# Preparation of Polyfunctionally Substituted Pyridine-2(1H) thione Derivatives as Precursors to Bicycles and Polycycles 

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#### Abstract

Reaction of acetylacetone with 1 mole of dimethylformamide dimethyl acetal (DMFDMA) affords enamine $2 a$ which react with cyanothioacetamide to give pyridinethione 3 a. Pyridinethione 3 a reacts with methyl iodide, halogenated compounds, aromatic aldehyde and malononitrile / elemental sulfur to yiled compounds 7-10 respectively. Reactions of thioether (7) in ethanolic $\mathrm{K}_{2} \mathrm{CO}_{3}, 1$ mole DMFDMA and 4-(dimethylamino) benzaldehyde gave compounds (11, 13, 14) respectively. Enaminone (12) can be prepared by reaction of compound (11) with DMFDMA. Also, compounds (13) and (14) can be obtained by reaction of compounds (3a) and (9a) with 2 mole of DMFDMA and methyl iodide respectively. Reactions of enaminone (9) with hydrazine hydrate and cyanothioacetamide gave compounds (11) and (12) respectively. Malononitrile dimmer when reacted with chalcones (9a), (14) and enaminone (12), (13) gave bipyridyl (15), (16) and (19a,b) respectively. Bipyridyl (15) can be converted to (16) by methylation using methyl iodide. Reactions of thienopyridine (8) with DMFDMA and sodium nitrite in acetic acid gave tricyclic compounds (20) and (21) respectively. Finally, reactions of (21a) with malononitrile / elemental sulfur and DMFDMA gave compounds (22) and (23) respectively.


Keywords: Acetyl acetone; DMFDMA; Malononitrile dimmer; Bipyridyl; 5-Acetylpyridinethione.

## Introduction

Formamide acetals are useful reagents in organic synthesis; [1,2] their main application has been used for functional group transformations [3], but they may also be regarded as one-carbon synthons in the construction of carbon skeletons. One type of reaction, which is potentially valuable for the future purpose, is the reaction of $N, N^{\prime}$ - dimethylformamide dimethyl acetal (DMFDMA) with 1,3-dicarbonyl compounds $\mathbf{1}$ to give enamines $2[2,4]$.

We have reported that enamines 2 were used as precursors in the synthesis of pentasubstituted pyridines 3-6 [5-8].


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\mathrm{X}=\mathrm{O}, \mathrm{~S}, \mathrm{C}(\mathrm{CN})_{2}
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## Results and Discussion

In conjunction of this work we report here the reaction of acetylacetone 1a with one mole of $N, N^{\prime}-$ dimetylformamide dimethyl acetal (DMFDMA) in dry dioxane gave the corresponding enamine 2a. Treatment of the lasted compound with cyanothioacetamide in ethanol in the presence of sodium ethoxide gave 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a [7], scheme 1.

We have found that the prepared compound 3a contains three functional groups which are thioamido group, nitrile group and acetyl group. These functional groups can be used for the preparation of bicyclic or polycyclic compounds of biological interest. Thus, some illustrative reactions designed to demonstrate the potential usefulness of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a for further heterocyclic synthesis. Therefore the reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a with methyl iodide in alcoholic sodium hydroxide afforded the corresponding thioether derivative 7, which in turn is a good intermediate for the preparation of further heterocyclic compounds of biological interest. The structure of the isolated compound 7 is conformed by spectral analysis. The IR spectrum shows the disappearance of (NH) group. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum shows the disappearance of the thioamide proton and the appearance of a singlet signal corresponding to $\left(\mathrm{SCH}_{3}\right)$ at $\delta_{\mathrm{H}}=2.63 \mathrm{ppm}$. Also, the mass spectrum shows the molecular ion peak at $\mathrm{m} / \mathrm{e}$ 206. The reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a with ethyl chloroacetate or chloroacetamides in ethanolic sodium ethoxide afforded the corresponding 5-acetyl-3-amino-6-methylthieno[2,3-b]pyridine derivatives 8a-c in a good yield. The structure of the isolated compounds is conformed by elemental and spectral analysis. The IR spectrum shows the disappearance of cyano group and appearance of amino group at $v_{\max }=3427 \mathrm{~cm}^{-1}$ in compound $\mathbf{8 a}$ as example beside the other functional groups. Also, the mass spectra show the molecular ion peaks fit to all compounds $\mathbf{8 a}-\mathbf{c}$. Also, the ${ }^{1} \mathrm{H}$ NMR spectra show signals fit to the structure of all compounds 8a-c. The presence of acetyl group in 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a is useful for the preparation of fused heterocyclic compounds. So that the reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a with aldehydes like 4-(dimethylamino)benzaldehyde and 4methylbenzaldehyde in ethanolic sodium hydroxide afforded the corresponding chalcones $\mathbf{9 a}, \mathbf{b}$.

The structure of the isolated chalcones is conformed by elemental analysis as well as spectral analysis. The mass spectra show the molecular ion peak fit to all compounds $\mathbf{9 a}, \mathbf{b}$. As an example compound $\mathbf{9 a}$ shows the molecular ion peak at m/e 323 which corresponding to the molecular formula $\left(\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{OS}\right)$.

Also, the ${ }^{1} \mathrm{H}$ NMR spectra of these compounds $\mathbf{9 a , b}$ show the disappearance of the signal corresponding to the methyl of acetyl group and the appearance of two doublets signals corresponding to the two proton of double bond of chalcone. Finally, 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a was treated with malononitrile and sulfur element (Gewald's reaction) in ethanol in the presence of triethylamine as a base to afford 5-(5-amino-4-cyanothiophen-3-yl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile $\mathbf{1 0}$ in a good yield, scheme 1 . The IR spectrum of compound $\mathbf{1 0}$ shows the appearance of amino group at $v_{\max }=3435 \mathrm{~cm}^{-1}$ beside the other functional groups. Also, ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 0}$ shows singlet signal at $\delta_{H}=2.45 \mathrm{ppm}$ corresponding to methyl group and singlet signal at $\delta_{\mathrm{H}}=6.95 \mathrm{ppm}$ corresponding to amino group and singlet signal at $\delta_{\mathrm{H}}=7.07 \mathrm{ppm}$ corresponding to CH thiophene ring and singlet signal at $\delta_{\mathrm{H}}=7.2 \mathrm{ppm}$ corresponding to CH pyridine ring.

