

## UvA-DARE (Digital Academic Repository)

## Analysis of Vibrational Circular Dichroism Spectra of Peptides: A Generalized Coupled Oscillator Approach of a Small Peptide Model Using VCDtools

Koenis, M.A.J.; Visscher, L.; Buma, W.J.; Nicu, V.P.

DOI 10.1021/acs.jpcb.9b11261 Publication date

2020 Document Version Final published version

Published in Journal of Physical Chemistry B License CC BY-NC-ND

Link to publication

### Citation for published version (APA):

Koenis, M. A. J., Visscher, L., Bumá, W. J., & Nicu, V. P. (2020). Analysis of Vibrational Circular Dichroism Spectra of Peptides: A Generalized Coupled Oscillator Approach of a Small Peptide Model Using VCDtools. *Journal of Physical Chemistry B*, *124*(9), 1665-1677. https://doi.org/10.1021/acs.jpcb.9b11261

#### General rights

It is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), other than for strictly personal, individual use, unless the work is under an open content license (like Creative Commons).

#### **Disclaimer/Complaints regulations**

If you believe that digital publication of certain material infringes any of your rights or (privacy) interests, please let the Library know, stating your reasons. In case of a legitimate complaint, the Library will make the material inaccessible and/or remove it from the website. Please Ask the Library: https://uba.uva.nl/en/contact, or a letter to: Library of the University of Amsterdam, Secretariat, Singel 425, 1012 WP Amsterdam, The Netherlands. You will be contacted as soon as possible.

UvA-DARE is a service provided by the library of the University of Amsterdam (https://dare.uva.nl)



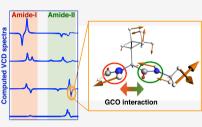
Article

## Analysis of Vibrational Circular Dichroism Spectra of Peptides: A Generalized Coupled Oscillator Approach of a Small Peptide Model Using VCDtools

Mark A. J. Koenis, Lucas Visscher, Wybren J. Buma, and Valentin P. Nicu\*

Cite This: J. Phy	rs. Chem. B 2020, 124, 1665–167	7	Read Online	
ACCESS	III Metrics & More		E Article Recommendations	s Supporting Information

**ABSTRACT:** Vibrational circular dichroism (VCD) is one of the major spectroscopic tools to study peptides. Nevertheless, a full understanding of what determines the signs and intensities of VCD bands of these compounds in the amide I and amide II spectral regions is still far from complete. In the present work, we study the origin of these VCD signals using the general coupled oscillator (GCO) analysis, a novel approach that has recently been developed. We apply this approach to the ForValNHMe model peptide in both  $\alpha$ -helix and  $\beta$ -sheet configurations. We show that the intense VCD signals observed in the amide I and amide II spectral regions essentially have the same underlying mechanism, namely, the through-space coupling of electric dipoles. The crucial role played



by intramolecular hydrogen bonds in determining VCD intensities is also illustrated. Moreover, we find that the contributions to the rotational strengths, considered to be insignificant in standard VCD models, may have sizable magnitudes and can thus not always be neglected. In addition, the VCD robustness of the amide I and II modes has been investigated by monitoring the variation of the rotational strength and its contributing terms during linear transit scans and by performing calculations with different computational parameters. From these studies—and in particular, the decomposition of the rotational strength made possible by the GCO analysis—it becomes clear that one should be cautious when employing measures of robustness as proposed previously.

#### INTRODUCTION

The study of peptides using vibrational circular dichroism (VCD) spectroscopy has been the focus of many studies since the introduction of this powerful technique.<sup>1-3</sup> These studies have established that the amide I and amide II bands provide important insight into the dominant and/or fractional average content of the secondary structures of peptides and proteins. Traditionally, the analysis of these bands was done using the coupled oscillator (CO) model.<sup>4</sup> However, after the development of density functional theory (DFT) programs that can accurately and routinely compute VCD spectra, more elaborate computational methods have been developed that now allow one to compute VCD spectra for very large systems.<sup>5–8</sup>

Subsequent computational studies on peptides and proteins<sup>6,9–15</sup> have led to two important conclusions. First, it was concluded that the normal modes associated with the amide I and amide II bands are delocalized modes involving motion of several interacting groups. The general consensus is, therefore, that the observed amide I and amide II VCD signals are caused by distance effects associated with the coupling of groups on different strands. In the case of the amide I VCD bands, a number of computational studies<sup>6,12,13,16</sup> have demonstrated that the very intense VCD bands<sup>17,18</sup> originate from a more general degenerate CO mechanism. The amide II VCD bands, on the other hand, are typically not interpreted in terms of a CO model even though they appear to be governed by a similar mechanism. Second, it has been found that it is

more difficult to identify  $\beta$ -sheets than  $\alpha$ -helices. This is thought to be a consequence of the fact that in  $\beta$ -sheets, the amide groups that are coupled have almost the same planar arrangement and thus often may not form a chiral arrangement.<sup>14,15</sup>

To investigate this last aspect, Magyarfalvi et al. recently performed a VCD robustness analysis of the amide I and amide II modes for a series of For(AA), NHMe (AA being Val, Asn, Asp, or Cys, n = 1-5 for Val and Asn, and n = 1 for Asp and Cys) model peptides with  $\alpha$ -helix and  $\beta$ -sheet backbone conformations.<sup>19</sup> Apart from a few exceptions, it was concluded that for short peptide models, the computed VCD sign of the amide I and amide II modes is most often nonrobust.<sup>20</sup> On the other hand, in longer peptide models, the amide I and amide II modes were found to be robust if they involved couplings between similar amide vibrations localized on the various backbone amide groups. These findings seem to corroborate the experimental observations but do not provide any physical insight into why it is difficult to accurately compute the VCD sign of nonrobust modes. As discussed previously,<sup>21</sup> this is because the performed VCD robustness

```
Received:December 4, 2019Revised:February 7, 2020Published:February 9, 2020
```



analysis—which is based on the values computed for the dissymmetry factor<sup>20</sup>—is unable to judge properly the stability of the VCD sign computed for modes in which the dominant contribution to the total VCD intensity is determined by through-space coupling of electric dipoles.

To tackle this issue, we employed, in this work, the general CO (GCO) formalism, a novel approach for analyzing the computed VCD intensities that we have recently introduced.<sup>21,22</sup> The GCO expression is an exact formulation of the VCD CO model which is applicable for all types of normal modes. In addition, the GCO analysis can be used to judge the stability of the computed VCD signs. This makes the GCO analysis ideally suitable for studying the mechanisms that determine the magnitude and sign of the VCD signals observed in the amide I and amide II spectral regions. To make the GCO analysis even more efficient and useful, we have embedded it into the graphical user interface (GUI) of the Amsterdam density functional (ADF) software package.<sup>23,25</sup> This has allowed us to implement a novel normal mode analysis, which makes it possible to monitor on-the-fly the nuclear displacement vectors and the atomic contributions of the vibrational electric DTMs.

In this work, we will use these newly developed tools to analyze ForValNHMe model peptides that have previously been studied by Magyarfalvi et al.<sup>19</sup> The relatively small size of the system allows us to perform a large number of careful calculations on the dependence of the amide I and amide II VCD signals on structural and electronic parameters, as well as detailed normal mode analyses. Clearly, such calculations do not aim to interpret the sign and intensity patterns of VCD spectra of large peptides and proteins. What they do allow us to show is that the newly developed tools enable us (i) to pinpoint the origin of the VCD signal and, at the same time, judge the stability of its sign, (ii) to explain the differences observed between the VCD spectra computed for  $\alpha$ -helix and  $\beta$ -sheet conformers, and (iii) to understand why some of the normal modes have a nonrobust VCD sign. Importantly, they also allow us to get a detailed insight into the VCD GCO mechanism. As a result, it will become clear that GCO analyses such as the ones performed in the present work on peptides are much more generally applicable and are in fact an indispensable tool for unraveling the origin and sensitivity of VCD signals in any (supra)molecular system.

#### THEORY

A detailed description of the GCO formalism can be found in ref 22. Here, it will be briefly summarized. The IR and VCD intensities of the fundamental vibrational transition of a normal mode j are given by the dipole  $(D_{01})$  and rotational  $(R_{01})$  strengths<sup>24</sup>

$$D_{01}(j) = \vec{E}_{01}^{\text{tot}}(j) \cdot \vec{E}_{10}^{\text{tot}}(j)$$
(1)

$$R_{01}(j) = -i \cdot \vec{E}_{01}^{\text{tot}}(j) \cdot \vec{M}_{10}^{\text{tot}}(j)$$
(2)

where  $\vec{E}_{01}^{\text{tot}}(j)$  and  $\vec{M}_{10}^{\text{tot}}(j)$  are the total electric and magnetic dipole transition moments (DTMs), respectively, associated with the fundamental vibrational transition (01) of mode j while i is the unit imaginary number.

