sec			
1.5	12.3	6.6	8.6	10.0	10.4	9.6	8.0	9.0	8.1.			
grm/min	135 grm/min	299 grm/2 min	146 grm/2 min	151 grm/2 min	355 grm/2 min	342 grm/2 min	347 grm/2 min	353 grm/2 min	374 grm/2 min			
51jī	211	230	235	236	239	237	233	235	234			
1112	J.110	135	95.5	86	150.5	111	137	1.39	138			
25.2.1	25.5	26.2	29 . 4	29.6	23.6	25.3	25.8	25.6	25.7			
13.7	13.7	13.	10.1	10.1	10.1	10,1	10,1	1.0.1				
29.93	29.93	29.93	29.98	29.98	29.96	29.98	29.98	29.98	29.98			
25.6	128.0	and the second	98.0	103.4	107.8	111.9	103.2	108.4	101.4			
348	3.8	1.0	nee say the	1.3	1. e C	7.2		12	10 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0			
	4.2	4.3	1.8	1.0	1.6	1.5	207	1.6	1.8			
5.2	5.2	6.0	3.2	3.1		3.0	3.1	3.1	J.L			
6.9	6.9	7.2	the second	and the second sec	and the second sec	9	and the second sec	100 C	and the second sec	18 A.		
7.9	7.9	8.1	1.9		1:•6	2.03	2:07	4.6	1.09			
9.3	9.3	lo.l	6.8	6.3	6.1	5.9	6.3	6.1	6.8	* *		



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Experimental Results

Table II is a tabulation of the experimental results calculated from the original data presented in Table I.

The heat balance for 40 of the 51 runs were within 5%. The remaining 11 runs were within 10%. An analysis of the heat balances showed that the amount of heat transferred based on the condensate rate and steam quality are more accurate and more consistent than those based on the temperature rise of the slurry. Therefore, the heat transfer coefficients were based on the heat given up by the steam and not on the temperature rise of the < slurry. (See Discussion of Results)

The six thermocouples, cemented into the brass pipe of the heat exchanger, indicated that the pipe wall temperature was no more than 5° F. below the temperature of the steam at the higher slurry flow rates. At the lower flow rates, the pipe temperature was close to the steam temperature.

Table III is a tabulation of the slurry viscosities at corresponding shear rates calculated from the original Fann viscometer data presented on Table I. Figures 7 through 1h are plots of the data from Table III. The curve or curves on each graph are for only one percentage concentration of solids. In most cases, one viscosity curve is used for more than one heat transfer run. A small variation in average bulk temperature had very little effect on the viscosity and, therefore, can be represented by one curve. Hore than one curve on a graph is due to a larger variation in average bulk temperature. These curves show the non-Newtonian behavior of the slurries. A Newtonian fluid would be represented by a horizontal line. The bulk viscosity of the slurries flowing through the heat exchanger were determined by calculating the shear rate and finding the corresponding viscosity from the viscosity curves. The shear rate was calculated from the following equation (11):

$$\dot{\sigma} = \frac{\delta V_b}{D}$$
 Equation (12)

The viscosity determined by this method was used to correlate the heat transfer data.

The Reynolds numbers ranged from 0.9 to 2020 for the 51 runs.

The thermal conductivity of the slurries were calculated using the following equation (22):

$$K_{b} = K_{1} \left[\frac{2K_{1} + K_{p} - 2 Xv (K_{1} - K_{p})}{2K_{1} + K_{p} + Xv (K_{1} - K_{p})} \right]$$
 Equation (11)

Where: Xv = fraction of solids by volume

- Kb = thermal conductivity of slurry
- K1 = thermal conductivity of liquid
- K_D = thermal conductivity of solids

The thermal conductivity of the slurries did not vary to any great extent with change in solids content and were within 10% of the value for water.

The effect of natural convection on the heat transfer coefficients was investigated by using the following equation (7):

$$\frac{hD}{k} = 1.75 F_1 \left[\frac{WC}{kL} + 0.0722 \left(\frac{D}{L} - R_{Gr} R_{Pr} \right)^{0.75} F_2 \right]^{1/3}$$

iquation (10)

Where: F_1 and F_2 are dimensionless factors that can vary between 0 and 1.

N_{Gr} is the Grashof number - $\left(\frac{D^3 q^2 E}{A^2}\right)$ (BAT) N_{Pr} is the Grandtl number - $\left(\frac{C A}{E}\right)$

The empirical equation (8)

$$\frac{hD}{k} = 1.75 \qquad \left(\frac{wC}{kL}\right) \qquad \qquad \text{Equation (5)}$$

is a standard equation that is used to calculate heat transfer coefficients in the laminar region. The additional term in equation (10), not included in equation (5), represents the correction for natural convection. Using equation (10), the correction for natural convection for 33 runs was less than 1%, for 11 runs it was less than 2% and for the remaining 7 runs it was less than 5%. The effect of natural convection was within experimental error and, therefore, could be disregarded.

Equations (5) and (10) were not used to correlate the heat transfer data because it was desired to use an equation having the Reynolds number as one of the dimensionless groups. Correlating the data with this type of equation would make it possible to put the Reynolds number on the abscissa of the graph which would easily show the range of the laminar region in which the work was done.

The following equation was used to correlate the heat transfer data:

There: d and e are constants determined from the plot of the data.

Figure 15 is a plot of the heat transfer data for all 51 runs. From this plot, constants d and e were found to be 0.70 and - 2/3 respectively. The final heat transfer equation developed from this work is as follows:

$$\left(\frac{h_{f}}{C_{b}V_{b}\rho_{b}}\right) \left(\frac{C_{b}\mu_{b}}{K_{b}}\right) \stackrel{2/3}{= 0.70} \left(\frac{DV_{b}\rho_{b}}{\mu_{b}}\right)^{-2/3}$$
 Equation (14)

for an L ratio of 68.7.

