## Luminescence of CaGd<sub>2(1-x)</sub>Eu<sub>2x</sub>(WO<sub>4</sub>)<sub>4</sub> scheelites

<u>Katrien W. Meert</u><sup>a#</sup>, Philippe F. Smet<sup>a</sup>, Anne Bertha<sup>b</sup>, Joke Hadermann<sup>b</sup>, Dirk Poelman<sup>a</sup>,

<sup>a</sup>LumiLab, Department of Solid State Sciences, Ghent University, Krijgslaan 281-S1,9000 Gent, Belgium

Center for Nano- and Biophotonics (NB Photonics), Ghent University, Gent, Belgium

<sup>b</sup>EMAT, University of Antwerp, Groenenborgerlaan 171, 2020 Antwerpen, Belgium

Scheelites are ABO<sub>4</sub> compounds (A = alkali, alkaline-earth or rareearth element; B = Mo, W) with the most well-known scheelite being CaWO<sub>4</sub>. In scheelite related compounds there is a partial substitution of the A and/or B cation and crystals can be aperiodic in 3dimensional space. The advantage of this so-called incommensurate modulation is that varying the composition results in various order patterns which yield a wide range of materials with often good optical properties, good stability and a relatively simple preparation method [1]. Since the order directly affects the position of the luminescence centers, and thus the efficiency of the phosphor, there could be a relation between this order and the optical properties. Here, the main focus will be on the optical properties of CaGd<sub>2(1-x)</sub>Eu<sub>2x</sub>(WO<sub>4</sub>)<sub>4</sub>. The

luminescence and decay pathways of the materials are investigated over a wide concentration range. In this way the potential of the

temperature-dependent

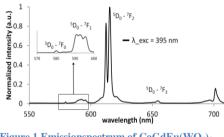


Figure 1 Emissionspectrum of CaGdEu(WO<sub>4</sub>)<sub>4</sub>

materials as LED-phosphors or even as thermometric phosphors are evaluated.

[1] C. Guo, H.K. Yang, and J.-H. Jeong, *Preparation and luminescent properties of the phosphor*  $MGd_2(MoO_4)_4$  : $Eu^{3+}$ . Journal of Luminescence, 2010. **130**: p. 1390-1393.