

1 **Photorespiration in the context of Rubisco biochemistry, CO<sub>2</sub>**  
2 **diffusion, and metabolism**

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24 **Key words**

25 Rubisco, CO<sub>2</sub> diffusion, mesophyll conductance, photorespiratory C<sub>2</sub> cycle,  
26 photosynthetic efficiency, photorespiratory bypass, yield improvement, photosynthesis

27 **Abstract**

28

29 Photorespiratory metabolism is essential for plants to maintain functional  
30 photosynthesis in an oxygen-containing environment. Because the oxygenation reaction  
31 of Rubisco is followed by the loss of previously fixed carbon, photorespiration is often  
32 considered a wasteful process and considerable efforts are aimed at minimizing the  
33 negative impact of photorespiration on the plant's carbon uptake. However, the  
34 photorespiratory pathway has also many positive aspects, as it is well integrated within  
35 other metabolic processes, such as nitrogen assimilation and C<sub>1</sub> metabolism, and it is  
36 important for maintaining the redox balance of the plant. The overall effect of  
37 photorespiratory carbon loss on the net CO<sub>2</sub> fixation of the plant is also strongly  
38 influenced by the physiology of the leaf related to CO<sub>2</sub> diffusion. This review outlines the  
39 distinction between Rubisco oxygenation and photorespiratory CO<sub>2</sub> release as a basis to  
40 evaluate the costs and benefits of photorespiration.

41

42 **Introduction**

43

44 Photosynthesis is quantitatively the most important biochemical pathway on the planet,  
45 by which plants turn sunlight and atmospheric CO<sub>2</sub> into organic biomass. It thereby  
46 provides the foundation of most life on Earth. The key enzyme involved in the  
47 carboxylation of ribulose 1,5-bisphosphate (RuBP), which drives the assimilation of CO<sub>2</sub>  
48 in photosynthetic organisms, is RuBP carboxylase/oxygenase (Rubisco). For every

49 carboxylation reaction two molecules of 3-phosphoglycerate (3-PGA) are produced that  
50 can be metabolized in the Calvin-Benson-Bassham (CBB) cycle to either regenerate  
51 RuBP, or to be exported as triose phosphate and become the substrate for most other  
52 organic compounds that make up a plant. Almost a century ago, Otto Warburg  
53 discovered that the rate of CO<sub>2</sub> assimilation is suppressed by increasing the external  
54 oxygen concentration (Warburg, 1920; Nickelsen, 2007). Only much later, and more  
55 than a decade after the discovery of Rubisco's capacity to carboxylate its substrate RuBP  
56 (Quayle et al., 1954; Weissbach et al., 1954; Mayaudon et al., 1957), it was established  
57 that Rubisco is an enzyme that can also react with oxygen to produce one molecule of 2-  
58 phosphoglycolate (2-PG) along with one molecule of 3-PGA (Bowes et al., 1971).

59         The two-carbon molecule 2-PG is a potent inhibitor of several enzymes, including  
60 triose phosphate isomerase (Anderson, 1971), sedoheptulose 1,7-bisphosphatase  
61 (Flügel et al., 2017), and phosphofructokinase (Kelly and Latzko, 1976) and therefore  
62 needs to be metabolized quickly. This is accomplished via the photorespiratory pathway,  
63 also known as C<sub>2</sub> pathway or the oxidative photosynthetic carbon cycle. This pathway  
64 recovers 75% of the 2-PG carbon by converting it to 3-PGA, which is then fed back into  
65 the CBB cycle where it can be used to regenerate RuBP (Berry et al., 1978). The  
66 remaining 25% of the carbon is released as CO<sub>2</sub> from photorespiration, meaning  
67 'respiration in the light', as this process is light-dependent. Photorespiration tends to be  
68 higher at high temperatures, low CO<sub>2</sub> concentrations and high O<sub>2</sub> concentrations  
69 (Sharkey, 1988).

70           Because of the release of previously fixed CO<sub>2</sub> and the energetic cost involved in  
71 metabolizing 2-PG, photorespiration has often been considered wasteful and it has been  
72 estimated that decreasing photorespiration could significantly increase food production  
73 (Walker et al., 2016). However, considering photorespiration as merely an unwanted  
74 side reaction is too simplistic, as the photorespiratory pathway is linked with several  
75 other metabolic processes that are essential to the plant. In addition to being connected  
76 with the carbon metabolism of the CBB cycle via Rubisco activity and the returned 3-  
77 PGA, photorespiration is also involved in nitrogen assimilation (Bloom, 2015; Busch et  
78 al., 2018), C<sub>1</sub> metabolism, and amino acid and phospholipid biosynthesis (Hanson and  
79 Roje, 2001; Ros et al., 2014). The photorespiratory pathway may also be beneficial  
80 through the dissipation of excess energy (Kozaki and Takeba, 1996) and has important  
81 implications in balancing the ratio of NADPH and ATP within the cell (Kramer and Evans,  
82 2011).

83           Photorespiration is the metabolic consequence of the generation of 2-PG via the  
84 oxygenation reaction of Rubisco that is recycled via the photorespiratory pathway. As  
85 such, the oxygenation reaction is the ultimate cause for photorespiration and, to a large  
86 extent, also determines the rate of photorespiratory CO<sub>2</sub> release. Although often  
87 conflated with the oxygenation reaction by Rubisco, the CO<sub>2</sub> release from  
88 photorespiration is an interrelated, but independent process and the stoichiometry of  
89 oxygenation and photorespiration is not fully fixed. Metabolic pathways that drain  
90 carbon from the photorespiratory pathway, such as glycine being exported for protein  
91 synthesis (Busch et al., 2018) or serine used for C<sub>1</sub> metabolism (Mouillon et al., 1999),

92 will alter the amount of CO<sub>2</sub> released per oxygenation reaction. It is therefore instructive  
93 to consider the generation of 2-PG and its metabolism separately. In the following I  
94 outline the two sides of photorespiration: (1) the enzymatic properties of Rubisco and  
95 aspects of the leaf affecting CO<sub>2</sub> diffusion that determine the generation of 2-PG; and (2)  
96 the metabolic context in which photorespiratory recycling of 2-PG occurs. It is these two  
97 sides of photorespiration that in combination define photorespiratory CO<sub>2</sub> release and  
98 thereby the costs and benefits of photorespiration. I conclude by discussing whether  
99 and under which conditions photorespiration might be considered wasteful.

100

## 101 **The oxygenation of RuBP by Rubisco**

102

103 The rate limiting step of CO<sub>2</sub> fixation in plants is catalyzed by Rubisco, a hexadecameric  
104 enzyme that consists of eight large subunits (RbcL) containing the catalytic site and eight  
105 small subunits (RbcS) that have some influence over the kinetic properties of Rubisco  
106 (Pottier et al., 2018). The key kinetic parameters of Rubisco include the Michaelis-  
107 Menten constants for CO<sub>2</sub> ( $K_c$ ) and O<sub>2</sub> ( $K_o$ ), and the catalytic turnover speed for  
108 carboxylation and oxygenation ( $k_{cat}^c$  and  $k_{cat}^o$ , respectively, in turnovers per second). The  
109 value of  $k_{cat}^c$  is only  $\sim 2-3 \text{ s}^{-1}$  in C<sub>3</sub> plants (Whitney et al., 2011), which is the reason why  
110 Rubisco constitutes around 20-30% of soluble leaf protein, but may reach more than  
111 50% depending on the growth condition (Makino and Osmond, 1991; Galmés et al.,  
112 2014), corresponding to 10-25% of total leaf nitrogen (Onoda et al., 2017; Evans and  
113 Clarke, 2019). Its high abundance across all photosynthetic organisms means Rubisco

114 may be the most abundant protein on Earth (Ellis, 1979; Bar-On and Milo, 2019). Both  
 115 the slow turnover rate and the competitive inhibition by O<sub>2</sub> give Rubisco the reputation  
 116 of being an inefficient enzyme. However, a recent comparison with chemically related  
 117 enzymes showed that Rubisco's catalytic performance is not unusual and the perception  
 118 of it being a particularly sluggish enzyme may be unwarranted (Bathellier et al., 2018).

119 We can mathematically relate the maximum carboxylation capacity ( $V_{cmax}$ ) and  
 120 oxygenation capacity ( $V_{omax}$ ) of Rubisco to the total concentration of enzyme sites ( $E_t$ ) by  
 121 defining  $V_{cmax} = k_{cat}^c E_t$  and  $V_{omax} = k_{cat}^o E_t$ . From these equations we can calculate the ratio  
 122 of the actual Rubisco carboxylation rate ( $V_c$ ) relative to its oxygenation rate ( $V_o$ ) as (Laing  
 123 et al., 1974):

$$124 \quad \frac{V_c}{V_o} = \frac{V_{cmax}}{K_c} \frac{K_o}{V_{omax}} \frac{C_c}{O}, \quad (1)$$

125 where  $C_c$  is the CO<sub>2</sub> concentration and  $O$  the O<sub>2</sub> concentration at the Rubisco active site.  
 126 The ratio of the carboxylation rate to the oxygenation rate when  $C_c$  and  $O$  are equal is  
 127 called the relative specificity of Rubisco ( $S_{c/o}$ ) and is given by

$$128 \quad S_{c/o} = \frac{V_{cmax}}{K_c} \frac{K_o}{V_{omax}}. \quad (2)$$

129  $S_{c/o}$  is therefore a property that is determined by the fundamental enzyme kinetics of  
 130 Rubisco. See Tcherkez (2016) for a detailed discussion of the reaction mechanisms  
 131 responsible for the partitioning between carboxylation and oxygenation reactions of  
 132 Rubisco. For given  $C_c$  and  $O$ , the ratio of the oxygenation to the carboxylation rate is  
 133 determined by

134 
$$F = \frac{V_o}{V_c} = \left( \frac{1}{S_{c/o}} \right) \frac{O}{C_c} . \quad (3)$$

135 Eqn. (3) demonstrates that the production of 2-PG through RuBP oxygenation is in a  
136 fixed relationship with the production of 3-PGA for any given  $S_{c/o}$ ,  $C_c$ , and  $O$ . In other  
137 words, higher rates of CO<sub>2</sub> fixation are inextricably linked to higher rates of 2-PG  
138 production, which then require a higher capacity of 2-PG recycling in the  
139 photorespiratory pathway. Consequently, an uncoupling of 3-PGA from 2-PG synthesis  
140 can only be achieved by altering  $S_{c/o}$ ,  $C_c$ , or  $O$ .

