

Development, implementation and use of a generic nutrient recovery model (NRM) library

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Quality Research**

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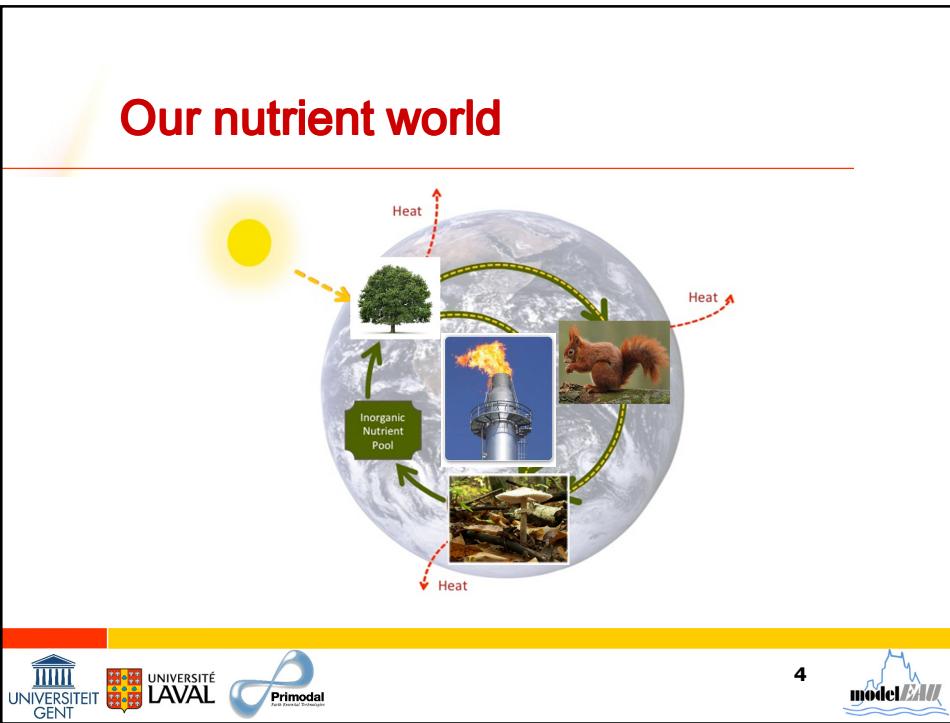
Presentation outline

-  Problem statement
-  Objectives
-  Model development methodology
-  Model simulation & validation results
-  Recommendations for research
-  Conclusions & perspectives

PROBLEM STATEMENT



Our nutrient world



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Our nutrient world

The diagram illustrates the global nutrient cycle, centered around an 'Inorganic Nutrient Pool' represented by a globe. Key components include:

- Human Impact:** A person spraying a field leads to a yellow lightning bolt pointing to a tree, labeled 'Heat'.
- Natural Processes:** A squirrel on a pile of trash leads to a yellow lightning bolt pointing to a fire at an oil refinery, labeled 'Heat'.
- Industrial Impact:** A pipe dumping waste into a river leads to a yellow lightning bolt pointing to a forest, labeled 'Heat'.
- Organic Nutrient Pool:** The globe shows a green arrow indicating the flow of nutrients between the inorganic pool and organic forms like trees, mushrooms, and a squirrel.

Logos at the bottom left: Universiteit Gent, Université Laval, Primodal. Number 5 is at the bottom right.

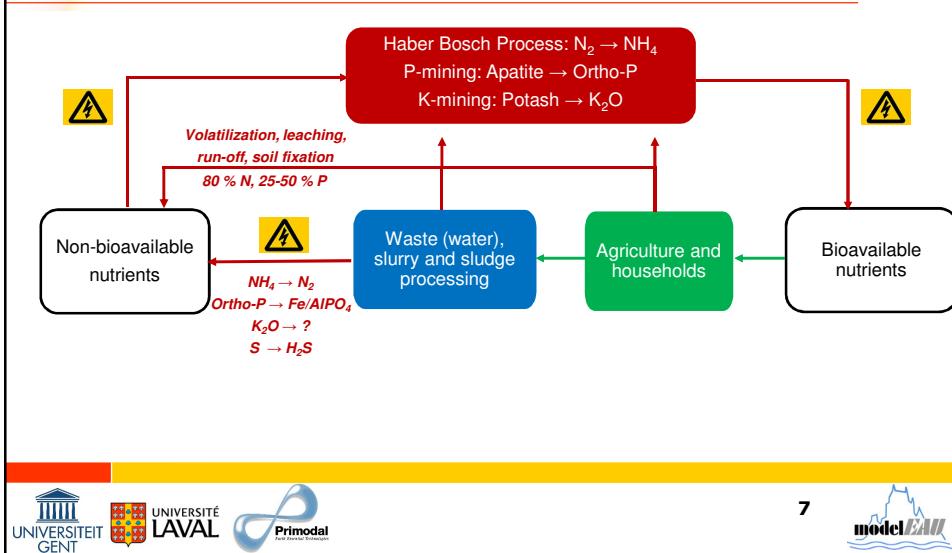
Our nutrient world

The photographs illustrate the visible impacts of nutrient pollution:

- A coastal area where red-tide algae has turned the water and sand a vibrant red color.
- A dead fish lying on a green, algae-covered shore.
- A city skyline obscured by thick, hazy air, likely due to nutrient-related atmospheric pollution.

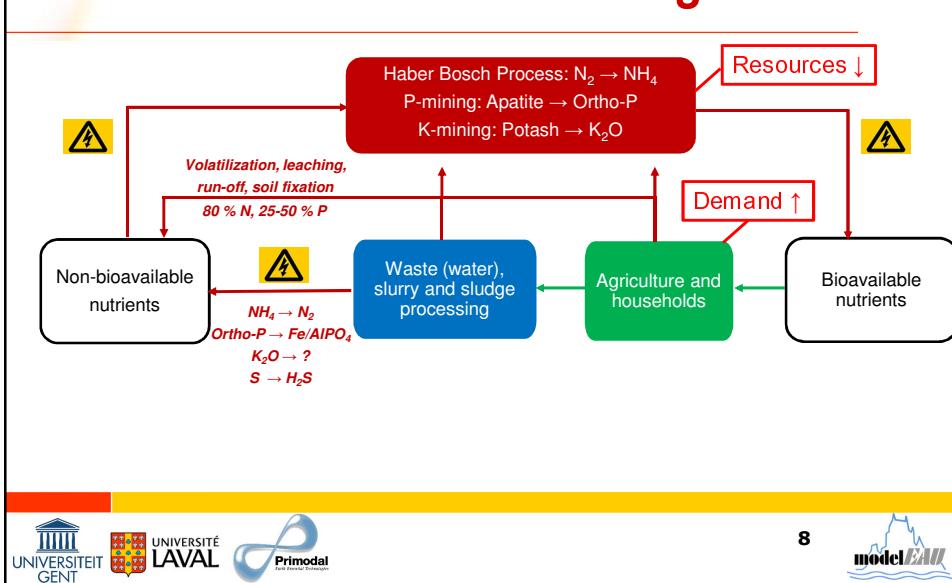
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Need for sustainable resource management