5-Acetyl-6-methyl-2-(methylthio)nicotinonitrile 7 can be used as intermediate for further preparation of heterocyclic compounds.

So that compound 7 was treated with potassium carbonate in ethanol to afford 5-acetyl-2-ethoxy-6methylnicotinonitrile 11. This compound was formed by nucleophilic substitution of SMe by OEt group.

The structure of the isolated compound is conformed by elemental and spectral analyses. The mass spectrum shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 204$ corresponding to the molecular formula $\left(\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum shows the disappearance of SMe signal and appearance of two signals; a triplet at $\delta_{\mathrm{H}}=1.43 \mathrm{ppm}$ and a quartet at $\delta_{\mathrm{H}}=4.54 \mathrm{ppm}$ corresponding to the OEt moiety, in addition to the rest of signals corresponding to the other protons in the molecule. Compound $\mathbf{1 1}$ was reacted with $N, N^{\prime}$-dimetylformamide dimethyl acetal (DMFDMA) in dry xylene to give the corresponding enamine $\mathbf{1 2}$ in a good yield. The mass spectrum of compound $\mathbf{1 1}$ shows the molecular ion peak at m/e 259 which corresponding to the molecular formula $\left(\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 2}$ shows the disappearance of the singlet signal which is related to the methyl of acetyl group and the appearance of two singlet signals at $\delta_{\mathrm{H}}=2.68$ and 3.04 ppm corresponding to the two methyl groups of $\mathrm{NMe}_{2}$ moiety. Consequently the ${ }^{1} \mathrm{H}$ NMR spectrum shows the appearance of two doublets at $\delta_{\mathrm{H}}=6.25 \mathrm{ppm}$ and 7.87 ppm corresponding to the two protons of the enamine double bond.

Enamine 13 can be prepared in a good yield by reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a with two moles of $N, N^{\prime}$-dimetylformamide dimethylacetal (DMFDMA) in dry xylene or by reaction of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile 7 with one mole of $N, N^{\prime}$-dimetylformamide dimethylacetal (DMFDMA) in dry xylene. The structure of the isolated compound is conformed by elemental and spectral analysis. Where as the mass spectrum shows the molecular ion peak at m/e 261 which corresponding to the molecular formula $\left(\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of it shows the disappearance of the singlet signal which is related to the methyl of acetyl group and appearance of two singlet signals at $\delta_{H}=2.62$ and 2.64 ppm corresponding to the two methyl groups of $\mathrm{NMe}_{2}$ moiety. Consequently the ${ }^{1} \mathrm{H}$ NMR spectrum shows the appearance of two doublets at $\delta_{\mathrm{H}}=5.28 \mathrm{ppm}$ and 7.75 ppm corresponding to the two protons of double bond of enamine.

Chalcone 14 can be prepared by either the reaction of compound 7 with (4(dimethylamino)benzaldehyde) in ethanolic sodium hydroxide or by treatment of compound $9 \mathbf{a}$ with methyl iodide in ethanolic sodium hydroxide. The mass spectrum of compound $\mathbf{1 4}$ shows the molecular ion peak at m/e 337 corresponding to the molecular formula $\left(\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{OS}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 4}$ shows singlet signal at $\delta_{\mathrm{H}}=2.62 \mathrm{ppm}$ corresponding to methyl group and singlet signal at $\delta_{\mathrm{H}}=2.66 \mathrm{ppm}$ corresponding to $\mathrm{SCH}_{3}$ and two singlet signal at $\delta_{\mathrm{H}}=2.9,3.04 \mathrm{ppm}$ corresponding to $\mathrm{NMe}_{2}$ moiety and appearance of some signals of other protons in molecule. For preparation of bipyridyl derivatives, we have carried out the reaction of chalcones 5-(3-(4(dimethylamino) phenyl) acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-arbonitrile 9a and 5-(3-(4(dimethylamino) phenyl) acryloyl)-6-methyl-2-(methylthio)nicotinonitrile $\mathbf{1 4}$ with malononitrile dimmer [9] in acetic acid and ammonium acetate afforded the corresponding bipyridyl derivatives

6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile 15 and 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-(methylthio)-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile 16 respectively. The reaction proceeds by Michael addition followed by cyclization through condensation as shown in scheme 2 . The compound 16 can also be obtained by the reaction of 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile 15 with methyl iodide in alcoholic sodium hydroxide scheme 2. The structure of the isolated compounds $\mathbf{1 5}$ and $\mathbf{1 6}$ is conformed by elemental and spectral analysis. Where as the mass spectra of these compounds show the molecular ion peaks at m/e 435 corresponding to the molecular formula $\left(\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{7} \mathrm{~S}\right)$, and at m/e 449 corresponding to the molecular formula $\left(\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{7} \mathrm{~S}\right)$ for 15 and $\mathbf{1 6}$ respectively. The IR spectra of both compounds 15 and 2.14 show the disappearance of the carbonyl group and the appearance of NH group. Also, the ${ }^{1} \mathrm{H}$ NMR spectra of these compounds show signals fit to structures $\mathbf{1 5}$ and $\mathbf{1 6}$.

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For further preparation of heterocyclic compounds [10] we carried out the following reactions. The reaction of enamine $\mathbf{1 3}$ with excess hydrazine hydrate in ethanol afforded 6-methyl-5-(1H-pyrazol-3-yl)-1H-pyrazolo[3,4-b] pyridin-3-amine 17 in a good yield as shown in scheme 3.

