

XPS – as a versatile tool in the research of the Department of Inorganic and Physical chemistry, UGent

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Content

- Thin film superconducting materials by sol gel chemistry
- Fibrous transistors, in cooperation with dep. of Textiles, Ugent
- Biogenic nanoparticles, in cooperation with labMET
- Supported metal oxide catalyst for liquid oxidation reactions

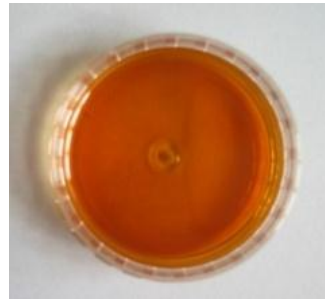
Principle



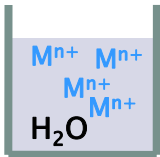
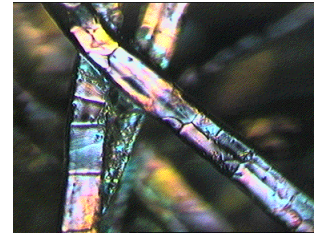
Deposition on substrates by
dipcoating, printing



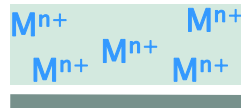
Gelation, $T \leq 60^\circ\text{C}$



Heat treatment under
controlled atmosphere



Sol



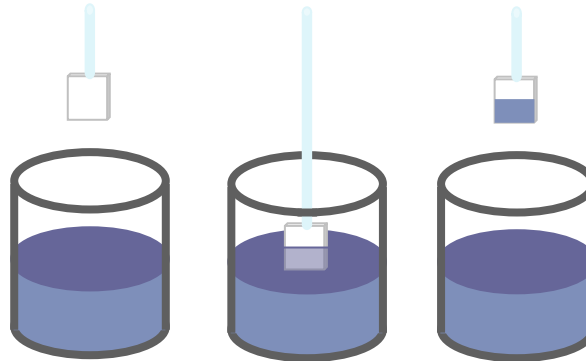
Gel



oxide thin film
(5-100 nm)

Deposition

- **dip-/spin-coating** from aqueous solutions
- **ink-jet printing** from water-based inks



liquid
precursor-
layer

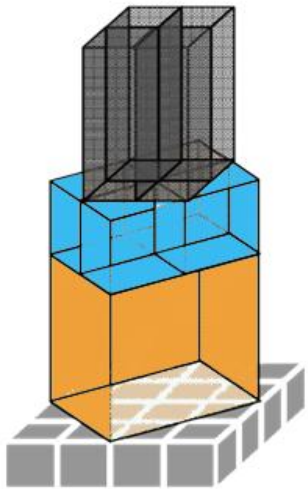
Analysis of solutions, gels and thin films:

- optical analysis : UV/Vis/IR/Raman spectroscopy
- rheology : viscosity, surface tension, contact angle, particle size, ...
- thermal analysis : TGA/DTA/DSC/TMA
- microscopy : AFM, optical, SEM/EDX, (HR)TEM with EELS, STEM, EDX
- structural analysis : XRD, pole figures, BET
- electromagnetic : resistivity measurements
- complexometry : potentiometric titrations
- XPS

Materials

- ceramic high T_c superconductor architectures on NiW tapes :
YBCO, CeO_2 , $\text{La}_2\text{Zr}_2\text{O}_7$
- TiO_2 for self-cleaning surfaces : on ceramic tiles and steel
- YSZ for thermal barrier coatings and solid oxide fuel cells
- mesoporous organosilica layers for low-k dielectrics (through ...)
- Diesel soot catalyst

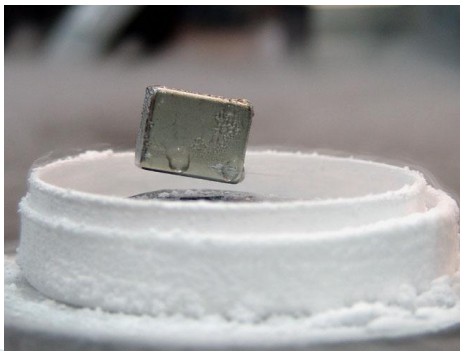
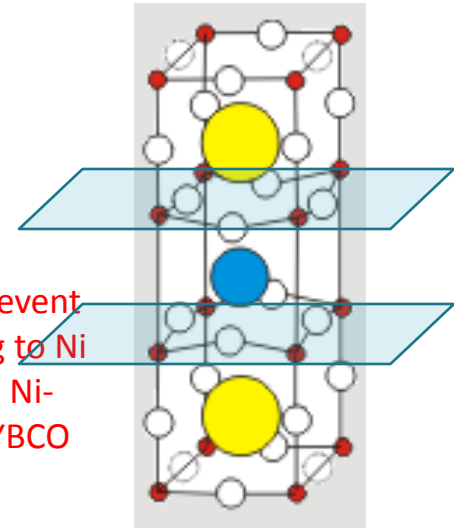
High T_c superconductor architecture :