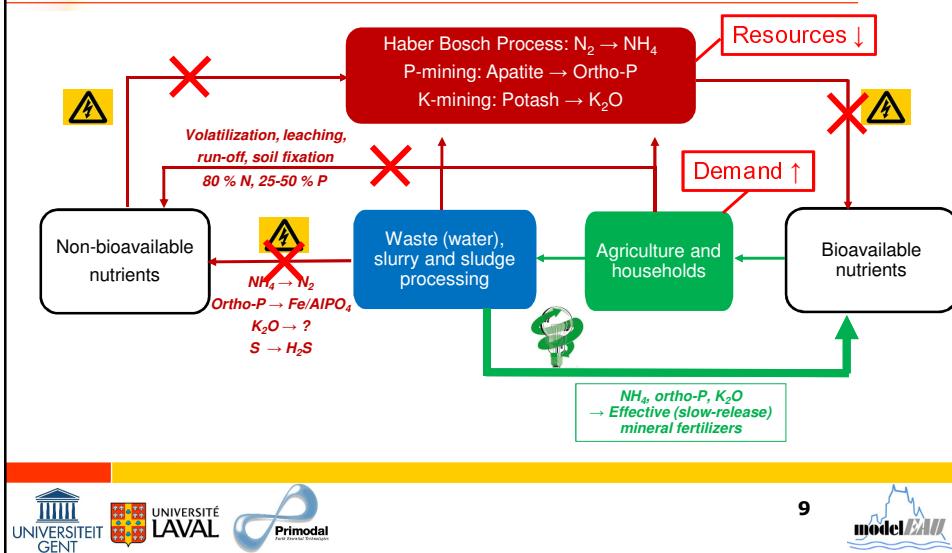
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Need for sustainable resource management



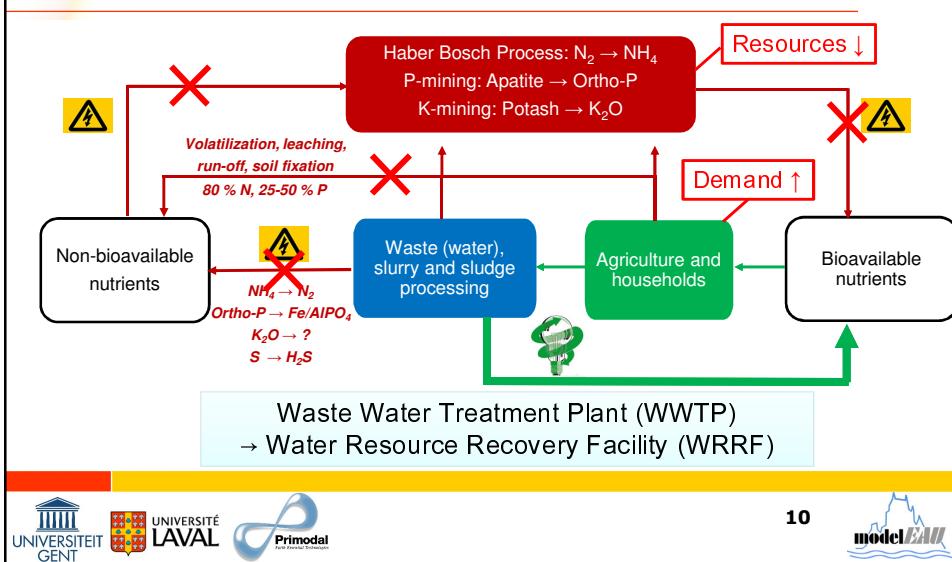
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Need for sustainable resource management



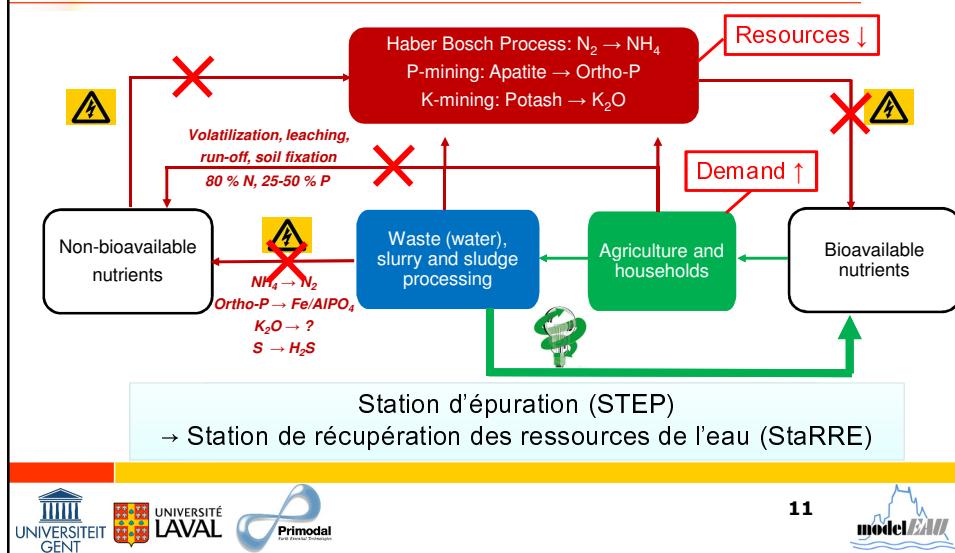
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Need for sustainable resource management



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Need for sustainable resource management



Nutrient recovery processes

- Precipitation \rightarrow struvite, calciumphosphates



Nutrient recovery processes

- Precipitation → struvite, calciumphosphates
- Ammonia stripping → NH₃



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Nutrient recovery processes

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- Membrane filtration → H₂O, N-K concentrates



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Nutrient recovery processes

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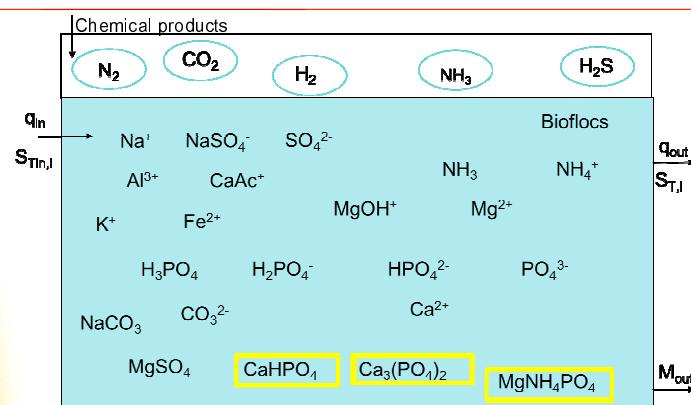
⇒ Mainly physicochemical unit processes !



