

IONIC LIQUIDS FOR CAPTURE AND ELECTROCHEMICAL CONVERSION OF CO₂ - CIS 2021

Original

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IONIC LIQUIDS FOR CAPTURING ELECTROCHEMICAL CONVERSION

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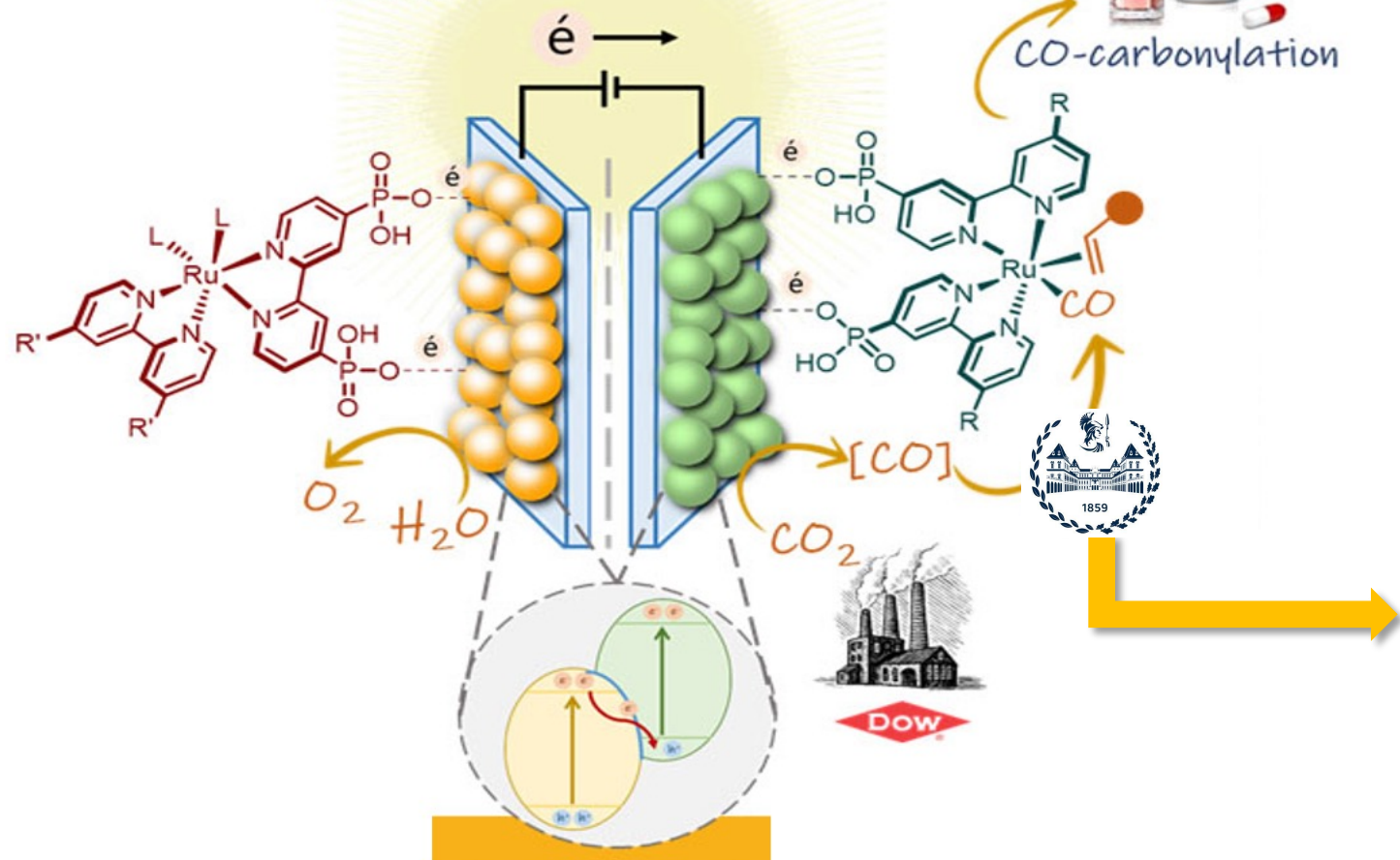
CIS-2021
Young Researchers