141

#### 142 ***Variability of $S_{c/o}$***

143

144 The  $S_{c/o}$  of Rubisco is largely determined by the protein sequence of its subunits and  
145 even single amino acid substitutions can impact the kinetic properties of the enzyme. It  
146 has been shown that a substitution of Met309 to Ile309 of RbcL turns a C<sub>3</sub>-like enzyme  
147 with lower  $k_{cat}^c$  and higher CO<sub>2</sub> affinity (i.e. lower  $K_c$ ) into a C<sub>4</sub>-like enzyme with higher  
148  $k_{cat}^c$  and lower CO<sub>2</sub> affinity (Whitney et al., 2011). Other examples exist where small  
149 changes in the protein sequence affected Rubisco kinetic properties, including  $S_{c/o}$   
150 (Whitney and Sharwood, 2008) and residues of RbcL have been determined that are  
151 under positive selection towards specific kinetic parameters, indicating that RbcL is  
152 instrumental in determining the catalytic properties of the enzyme (Kapralov and  
153 Filatov, 2007; Kapralov et al., 2010; Galmés et al., 2014).

154           The role of the small subunit so far is not as well understood, but it has become  
155   apparent that it, too, has an impact on the catalytic properties of Rubisco (Spreitzer,  
156   2003; Kapralov et al., 2010). A transformation of the small subunit of *Sorghum bicolor*, a  
157   C<sub>4</sub> plant, into the C<sub>3</sub> plant rice has revealed a role for RbcS in controlling catalytic  
158   properties (Ishikawa et al., 2011). The transgenic rice plants obtained in these  
159   experiments expressing a C<sub>4</sub>-type RbcS alongside the native rice RbcL exhibited a higher  
160    $k_{cat}^c$  and lower  $S_{c/o}$  than the non-transgenic rice, making Rubisco's catalytic properties  
161   much more C<sub>4</sub>-like. Other hybrid enzymes consisting of *Chlamydomonas* RbcL and RbcS  
162   from spinach, sunflower, or *Arabidopsis* also produced enzymes with specificities that  
163   were intermediate between the wild-type *Chlamydomonas* and the higher plant Rubisco  
164   (Genkov et al., 2010). Similarly, a hybrid Rubisco with RbcL from sunflower and RbcS  
165   from tobacco showed  $S_{c/o}$  values intermediate between sunflower and tobacco  
166   (Sharwood et al., 2008). In contrast to RbcL, which is usually encoded by a single-copy  
167   gene in the chloroplast, RbcS is a multigene family that encodes several different  
168   isoforms in the nuclear genome. The number of rbcS genes can range from as few as 2 in  
169   the green alga *Chlamydomonas reinhardtii* to more than 22 in wheat (Spreitzer, 2003).  
170   Some of these RbcS variants that are mostly expressed in the mesophyll (M-type) share  
171   up to 100% amino acid identity. Another type of RbcS that exists in some plant species  
172   and was first found in trichomes (T-type), is distinctly different from the M-type (Pottier  
173   et al., 2018). So far, this T-type RbcS has been found in non-photosynthetic tissues such  
174   as roots, seeds, fruits, and secretory organs such as trichomes, but appears to be largely  
175   absent in photosynthetic tissues (Morita et al., 2016). It can be speculated that this type



176 of RbcS may be important when Rubisco is not being used in the context of the CBB  
177 cycle, such as in the fatty acid synthesis during seed development (Schwender et al.,  
178 2004). In rice the overexpression of its T-type *RbcS*, *OsRbcS1*, altered the effective  
179 catalytic properties of extracted Rubisco (Morita et al., 2014). In these experiments,  
180 *OsRbcS1* overexpression increased  $k_{cat}^c$  and  $K_c$ , with a concomitant decrease in  $S_{c/o}$ ,  
181 shifting Rubisco's catalytic properties towards those found in C<sub>4</sub> plants. Similar  
182 differences were observed when tobacco M-type and T-type *RbcS* genes were expressed  
183 in *Chlamydomonas* lacking its own small subunits (Laterre et al., 2017). Small subunits  
184 do not have to be as different as M-type and T-type RbcS to confer differences in  
185 Rubisco kinetic parameters. Substitutions of single *Chlamydomonas* RbcS residues have  
186 been shown to account for decreases in  $S_{c/o}$  of up to 8% (Spreitzer et al., 2001).

187         Although direct experimental evidence is lacking to date that the different M-  
188 type small subunits within a plant mediate substantial differences in Rubisco kinetic  
189 properties *in vivo*, it can be speculated that they do so. Several studies have shown that  
190 changes in growth environment, such as temperature (Yoon et al., 2001), light intensity  
191 (Wanner and Gruissem, 1991; Dedonder et al., 1993) and light quality (Eilenberg et al.,  
192 1998), CO<sub>2</sub> concentration (Cheng et al., 1998), as well as the developmental stage of the  
193 plant (Wanner and Gruissem, 1991), result in differential expression of *RbcS* variants.  
194 Plants may therefore acclimate to their environment by expressing a Rubisco isoform  
195 that is most suitable, e.g. one with high  $S_{c/o}$  under conditions where photorespiration  
196 would be high, such as high temperatures. An increase in  $S_{c/o}$  under high growth  
197 temperatures has been experimentally observed in spinach (Yamori et al., 2006),

198 indicating that the acclimation of Rubisco kinetics via expression of different *RbcS*  
199 isoforms is at least plausible.

200         Because the half-life of Rubisco inside leaves is in the order of one week  
201 (Simpson et al., 1981), any acclimation on the protein level of its kinetic properties  
202 cannot happen quickly. However, this does not mean that  $S_{c/o}$  necessarily needs to be  
203 considered constant over short periods of time. Recent evidence points to the possibility  
204 that there may be mechanisms through which Rubisco's kinetic properties can be  
205 altered post-translationally. The catalytic site of Rubisco (E) is inactive in its native form  
206 (Fig. 1). Its activation requires the binding of non-substrate  $\text{CO}_2$  to the Lys201 residue of  
207 the active site forming a carbamate (EC) (Sharwood, 2017). This is followed by the  
208 binding of  $\text{Mg}^{2+}$  to create a stable ECM complex, to which the substrate RuBP can bind  
209 resulting in the formation of ECMR. The presence of  $\text{Mg}^{2+}$  and its binding to the catalytic  
210 site therefore plays a critical role in the activation of Rubisco. It has been speculated  
211 that  $\text{Mg}^{2+}$  could be replaced by  $\text{Mn}^{2+}$  inside the chloroplast, which would slightly alter  
212 the geometry of the active site (Bloom and Lancaster, 2018). This may differentially  
213 change the catalytic properties and thereby favor the oxygenation reaction over the  
214 carboxylation reaction. However, so far it is uncertain how much this replacement  
215 contributes to the regulation of Rubisco kinetic properties inside the leaf. Experimental  
216 evidence of the effect of  $\text{Mn}^{2+}$  *in vivo* is required, especially since the change in  
217 specificity is accompanied by a drastic decrease in the overall catalytic rate (Jordan and  
218 Ogren, 1983; Bloom and Kameritsch, 2017).

219 From an evolutionary perspective,  $S_{c/o}$  is fairly constrained within  $C_3$  plants (~85-  
220 110 mol mol<sup>-1</sup> at 25°C)(Hermida-Carrera et al., 2016; Orr et al., 2016), but varies  
221 substantially when other organisms such as green and non-green algae or  
222 photosynthetic bacteria are included in the comparison, due in part to their different  
223 forms of Rubisco (~10 mol mol<sup>-1</sup> in photosynthetic bacteria, up to ~240 mol mol<sup>-1</sup> in red  
224 algae)(Savir et al., 2010; Young et al., 2016; Flamholz et al., 2019). This variation in  
225 specificity is negatively correlated with a variation in  $k_{cat}^c$ , which has led to the  
226 conclusion that there is an unavoidable trade-off between  $S_{c/o}$  and  $k_{cat}^c$ , meaning  
227 Rubisco is well optimized to its environmental conditions (Tcherkez et al., 2006; Savir et  
228 al., 2010). However, more recent analyses based on data obtained from many more  
229 species weaken the apparent inverse relationship between  $S_{c/o}$  and  $k_{cat}^c$  (Galmés et al.,  
230 2014; Flamholz et al., 2019), suggesting that photosynthetic organisms have some  
231 capacity to adjust  $V_o$  relative to  $V_c$  without compromising the catalytic rate too much.