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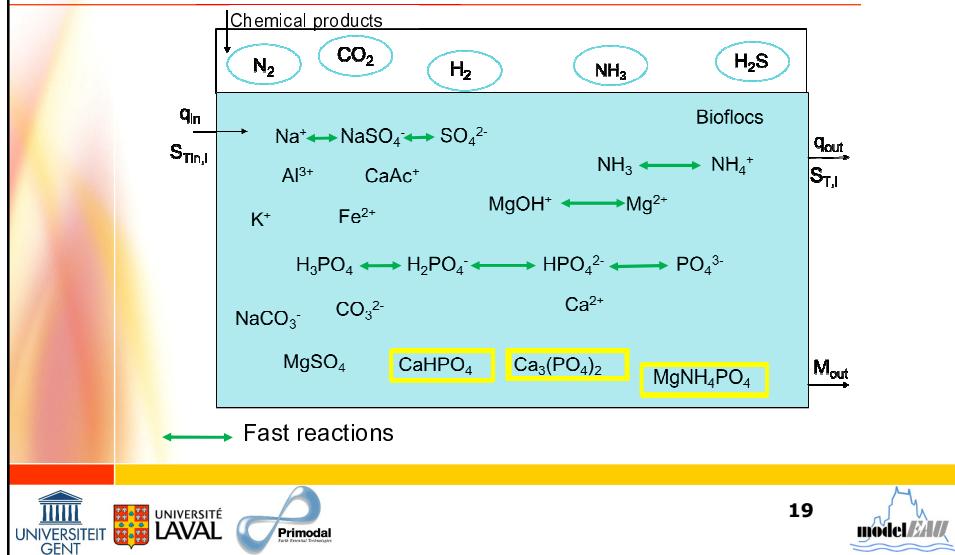
Prerequisite = marketable fertilizers



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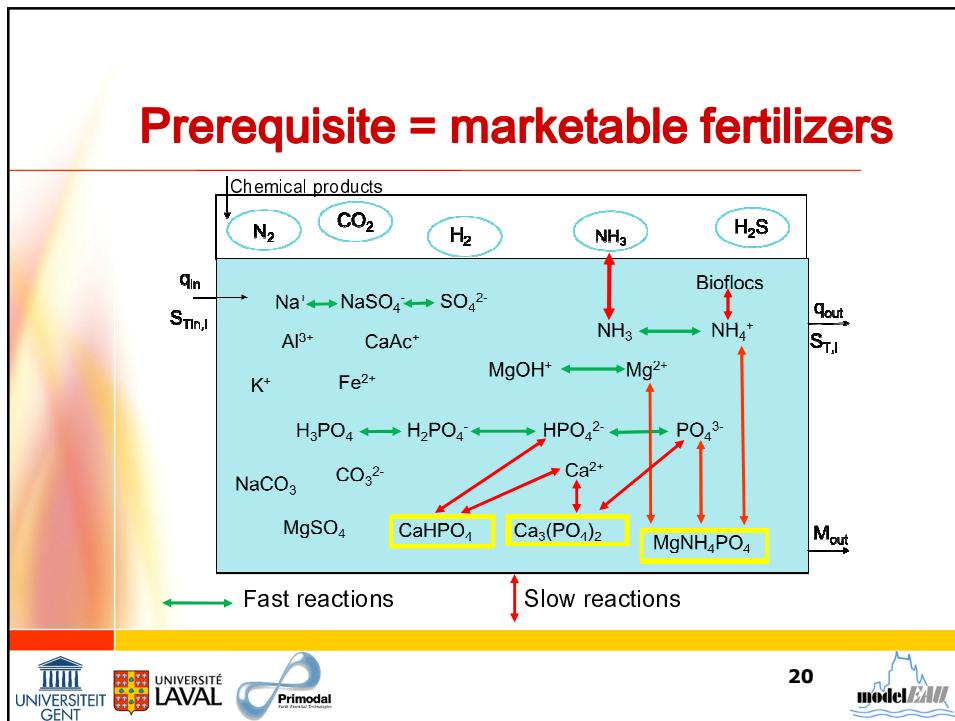
Prerequisite = marketable fertilizers