Short Talk ID:
Section 9: CO₂

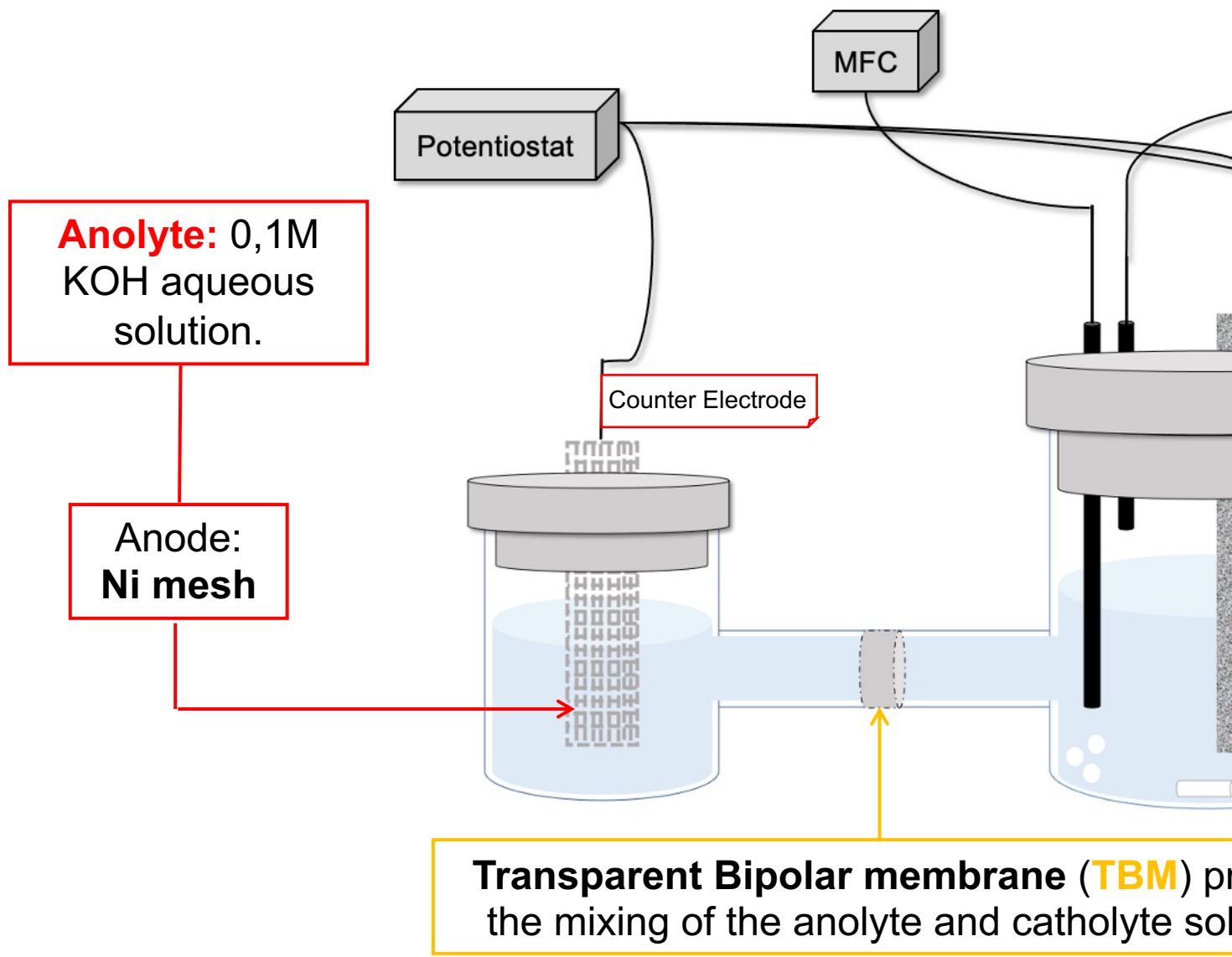
Introduction and aim of the work

SunCO₂Chem

Photoelectrochemical Reactor

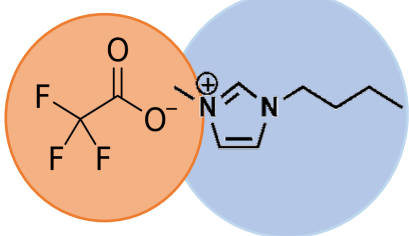


Materials and methods

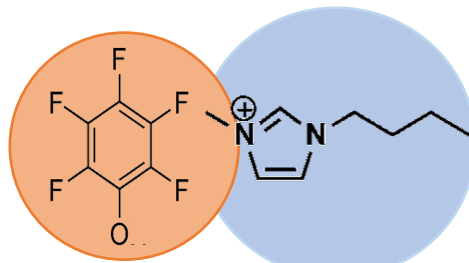


H-type cell configuration

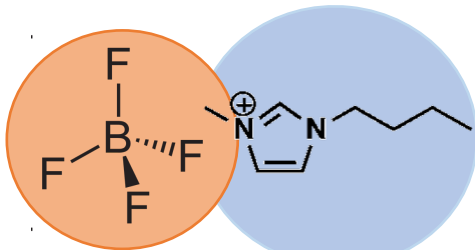