TABLE II

Run No. 10 No. 8 No. 9 No. 12 No. 11 No. 13 No. No. 1 No. 2 No. 3 No. 4 No. 5 No. 6 No. 7 11,80 1280 976 739 443 104 w = 1b/hr 1365 1110 863 600 426 135 199 50 - P 101.9 107.5 88.5 94.5 116.0 172.0 11 97.6 97.6 101.0 105.8 112.0 141.4 123.0 t1 = "F 83.4 83.4 73.5 80.4 84.0 84.0 83.4 84.0 82.3 82.8 82.8 82.8 81.5 18.5 24.1 88.0 15.0 14.1 32.0 $(t_0 - t_1) = {}^{o}F$ 14.2 13.6 18.7 58.6 23.0 41.5 29.2 9.6 9.6 9.6 9.6 9.6 9.6 X = % Solids 12.9 12.9 12.9 12.9 12.9 12.9 12.9 Ch = BTU/1B - OF 0.925 0.925 0.925 0.925 0.925 0,925 0, 0.900 0.900 0.900 0.900 0.900 0.900 0.900 16,600 -16,480 13,100 8,160 10 20,500 16,700 17,440 13,600 14,500 12,450 11,200 qh = BTU/hr 7,130 7,120 17.80 16,20 19,21 13.79 9.20 1 We = 1b/hr 21.00 17.50 15.50 14.40 13.60 12,91 7.57 8.80 1134 1134 1134 1134 1134 11.34 Hs = BTU/15 Taxa L 1138 1136 1137 1137 1137 1135 1133 206 206 805 202 204 He = BTU/1b 207 214 210 575 214 517 200 200 928 927 928 926 932 930 AH = BTU/1b 924 926 925 923 923 935 933 19,100 17,800 16,500 15,000 12,850 8,550 10 qg = BEU/hr 16,180 11,350 13,250 12,550 7,080 11,900 8,200 stm = op 159.0 147.5 148.8 146.8 136.1 104.9 1 156.5 151.2 154.0 153.8 149.1 118.1 131.0 , hr = BTU/hr-ft2-or 74.10 73.35 67.40 62.50 57.30 50.10 5 62.80 57.60 52.30 19.60 48.50 36.35 38.00 66.1. 66.0 66.0 65.9 65.5 66.0 Pp = 16/22 67.4 67.4 67.4 67.4 67.4 . 67.0 67.2 1.031 0.900 0.515 0.312 0.683 0.075 0 Vb = ft/sec 0.937 0.763 0.591 0.113 0.293 0.093 0.137 0.0382 0.0405 0.0508 0.101.0 0.0655 0.3650 0. Mb = 1b/sec-ft 0.0895 0.1060 0.1300 0.1800 0.2320 0.5950 0.1150 156.4 128.0 77.5 45.3 17.3 1.2 NRO 61.6 42.3 26.8 7.4 13.5 0.9 1.8 0.349 0.350 0.358 0.348 0.351 0.346 C K_h = ETU/hr-ft²⁰F/ft 0.344 0.344 0.344 0.315 0.344 0.351 0.348 368 388 485 3,375 2 Npr = Sola 623 988 842 998 1,691 1,225 2,185 4,190 4,145 Kh 2/3 53.0 51.2 61.5 72.5 99.0 221,0 1 Npr 88.5 114.5 99.9 141.5 167.0 310.0 258.0 $N_{st} \ge 10^{h} = h_{r}$ 3.71 4.50 5.53 8.37 30.45 2 3.27 3.07 3.45 4.05 5.50 7.49 17.90 12.70 CbVb6P 2/3 NPro 1.966 2.770 1.675 4.01 8.28 68,20 Nst 3.445 4,640 2,715 7.78 12.50 55.50 32.80

x 102

TABULATION OF EXPERIMENTAL RESULTS

tun 2. 14	Run No. 15	Run No. 16	Fun No. 17	Run No. 18	Run No. 19	Ro <u>e</u> 2
150	1500	1230	840	4.95	118	266
8.8	88 . 1	1.02.8	112.2	121.4	178.8	Lilia
3.4	71.0	82.3	85.7	86.6	86,6	85.
55.4	18.1	20.5	26.5	34.8	92.2	50. 1
9.6	5.6	5.6	5.6	5.6	5.6	5.1
.925	0.950	0.950	0.950	0.950	0.950	0.99
,580	25,800	24,000	21,200	16,100	10,300	16,7X
.0.90	27.10	24.95	22.60	17.39	10.80	15. 9
134	11.34	1134	1134	1133	1134	1133
203	206	207	207	505	208	300
991	928	927	927	931	926	993
,150	25,400	23,150	20,950	16,200	10,000	- 14,5K
16.8	159.L	147.7	110.1	130.1	102.3	113.1
2.80	96.90	95.25	90.80	75.55	59.25	76.5
5.6	64.2	61:.2	64.1	64.1	63.6	64 . (
.105	1.080	0.887	0.606	0.357	0.085	0.199
2600	0.0109	0.0127	0.0168	0.0252	0.0785	0.0h2ć
2.3	555.5	392.0	202.5	79.3	6.0	25.0
.356	0.351	0.354	0.356	0.357	0.366	, 0.362
, 135	106	123	161	5/15	73h	394
81.0	22 . L	21,.7	29.6	38.9	0.18	57.5
2.90	L. 09	L.90	6.82	9,65	31.80	18.20
112.11	0.915	, 1,210	2,020	3.76	25.8	10.15