- ➔ YBCO by dipcoating or printing from aqueous solutions
- ➔ CeO₂ (50-100 nm) by dipcoating from aqueous solutions
- ➔ LZO (250 nm) by dipcoating from aqueous solutions
- ➔ Textured, flexible NiW tape



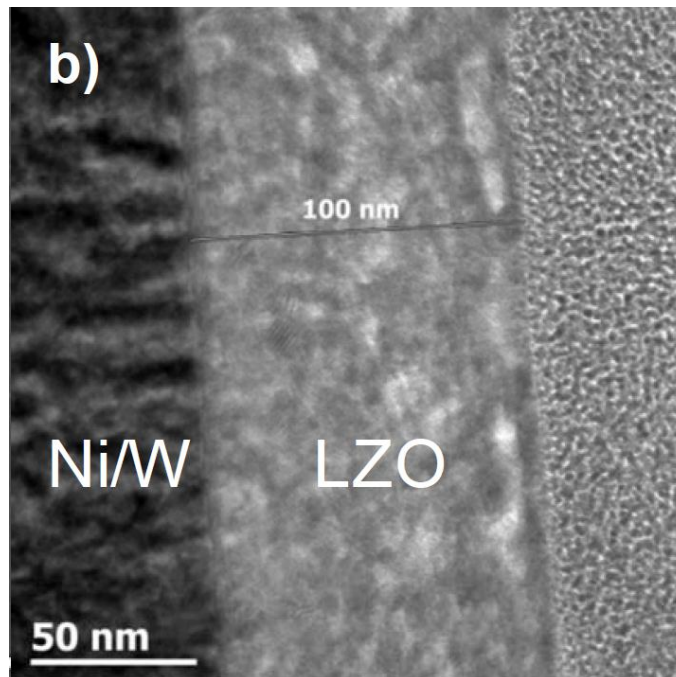
Buffer layers prevent O from diffusing to Ni substrate and Ni-diffusion into YBCO layer



Applications :

- resistance ~ 0 at -180°C
- second generation HTS wires
- fault current limiter
- magnets
- coils for renewable energy
- induction heaters ...

(HR)-TEM analysis of interface, surface,
crystallinity/texture

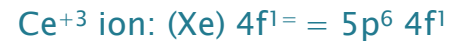
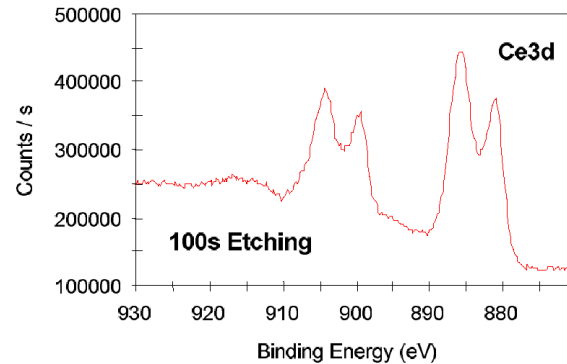
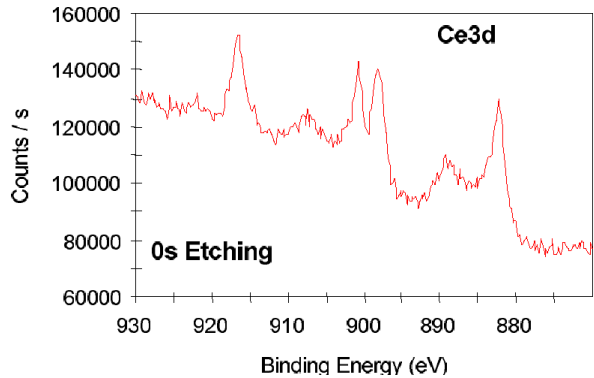


Cs-corrected TEM equipped with EELS, STEM, EDX ...

