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Hughes, John; Howind, Torsten

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## **SCIENCE and ART: A Future for Stone**

**Proceedings of the 13<sup>th</sup> International Congress on the  
Deterioration and Conservation of Stone – Volume I**

**Edited by  
John Hughes & Torsten Howind**

# SCIENCE AND ART: A FUTURE FOR STONE

PROCEEDINGS OF THE 13<sup>TH</sup> INTERNATIONAL CONGRESS ON THE  
DETERIORATION AND CONSERVATION OF STONE

6<sup>th</sup> to 10<sup>th</sup> September 2016, Paisley, Scotland

VOLUME I

Edited by  
John J. Hughes and Torsten Howind



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Cover image: The front door of the Paisley Technical College building, now University of the West of Scotland. T.G. Abercrombie, architect 1898. Photograph and cover design by T. Howind.

# EXPERIMENTAL STUDY OF THE AGEING OF BUILDING STONES EXPOSED TO SULFUROUS AND NITRIC ACID ATMOSPHERES

S. Gibeaux<sup>1\*</sup>, C. Thomachot-Schneider<sup>1</sup>, A. Schneider<sup>2</sup>, V. Cnudde<sup>3</sup>,  
T. De Kock<sup>3</sup>, V. Barbin<sup>1</sup> and P. Vazquez<sup>1</sup>

## Abstract

During the last few decades, due to remediation procedures, SO<sub>2</sub> emissions in the atmosphere have decreased, unlike NO<sub>x</sub>. Air pollution has changed. Indeed, the aim of this research is to assess the effect of NO<sub>x</sub> and their interactions with SO<sub>2</sub> on stones, particularly on limestones used in Champagne-Ardenne (France) during the restoration processes. Three French building limestones (Courville, Dom and Savonnières) and one reconstituted stone were exposed during 28 days to four strong acid atmospheres i.e. two H<sub>2</sub>SO<sub>3</sub> solutions with different concentrations and two mixed atmospheres with different proportions of HNO<sub>3</sub> and H<sub>2</sub>SO<sub>3</sub>. These tests produced an intensive acid attack on the stone, allowing the observation of short-term salt precipitation and the evolution of stone properties. Each day, one sample was removed from the acid atmosphere to measure the concentration of SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> by ion-chromatography. The surface changes were assessed before and after the tests by 3D scanning and observations with electron microscopy. X-ray microtomography has been performed in the Centre for X-ray Tomography (UGCT) and the Department of Geology at Ghent University (Belgium) in order to observe the penetration of salts and the consequences in stones porosity. First observations showed that exposure to acid atmosphere, led to gypsum efflorescences. Obvious colour changes occurred in all tests. Salt crystallization entailed a change in the porous system, which was evidenced by 3D, mercury porosimetry and X-ray microtomography. Difference between weathered and fresh stone was highlighted by Ion chromatography analyses.

**Keywords:** acid atmosphere ageing tests, limestone, non-destructive testing (NDT), sulphates, nitrates

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<sup>1</sup> S. Gibeaux\*, C. Thomachot-Schneider V. Barbin and P. Vazquez  
GEGENAA, University of Reims, France  
soizic.gibeaux@univ-reims.fr

<sup>2</sup> A. Schneider  
IFTS, University of Reims, France

<sup>3</sup> V. Cnudde and T. De Kock  
Dept. Geology and Soil Science, Ghent University, Belgium

