

LC-MS CHARACTERIZATION OF INTERMEDIATES AND PRODUCTS OF ACID ORANGE DYES AFTER LACCASE TREATMENT

¹Barbara Gornacka, ²Andrea Zille, ²Artur Cavaco-Paulo, ¹Astrid Rehorek*

¹Institute of Chemical Engineering and Plant Design, Betzdorfer Str. 2, D-50679 Cologne, Germany

²University of Minho, Department of Textile Engineering, 4800-058 Guimarães, Portugal

Introduction About 300.000 t/year azo dyes are used for textile dyeing operations, during which over 15% dyes are lost in the wastewater stream. The azo dyes are normally non-biodegradable under aerobic conditions. Under anaerobic conditions they are reduced to potentially toxic and mutagenic aromatic amines. The enzyme **Laccase** is a multicopper oxidase, which could decolorize azo dyes without cleavage of azo-bond and formation of aromatic amines. Degradation products of two **Acid Oranges dyes**, 3-(4-dimethylamino-1-phenylazo) benzene sulfonic acid and 3-(2-hydroxy-1-naphthylazo) benzene sulfonic acid by laccase *Trametes villosa* have been examined. Analyses of the products were performed using **ion pair chromatography coupled with a tandem mass spectrometer**.

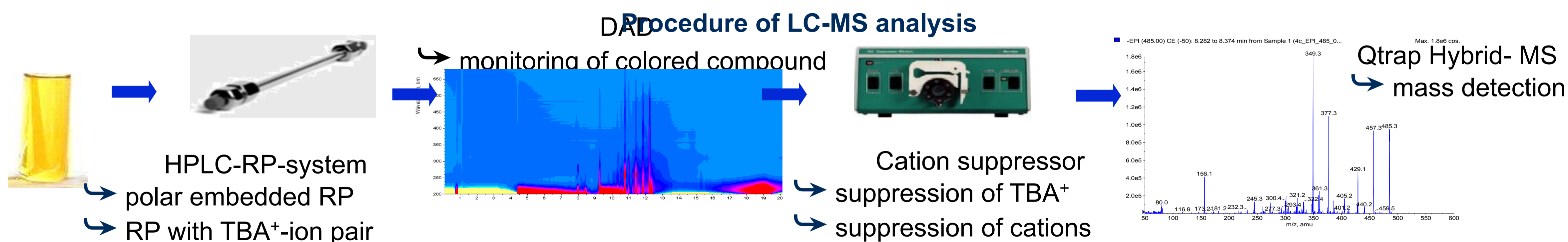
Experimental

Conditions of Laccase degradation of dyes

The investigated dyes 3-(4-dimethylamino-1-phenylazo) benzene sulfonic acid (**dye I**) and 3-(2-hydroxy-1-naphthylazo) benzene sulfonic acid (**dye II**) have been prepared in a concentration of 0,001 M (in citrate buffer 0,1M at pH 5). Dye solutions were incubated at room temperature with laccase solution from *Trametes villosa* (5,3 mg protein/L, 600 U/mL).

Experimental details of LC-MS analyses:

Analyses were performed on an Agilent 1100 HPLC system (degasser, binary pump, column compartment, Diode Array Detector) coupled with a Qtrap ESI-MS/MS from Applied Biosystems (Canada). For chromatographic separation the following analytical columns were used: Synergi Hydro (Phenomenex, USA); ProntoSIL AQ (Bischoff; Germany); Nucleosil 100 - 3 C18 HD (Macherey-Nagel, Germany) with different gradient methods. A 753 suppressor module (Methrom, Swiss) installed between DAD and MS was used for cation suppression. The MS analyses were performed under negative ionisation in the Enhanced Mass Scan-, Enhanced Product Ion-, Precursor Scan Ion- and MS³ Scan - mode.