Sputter XPS analysis
depth profiling

- Ni - diffusion through buffer layers for \neq processing
- determination of layer thickness
- compositional analysis
- degree of oxidation

Study of oxidation state of Cerium and Ni-diffusion, a single layer pHd Greet Penneman (2007) Escalab 250 (VG Scientific)

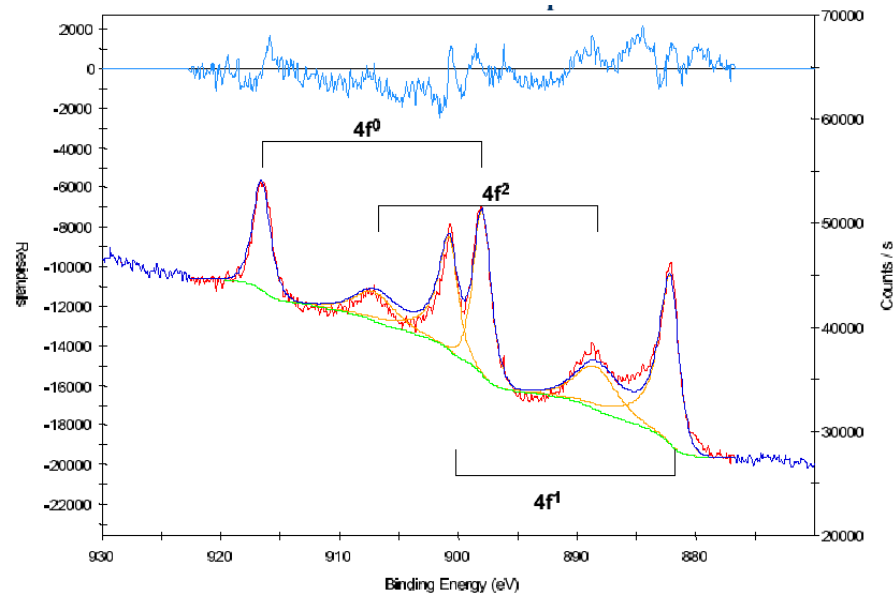


Ce-O is not 100 % ionic



Burrough's notation:		v ₀	v	v'	v''	v'''
Ce 3d _{5/2}	Origin	Ce ³⁺	Ce ⁴⁺	Ce ³⁺	Ce ⁴⁺	Ce ⁴⁺
	Shift (eV)	-36.1	-34.1	-30	-27.85	-18.3
	FWHM	4.11	5.77	3.76	2.69	3.96
		u ₀	u	u'	u''	u'''
Ce 3d _{3/2}	Origin	Ce ³⁺	Ce ⁴⁺	Ce ³⁺	Ce ⁴⁺	Ce ⁴⁺
	Shift (eV)	-17.8	-15.65	-13.65	-9.25	0
	FWHM	3.91	5.86	4.00	4.64	1.39

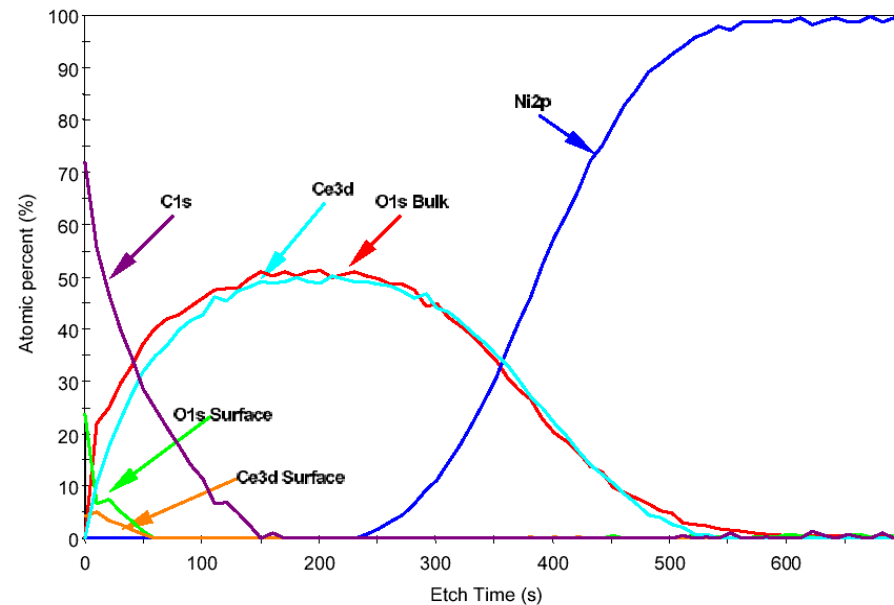
Study of oxidation state of Cerium and Ni-diffusion, a single layer pHd Greet Penneman (2007) Escalab 250 (VG Scientific)

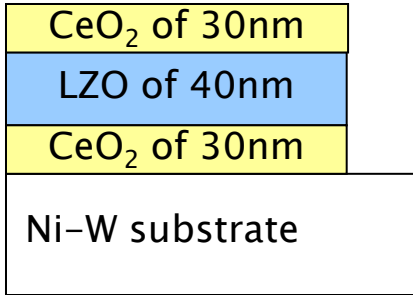


XPS narrow scan of the Ce 3d region showing the typical spectrum for Ce⁴⁺.