232 As with most other enzymes, the kinetic properties of Rubisco are highly  
233 dependent on temperature. With increasing temperatures Rubisco progressively loses  
234 its specificity for CO<sub>2</sub> (Badger and Collatz, 1977; Sharwood et al., 2016), meaning that  
235 the oxygenation reaction becomes more dominant relative to the carboxylation  
236 reaction. This results in generally higher rates of oxygenation and higher losses of  
237 carbon from photorespiration at higher leaf temperatures. Surveying a large diversity of  
238 species has uncovered a substantial variability in how  $S_{c/o}$  is affected by temperature  
239 (Hermida-Carrera et al., 2016; Orr et al., 2016; Sharwood et al., 2016). Interestingly,  $C_3$   
240 plants from cool habitats contain Rubiscos that perform better at low temperatures,

241 while Rubiscos from warm-habitat plants perform better at high temperatures (Galmés  
242 et al., 2016). Despite these insights into the variability of Rubisco kinetics, the exact  
243 molecular bases for what makes  $S_{c/o}$  of some Rubiscos more invariant to temperature  
244 than others, and which subunit is the main driver for this effect, remain elusive.

245 Overall,  $S_{c/o}$  is variable to some degree, both through acclimation to current  
246 environmental conditions and adaptation to the plant's habitat. This necessarily impacts  
247  $V_o$  and thereby the rate of 2-PG recycling and consequentially the rate of  
248 photorespiration. It also means that despite the Rubisco oxygenation reaction  
249 potentially being wasteful and unavoidable, plants have some control over how much  
250 oxygenation of RuBP they allow to happen. It is interesting to note at this point that  
251 plants do not always seem to opt for the highest possible specificity. It is therefore  
252 worthwhile to discuss the benefits that plants receive from RuBP  
253 oxygenation/photorespiration.

254

### 255 ***Delivery of CO<sub>2</sub> into the chloroplast***

256

257 Besides  $S_{c/o}$ , the other important factor determining  $V_o$  inside the leaf is  $O/C_c$  (Eqn. (3)).  
258 Because the O<sub>2</sub> concentration in the atmosphere surrounding the leaf (~ 210,000 μmol  
259 mol<sup>-1</sup>) is much higher than the CO<sub>2</sub> concentration ( $C_a$ ; ~ 415 μmol mol<sup>-1</sup>), photosynthetic  
260 activity, which releases O<sub>2</sub> at a similar rate as it takes up CO<sub>2</sub>, does not substantially alter  
261  $O$ . In contrast, in photosynthesizing leaves  $C_c$  can vary considerably and reach values  
262 much below  $C_a$ . This is due to the rate of CO<sub>2</sub> uptake relative to  $C_a$  being much larger

263 than the rate of O<sub>2</sub> release relative to O, which according to Fick's law causes a much  
264 greater concentration gradient for CO<sub>2</sub> as compared to O<sub>2</sub>. We can therefore consider O  
265 to be a constant in the natural environment and focus here on the variability of C<sub>c</sub>.

266 The CO<sub>2</sub> needed for photosynthesis inside the chloroplast enters the leaf through  
267 stomata that tightly regulate the gas exchange between the atmosphere and the inside  
268 of the leaf. Thus, stomata impose a resistance to CO<sub>2</sub> diffusion ( $r_s$ ) that causes the  
269 intercellular CO<sub>2</sub> concentration ( $C_i$ ) to be lower than  $C_a$  (Fig. 2a). A similar effect occurs  
270 during the diffusion of CO<sub>2</sub> from the intercellular air space to the site of carboxylation in  
271 the chloroplast. Here, physical components such as cell walls and membranes impose an  
272 additional resistance, termed mesophyll resistance ( $r_m$ ). According to Fick's law of  
273 diffusion, the net CO<sub>2</sub> assimilation rate of a plant ( $A$ ) can be related to changes in the  
274 CO<sub>2</sub> concentrations along the diffusive path by making use of these resistances:

$$275 \quad A = \frac{C_a - C_i}{r_s} = \frac{C_i - C_c}{r_m} \quad (4)$$

276 Under normal growth conditions, both  $C_i/C_a$  and  $C_c/C_i$  tend to be around 0.7, resulting in  
277  $C_c/C_a$  of about 0.5 (Wong et al., 1978; von Caemmerer and Evans, 1991). This indicates  
278 that these resistances have a large impact on  $C_c$  and therefore  $V_o$ . Stomata respond to a  
279 range of environmental signals, such as light intensity and quality, CO<sub>2</sub> concentration,  
280 the humidity of the air, and the general water status of the plant (Assmann, 1993;  
281 Vavasseur and Raghavendra, 2005; Busch, 2014; Buckley, 2019). In general, stomata  
282 open when there is a metabolic need for CO<sub>2</sub> inside the leaf (i.e. the plant is  
283 photosynthesizing), and close when the water loss from transpiration exceeds the water  
284 availability (Wong et al., 1979; Wong et al., 1985, 1985; Hetherington and Woodward,

285 2003), making  $r_s$  variable on short time scales. Similarly, there is evidence that  $r_m$  is also  
286 variable, which may be at least in part due to the effective location of (photo)respiratory  
287 CO<sub>2</sub> release (Tholen et al., 2012). In the case of chloroplast covering the cell periphery  
288 tightly, this forces all (photo)respired CO<sub>2</sub> to diffuse through the chloroplast, effectively  
289 resulting in a CO<sub>2</sub> release ‘inside the chloroplast’ (Fig. 2a). If, however, chloroplast cover  
290 is incomplete and the (photo)respiratory CO<sub>2</sub> mixes with the CO<sub>2</sub> coming in from the  
291 intercellular air space mostly inside the cytosol, then  $r_m$  is an apparent resistance that  
292 depends on the diffusive resistance across the cell wall and plasmamembrane ( $r_{wp}$ ) and  
293 the resistance across the chloroplast membrane and stroma ( $r_{ch}$ ) (Fig. 2b) (for discussion  
294 and further explanation see Tholen et al., 2014; Yin and Struik, 2017; Ubierna et al.,  
295 2019):

$$296 \quad r_m = r_{wp} + r_{ch} \frac{V_c}{A} \quad (5)$$

297 It becomes evident from Eqn. (5) that  $r_m$  increases (and the inverse, mesophyll  
298 conductance ( $g_m$ ) decreases) dramatically as  $A$  approaches zero (Fig. 2c). This has also  
299 been shown experimentally (Busch et al., Accepted). Thus, the location of  
300 (photo)respiratory CO<sub>2</sub> release has some impact on  $C_c$ , especially when  $A$  is low. This  
301 idea is exploited by bioengineering approaches that introduce photorespiratory  
302 bypasses to relocate photorespiratory CO<sub>2</sub> release from the mitochondria to the  
303 chloroplast (Kebeish et al., 2007; Maier et al., 2012; Shen et al., 2019; South et al.,  
304 2019). This effectively changes the CO<sub>2</sub> diffusive properties from a “two-resistance” case  
305 (Fig. 2b) to a “single-resistance” case (Fig. 2a), which is associated with increases in  $g_m$  at  
306 low CO<sub>2</sub> concentrations (Fig. 2c).

307 Furthermore,  $r_m$  is influenced by anatomical traits, such as cell wall thickness  
 308 (Ellsworth et al., 2018), chloroplast shape and cover (Busch et al., 2013; Weise et al.,  
 309 2015), cell density, and the relative amount of intercellular air space (Lehmeier et al.,  
 310 2017). Mesophyll  $\text{CO}_2$  diffusion properties, and consequently the  $\text{CO}_2$  concentration  
 311 inside the chloroplast, therefore are mediated by the 3D anatomy of the mesophyll cells  
 312 and the leaf as a whole (Earles et al., 2019).