**Fragment Decomposition of the Rotational Strength.** In the harmonic approximation,  $\vec{E}_{01}^{\text{tot}}(j)$  and  $\vec{M}_{10}^{\text{tot}}(j)$  can be written as a sum of atomic components.<sup>24</sup> Consequently, by grouping the atoms of a molecule in three molecular fragments (labeled as A, B, and R) one can write

$$\vec{E}_{01}^{\text{tot}}(j) = \sum_{\lambda=1}^{N} \vec{E}_{01}^{\lambda}(j) = \sum_{\lambda=1}^{N_{A}} \vec{E}_{01}^{\lambda}(j) + \sum_{\lambda=1}^{N_{B}} \vec{E}_{01}^{\lambda}(j) + \sum_{\lambda=1}^{N_{R}} \vec{E}_{01}^{\lambda}(j) = \vec{E}_{01}^{A}(j) + \vec{E}_{01}^{B}(j) + \vec{E}_{01}^{R}(j)$$
(3)  
$$\vec{M}_{01}^{\text{tot}}(j) = \sum_{\lambda=1}^{N} \vec{M}_{01}^{\lambda}(j) = \sum_{\lambda=1}^{N_{A}} \vec{M}_{01}^{\lambda}(j) + \sum_{\lambda=1}^{N_{B}} \vec{M}_{01}^{\lambda}(j) + \sum_{\lambda=1}^{N_{R}} \vec{M}_{01}^{\lambda}(j)$$

$$= \vec{M}_{01}^{\mathbf{A}}(j) + \vec{M}_{01}^{\mathbf{B}}(j) + \vec{M}_{01}^{\mathbf{R}}(j)$$
(4)

In eqs 3 and 4,  $\vec{E}_{01}^{X}$  and  $\vec{M}_{01}^{X}$  are the EDTM and MDTM associated with a molecular fragment (X is A, B, or R),  $\lambda$  runs over nuclei,  $N_A$ ,  $N_B$ , and  $N_R$  are the number of atoms in the molecular fragments A, B, and R, respectively, while N is the total number of atoms in the molecule ( $N = N_A + N_B + N_R$ ). By inserting eqs 3 and 4 in eq 2, we obtain<sup>22</sup>

$$R_{01}(j) = -i[\vec{E}_{01}^{A}(j) + \vec{E}_{01}^{B}(j) + \vec{E}_{01}^{R}(j)] \cdot [\vec{M}_{10}^{A}(j) + \vec{M}_{10}^{B}(j) + \vec{M}_{10}^{R}(j)]$$
$$= R_{01}^{GCO}(j) + R_{01}^{IF}(j) + R_{01}^{R}(j)$$
(5)

 $R_{01}^{GCO}$  is the generalized CO term. This term is associated with the through-space interaction of fragments **A** and **B** and is given by

$$R_{01}^{\mathbf{GCO}}(j) = -i[\vec{E}_{01}^{\mathbf{A}}(j) \cdot \vec{M}_{10}^{\mathbf{B}}(j) + \vec{E}_{01}^{\mathbf{B}}(j) \cdot \vec{M}_{10}^{\mathbf{A}}(j)]$$
(6)

The  $R_{01}^{IF}$  term represents the contribution of the individual fragments A and B

$$R_{01}^{\mathbf{IF}}(j) = -i[\vec{E}_{01}^{\mathbf{A}}(j) \cdot \vec{M}_{10}^{\mathbf{A}}(j) + \vec{E}_{01}^{\mathbf{B}}(j) \cdot \vec{M}_{10}^{\mathbf{B}}(j)]$$
(7)

Finally, the  $R_{01}^{\mathbf{R}}(j)$  contribution contains the coupling between fragment **R** and fragments **A** and **B**, and the individual contributions from fragment **R** 

$$R_{01}^{\mathbf{R}}(j) = -i[\vec{E}_{01}^{\mathbf{A}}(j)\cdot\vec{M}_{10}^{\mathbf{R}}(j) + \vec{E}_{01}^{\mathbf{R}}(j)\cdot\vec{M}_{10}^{\mathbf{A}}(j)] + -i[\vec{E}_{01}^{\mathbf{B}}(j)\cdot\vec{M}_{10}^{\mathbf{R}}(j) + \vec{E}_{01}^{\mathbf{R}}(j)\cdot\vec{M}_{10}^{\mathbf{B}}(j)] + -i[\vec{E}_{01}^{\mathbf{R}}(j)\cdot\vec{M}_{10}^{\mathbf{R}}(j)]$$
(8)

The fragment decomposition of the rotational strengths defined by eqs 5–8 is exact within the harmonic approximation and can be performed for any type of normal mode in both symmetric and asymmetric chiral molecules. The **A** and **B** fragments are the important fragments and as such contain atoms that exhibit large EDTMs and MDTMs. Fragment **R** contains the remaining atoms, that is, atoms with small or zero EDTMs and MDTMs, and as a result, its contribution to the total rotational strength is expected to be small. Indeed, as shown, one can almost always define the **A**, **B**, and **R** fragments such that the  $R_{01}^{R}$  term is very small and negligible, irrespective of the type of the normal mode.<sup>22</sup> More importantly, however, we found that large VCD signals are always due to dominant coupling  $R_{01}^{GCO}$  terms.

**GCO VCD Analysis.** The fragment decomposition of the rotational strength defined in the previous section allows one to interpret VCD signals in terms of the interaction of two molecular fragments and to identify two sources that contribute dominantly to the rotational strength of a particular transition. The first contribution  $(R_{01}^{\text{IF}})$ —which is expected to be small when the two fragments are achiral—is associated with the two isolated fragments and the second  $(R_{01}^{\text{GCO}})$  with the through-space coupling of their EDTMs and MDTMs. This perspective offers more insight into the origin of the VCD signal than eq 2, although a simple and intuitive interpretation for the magnitude and sign of  $R_{01}^{\text{GCO}}$  remains practically impossible because the MDTMs are origin-dependent and imaginary. Recently, it has been shown<sup>22</sup> that more insight into the properties of  $R_{01}^{\text{GCO}}$  can be gained by casting eq 6 into the "coupled oscillator" form<sup>4</sup>

$$R_{01}^{\mathbf{GCO}}(j) = -i[\vec{E}_{01}^{\mathbf{A}}(j) \cdot \vec{M}_{10}^{\mathbf{B}}(j) + \vec{E}_{01}^{\mathbf{B}}(j) \cdot \vec{M}_{10}^{\mathbf{A}}(j)] = \frac{\pi \nu_{j}}{c} \vec{Y}^{\mathbf{GCO}}(j, \mathbf{A}, \mathbf{B}) \cdot [\vec{E}_{01}^{\mathbf{A}}(j) \times \vec{E}_{01}^{\mathbf{B}}(j)]$$
(9)

where  $\nu_j$  is the frequency of mode *j*, *c* the speed of light in vacuum, and  $\vec{Y}^{GCO}(j,\mathbf{A},\mathbf{B})$ , the generalized CO vector, which is defined per normal mode and depends on the definition of the GCO fragments **A** and **B**. Unlike eqs 2 and 6, which depend explicitly on the MDTM, the second equality in eq 9 allows us to interpret the VCD intensities in terms of couplings between the EDTMs associated with the GCO fragments **A** and **B**, which are basically charge movements separated by the  $\vec{Y}^{GCO}$  distance. As we have shown, <sup>21,22,25</sup> such an analysis provides detailed insight into the magnitude and sign of the calculated VCD intensities.

A key step in the VCD GCO analysis is the identification of the molecular fragments **A**, **B**, and **R**, providing the most simple and intuitive physical insight. The procedure used to identify the GCO fragments maximizes the  $R_{01}^{GCO}$  term while, at the same time, minimizing the  $R_{01}^{R}$  term.<sup>25</sup> Formally, the molecular fragments **A**, **B**, and **R** are defined per normal mode. In practice, it is found, however, that for most modes, the atoms involved in polar bonds provide the most important contributions to the VCD intensities. For this reason, one can often group the atoms of a molecule into VCD-active and VCD-inactive sites. This simplifies the interpretation of the VCD spectra even further because according to eq 9, intense VCD signals arise when the VCD-active sites of a molecule are in a favorable orientation.

It is important to recognize that the standard CO model is a model used to predict VCD intensities, while the VCD GCO analysis is a postanalysis method performed after running a DFT VCD calculation that is aimed to gain physical insight into the origin of the computed VCD intensities. To connect the  $R_{01}^{\rm GCO}$  term with the standard CO model, we define the CO correction vector  $\vec{Y}^{\rm COC}$  as

$$\vec{Y}^{\text{COC}}(j, \mathbf{A}, \mathbf{B}) = \vec{Y}^{\text{GCO}}(j, \mathbf{A}, \mathbf{B}) - \vec{Y}^{\text{AB}}$$
(10)

where  $\vec{Y}^{AB}$  is the distance vector determined by the mass centers of the **A** and **B** fragments. Combining eq 9 with eq 10 allows one to link the  $R_{01}^{GCO}$  term directly to the standard CO term  $R_{01}^{CO}$  via

$$R_{01}^{GCO}(j) = \frac{\pi \nu_j}{c} \cdot \vec{Y}^{GCO}(j, \mathbf{A}, \mathbf{B}) \cdot [\vec{E}_{01}^{A}(j) \times \vec{E}_{01}^{B}(j)]$$
  
=  $\frac{\pi \nu_j}{c} \cdot [\vec{Y}^{AB} + \vec{Y}^{COC}(j, \mathbf{A}, \mathbf{B})] \cdot [\vec{E}_{01}^{A}(j) \times \vec{E}_{01}^{B}(j)]$   
=  $R_{01}^{CO}(j) + R_{01}^{COC}(j)$  (11)

where

$$R_{01}^{CO}(j) = \frac{\pi \nu_j}{c} \vec{Y}^{AB} \cdot [\vec{E}_{01}^{A}(j) \times \vec{E}_{01}^{B}(j)]$$
(12)

$$R_{01}^{\text{COC}}(j) = \frac{\pi \nu_j}{c} \vec{Y}^{\text{COC}}(j, \mathbf{A}, \mathbf{B}) \cdot [\vec{E}_{01}^{\mathbf{A}}(j) \times \vec{E}_{01}^{\mathbf{B}}(j)]$$
(13)

The magnitude of the errors made when using the standard CO model can be assessed by comparing the values computed with DFT for  $R_{01}^{CO}$ ,  $R_{01}^{COC}$ ,  $R_{01}^{GCO}$ ,  $R_{01}^{IR}$ , and  $R_{01}^{R}$ . **Magnitude and Sign of the VCD Intensities.** As

**Magnitude and Sign of the VCD Intensities.** As mentioned above, typically all intense bands that stand out in a VCD spectrum are characterized by large  $R_{01}^{GCO}$  values which are significantly larger than their associated  $R_{01}^{IF}$  contributions. Small- and medium-sized VCD bands, on the other hand, are most often characterized by comparable  $R_{01}^{GCO}$  and  $R_{01}^{IF}$  values. On the basis of eq 9, one could thus conclude that large VCD intensities will typically be observed when the EDTMs associated with fragments **A** and **B** are in a favorable orientation.