	S. A.				and the second		A Charles and				and the second second				같은 말을 수 있어? 같은 말을 수 있어? 같은 말을 수 있어?		And the second second				A Contraction of the second state	States and the	St. Squarter	Contraction Compared
16 1	Run 10.17	Run No. 18	Run No. 19	Run No. 20	Run No. 21	han No. 22	Run 116•22	1899 1999 - 291	Run No. 25	Run No. 26	Run Ro. A.	Run Ros 28	Run <u>No. 22</u>	Aun No. 30	kun <u>ko. 31</u>	Run No. 3	Run No. 33	Ran No. 34	Run <u>No. 35</u>	Run <u>No. 36</u>	Rum No. 37	Run 10. 38	Run No. 39	hon No. 10
0	81,0	495	118	266	1590	997	870	289	1.80	1,3%0	364	300	832	1127	577	11.62	990	750	570	315	1440	1185	1470	1005
8 1	12.2	121.1	178.8	alili.0	95.5	113,0	121.3	16h.1	192.1	113.9	125.6	141.5	115.7	111.5	120.2	116.	120.7	120.2	124.1	134.7	112.0	115.7	112.1	122.0
3	85.7	86.6	86.6	85.9	73.8	80.8	84.0	87.3	87.2	95.8	9h. h	92*0	90.6	91.2	90.6	98.	5 98.6	97.4	95.7	94.6	94.7	96.6	92.1	95.9
5	26.5	34.8	92.2	58.1	21.7	32.02	37.3	76.8	101°8	10,1	31.2	49.5	25.1	20.3	29.6	18.	22.1	22.8	28.lı	40.1	17.3	19.1	20,0	26.1
6	5.6	5.6	5.6	5.6	3.0	3.0	3.0	3.0	3.0	23.1	23.2	23.1	23.1	23.1	23.1	20.	4 20.li	20.4	20.li	20.h	20 . 4	20.4	13.7	23.7
0 (0.950	0.950	0.950	0,950	0.977	0.977	0.977	0.977	0.977	0,825	0.825	0.825	0.825	0.825	0.825	0.81	3 0.813	0.813	0.843	0.843	0.813	0.813	0.896	0.896
00	21,200	16,400	10,300	11,700	33,700	IL, X)()	31,700	21,500	17,800	20,600	9,370	12,250	17,260	18.890	14,100	21,60	18,150	Ili, 400	13,650	10,680	21,000	19,100	26,400	23,550
95	22.60	17.39	10.80	15.59	35 . 15	32.90	35.70	23.60	20.50	22 . 80	14.05	12.95	17.79	19.90	15.09	22.	50 18.07	16.20	14.67	12.10	21.40	19.75	27.65	24.50
h.	1134	1133	باودد	1133	1134	1136	11.34	22.34		11.26	1152	1126	1152	1156	1759	1126	11.25	1125	1126	1156	1125	1125	1128	1121
7	207	202	208	200	502	208	308	206	208	211	209	207	206	2005	199	201	202	201	204	205	198	200	208	20F
7	927	931	926	933	927	926	926	928	926	915	917	919	920	926	227	922	923	924	922	921	927	925	920	917
50	20,950	16,200	10,000	16,550	32,850	30,400	29,700	21,990	18,950	20,000	12,890	12,900	16,350	18,100	14,000 ·	-20,65	0 16,670	14,980	13,520	11,140	19,850	18,250	25,150	22,500
•7	110.1	130.1	102.3	115.7	253.2	Illia7	135.3	109.0	92.7	131.1	1,32.0	125.7	132.0	1,92.0	1274	128.	1 123.1	123.8	125.2	121.8	129.0	126.2	136.8	124.7
25	90.80	;75.55	59.25	76.50	130.2	130.2	133.2	122.1	15/101	91.,20	59.30	57.60	75.30	64 .7 0	67.00	99.(0 82.25	73.50	65.70	55.60	93.50	88.00	117.1	109.1
2	64.1	64.1	63.6	61.0	63.h	63.0	63.0	62.7	62 . 5	72.2	72.2	72.0	72.2	72.2	72.2	70.	6 70.6	70.6	70.6	70.5	70.7	70.7	67.9	67.7
187	0.606	0.357.	0.085	0,192	1.169	0.734	0.638	0,223	0.133	0,886	0.233	0.193	0.534	0.722	0.370	0.936	0.619	0.192	0.374	0.207	0.942	0.777	1,001	0.687
.27	0.0168	0.0252	0.0785	0.0126	0.0032	0.0010	0.0013	0.0079	0.0102	0,1030	0.3010	0.3500	0,1580	0.1210	0.2100	0.71	0.088	5 0.1110) 0.1400	0.2270	0.0669	6 0.0770	0.0202	0,0278
0.9	202.5	79.3	6.0	25.8	5050*0	1010,0	816.0	147.9	72.07	54.3	4.8	3.5	21.3	37.7	II.I.	81.	15.3	27.li	16.5	5.6	87.5	62.8	296.1	146.4
154	0.356	0.357	0.366	0.362	0.356	0,359	0.360	0.367	0.372	0.333	0.335	0.336	0,333	0.332	Ó . 332	0.3	8 0.338	0.338	0.338	0.340	0.337	0.337	0.316	0.348
.23	161	21,2	734	394	31.6	39.1	15.0	75.6	96.7	91.9	2,695	3,090	1,410	1,082	1,879	63	795	995	1,258	2,030	599	691	188	258
-		• ~ A	0.0		*/1 /1	** 61	19 09 13 19	1 1 2 A	177 A	1914. m	105 0	011 6	. 105 A	1 AC 11	101 6	~								
1.7	29.6	38.9	0.10	3102	awaw r ri	aloj) Jan	1.6.41.3 f	Ligu ne en	CLAV In nn	7403. 1. pr	72.00 72.00	chię? Na ca	26749 2 CM	27. 14 17 14	2 1 E	73	.7 85.1	99.8	116.5	160.5	71.0	78.3	32.8	10.1
.90	6.82	9.65	31.80	10.00	2*44	VoUS	X #114	et and	na o chi	407D	4.4.00	4.3474	(0.)	2041	vell?	Control of the second s	.93 5.5	1 6.9	5 8.2	0 12.5	2 4.6) 5.28	5.32	7.30
.210	2.020	3.76	25.8	10.15	0.501	0,928	1.139	L.61	6.86	4.67	22.93	29 . 60	8.22	5.75	12,80	3	63 5.0	6 6.91	9.5	5 20.20	3.2°	7 4.11	1.716	2.95