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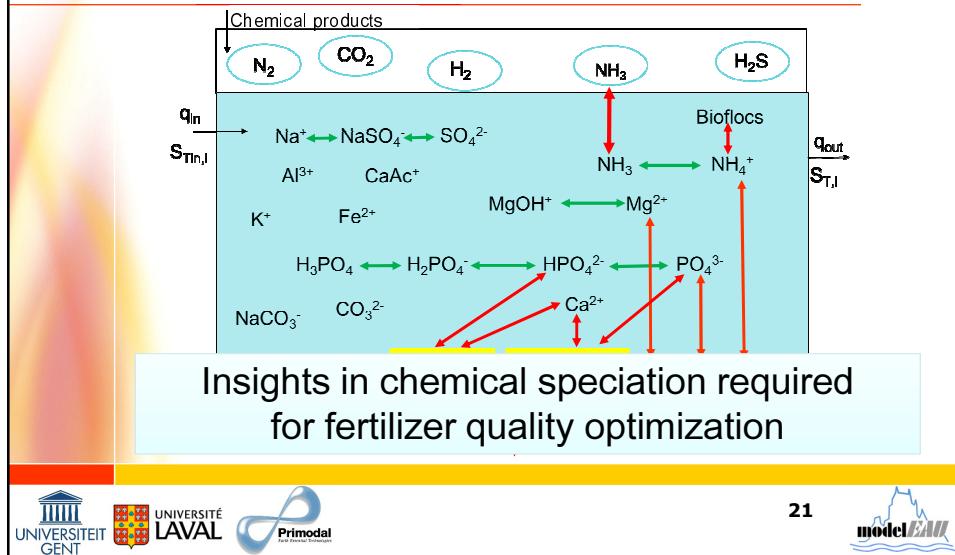
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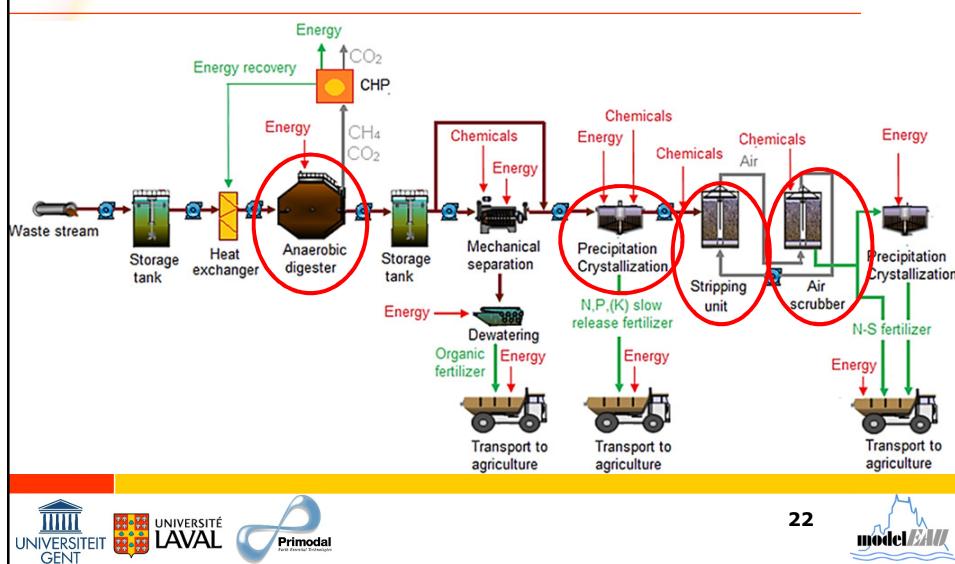
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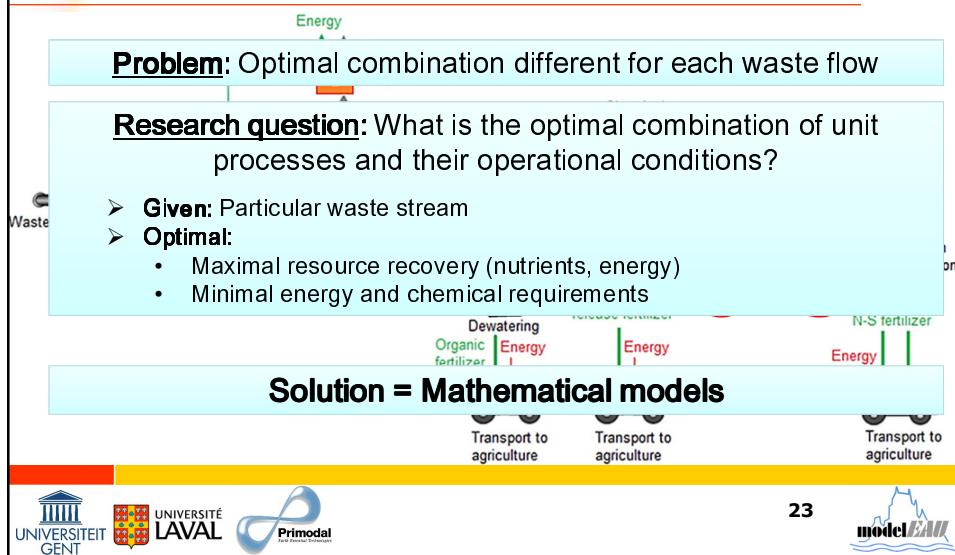
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Potential flow diagram of a WRRF



Potential flow diagram of a WRRF



Modelling challenges

Existing WWTP models

- Good description of biological principles for N & COD removal
- Lack of physicochemical reactions and speciation
- No models for nutrient recovery systems



Modelling challenges

Existing WWTP models → WRRF models

- Good description of biological principles for N & COD removal
- Lack of physicochemical reactions and speciation
- No models for nutrient recovery systems
- Physicochemical principles essential to describe nutrient recovery
 - pH ? Ion-pairing ? Precipitation ? Redox ? ...
- Some progress made for an. digestion (CEIT, UCT, etc.)
- No models for selective nutrient recovery based on detailed speciation and reaction dynamics

Modelling challenges

Existing WWTP models → WRRF models

⇒ Lack of models to adequately put together optimal treatment trains and their operating conditions

OBJECTIVES

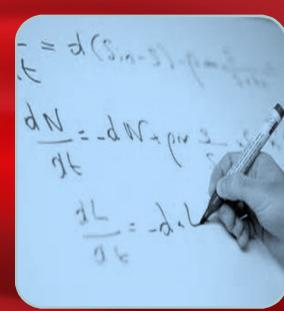


Specific research objectives

1. To develop generic models for the best available resource recovery systems including:
 - detailed chemical speciation
 - biological and physicochemical reaction kinetics
 - interactions between three phases (liquid-solid-gas)

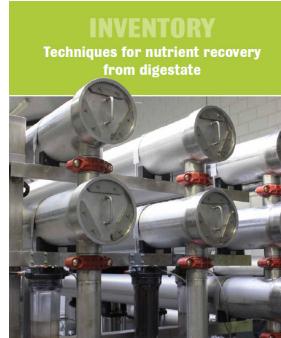
2. To apply the models as a tool for optimization of single processes and treatment trains in order to:
 - maximize resource recovery (nutrients, energy)
 - minimize energy and chemical requirements

MODEL DEVELOPMENT METHODOLOGY



I. Selection of unit processes and input streams

- Selection of unit processes (Vaneeckhaute et al., submitted):



This report has been elaborated in the framework of the modelEU Interreg V-A project, with financial support of the Interreg V-A programme, the province of West Flanders and the French Community.

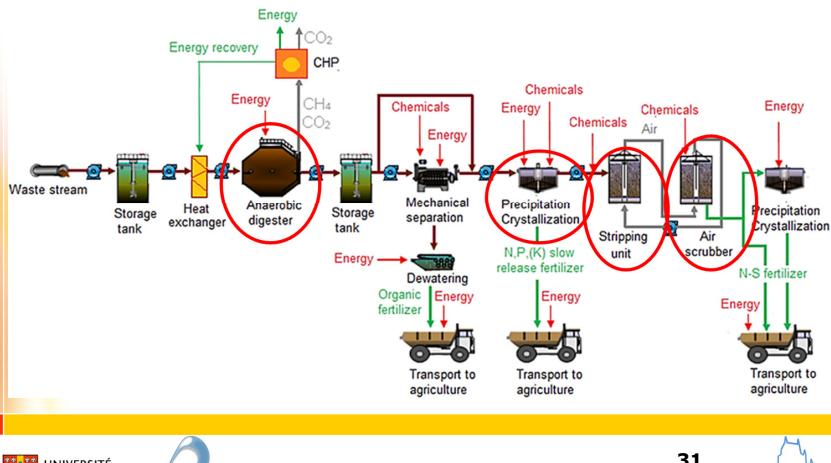


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I. Selection of unit processes and input streams

