

Electrochemical Characterization of Imidazole-based Carboxamidrazones in Aqueous and Organic Solutions and Structure-Activity Relationship

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The search for new antibiotics is an emergent topic in the scientific community due to the appearance of pathogenic organisms multiresistant to the available drugs and therapy [1]. The worrying increase in infections caused by pathogenic fungi in immunosuppressed patients prompted the search for new antifungal agents with new mechanisms of action and less side effects [2]. The imidazole ring is a crucial scaffold for drug design due to its vital role in various biological processes [3]. Thus, imidazole-based molecules have been widely used to synthesize promising heterocyclic ligands for several drug targets. Furthermore, amidrazones are known for their high reactivity, being useful intermediates for the synthesis of compounds with a wide range of biological activities, including antimicrobial action. The amidrazone derivatives have also been applied in different areas of chemistry, specifically in the synthesis of azo molecules [4].

In a previous work, novel imidazole based 5-aminoimidazole-4-carboxamidrazones were synthesized, which exhibited great antifungal activity evaluated against three yeast species, namely *Candida krusei*, *Candida albicans* and *Candida parapsilosis* [5].

In this communication, the electrochemical characterization of some carboxamidrazones performed by cyclic voltammetry in two different media is compared. The effects of substituent groups in the different carboxamidrazones on the variation of oxidation potential of these compounds in aqueous and organic media were also analyzed. To find a relationship between chemical structure and activity, the minimum inhibitory concentration was correlated with their oxidation potentials. The effect of the variation of pH on the oxidation potential of substituted carboxamidrazones was also analyzed. In addition, kinetic studies of the evolution of some carboxamidrazones in aqueous media characterized by UV-Vis spectroscopy were performed and the obtained results were correlated with the electrochemical and biological data.

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