The IR spectrum of compound $\mathbf{1 7}$ shows the disappearance of the cyano group and the appearance of $\mathrm{NH}_{2}$ and NH groups at $\mathrm{v}_{\max }$ at $3405 \mathrm{~cm}^{-1}, 3329 \mathrm{~cm}^{-1}$ and $3136 \mathrm{~cm}^{-1}$ respectively. Also, the mass spectrum of compound $\mathbf{1 7}$ shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 214$ corresponding to the molecular formula $\left(\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{6}\right)$.

Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 7}$ shows signals fit to the structure. Also the enamine $\mathbf{1 3}$ is treated with cyanothioacetamide in acetic acid and ammonium acetate afforded 2'-methyl-6'-(methylthio)-6-thioxo-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile 18. The reaction is started by Micheal addition of cyanothioacatamide on the double bond followed by elimination of dimethylamine ( $\mathrm{HNMe}_{2}$ ) and cyclization with the carbonyl group.

The structure of the isolated compound $\mathbf{1 8}$ is conformed by elemental and spectral analysis. The IR spectrum of compound $\mathbf{1 8}$ shows the disappearance of carbonyl group and appearance of NH group at $v_{\max }$ at $3428 \mathrm{~cm}^{-1}$. The mass spectrum of compound $\mathbf{1 8}$ shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 298$ corresponding to the molecular formula $\left(\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{~S}_{2}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 8}$ shows the disappearance of protons of $\mathrm{NMe}_{2}$ moiety and appearance of NH proton beside the other protons.

Another type of bipyridyl derivatives $\mathbf{1 9 a , b}$ can be prepared by the reaction of the enamines $\mathbf{1 2}$ and $\mathbf{1 3}$ with malononitrile dimmer in acetic acid and ammonium acetate. This reaction proceeds by Michael addition of malononitrile dimmer, followed by elimination of dimethylamine ( $\mathrm{HNMe}_{2}$ ) and cyclization through condensation of amino group with carbonyl group as shown in scheme 3.

The mass spectrum of compound 19a shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 328$ corresponding to the molecular formula $\left(\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{6} \mathrm{O}\right)$, and compound 19b shows the molecular ion peak at m/e 330 corresponding to the molecular formula $\left(\mathrm{C}_{17} \mathrm{H}_{10} \mathrm{~N}_{6} \mathrm{~S}\right)$.

The IR spectra of the compounds $\mathbf{1 9} \mathbf{a}, \mathbf{b}$ show the disappearance of the carbonyl group and the appearance of NH group beside the other groups. Also, the ${ }^{1} \mathrm{H}$ NMR spectra of compounds 19a,b show the disappearance of protons of $\mathrm{NMe}_{2}$ moiety and appearance of NH proton beside the other protons.

The tricyclic heterocyclic compounds are biologically interest compounds. They are examples of uncommon ring system [11,12]. Therefore we are interested for the preparation of this type of heterocyclic compound. Thus 5-acetyl-3-amino-6-methyl-N-(p-tolyl)benzo[b]thiophene-2-carboxamide $\mathbf{8 b}$ is reacted with $N, N^{\prime}$-dimetylformamide dimethyl acetal (DMFDMA) in dry dioxane afforded 8-acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d]pyrimidin-4( $3 H$ )-one $\mathbf{2 0}$. The $\mathbb{R}$ spectrum of compound $\mathbf{2 0}$ shows the disappearance of $\left(\mathrm{NH}_{2}\right)$ and $(\mathrm{NH})$ groups. The mass spectrum of compound $\mathbf{2 0}$ shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 349$ which corresponding to the molecular formula $\left(\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 20 shows the appearance of two singlet signals at $\delta_{\mathrm{H}}=8.43 \mathrm{ppm}$, and 8.52 ppm corresponding to two protons of pyrimidinone and pyridine rings respectively beside other signals for other protons. For further reaction of 5-acetyl-3-amino-6-methyl-N-substituted[b]thiophene-2-carboxamide 8b,c it reacted with nitrous acid in acetic acid, under cooling, afforded the tricyclic compounds 21a,b in a good yield as shown in scheme 4. The structures of the compounds 21a,b are conformed by elemental and spectral analysis. Where as the IR spectrum of both compounds 21a,b show the disappearance of the bands corresponding to $\left(\mathrm{NH}_{2}\right)$ and (NH) groups. The mass spectrum of the compound 21a as an example shows the molecular ion peak at m/e 350 corresponding to molecular formula $\left(\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}\right)$.

Also, the ${ }^{1} \mathrm{HNMR}$ spectra of compounds 21a,b shows the disappearance of the signals which corresponding to $\left(\mathrm{NH}_{2}\right)$ and $(\mathrm{NH})$ groups beside the appearance the other signals for other groups. We have found that the prepared tricyclic compounds 20 and 21a,b contain acetyl group which is very important for the preparation of new heterocyclic compounds. So that the reaction of 21a with malononitrile and sulphur element in ethanol and triethylamine (Geweld reaction) afforded 2-amino-4-(7-methyl-4-oxo-3-(p-tolyl)-3,4-dihydropyrido[3', 2':4,5]thieno[3,2-d][1,2,3]triazin-8-yl)thiophene-3carbonitrile 22. The IR spectrum of compound 22 shows the disappearance of the carbonyl group of acetyl moiety and the appearance of amino and cyano groups at $v_{\text {max }}$ at $3427 \mathrm{~cm}^{-1}$ and $2208 \mathrm{~cm}^{-1}$ respectively. Also, the mass spectrum of this compound $\mathbf{2 2}$ shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 430$ which corresponding to the molecular formula $\left(\mathrm{C}_{21} \mathrm{H}_{14} \mathrm{~N}_{6} \mathrm{OS}_{2}\right)$. Also, the compound 21a is treated with $N, N^{\prime}$-dimetylformamide dimethyl acetal (DMFDMA) in dry xylene afforded the corresponding enamine 8-(3-(dimethylamino)acryloyl)-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)one $\mathbf{2 3}$ in a good yield, scheme 4 . The mass spectrum of compound $\mathbf{2 3}$ shows the molecular ion peak at $\mathrm{m} / \mathrm{e} 405$ corresponding to molecular formula $\left(\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}\right)$. Also, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{2 2}$ shows the disappearance of the methyl of acetyl moiety and appearance instead of it two singlet signals at $\delta_{H}=3.63 \mathrm{ppm}$ and 3.67 ppm corresponding to $\left(\mathrm{NMe}_{2}\right)$ moiety. Also, it shows the appearance of two doublet signals at $\delta_{\mathrm{H}}=5.42 \mathrm{ppm}$ and 7.82 ppm respectively corresponding to the double bond protons of enaminone moiety beside signals for other protons.