This interpretation of eq 9 explains in very simple terms why VCD spectra depend so sensitively on the molecular conformation. Besides this, it also provides insight into the stability of the computed VCD sign, that is, the VCD robustness<sup>26</sup> of a mode. In a series of studies,<sup>19,27–29</sup> Magyarfalvi et al. proposed to use the magnitude of the ratio  $\zeta(j)$  between the rotational and dipole strengths of the mode *j*—a quantity that is very similar to Kuhn's dissymmetry factor—as a measure of its VCD robustness. Modes with  $\zeta(j) > 10$  ppm are considered to be robust and their computed VCD sign reliable, whereas modes with  $\zeta(j) < 10$  ppm should be considered as nonrobust and their computed sign unreliable.

Using eqs 1, 2, 5, and 9, the  $\zeta(j)$  ratio is expressed as

$$\zeta(j) = \frac{R_{01}(j)}{D_{01}(j)} = \frac{R_{01}^{GCO}(j) + R_{01}^{IF}(j) + R_{01}^{R}(j)}{D_{01}(j)}$$
$$= \frac{\frac{\pi v_j}{c} \vec{Y}^{GCO} \cdot [\vec{E}_{01}^{A}(j) \times \vec{E}_{01}^{B}(j)]}{D_{01}(j)} + \frac{R_{01}^{IF}(j)}{D_{01}(j)} + \frac{R_{01}^{R}(j)}{D_{01}(j)}$$
(14)

When  $R_{01}^{\text{GCO}}(j)$  is the dominant contribution to  $R_{01}(j)$ , eq 14 can be approximated as

$$\begin{aligned} \zeta(j) &\simeq \frac{R_{01}^{\mathbf{GCO}}(j)}{D_{01}(j)} \\ &= \frac{\frac{\pi \nu_j}{c} |\vec{Y}^{\mathbf{GCO}}| |\vec{E}_{01}^{\mathbf{A}}(j)| |\vec{E}_{01}^{\mathbf{B}}(j)| \sin[\theta(j)] \cos[\varphi(j)]}{D_{01}(j)} \end{aligned}$$
(15)

where  $\theta(j)$  is the angle between the  $\vec{E}_{01}^{A}(j)$  and  $\vec{E}_{01}^{B}(j)$  and  $\varphi(j)$ , the angle between  $\vec{Y}^{\text{GCO}}$  and  $\vec{E}_{01}^{A}(j) \times \vec{E}_{01}^{B}(j)$ . When  $\sin(\theta)$  or  $\cos(\varphi)$  is close to zero, small changes in the relative orientation of the fragments **A** and **B** can easily induce a sign change and/

or a large change in the magnitude of  $R_{01}^{\text{GCO}}(j)$ . Equation 15, therefore, suggests that when evaluating the robustness of a mode whose VCD intensity is dominantly determined by  $R_{01}^{\text{GCO}}$ , it is not sufficient to consider only the magnitude of the associated  $\zeta$  ratio. Instead, it is necessary to identify the GCO fragments that provide the main contribution to the rotational strength and to investigate whether at ambient temperatures the relative orientation of these two fragments can be changed easily. This is because the sign and magnitude of  $R_{01}^{\text{GCO}}$  can critically depend on small variations in the angles  $\theta$  and  $\varphi$ .

**Molecular Structure.** A schematic representation of the ForValNHMe peptide is shown in Figure 1. Eight

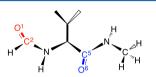


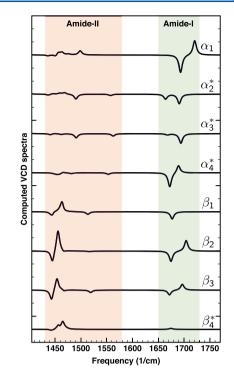
Figure 1. Schematic representation of the ForValNHMe peptide as well as definition of the fragments used in the GCO analysis of the amide I modes. The CO fragments A and B are highlighted in red and blue, respectively, while the atoms in the fragment R are shown in black.

ForValNHMe conformers have been considered of which four are in the  $\alpha$ -helix and four in the  $\beta$ -sheet configuration. Six of the conformers have been taken from ref 19. Following the labeling convention used in ref 19, these conformers are labeled as  $\alpha_1$ ,  $\alpha_2$ ,  $\alpha_3$  and  $\beta_1$ ,  $\beta_2$ ,  $\beta_3$ . For a given configuration (i.e.,  $\alpha$  or  $\beta$ ), these conformers differ only in the orientation of the isopropyl group with respect to the backbone. The  $\alpha$ conformers taken from ref 19 have been mirrored to assure that all considered conformations have the same absolute configuration.

The linear transit (LT) scans that will be discussed below revealed two additional conformational energy minima, one in the vicinity of the conformer  $\alpha_1$  and one in the vicinity of the conformer  $\beta_1$ , which have, therefore, also been considered and labeled as  $\alpha_4$  and  $\beta_4$  (see Figure S1 for the structures of all conformers). The conformers  $\alpha_2$ ,  $\alpha_3$ ,  $\alpha_4$ , and  $\beta_4$  exhibit an intramolecular hydrogen bond between the O<sub>1</sub> atom and the N–H bond on the C-terminal, which plays a crucial role in the present study. To emphasize this aspect, these conformers will, therefore, be referred to as  $\alpha_2^*$ ,  $\alpha_3^*$ ,  $\alpha_4^*$ , and  $\beta_4^*$ . The molecular structures of all conformers have been optimized at various levels of theory with structures determined at the BP86/TZP level of the theory reported in Table S1 of the Supporting Information.

#### RESULTS AND DISCUSSION

Figure 2 shows the amide I and amide II regions of the VCD spectra computed for the various ForValNHMe conformers. Interestingly, in the amide I region, some conformers exhibit the expected bisignate VCD pattern  $(\alpha_1, \alpha_4^*, \beta_2, \text{ and } \beta_3)$  while the other conformers do not show this feature  $(\alpha_2^*, \alpha_3^*, \beta_1, \text{ and } \beta_4^*)$ . Moving to the amide II region, we find that the spectra of all  $\beta$  conformers exhibit very intense VCD bands with magnitudes comparable to, or even larger than, those of the amide I bands, while the  $\alpha$  conformers display only very weak signals. To understand these differences, we will analyze in the following the amide I and amide II VCD bands using the GCO analysis introduced above. Apart from revealing the differences observed between the considered conformers, this analysis will

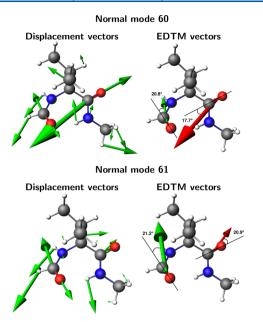


**Figure 2.** Amide I and amide II regions of the VCD spectra computed for the eight ForValNHMe conformers considered in this study.

also provide insight into the robustness of the amide I and amide II VCD bands. In the following section, we will discuss these results and refer to the rotational strengths and its various contributions in units of  $10^{-44}$  esu<sup>2</sup>·cm<sup>2</sup> while the EDTMs are referred to in units of  $10^{-20}$  esu·cm.

Amide I Modes. Normal Mode Analysis. The For-ValNHMe peptide has two amide moieties (see Figure 1). Normal mode analysis finds that in all conformers, the amide I modes (i.e., modes 60 and 61) dominantly consist of stretching motions of the two C=O bonds but also involve C-H and N–H bending motions. Except in the conformer  $\alpha_4^*$  where the two carbonyl bonds move with similar amplitudes, we find that the normal mode motion is not localized equally on the two carbonyl bonds. One can thus distinguish between a dominant C=O bond that moves with large amplitude and a secondary C=O bond that oscillates with significantly smaller amplitude. Furthermore, modes 60 and 61 can be loosely classifiedbecause there is no symmetry-as in- and out-of-phase with respect to the C=O stretch motions. As can be concluded from Table S9 in the Supporting Information, the out-of-phase mode has a lower frequency than the in-phase mode in conformers without an intramolecular hydrogen bond, whereas the reverse holds for the conformers with an intramolecular H bond. However, irrespective of the conformation, the  $C_5 = O_6$ stretch is always dominant in the mode with lower frequency and the  $C_2 = O_1$  stretch is dominant in the mode with higher frequency as exemplified in Figure 3 where the nuclear displacement vectors of modes 60 and 61 of the conformer  $\alpha_3^*$ are shown.

Generalized CO Analysis. A GCO analysis was performed by decomposing  $R_{01}$  into the contributions defined by eqs 5–8 with molecular fragments as defined in Figure 1. We first analyze the magnitude and the sign of the  $R_{01}^{CO}$ ,  $R_{01}^{CO}$ ,  $R_{01}^{F}$ , and  $R_{01}^{R}$  contributions to the rotational strength  $R_{01}$  of modes 60 and 61. Table S9 shows that the magnitude of  $R_{01}^{CO}$  varies



**Figure 3.** Analysis of the two amide I modes of the conformer  $\alpha_3^*$ . Left: Nuclear displacement vectors. Right: EDTMs  $(\vec{E}_{01}^A \text{ and } \vec{E}_{01}^B)$  associated with the two carbonyl groups.

significantly between the conformers considered and that it is not always the dominant contribution to  $R_{01}$ . Similarly, we find that the relative magnitude and sign of  $R_{01}^{CO}$  and  $R_{01}^{COC}$ associated with a given normal mode depend sensitively on the molecular conformation. Regarding the magnitude of these two terms,  $R_{01}^{CO}$  is found to be larger than  $R_{01}^{COC}$  for all the conformers except for  $\alpha_4^*$  and  $\beta_4^*$ . Further, because modes 60 and 61 are not pure carbonyl stretching modes, the  $R_{01}^{IF}$  and  $R_{01}^{R}$ contributions to the rotational strength can have sizeable magnitudes and cannot always be neglected. Indeed, the analysis in Table S9 shows that these two terms are the reason why the VCD spectra of the conformers  $\alpha_2^*$ ,  $\alpha_3^*$ ,  $\beta_1$ , and  $\beta_4^*$  do not exhibit a bisignate VCD pattern in the amide I region. To illustrate this, we note that the generalized CO term  $R_{01}^{\text{GCO}}$  (the sum of  $R_{01}^{\text{CO}}$  and  $R_{01}^{\text{COC}}$ ) always has opposite signs for the two amide I modes of a given conformer. VCD spectra that exhibit a bisignate feature are thus associated with conformers in which  $R_{01}^{GCO}$  is the dominant contribution to the rotational strength. In contrast, for conformers where the sum of  $R_{01}^{IF}$  and  $R_{01}^{R}$  is comparable in magnitude with  $R_{01}^{GCO}$  and also has an opposite sign, VCD spectra may not exhibit a bisignate VCD feature, and this is exactly the case for the conformers  $\alpha_2^*$ ,  $\alpha_3^*$ ,  $\beta_1$ , and  $\beta_4^*$ .