- Selection of unit processes (Vaneeckhaute et al., submitted):



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I. Selection of unit processes and input streams

- Selection of unit processes (Vaneeckhaute et al., submitted):

Type	Unit	Model name
Key units	1. Anaerobic digester	NRM-AD
	2. Precipitation/crystallization unit	NRM-Prec
	3. Stripping unit	NRM-Strip
	4. Air scrubber	NRM-Scrub

- Full-scale application
- Economic feasibility

Ancillary units	1. Separation unit 2. Storage tank 3. Chemical dosing unit 4. Heat exchanger	NRM-Settle NRM-Store NRM-Chem NRM-Heat
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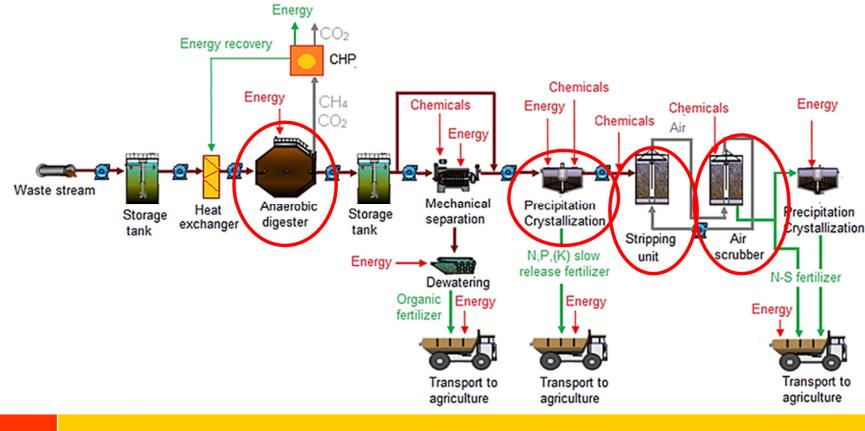


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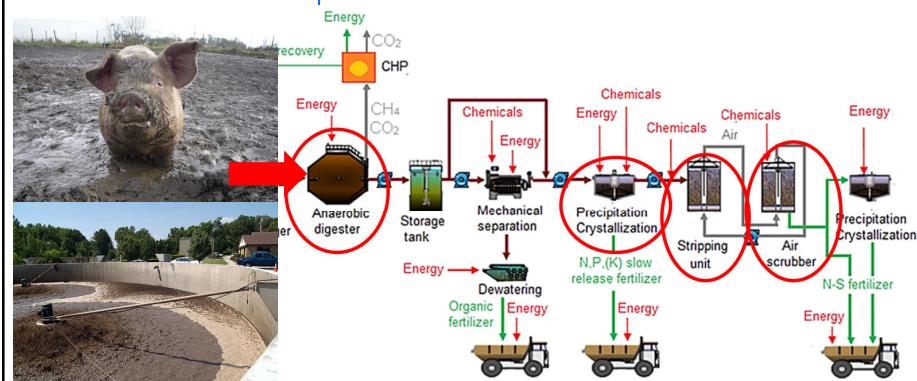
I. Selection of unit processes and input streams

- Selection of input waste streams:



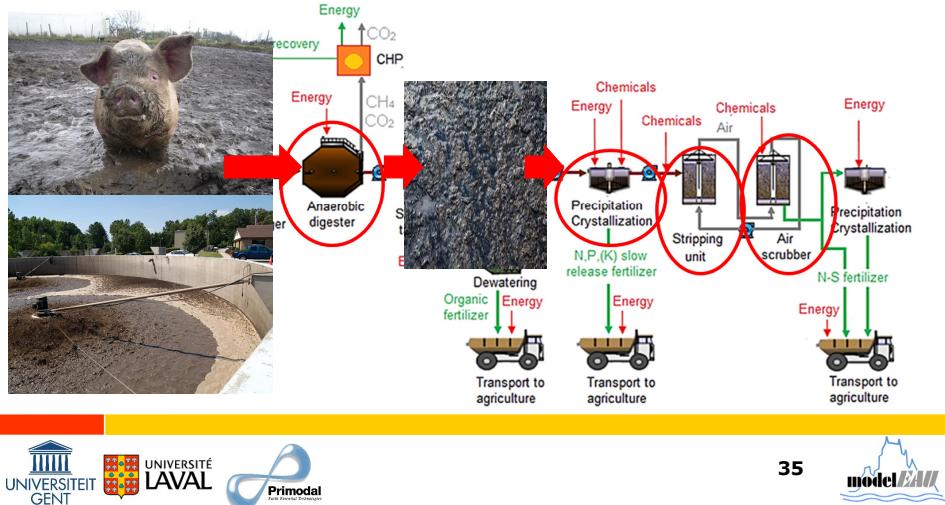
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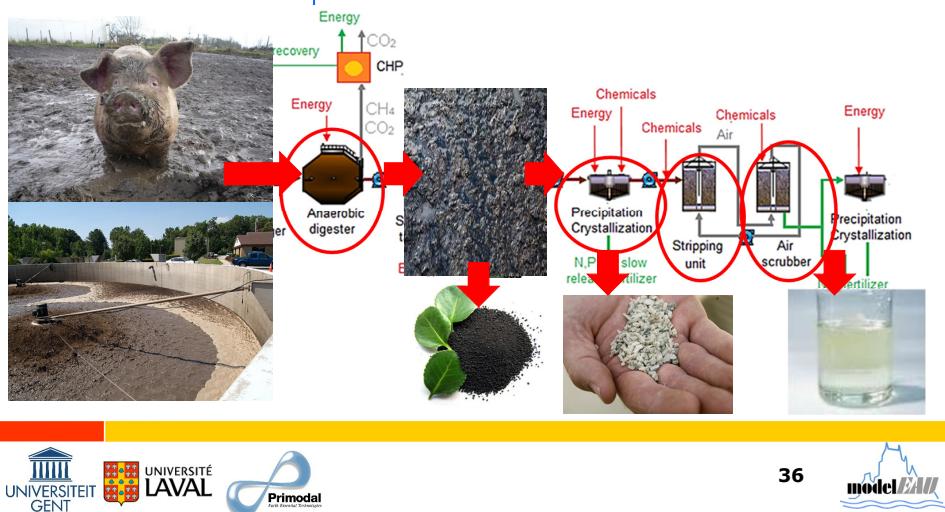
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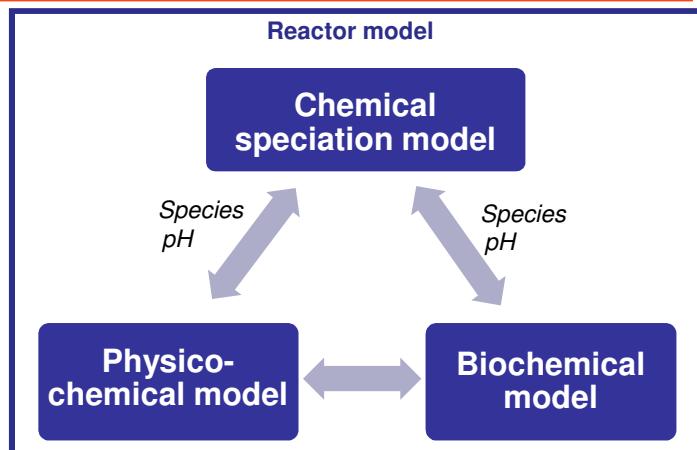
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II. Model theory and implementation