## Experimental

All melting points are uncorrected. IR spectra were recorded on a Perkin-Elmer 17100 FTIR spectrometer as KBr disks. NMR spectra were recorded on Bruker AC300 spectrometer at 400 MHz for solutions in $\mathrm{CDCl}_{3}$ or DMSO with tetramethylsilane (TMS) as an internal standard unless otherwise recorded at Department of Chemistry, College of Science, Sultan Qaboos University, P.O. Box 36, AlKhod23, Oman. Mass spectra were obtained on Finnigan 4500 (low resolution) spectrometers using electron impact (EI) at Micro-analytical Center Cairo University Giza Egypt.

## Preparation of 5-acetyl-6-methyl-2-(methylthio) nicotinonitrile

Mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile $3 \mathrm{a}(1.92 \mathrm{~g}, 10 \mathrm{mmol}$ ) in ethanol as solvent and sodium hydroxide ( $0.4 \mathrm{~g}, 10 \mathrm{mmol}$ ) with stirring for 1 hr ., and add methyl iodide $(0.63 \mathrm{ml}, 10 \mathrm{mmol})$ with stirring until precipitate formed. The product was recovered by filtration and recrystallised from ethanol as white crystals (1.52g, 74\%), Mp. 140-142 oC; 1H-NMR (CDCl3): $\delta=$ $2.54(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}),. 2.63(3 \mathrm{H}, \mathrm{s}, \mathrm{SCH} 3), 2.77(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}), 8.07(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v 2227 (CN), $1685 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $206 \mathrm{M}+$; Anal. Calcd for C10H10N2OS (206.27): C, 58.23; H, 4.89; N, 13.58. Found: C, 58.03; H, 4.73; N, 13.41.

## General procedure for the preparation of compounds 8a-c

In dry flask a mixture 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile 3a (1.92g, 10 mmol ) and $\alpha$-chloro compounds ( 10 mmol ) in ethanol and sodium ethoxide ( 20 mmol ) was left under reflux for two hours. The mixture was left for cooling and poured onto ice cold water. The solid product was recovered by filtration and recrystallised from the proper solvent.

Ethyl 5-acetyl-3-amino-6-methylthieno[2,3-b]pyridine-2-carboxylate (8a): Obtained using ethyl 2chloroacetate $(1.06 \mathrm{ml}, 10 \mathrm{mmol})$. The product was recrystallised from acetic acid as yellow crystals ( $2.16 \mathrm{~g}, 77.7 \%$ ), Mp. 220-222 oC; 1H-NMR (DMSO): $\delta=1.25$ ( $3 \mathrm{H}, \mathrm{t}, \mathrm{CH} 3$ ethyl), 4.25 ( $2 \mathrm{H}, \mathrm{q}, \mathrm{CH} 2$ ethyl), 2.6 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}$ ), 2.66 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}$ ), 7.29 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{NH} 2$ ), 8.95 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v 3427, 3328 (NH2), $1679 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $278 \mathrm{M}+$; Anal. Calcd for C13H14N2O3S (278.33): C, 56.10; H, 5.07; N, 10.06, Found: C, 55.96; H, 4.94; N, 9.97.

5-Acetyl-3-amino-6-methyl-N-(p-tolyl)thieno[2,3-b]pyridine-2-carboxamiden(8b): Obtained using 2-chloro-N-(p-tolyl)acetamide ( $1.83 \mathrm{~g}, 10 \mathrm{mmol}$ ). The product was recrystallised from ethanol as yellow crystals ( $2.7 \mathrm{~g}, 79 \%$ ), Mp. 218-220 oC; 1H-NMR (DMSO) $\delta=2.26$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{Ar}$ ), 2.64 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3$ py.), $2.73(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}), 7.12(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.55(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}) 7.47(2 \mathrm{H}, \mathrm{s}, \mathrm{NH} 2), 9.04(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ ру.), $9.4(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$; IR (KBr) v 3428,3312 (NH2, NH), $1685 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $339 \mathrm{M}+$; Anal. Calcd for C18H17N3O2S (339.42): C, 63.70; H, 5.05; N, 12.38, Found: C, 63.56; H, 4.93; N, 12.15.

## 5-Acetyl-3-amino-N-(4-methoxyphenyl)-6-methylthieno[2,3-b]pyridine-2-carboxamide(8c):

Obtained using 2 -chloro-N-(4-methoxyphenyl)acetamide $(1.99 \mathrm{~g}, 10 \mathrm{mmol})$. The product was recrystallised from ethanol as yellow crystals ( $2.8 \mathrm{~g}, 79 \%$ ), Mp. 240-242 oC; 1H-NMR (DMSO) $\delta=$ 2.65 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}$ ), 2.73 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}$ ), 3.76 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{O}$ ), 6.9 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar)}$,7.56 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar)} 7.45$

## General procedure for the preparation of compounds 9a,b

A mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile $3 \mathrm{a}(1.92 \mathrm{~g}, 10 \mathrm{mmol}$ ) in ethanol as solvent in presence of sodium hydroxide ( $0.4 \mathrm{~g}, 10 \mathrm{mmol}$ ) with aromatic aldehydes ( 10 mmol ) with stirring for 2 hr . then poured onto ice, cold water and acidified with conc. Hydrochloric acid until the precipitate was formed. The solid product was recovered by filtration and recrystallised from ethanol.