Results and discussion



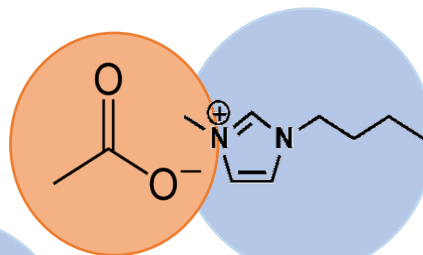
[BMIM][CO₂CF₃]



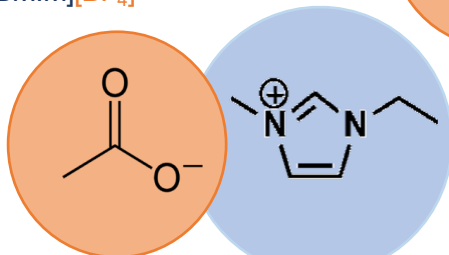
[BMIM][5FF]



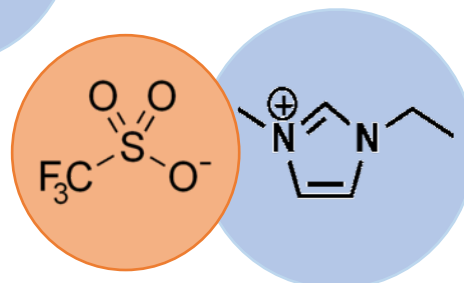
[BMIM][BF₄]



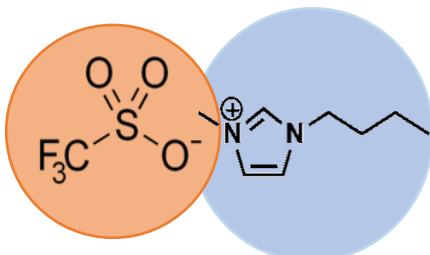
[BMIM][CO₂CH₃]



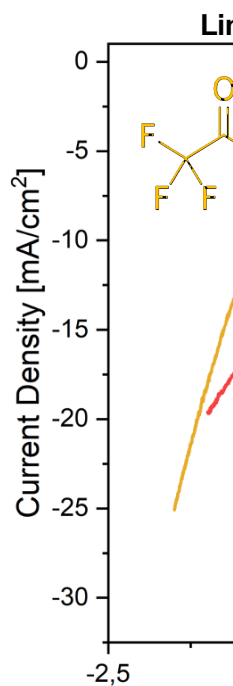
[EMIM][CO₂CH₃]



[EMIM][SO₃CF₃]