\*corresponding author

## 1. Introduction

The 21<sup>st</sup> century is known as a financial crisis period and an air pollution evolution era (CITEPA 2015). In this context we have to respect sustainability of our monuments choosing the best restoration processes while finding the cheapest solution. The two determining factors to take into account are the choice of the material and the type of environment. One of the most important weathering agent is the dry deposition of gaseous pollutants on stones which results in salt weathering (Török *et al.*, 2004; Charola *et al.*, 2007; Grossi *et al.*, 2007; Monna *et al.*, 2008). Crust formation, blackening, yellowing and carbonate dissolution are the consequences of this phenomenon (Grossi *et al.*, 2007; Monna *et al.*, 2008). Sulphur dioxide is particularly known as the main gaseous pollutant responsible for salt weathering but last decades, ratios in emissions of pollutants have changed with a SO<sub>2</sub> decrease and an increase in NO<sub>x</sub>). The consequences of these changes in atmospheric composition on the type of salt crystallization are scarcely investigated (Sikiotis and Kirkitsos, 1995; Camaiti *et al.*, 2007; Vazquez *et al.*, 2016) unlike the effects of NO<sub>x</sub> as catalysts for the SO<sub>2</sub> oxidation (Massey 1999). The alteration processes of the stones also depend on the material characteristics like chemical composition, porosity and texture (Massey 1999; Grossi *et al.*, 1995; Vazquez *et al.*, 2015). The goal of this study is to compare, quantitatively and qualitatively, the reactivity of fresh and weathered stones exposed to different acid atmospheres.

## 2. Material

### 2.1. Samples

Three natural stones and one artificial stone were tested. Samples of reconstituted stone and Savonnières limestone are not fresh because they stem from historical buildings, respectively from Orval Abbey (Belgium) and Saint-André Church (Reims, France), unlike the Courville and Dom samples which are fresh stones particularly used in the North of France and the South of Belgium (Fronteau 2000; Thomachot *et al.*, 2011; 2015). These stones present different composition and porosity detailed in Tab. 1.

Tab. 1: Characteristics and composition of the four studied stones.

Stone	Limestone type	Colour	CaCO <sub>3</sub> wt. %	SiO <sub>2</sub> wt. %	Porosity %	Named as
Reconstituted Stone	Lithoclastic	Orange	94.6	1.09	23	RS
Savonnières	Oolitic vacuolar	Grey	90.3	4.90	30	S
Courville	Micritic fine	Russet	86.9	6.64	16	C
Dom	Oolitic to bioclastic	Yellow ochre	96.3	0.68	20	D

### 2.2. Ageing tests





The dry deposition process was evaluated by means of accelerated ageing tests, based on the standard UNE-EN 13919:2003 procedures (CEN, 2002). They were performed with the atmospheres created by four solutions using nitric and/or sulphurous acids in different proportions (Vazquez *et al.*, 2016) as described in Tab. 2.

Tab. 2: Ageing tests solutions concentrations.

	HNO <sub>3</sub> , mol l <sup>-1</sup>	H <sub>2</sub> SO <sub>3</sub> , mol l <sup>-1</sup>	HNO <sub>3</sub> /H <sub>2</sub> SO <sub>3</sub> ratio	Named as
Solution 1	4.8 10 <sup>-1</sup>	1.7 10 <sup>-1</sup>	3/1	H <sub>2</sub> SO <sub>3</sub> < HNO <sub>3</sub>
Solution 2	1.4 10 <sup>-1</sup>	5.6 10 <sup>-1</sup>	1/4	H <sub>2</sub> SO <sub>3</sub> > HNO <sub>3</sub>
Solution 3	-	5.6 10 <sup>-1</sup>	-	H <sub>2</sub> SO <sub>3</sub> HC
Solution 4	-	1.7 10 <sup>-1</sup>	-	H <sub>2</sub> SO <sub>3</sub> LC

The stock solution for nitric acid was Panreac 69% w/w M = 14.4 g mol<sup>-1</sup>, and for sulphurous acid was Panreac 6% w/w M = 6 g mol<sup>-1</sup>.

Tab. 3: Samples type in function of analysis.

Number of samples	Size	Analysis	Analysis frequency
28		Ion chromatography	One per day
7	1×1×0.4 cm	SEM/EDS	After 1, 3, 5, 8, 15, 22 and 28 days
1	 1×1×2 cm	Hg porosimetry	At the end of the experiment
1	 5×5×1 cm	Macroscopic evaluation Weight 3D scanning	At the end of the experiment
1	 Ø 0.4 cm	X-ray Microtomography	After 1, 10 and 28 days