## 5-(3-(4-(Dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile

 (9a):Obtained using 4-(dimethylamino)benzaldehyde ( $1.49 \mathrm{~g}, 10 \mathrm{mmol}$ ). Mp. 140-142 oC as yellow crystals ( $2.45 \mathrm{~g}, 76 \%$ ); 1H-NMR (CDCl3) $\delta=2.65(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3), 2.79(6 \mathrm{H}, \mathrm{s}, \mathrm{NMe} 2), 6.70(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.74$ $(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.06(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ chalcone), $7.85(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ chalcone), $8.07(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $13.2(1 \mathrm{H}, \mathrm{br}$. , NH); IR (KBr) v 3437 (NH), 2225 (CN), $1685 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $323 \mathrm{M}+$; Anal. Calcd for C18H17N3OS (323.42): C, 66.85; H, 5.30; N, 12.99, Found: C, 65.4; H, 5.17; N, 12.86.

6-Methyl-2-thioxo-5-(3-(p-tolyl)acryloyl)-1,2-dihydropyridine-3-carbonitrile (9b):
Obtained using 4-methylbenzaldehyde ( $1.2 \mathrm{~g}, 10 \mathrm{mmol}$ ). Mp. $=240-242$ oC as yellow crystals $(2.2 \mathrm{~g}$, $74.8 \%$ ); 1H-NMR (DMSO) $\delta=2.31$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{Ar}$ ), 2.58 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3$ ру.), 7.24 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}$ ), 7.69 $(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.49(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ chalcone $), 7.6(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ chalcone $), 8.58(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.) $13(1 \mathrm{H}, \mathrm{br}$., NH); IR (KBr) v 3434 (NH), 2231 (CN), $1659 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $294 \mathrm{M}+$; Anal. Calcd for C17H14N2OS (294.38): C, 69.36; H, 4.79; N, 9.52, Found: C, 69.19; H, 4.81; N, 9.45.

## 5-(5-Amino-4-cyanothiophen-3-yl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile (10)

In dry flask a mixture 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile $3 \mathrm{a}(1.92 \mathrm{~g}, 10$ $\mathrm{mmol})$, malononitrile $(0.66 \mathrm{~g}, 10 \mathrm{mmol})$ and sulfur $(0.32 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol and few drops of triethylamine as base was left under reflux for three hours. The mixture was left for cooling then poured onto ice cold water. The product obtained was recrystallised from a mixture of ethanol/DMF (3:1) as brown crystals $(1.9 \mathrm{~g}, 69.8 \%), \mathrm{Mp} .>300 \mathrm{oC} ; 1 \mathrm{H}-\mathrm{NMR}(\mathrm{DMSO}) \delta=2.45(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3), 6.95(2 \mathrm{H}, \mathrm{s}$, NH2), 7.07 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ thiophene), 7.2 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v 3435, 3350 (NH2), $3250(\mathrm{NH})$, $2210 \mathrm{~cm}-1$ (CN); Anal. Calcd for C12H8N4S2 (272.35): C, 52.92; H, 2.96; N, 20.57, Found: C, 52.85; H, 2.92; N, 20.15.

## 5-Acetyl-2-ethoxy-6-methylnicotinonitrile (11)

In dry flask a mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile $7(2.06 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol and potassium carbonate was left under reflux for 3 hr . after cooling the mixture was poured onto ice cold water. The product was recovered and recrystallised from $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(1: 1)$ as yellowish crystals ( $1.6 \mathrm{~g}, 78 \%$ ), Mp. $78-80{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta=1.43\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right.$ ethyl), $4.54\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right.$ ethyl), $2.66\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.04\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{CO}\right), 7.85\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}\right.$ py.); IR (KBr) v $2228(\mathrm{CN}), 1688 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O})$; MS (EI) ${ }^{+}: \mathrm{m} / \mathrm{z} 204 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ (204.23): C, 64.69; H, 5.92; N, 13.72, Found: C, 64.51; H, 5.83; N, 1

## (E)-5-(3-(Dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile (12)

In dry flask a mixture of 5-acetyl-2-ethoxy-6-methylnicotinonitrile $11(2.04 \mathrm{~g}, 10 \mathrm{mmol})$ in dry xylene as solvent and $N, N$-dimethylformamide dimethyl acetal (DMFDMA) ( $1.32 \mathrm{ml}, 10 \mathrm{mmol}$ ) was left under reflux for 2 hr ., cool and the solvent was evaporated. The product was recovered and recrystallised from $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(1: 1)$ as yellow crystals ( $1.9 \mathrm{~g}, 73.3 \%$ ), Mp. $68-70{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ $\delta=1.3\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right.$ ethyl), $4.58\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right.$ ethyl), $2.62\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.68,3.04\left(6 \mathrm{H}, 2 \mathrm{~s}, \mathrm{NMe}_{2}\right), 6.25$ $(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 7.87(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 8.2\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}\right.$ py.); IR (KBr) v $2230(\mathrm{CN}), 1684 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; MS (EI) ${ }^{+}$: $m / z 259 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2}$ (259.31): C, 64.85; H, 6.61; N, 16.20, Found: C, 64.56; H, 6.47; N, 16.11.