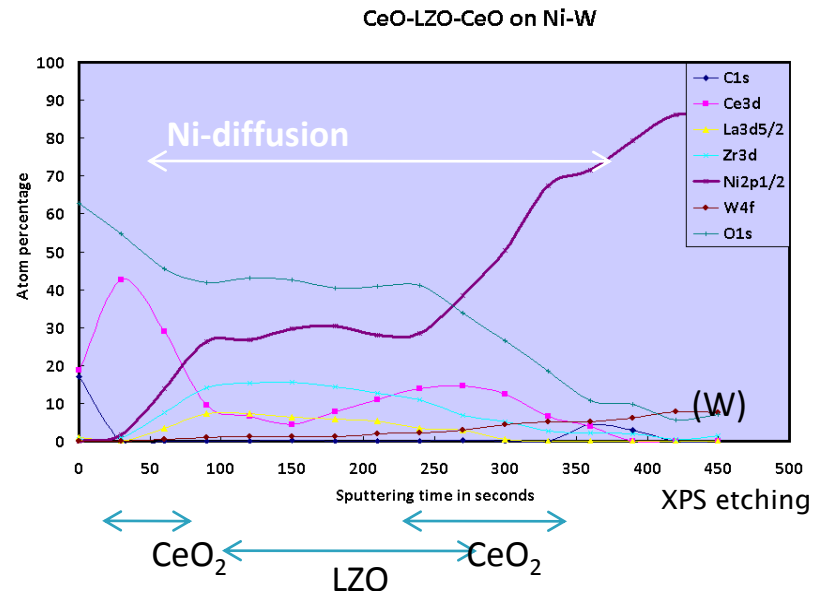
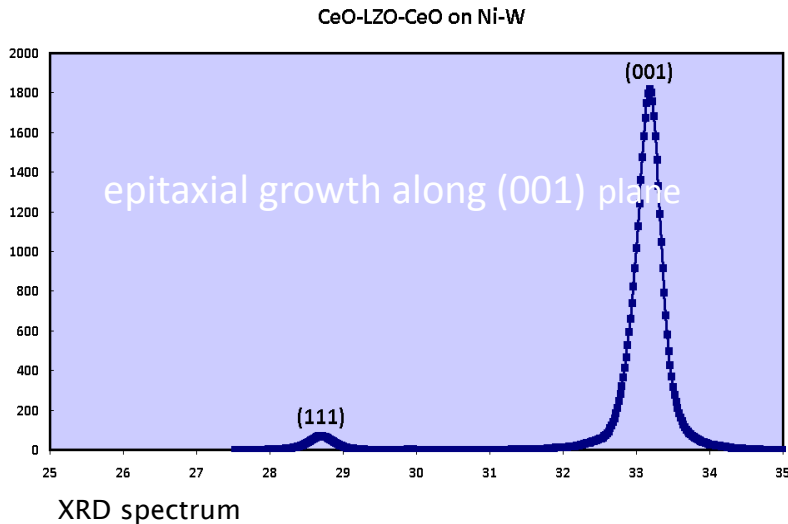
Fitting: constraints: doublet separation, peak position, FWHM, peak shape

Top: Ce (IV)
Core: Ce (III)
Due to sputtering?





The objective is to study the role of buffer layers in transferring the texture of Ni-W substrate to YBCO (XRD) and its effective role in prevention of Ni into YBCO



Biogenic metals: XPS analysis of cerium from organic origin

B. De Gusseme, Prof. Verstraete (LabMET)

Goal:

In a biological process cerium-ions are removed from a solution into a solid state. The question was to identify the oxidation number of the cerium in the solid state.

