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Citation: Journal of Rheology **66**, 853 (2022); doi: 10.1122/8.0000515 View online: https://doi.org/10.1122/8.0000515 View Table of Contents: https://sor.scitation.org/toc/jor/66/5 Published by the The Society of Rheology

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Granulation and suspension rheology: A unified treatment

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(Received 1 June 2022; final revision received 22 June 2022; published 3 August 2022) Published open access through an agreement with JISC Collections 128554

Abstract

Mixing a small amount of liquid into a powder can give rise to dry-looking granules; increasing the amount of liquid eventually produces a flowing suspension. We perform experiments on these phenomena using Spheriglass, an industrially realistic model powder. Drawing on recent advances in understanding friction-induced shear thickening and jamming in suspensions, we offer a unified description of granulation and suspension rheology. A "liquid incorporation phase diagram" explains the existence of permanent and transient granules and the increase of granule size with liquid content. Our results point to rheology-based design principles for industrial granulation. © 2022 Author(s). All article content, except where otherwise noted, is licensed under a Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/). https://doi.org/10.1122/8.0000515

I. INTRODUCTION

Incorporating a small amount of liquid into powders is a ubiquitous unit operation in industrial materials processing. In some cases, a minimal amount of liquid is used to produce matt solid granules. Such "wet granulation" [1] has long been used in the manufacturing of, e.g., detergents, drugs, [2] and gunpowder [3]. When too much liquid is added, the mixture becomes (in granulation jargon) "overwet" [4]: it turns into a flowing suspension and granulation fails. However, such high-solid-content dispersions are the desired end point for other applications, e.g., ceramic pastes, construction and medical cements, and molten chocolate [5]. If insufficient liquid is added in these systems, flow fails; they fracture and jam, e.g., near constrictions.

Applied research into these two areas has hitherto proceeded separately: where the interest of one community ends ("overwet"), the attention of the other begins. However, in terms of the amount of liquid incorporated, the preparation of granules and high-solid dispersions form a continuum, as is visually apparent in published snapshots taken from the process of "conching" chocolate [5] (see also supplementary material [6]) and mixing concrete [7]. Thus, it is of interest to inquire whether, on some fundamental level, these two areas of industrial practice may be amenable to a single, unified description, with insights from each enriching the understanding of the other. Indeed, it has been suggested that shearinduced jamming may be related to granulation [8,9]: the

© Author(s) 2022. J. Rheol. 66(5), 853-858 September/October (2022) fracturing of an "underwet dispersion" into granules is the system's response to the impossibility of homogeneous flow.

We report an experimental study of the entire range of phenomena as progressively more liquid is incorporated into a model industrial powder of $\sim 10 \mu m$ glass particles. We observe successively the formation of two different kinds of granules, which, at the point of "overwetting," merge into a high-solid-content dispersion (Fig. 1). We characterize the rheology of these dispersions, which shear thicken (Fig. 2). Interpreting our data in light of recent advances in suspension rheology [10–16], we construct a "liquid incorporation phase diagram" (Fig. 3), which unifies the various possible states into a single conceptual scheme arising from shear-induced jamming. A "tie line construction" satisfying the "lever rule" explains the dependence of granule size on solid content (Fig. 4). Our results confirm theoretical proposals [8,9] that link suspension rheology and wet granulation.

Suspensions of glass spheres in a mixture of glycerol and water at various solid volume fractions, ϕ , were mixed using different methods to apply "high" and "low" stresses (see Sec. II for details). In all cases, liquid droplets with diameters much larger than that of the particles were added to powder with good wettability, such that small drops of suspension are created. In granulation terminology, granules are formed in the "immersion" regime [2]. Particles in the core of the granule are surrounded by liquid, termed the "capillary state," as opposed to granules in the "pendular" state, which are dominated by liquid bridges [17].