## (E)-5-(3-(Dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile (13)

(A) In dry flask a mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile $7(2.06 \mathrm{~g}, 10 \mathrm{mmol})$ in dry xylene as solvent and $N, N^{\prime}$-dimethylformamide dimethyl acetal (DMFDMA) ( $1.32 \mathrm{ml}, 10 \mathrm{mmol}$ ) was left under reflux for 2 hr ., cool and poured in dry backer and the solvent was evaporated. The product was recovered and recrystallised from $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ (1:1) as yellow crystals ( $2 \mathrm{~g}, 76.6 \%$ ), Mp. 100-102 ${ }^{\circ} \mathrm{C}$; (B) In dry flask a mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3carbonitrile 3a ( $1.92 \mathrm{~g}, 10 \mathrm{mmol}$ ) in dry xylene as solvent and $N, N$-dimethylformamide dimethyl acetal (DMFDMA) ( $2.64 \mathrm{ml}, 20 \mathrm{mmol}$ ) was left under reflux for 2 hr ., cool and poured in dry backer and the solvent was evaporated. The product was recovered and recrystallised from $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(1: 1)$ as yellow crystals ( $2.1 \mathrm{~g}, 80.4 \%$ ), Mp. and mixed Mp. 100-102 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta=2.62,2.64(6 \mathrm{H}$, $\left.2 \mathrm{~s}, \mathrm{NMe}_{2}\right), 2.9\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right.$ py.), $3.15\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SCH}_{3}\right), 5.28(1 \mathrm{H}$, d, trans CH$), 7.75(1 \mathrm{H}$, d, trans CH$)$, $6.28(1 \mathrm{H}, \mathrm{d}$, cis CH$), 10.15(1 \mathrm{H}, \mathrm{d}$, cis CH$), 8.07\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}\right.$ py.); IR (KBr) v $2227(\mathrm{CN}), 1685 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O})$; MS (EI) ${ }^{+}: m / z 261 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}$ (261.35): C, 59.74; H, 5.79; N, 16.08, Found: C, 59.63; H, 5.45; N, 15.8.

## (E)-5-(3-(4-(Dimethylamino)phenyl)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile (14)

(A) Mixture of 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3carbonitrile $9 \mathrm{a}(3.23 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol as solvent and sodium hydroxide $(0.4 \mathrm{~g}, 10 \mathrm{mmol})$ with stirring for 1 hr ., and add methyl iodide ( 10 mmol ) with stirring until precipitate was formed. The product was recovered by filtration and was purified by recrystallised from ethanol as yellow crystals ( $2.5 \mathrm{~g}, 74 \%$ ), Mp. 160-162 oC; (B) mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile 7 ( 2.06 g , 10 mmol ) in ethanol as solvent in presence of sodium hydroxide ( $0.4 \mathrm{~g}, 10 \mathrm{mmol}$ ) with 4(dimethylamino)benzaldehyde ( $1.49 \mathrm{~g}, 10 \mathrm{mmol}$ ) with stirring for 2 hr ., until precipitate formed and dilute with water. The product was recovered by filtration and purified by recrystallised from ethanol as yellow crystals ( $2.4 \mathrm{~g}, 71 \%$ ), Mp. and mixed Mp. 160-162 oC; 1H-NMR $(\mathrm{CDCl} 3) \delta=2.62(3 \mathrm{H}, \mathrm{s}$, CH3), $2.66(3 \mathrm{H}, \mathrm{s}, \mathrm{SCH} 3), 2.9,3.04(6 \mathrm{H}, 2 \mathrm{~s}, \mathrm{NMe} 2), 6.83(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.46(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 6.67(1 \mathrm{H}, \mathrm{d}$, $\mathrm{CH}), 7.38(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 7.85(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v $2217(\mathrm{CN}), 1648 \mathrm{~cm}-1(\mathrm{C}=\mathrm{O})$; MS (EI)+: m/z 337 M+; Anal. Calcd for C19H19N3OS (337.45): C, 67.63; H, 5.68; N, 12.45, Found: C, 67.49; H, 5.62; N, 12.48.

## 6-(Dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (15)

In dry flask a mixture 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile $9 \mathrm{a}(3.23 \mathrm{~g}, 10 \mathrm{mmol})$ and malononitrile dimmer ( $1.32 \mathrm{~g}, 10 \mathrm{mmol}$ ) in acetic acid and presence of ammonium acetate was left under reflux for three hours. The mixture was left for cooling and poured onto ice, cold water. The product was recovered by filtration and recrystallisation from ethanol as brown crystals ( $3.25 \mathrm{~g}, 74.7 \%$ ), Mp. 260-262 oC; 1H-NMR (DMSO) $\delta$ $=2.38(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3), 3.06(6 \mathrm{H}, \mathrm{s}, \mathrm{NMe} 2), 7.5(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $8.21(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), 6.83 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}$ ), $7.93(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 11.93(1 \mathrm{H}, \mathrm{br}, \mathrm{NH}), 12.4(1 \mathrm{H}, \mathrm{br}, \mathrm{NH})$; IR (KBr) v 3334, $3207(2 \mathrm{NH}), 2206 \mathrm{~cm}-1$ (CN); MS (EI)+: m/z $435 \mathrm{M}+$; Anal. Calcd for C24H17N7S (435.51): C, 66.19; H, 3.93; N, 22.51, Found: C, 66.06; H, 3.78; N, 22.3.

## 6-(Dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-(methylthio)-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (16)

(A) In dry flask a mixture of 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2(methylthio)nicotinonitrile $14(3.37 \mathrm{~g}, 10 \mathrm{mmol})$ and malononitrile dimmer $(1.32 \mathrm{~g}, 10 \mathrm{mmol})$ in acetic acid acid and ammonium acetate was left under reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from acetic acid as brown crystals ( $3.4 \mathrm{~g}, 76 \%$ ), Mp. 220-222 ${ }^{\circ} \mathrm{C}$; (B) mixture of 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1',5,6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (15) ( $4.35 \mathrm{~g}, 10 \mathrm{mmol}$ ) in ethanol as solvent in presence of sodium hydroxide $(0.4 \mathrm{~g}, 10 \mathrm{mmol})$ and methyl iodide ( 10 mmol ) with stirring until precipitate formed. The product was recovered by filtration and recrystallised from acetic acid as brown crystals (3.2g, 71.5\%), Mp. and mixed Mp. 220-222 ${ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) $\delta=2.61\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.65(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{SCH}_{3}\right), 2.99,3.01\left(6 \mathrm{H}, 2 \mathrm{~s}, \mathrm{NMe}_{2}\right), 7.09(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $8.66(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $6.82(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.73$ $(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 10.3(1 \mathrm{H}, \mathrm{br}, \mathrm{NH})$; IR (KBr) v $3345(\mathrm{NH}), 2213 \mathrm{~cm}^{-1}(\mathrm{CN})$; MS (EI) ${ }^{+}: \mathrm{m} / \mathrm{z} 449 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{7} \mathrm{~S}$ (449.54): C, 66.80; H, 4.26; N, 21.81, Found: C, 66.69; H, 4.18; N, 21.6.