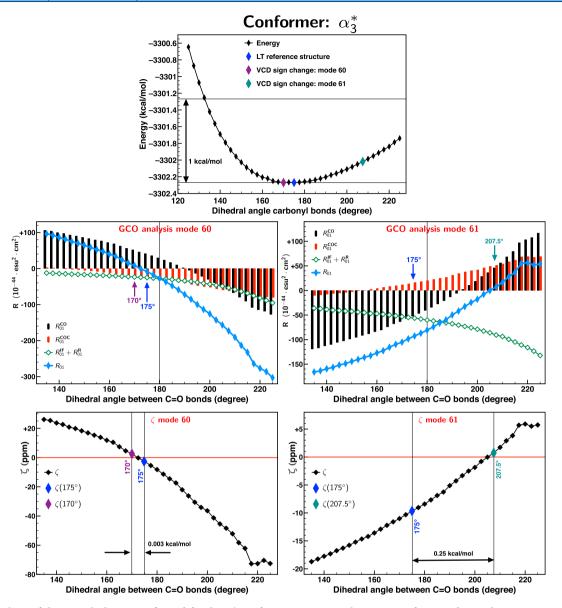
To understand why the  $R_{01}^{CO}$  term depends so sensitively on the molecular conformation, we consider the quantities that determine its magnitude, that is, the magnitude of the EDTMs associated with the two fragments as well as the angle  $\theta$ , the angle between the  $\vec{E}_{01}^A$  and  $\vec{E}_{01}^B$ , and  $\phi$ , the angle between  $\vec{Y}^{AB}$ and  $\vec{E}_{01}^A(j) \times \vec{E}_{01}^B(j)$ . We then find (see Table S10) that the EDTM associated with the dominant carbonyl bonds has a similar magnitude in all modes while the EDTM associated with the secondary carbonyl bonds exhibits much larger variations. This is a consequence of the fact that the involvement of the secondary C=O bonds in the mode motion varies from one conformer/mode to the other. The comparison of the  $|\mathbf{E}_{01}^{C}|\cdot|\mathbf{E}_{01}^{B}|$ ,  $\sin(\theta)\cos(\phi)$ , and  $R_{01}^{CO}$  values moreover shows that the large values observed for  $R_{01}^{CO}$  in the conformers  $\alpha_1$  and  $\beta_2$  are a consequence of the fact that in these conformers  $\vec{E}_{01}^{A}$  and  $\vec{E}_{01}^{B}$  are favorably oriented. Therefore, it should be clear that in the considered conformers, the relative orientation of  $\vec{E}_{01}^{A}$  and  $\vec{E}_{01}^{B}$  is more important than their magnitude. For example, the conformer  $\alpha_1$  has the largest  $R_{01}^{CO}$ values but these values are obtained by combining the smallest  $|\mathbf{E}_{01}^{A}|\cdot|\mathbf{E}_{01}^{B}|$  values with the largest  $\sin(\theta)\cos(\phi)$  values. On the other hand, because  $\alpha_{4}^{*}$  has the smallest  $\sin(\theta)\cos(\phi)$  value, the conformer  $\alpha_{4}^{*}$  has the smallest  $R_{01}^{CO}$  values even though it shows the largest  $|\mathbf{E}_{01}^{A}|\cdot|\mathbf{E}_{01}^{B}|$  value.

The analysis performed in this section has clearly shown why the standard CO model cannot be used to interpret the VCD bands in the amide I region. Furthermore, the GCO terms have the potential to be the dominant contributions to the rotational strength and thus determine the observed VCD pattern. Whether or not that is the case depends, however, on the structural details associated with the various conformations, that is, on whether the  $\vec{E}_{01}^A$ ,  $\vec{E}_{01}^B$ , and  $\vec{Y}^{GCO}$  are in favorable orientations. In addition, we have shown that hydrogen bonding may also influence the observed VCD patterns as it affects the relative position of the in-phase and out-of-phase modes.

LT Calculations. To investigate the robustness of the VCD sign computed for modes 60 and 61, we performed LT calculations in which a sequence of related constrained geometry optimizations is performed along a predefined geometrical path. Such calculations are quite useful as they serve as the first approach to account for structural variations as occurring at room temperature instead of performing full molecular dynamics calculations. Furthermore, they enable one to assess how the various factors contributing to the rotational strength of a particular mode depend on the geometry. Both aspects are highly important for the present study as they allow us to determine and understand the robustness of the amide I modes. In this respect, the dihedral angle  $\tau$  defined by the two carbonyl bonds is a parameter that can be expected to be particularly important.

Starting from the value in the fully optimized structure,  $\tau$  was varied by taking 20 steps of 2.5° in each direction. For LT structures that fell within an energy window of 1 kcal/mol with respect to the fully relaxed structure, VCD calculations were performed allowing us to monitor how  $R_{01}$  and its contributions as well as  $\zeta$  change during the LT scans. Although such an analysis was done for all eight conformers, we will in the following focus on the conformer  $\alpha_3^*$  because the GCO analysis of its VCD spectrum is representative for the analysis of the spectra of the other conformers and provides insight into both the stability of the amide I bands and the GCO mechanism. The analyses performed for the other conformers are given in Section 3 of the Supporting Information. These analyses lead to similar observations and conclusions as drawn here for the conformer  $\alpha_3^*$  and fully support the subsequent analysis.

The upper panel of Figure 4 shows the variation of energy during the LT scan. As can be seen, the energy varies smoothly with energy variations of less than 0.1 kcal/mol for angles between 160 and 190°. This is important as it implies that the relative orientation of the two carbonyl bonds may easily be influenced by perturbations such as the interaction with a solvent but will also be critically dependent on the use of different computational parameters. The middle panel of Figure 4 shows the variation of  $R_{01}$ ,  $R_{01}^{CO}$ ,  $R_{01}^{COC}$ , and  $R_{01}^{IF} + R_{01}^{R}$  associated with modes 60 and 61. We find that the overall change of  $R_{01}$  in the magnitude and the sign during the entire



**Figure 4.** Analysis of the LT calculations performed for the  $\alpha_3^*$  conformer. Upper panel: Variation of energy during the LT scan. Middle panel: Variation of the rotational strength  $R_{01}$  and its contributions during the LT scan. Lower panel: Variation of  $\zeta$  during the LT scan.

LT scan is quite significant:  $R_{01}$  changes from +96.5 to -303.2 for mode 60, and from -166.0 to +53.8 for mode 61. In contrast, the magnitude of the dipole strength  $D_{01}$  does not change appreciably during the LT scan (see Table S11). Therefore, the large variations observed for  $R_{01}$  are expected to be directly reflected in the  $\zeta$  ratios as is confirmed by the middle and lower panels of Figure 4.

These results thus indicate that in order to assess the robustness of the amide I modes, one should consider the profile of the potential energy surface and the position of the relaxed structure with respect to the points where  $R_{01}$  changes sign. For this purpose, we highlight in Figure 4 the position of the relaxed structure ( $\tau = 175^\circ$ ) in blue while the LT structures where  $R_{01}$  sign changes is highlighted in purple ( $\tau = 170^\circ$ ) and teal ( $\tau = 207.5^\circ$ ) for modes 60 and 61, respectively. As can be seen from the figure, the relaxed structure is located quite close to the point where  $R_{01}$  changes sign for mode 60 but far from the point where  $R_{01}$  changes sign for mode 61. Besides explaining the small VCD intensity in mode 60 (-8.5) and the large VCD intensity in mode 61 (-92.6), this analysis also

provides insight into the robustness of the two VCD bands. To change the sign of mode 60, it is sufficient to change  $\tau$  from 175 to 170°. Such a change is accompanied by a very small energy change (0.003 kcal/mol) and the sign computed for this mode is thus likely to depend on the choice of computational parameters. Mode 61, on the other hand, should be assigned as a robust mode because for a sign change to occur,  $\tau$  should be changed by more than 32.5°. Even though this change is accompanied by a relatively small energy change (0.25 kcal/mol), it is unlikely that the use of different basis sets or functionals will yield such a large variation in  $\tau$ .

The VCD robustness criterion proposed by Magyarfalvi and Gobi, on the other hand, would lead to the conclusion that both modes should be considered as nonrobust as  $\zeta(60) = -2.6$  and  $\zeta(61) = -9.7$ . To test the validity of our predictions, we have performed VCD calculations for  $\alpha_3^*$  using different computational parameters, that is, vacuum/BP86/OLYP/TZP/TZTP/DZP, COSMO(DMSO)/BP86/TZP, and COSMO(chloroform)/BP86/TZP. These calculations show that the sign changes only occur for mode 60, and even then

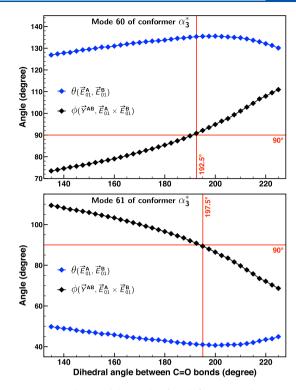
only between vacuum and COSMO calculations, for example, from -8.5 in the vacuum/BP86/TZP calculations to +23.5 in COSMO(chloroform) and +33.0 in COSMO(DMSO) calculations. These results thus support the robustness assignment based on the analysis shown in Figure 4. Similar conclusions are reached when the other conformers are considered (see Section S3 of the Supporting Information).

