

Chemical Verification of Variationally Optimized Second Order Density Matrices

Helen van Aggelen^a, Brecht Verstichel^b, Patrick Bultinck^a, Dimitri Van Neck^b, Paul Ayers^c, David Cooper^d

^aDepartment of Inorganic and Physical Chemistry, Ghent University, 281 S3, Krijgslaan, Ghent 9000, Belgium

^bCenter for Molecular Modeling, Ghent University, 86, Proeftuinstraat, Ghent 9000, Belgium

^cDepartment of Chemistry, McMaster University, Hamilton L8S 4M1, Canada

^dDepartment of Chemistry, University of Liverpool, Liverpool L69 76D, UK

Size-consistency is a highly desirable property for any quantum chemical method. Variational second order density matrix theory aims to replace the role of the wavefunction in quantum chemical calculations by a variationally optimized second order density matrix [1,2]. Practical implementations of this method are not size-consistent, because they apply a necessary but not sufficient subset of N-representability constraints.

We identify several problems of this method in describing dissociating molecular systems under the approximate 2-positivity conditions [1,3] for N-representability. First of all, it tends to dissociate molecules incorrectly into fractionally charged dissociation products with too low energies, thereby violating size-consistency [4]. Secondly, the second order density matrix of a non-interacting system is not separable into density matrices for each of the non-interacting units. The first shortcoming, dissociation into fractionally charged products, is a consequence of the method's faulty relationship between energy and the number of electrons. We propose simple 2-index constraints to solve this problem [5]. Nonetheless, these constraints do not solve the second problem, non-separability of the second-order density matrix for a system of non-interacting units.

References:

1. J. Coleman, *Rev. Mod. Phys.*, **35**, 668 (1963)
2. H. Kummer, *J. Math. Phys.*, **8**, 2063 (1967)
3. C. Garrod, J.K. Percus, *J. Math. Phys.*, **5**, 1756 (1964)
4. H. van Aggelen, B. Verstichel, P. Bultinck, D. Van Neck, P.W. Ayers, *J. Chem. Phys.*, **132**, 114112 (2010)
5. B. Verstichel, H. van Aggelen, D. Van Neck, P.W. Ayers, P. Bultinck, *J. Chem. Phys.*, **132**, 114112 (2010)