II. EXPERIMENTAL SYSTEM AND METHODS

We used soda-lime Potters Spheriglass (A-Type 5000, dried at 120 °C for 3 h), which consists mostly of polydisperse

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FIG. 1. The result of mixing at different volume fractions and stresses. (a) Output from the high-shear mixer and (b) and subsequent low stress vortex mixing. Tube diameter = 20 mm. Colors correspond to the ϕ regimes in Fig. 2. $\phi < \phi_{\rm m}$: samples are flowing, liquid suspensions in both cases. $\phi_{\rm m} < \phi < \phi_{\rm rep}$; transient granulation. High-shear mixing produces solid-like granules, which melt to a flowing liquid suspension upon vortex mixing. $\phi > \phi_{\rm rep}$; permanent granulation. Solid granules from the high-shear mixing remain solid upon vortex mixing.

hard spheres (HSs) (mean diameter $d = 7.2\,\mu$ m, polydispersity = 147%), but with some irregular shards [6]. The powder was dispersed into a 9:1 by volume glycerol-water mixture (viscosity $\eta_s = 0.336 \text{ Pa s}$) using a two-stepped protocol at various solid volume fractions $\phi = V_{\text{solid}}/(V_{\text{solid}} + V_{\text{liquid}})$ (so that in a dry powder $\phi = 1$) calculated from mass fractions [6]. High stress mixing using a bespoke mixer is followed by lower stress mixing using an Ika Vortex Genius 3 mixer.

Our bespoke device consisted of a high-torque overhead mixer (Ika Eurostar Power Control-Visc) driving an aluminum impeller with three equi-spaced blades with a 45° rake at 500 rpm. Liquid drops [diameter 4.57(7) mm $\gg d$] were added at 15 ml min⁻¹ using a syringe pump (New Era Pump Systems Inc. NE-1000). Mixing occurred in glass cylinders with diameter 110 mm (75 cm³ batches) or 75 mm (45 cm³ batches) [6]. Rheology measurements were made using an Anton-Paar MCR-301 stress-controlled rheometer with 50 mm roughened parallel plates (surface roughness $\approx 50 \mu$ m) separated by 1 mm. Experiments were performed at 19 °C.

III. RESULTS

The result of high and low stress mixing at $0.5 \le \phi \le 0.75$ is shown in Fig. 1, where we have color coded different regimes. Samples with $\phi = 0.75$ and 0.70 formed permanent granules; solid granules were produced upon high-stress mixing, which remained solid but grew in size after subsequent low-stress mixing. Samples with $\phi = 0.65$ and 0.60 exhibited transient granulation; they produced solid granules after high-stress mixing, but these merged to give flowing suspensions after low-stress mixing. Agitating the flowing suspensions under high shear regenerated the solid granules. Samples with $\phi = 0.55$ and 0.50 always produced flowing suspensions.



FIG. 2. Relative viscosity, η_r , as a function of volume fraction, ϕ . The lowstress, η_L , (\bullet) and high-stress (or thickened) η_H , (\bullet) branches with fits (solid lines) to Eq. (1) diverging at ϕ_{rep} and ϕ_m , respectively. $\phi < \phi_m$: the system flows at all applied stresses. $\phi_m < \phi < \phi_{rep}$: the system can only flow for applied stresses $\lesssim \sigma^*$. $\phi \ge \phi_{rep}$: samples never flow. *Inset*: η_r as a function of σ for various ϕ . ϕ , $\phi =$ points plotted using the same symbols in the main graph.

We next characterized the rheology of the flowing suspensions produced in the range of $0.3 \le \phi \le 0.65$. These samples were prepared from granules made in the high-shear mixer at $\phi = 0.65$, which were then vortex mixed to a flowing suspension and diluted to the required volume fraction. These suspensions shear thicken (Fig. 2, inset): the viscosity increases from a lower to a higher Newtonian plateau as stress increases, with the onset stress for thickening, $\sigma^{\star} \approx 1$ Pa, being approximately independent of ϕ . The viscosity at a fixed stress, $\eta(\sigma; \phi)$, diverges at some $\phi_{I}(\sigma)$. Figure 2 shows the two limiting plots, $\eta_{\rm L}(\phi)$ for the lowstress ($\sigma \ll \sigma^{\star}$) Newtonian plateau and $\eta_{\rm H}(\phi)$ for the highstress ($\sigma \gg \sigma^{\star}$) Newtonian plateau. A fully developed high- σ plateau is typically curtailed by sample fracture, evidenced by $d\eta/d\sigma < 0$; we estimate the high- σ plateau using the highest viscosity reached.