The paste like substance was spread onto a Sn-substrate. It is shown that the presence of cerium in the sample could be confirmed. The oxidation state of the cerium is Ce(III)

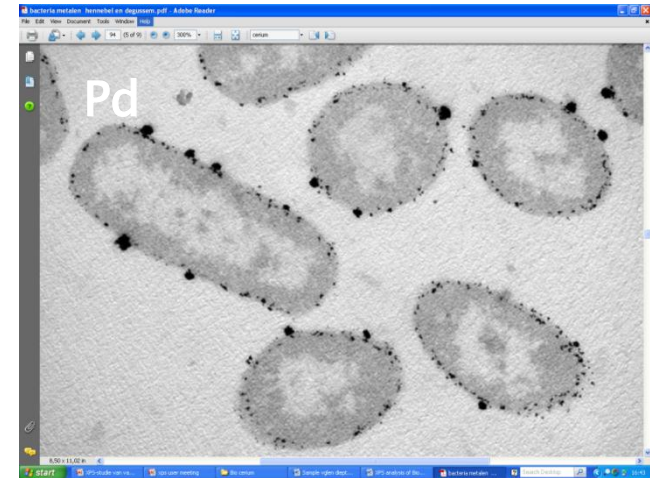


Illustration: TEM of zerovalent palladium particles precipitated at the cell surface of *Shewanella oneidensis*

Oxidation state	BE (eV)
Ce(IV)	898.4 $3d^{5/2}$
	916.9 $3d^{3/2}$
Ce(IV)	888.9 $3d^{5/2}$
	907.5 $3d^{3/2}$
Ce(IV)	882.3 $3d^{5/2}$
	901.0 $3d^{3/2}$
Ce(III)	880 $3d^{5/2}$
	899 $3d^{3/2}$
Ce(III)	886 $3d^{5/2}$
	904 $3d^{3/2}$

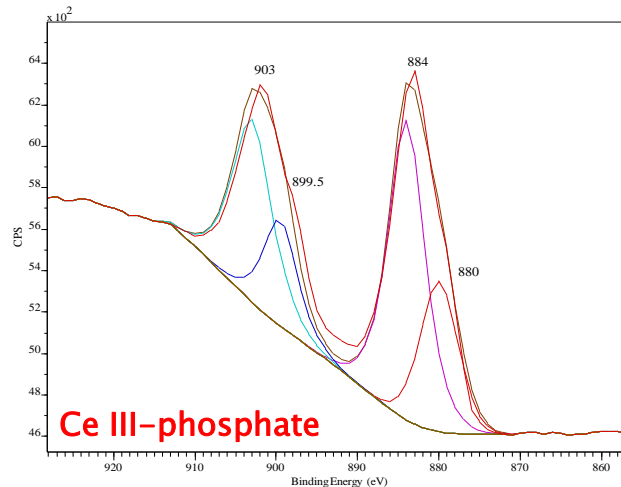
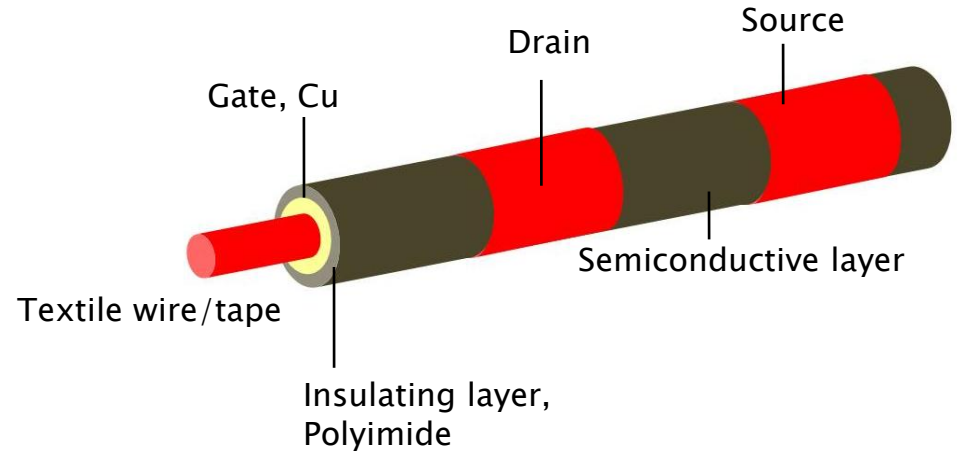
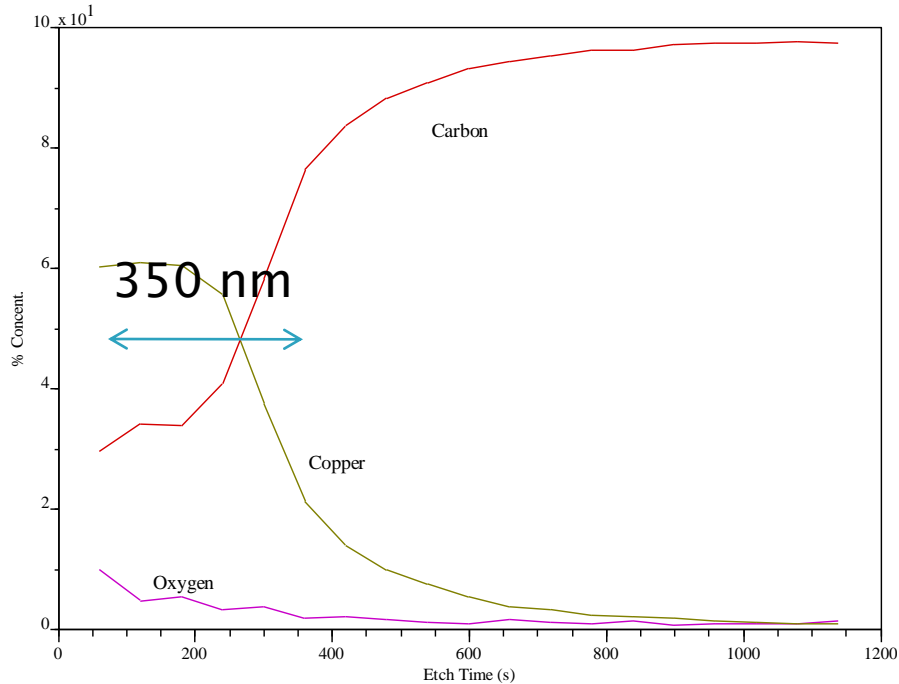