313 Overall, the effective total resistance for  $\text{CO}_2$  diffusion ( $r_t$ ) from  $C_a$  to  $C_c$  is highly  
 314 variable with the environment, causing  $C_c$  to change according to

$$315 \quad C_c = C_a - Ar_t . \quad (6)$$

316  $A$ , in turn, is the difference between the  $\text{CO}_2$  taken up by Rubisco and the  $\text{CO}_2$  released  
 317 from photorespiration ( $F$ ) and mitochondrial respiration ( $R_d$ ) and can therefore be  
 318 described by (Farquhar et al., 1980)

$$319 \quad A = V_c - F - R_d . \quad (7)$$

320 Here,  $F$  relates to  $V_o$  through  $F = \Gamma V_o$ , where  $\Gamma$  is the amount of  $\text{CO}_2$  released from  
 321 photorespiration per oxygenation reaction. The  $\text{CO}_2$  concentration, at which the  $\text{CO}_2$   
 322 uptake by Rubisco carboxylation equals  $\text{CO}_2$  release from photorespiration is called the  
 323  $\text{CO}_2$  compensation point in the absence of mitochondrial respiration ( $\Gamma^*$ ), and is  
 324 calculated according to

$$325 \quad G^* = \frac{\Gamma O}{S_{c/o}} \quad (8)$$

326 If the photorespiratory pathway functions as a closed cycle  $\Gamma = 0.5$ , corresponding to  
 327 25% of the 2-PG carbon lost as  $\text{CO}_2$ . Combining Eqn. (7) with Eqns. (3) and (6) we can  
 328 obtain an expression for  $V_c$

329 
$$V_c = \frac{A + R_d}{1 - F} = \frac{A + R_d}{1 - \frac{1}{S_{c/o}} \frac{O}{C_c}} = \frac{A + R_d}{1 - \frac{1}{S_{c/o}} \frac{O}{C_a - A r_t}} \quad (9)$$

330 and one for  $F$

331 
$$F = 1 - V_o = \frac{A + R_d}{\frac{1}{1 - F} - 1} = \frac{A + R_d}{\frac{S_{c/o} C_c}{1 - O} - 1} = \frac{A + R_d}{\frac{S_{c/o} C_a - A r_t}{1 - O} - 1} \quad (10)$$

332 Eqn. (10) highlights that  $F$  is not linearly related to  $A$  because of the additional factor  
 333  $C_a - A r_t$ . In other words, decreasing  $F$  by a certain amount does not increase  $A$  by the  
 334 same amount, because  $C_c$  will change at the same time as a consequence. This can be  
 335 assessed quantitatively by solving Eqn. (10) for  $A$ , yielding a quadratic equation in  $A$  with  
 336  $C_a$  as the reference  $\text{CO}_2$  concentration (see e.g. Farquhar and Busch, 2017). A graphical  
 337 representation and detailed description of this effect is displayed in Box 1. Most studies  
 338 to date estimate the impact of photorespiration on  $A$  by comparing  $\text{CO}_2$  assimilation  
 339 rates with and without photorespiration modelled at a common  $C_i$  or  $C_c$ , or investigate  
 340 the rate of photorespiratory  $\text{CO}_2$  release at a given internal  $\text{CO}_2$  concentration (see e.g.  
 341 Sharkey, 1988; Valentini et al., 1995; Walker et al., 2016). While these approaches may  
 342 account for  $g_m$ , they disregard that diffusion resistances simultaneously affect both  $A$   
 343 and  $C_c$ , and thus ignore that some of the photorespired  $\text{CO}_2$  is refixed. Box 1  
 344 demonstrates that the modelling of  $A$  has to be coupled to  $\text{CO}_2$  diffusion through the  
 345 stomata and the mesophyll when estimating the effect of photorespiration on net  $\text{CO}_2$   
 346 assimilation rate. The costs of photorespiration on food production as estimated by  
 347 Walker et al. (2016) is therefore likely an overestimation and should be revisited.

348



## 349 **Recycling of 2-PG and release of CO<sub>2</sub> by glycine decarboxylation**

350

### 351 *The primary photorespiratory pathway*

352

353 Under normal conditions, the oxygenation of RuBP comprises about a quarter of the  
354 total use of RuBP by Rubisco, and it can be considerably more under conditions where  
355 stomata are substantially impacting CO<sub>2</sub> exchange with the atmosphere. This makes the  
356 flux through the photorespiratory pathway the largest biochemical process in plants  
357 second only to the flux through the CBB cycle. The 2-PG produced in the oxygenation  
358 reaction is dephosphorylated to glycolate by phosphoglycolate phosphatase 1 (PGLP1) in  
359 the chloroplast (Fig. 3). Glycolate is exported from the chloroplast by the plastidial  
360 glycolate/glycerate translocator 1 (PLGG1), which exchanges chloroplastic glycolate for  
361 cytosolic glycerate (Pick et al., 2013). A second transporter, the bile acid sodium  
362 symporter BASS6, has recently been implicated in the glycerate-independent export of  
363 glycolate, ensuring the balance of glycolate export and glycerate import for varying flux-  
364 ratios of these metabolites (South et al., 2017).

365 In the peroxisome glycolate reacts irreversibly with O<sub>2</sub> and is converted to  
366 glyoxylate by glycolate oxidase (GOX), producing H<sub>2</sub>O<sub>2</sub> as a byproduct. Catalase (CAT)  
367 then decomposes H<sub>2</sub>O<sub>2</sub> to water and oxygen. Next, the enzyme glutamate:glyoxylate  
368 aminotransferase (GGAT) transaminates glyoxylate to form glycine. A second enzyme  
369 that catalyzes the conversion of glyoxylate to glycine is serine:glyoxylate  
370 aminotransferase (SGAT), which returns the amino group from the three-carbon branch

371 of the photorespiratory pathway to the two-carbon branch by converting serine into  
372 hydroxypyruvate (Fig. 3; Bauwe et al., 2010). The glycine produced is subsequently  
373 exported from the peroxisome and taken up by mitochondria. So far, no transporters or  
374 channels facilitating the exchange of photorespiratory metabolites between the  
375 peroxisomes and the cytosol, or the mitochondrial glycine import and serine export,  
376 have been identified (Eisenhut et al., 2013). In the mitochondria glycine is  
377 decarboxylated (the source of 'photorespiration') by the multi-enzyme glycine  
378 decarboxylase complex (GDC) containing the four cooperating enzymes P-, H-, T-, and L-  
379 protein (Douce et al., 2001). This step catalyzes the conversion of the co-factor  
380 tetrahydrofolate (THF) to 5,10-methylene-THF (CH<sub>2</sub>-THF), which acts as the leaf's  
381 currency for activated one-carbon (C<sub>1</sub>) units. As a side product, NH<sub>3</sub> is released and  
382 NADH generated. CH<sub>2</sub>-THF then reacts with a second glycine to form serine in a reaction  
383 catalyzed by serine hydroxymethyltransferase 1 (SHMT1). Because the activity of GDC is  
384 higher than that of SHMT1, CH<sub>2</sub>-THF accumulates in the mitochondria relative to THF  
385 (Rebeille et al., 1994).

386 Photorespiratory serine is shuttled back to the peroxisome, where SGAT  
387 facilitates the transfer of the amino group to glyoxylate. The resulting hydroxypyruvate  
388 is converted to glycerate under the consumption of NADH. This happens either in the  
389 peroxisome, facilitated by hydroxypyruvate reductase (HPR1), or in the cytosol by a  
390 second hydroxypyruvate reductase (HPR2; Timm et al., 2008). A third enzyme, HPR3,  
391 can also react with glyoxylate to form glycolate and operates in the chloroplast (Timm et  
392 al., 2011). While the bulk of the flux goes through HPR1 in the peroxisome, the enzymes

393 in the other compartments allow for a redirection of the flux depending on the  
394 availability of NADH in the peroxisome (Timm et al., 2008). As the final step in the  
395 photorespiratory pathway, the chloroplastic glycerate kinase (GLYK) phosphorylates  
396 glycolate to 3-PGA, which can then enter the CBB cycle. In the case of the  
397 photorespiratory pathway operating as a full cycle, three out of four 2-PG carbon atoms  
398 will be returned to the CBB cycle and one will be released as CO<sub>2</sub>. A detailed description  
399 of the enzymatic steps involved in the photorespiratory pathway outlined above can be  
400 found in Bauwe (2018).

401

#### 402 ***Reassimilation of NH<sub>3</sub>***

403

404 Glycine decarboxylation in the mitochondria releases NH<sub>3</sub> as a byproduct (Fig. 3). Plants  
405 have evolved an efficient mechanism that recaptures and recycles photorespiratory  
406 ammonia and it has been estimated that less than 1% is lost to the atmosphere in the  
407 process (Mattsson and Schjoerring, 1996). The reassimilation of ammonia is facilitated  
408 by two enzymes in the chloroplast, glutamine synthetase (GS2) and ferredoxin-  
409 dependent glutamine:oxoglutarate aminotransferase (GOGAT; Coschigano et al., 1998).  
410 GS2 converts ammonia and glutamate to glutamine in an ATP-consuming process (Fig.  
411 3). Glutamine then donates an amino group to 2-oxoglutarate (2-OG) to form glutamate  
412 in a reaction catalyzed by GOGAT. The glutamate is subsequently transported to the  
413 peroxisome, where it is used by GGAT to convert glyoxylate to glycine. The product of  
414 this reaction, 2-OG, is returned to the chloroplast to close the cycle.

415

## 416 **Interaction with other metabolic pathways**

417

418 The photorespiratory pathway is essential for all oxygenic phototrophs as a way to  
419 metabolize 2-PG and salvage the majority of its carbon. Its importance is demonstrated  
420 when the pathway is disrupted by deletion of any of its enzymes, resulting in a  
421 photorespiratory phenotype, which may even be lethal (summarized e.g. by Timm and  
422 Bauwe, 2013; Eisenhut et al., 2019). The photorespiratory pathway, however, also has  
423 other important functions owing to its central position within plant metabolism, and is  
424 coupled to the nitrogen-, sulfur-, and C<sub>1</sub>- metabolisms by supplying key metabolites and  
425 coupled to the TCA cycle and respiration by changing the cell's redox and energy  
426 balance (Abadie et al., 2017). For recent reviews on the metabolic integration of the  
427 photorespiratory pathway see Hodges et al. (2016) and Obata et al. (2016).