## 6-Methyl-5-(1H-pyrazol-3-yl)-1H-pyrazolo[3,4-b]pyridin-3-amine (17)

In flask a mixture of ( $E$ )-5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile 13 ( $2.61 \mathrm{~g}, 10 \mathrm{mmol}$ ) and excess of hydrazine hydrate was left reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from ethanol as yellowish crystals $(1.6 \mathrm{~g}, 75 \%)$, Mp. 260-262 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathbf{H}$-NMR (DMSO) $\delta=2.49\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 6.47(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ pyrazole), $8.3(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ pyrazole), 7.8 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $5.54\left(2 \mathrm{H}, \mathrm{s}, \mathrm{NH}_{2}\right), 11.75(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 12.91(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$; IR (KBr) v at 3405, 3329, $3136 \mathrm{~cm}^{-1}\left(\mathrm{NH}_{2}, \mathrm{NH}\right)$; MS (EI $)^{+}: \mathrm{m} / \mathrm{z} 214 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{6}$ (214.23): C, 56.07; H, 4.71; N, 39.23, Found: C, 55.85; H, 4.56; N, 39.16.

## 5-(5-cyano-1,6-dihydro-6-thioxopyridin-2-yl)-6-methyl-2-(methylthio)pyridine-3-carbonitrile (18)

In dry flask a mixture of ( $E$ )-5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile 13 ( $2.61 \mathrm{~g}, 10 \mathrm{mmol}$ ) and cyanothioacetamide ( $1 \mathrm{~g}, 10 \mathrm{mmol}$ ) in acetic acid and ammonium acetate was left under reflux for four hours. Cool and poured the mixture into ice cold water.

The product was recovered by filtration and recrystallised from ethanol as brown crystals ( 2.3 g , $77.1 \%)$, Mp. 170-172 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}(\mathrm{DMSO}) \delta=2.63\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SCH}_{3}\right), 7.7(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ py.), $8\left(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}\right.$ ру.), $8.14\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}\right.$ py.), $12.25(1 \mathrm{H}, \mathrm{br}, \mathrm{NH})$; IR (KBr) v $=3428(\mathrm{NH}), 2221 \mathrm{~cm}^{-1}$

## General procedure for the preparation of compounds 19a,b

In dry flask a mixture of (E)-5-(3-(dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile 12 ( $2.59 \mathrm{~g}, 10 \mathrm{mmol}$ ) or (E)-5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile 13 $(2.61 \mathrm{~g}, 10 \mathrm{mmol})$ and malononitrile dimmer $(1.32 \mathrm{~g}, 10 \mathrm{mmol})$ in acetic acid and ammonium acetate was heated under reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from ethanol

5-(5-Cyano-6-(dicyanomethylene)-1,6-dihydropyridin-2-yl)-2-ethoxy-6-methylpyridine-3-carbonitrile (19a): Obtained using ( $E$ )-5-(3-(dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile 12. Mp. $200-202{ }^{\circ} \mathrm{C}$ as brown crystals ( $2.4 \mathrm{~g}, 73.1 \%$ ); ${ }^{1} \mathrm{H}-\mathrm{NMR}(\mathrm{DMSO}) \delta=1.39\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right), 4.50(2 \mathrm{H}, \mathrm{q}$, $\left.\mathrm{CH}_{2}\right), 2.62\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 7.58(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ py.), $8.48(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ py.), $8.7(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py. ring $), 11.3(1 \mathrm{H}$, br, NH); IR (KBr) v $3330(\mathrm{NH}), 2218 \mathrm{~cm}^{-1}(\mathrm{CN})$; MS (EI) $)^{+}: \mathrm{m} / \mathrm{z} 328 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{6} \mathrm{O}$ (328.34): C, 65.85; H, 3.68; N, 25.60, Found: C, 65.71; H, 3.52; N, 25.43.

5-(5-Cyano-6-(dicyanomethylene)-1,6-dihydropyridin-2-yl)-2-(methylthio)-6-methylpyridine-3carbonitrile (19b): Obtained using (E)-5-(3-(dimethylamino)acryloyl)-6-methyl-2(methylthio)nicotinonitrile 13. Mp. $=190-192{ }^{\circ} \mathrm{C}$ as brown crystals (2.3g, 69.7\%); ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) $\delta$ $=2.58\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.64\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SCH}_{3}\right), 6.5(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ py.), $8.2(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}$ py.), $8.69(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.), $11.31(1 \mathrm{H}, \mathrm{br}, \mathrm{NH})$; IR (KBr) v $3340(\mathrm{NH}), 2212 \mathrm{~cm}^{-1}(\mathrm{CN})$; MS (EI) ${ }^{+}: \mathrm{m} / \mathrm{z} 330 \mathrm{M}^{+}$; Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{10} \mathrm{~N}_{6} \mathrm{~S}$ (330.37): C, 61.80; H, 3.05; N, 25.44, Found: C, 61.63; H, 2.89; N, 25.27.