Variation of R<sub>01</sub> and Its Contributions during LT Scan. To further elucidate the dependence of the amide I VCD spectrum on the molecular structure, it is of interest to consider how R<sub>01</sub> varies during the LT scan from the perspective of the CO model (see Figure 4). Because the standard CO term  $R_{01}^{CO}$  and  $R_{01}$  exhibit similar trends, it is clear that  $R_{01}^{CO}$  is the driving force behind the changes observed in  $R_{01}$ . At the same time, it is important to note that  $R_{01}$  and  $R_{01}^{CO}$ often have different magnitudes and can also have different signs. This occurs because  $R_{01}^{\text{COC}}$ ,  $R_{01}^{\text{IR}}$ , and  $R_{01}^{\text{R}}$  are not negligible. As shown in Figure 3, the two amide I modes involve C=O stretches along with N-H and C-H bending motions. Because the latter contributions are not the same in the two modes,  $R_{01}$  changes sign at different  $\tau$  values for the two amide I modes, viz.  $R_{01}^{IF} + R_{01}^{R}$  has a very different value for both modes. This explains why the standard CO model is not able to predict accurately the sign and magnitude of  $R_{01}$ .

The  $R_{01}^{CO}$  term exhibits a behavior during the LT scan that is typical for CO terms. First, it is observed that when the two carbonyl bonds are in a favorable orientation, the magnitude of  $R_{01}^{CO}$  is very large. As the relative orientation of the C=O bonds changes,  $R_{01}^{CO}$  becomes gradually smaller and eventually changes sign. After the sign change, the magnitude of  $R_{01}^{CO}$ increases steadily as the relative orientation of the two bonds improves. Second, for a given LT structure,  $R_{01}^{CO}$  has different signs in the in-phase and out-of-phase modes. The only exception is found for the LT structure with  $\tau = 192.5^{\circ}$ , which is very close to the point where  $R_{01}^{CO}$  changes sign. This small discrepancy is due to the perturbed in-phase/out-of-phase behavior resulting from the involvement of the C-H and N-H bending motions. It is also important to notice that  $R_{01}^{CO}$ does not change sign when  $\tau$  is 180° as the standard CO model would predict. This is a consequence of the fact that the EDTM associated with a C=O bond is not oriented exactly along the bond. As a result, the angle  $\theta$  between  $\vec{E}_{01}^{A}$  and  $\vec{E}_{01}^{B}$ can be very different from the dihedral angle  $\tau$  determined by the two C=O bonds. Indeed, as illustrated in Figure 3, using the VCDtools GUI, the  $\vec{E}_{01}^{A}$  and  $\vec{E}_{01}^{B}$  vectors deviate from the direction of the bonds by roughly 20°.

To explain the sign change observed for  $R_{01}^{CO}$  during the LT scan, the variation of the angles  $\theta$  (the angle between  $\vec{E}_{01}^A$  and  $\vec{E}_{01}^B$ ) and  $\phi$  (the angle between  $\vec{Y}^{AB}$  and  $\vec{E}_{01}^A \times \vec{E}_{01}^B$ ) during the LT scan has been plotted and is shown in Figure 5. Surprisingly, for both amide I modes,  $\theta$  varies only minimally during the LT scan: variations of 90° in  $\tau$  lead to variations in  $\theta$ of less than 10°. The sign changes observed in Figure 4 for  $R_{01}^{CO}$ are thus not caused by  $\theta$  but by  $\phi$ . Indeed,  $\phi$  exhibits significant changes during the LT scan and crosses the 90° line at approximately 192.5 and 197.5° for modes 60 and 61, respectively. These angles coincide exactly with the angles  $\tau$  at which  $R_{01}^{CO}$  changes sign for both modes.

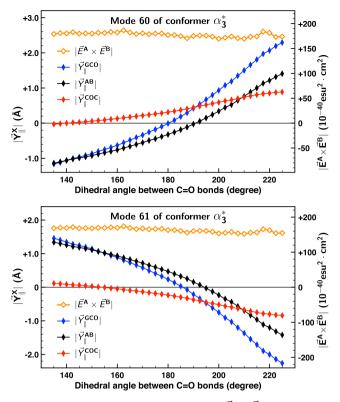
The analysis in Figure 4 also provides further insight into the CO correction term  $R_{01}^{\text{COC}}$ . During the LT scan, the  $R_{01}^{\text{COC}}$  term exhibits a typical CO behavior as its magnitude varies gradually, eventually changing sign and also has different signs for modes 60 and 61. Interestingly, it is found that  $R_{01}^{\text{COC}}$ 



**Figure 5.** Dependence of the angles  $\theta$  and  $\phi$  associated with normal modes 60 and 61 of conformer  $\alpha_3^*$  on the dihedral angle between the carbonyl bonds.

and  $R_{01}^{CO}$  change sign at different geometries. The latter observation has two important implications. First, the relative magnitudes and signs of  $R_{01}^{CO}$  and  $R_{01}^{COC}$  depend on the relative orientation of the C=O bonds. This means that in certain  $\tau$ ranges,  $R_{01}^{CO}$  and  $R_{01}^{COC}$  not only have opposite signs, but also the correction term is larger than the standard term. This situation occurs for mode 60, when  $\tau$  is in the (187.5, 192.5°) interval and for mode 61, when  $\tau$  is in the (182.5, 190°) interval. Second, the  $\vec{Y}^{AB}$  and  $\vec{Y}^{COC}$  vectors are not parallel. In refs 21 and 22, these vectors were considered to be parallel when quantifying the errors made in the standard CO model. While from a mathematical point of view this assumption is correct, the observation made above shows that in reality the two vectors do not need to be parallel. To this end, we note that—because the vector  $\vec{Y}^{COC}$  associated with a given mode is not uniquely defined<sup>22</sup>—it is better to consider its projection on the associated  $\vec{E}_{01}^{A} \times \vec{E}_{01}^{B}$  vector, which is fixed according to eq 13. We have, therefore, monitored in Figure 6 how the magnitude of the  $\vec{E}^A \times \vec{E}^B$  vector ( $|\vec{E}^A \times \vec{E}^B|$ ), and the projections of the associated  $\vec{Y}^{AB}$ ,  $\vec{Y}^{COC}$ , and  $\vec{Y}^{GCO}$  vectors on  $\vec{E}^A \times \vec{E}^B$  ( $|\vec{Y}^{AB}_{\parallel}|$ ,  $|\vec{Y}^{COC}_{\parallel}|$ , and  $|\vec{Y}^{GOC}_{\parallel}|$ , respectively) vary. Figure 6 shows that for both modes,  $|\vec{E}_{01}^A \times \vec{E}_{01}^B|$  shows only small variations during the LT scans. On the other hand,  $|\vec{Y}_{\parallel}^{AB}|$ ,  $|\vec{Y}_{\parallel}^{COC}|$ , and their sum  $|\vec{Y}_{\parallel}^{GOC}|$  vary gradually during the LT scan and eventually all change sign. This behavior, which mirrors closely the variations observed in Figure 5 for the angles  $\theta$  and  $\phi$ , further supports the conclusion that  $R_{01}^{COC}$  behaves like a typical CO term.

The above analysis has clearly shown that it is far from trivial to explain the intensities and signs of the amide I VCD bands in peptides. It is only by careful and detailed calculations that we can get a hold on the origin of the intensities and the robustness of the bands. GCO analyses in which the geometry



**Figure 6.** LT dependence of the magnitude of  $\vec{E}^{A} \times \vec{E}^{B}$  for modes 60 and 61 of conformer  $\alpha_{3}$  as well of the projections (||) of the associate  $\vec{Y}^{AB}$ ,  $\vec{Y}^{COC}$ , and  $\vec{Y}^{GCO}$  vectors on  $\vec{E}^{A} \times \vec{E}^{B}$ .

dependence and relative magnitudes of the  $R_{01}^{GCO}$ ,  $R_{01}^{CO}$ ,  $R_{01}^{CO}$ ,  $R_{01}^{IF}$ , and  $R_{01}^{R}$  contributes to  $R_{01}$  are unraveled and are in this respect indispensable.

**Amide II Modes.** As discussed, the comparison of the computed VCD spectra in Figure 2 shows that in contrast to the  $\alpha$  conformers, the  $\beta$  conformers display very intense bands in the amide II region. In this section, we aim to explain these differences.

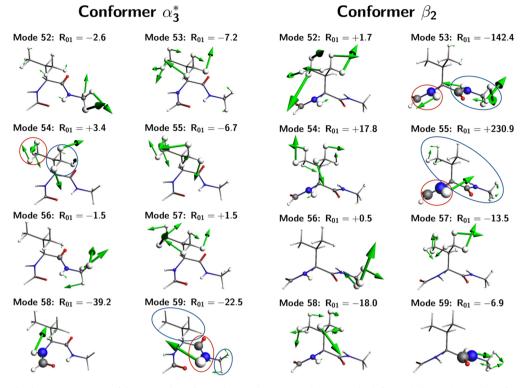
Normal Mode Analysis. The conformers of the For-ValNHMe peptide have eight normal modes in the 1430–  $1580 \text{ cm}^{-1}$  amide II spectral region. A detailed normal mode analysis (vide infra, see Section S4 in the Supporting Information) reveals that these modes can be grouped into three different categories (A, B, and C). Type A modes are standard amide II modes in which motion is mostly localized on the C-N-H groups and in which coupling between the motions of C-N-H groups of the C- and N-termini is typically very small. The frequencies of these modes show a clear difference between conformers with or without intramolecular H-bonding, the former being blue shifted with respect to the latter. This blue shift is well-known and finds its origin in the stabilization of the double-bond character of the C-N bond.<sup>30</sup> Type **B** modes are mixed modes not only consisting of C-H bending but also involving motion of the C-N-H groups. Because the magnitude of the atomic EDTMs associated with the C-N-H groups are comparable or even larger than those of the moving C-H bonds, B modes often have a strong amide II character. Two factors appear to determine the number of type **B** modes in a given conformer. The first one is the configuration of the conformer as  $\beta$ -sheet conformers have more type **B** modes than  $\alpha$ -helix conformers. The second one is the presence or absence of intramolecular H-bonding: conformers without intramolecular H-bonding are expected to have more B-type modes because in these conformers, the amide II modes (the A-type modes) have lower frequencies and thereby mix more with the C-H bending modes. Type C modes are typical C-H bending modes that do not mix appreciably with the amide II modes. For the present study, these modes are less important as they generally only have low VCD intensities. As summarized in Table 1 the 64 modes of the eight conformers in the amide-II spectral region can be grouped into 15 A-type, 19 B-type, and 30 C-type modes.