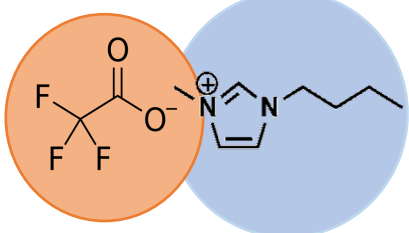


[BMIM][SO₃CF₃]

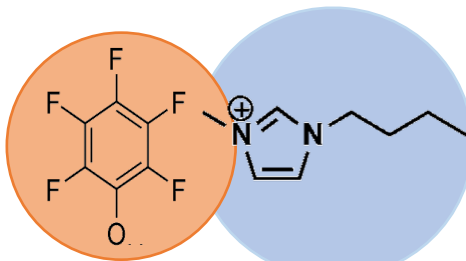


- **CO₂ solubility** s
- A higher fluorinat
- CO₂ solubility an
- It might be relate

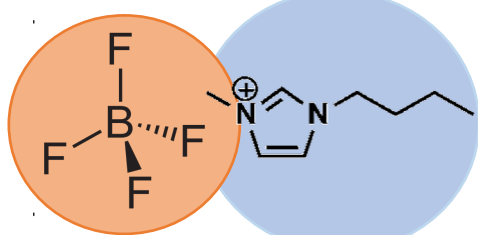
Results and discussion



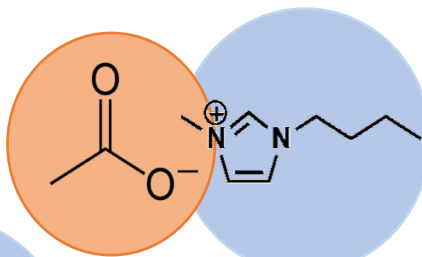
[BMIM][CO₂CF₃]



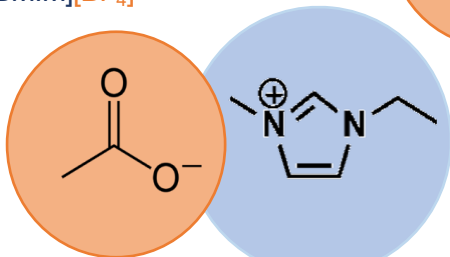
[BMIM][5FF]



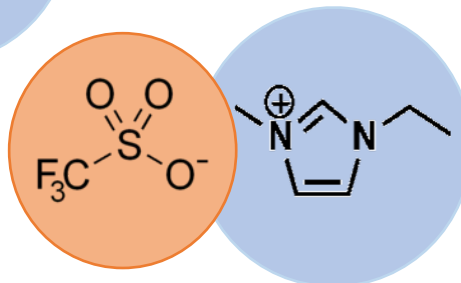
[BMIM][BF₄]



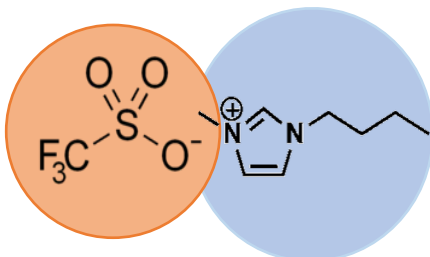
[BMIM][CO₂CH₃]



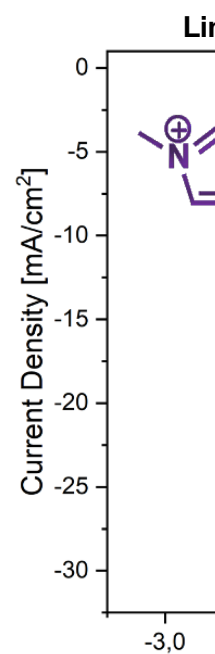
[EMIM][CO₂CH₃]



[EMIM][SO₃CF₃]