For all σ , we find that $\eta(\sigma; \phi)$ follows

$$\eta_{\rm r}(\sigma;\,\phi) = \frac{\eta(\sigma;\,\phi)}{\eta_{\rm s}} = \left[1 - \frac{\phi}{\phi_{\rm J}(\sigma)}\right]^{-\lambda},\tag{1}$$

with $\lambda = 1.74 \pm 0.02$. This exercise yields the jamming point as a function of stress, $\phi_J(\sigma)$ (Fig. S3 [6]). Fitting the limiting branches $\eta_L(\phi)$ and $\eta_H(\phi)$ returns $\phi_J(\sigma \ll \sigma^*) = 0.662$ and $\phi_J(\sigma \gg \sigma^*) = 0.568$, respectively.

The rheology shown in Fig. 2 is qualitatively identical to the shear thickening behavior of nearly monodisperse HSs [13,18], which is commonly caused by the stress-driven transition from lubricated to frictional interparticle contacts [10-16], although other mechanisms of constraining interparticle sliding also give rise to such behavior [19-21]. At



FIG. 3. A ϕ - σ phase diagram for liquid incorporation into powders. Bold curve and \bullet : $\sigma_J(\phi)$ derived from fitting the values of $\phi_J(\sigma)$ [6]. To its left and right, we find flowing and jammed states, respectively. Color coding: as in Figs. 1 and 2. Filled triangles; data from Fig. 1 representing granules (\blacktriangle) and flowing suspensions (\bigstar). Mixing a sample with average composition ϕ_0 at high stress (point *A*) gives jammed granules at composition ϕ_m and dry powder at composition $\phi = 1$, with relative proportions AC/AB. Mixing the same average composition at a lower stress (say, 0.05 Pa), *D*, gives granules at composition *E* ($\phi = \phi_{rep}$) and dry powder (*F*), with relative volumes of DE/DF.

 $\sigma \ll \sigma^{\star}$, all contacts are lubricated, and $\eta(\phi)$ diverges at random close packing, $\phi_{\rm rcp}$. On the other hand, when $\sigma \gg \sigma^{\star}$, all contacts are frictional, and $\eta(\phi)$ diverges at a jamming point $\phi_{\rm m} < \phi_{\rm rcp}$. In monodisperse HSs, $\phi_{\rm rcp} = 0.64$ and $\phi_{\rm m} = 0.55$ in the limit of large interparticle friction coefficient [11]. In our polydisperse system, $\phi_{\rm rcp} = 0.662$ and $\phi_{\rm m} = 0.568$.

These two volume fractions demarcate three regimes in Fig. 2. At $\phi < \phi_m$ (red), the system flows at all applied stresses, transitioning from a low- to a high-viscosity state as σ increases beyond σ^* . At $\phi_m \le \phi < \phi_{rcp}$ (yellow), the system flows at $\sigma < \sigma^*$ but jams into a solid-like state [9,22] at higher stresses. Finally, at $\phi > \phi_{rcp}$ (green), there is inadequate liquid to disperse all the particles, and there is no meaningful "suspension rheology." Significantly, the three types of behavior shown in Fig. 1 occur in precisely the three concentration regimes in Fig. 2, pointing to a connection between rheology and liquid incorporation via the physics of jamming [8,9].

To elucidate this connection, we turn to the measured jamming point as a function of applied stress, $\phi_J(\sigma)$, which can be fitted well by a stretched exponential [6]. The inverse function $\sigma_J(\phi)$ (Fig. 3) gives the "jamming phase boundary" separating flowing and jammed state points. Red and green regions flow and jam, respectively, at all stresses, whereas the yellow region will flow at low stresses and form granules at high stress, with the transition between the two states beginning at $\sigma \sim \sigma^*$.