### 3. Analytical methods

Dry deposition on stones lead to salt precipitation and stone dissolution. Different technics were used to assess change of the samples (Tab. 2). Photos and images of the upper face were taken with a Nikon Coolpix 4500 and a SEM Hitachi TM-1000 Tab. Top equipped with an EDS. The roughness change was assessed with the ZScan 700Cx device for the 3D scanning and the 3D Reshaper software for the processing of the point cloud with an accuracy of 0.05 mm and a resolution of 0.1 mm. Quantity of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> was determined by Ion Chromatography (IC) (Dionex ICS 2000). The high resolution X-ray Computed Microtomography was performed for RS and S tested with H<sub>2</sub>SO<sub>3</sub> < HNO<sub>3</sub> using the in-house scanner HECTOR (Masschaele *et al.*, 2013) developed in the Centre for X-ray Tomography of Ghent University (UGCT). The tomographs were reconstructed with the Octopus Reconstruction software (Inside Matters, Belgium; Vlassenbroeck *et al.*, 2007), while the reconstructed images were subsequently analyzed using Octopus Analysis, formerly called Morpho+ (Brabant *et al.*, 2011). Porosity characterization was obtained using an Hg intrusion porosimeter (MIP) Micromeritics Autopore IV 9500.

## 4. Results

### 4.1. Visual and SEM observations

The observation showed a yellowing of every stone for every test. Only a located blackening of RS and D appeared in the  $\text{H}_2\text{SO}_3 < \text{HNO}_3$  test and slightly in the  $\text{H}_2\text{SO}_3$  LC test. In the former, white spots arise on C sample. The SEM images highlighted different salt crystal habits (Fig. 1) independently of the time of exposure. Moreover no evolution in the crystals shape or size was detected between the tests or the samples.

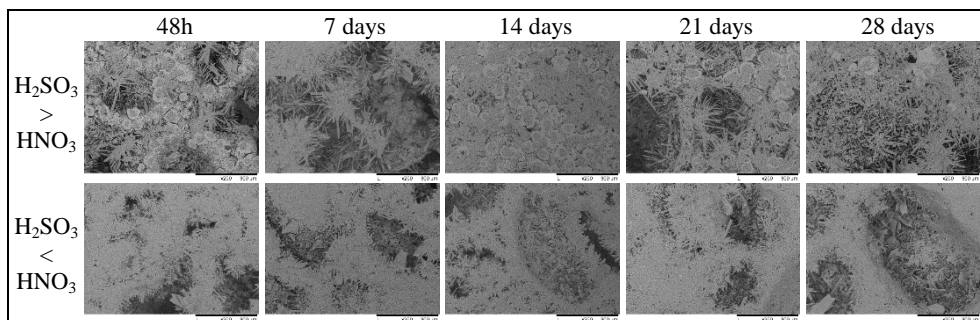


Fig. 1: SEM images of Savonnières sample during the mixed atmospheres tests.

### 4.2. Weight changes

After the  $\text{H}_2\text{SO}_3 < \text{HNO}_3$  and  $\text{H}_2\text{SO}_3$  LC tests, the weight gain did not exceed 0.5% with few differences between the stones. The weight gain was notable after the  $\text{H}_2\text{SO}_3 > \text{HNO}_3$  and  $\text{H}_2\text{SO}_3$  HC tests (from more than 0.5 to 2.0%), the latter with the highest values for all the stones except for S.

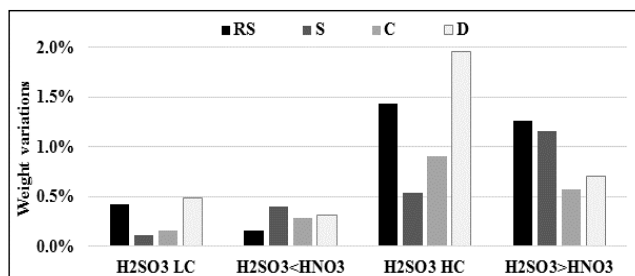


Fig. 2: Weight variations between initial and final states.

### 4.3. Roughness

Dry deposition created a new surface roughness. No spatial differences were observed among the tests. Fig. 3a showed the numerical subtraction of the surface before and after the tests. Green colour indicated that no change took place on the surface. Salt crystallized mainly in the sample's limits and in preexistent defects and pores (red colour). Dissolution occurred in carbonate grains in the case of RS (blue colour). In other stones, dissolution could be observed disseminated through the surface without a defined pattern.