Table 1: Overview literature $Ce3d^{3/2}$ and $3d^{5/2}$

Textile based transistor

Dep. Textiles A. Swhwarz, prof. Van Langenhove

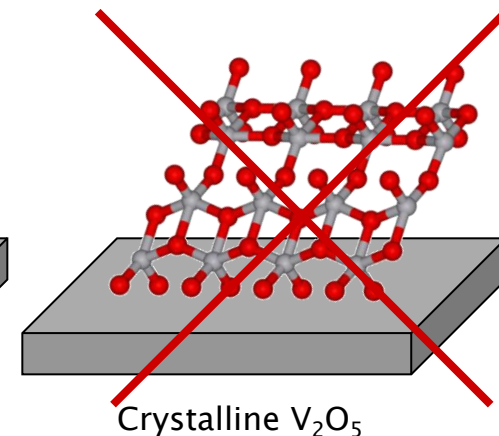
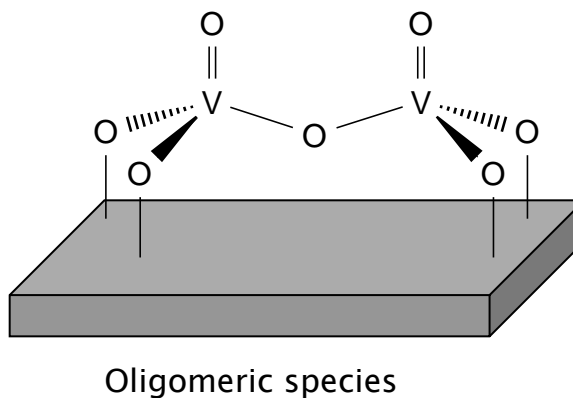
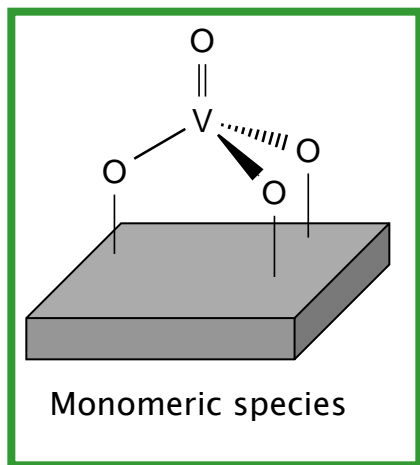
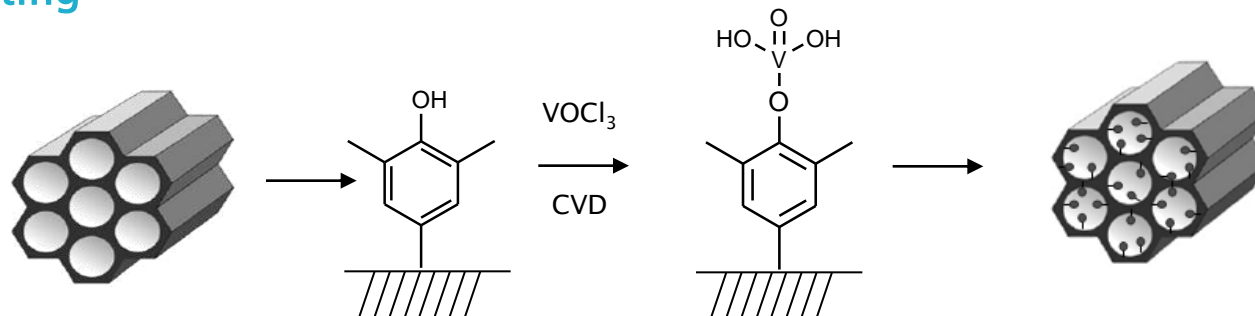