							2									1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1					en in
Run No. <u>30</u>	Run No. 31	No. 32	Run No. 33	Sun No. 34	Run <u>No. 35</u>	Run No. 36	Run No. 37	Run <u>No. 38</u>	Run No. 39	Run <u>No. 10</u>	Run No <u>. I</u> A	Run <u>No. 112</u>	Bun <u>No. 13</u>	Ron No. LL	Run No. 15	Run No. 16	Run <u>No. h7</u>	Run No. 18	Run No. 119	Run No. 50	Run No. 51
1754	577	11.62	990	750	570	315	1440	1185	11,70	1005	1230	51,0	385	802	1500	1200	840	720	1365	840	11,55
111.5	120,2	116.6	120.7	120.2	124.1	134.7	112.0	115.7	. 112.1	122,0	125.1	147.1	152.7	126 . li	108.3	115.2	121.5	127.2	112.0	122.7	110.2
91.2	90.6	98.6	98.6	97 . L	95.7	94.6	94.7	96.6	92 . 1	95.9	103.0	104.2	103.5	1.00.0	87.8	91.6	94.1	96.6	9l1 . 5	94.1	92.6
20.3	29.6	18.0	22.1	22.8	28.4	10.1	17.3	19.1	20.0	26.1	22.1	42.9	19.2	26.4	20.5	23.6	27.4	30.6	17.5	28.6	17.6
23.1	23.1	20.1	20.4	20.4	20.4	20.4	20.4	20.1	13.7	13.7	13.7	13.7	13.7	13.7	10.1	10.1	10.1	10.1	10.1	10.1	10.1
0.825	0.825	0.813	0.843	0.813	0.813	0.813	0.843	0.843	0.896	0.896	0.896	0.896	0.896	0.896	0,923	0.923	0.923	0.923	0.923	0.923	0.923
18,890	14,100	21,600	18,150	14,100	13,650	10,680	21,000	19,100	26,400	23,550	24,400	20,750	16,930	18,970	28,100	26,100	21,250	20,350	22,100	22,200	23,650
19,90	15.09	22.60	18.07	16.20	14.67	12.10	21.10	19.75	27.65	24.50	26.90	21.95	17.85	19.80	29.50	29.80	23.15	22,60	24.70	23.35	24.70
1126	1152	1126	11.25	1125	1126	1126	1125	1125	1128	1151	1122	1155	1122	11.21	1121	1127	1128	1122	1121	1122	1122
200	199	201	202	201	201	205	198	200	208	201	201.	209	209	108	203	201	207	205	201	203	202
926	927	922	923	921	922	921	927	925	920	917	918	91.3	913	023	918	923	921	91.7	920	919	920
28,100	14,000	20.850	16.670	11.980	13.520	11.110	19.850	18.250	25.150	22.500	24,700	20.050	16.300	18.300	27.100	27.500	21.600	20.700	22,700	21.150	22,700
132.0	127.1	198.1	193.1	192.8	195.9	N. 191.8	129.0	126.2	136.8	191. 7	110.8	116.9	112.1	115.1	133.8	132.0	127.0	122.h	126.7	126.0	130.6
84.70	67.00	. 00 M	80.05	72.50	25 70	EE.AO	09. dh	88.00	117 1	100 L	105 0	105 8	88.1	700.9	102 0	1 00 0	103.9	102.7	108.8	103.5	110.0
72.2	72.02	77.00	70 6	90 6	70 6	77400 78 2	73000	- 00800	den den 6 is den L. Pro - Pro	20700	2. C. 7 & C.	2020	(m)	L'Mec	1. C.	LCVOV	66 0	66 0	66.0	66.0	66.0
0.722	0.370	1000	0.01	1V.0U	(UeU	(Vs)	10.10	V (2000)	ו)0	0 / e 7	C+10	Olell	01.4	.0102 Co102	. 02+1	0.0LA	n enn	n tht	n ocr	n dan	1 020
0.1210	0.2100	0.930	0.009	0.472	0.374	0.207	0.942	0.111	L.WI	0.007	0.045	216.0	6.509	0.551	1.0555	S170•0	0.370	V+2V2	0.750	0.001.6	n nnoo
37.7	11.1	0.710	0.0885	0.1110	0.1400	0,2270	0.0665	0.0770	0.0202	0.0276	0.0552	0.0000	0.0600	0.0331	0.0097	U.ULLII	O .OTTO	vellov	U.ULUU	Veveluo Man A	rol 0
0.332	0.332	81.3	15.3	27.4	15.5	5.6	87.5	62.8	296.1	146.4	221.0	49.9	26.0	98.2	625.0	424.0	233.0	104.0	532.U	233.0	>>U.v
1, 082	1,879	0.338	0.338	0.335	0.338	0.340	0.337	0.337	0.346	0.348	0.349	0.353	0.353	0.349	0.350	0.351	0.853	0,354	0.351	0.353	0.350
		637	795	995	1,258	2,030	599	694	188	258	208	705	519	306	92.1	108.1	. 137.5	: 140.J	90.6	137.5	94.0
105.0	151.5	73.7	85.7	99.8	116.5	160.5	71.0	78.3	32.8	40.L	35.L	54.7	67.0	15.3	20.1	22.6	26.6	28.0	21.4	26.6	20.6
5.47	8.15	4.93	5.91	6.95	8.20	12.52	4.60	5.28	5.32	7.30	6.78	13.02	15.22	8.32	5.30	6.18	7.96	9.25	5.16	7.98	4.90
5.75	12.80	3.63	5.06	6.94	9.55	20.10	3.27	A state	1.746	2.95	2.38	7.12	10.20	3.77	1.08	1.165	5 2.12	2.59	1,105	2.12	1.216

TABULATION OF SLUREY VISCOSITY VERSUS SHEAR BATE

RPM Shear Rate	600 942		300 471		20 31	0 4	10 15	0 7	6 9.1	42	3 4.1	71
(Sec -1)	********	Visc		Visc	~ ~	Visc		Visc		Visc		Visc
	Scale Rdg	10 ³	Scale Rdg	10 ³	Rdg	10 ³	Rdg	10 ³	Rdg	10 ³	Rdg	10 ³
Run No.	LB FT ²	LB FT.SEC	LB FT ²	LB FT.SEC	LB FT2	LB FT.SEC	LB FT ²	LB FT.SEC	$\frac{LB}{FT^2}$	LB T.SEC		LB T.SEC
l	0.365	12.5	0.315	21.5	0.298	30.6	0.267	54.8	0.170	581.0	0.135	922.0
2	0.365	12.5	0.315	21.5	0.298	30.6	0.267	54.8	0.170	581.0	0.135	922.0
3	0.365	12.5	0.315	21.5	0.298	30.6	0.267	54.8	0 ,170	581.0	0 .13 5	922.0
4	0.362	12.4	0,315	21.5	0.300	30.8	0.275	56.4	0.172	587 .0	0.132	902.0
5	0.365	12.5	0,315	21.5	0.298	30.6	0.267	54.8	0.170	581.0	0.135	922.0
6	0.352	12.1	0 .305	20.8	0.305	31.3	0.272	55.8	0.165	564.0	0.125	855.0
7	0.360	12.3	0.315	21.5	0.315	32.3	0.276	56.6	0.180	615.0	0.135	922.0
8	0.198	6.8	0.162	11.1	0.137	14.0	0.118	24.2	0.085	290.0	0.075	512.0
9	0.185	6.3	0.147	10.1	0.132	13.5	0.115	23.6	0.075	256.0	0.067	458 . 0
10	0.185	6.3	0.147	10.1	0.132	13.5	0.115	23.6	0.075	256.0	0.067	458.0
11	0.185	6.3	0.147	10.1	0.132	13.5	0.115	23.6	0.075	256.0	0.067	158.0
12	0.175	6.0	0.145	9.9	0.132	13.5	0.115	23.6	0,080	273.0	0.070	1,79.0
13	0.165	5.6	0,135	9.2	0.125	12.8	0.115	23.6	0.085	290.0	0.072	492.0
14	0.165	5.6	0.132	9.0	0.128	13.1	0.112	23.0	0.080	273.0	0.068	465 .0
15	0.080	2.7	0.055	3.8	0 .050	5.1	0.035	7.2	0.020	68.4	0.015	103.0
16	0.075	2.6	0.055	3.8	0.048	4.9	0.035	7.2	0 .022	75 .1	0.020	137.0
17	0.072	2.5	0 .055	3.8	0.047	4.8	0.035	7.2	0.020	68.4	0.020	137.0
18	0.072	2.5	0.055	3.8	J.047	L.8	0.035	7.2	0.020	68.4	0.020	137.0
19	0.060	2.0	0.01/8	3.3	0.042	4.3	0.035	7.2	0.022	75.1	0.015	103.0
20	0.065	2.2	0.050	3.4	0.045	4.6	0.035	7.2	0.022	75.1	0.018	123.0
21	0.038	1.3	0.025	1.7	0.020	2,0	0.013	2.7	0.005	17.1	0.002	13.7

TABLE III (cont.)