GCO Analysis. GCO analyses of the amide II modes of the eight conformers show that there are three factors that determine the magnitude of the VCD signals of these modes. The first one is the type of normal motion. Localized normal modes cannot trigger the GCO mechanism and as such most often have small or medium intensities. Mode 59 of  $\alpha_3^*$  is in this regard a typical example. Figure 7 shows that it is localized entirely on the N-terminal and has a medium-sized rotational strength even though it has a very large total EDTM. However,

Table 1. Summary of the Robustness Analysis Performed for the Modes in the Amide II Spectral Region: the Number of Normal Modes (No. NMs) and the Number of VCD Sign Changes (No. Sign Changes) Observed When Comparing Calculations Performed with Six Different Computational Parameters (BP86/DZP, BP86/TZP, BP86/TZ2P, OLYP/DZP, OLYP/DZP, and OLYP/TZ2P)

	A-type modes		B-type modes		C-type modes		all modes	
conformer	no. NMs	no. sign changes	no. NMs	no. sign changes	no. NMs	no. sign changes	no. NMs	no. sign changes
$\alpha_2^*$	2	1	0	0	6	1	8	2
$\alpha_3^*$	2	0	0	0	6	5	8	5
$lpha_4^*$	2	1	1	0	5	1	8	2
$eta_4^*$	1	1	5	1	2	1	8	3
total H bonding	7	3	6	1	19	8	32	12
$\alpha_1$	2	0	4	2	2	0	8	2
$\beta_1$	2	1	3	2	3	1	8	4
$\beta_2$	2	0	4	1	2	2	8	3
$\beta_3$	2	0	2	0	4	3	8	3
total rest	8	1	13	5	11	6	32	12
total all modes	15	4	19	6	30	14	64	24
total all $\alpha$ modes	8	2	5	2	19	7	32	11
total all $\beta$ modes	7	2	14	4	11	7	32	13

pubs.acs.org/JPCB



**Figure 7.** Nuclear displacement vectors of the normal modes in the amide II spectral region. The left panel shows the modes of conformer  $\alpha_3^*$  while the right panel shows the modes of conformer  $\beta_2$ . To illustrate the amide II character of each mode the radius of the atoms has been scaled with the magnitude of the associated atomic EDTMs. The fragments used in the GCO analysis performed in Table 2 for modes  $54(\alpha_3^*)$ ,  $59(\alpha_3^*)$ ,  $53(\beta_2)$ , and  $55(\beta_2)$  are highlighted in red and blue.

because the rest of the atoms in the molecule are not involved in the normal mode motion, the EDTM associated with the rest of the molecule is for all practical purposes considered to be zero. As a result, the large EDTM associated with the C– N–H atoms does not have the possibility to couple with the EDTM of another moiety (see GCO analysis in Table 2). Delocalized modes, on the other hand, have the potential to trigger the GCO mechanism and to have very large VCD intensities.

# Table 2. GCO Analysis Results of Amide II Modes of Conformers $\alpha_1^*$ and $\beta_2^{\ a}$

conf.	NM	freq.	$D_{01}$	<i>R</i> <sub>01</sub>	$R_{01}^{GCO}$	$R_{01}^{\mathrm{IF}}$	$R_{01}^{\mathbf{R}}$		
$\beta_2$	53	1445.6	416.7	-142.4	-145.8	+16.1	-12.6		
$\beta_2$	55	1454.2	279.6	+230.9	+185.4	+49.1	-3.6		
$\alpha_3^*$	54	1449.3	4.3	+3.7	+0.2	+4.0	-0.5		
$\alpha_3^*$	59	1565.6	387.8	-24.5	+2.8	-21.5	-5.8		
<sup>a</sup> GCO fragments are defined in Figure 7. Frequencies are given in									

cm<sup>-1</sup>, dipole strength  $D_{01}$  in 10<sup>-40</sup> esu<sup>2</sup>·cm<sup>2</sup>, and rotational strength  $R_{01}$  and its contributions  $R_{01}^{\text{GCO}}$ ,  $R_{01}^{\text{IF}}$ , and  $R_{01}^{\text{R}}$  in 10<sup>-44</sup> esu<sup>2</sup>·cm<sup>2</sup>.

The second factor is the involvement of the C–N–H group in the normal mode motion. Typical amide II modes involve stretches of the C–N bond along with bending of the N–H bond. Because of the polar character of these two types of bonds, the associated atomic EDTMs are significantly larger than those associated with the C–H bending motions. As indicated previously, even small motions of the atoms in the C–N–H groups induce atomic EDTMs that are comparable to—or larger than—those associated with C–H bending motions with large amplitudes. Because large  $R_{01}^{\rm GCO}$  values require at least one of the interacting EDTMs to be large, intense VCD bands are thus associated with normal modes that are localized on at least one of the C–N–H groups. This point is further illustrated by the GCO analyses reported in Table 2. Modes 53 and 55 of conformer  $\beta_2$  exhibit very intense VCD signals. In line with the previous reasoning, we find that in both cases,  $R_{01}^{\text{GCO}}$  is significantly larger than the other two contributions to the rotational strengths. We can, therefore, conclude that the large VCD intensities are the result of the GCO coupling between the EDTMs associated with the C–N–H groups. In mode 54 of the conformer  $\alpha_3^*$ , on the other hand, we have only C–H bending movements which generate weak EDTMs that are unable to couple. As a result, the mode is accompanied by only a small  $R_{01}^{\text{GCO}}$  value and a small total VCD intensity.

The final factor concerns the relative orientation of the interacting EDTMs. This is a consequence of eqs 9, 12 and 13. As can be seen in Figure 7, not all modes involving movements of the C–N–H group along with C–H bending motions have large rotational strengths. Instead, large rotational strengths are only observed for modes where the interacting EDTMs have a favorable orientation with respect to each other.

The normal mode and GCO analyses discussed above clearly explain why the  $\beta$  conformers exhibit very intense bands in the amide II region while  $\alpha$  conformers do not. The intense VCD signals observed in this region are determined by large GCO terms which in turn are determined by the through-space interaction of large EDTMs. As a result, intense VCD signals are more likely to be observed in mixed modes that are delocalized and characterized by large EDTMs, that is, in **B**-type modes. The **A**-type modes—which are localized—and the

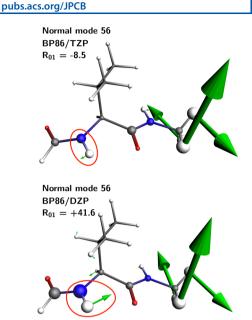
C-type modes—which are delocalized but have small EDTMs—most often have small VCD intensities.

VCD Robustness in the Amide II Region. Unlike the amide I modes, the stability of most of the intense VCD bands in the amide II spectral region cannot be investigated consistently by LT calculations because the delocalized nature of the GCO fragments makes it difficult to find LT parameters that allow one to change the computed VCD sign without changing the conformation. For this reason, the robustness of the computed VCD sign was investigated by performing geometry optimizations and VCD calculations with six different computational parameters (BP86/DZP, BP86/TZP, BP86/TZ2P, OLYP/DZP, OLYP/TZP, and OLYP/TZ2P). To identify VCD sign changes induced by the use of different computational parameters, the normal modes predicted by the six sets of calculations have been mapped by computing normal mode overlaps.<sup>26</sup>

The results of these analyses are summarized in Table 1, which shows that 24 of the 64 modes in the amide II spectral region change sign upon changing computational parameters. Interestingly, no significant differences are observed between  $\alpha$ -helix and  $\beta$ -sheet conformers, nor between structures with and without intramolecular hydrogen bonding. When grouping the modes into **A**, **B**, and **C** types, VCD sign changes are observed for 4 out of the 15 A-type modes (27%), 7 of the 19 B-type modes (37%), and 14 of the 30 C-type modes (47%). It is important to realize that the number of **B** and **C** modes found in a given conformer depends on the computational parameters used in the calculations, in contrast to the number of **A** modes, which is not affected by the change of computational parameters.

Type **B** modes are predominantly C–H bending modes that also involve small movements of the C-N-H group(s). This weak coupling between the C-H bending motions and the movements localized on the C-N-H group is very sensitive to the computational parameters, and this has far-reaching consequences. To illustrate this point, we consider mode 56 of the conformer  $\beta_3$  whose nuclear displacement vectors calculated at the BP86/DZP and BP86/TZP levels are compared in Figure 8. As can be seen, both calculations predict the mode to be a C-H bending mode localized on the C-terminal methyl group and involving also small N-H bending contributions. For the H atoms moving with large amplitudes, the two sets of calculations make similar predictions for the nuclear displacement vectors. In contrast, the two sets of calculations make quite different predictions for the small-amplitude C-H and N-H bending motions. Not only is localization of the motion on the N-H bond different (4.4% in BP86/TZP vs 11.1% in BP86/DZP), but the two calculations also predict a different phase for the coupling between the C–H and N–H bending motions (see Figure 8). The consequence of this in-phase/out-of-phase behavior is quite dramatic as it leads to opposite VCD signs for the two modes.