428

### 429 ***Photorespiration and nitrogen assimilation***

430 The photorespiratory pathway is not only linked to nitrogen assimilation via the  
431 reassimilation of photorespiratory ammonia through GS/GOGAT, but also through the  
432 reduction and assimilation of new nitrogen, most importantly NO<sub>3</sub><sup>-</sup> (Bloom, 2015).  
433 Decreasing the flux through the photorespiratory pathway results in decreased rates of  
434 NO<sub>3</sub><sup>-</sup> uptake and assimilation, demonstrating a clear link between *de novo* nitrogen  
435 assimilation and photorespiration (Rachmilevitch et al., 2004; Bloom et al., 2010; Bloom  
436 et al., 2012). The photorespiratory pathway is the main biosynthetic pathway for the

437 amino acids glycine and serine, which are used for many purposes in plant metabolism  
438 other than regenerating 3-PGA, such as protein synthesis and as precursors for several  
439 other amino acids and phospholipids (Ros et al., 2014). Amino acids produced via the  
440 photorespiratory pathway also have a role in stress mitigation as they are key  
441 constituents of dehydrins, glutathione, and glycine betaine that increase tolerance to  
442 desiccation and prevent damage from reactive oxygen species (Close, 1997; Sakamoto  
443 and Murata, 2002; Layton et al., 2010; Noctor et al., 2012). If  $\text{NO}_3^-$  is available to support  
444 *de novo* nitrogen assimilation into glycine and serine, a considerable amount of these  
445 amino acids can be diverted from the photorespiratory pathway for other uses, meaning  
446 the pathway does not exhibit a fully cyclic nature (Abadie et al., 2016; Busch et al.,  
447 2018). Remarkably, because the diverted amino acids contain carbon in addition to  
448 nitrogen, the carbon and nitrogen metabolism act synergistically, and this means that  
449 photorespiration can increase the overall carbon uptake of a plant despite reducing the  
450 efficiency of Rubisco carboxylation (Busch et al., 2018).

451

#### 452 ***Photorespiration and C<sub>1</sub> metabolism***

453 The photorespiratory pathway also interacts with the C<sub>1</sub> metabolism as the main supply  
454 of activated C<sub>1</sub> units in the form of CH<sub>2</sub>-THF (Li et al., 2003). CH<sub>2</sub>-THF has numerous uses  
455 throughout the plant's metabolism, such as in the synthesis of nucleic acids, proteins,  
456 lipids, chlorophyll, pantothenate, and other methylated molecules (Cossins, 2000;  
457 Hanson and Roje, 2001; Gorelova et al., 2017). It is the precursor of several derivatives  
458 of THF that are interconverted between 10-formyl-THF (10-CHO-THF), 5,10-methenyl-

459 THF (5,10-CH<sup>+</sup>-THF), 5,10-methylene-THF (CH<sub>2</sub>-THF), and 5-methyl-THF (5-CH<sub>3</sub>-THF),  
460 differing by their oxidation state (Fig. 4). C<sub>1</sub> units are drawn from these pools for the  
461 synthesis of purines, thymidylate, and pantothenate. The 5-CH<sub>3</sub>-THF that is derived from  
462 CH<sub>2</sub>-THF in a reaction catalyzed by 5,10-methylene-THF reductase (MTHFR) can be used  
463 to convert homocysteine to methionine. In addition to protein synthesis, methionine is  
464 also involved in the production of S-adenosyl-methionine (SAM), which is used by  
465 different methyltransferases (MT) for the methylation of DNA, RNA, proteins,  
466 phospholipids, and other substrates (Crider et al., 2012). Both methionine and SAM are  
467 assumed to be produced predominantly in the cytosol (Isegawa et al., 1993), supported  
468 by the SHMT1-mediated conversion of photorespiratory serine to glycine. The C<sub>1</sub>  
469 metabolism also plays a major role in the biosynthesis of many secondary products,  
470 such as glycine betaine, nicotine, and lignin. It was estimated that the carbon demand in  
471 the form of C<sub>1</sub> units for these secondary metabolites can be in the order of 2 mmol C<sub>1</sub>  
472 units/g dry weight, or about 5% of the total assimilated carbon (Hanson and Roje, 2001).

473

#### 474 ***Photorespiration and sulfur metabolism***

475 The photorespiratory pathway further intersects with, and stimulates, sulfur metabolism  
476 (Abadie and Tcherkez, 2019). The synthesis of O-acetylserine by serine O-  
477 acetyltransferase (SAT3) draws from the serine pool in the mitochondria. Reduced sulfur  
478 is then incorporated into O-acetylserine via O-acetylserine (thiol) lyase (OAS-TL) to form  
479 cysteine. Cysteine is the primary product of S-assimilation and is required for, among  
480 other things, methionine synthesis and glutathione metabolism (Rausch and Wachter,

481 2005). While the photorespiratory flux supporting this S-metabolism is fairly small  
482 compared to that supporting the N- or C<sub>1</sub>-metabolism, it should not be considered  
483 negligible (Tcherkez and Tea, 2013).

484

#### 485 **The stoichiometry of CO<sub>2</sub> release per oxygenation reaction**

486

487 The way photorespiration is embedded with other metabolic pathways shows that  
488 numerous biochemical processes rely on the supply of metabolites from the  
489 photorespiratory pathway. It is therefore likely that a sizeable proportion of  
490 photorespiratory carbon leaves the pathway in the form of glycine, serine, and CH<sub>2</sub>-THF  
491 and does not get recycled back to 3-PGA. In support of this, the release of CO<sub>2</sub> per  
492 oxygenation reaction ( $\Gamma$ ) was shown to be variable and not fixed at 0.5 (Hanson and  
493 Peterson, 1985, 1986). In their early attempt to measure the stoichiometry of  
494 photorespiratory CO<sub>2</sub> release per oxygenation reaction *in vivo* they determined values  
495 ranging from 0.30 to 0.84, depending on environmental conditions such as temperature  
496 and light intensity. Since then, other studies have reported a departure of  $\Gamma$  from 0.5 in  
497 mutants that had an impairment in the photorespiratory pathway, such as plants lacking  
498 the peroxisomal malate dehydrogenase (PMDH; Cousins et al., 2008), HPR1 (Cousins et  
499 al., 2011; Timm et al., 2011), or the thioredoxins THXh2 or THXo1 (da Fonseca-Pereira et  
500 al., 2019; Reinholdt et al., 2019), and in WT tobacco, wheat, soybean, and Arabidopsis  
501 (Walker and Cousins, 2013; Walker et al., 2017). A value of  $\Gamma$  lower than 0.5 would be  
502 expected if carbon is diverted from the photorespiratory pathway as glycine (Busch et

503 al., 2018). In contrast, the export in the form of CH<sub>2</sub>-THF would increase this value  
504 towards  $\Gamma = 1$ , because every carbon exported as CH<sub>2</sub>-THF releases one carbon as CO<sub>2</sub> in  
505 its production by GDC (see Box 2). Based on this, one might expect values of  $\Gamma$  that are  
506 lower than 0.5 in plants that draw more carbon from the photorespiratory pathway in  
507 the form of glycine, and values greater than 0.5 in plants that use large amounts of CH<sub>2</sub>-  
508 THF to synthesize metabolites such as lignin or nicotine. Related to this, Walker et al.  
509 (2017) reported that the value of  $\Gamma$  increases with temperature much more in tobacco  
510 than in wheat and soybean. Interestingly, this increase in  $\Gamma$  corresponds well with an  
511 increase in nicotine biosynthesis in tobacco plants exposed to high temperatures (Chen  
512 et al., 2016). While the reasons for changes in  $\Gamma$  *in vivo* are mostly speculative at the  
513 moment, further research on the extent of use of photorespiratory carbon for other  
514 metabolic processes will undoubtedly lead to significant advances in our understanding  
515 of the costs and benefits of photorespiration.

516

### 517 **Costs vs. benefits of photorespiration**

518

519 The interactions of the photorespiratory pathway with other metabolic pathways that  
520 draw from photorespiratory metabolites shine a light on which aspects of  
521 photorespiration, if any, should be considered wasteful. As outlined above, oxygenation  
522 itself is not wasteful *per se* and can provide substantial benefits towards maintaining the  
523 redox homeostasis of the cell. The release of photorespiratory CO<sub>2</sub> by GDC should also  
524 not be considered a wasteful process as such, as long as it is related to producing



525 metabolites that are needed elsewhere in metabolism, such as glycine, serine, or CH<sub>2</sub>-  
526 THF. Producing these compounds independently of the photorespiratory pathway would  
527 also necessitate reducing equivalents and ATP of similar magnitude – or potentially  
528 more – than when going via the photorespiratory pathway. It would also require a  
529 steady supply of C<sub>1</sub> compounds that have to be generated with an associated CO<sub>2</sub>  
530 release. Any cost/benefit analysis therefore has to include an analysis of the possible  
531 alternatives. With this in mind, the only portion of photorespiration that we may want  
532 to consider ‘wasteful’ is the flux of 2-PG that actually makes it back to the CBB cycle as  
533 3-PGA, as the 25% of carbon lost along the way is not compensated for by other  
534 processes (Fig. 3, Box 2). While this ‘wasteful’ proportion of 2-PG may comprise most of  
535 the flux coming from Rubisco’s oxygenation reaction under some conditions, such as  
536 high temperatures or low CO<sub>2</sub> supply to the chloroplast, it potentially makes up only a  
537 small part under other conditions. I am not aware of any direct measurements of  
538 *photorespiratory* 3-PGA production relative to 2-PG synthesis, i.e. the effective activities  
539 of GLYK relative to PGLP1, that would be useful to evaluate how much of the  
540 photorespiratory carbon is put to good use. Indirect evidence that photorespiration is  
541 not all wasteful, however, comes from photorespiratory mutant studies. As discussed  
542 above, values of  $\Gamma$  deviating from 0.5 have been previously observed experimentally  
543 (Hanson and Peterson, 1985, 1986; Cousins et al., 2008; Cousins et al., 2011; Timm et  
544 al., 2011; Walker and Cousins, 2013; da Fonseca-Pereira et al., 2019), which would be  
545 expected if carbon is exported from the photorespiratory pathway. These observations  
546 hint at that an important role of the photorespiratory pathway is to supply the demand

547 for metabolites for other processes and that it acts as an open pathway rather than a  
548 closed cycle when rates of oxygenation are not exceeding this demand. Future studies  
549 that aim at quantifying carbon export from the photorespiratory pathway could provide  
550 new insights into how leaky the pathway is under varying environments.