Combined three-phase physicochemical-biological models

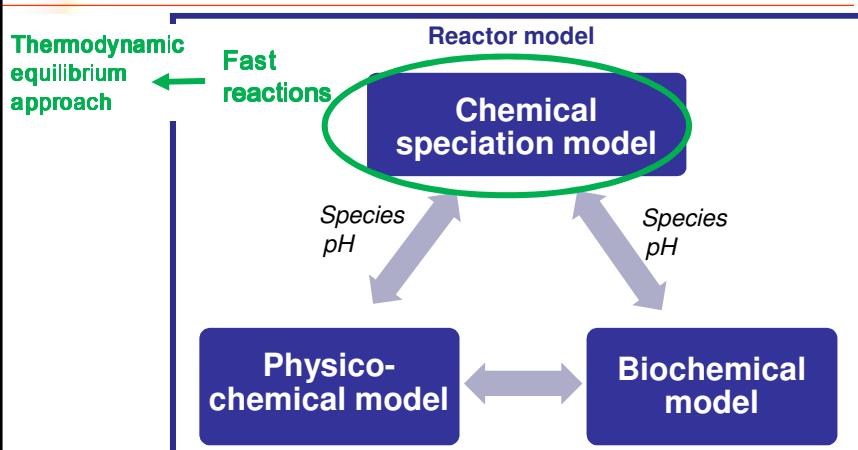


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II. Model theory and implementation

Combined three-phase physicochemical-biological models

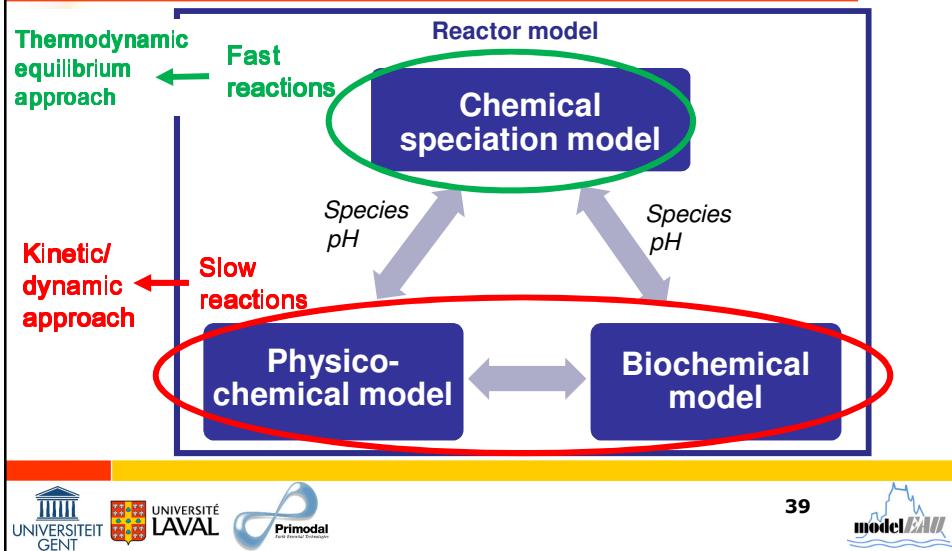


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II. Model theory and implementation

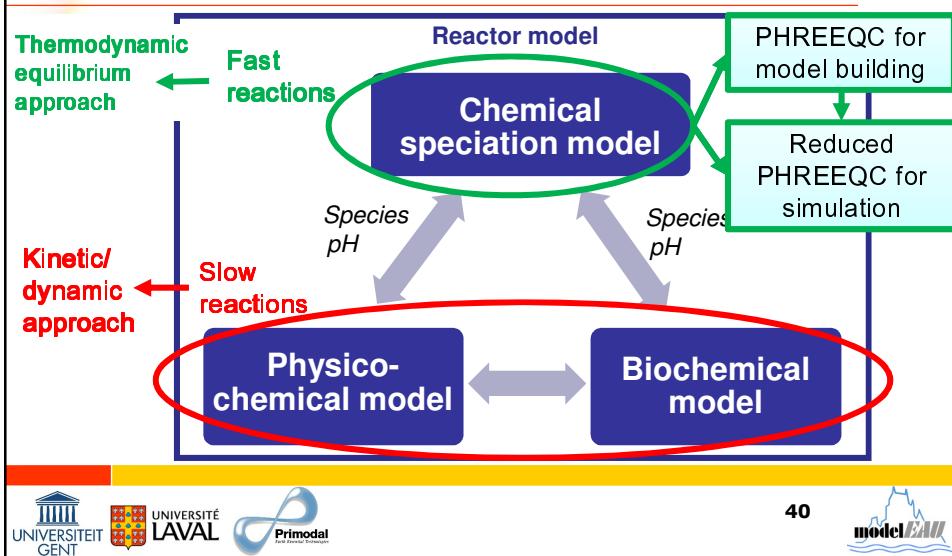
Combined three-phase physicochemical-biological models