From the perspective of the GCO analysis, the two modes in Figure 8 are quite different. Both calculations predict that the  $R_{01}^{GCO}$  term is the largest contribution to the rotational strength, but its sign and magnitude are different in the two calculations (see Table 3). The difference in the sign is a consequence of the different phase observed in the mode coupling, while the difference in magnitude is caused by the different involvement of the N–H bending motion in the normal mode motion. Further, we note that both sets of calculations predict large



**Figure 8.** Normal mode 56 of ForValNHMe- $\beta_3$  calculated at the BP86/TZP (top) and BP86/DZP (bottom) levels. The green arrows indicate the direction and amplitude of the vibration. The size of the atoms corresponds to the magnitude of the associated atomic EDTM.

absolute  $\zeta$  values which provides further evidence that the magnitude of  $\zeta$  does not reflect the stability of the computed VCD sign when the VCD intensity of a mode is determined by the GCO mechanism.

Mode 56 of the conformer  $\beta_3$  is representative for most modes in the amide II spectral region of all conformers. Analysis of the data shows that 15 of the 24 observed VCD sign changes are caused by computational uncertainties related to the localization of the normal mode motion on the C–N–H group while the remaining sign changes are associated with modes that have very weak VCD intensities (e.g.  $|R_{01}| < 10$ ). Similarly, there is hardly any correlation between the magnitude of  $\zeta$  and the observed VCD sign changes which occur for both small and high  $\zeta$  values. In fact, in some conformers ( $\alpha_3^*$ ,  $\beta_1$ ,  $\beta_3$ , and  $\beta_4^*$ ) the modes with a high  $\zeta$  value have a nonrobust VCD sign. From the perspective of the GCO mechanism, this is not surprising. Large  $\zeta$  values are often correlated with large  $R_{01}^{GCO}$  contributions but  $R_{01}^{GCO}$  depends sensitively on the computational parameters used in calculations as has been shown above.

Finally, we note that the present study has also shown that differences between the relaxed structures predicted by various computational parameters may induce VCD sign changes. For example, the optimized structures with the DZP basis set do not contain the H-bond in the conformers  $\alpha_2^*$  and  $\alpha_3^*$ . As a result, some of the normal modes show VCD bands with opposite signs. To avoid such errors, we have employed, in the present study, the relaxed TZP structures as the starting point for the geometry optimizations performed using DZP. With this approach, we find very similar relaxed structures predicted by the six sets of calculations.

#### CONCLUSIONS

Using the ForValNHMe peptide as a prototypical peptide, we have presented in this work a thorough analysis of the amide I and amide II VCD bands of peptides. By combining the recently developed VCD GCO formalism with a novel normal

Table 3. Normal Mode and GCO Analyses of Mode 56 of Conformer  $\beta_3$  Predicted at the BP86/TZP and BP86/DZP Levels<sup>*a*</sup>

calc.	conf.	mode	overlap	freq.	ζ	$D_{01}$	<i>R</i> <sub>01</sub>	$R_{01}^{GCO}$	$R_{01}^{\mathrm{IF}}$	$R_{01}^{\mathbf{R}}$
BP86/TZP	$\beta_3$	56	1.00	1457.5	-117.7	7.2	-8.5	-5.33	-4.42	+1.24
BP86/DZP	$\beta_3$	56	0.87	1457.3	+44.6	93.3	+41.6	+41.58	-0.47	+0.49
<sup>a</sup> C-terminal methyl group was chosen as GCO fragment A and the N-terminal C–N–H group as GCO fragment B (see Figure 8). Frequency is										

given in cm<sup>-1</sup>, dipole strength  $D_{01}$  in  $10^{-40}$  esu<sup>2</sup>·cm<sup>2</sup>, rotational strength  $R_{01}$  and its contributions  $R_{01}^{GCO}$ ,  $R_{01}^{IF}$ , and  $R_{01}^{R}$  in  $10^{-44}$  esu<sup>2</sup>·cm<sup>2</sup>, and the dissymmetry factor  $\zeta$  in ppm. The normal mode overlap is dimensionless.

modes analysis that monitors the magnitude of the nuclear displacement together with the magnitude of the vibrational DTMs, we have been able to explain all the differences observed for the  $\alpha$  and  $\beta$  conformers in these spectral regions. The performed analyses have revealed which electronic and spatial parameters determine the intensity of the amide I and amide II VCD bands, and have highlighted the critical role played by intramolecular hydrogen-bonding. As shown, the GCO mechanism is responsible for all VCD bands that stand out in both spectral regions. Essentially, this means that the same underlying mechanism, that is, the through-space coupling of electric dipoles, is responsible for the intense VCD bands observed in the amide I and amide II spectral regions. As a result, both amide I and amide II VCD bands are found to be very sensitive to the relative orientation of the Cand N-termini and on the localization of the normal mode motion on these two termini. This, in turn, implies that one should be cautious to determine the stability of computed VCD signs by evaluating the magnitude of the  $\zeta$  ratio, as proposed previously.

The present study has also provided additional insight into the VCD GCO mechanism. By studying the trends of the  $R_{01}^{CO}$ and  $R_{01}^{COC}$  terms during LT scans, new insight was gained regarding the CO correction term  $R_{01}^{COC}$ . It was shown that  $R_{01}^{COC}$  and the standard  $R_{01}^{CO}$  CO term behave in a very similar fashion during the LT scans. However, the behavior of the  $R_{01}^{COC}$  term is shifted with respect to  $R_{01}^{CO}$ . Therefore,  $R_{01}^{COC}$  can be larger than  $R_{01}^{CO}$  while at the same time have a different sign. In other words, terms that are neglected in the standard CO model can not only be larger than what this model considers to be important but also have the opposite sign.

The detailed insight that has been obtained in the VCD properties of the ForValNHMe peptide is essential to further our understanding of the VCD spectra of peptides in general, and how these spectra reflect key details of their electronic and spatial structure. At an even more general level, it is clear that supported by theoretical GCO analyses VCD spectroscopy is coming of age and becoming an indispensable tool for studies of the stereochemistry of chiral systems.

#### METHODS

All geometry optimizations,<sup>31</sup> VCD calculations,<sup>32</sup> and LT scans were performed at the DFT level of theory using the ADF program package.<sup>23,33</sup> Two exchange–correlation functionals (BP86<sup>34,35</sup> and OLYP<sup>36,37</sup>) and three different basis sets (the ADF DZP, TZP, and TZ2P basis sets<sup>38</sup>) were used to investigate the effects of different parameters. For the larger part of the calculations, BP86/TZP was used, and these as such served as a reference when comparing between different sets of parameters. In order to investigate solvent effects, the COSMO-solvation model was used as well.<sup>39,40</sup>

To determine the dependence of the calculated VCD sign on the relative orientation of the coupled molecular fragments, LT calculations<sup>41,42</sup> were performed. In these calculations, the molecule structure was changed systematically starting from the fully relaxed structure by varying a LT parameter (e.g., bond length, angle, or dihedral angle) in predefined steps. After each LT step, a constrained geometry optimization was performed by relaxing all structural parameters except the LT parameter which was kept at the predefined value.

The VCD spectra, normal modes, and EDTMs were analyzed and visualized using a new implementation of the VCDtools program<sup>43</sup> in the ADF GUI program.<sup>25</sup> VCDtools computes the rotational strength  $R_{01}$  and its various contributions, for example  $R^{GCO}$ ,  $R^{IF}$ ,  $R^R$ ,  $R^{CO}$ , and  $R^{COC}$  using atomic polar tensors, atomic axial tensors, and nuclear displacement vectors obtained from DFT calculations.

#### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jpcb.9b11261.

Details about the used molecular structures, the amide I GCO analysis, the LT calculations, and normal mode analyses of amide-II (PDF)

#### AUTHOR INFORMATION

#### **Corresponding Author**

Valentin P. Nicu – Department of Environmental Science, Physics, Physical Education and Sport, Lucian Blaga University of Sibiu, 550012 Sibiu, Romania; Email: vp.nicu@gmail.com

#### Authors

- Mark A. J. Koenis Van't Hoff Institute for Molecular Sciences, University of Amsterdam, 1098 XH Amsterdam, The Netherlands
- Lucas Visscher Amsterdam Center for Multiscale Modeling, Section Theoretical Chemistry, Faculty of Sciences, Vrije Universiteit Amsterdam, 1081 HV Amsterdam, The Netherlands; Ocid.org/0000-0002-7748-6243
- Wybren J. Buma Van't Hoff Institute for Molecular Sciences, University of Amsterdam, 1098 XH Amsterdam, The Netherlands; Institute for Molecules and Materials, FELIX Laboratory, Radboud University, 6525 ED Nijmegen, The Netherlands; ◎ orcid.org/0000-0002-1265-8016

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.jpcb.9b11261

#### Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

M.A.J.K., L.V., and W.J.B. acknowledge financial support from NWO in the framework of the Fund New Chemical Innovations (NWO project no. 731.014.209). V.P.N. acknowledges funding from the UEFISCDI (PN-III-P1-1.1-TE-2016-1049, contract no. 46/2018).

pubs.acs.org/JPCB

#### REFERENCES

(1) Wang, F.; Polavarapu, P. L.; Schurig, V.; Schmidt, R. Absolute configuration and Conformational Analysis of a Degradation Product of Inhalation Anesthetic Sevoflurane: A Vibrational Circular Dichroism Study. *Chirality* **2002**, *14*, 618–624.

(2) Keiderling, T. A. Protein and Peptide Secondary Structure and Conformational Determination with Vibrational Circular Dichroism. *Curr. Opin. Chem. Biol.* **2002**, *6*, 682–688.

(3) Urbanová, M. Bioinspired Interactions Studied by Vibrational Circular Dichroism. *Chirality* **2009**, *21*, E215–E230.