Determination of the thickness of the Cu-layer. Based on differences in etch-rate between Ta_2O_5 and Cu

Ultra-stable and zero leaching supported metal oxide catalyst for liquid oxidation reactions

Study of Vanadiumoxide on mesoporous fenol resins
Ilke Muylaert, prof. Van Der Voort

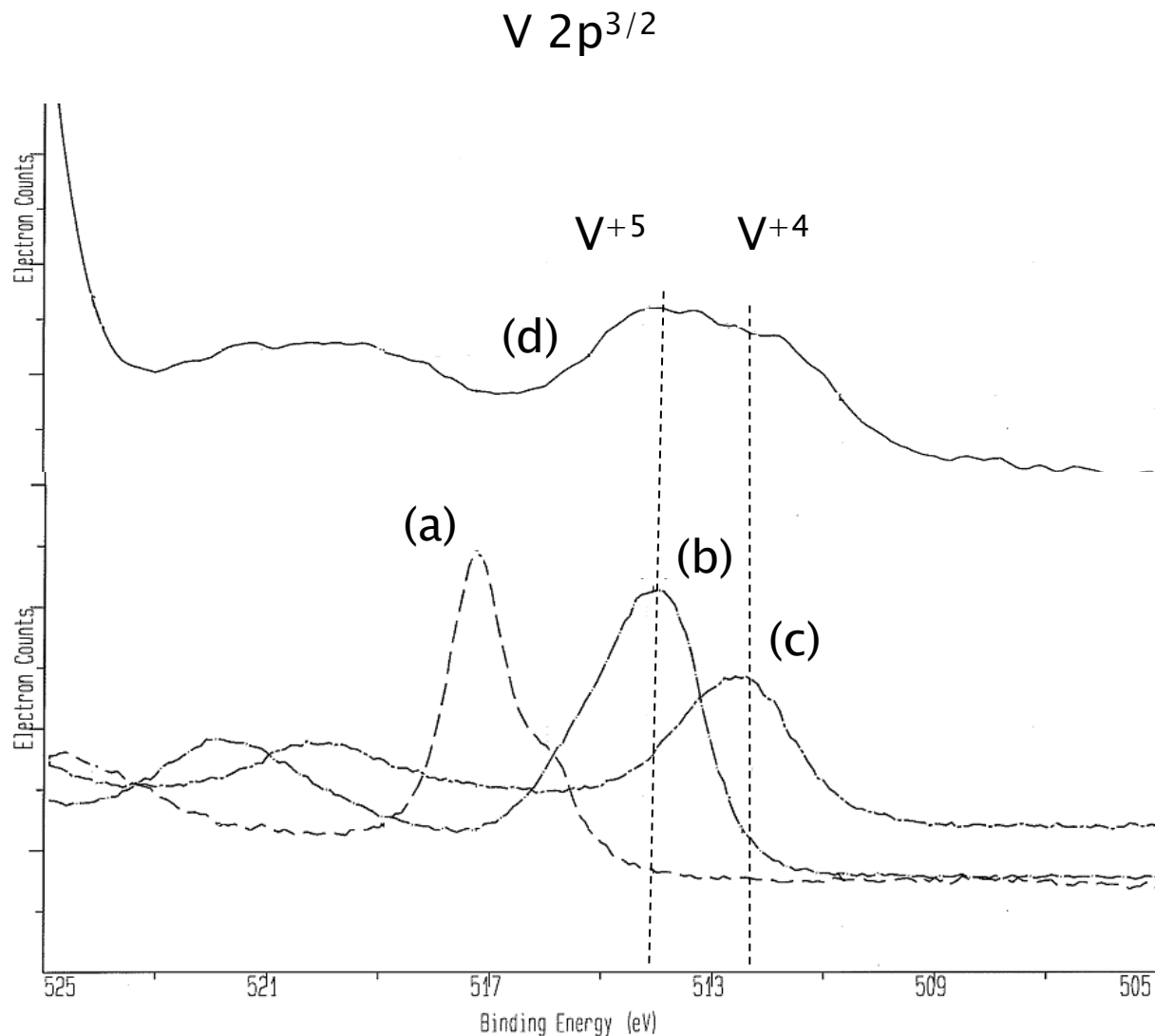
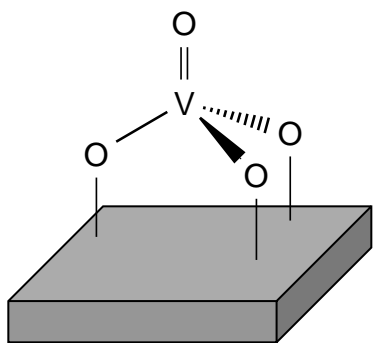
Vanadium grafting