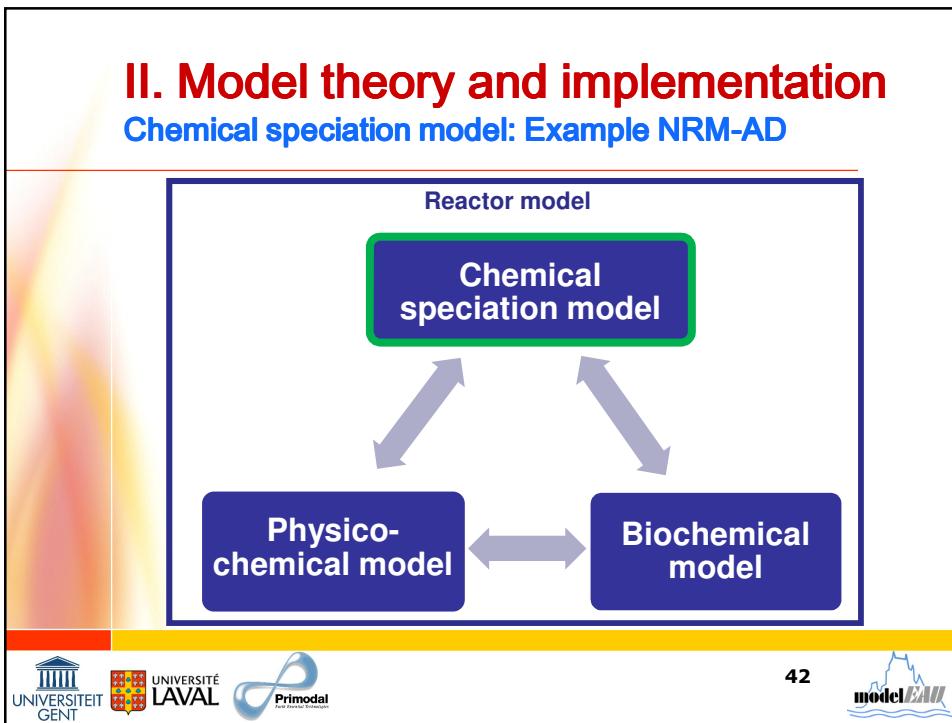
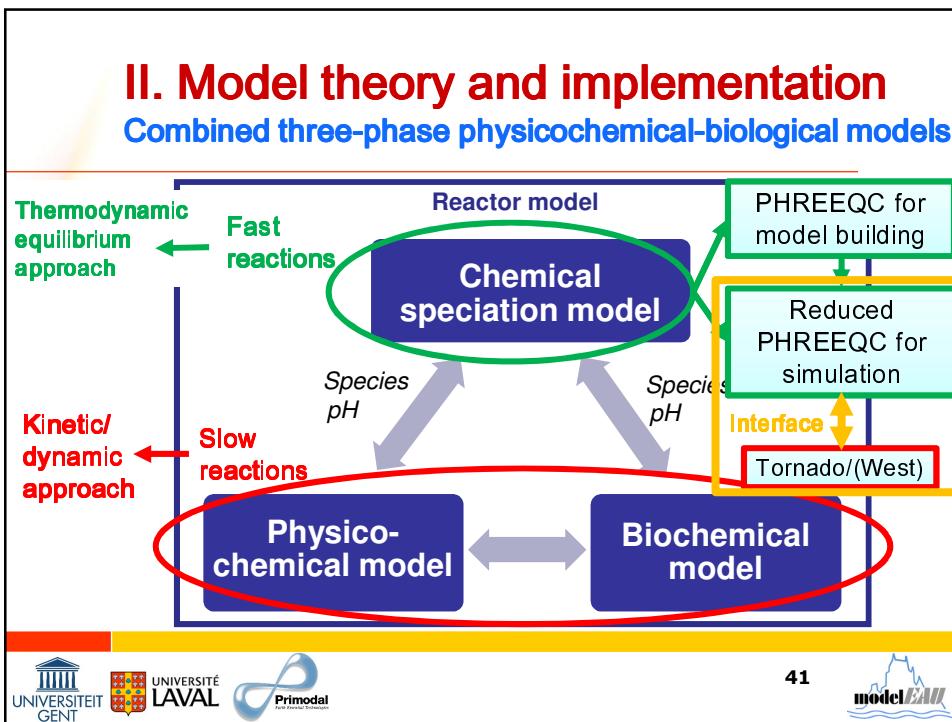
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II. Model theory and implementation

Combined three-phase physicochemical-biological models



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II. Model theory and implementation

Chemical speciation model: Example NRM-AD

I. Physicochemical component selection (21)

Ac, Al, Bu, Ca, C(IV), Cl, C(-IV), Fe, H(0), H(I), K, Mg, Na, N(-III), N(0), N(IV), O(0), P, Pro, S(-II), S(+VI), Va

II. Speciation calculation

PHREEQC/
MINTEQC

Corrections:
- Ion activity
- Temperature

III+IV. Selection species/reactions ⇒ reduced model

77 species
± > 3000 species

12 acid-base reactions
43 ion-pairing reactions
23 precipitation reactions
7 gas-liquid reactions



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II. Model theory and implementation

Physicochemical transformation model

Reactor model

Chemical speciation model

Physico-
chemical model

Biochemical
model



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II. Model theory and implementation

Physicochemical transformation model

- Liquid-gas transfer:

$$\rho_{CO_2,T} = k_L a_{CO_2} * (S_{CO_2,liq} - H_{CO_2} * p_{CO_2,gas})$$

Calculated by PHREEQC at every time step

- Liquid-solid transfer:

$$\rho_{struv} = k_{r,struv} * [S_{r,struv} - 1]^2$$

Relative supersaturation

Calculated by PHREEQC at every time step



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II. Model theory and implementation

Biochemical transformation model (NRM-AD)

Reactor model

Chemical speciation model

Physico-chemical model

Biochemical model



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II. Model theory and implementation

Biochemical transformation model (NRM-AD)

BIOCHEMICAL

- ADM 1 (19): Disintegration, hydrolysis, acidogenesis, acetogenesis, methanogenesis
Batstone et al. (2002)

COD-conversions
N-release/uptake



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II. Model theory and implementation

Biochemical transformation model (NRM-AD)

BIOCHEMICAL

- ADM 1 (19): Disintegration, hydrolysis, acidogenesis, acetogenesis, methanogenesis
Batstone et al. (2002)

PHYSICOQUIMICAL

- Acid-base systems (9)
- Ion-pairing (43)
- Precipitation-dissolution (23)
- Gas-liquid exchange (7)



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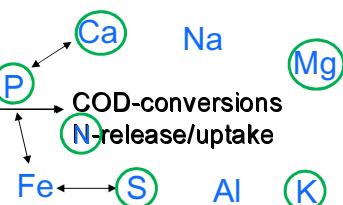


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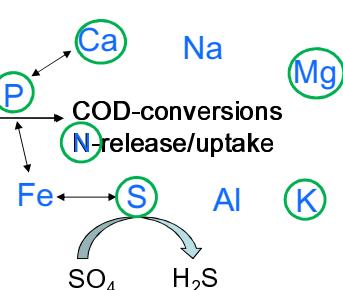


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Biochemical transformation model (NRM-AD)