## 8-Acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d]pyrimidin-4(3H)-one (20)

A mixture of 5-acetyl-3-amino-6-methyl-N-(p-tolyl)thieno[2,3-b]pyridine-2-carboxamide 8 b ( $3.39 \mathrm{~g}, 10$ $\mathrm{mmol})$ in dry dioxane and DMFDMA $(1.32 \mathrm{ml}, 10 \mathrm{mmol})$ with stirring for 12 hrs . The product was recovered by filtration and recrystallised from acetic acid as gray crystals ( $2.6 \mathrm{~g}, 74.5 \%$ ), Mp. 200.202 oC; 1H-NMR (DMSO) $\delta 2.26$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{Ar}$ ), 2.68 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}$. ), 2.69 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}$ ), 7.16 ( 2 H , d, Ar), $7.52(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 8.43(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ pyrimidinone), $8.52(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v at 1649,1691 cm-1 (2C=O); MS (EI)+: m/z $349 \mathrm{M}+$; Anal. Calcd for C19H15N3O2S (349.41): C, 65.31; H, 4.33; N, 12.03, Found: C, 65.19; H, 4.26; N, 11.95 .

## General procedure for the preparation of compounds 21a,b

A mixture of N -substituted-5-acetyl-3-amino-6-methylthieno[2,3-b]pyridine-2-carboxamide 8b,c (10 $\mathrm{mmol})$ in acetic acid and sodium nitrite $(1.38 \mathrm{~g}, 20 \mathrm{mmol})$ with stirring for 1 hr . the precipitate was formed and dilute with water. The product was recovered by filtration and recrystallised from ethanol.

8-Acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one (21a):
Obtained using 5-acetyl-3-amino-6-methyl-N-(p-tolyl)thieno[2,3-b]pyridine-2-carboxamide 8b (3.39g, 10 mmol ). Mp. 170-172 oC as gray crystals ( $3 \mathrm{~g}, 85.7 \%$ ); 1HNMR (DMSO) $\delta=2.4$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{Ar}$ ), 2.74 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}.), 2.77$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}$ ), $7.4(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}$ ), 7.54 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}$ ), 9.17 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v 1687, $1700 \mathrm{~cm}-1$ (2C=O); MS (EI)+: m/z $350 \mathrm{M}+$; Anal. Calcd for C18H14N4O2S (350.40): C, 61.70; H, 4.03; N, 15.99, Found: C, 61.56; H, 3.94; N, 15.78.

## 8-Acetyl-3-(4-methoxypheny))-7-methylpyrido[3', $\left.\mathbf{2}^{\prime}: 4,5\right]$ thieno $[3,2-\mathrm{d}][1,2,3]$ triazin-4(3H)-one

(21b): Obtained using 5-acetyl-3-amino-N-(4-methoxyphenyl)-6-methylthieno[2,3-b]pyridine-2carboxamide $8 \mathrm{c}(3.55 \mathrm{~g}, 10 \mathrm{mmol}) . \mathrm{Mp} .=220-222 \mathrm{oC}$ as gray crystals $(2.9 \mathrm{~g}, 79.4 \%) ; 1 \mathrm{H}-\mathrm{NMR}$ (DMSO) $\delta=2.74(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}),. 2.81(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{CO}), 3.85(3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{O}), 7.15(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.61$ ( $2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}$ ), 9.26 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v $1687 \mathrm{~cm}-1$ (C=O); Anal. Calcd for C18H14N4O3S (366.40): C, 59.01; H, 3.85; N, 15.29, Found: C, 58.96; H, 3.76; N, 15.17.

2-Amino-4-(7-methyl-4-oxo-3-(p-tolyl)-3,4-dihydropyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-8-yl)thiophene-3-carbonitrile (22): In dry flask a mixture 8-acetyl-7-methyl-3-(ptolyl)pyrido[ $\left.3^{\prime}, 2^{\prime}: 4,5\right]$ thieno[3,2-d][1,2,3]triazin- $4(3 \mathrm{H})$-one $21 \mathrm{a}(3.5 \mathrm{~g}, 10 \mathrm{mmol}$ ), malononitrile $(0.66 \mathrm{~g}$, $10 \mathrm{mmol})$ and elemental sulfer $(0.32 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol and few drops of triethylamine as base was heated under reflux for three hours. The mixture was left for cooling and poured onto ice cold water. The product was recovered by filtration and recrystallised from a mixture of ethanol/DMF (3:1) as brown crystals (3g, 69.7\%), M.p 260-262 oC; IR (KBr) v 3427 (NH2), 2208 (CN), $1683 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $430 \mathrm{M}+$; Anal. Calcd for C21H14N6OS2 (430.51): C, 58.59; H, 3.28; N, 19.52, Found: C, 58.43; H, 3.14; N, 19.36 .

8-(3-(Dimethylamino)acryloyl)-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-
4(3H)-one (23): In dry flask a mixture 8-acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2d] [1,2,3]triazin- $4(3 \mathrm{H})$-one $21 \mathrm{a}(3.5 \mathrm{~g}, 10 \mathrm{mmol})$ and DMFDMA $(1.32 \mathrm{ml}, 10 \mathrm{mmol})$ in dry dioxane was left under reflux for two hours. The mixture was left for cooling and evaporates the solvent. The product was recovered by filtration and recrystallised from ethanol as brown crystals ( $2.9 \mathrm{~g}, 71.6 \%$ ), Mp. 210-212 oC; 1H-NMR (DMSO) $\delta=2.39$ ( $3 \mathrm{H}, \mathrm{s}$, CH3 Ar), 2.66 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH} 3 \mathrm{py}$. ), 3.63, 3.67 ( 6 H , $2 \mathrm{~s}, \mathrm{NMe} 2), 5.42(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 7.82(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 7.41(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 7.54(2 \mathrm{H}, \mathrm{d}, \mathrm{Ar}), 9.12(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}$ py.); IR (KBr) v 16.44, $1693 \mathrm{~cm}-1$ (C=O); MS (EI)+: m/z $405 \mathrm{M}+$; Anal. Calcd for C21H19N5O2S (405.48): C, 62.21; H, 4.72; N, 17.27, Found: C, 62.12; H, 4.59; N, 17.11.

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