551         The interactions of the photorespiratory with other metabolic pathways, as well  
552 as the impact that leaf anatomy and physiology have on determining the net effect of  
553 photorespiration on carbon balance, make it difficult to assess the impact of these  
554 aspects individually on plant performance (see Box 3 for techniques that may be used to  
555 quantify photorespiratory carbon fluxes at different points along the pathway). We can,  
556 however, gauge the overall short-term effect of photorespiration on carbon uptake,  
557 ignoring potential effects due to long-term acclimation. The rate of oxygenation can be  
558 instantaneously decreased by experimentally decreasing  $O$  (see Eqn. (3)). This allows us  
559 to compare  $A$  as an integrative parameter for the net cost of photorespiration on  $\text{CO}_2$   
560 uptake under conditions with and without photorespiration. Despite a substantial  
561 proportion of Rubisco activity being used to support the oxygenation reaction, the net  
562 effect on  $A$  is much smaller for a wide range of environmental conditions (Fig. 5).  
563 Especially at  $C_a$  of ambient or higher concentrations, the net effect of photorespiration is  
564 negligible or even positive. Similarly, at temperatures below about  $20^\circ\text{C}$   
565 photorespiration has little or no negative consequences on  $A$  in the example shown in  
566 Fig. 5. In contrast, photorespiration quickly becomes disadvantageous as leaf  
567 temperatures rise above  $25^\circ\text{C}$ . Based on these examples, photorespiration appears to be  
568 beneficial for the overall carbon uptake for a  $V_o/V_c$  of up to approximately 0.25.

569 Photorespiration has further benefits that go beyond the carbon balance of the  
570 plant, e.g. due to its contribution to maintaining the redox homeostasis of the cell under  
571 abiotic stress conditions such as drought or chilling (Voss et al., 2013). Under certain  
572 conditions NADPH and ATP production by the photosynthetic light reactions may exceed  
573 their consumption in the CBB cycle. This is likely the case during light induction of  
574 photosynthesis under the fluctuating light environments that plants experience e.g. in  
575 the understory. Under low light, Rubisco and other CBB cycle enzymes tend to be  
576 deactivated and stomata to be relatively closed. If a leaf experiences a sudden increase  
577 in light intensity, NADPH and ATP production is induced rapidly (Björkman and Demmig-  
578 Adams, 1995), while their consumption is restricted due to slow CBB cycle activation and  
579 stomatal opening (Deans et al., 2019; Deans et al., 2019). Under these conditions, the  
580 photosynthetic electron transport chain can become over-reduced, causing an increased  
581 production of superoxide and other reactive oxygen species, which exacerbates the  
582 potential for photodamage (Niyogi, 1999). Photorespiration provides an extra outlet for  
583 NADPH and ATP, mitigating the negative effects of excessive light (Kozaki and Takeba,  
584 1996; Wingler et al., 2000; Takahashi and Badger, 2011; Eisenhut et al., 2017). While this  
585 might come at a carbon cost, the loss in  $A$  from photorespiration ( $DA$ ) is small at least in  
586 the case of low stomatal conductance (Box 1), seen e.g. during photosynthetic induction  
587 or drought stress. Independent of the magnitude of the actual carbon cost, the energy  
588 dissipation aspect of photorespiration increases the resilience of plants under adverse  
589 conditions, such as under variable light environments or when limited water availability  
590 causes stomata to close.

591 While photorespiration has certain benefits, it also has costs other than the  
592 direct impact on the carbon balance. At a  $V_o/V_c$  of 0.33 (an approximate ratio under  
593 normal growing conditions, see Fig. 5), one quarter of the nitrogen investment in  
594 Rubisco supports the oxygenation reaction, which negatively impacts the photosynthetic  
595 nitrogen use efficiency (PNUE) of  $C_3$  plants. Due to higher  $CO_2$  concentrations around  
596 Rubisco  $C_4$  plants have much lower rates of oxygenation, which allows them to  
597 compromise their Rubisco's  $S_{c/o}$  for a higher  $k_{cat}^c$ . Thus their nitrogen investment in  
598 Rubisco (and other enzymes in the photorespiratory pathway) is much lower than that  
599 of  $C_3$  plants, increasing their PNUE (Rotundo and Cipriotti, 2017). It has been shown that  
600 an efficient use of nitrogen in  $C_4$  photosynthesis lessens nitrogen cost constraints on  
601 molecular sequence evolution (Kelly, 2018). Photorespiration therefore slows both the  
602 rate of speciation and extinction in  $C_3$  plants as compared to  $C_4$  plants. Thus, the  $C_4$   
603 photosynthetic pathway may be just as much an adaptation to make the most of the  
604 available nitrogen as it is to increase carbon uptake of the plant.

605

### 606 **How have plants dealt with photorespiration evolutionarily?**

607

608 Despite the positive side effects of photorespiration, certain environmental conditions,  
609 specifically low  $CO_2$  concentrations, higher temperatures, and conditions inducing  
610 stomatal closure, will increase Rubisco oxygenation beyond a flux beneficial for overall  
611 plant performance. Plants have evolved several strategies to limit photorespiratory  
612 carbon loss, and species with mechanisms that counter photorespiration dominate

613 environments that are particularly prone to Rubisco oxygenation. It is assumed that a  
614 gradual decrease in the atmospheric CO<sub>2</sub>:O<sub>2</sub> ratio throughout Earth's history and the  
615 associated increase of photorespiration at the cost of carboxylation prompted an  
616 adaptation response to these conditions (Moroney et al., 2013). The mechanisms that  
617 many plant lineages have evolved can be categorized as either minimizing the rate of  
618 oxygenation, resulting in relatively higher carboxylation rates, or minimizing the loss of  
619 photorespiratory CO<sub>2</sub> downstream.

620

#### 621 ***Minimizing the rate of oxygenation***

622 The first category includes the carbon concentrating mechanisms (CCMs) of C<sub>4</sub> plants  
623 (evolved ~30 Ma; Christin et al., 2011), plants operating a crassulacean acid metabolism  
624 (CAM), and the carboxysomes and pyrenoids found in many cyanobacteria and algae,  
625 respectively (evolved ~350 Ma; Badger et al., 2002). The common principle by which  
626 these CCMs operate includes biochemical and anatomical changes that allow the  
627 concentration of CO<sub>2</sub> around Rubisco above ambient levels. This not only decreases the  
628 oxygenation rate of Rubisco, but also increases Rubisco's carboxylation rate due to its  
629 dependence on the absolute CO<sub>2</sub> concentration following Michaelis-Menten kinetics.

630         In the C<sub>4</sub> photosynthetic pathway the initial fixation of CO<sub>2</sub> is accomplished in the  
631 mesophyll cells by the enzyme phosphoenolpyruvate carboxylase (PEPC) producing the  
632 C<sub>4</sub> acid oxaloacetate (OAA). OAA is then converted to malate (or aspartate in some  
633 variants of the pathway), which is transported to the bundle sheath cells where Rubisco  
634 is located. There the C<sub>4</sub> acid is decarboxylated, producing pyruvate and CO<sub>2</sub>, which

635 increases the CO<sub>2</sub> concentration around Rubisco several-fold (Langdale, 2011). This  
636 biochemical pathway is usually accompanied by a C<sub>4</sub> Kranz leaf anatomy, which  
637 efficiently manages the separation of the C<sub>4</sub>- and C<sub>3</sub>-pathways (Edwards and  
638 Voznesenskaya, 2011). Although the interplay between changes in biochemistry and  
639 anatomy is very complex, C<sub>4</sub> photosynthesis is a highly convergent trait that has evolved  
640 at least 66 times independently, demonstrating its effectiveness for reducing  
641 photorespiration (Sage et al., 2011; Sage et al., 2012). The CAM pathway has similar  
642 features to the C<sub>4</sub> pathway, but with a temporal rather than spatial separation of C<sub>3</sub> and  
643 C<sub>4</sub> metabolisms (Dodd et al., 2002). In CAM plants CO<sub>2</sub> is initially fixed at night into OAA  
644 by PEPC, which is then converted to malate. Malate is stored in the vacuole until  
645 daytime, when the conversion back to CO<sub>2</sub> and pyruvate occurs. During the day, the  
646 plant's stomata remain closed, resulting in a similar increase in CO<sub>2</sub> concentration  
647 around Rubisco as in the bundle sheath cells of C<sub>4</sub> plants. Because the stomata are  
648 closed during mid-day when potential rates of transpiration are large, CAM plants can  
649 achieve high water-use efficiencies (Szarek and Ting, 1975).