#### 4.4. X-ray Computed Micro tomography ( $\mu$ CT)

After solely one day of exposure to the  $\text{H}_2\text{SO}_3 < \text{HNO}_3$  test, a homogeneous crust and some areas showing dissolution process appeared on RS and S. After 10 and 28 days, the growth crust was negligible. The fine crust large of around  $50\mu\text{m}$  for RS and  $100\mu\text{m}$  for S seemed to be mainly composed of needle-like crystals (Fig.3b).

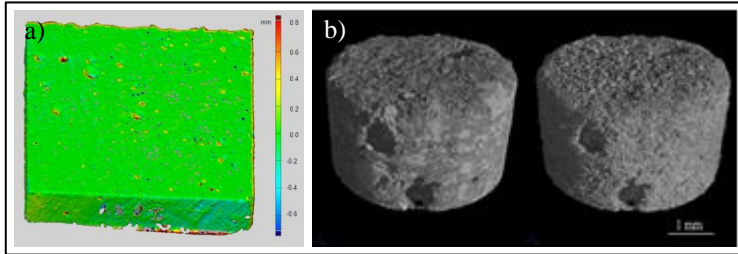


Fig. 3: a) Roughness measurements (subtractions between initial and after test states) on RS after  $\text{H}_2\text{SO}_3 < \text{HNO}_3$ ; b) 3D reconstruction of RS  $\mu$ CT scans before (left) and after 28 days (right) exposed to the  $\text{H}_2\text{SO}_3 < \text{HNO}_3$  test.

#### 4.5. Pore access radii

The total porosity varied slightly, between 0 and 6%, unlike the pore radius access (Fig.4). The volume of pore access radii could increase according to the stone and the test: C for every test except  $\text{H}_2\text{SO}_3 > \text{HNO}_3$ , S for  $\text{H}_2\text{SO}_3 > \text{HNO}_3$  and  $\text{H}_2\text{SO}_3$  HC tests, D for  $\text{H}_2\text{SO}_3 < \text{HNO}_3$  tests. Some pore access radii could also disappear like the macropores of RS and S submitted to  $\text{H}_2\text{SO}_3$  LC test and those of D submitted to  $\text{H}_2\text{SO}_3$  LC and HC tests. The  $\text{H}_2\text{SO}_3$  LC led to the closing of pore access radii of all stones except the Courville stone for which the pore access radii widened ( $\text{H}_2\text{SO}_3 > \text{HNO}_3$ ).

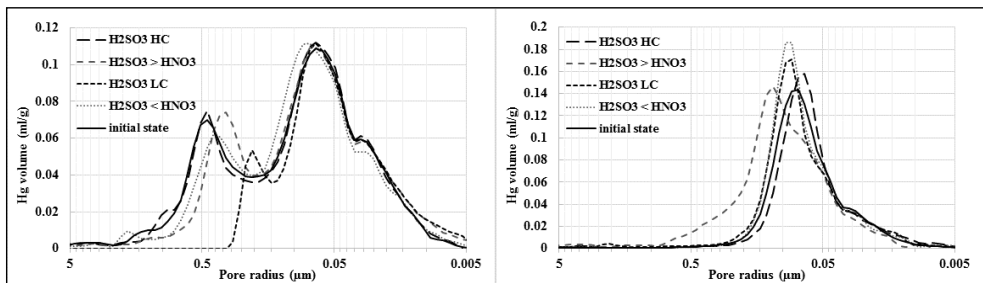


Fig. 4: Pore distribution of RS (left) and C (right) samples before and at the end of the tests.

#### 4.6. Ion chromatography

S and RS were initially S-rich, because they were not fresh stones (from 5000 to 6000 mg/g, Fig. 5). In all tests, the  $\text{NO}_3^-$  concentration was less than 250 mg/g, therefore negligible if compared to  $\text{SO}_4^{2-}$  concentration which could reach up to 8000 mg/g. For C and D samples, there was a remarkable increase of the  $\text{SO}_4^{2-}$  concentration after the first

day, after which the concentration stabilized. The  $\text{SO}_4^{2-}$  concentration did not change during mixed atmospheres exposures for RS and S samples.