In order to situate our observations (Fig. 1) on Fig. 3, we need to estimate typical stresses encountered in our protocol. The main mechanical action in our high-shear mixer occurs in the $h \approx 1 \text{ mm}$ gap between the tip of the blade (speed

 $v \approx 3 \,\mathrm{m \, s^{-1}}$) and the bowl, at a shear rate of $\sim v/h \sim 3000 \,\mathrm{s^{-1}}$. Multiplying by the appropriate viscosity turns this into a stress. Below $\phi_{\rm m}$, we use the high-stress plateau value; above $\phi_{\rm m}$, we use the high-stress plateau value; above $\phi_{\rm m}$, we use the high-stress plateau value of the most concentrated sample below $\phi_{\rm m}$ as a lower bound. In our vortex mixer, the centrifugal stress acting on a particle is $\approx 0.2 \,\mathrm{Pa} \leq \sigma^{\star}$ [6]. These crude estimates locate the observations shown in Fig. 1 on the jamming phase diagram to within order of magnitude on the stress axis (Fig. 3). It is clear that the jamming phase boundary demarcates granules (\blacktriangle) and flowing suspensions (\bigstar) in our experiments. Granulation is indeed directly related to jamming.

Cates and others have outlined a physical mechanism for this connection [8,9]. A suspension droplet at $\phi_m < \phi < \phi_{rcp}$ subjected to stress $\sigma > \sigma_J(\phi)$ will be in a jammed, solid state. Trying to shear a jammed state at constant $\dot{\gamma}$ generates very large stresses, which we assume exceed $\sigma_{cap} \sim \Sigma/a$, a capillary stress scale (where Σ is the interfacial tension and a = d/2 is the particle radius). Particles will, therefore, protrude from the droplet–air interface [23]. If $\sigma_{cap} > \sigma_J(\phi)$, then upon removal of the applied stress, the negatively curved interfaces between protruding particles continue to exert a high enough stress on the droplet to trap it in a jammed, solid state; it is a granule, which is matt because of protruding particles. Note, in passing, that there has long been experimental evidence for a critical Σ/a ratio for granulation [24].

Such granules are "fragile" [25]: each offers solid-like resistance only to the particular configuration of stresses that jammed it in the first place. Even very low stresses in other directions will unjam the state [26]. Thus, a granule with $\phi < \phi_{rep}$ subjected to random stresses below $\sigma_J(\phi)$ will unjam and remain fluid, because the external stresses are now insufficient to re-jam it. A collection of such droplets will merge into a bulk, flowing suspension. This mechanism explains the transient granulation observed in Fig. 1 (yellow region). When $\phi > \phi_{rep}$, there is no low stress flowing state available; the granules are permanent.

To confront observations, note that for us, $\Sigma \lesssim 65 \text{ mN m}^{-1}$ [27] and $a \approx 3 \,\mu\text{m}$, so that $\sigma_{\text{cap}} \sim \Sigma/a \sim 10^4 \text{ Pa}$. Indeed, in all granulated cases, $\sigma_{\text{cap}} > \sigma_{\text{J}}(\phi)$ (Fig. 3) providing direct confirmation of the link between granulation and shearinduced jamming [8,9].

As in a conventional phase diagram, a "lever rule" applies to Fig. 3, in both cases reflecting mass conservation. Consider a mixture with average composition $\phi = \phi_0$ (point A) consisting initially of dry powder at $\phi = 1$ (point C) and liquid sprayed into the mixer at $\phi = 0$. We propose that as mixing progresses, particles are incorporated into droplets (all $\gg d$) until their composition reach point B, which for $\sigma \gg \sigma^*$ will be at $\approx \phi_m$. Here, the droplets jam solid and will not incorporate any more particles, so that some unincorporated dry powder remains. Mass conservation then predicts that the proportion of granules to dry powder is given by the lever-rule construction, viz., AC/AB. Similarly, granulating a mixture starting at state point D gives granules at ϕ_{rcp} (point E) coexisting with powder (point F) in proportion DF/DE.