650 CO<sub>2</sub> diffusion in water is several orders of magnitude slower than in air,  
651 restricting the ability of plants to access CO<sub>2</sub> for photosynthesis. Aquatic photosynthetic  
652 organisms therefore frequently contain CCMs based on encapsulating Rubisco in  
653 organelles that restrict outward CO<sub>2</sub> diffusion by a protein shell (in cyanobacterial  
654 carboxysomes; Rae et al., 2013) or a starch sheath that may be supplemented by an  
655 additional layer of proteins (in algal pyrenoids; Ramazanov et al., 1994). In both cases,  
656 CO<sub>2</sub> is converted to bicarbonate in the cytosol before being transported into the

657 carboxysome or pyrenoid, where it is converted back to CO<sub>2</sub> by the enzyme carbonic  
658 anhydrase. This results in an increase in CO<sub>2</sub> concentration around Rubisco, limiting the  
659 oxygenation of RuBP.

660 Despite the existence of a CCM, all these photosynthetic types possess, and  
661 require, a fully functioning photorespiratory pathway, though operating at lower fluxes  
662 due to decreased Rubisco oxygenation rates that come along with an increased CO<sub>2</sub>  
663 concentration around Rubisco (Eisenhut et al., 2008; Zelitch et al., 2009; Levey et al.,  
664 2019). Here, the metabolic demand for glycine and serine may at least partially be met  
665 through other metabolic pathways, such as the phosphorylated pathway (Igamberdiev  
666 and Kleczkowski, 2018). The essential nature of the photorespiratory pathway in C<sub>4</sub>  
667 plants, however, might point towards its contribution to providing metabolites for other  
668 pathways also in C<sub>4</sub> plants. In any case, increasing the CO<sub>2</sub> concentration inside the cell  
669 relaxes the need for a highly substrate-specific Rubisco. Organisms with a CCM usually  
670 express a Rubisco that trades off an increased  $k_{cat}^c$  for a decreased  $S_{c/o}$  (Badger et al.,  
671 1998; Sharwood et al., 2016; Heureux et al., 2017; Sharwood, 2017). Non-CCM ways to  
672 reduce Rubisco oxygenation include increasing  $S_{c/o}$  by lowering the leaf temperature,  
673 which can be achieved by decreasing the absorbance of solar radiation or increasing  
674 transpiration (Sage, 2013).

675

### 676 ***Minimizing the loss of photorespiratory carbon***

677 The second category, minimizing the loss of photorespiratory carbon downstream of the  
678 production of 2-PG, is a much more boutique approach that has not been as widely

679 adopted in nature. Considered an evolutionary intermediate condition between C<sub>3</sub> and  
680 C<sub>4</sub> photosynthesis, some plant species shuttle photorespiratory glycine from the  
681 mesophyll to the bundle sheath by restricting GDC to the mitochondria of the bundle  
682 sheath cells (Sage et al., 2012). These mitochondria are positioned against the  
683 centripetal wall, with chloroplasts covering the cell periphery towards the mesophyll  
684 cells (Sage et al., 2013; Sage et al., 2014). This increases the probability of refixation of  
685 the photorespiratory CO<sub>2</sub>, evidenced by a decrease in the apparent CO<sub>2</sub> compensation  
686 point (Sage et al., 2013; Khoshravesh et al., 2016), while at the same time increasing the  
687 CO<sub>2</sub> concentration around the population of Rubisco in the bundle sheath.  
688 Photorespiratory CO<sub>2</sub> scavenging via this glycine shuttle has been termed C<sub>2</sub>  
689 photosynthesis, which refers to the number of carbons in the shuttling metabolite  
690 glycine (Vogan et al., 2007). The arrangement of chloroplast around the cell periphery as  
691 barrier against outward CO<sub>2</sub> diffusion has been shown to be beneficial also in C<sub>3</sub> plants  
692 (Busch et al., 2013).

693 Thus, many different approaches exist in nature to combat excessive rates of  
694 photorespiration, with the CCMs being the most successful, both in terms of the number  
695 of species employing them and their global biomass production. The C<sub>4</sub> photosynthetic  
696 pathway stands out, occurring in 3% of the world's plant species and accounting for 23%  
697 of the terrestrial primary productivity (Sage et al., 1999; Still et al., 2003). It is a curious  
698 observation that the C<sub>4</sub> photosynthetic pathway is, with very few exceptions, limited to  
699 herbaceous species. This was hypothesized to be due to constraints associated with the  
700 evolutionary history of the C<sub>4</sub> lineages (Sage and Sultmanis, 2016). However, a



701 contributing factor could be related to the heavy reliance of the arborescent life form on  
702 lignin as structural support. Lignin is a large sink for C<sub>1</sub> units (Hanson and Roje, 2001)  
703 and one might speculate that this large demand for C<sub>1</sub> units of photorespiratory origin is  
704 the reason why the C<sub>4</sub> photosynthetic pathway is largely absent in trees.

705

## 706 **Concluding remarks**

707

708 In summary, predicting the net effect of photorespiration on the CO<sub>2</sub> uptake of plants  
709 requires knowledge of the anatomical and physiological properties of the leaf  
710 influencing CO<sub>2</sub> diffusion, the kinetic properties of the CO<sub>2</sub>-fixing enzyme Rubisco, and  
711 the biochemistry of the photorespiratory pathway and how it is connected to other  
712 metabolic processes. Each of these aspects influences the rate of photorespiration,  
713 resulting in the fact that knowledge of the photorespiratory CO<sub>2</sub> release alone is not  
714 sufficient to estimate the carbon cost of photorespiration. In particular, CO<sub>2</sub> diffusive  
715 resistances have a large impact on the carbon balance of the photorespiratory pathway.  
716 The price a plant has to pay in terms of forgone carbon assimilation is not equal to the  
717 carbon lost from photorespiration or the additional carbon that could be assimilated if  
718 the RuBP oxygenation reaction were to be replaced by its carboxylation. The actual  
719 carbon costs of photorespiration that come into play when  $V_o/V_c$  exceeds ~0.25 should  
720 be evaluated against the benefits of providing protection against photodamage and  
721 helping to balance the ATP to NADPH ratio of the cell, increasing the overall resilience of  
722 the plant operating in a variable environment.

723           Given that the photorespiratory pathway is so well integrated within the plant's  
724 metabolism, it may be instructive to view its main purpose as a biosynthetic pathway  
725 using a substrate that is available in excess, rather than a pathway for the detoxification  
726 of 2-PG. Rubisco can then be considered a dual-functioning enzyme that does not only  
727 facilitate the production of primary carbohydrates, but is also the first step in the  
728 biosynthesis of several amino acids and compounds relying on the C<sub>1</sub> metabolism. An  
729 integrated view of photorespiration within the context of the leaf's biochemical and  
730 diffusional properties will ultimately allow us to better target our research efforts  
731 towards modifying photorespiration in pursuit of increasing crop productivity.

732

733

#### 734 **Acknowledgements**

735 This work was supported by the Australian Government through the Australian Research  
736 Council Centre of Excellence for Translational Photosynthesis. I thank Graham Farquhar  
737 from The Australian National University for critical feedback on a draft of this  
738 manuscript.

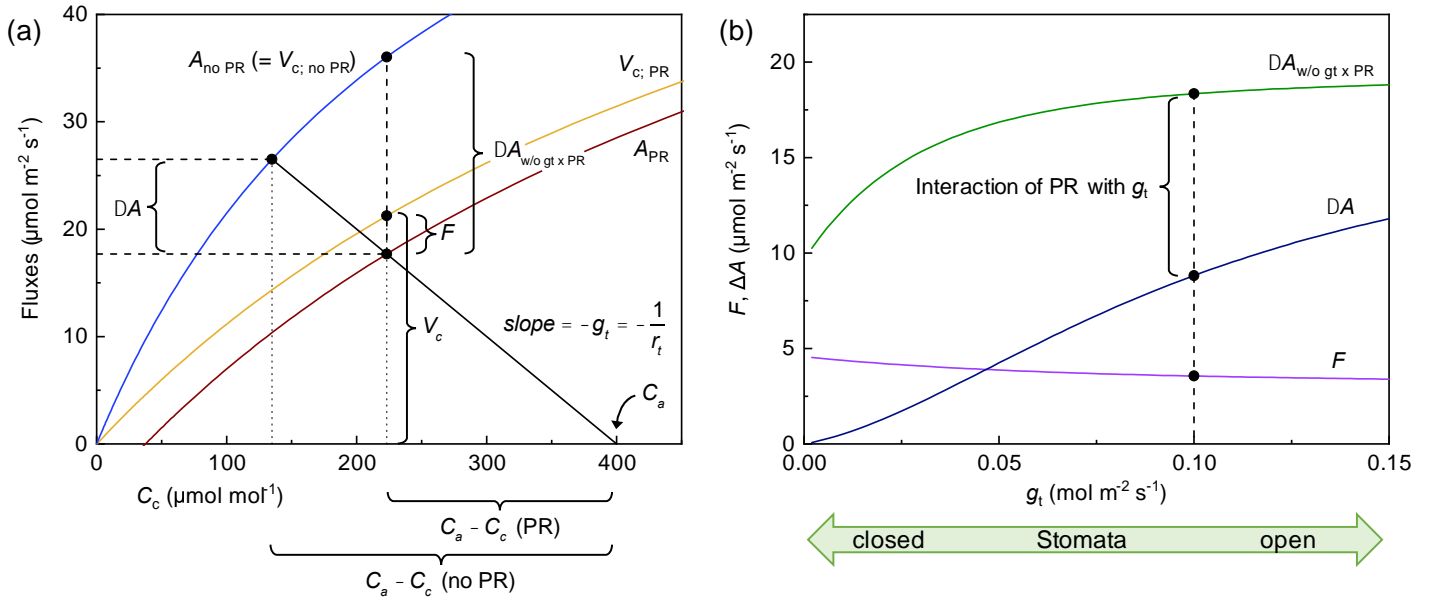
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#### 740 **Conflict of interest**