TABULATION OF SLUEPY VISCOSITY VERSUS SHEAR RATE

RPM Shear Rate	600 9142		300 471		200 31.)) Ц	10 15	0 7	6 9.1	42	3 4.1	71
(Sec -1)	Scale Rdg	visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	V18C x 10 ³
Run No.	$\frac{LB}{FT^2}$	LB FT.SEC	LB FT2	LB T.SEC	LB FT2	LB FT.SEC	$\frac{LB}{FT^2}$	LB FT.SEC	LB FT2 1	LB FT.SEC	LB FT2 1	LB T.SEC
22	0.035	1.2	0.022	1.5	0.017	1.7	0.012	2.5	0.005	17.1	0.002	13.7
23	0.035	1.2	0.022	1.5	0.017	1.7	0.012	2.5	0 .005	17.1	0,002	13.7
24	0.030	1.0	0.020	1.4	0.017	1.7	0.012	2.5	0.005	17.1	0.002	13.7
25	0.030	1.0	0.020	1.4	0.017	1.7	0.012	2.5	0 .005	17.1	0,002	13.7
26	0.419	14.3	0.369	25.2	0.340	34.8	0.301	61.7	0.177	604 .0	0.135	922.0
27	0.119	14.3	0.369	25.2	0.340	34.8	0.301	61.7	0.177	604 .0	0.135	922 .0
28	0.402	13.7	0 .35 5	24.3	0,330	33.8	0.290	59 .5	0.175	598.0	0.090	6 15.0
29	0.410	14.0	0.362	24.8	0,335	34.3	0,300	61.5	0.180	615.0	0.130	889.0
30	0.419	14.3	0.369	25.2	0.340	34.8	0.301	61.7	0.177	60 4.0	0.135	922.0
31	0.419	14.3	0.369	25.2	0.340	34.8	0.301	61.7	0.177	604 .0	0.135	922.0
32	0.280	9.6	0.211	17.9	0.221	22.7	0.194	39.8	0.120	410.0	0,100	684.0
33	0.280	9.6	0.241	17.9	0.221	22.7	0.194	39.8	0.120	410.0	0.100	684.0
34	0.280	9.6	0.24	17.9	0.221	22.7	0.194	39.8	0.120	410 .0	0.100	684 .0
35	0,280	9.6	0.211	17.9	0.221	22.7	0.194	39.8	0.120	410.0	0.100	684.0
36	0.275	9.4	0.235	16.1	0.220	22.6	0.192	39.4	0.120	110.0	0.100	684.0
37	0.287	9.8	0.215	16.8	0.223	22.9	0.200	山.0	0.125	427.0	0.100	68h.0
38	0.283	9.7	0.243	16.6	0.222	22.8	0.197	40.4	0.123	420.0	o .100	684.0
39	0,109	3.7	0.088	6 .0	0.075	7.7	0.062	12.7	0.045	154.0	0.011	280.5
110	0.103	3.5	0.083	5.7	0.073	7.5	0.061	12.5	0.044	150.0	0.041	280.5
抑	0.101	3.5	0.081	5.5	0.072	7.4	0.060	12.3	0.043	147.0	0.040	273.5
L2	0.093	3.2	0.079	5.4	0.069	7.1	0.052	10.7	0.042	143.5	0.038	260.0

TABLE III (Cont.)

TABULATION OF SLURAY VISCOSITY VERSUS SHEAR RATE

RPM Shear Rate	600 942		300 1,71		200 31) L	100 15	5 7	6 9.1	42	3 4•	3 4.71		
(Sec -1)	Sca le Edg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³	Scale Rdg	Visc x 10 ³		
Run No.	IB FT2	LB T.SEC	FLS H	LB T.SEC	$\frac{LB}{FT^2}$	LB FT.SEC	I.B FT ² 1	LB ST.SEC	LB FT2	LB FT.SEG	$\frac{LB}{FT^2}$	LB FT.SEC		
13	0.093	3.2	0.079	5-4	0.069	7.1	0.052	10.7	0 .042	143.5	0.038	260.0		
44	0.101	3.5	0,081	5.5	0.072	7.h	0.060	12.3	0.043	147.0	0 . 010	273.5		
1.5	0.068	2.3	0.049	3.4	0.0l1	4.2	0.031	6.ь	0.018	61.5	0.013	89 .0		
16	0.063	2.1	0.017	3.2	0.0/1	4.2	0.031	6.կ	0.017	58.1	0.013	89 .0		
47	0.061	2.1	0.027	3.2	0.011	4.2	0.031	6 . k	0.016	54.7	0.012	82.1		
43	0.059	2.0	0.013	2.9	0.039	1.0	0.030	6.2	0.015	51.3	0.012	82.1		
149	0 .063	2.1	0.047	3.2	0.011	4.2	0.031	6.4	0.017	58.1	0.013	89 .0		
50	0.061	2.1	0.046	3.1	0.001	b.2	0.031	6.4	0.016	54.7	0.012	82.1		
51	0.068	2.3	0.0],9	3.4	0.0/1	h.2	0.031	6.4	0,018	61.5	0,013	89 .0		

. .

VISCOSITY OF 12.9% HALF MICRON AND FINER KAOLIN SLURRY

Viscosity vs. Shear Rate



FIGURE 7

VISCOSITY OF 5.6% HALF MICRON AND FINER KAOLIN SLURRY

Viscosity vs. Shear Rate



VISCOSITY OF 3.0% HALF MICRON AND FINER KAOLIN SLURRY

Viscosity vs. Shear Rate



Logarithmic, o X o Uyeles MADE IN U.S.A.

Shear Rate, sec.1

FIGURE 10

VISCOSITY OF 23.15 TWO MICRON AND FINER KAOLIN SLURRY

Viscosity vs. Shear Rate



Shear Rate, sec.1

Logarithmic, 3×3 Cycles. MABE IN U.S.A. VISCOSITY OF 20.4% TWO MICRON AND FINER RAOLIN SLURRY

Viscosity vs. Shear Rate



Lugarithinic, 3 × 3 Uycles. Nadeir U.s.a.