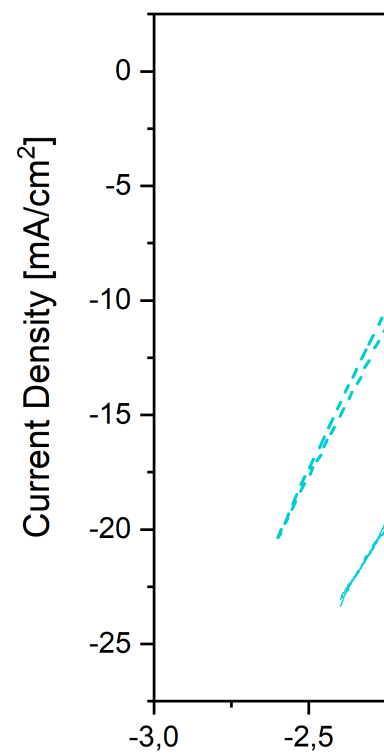
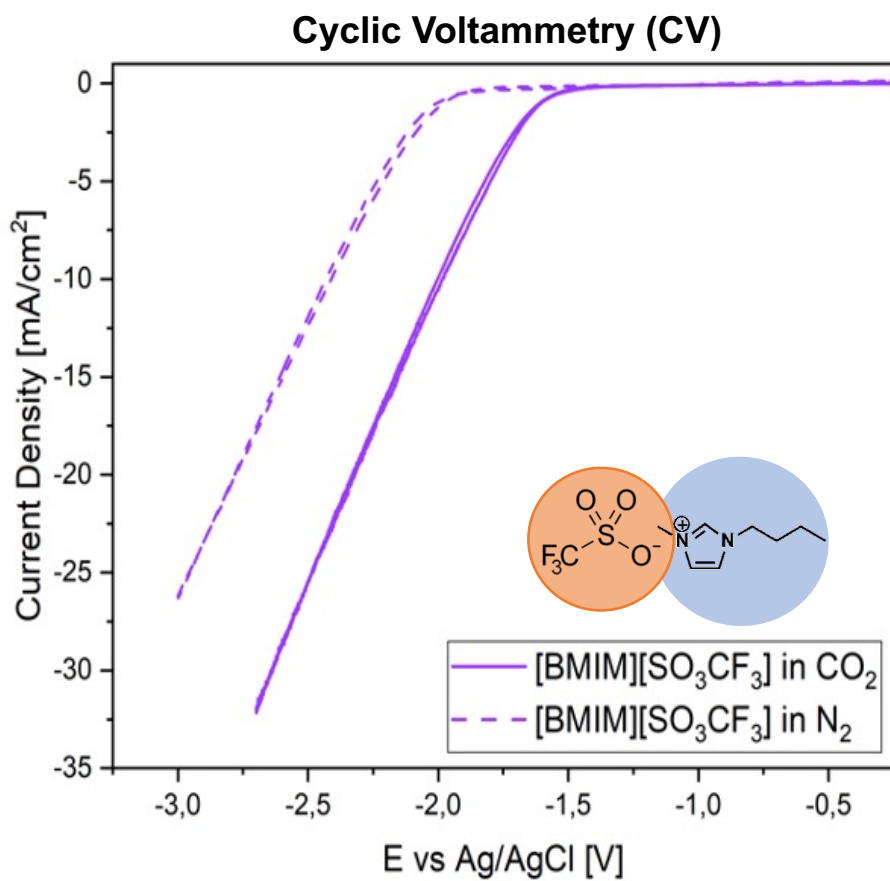


[BMIM][SO₃CF₃]



- The alkyl chain o
- When the **alky** ring finds a **mor** to reduce and dioxide molecule negative onset p