Results

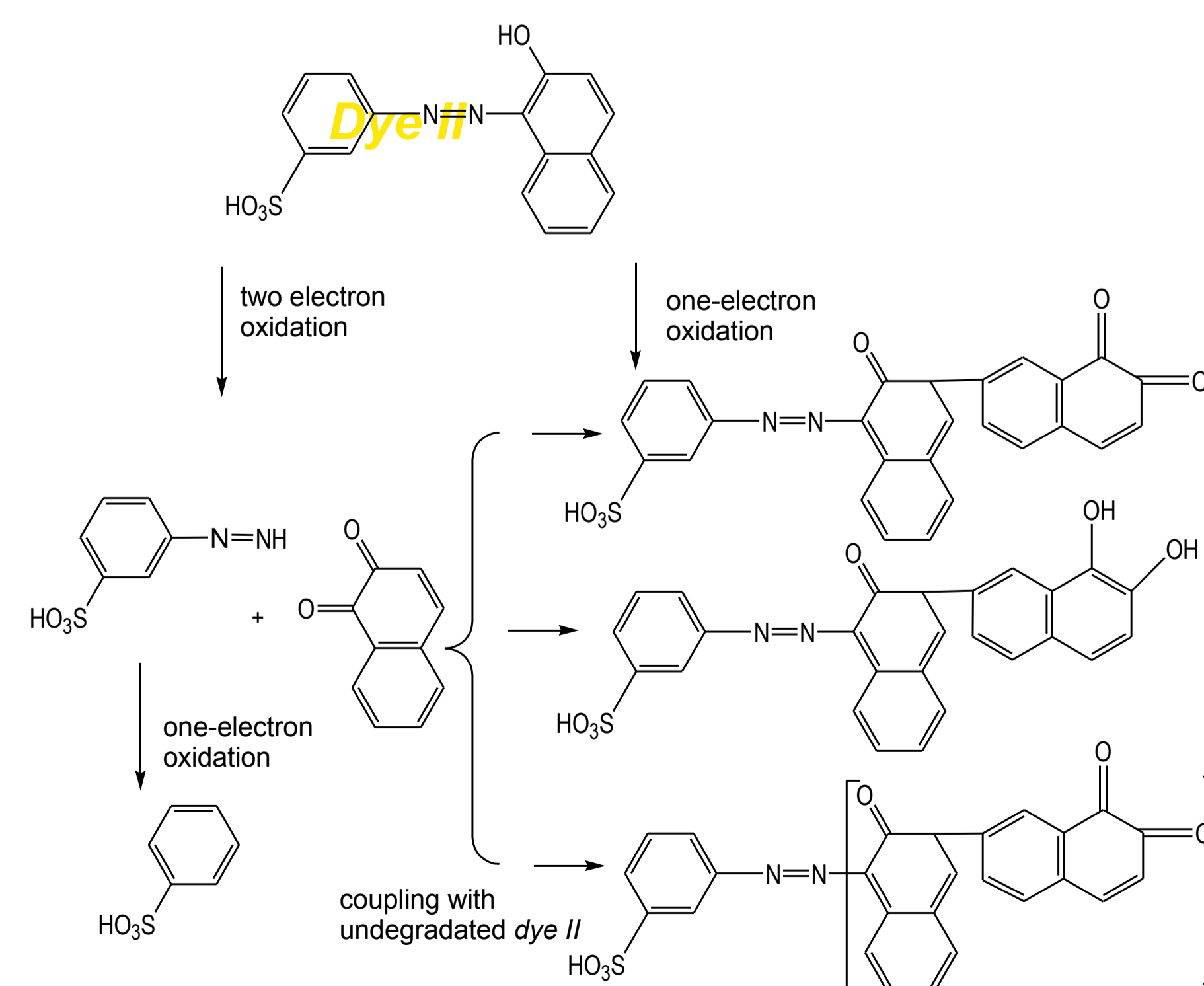
In the investigated samples **products of oxidation** by C-N bond cleavage have been observed and **oligomeric products of coupling of this oxidation products with undegraded molecules of dye I and dye II**.

In both samples oxidation products, like benzenesulfonic acid, 3-hydroxy-benzenesulfonic acid and 3-diazenyl-benzenesulfonic acid have been identified.

Degradation of **dye I** leads to the formation of dimeric products of coupling of **dye I** and one or two benzenesulfonic acid molecules.

Laccase treatment of **dye II** results in the formation of coupling products of **dye II** and one or two 1,2-naphthoquinone molecules and otherwise in the formation of coupling products of hydroxy-benzenesulfonic acid and one 1,2-naphthoquinone molecule. Additionally coupling of nitrogen molecules with degradation products has been observed.

Major products of laccase treatment of dye II



Conclusions

Several products of laccase treatment including several consecutive reaction products have been identified by an optimized LC-DAD-IC-MS technique. Attempts should be made to find conditions for complete degradation of the investigated azo dyes by laccase treatment without oligomer formation.