VISCOSITE OF 13.10 TWO . ICRONS AND FILER RACLIN SLERY

Viscosity vs. Shear Rate



Shear Rate, sec.1

KEUFFEL & ESSER CO. MADEIN ULSLALT A X 3 CYCLER

Zavi

VISCOUTTY OF 10.1% TWO WIGRON AND MINER KAOLIM SLURRY

Viscosity vs. Shear Rate



大学位 House a sesen co. MADEIN U.S.A. 人体位 KEUPPEL & SSEE CO. MADEIN U.S.A. 小文 3 CVCLES

Shear Rate, sec.1



19116

Reynolds Number = $\left[\frac{DVP}{4}\right]$

LOGARITHMIC 359-125LG KEUFFEL & ESSER CO. MADE IN U.S.A. 3 X 5 CYCLES 3#1



Discussion of Results

The heat balances were rather satisfactory. The heat transferred based on the steam quality and condensate rate was accurate to three significant figures. Based on the temperature rise of the slurry at the higher flow rates, it was also accurate to three figures, however, at the lower flow rates it was accurate to only two figures. This was due to a fluctuation in the outlet temperature which varied as much as 10% of the temperature rise of the slurry even though the temperature was taken after the slurry flowed through a mixing valve. Without any method of mixing, even at the higher flow rates, it was found that there was much error in the outlet temperature. Plotting the heat balances versus the flow rate showed that the data based on the heat content of the steam was more reliable. The plots of this data always formed a smooth curve, while the data based on the temperature rise of the slurry often straddled on both sides of this curve, especially at the lower flow rates. As a result, the heat transfer coefficients were based only on the heat given up by the steam and not on the temperature rise of the slurry. It should be pointed out that with a guard jacket on the heat exchanger and the entire unit well insulated, including the steam trap, the heat losses were quite small. Also, the steam calorimeter was located at the entrance of the heat exchanger steam jacket.

The first approach in determining the viscosity of the slurries was by means of a pipeline viscometer. The pressure drop due to friction, through a $1^{n} \times 9^{1}$ long pipe was so small that it could not be measured to any degree of accuracy, especially at the lower flow rates. This method was, therefore, not used. A literature search was made to find other methods of determining viscosity that could be used instead. A method used by many investigations was to use laboratory viscometer data and relate these data to pipeline flow.

A Fann viscometer was used to determine the viscosity of the slurries. This instrument, which is rather an expensive piece of laboratory equipment, was just reconditioned before it was used. This is one of the types of viscometers used in the clay industry to measure viscosities of The data from the Fann viscometer was significant to clay slurries. three figures. The viscosity curves, shown on Figures 7 through 14, are, no doubt, a rather accurate representation of the non-Newtonian behavior of the slurries. Using the equation $\dot{\nabla} = \frac{8v_b}{n}$, the shear rates of the slurries flowing through the heat exchanger were calculated using the average velocity. From the viscosity curves, which were determined at the average temperature the slurry was at in the heat exchanger, the shear rate was used to find the corresponding bulk viscosity. This viscosity was used to correlate the heat transfer data. A variation in the mass rate through the heat exchanger by a factor of ten caused the bulk viscosity to change by about a factor of thirty. Therefore, by increasing the mass rate by a factor of 10 caused the Reynolds number to increase by about a factor of 300, since the slurries were pseudoplastic. The non-Newtonian behaviors of the fluids used were not known at the time of the equipment design. This made sizing the equipment and establishing flow rates difficult.

Several additional comments can be made about the viscosity curves shown on Figures 7 through 11. Generally speaking, the slopes of all the viscosity curves are approximately the same. However, as the solids content of the slurry became greater, there is a tendency for the viscosity curves to have a slightly greater slope, indicating a more non-Newtonian behavior. Also, an increase in solids caused the viscosity curves to shift upward on the graphs, indicating higher viscosity. At the same solids content, the finer kaolin produced a higher slurry viscosity than the coarser material, but the non-Newtonian behavior (i.e. slope of curve) was about the same for both materials.

The following equation (7) was used to determine the effect of a natural convection on the heat transfer coefficients:

$$\frac{hD}{K} = 1.75 \text{ F1} \left[\frac{WC}{KL} + 0.0722 \left(\frac{D}{L} \text{ N}_{Gr}, \text{ N}_{T} \right)^{-0.75} \text{ F2} \right]^{1/3}$$
Equation (10)

There: F_1 and F_2 are dimensionless factors that can vary between 1 and 0.

 N_{Gr} is the Grashof number $\left(\frac{D^3 e^2 g}{\mu^2}\right)$ (BAt) N_{Pr} is the Frandtl number $\left(\frac{C \mu}{K}\right)$

This equation is for heating fluids flowing vertically upward. The Husselt numbers varied from 9.07 to 32.35. The Grashof numbers varied from 0.188 to 6300. The Frandtl numbers varied from 31.6 to 5490.

The correction for natural convection was quite small for all 51 runs and within experimental error. The correction for natural convection was, therefore, disregarded. It was found, however, that at the higher flow rates along with higher viscosities, the correction for natural convection was very small. Conversely, at the lower flow rates along with lower viscosity, the correction became more significant. However, it was still less than 5% of the Musselt number.

It was brought out in the Experimental Results that it was desired to correlate the heat transfer data using an equation which had the Reynolds number as one of the dimensionless groups. This made it possible to plot the data with the Reynolds number on the abscissa of the graph easily showing in what range of the laminar region the work was done. The equation that was selected to correlate the data was of the same form as that developed by Sieder and Tate. This equation can be expressed in two different ways:

$$\left(\frac{h_{f}}{C_{b}V_{b}}\right) \left(\frac{C_{b}H_{b}}{K_{b}}\right)^{2/3} \left(\frac{H_{W}}{H_{b}}\right)^{0.14} \left(\frac{L}{D}\right)^{1/3} = d\left(\frac{DV_{b}P_{b}}{H_{b}}\right)^{0}$$
Equation (13)

or

$$\left(\frac{h_{f}D}{K_{f}}\right) \left(\frac{C_{b}\mathcal{H}_{b}}{K_{b}}\right) = \frac{1/3}{\binom{\mathcal{H}_{W}}{\mathcal{H}_{b}}} \stackrel{0.1h}{=} \left(\frac{L}{D}\right) \stackrel{1/3}{=} f\left(\frac{DV_{b}P_{b}}{\mathcal{H}_{b}}\right)^{g}$$
Equation (13a)

Where: d, e, f, and g are constants determined from the plot of the data. As can be seen from Figure 15, the heat transfer data was correlated using Equation (13). The data gave a rather satisfactory plot and, therefore, it was felt that there was no need to develop a special equation based on this work.