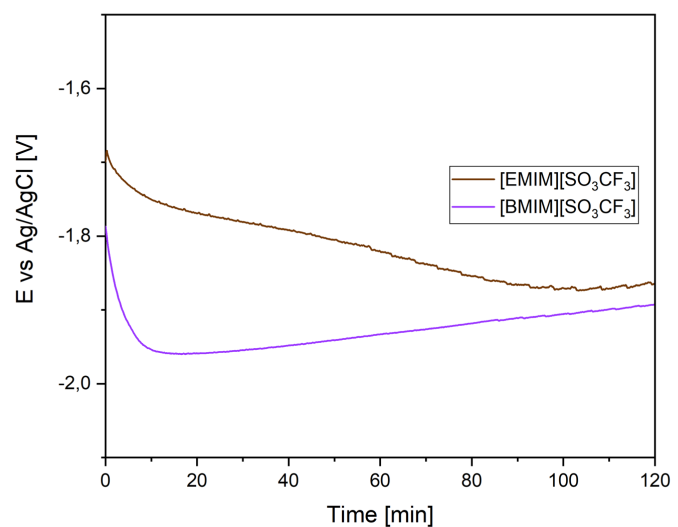
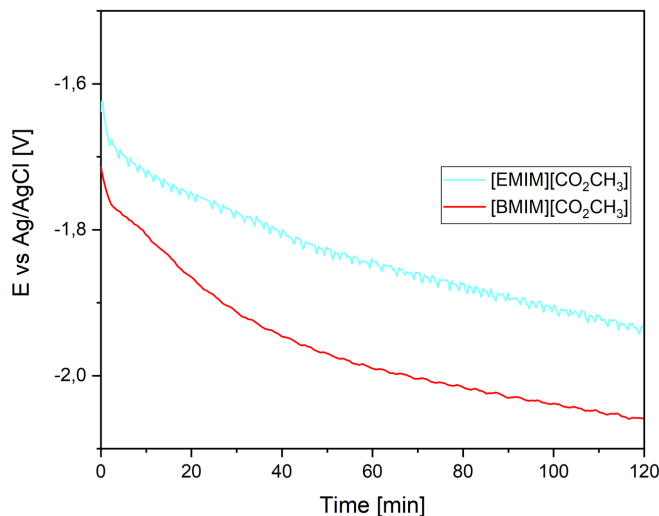
Results and discussion



✓ **CVs' highlights:** Onset potential of all the ILs is shifted to less negative potentials with CO₂.