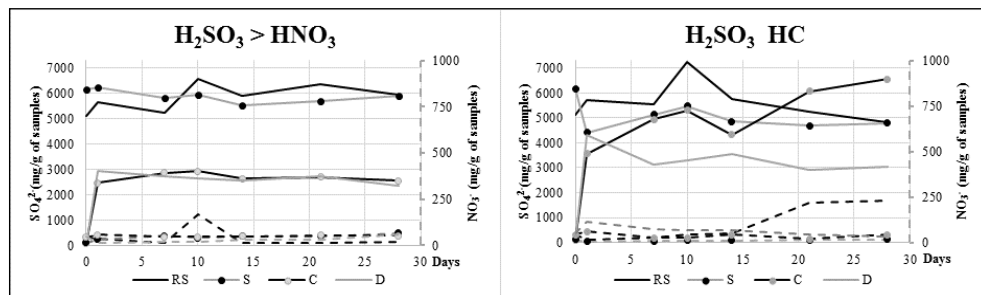


Fig. 5:  $\text{SO}_4^{2-}$  (solid lines) and  $\text{NO}_3^-$  (dotted lines) concentrations.

## 5. Discussion and conclusion

### 5.1. Type of atmosphere

Every test led to salt efflorescence growth on each stone. It formed a fine and homogeneous crust highlighted by the X-ray computed microtomography. Additionally, 3D measurements revealed the location of crystallization in the stone defects while dissolution occurred mainly at the contact with calcite elements. Weight variations showed the reactivity between stones and acid atmospheres seemed directly linked to the quantity of  $\text{H}_2\text{SO}_3$  regardless pure or mixed with  $\text{HNO}_3$ . MIP showed that pure  $\text{H}_2\text{SO}_3$  atmospheres led to the closing of pore access radii while the mixed atmospheres ( $\text{H}_2\text{SO}_3$  combined with  $\text{HNO}_3$ ) led to the widening of the pore access radii. The ion chromatography showed the maximum production of sulfates during the first day of exposure regardless the test and the stone. A low quantity of nitrates was measured in both mixed test because of their high instability in water and their hygroscopicity (Cheng *et al.*, 1987). Or nitrates play a catalytic role in the oxidation of sulfur compounds. The variations of  $\text{SO}_4^{2-}$  concentrations were similar from one test to another for each stone except for Courville limestone. The bad correlation between the weight variations and salts quantity can be due to the difference in size scale of samples.

### 5.2. Behaviour of fresh stone and exposed stone

The presence of sulphates in the initial state of the tested weathered samples did not differ in the degree or type of decay to the fresh ones. The only difference was found in ion chromatography analyses. The initial concentration of sulphates in the initially weathered stones (RS and S) is much higher than in the fresh ones (C, D). However, this high concentration in RS and S remained constant through the whole test, while in C and D sometimes an increase was observed after the first day, and then the concentration kept almost constant.

### 5.3. Suitable techniques in the study of stone decay by exposure to acid atmosphere

The exposure to an acid atmosphere enriched in  $\text{SO}_2$  entailed the formation of a gypsum crust from the very first day of test. The different techniques to assess the evolution of the decay gave rich but varied information about the process. Visual observation gave

information about the general changes on the stone surface, such as colour changes or crystal formation. On one hand, visual observation resulted fast and suitable techniques to evaluate the decay (Vazquez *et al.*, 2015). On the other hand, SEM observations only showed the crystal shapes since the salt crust was formed during the first day. Weight variations give information about the new precipitated crystals. However, this information should be taken with care. Dry deposition in controlled atmosphere involves precipitation and dissolution, two processes that depend strongly on stone properties. A weight gain reflects that salt precipitation was more important than dissolution. To have detailed information about where precipitation or dissolution occurs, 3D measurements is an optimal technique to study the sample's surface (Vazquez *et al.*, 2016), MIP to assess the variation in the pore radii access and the  $\mu$ CT to observe exactly in which grain or pore the process took place (Dewanckele *et al.*, 2012). Ion chromatography gives a general idea about the anion quantity, however, only the strong changes in concentration must be taken into account due to the heterogeneous reactivity of each sample.

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