The heat transfer equation resulting from this work is as follows:

$$\left(\frac{\mathbf{h_f}}{C_{\mathbf{b}}V_{\mathbf{b}}\mathbf{R}_{\mathbf{b}}}\right)\left(\frac{C_{\mathbf{b}}\mu_{\mathbf{b}}}{K_{\mathbf{b}}}\right)^{2/3} = 0.70 \left(\frac{\mathbf{b}V_{\mathbf{b}}\mathbf{R}_{\mathbf{b}}}{\mu_{\mathbf{b}}}\right)^{-2/3}$$

This equation is for an L/D ratio of 68.7. The Sieder and Tate Equation (21) for laminar flow is as follows:

$$\left(\frac{h}{UVP}\right)\left(\frac{C_{\mu}}{K}\right)^{2/3} \left(\frac{L}{D}\right)^{1/3} \left(\frac{\mu_{W}}{\mu_{D}}\right)^{0.114} = 1.86 \quad \left(\frac{DVP}{\mu}\right)^{-2/3}$$
Equation (15)

For an L/D of 68.7, this equation becomes:

$$\left(\frac{h}{CVQ}\right)\left(\frac{C}{K}\mu\right)^{2/3}\left(\frac{\mu_W}{\mu_D}\right)^{0.14} = 0.155 \left(\frac{DVQ}{\mu}\right)^{-2/3}$$
 Equation (16)

The Reynolds number exponent was found to be the same from this work as that of the Sieder and Tate equation. However, there is a difference in the intercepts. Substituting 68.7 in the Sieder and Tate equation for the L/D term, the constant becomes 0.455. From this work it was found to be 0.70. This difference, is no doubt, mostly due to the viscosity correction term included in the Sieder and Tate equation but not included in the equation developed from this work. The reason for not including this term was because there was no data available and to obtain the necessary data would have been quite a difficult task. Also, it was felt that this term would make only a small difference in the final correlation and, therefore, could be disregarded.

In order to get a constant of 0.155 in the equation resulting from this work, the $\frac{\mu_W}{\mu_b}$ ^{0.11} term, if included in this equation, would have to be equal to 0.65. In the case of a dewtonian fluid, the viscosity term is usually less than one. For example, with an average bulk temperature of 100°F and a pipe wall temperature of 235°F, this viscosity term for water is equal to 0.839. At these two temperatures, which are in the same range as the sharpy to perstures, the viscosity of eater charges by a factor of 3.5. However, the average change between the bulk viscosity and the wall viscosity of the shurries, calculated from $\frac{h_{\rm ev}}{h_{\rm D}} = 0.65$ is found to be 21.6 times. The effect of temperature on the viscosity of the shurries is probably close to that for water. From what is known about the shurries, it is believed that the shear rate has more effect on the viscosity than the temperature. The difference between the values of 21.6, calculated for the shurries, and 3.5, which is probably close to the average value for the shurries, and 3.5, which is probably close to the average value for the temperature effect on the viscosities of the shurries, is probably due to their non-lewtonian schavior.

The groatest shear rate must be close to the pipe sall and not in the conter of the stream. Close to the walk, both the higher teoperature and greater shear rate could cause the viscosity to decrease by this large (21.8 times) amount. This could account for a large difference between the bulk viscosity and the walk viscosity. In order to study this more thoroughly, more heat transfer and viscosity data are needed. The omission of this viscosity correction factor given possibly account for some of the spread of the data.

Coveral comments can be made about the effect of solids content and particle size on the heat transfer coefficients. The coefficients had a tendency to increase with increase in particle size and decrease with inerease in solids content. These two variables more taken indirectly into account to some degree in the viscosity term when the heat transfer data uas correlated.

1,2

CONCLUSIONS

An equation similar in form to the Sieder-Tate equation for Newtonian fluids can be used to calculate the heat transfer coefficients in terms of various dimensionless groups of the non-Newtonian slurries studied in this work in laminar flow. The heat transfer coefficients for slurries of half micron and finer kaolin and water are as follows:

- For a solids content of 12.9 weight percent the heat transfer coefficients ranged from 62.80 to 36.35 for a Reynolds number range of 61.6 to 0.09.
- For a solids content of 9.6 weight percent the heat transfer coefficients ranged from 74.10 to 50.10 for a Reynolds number range of 156.4 to 1.2.
- 3. For a solids content of 5.6 weight percent the heat transfer coefficients ranged from 96.90 to 59.25 for a Reynolds number range of 555.5 to 6.0.
- 4. For a solids content of 3.0 weight percent the heat transfer coefficients ranged from 130.2 to 124.1 for a ^heynolds number range of 2020 to 71.7.

The heat transfer coefficients for slurries of two microns and finer kaolin and water are as follows:

- 1. For a solids content of 23.1 weight percent the heat transfer coefficients ranged from 9h.20 to 57.60 for a Reynolds number range of 5h.3 to 3.5.
- 2. For a solids content of 20.4 weight percent the heat transfer coefficients ranged from 99.00 to 55.60 for a Reynolds number range of 81.3 to 5.6.

- 3. For a solids content of 13.7 weight percent the heat transfer coefficients ranged from 125.2 to 88.10 for a Deynolds number range of 296.1 to 49.9.
- 4. For a solids content of 10.1 weight percent the heat transfer coefficients ranged from 133.8 to 122.4 for a Reynolds number range of 625.0 to 184.0.

The degree of pseudoplasticity did not significantly change with solids content or particle size for the kaolin slurries studied in this work. However, an increase in solids content increased the viscosity at any given shear rate. Also, at identical solids contents the kaolin of finer particle size gave higher values of viscosity than the coarser material. HOMENCLATURE

A - Area for heat transfer - ft^2 C - Specific heat - BTU/1b - oF D - Pipe diameter - ft f - Friction factor - dimensionless $G - Mass velocity - 1b/hr - ft^2$ of cross section g - Acceleration due to gravity - μ .17 x 10⁸ ft/hr² h - Coefficient of heat transfer between fluid and surface - BTU/hr - ft^2 - o_F K - Thermal conducitivity - BTU/hr - ft² - ^oF/ft L - Length of heat transfer tube - ft. N - Dimensionless group - i.e. - N_{Re} - Leynolds number $P - Pressure - lb/ft^2$ q - Heat transfer rate - BTU/hr r - Pipe radius - ft S - Cross section of stream in a tube - ft^2 T - Absolute temperature - OR t - Temperature - °F U - Overall heat transfer coefficient - $BTU/hr - ft^2 - {}^{o}F$ V - Average velocity - ft/sec w - Mass rate of flow - 1b/hr x - Percent of dry solids by weight in slurry Xv - Fraction of solids by volume

Subscripts

b - For bulk

c - For condensate

f - For film

i - For inlet

1 - For liquid

m - For log mean temperature

o - For outlet

- p For particle
- s For steam
- v For volume

w - For wall

Greek

Δ - For difference
κ - /iscosity of fluid - lb/hr - ft centipoises x 2.42
π - 3.1416
ℓ - Density of fluid - lb/ft³
γ - Shear rate - sec⁻¹
τ - Shear stress - lb/ft²