(4) Holzwarth, G.; Chabay, I. Optical Activity of Vibrational Transitions: A Coupled Oscillator Mode. J. Chem. Phys. 1972, 57, 1632–1635.

(5) Bouř, P.; Sopková, J.; Bednárová, L.; Maloň, P.; Keiderling, T. A. Transfer of Molecular Property Tensors in Cartesian Coordinates: A New Algorithm for Simulation of Vibrational Spectra. *J. Comput. Chem.* **1997**, *18*, 646–659.

(6) Choi, J.-H.; Kim, J.-S.; Cho, M. Amide I Vibrational Circular Dichroism of Polypeptides: Generalized Fragmentation Approximation Method. *J. Chem. Phys.* **2005**, *122*, 174903.

(7) Jose, K. V. J.; Raghavachari, K. Molecules-in-molecules fragmentbased method for the calculation of chiroptical spectra of large molecules: Vibrational circular dichroism and Raman optical activity spectra of alanine polypeptides. *Chirality* **2016**, *28*, 755–768.

(8) Kessler, J.; Andrushchenko, V.; Kapitán, J.; Bouř, P. Insight into vibrational circular dichroism of proteins by density functional modeling. *Phys. Chem. Chem. Phys.* **2018**, *20*, 4926–4935.

(9) Huang, R.; Kubelka, J.; Barber-Armstrong, W.; Silva, R. A. G. D.; Decatur, S. M.; Keiderling, T. A. Nature of Vibrational Coupling in Helical Peptides: An Isotopic Labeling Study. *J. Am. Chem. Soc.* **2004**, *126*, 2346–2354.

(10) Welch, W. R. W.; Kubelka, J.; Keiderling, T. A. Infrared, Vibrational Circular Dichroism, and Raman Spectral Simulations for  $\beta$ -Sheet Structures with Various Isotopic Labels, Interstrand, and Stacking Arrangements Using Density Functional Theory. *J. Phys. Chem. B* **2013**, *117*, 10343–10358.

(11) Bour, P.; Keiderling, T. A. Structure, Spectra and the Effects of Twisting of  $\beta$ -Sheet Peptides. A Density Functional Theory Study. *J. Mol. Struct.: THEOCHEM* **2004**, 675, 95–105.

(12) Measey, T. J.; Schweitzer-Stenner, R. Vibrational Circular Dichroism as a Probe of Fibrillogenesis: The Origin of the Anomalous Intensity Enhancement of Amyloid-like Fibrils. *J. Am. Chem. Soc.* **2011**, *133*, 1066–1076.

(13) Schweitzer-Stenner, R. Simulated IR, Isotropic and Anisotropic Raman, and Vibrational Circular Dichroism Amide I Band Profiles of Stacked  $\beta$ -Sheets. J. Phys. Chem. B **2012**, 116, 4141–4153.

(14) Barth, A.; Haris, P. I. Biological and Biomedical Infrared Spectroscopy; IOS Press: BV Amsterdam, 2009; Vol. 2.

(15) Choi, J.-H.; Hahn, S.; Cho, M. Vibrational Spectroscopic Characteristics of Secondary Structure Polypeptides in Liquid Water: Constrained MD Simulation Studies. *Biopolymers* **2006**, *83*, 519–536.

(16) Andrushchenko, V.; Bouř, P. Circular Dichroism Enhancement in Large DNA Aggregates Simulated by a Generalized Oscillator Model. J. Comput. Chem. **2008**, *29*, 2693–2703.

(17) Ma, S.; Cao, X.; Mak, M.; Sadik, A.; Walkner, C.; Freedman, T. B.; Lednev, I. K.; Dukor, R. K.; Nafie, L. A. Vibrational Circular Dichroism Shows Unusual Sensitivity to Protein Fibril Formation and Development in Solution. *J. Am. Chem. Soc.* **2007**, *129*, 12364–12365.

(18) Goncharova, I.; Sykora, D.; Urbanová, M. Association of Biotin with Silver (I) in Solution: a Circular Dichroism Study. *Tetrahedron: Asymmetry* **2010**, *21*, 1916–1920.

(19) Góbi, S.; Magyarfalvi, G.; Tarczay, G. VCD Robustness of the Amide-I and Amide-II Vibrational Modes of Small Peptide Models. *Chirality* **2015**, *27*, 625–634.

(20) Góbi, S.; Magyarfalvi, G. Reliability of Computed Signs and Intensities for Vibrational Circular Dichroism Spectra. *Phys. Chem. Chem. Phys.* **2011**, *13*, 16130–16133.

(21) Nicu, V. P. Revisiting an Old Concept: the Coupled Oscillator Model for VCD. Part 2: Implications of the Generalised Coupled Oscillator Mechanism for the VCD Robustness Concept. *Phys. Chem. Chem. Phys.* 2016, 18, 21213–21225.

(22) Nicu, V. P. Revisiting an Old Concept: the Coupled Oscillator Model for VCD. Part 1: The Generalised Coupled Oscillators Mechanism and its Intrinsic Connection to the Strength of VCD Signals. *Phys. Chem. Chem. Phys.* **2016**, *18*, 21202–21212.

(23) Amsterdam Density Functional program. http://www.scm.com (accessed Nov 8, 2017).

(24) Stephens, P. J. Theory of Vibrational Circular Dichroism. J. Phys. Chem. 1985, 89, 748-752.

(25) Koenis, M. A. J.; Visser, O.; Visscher, L.; Buma, W. J.; Nicu, V. P. The GUI Implementation of VCDtools, a Program to Analyse Computed Vibrational Circular Dichroism Spectra. *J. Chem. Inf. Model.* **2020**, *60*, 259–267.

(26) Nicu, V. P.; Baerends, E. J. Robust Normal Modes in Vibrational Circular Dichroism Spectra. *Phys. Chem. Chem. Phys.* **2009**, *11*, 6107–6118.

(27) Magyarfalvi, G.; Tarczay, G.; Vass, E. Vibrational Circular Dichroism. *Wiley Interdiscip. Rev.: Comput. Mol. Sci.* 2011, 1, 403–425.

(28) Góbi, S.; Vass, E.; Magyarfalvi, G.; Tarczay, G. Effects of Strong and Weak Hydrogen Bond Formation on VCD Spectra: a Case Study of 2-Chloropropionic Acid. *Phys. Chem. Chem. Phys.* **2011**, *13*, 13972–13984.

(29) Góbi, S.; Magyarfalvi, G. Reliability of Computed Signs and Intensities for Vibrational Circular Dichroism Spectra. *Phys. Chem. Chem. Phys.* **2011**, *13*, 16130–16133.

(30) Myshakina, N. S.; Ahmed, Z.; Asher, S. A. Dependence of Amide Vibrations on Hydrogen Bonding. *J. Phys. Chem. B* 2008, 112, 11873–11877.

(31) Swart, M.; Bickelhaupt, F. M. QUILD: QUantum-regions Interconnected by Local Descriptions. J. Comput. Chem. 2008, 29, 724–734.

(32) Nicu, V. P.; Neugebauer, J.; Wolff, S. K.; Baerends, E. J. A Vibrational Circular Dichroism Implementation within a Slater-Type-Orbital Based Density Functional Framework and its Application to Hexa- and Hepta-Helicenes. *Theor. Chem. Acc.* **2008**, *119*, 245–263.

(33) te Velde, G.; Bickelhaupt, F. M.; Baerends, E. J.; Fonseca Guerra, C.; van Gisbergen, S. J. A.; Snijders, J. G.; Ziegler, T. Chemistry with ADF. J. Comput. Chem. 2001, 22, 931–967.

(34) Becke, A. D. Density-Functional Exchange-Energy Approximation with Correct Asymptotic Behavior. *Phys. Rev. A* **1988**, *38*, 3098–3100.

(35) Perdew, J. P. Density-Functional Approximation for the Correlation Energy of the Inhomogeneous Electron Gas. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1986**, 33, 8822–8824.

(36) Handy, N. C.; Cohen, A. J. Left-Right Correlation Energy. *Mol. Phys.* 2001, 99, 403-412.

(37) Hamprecht, F. A.; Cohen, A. J.; Handy, N. C. Development and Assessment of New Exchange-Correlation Functionals. *J. Chem. Phys.* **1998**, *109*, 6264–6271.

(38) Van Lenthe, E.; Baerends, E. J. Optimized Slater-Type Basis Sets for the Elements 1-118. *J. Comput. Chem.* **2003**, *24*, 1142–1156. (39) Klamt, A.; Schüürmann, G. COSMO: a New Approach to Dielectric Screening in Solvents with Explicit Expressions for the Screening Energy and its Gradient. *J. Chem. Soc., Perkin Trans.* **2 1993**, 799–805.

(40) Pye, C. C.; Ziegler, T. An Implementation of the Conductor-Like Screening Model of Solvation Within the Amsterdam Density Functional Package. *Theor. Chem. Acc.* **1999**, *101*, 396–408.

(41) Heshmat, M.; Baerends, E. J.; Polavarapu, P. L.; Nicu, V. P. The Importance of Large-Amplitude Motions for the Interpretation of Mid-Infrared Vibrational Absorption and Circular Dichroism Spectra: 6,6'-Dibromo-[1,1'-binaphthalene]-2,2'-diol in Dimethyl Sulfoxide. *J. Phys. Chem. A* **2014**, *118*, 4766–4777.

(42) Passarello, M.; Abbate, S.; Longhi, G.; Lepri, S.; Ruzziconi, R.; Nicu, V. P. Importance of C\*–H Based Modes and Large Amplitude Motion Effects in Vibrational Circular Dichroism Spectra: the Case of

pubs.acs.org/JPCB

Article

the Chiral Adduct of Dimethyl Fumarate and Anthracene. J. Phys.

*Chem. A* 2014, *118*, 4339–4350.
(43) Nicu, V. P.; Neugebauer, J.; Baerends, E. J. Effects of Complex Formation on Vibrational Circular Dichroism Spectra. J. Phys. Chem. A 2008, 112, 6978–6991.