Results and discussion

Chronopotentiometry (CP) in CO₂, t=120 min, -20 mA

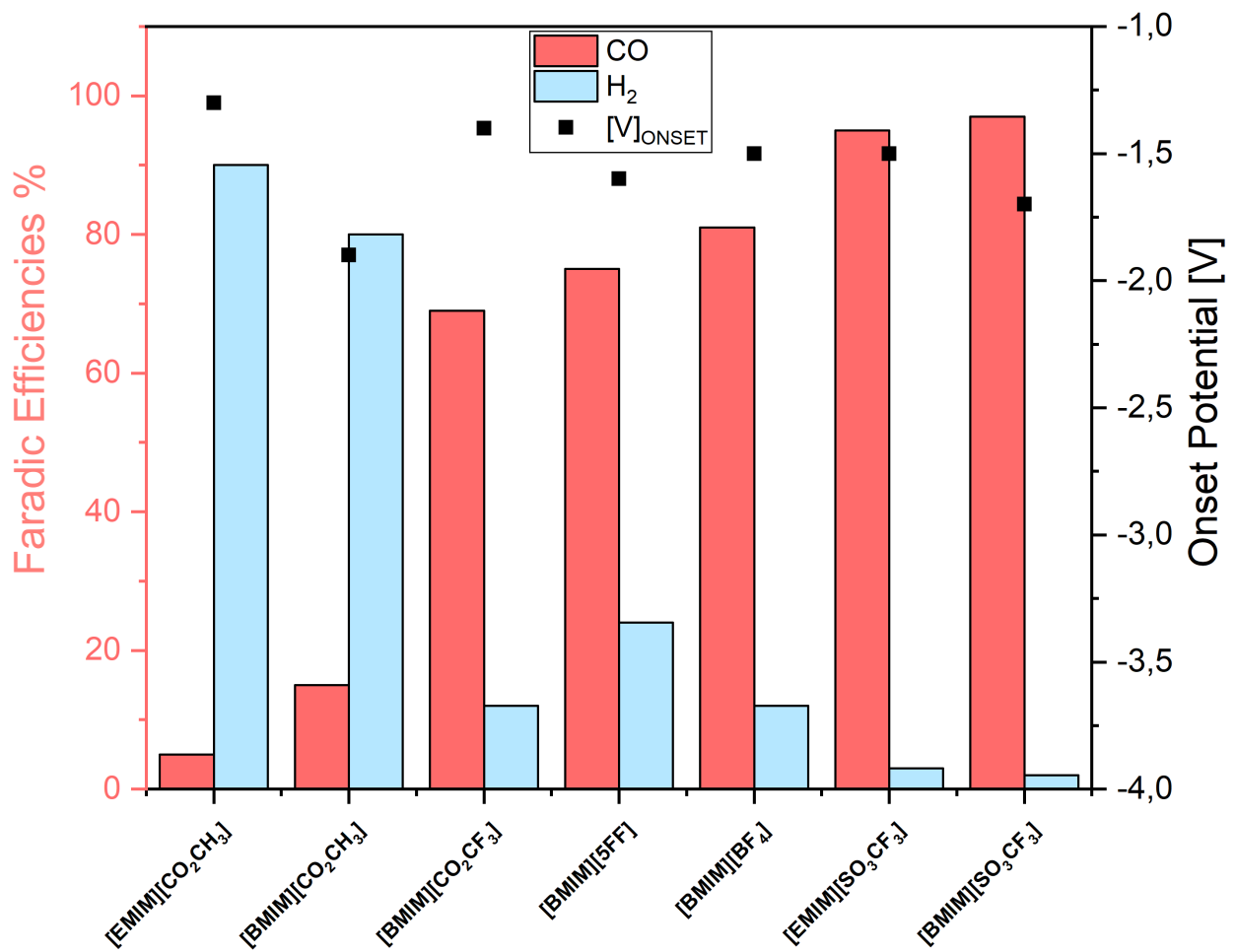


CP mean potential, [V] vs Ag/AgCl

- ✓ **CP's highlights:** comparing the CP's curves, for the same anion with EM with BMIM. Probably this trend is due to two aspects:
 - 1- A more convenient orientation reached by a shorter cation alkyl chain on t
 - 2- It can be linked to the conductivity of the catholyte. Catholyte conductivity [EMIM][SO₃CF₃] solutions are higher than [BMIM][CO₂CH₃] and [BMIM][S

Results and discussion

Chronopotentiometry (CP) in CO₂, t=120 min, -20 mA



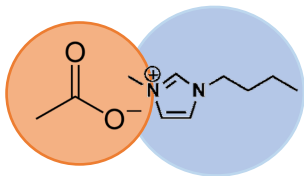
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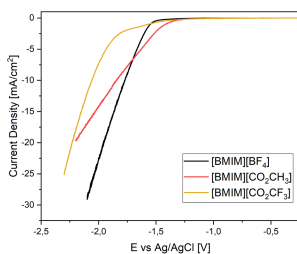
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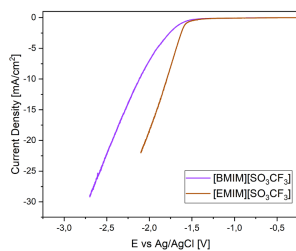
Conclusions



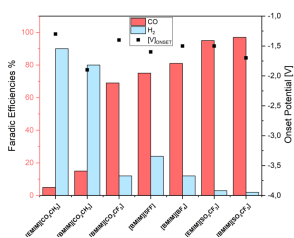
- ❖ Seven imidazolium salts were tested for the electroreduction of CO_2 .



- ❖ CO_2 solubility depends on the anion of the imidazolium salt, being higher for fluorinated anions.



- ❖ The cation has a steric effect and an orientation effect. As the steric bulkiness decreases, the imidazolium ring finds a more coordinated site on the electrode surface.



- ❖ Imidazolium salts of acetate are more selective for CO_2 reduction. [BMIM][SO₃CF₃] promotes the reduction of CO_2 to CO, while the most used [BMIM][BF₄] promotes the reduction of CO_2 to H₂.

Acknowledgements

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