Contrary to the proposal above, we observed only matt granules but no loose, dry powder up to $\phi \lesssim 0.85$. The most parsimonious assumption is to postulate granules in which

 10^{2}

 10^{1}

ill a



 ϕ

0.75

0.85

 $\phi_{\rm rcp}$

high- σ

low- σ

0.65

the outmost layer of particles protrude, which can be treated as a layer of dry powder with thickness *d*. Following from this assumption and using the lever rule (=mass conservation), we can predict the radius of such granules. Working in more general terms for later use, the volume of solid and liquid in a granule of radius *R* with a dry powder shell of thickness t_s , see the inset of Fig. 4, which in general may be dependent on stress or shear rate [28], and $\phi = 1$ (dry = no liquid) and a liquid-saturated jammed core ($\phi = \phi_J$) are

$$V_{\text{solid}} = \phi_{\text{J}}(\sigma) \left(\frac{4}{3}\pi R^3\right),\tag{2}$$

$$V_{\text{liquid}} = [1 - \phi_{\text{J}}(\sigma)] \left(\frac{4}{3}\pi [R - t_{\text{s}}(\sigma)]^3\right), \quad (3)$$

respectively. Since $\phi = V_{\text{solid}}/(V_{\text{solid}} + V_{\text{liquid}})$, we find

$$R = \frac{t_{\rm s}(\sigma)}{1 - \left[\left(\frac{\phi_{\rm J}(\sigma)}{1 - \phi_{\rm J}(\sigma)} \frac{1 - \phi}{\phi} \right)^{1/3} \right]}.$$
 (4)

The predictions for $t_s = d$ and $\sigma_J = \phi_{rcp}$ and ϕ_m are shown in Fig. 4 (dotted–dashed). In both cases, *R* decreases with ϕ , starting from a divergence at ϕ_J . The latter is qualitatively consistent with previous reports of diverging granule size "at 100% liquid saturation" [29,30].

To test Eq. (4) quantitatively, we measured the volumeweighted mean granule radius, $\bar{R}(\phi)$, prepared using a smaller version of the high-shear mixer used for the experiments reported so far (to conserve materials) [6]. We use the volume-weighted radius since the lever rule concerns mass conservation and we use volume as a surrogate for mass. Our results for granulating at high and low stresses are plotted in Fig. 4 (points). Note that experiments at high stress below ϕ_{rcp} proved impractical: granules fluidized into a high viscosity suspension in the low-stress regions of our instrument, which required a higher torque to flow than this mixer could generate. Nevertheless, *R* does indeed increases with decreasing ϕ , with clear evidence for a divergence for the low- σ data at $\approx \phi_{rcp}$. The predictions of Eq. (4) with $t_s = d$ capture the form of the experimental data but are numerically two orders of magnitude too small.

To investigate the cause of this discrepancy, we cut open a granule [6], revealing a wet core and an essentially dry shell that was many particles thick. Homogeneous granules in which all particles are slightly wetted and cohere by capillary bridges could be envisaged, but are not observed under our conditions. Such core–shell granules have been observed before [31,32]. Presumably, there is a very small amount of liquid in this dry shell forming capillary bridges to bind the particles together [2,33]; but the liquid content is too small to visualize, justifying the assumption of $\phi = 1$ in the shell we used in arriving at Eq. (4). The assumption that the jammed core is at ϕ_J is justified by x-ray tomography [6], which detects $\leq 2\%_{vol.}$ air inside granules.