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Samle saleslations

Sær	mie Am - Con do. 1 (defor to Cable I for Griginal data)
1.	Flow Tate (a) = 11 3/8 1hs./30 sec = 1,365 1b/hr
2.	Outlet Cemperature of Clurry (t _o) 1.90 millivolts thermocouple reading from Figure 1 is 97.6°F.
3.	inlet Desperature of Slorry (t ₁) 1.68 millivolus thermocouple reading from Figure h is 83.6°F.
4.	Temperature line of filmry $(t_0 - t_1) = 97.6 - 83.1 = 1)_{0.2}$
5.	Cercent Solids (x) Setermined by & Cenco Coistore Calance . x = 12.9% Kaolin
б.	Specific heat of Clurry $(O_b) = C_1(-x) + C_3$ $C_1 = Neat Canacity of Cater - SU/1b - O_F$ $C_p = Cest Canacity of Kaclin - SU/1b - O_F$ $X = Seight Prection of solid C_b = 1 (1 - 0.129) + 0.22b (0.129) = 0.900 SU/1b - O_F$
7.	Slarry boost (qb) = (w6b) (Temp. Gise) qb = (1.365) (0.900) (12.2) = 17,120 ST /hr.
8.	Condensate Flow date (we) = 530 gracs/h min = 17.00 lbs/hr.
9.	Enthalpy of Stean (Hg) - Smottling Stean Calorimeter Sata - Temp. of Steam Sefere Wrifice = 266 °F Temp. of Steam After Swifice = 180 °F = 14.70° and Samum Sarometric Treampre = 29.65° Mg.
	ressure After writice = (29.65 - 10.70) $\frac{11.7}{25.52}$ = 0.05 psia Fluid Enthalpy of Steam of 1809F and 6.85 psia From Follter Magram S _s = 1138.0 STO/16.
10.	Inthalpy of Condensate (He) Condensate Temperature = 21.6°F Condensate Temperature = 21.0 CTL/16.
11.	Heat Miven op By Steam (Ha - Hc) = (1136 - 21h) = 921 SEAD of condensate.
12.	Steam west $(q_8) = (w_6) (\Delta H) = (17.00) (12h) = 16,100 bro/hr.$
Log Mean Temperature Difference Detween Slurry Bulk and Pipe Temperature (Δtm)

	Pipe Tem			
#	MV	o _F		
2	6.02	236	Bottom	of lxchanger
3	6.35	21,5		
4	6.30	244		
5	6.03	236		
6	6 .30	244		
7	6.37	246	Top of	Exchanger
Bulk	Inlet Tempera	ature = $83.4^{\circ}F$	-	
Bulk	Outlet Tempe:	rature = 97.6°F		

Plot Data As Shown On Following Page (Figure 16) Area Between Curves = $\frac{1}{16.90 \text{ in}^2}$ Therefore $\Delta t_m = \frac{1}{6.90 \text{ in}^2} \times \frac{20^{\circ}\text{F}}{10} = 156.5^{\circ}\text{F}$

14. Experimental Film Coefficient of Heat Transfer (hf)

$$h_{f} = \frac{q_{s}}{A\Delta t_{m}} = \frac{(16,180)}{(1.049\pi)} = 62.80 \text{ BTU/hr} - ft^{2} - \sigma_{F}$$

15. Slurry Density (Pb) Found From Figure 3.

$$X = 12.9\%$$
 Kaolin
Pb = 67.4 lb/ft³

16. Velocity of Slurry Through Heat Exchanger (Vb)

$$V_{b} = \frac{(w) (1hh)}{(e_{b}) (\pi D^{2}) (3600)}$$

$$V_{b} = (1365 1b/hr) (1hh in2
$$\frac{ft^{2}}{(67.4 1b/ft^{3}) (\pi) (1.0h9 in)^{2} (3600 \frac{sec}{hr})} = 0.937 \text{ ft/sec}$$$$

17. Bulk Viscosity (Kb)



К. 10 X 10 TO THE INCH 359-5 Ке∪ГРЕСА ESSER CO. КАРЕИИ. 8.А.

From Aboon Cate (a) Chear rate (*) = Mocalty (4) =	culate: (DBC) (1.569) 1b (100) (32.2) 100 ft2		
•	1200		

1.47.7 68 1	0.00	101200	47 BPC (19 7 (1)	
			lb	
Sec			<u>ít - sec</u>	
91 .2			0.0125	
1.71			0.0215	
314			0.0306	
157			0.0318	
1.12			0.581	
1.71			0.922	

above Sata slotted on Figure 7. So Fine, satural to Shear Late (7) is Fine Viscosity of slarey in Fige, satural to Shear Late (7).

$$\dot{\gamma} = \frac{\partial V_0}{D} = \frac{(8) (0.937)}{(1.009) 1} = \partial .7 \ \text{aec}^{-1}$$

wefer to Humme 7 and Find Corresponding welk discosity (45) Naving a closer late of 20.7 sec $^{-1}$.

18. Seyncles Suster (S. e) = Nbeb

$$N_{0} = \frac{1.019}{12} = \frac{0.937}{0.0395} = (67.1i) = 62.6$$

19. Thermal Conductivity of Slurry $(\delta_{\mathbf{b}})$

$$\mathbf{k}_{\mathbf{b}} = \mathbf{k}_{1} \left[\frac{2\mathbf{k}_{1} + \mathbf{k}_{p} - 2 \mathbf{\lambda}_{v} (\mathbf{k}_{1} - \mathbf{k}_{p})}{2\mathbf{k}_{1} + \mathbf{k}_{p} + \mathbf{k}_{v} (\mathbf{k}_{1} - \mathbf{k}_{p})} \right]$$

$$\mathbf{k}_{v} = 0.0\%1 : \mathbf{k} = 0.3\%1 : \mathbf{k}_{v} = 0.3\%1$$

$$K_{b} = \left[\frac{0.361 \ 2(0.361) + 0.110 - 2(0.0(1))(0.361 - 0.110)}{2(0.3(1) + 0.110) + 0.0((1)(0.3(1 - 0.110))}\right] = 0.314 \frac{50}{br} \frac{50}{rt^{2}}$$

20. Franktl Humber
$$(\mathbb{N}_{p})$$

$$H_{\rm Tr} = \frac{(b_{\rm A'b} - (0.900) (0.0895) (3000)}{(0.3100)} = 312$$

21. Stanton Sumber (N_{Ct})

$$\frac{n}{16} = \frac{hr}{10^{10} b^{10} b^{10}} = \frac{(62.3)}{(0.900) (0.937) (07.1) (3600)} = 3.07 \times 10^{-11}$$