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Batstone et al. (2002)
- Extension 1 (8): Sulfurgenesis
Knobel & Lewis (2002)

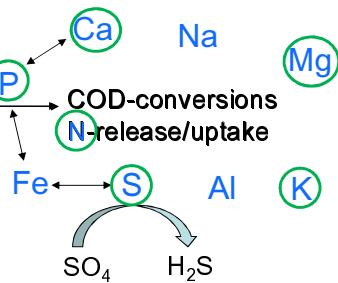


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Biochemical transformation model (NRM-AD)

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Batstone et al. (2002)
- Extension 1 (8): Sulfurgenesis
Knobel & Lewis (2002)
- Extension 2: Inclusion new components in stoich. matrix ADM 1 (P, K, S)
- Optional extension 3 (4): EBPR sludge (*Ikumi, 2011*)



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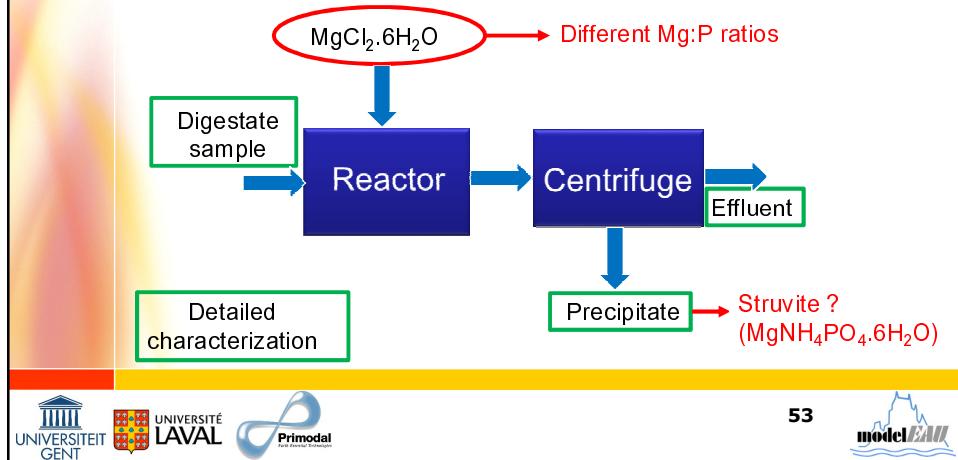


MODEL SIMULATION & VALIDATION RESULTS



NRM-Prec: Validation

- Lab-scale experiments P-precipitation



NRM-Prec: Validation

- Simulation vs. experimental results (after 12h)

Mg:P	Exp.	Digestate 1 % P-recovery		Digestate 2 % P-recovery	
		Model without NaH_2PO_4	Model with NaH_2PO_4	Exp.	Model with NaH_2PO_4
1:1	41	95.60	41.32	28	27.76
2:1	44	97.91	43.62	29	29.29

⇒ Importance of a detailed solution speciation and input characterization !



NRM-Prec: Scenario analyses and process optimization

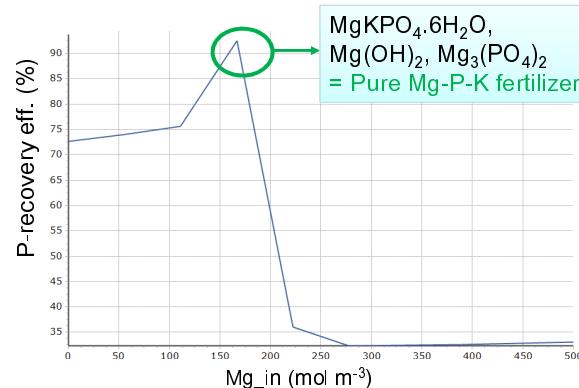
- Main precipitates: struvite, Ca- and Mg-phosphates, Ca- and Mg-carbonates, AlPO_4 , Fe- and Al-carbonates
⇒ No pure struvite fertilizer !
- Practical recommendations (if struvite is target):
 - Removal of CaCO_3 prior to precipitation (Huchzermeier et al., 2012)
 - Perform struvite precipitation upstream of Fe/Al dosing



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NRM-Prec: Scenario analyses and process optimization



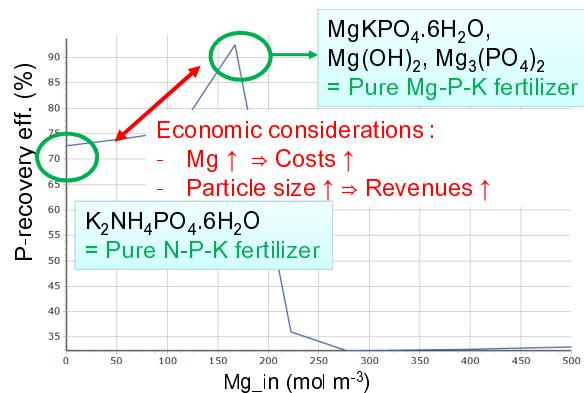
$\text{MgKPO}_4 \cdot 6\text{H}_2\text{O}$,
 $\text{Mg}(\text{OH})_2$, $\text{Mg}_3(\text{PO}_4)_2$
= Pure Mg-P-K fertilizer



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NRM-Prec: Scenario analyses and process optimization



RECOMMENDATIONS FOR RESEARCH

Recommendations for further experimental research

- Detailed characterisation input composition
→ Essential for process optimization
- Determination kinetic rates in real matrix ↔ pure solutions
→ Better prediction of fertilizer quality in time
- Development generic chemical analysis procedure for precipitate identification



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CONCLUSIONS & PERSPECTIVES



Conclusions

- Generic nutrient recovery model (NRM) library created
- PHREEQC-Tornado/West interface developed & verified
- Default parameters + proper input characterization
⇒ Good agreement with steady state experimental results (to be confirmed)
- Opportunities for better process understanding and optimization



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Perspectives

- Further dynamic data collection (on-going):
 - NRM-AD: Full-scale Greenwatt (BE)
 - NRM-Prec: Pilot-scale St. Hyacinthe (QC)
 - NRM-Strip/NRM-Scrub: Pilot-scale Waterleau (BE)



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Perspectives

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 - NRM-AD: Full-scale Greenwatt (BE)
 - NRM-Prec: Pilot-scale St. Hyacinthe (QC)
 - NRM-Strip/NRM-Scrub: Pilot-scale Waterloo (BE)
- Selection most important parameters for process optimization: Global sensitivity analysis (on-going)
- Process optimization (on-going):
 - Single units + treatment train



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Acknowledgements



Fonds de recherche
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technologies



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**THANK YOU FOR
THE ATTENTION**

QUESTIONS ?

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