We now fit Eq. (4) to our data, with the shell thickness $t_s(\sigma)$ as a stress-dependent parameter [Fig. 4 (dotted)]. This returns shell thicknesses of 74*d* and 54*d* for high- and low- σ , respectively, possibly suggesting that high-stress mixing makes available somewhat more liquid for capillary bridges to build a thicker dry shell. Fitting our two data sets using a single shell thickness returns $t_s = 60d$, while measurement of a single granule [6] gave $\approx 45d$. (Multiple measurements failed because granules typically disintegrated during sectioning.) Performing a similar analysis using a polydisperse granule distribution [6] does not materially change these conclusions.

IV. SUMMARY

To summarize, we have shown that the phenomenology of liquid incorporation into powders with repulsive interactions is dominated by shear thickening, which, *in extremis*, leads to jamming. Thus, the main control parameter is the stress to which the system is subjected during mixing, because there is a volume-fraction-dependent stress, $\sigma_J(\phi)$, above which frictional contacts result in jamming. Mixing below $\sigma_J(\phi)$ yields flowing suspensions. Mixing above $\sigma_J(\phi)$ yields wet, jammed homogeneous cores coexisting with dry powder. We observed these coexisting states as core and shell, respectively, in heterogeneous granules. Conservation of mass leads to the prediction of divergent granule sizes as $\phi \rightarrow \phi_J$ from above, as observed here and in previous work. Quantitatively, a stress-dependent shell size gives good fits to the observed granule size as a function of ϕ .

Measurement of power or torque, which we believe is a proxy of stress, is regularly used to determine granulation "end point" [34] and has previously been shown to correlate with granule saturation [35] and size [30]. Furthermore, stress has been explored explicitly as a control parameter in scale-up operations [36]. However, mixing stress in the



 10^3

0.55

 $\frac{a}{B}$

856

context of maximum packing fraction has not been identified as a key variable before in granulation: a well-known "granule regime map" [4] uses the "maximum pore saturation" and a "deformation number" as control parameters. The former roughly plays the role of ϕ^{-1} , while the latter measures kinetic energy density inside the mixer. These variables control kinetic processes, and affect, e.g., the thickness of the dry "shell" on core-shell granules.

Many other parameters are known to affect granule size [1]. Binder viscosity is an example. Its role may simply reflect the use of shear rate as a mixing variable, so that viscosity is simply another proxy for stress. Alternatively, previous work has shown that granules held together by liquid bridges have a higher dynamic strength with increasing viscosity [28]. This may lead to variations in shell thickness and, in turn, granule size.

Our incorporation phase diagram (Fig. 3) suggests novel design principles. For example, the formulation space in which granulation can occur could be tuned by varying $\phi_{\rm rcp}$ and $\phi_{\rm m}$ through modification of, e.g., the particle size or shape distributions. Separately, while the outside of a coreshell granule will always be at $\phi = 1$, the density of the core can be "tuned" by moving along $\sigma_{\rm J}(\phi)$ by using different mixing stresses. Furthermore, advances in shear-thickening rheology tell us how to modify the $\sigma_{\rm J}(\phi)$ curve itself: e.g., $\phi_{\rm m}$ can be tuned by changing the interparticle friction coefficient [10,11], e.g., by using a variety of surface additives [37]. A full understanding, of course, has to await future work that brings together our incorporation phase diagram and the kinetic regime map [4]. Note that the kinetic energy density axis in the latter has the dimensions of stress, which is an axis in the former. A deeper connection seems probable.

Finally, many systems exhibit more complex rheology such as shear thinning when additional constraints, e.g., adhesion, are considered [19]. The impact of such complexity on the incorporation of liquid into powders is not immediately clear and should be probed in future work.

ACKNOWLEDGMENTS

We thank François Lequeux (ESPCI, Paris) for directing us to [4]. D.H. held an EPSRC studentship. E.B. was funded by Mars Chocolates UK Ltd. M.H. and W.P. were funded by EPSRC Grant Nos. EP/J007404/1 and EP/N025318/1. D.H. held an EPSRC studentship. E.B. was funded by Mars Chocolates UK Ltd.

AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

DATA AVAILABILITY

The data that support the findings of this study are openly available at https://doi.org/10.7488/ds/2588.

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