Characterization – XPS

Determination oxidation state of Vanadium

- (a) V_2O_5 Crystalline
- (b) NH_4VO_3 +5
- (c) $VO(acac)_2$ +4
- (d) P/F Resin +5 (+4)

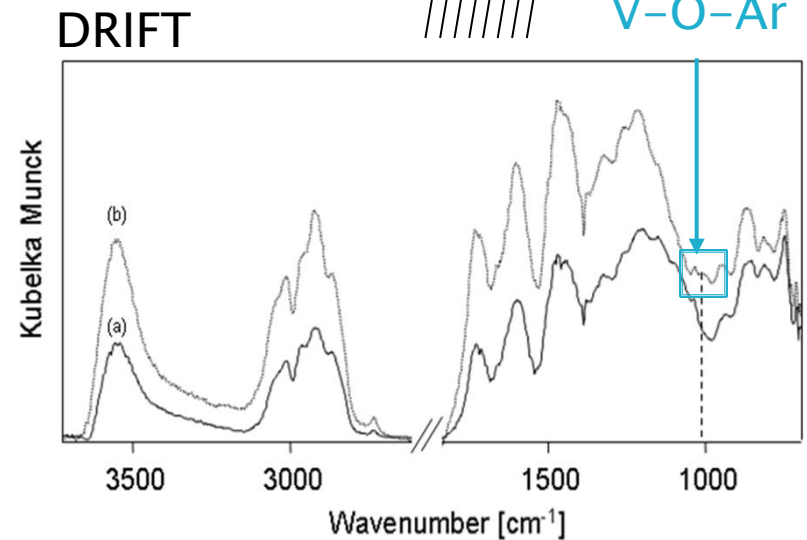
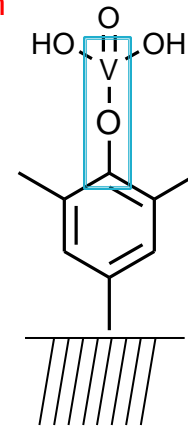
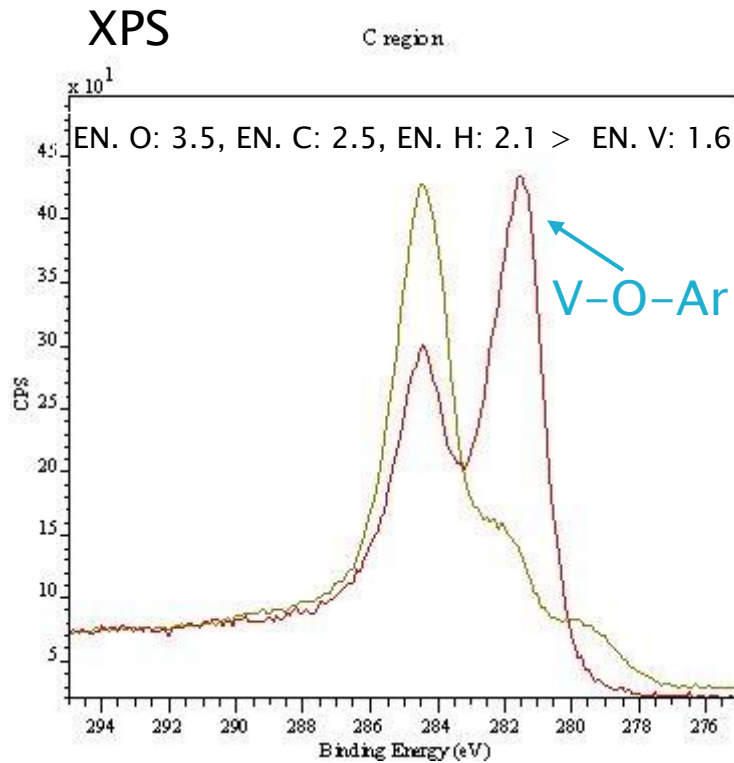


Structural information:

C- peak at low energy 281.5 eV: C-OV

Due to low electronegativity and soft center of vanadium atom

XPS and DRIFT spectra before and after grafting with Vanadium



Thanks