741 The author declares no conflict of interest.

742 **Box 1: The effect of diffusion properties on forgone carbon uptake due to**  
 743 **photorespiration**

744



745

746 The effect of CO<sub>2</sub> diffusion on carbon uptake forgone due to photorespiration. Panel (a)  
 747 shows a graphical representation of CO<sub>2</sub> assimilation rates with and without  
 748 photorespiration. In the simplest case, CO<sub>2</sub> uptake is limited by the rate of Rubisco  
 749 carboxylation (see Box 2 for the treatment of other biochemical limitations). Then,  $A$  can  
 750 be modelled according to Farquhar et al. (1980) as

751 
$$A = \frac{V_{c\text{max}} (C_c - \Gamma^*)}{C_c + K_c (1 + O/K_o)} - R_d, \quad (11)$$

752 where  $V_{c\text{max}}$  is the maximum carboxylation rate of Rubisco,  $K_c$  and  $K_o$  the Michaelis-  
 753 Menten constants for CO<sub>2</sub> and O<sub>2</sub>.  $O$  and  $C_c$  are the oxygen and CO<sub>2</sub> concentrations at  
 754 the site of carboxylation,  $R_d$  is the rate of mitochondrial respiration, and  $\Gamma^*$  is the CO<sub>2</sub>

755 compensation point in the absence of  $R_d$ , as defined by Eqn. (8). The red line  
756 corresponds to  $A$  when photorespiration is present, modeled with Eqn. (11) and  $V_{cmax} =$   
757  $80 \mu\text{mol m}^{-2} \text{s}^{-1}$ ,  $K_c = 272 \mu\text{mol mol}^{-1}$ ,  $K_o = 166 \text{mmol mol}^{-1}$  and  $O = 210 \text{mmol mol}^{-1}$ . Here,  
758  $R_d$  was assumed to be zero for simplicity.  $V_c$  in response to  $C_c$  when photorespiration is  
759 present is displayed by a yellow line. Because we ignore  $R_d$  in this example, the  
760 difference between  $V_c$  and  $A$  at a given  $C_c$  is then equal to  $F$  (Eqn. (7)). We can model  $A$   
761 for the case of no photorespiration occurring ( $A_{no\ PR}$ ; blue line), which is mathematically  
762 equivalent to setting  $O$  in Eqn. (11) to zero. Note that this also results in  $\Gamma^*$  being zero  
763 (Eqn. (8)). This has the effect of allowing Rubisco to put all of its activity towards the  
764 carboxylation reaction (i.e.  $A_{no\ PR} = V_c$ , as  $R_d = 0$ ). At any given  $C_c$ ,  $A_{PR}$  is lower than  $A_{no\ PR}$   
765 by  $DA_{w/o\ gt\ x\ PR}$ , which is the added effect of Rubisco forgoing carboxylation activity in  
766 favor of oxygenation of RuBP, and the loss of carbon from photorespiration. Here,  
767  $DA_{w/o\ gt\ x\ PR}$  represents the net effect of the Rubisco oxygenation reaction, if diffusion  
768 resistances are ignored.

769 The line originating at  $C_c = 400 \mu\text{mol mol}^{-1}$  and intercepting the  $A/C_c$  curve at the  
770 chloroplastic  $\text{CO}_2$  concentration corresponding to an ambient  $C_a$  of  $400 \mu\text{mol mol}^{-1}$  is  
771 called the supply function. It represents the decrease in  $C_c$  that will occur due to the  
772 assimilation of  $\text{CO}_2$  by Rubisco (indicated by the drop in  $\text{CO}_2$  concentration  $C_a - C_c$  (PR)).

773 From Eqn. (6) it follows that the slope of the supply function is  $-\frac{1}{r_t}$  ( $=-g_t$ ). This means  
774 that the  $C_c$  intersect of the supply function with the  $A/C_c$  curve decreases as  $r_t$  increases.  
775 Assuming  $r_t$  is the same under non-photorespiratory conditions, the supply function

776 intersects with the  $A_{\text{no PR}}$  line at a lower  $C_c$ , because the larger  $\text{CO}_2$  assimilation rate  
777 causes the  $\text{CO}_2$  draw-down ( $C_a - C_c$  (no PR)) to be larger. Therefore, the net effect of the  
778 Rubisco oxygenation reaction being present, if diffusion resistances are considered, is  
779 equal to  $DA$ .  $DA$  is generally less than  $DA_{\text{w/o } g_t \times \text{PR}}$ , emphasizing that diffusion  
780 resistances moderate the negative impact of photorespiration on  $A$ .

781 In a natural environment,  $C_a$  can be considered mostly constant, while  $C_c$  will  
782 vary on short timescales with changes in both  $\text{CO}_2$  diffusion properties and the capacity  
783 to fix  $\text{CO}_2$ . Panel (b) displays  $F$  and forgone rates of  $\text{CO}_2$  assimilation with ( $DA$ ) and  
784 without ( $DA_{\text{w/o } g_t \times \text{PR}}$ ) considering diffusion resistances as a function of the total  
785 conductance to  $\text{CO}_2$  diffusion ( $g_t$ ) at  $C_a = 400 \mu\text{mol mol}^{-1}$ . Here,  $g_t$  goes towards zero as  
786 stomata close – this is equivalent with the supply function in (a) becoming shallower –  
787 but it will also vary with varying mesophyll conductance. As stomata close, and thus  $C_c$   
788 decreases,  $F$  increases moderately. In contrast,  $DA$  decreases as  $g_t$  decreases, and  
789 reaches zero when the stomata are fully closed. This underscores, maybe somewhat  
790 counterintuitively, that the rate of photorespiratory  $\text{CO}_2$  release is not a good measure  
791 for the impact that photorespiration has on net  $\text{CO}_2$  uptake. Diffusive resistances have a  
792 big impact on the absolute amount of carbon uptake forgone due to photorespiratory  
793 processes, as demonstrated by  $DA_{\text{w/o } g_t \times \text{PR}}$  being considerably larger than  $DA$ .

794

795 **Box 2: Accounting for carbon export from the photorespiratory pathway**  
796 **when modelling CO<sub>2</sub> uptake**

797 Export of carbon from the photorespiratory pathway will influence how much carbon is  
798 assimilated by the plant overall (Busch et al., 2018). In addition, in the case of export of  
799 the nitrogen containing photorespiratory metabolites glycine and serine, it will also  
800 affect the nitrogen metabolism. Busch et al. (2018) have developed a model that  
801 describes net CO<sub>2</sub> uptake when considering carbon export from the photorespiratory  
802 pathway in the form of glycine and serine. Here, I provide a more generalized  
803 description that also accounts for the export of carbon through the C<sub>1</sub> metabolism (see  
804 Fig. 3).

805 Following Busch et al. (2018) we denote  $a_g$  the proportion of 2-PG carbon that  
806 is exported from the photorespiratory pathway as glycine and  $a_s$  the proportion  
807 exported as serine. Similarly, we define  $a_T$  as the proportion of 2-PG carbon exported as  
808 CH<sub>2</sub>-THF. The overall proportion of 2-PG carbon that is exported cannot exceed 1, and  
809 therefore the relation  $0 \leq a_g + 2a_T + \frac{4}{3}a_s \leq 1$  must hold. Adding up the electron  
810 requirement for carbon reduction, photorespiration, and nitrate reduction necessary to  
811 supply the nitrogen in the exported glycine and serine (Fig. S1, see Supporting  
812 Information), the actual photosynthetic electron transport rate can be calculated as

813 
$$J_a = \left(4 + \left(4 + 8a_g - 4a_T + 4a_s\right)F\right)V_c, \quad (12)$$

814 where  $F$  is defined as in Eqn. (3). Eqn. (12) indicates that  $J_a$  is both dependent on  $F$  as  
815 well as on the carbon exported from the photorespiratory pathway. The same is true for

816 the rate of ATP consumption, which is equal to  $\left(3 + \left(3.5 - 0.5a_G - a_T - 2/3a_S\right)F\right)V_c$ . If  
817 glycine is removed from the photorespiratory pathway, not only is the rate of electron  
818 transport used for nitrogen assimilation affected, but it also decreases the amount of  
819 CO<sub>2</sub> released per oxygenation reaction (  $l$  ). In contrast, the removal of CH<sub>2</sub>-THF from the  
820 pathway results in an increase in  $l$  , because for every carbon removed as CH<sub>2</sub>-THF one  
821 carbon is lost from glycine decarboxylation as CO<sub>2</sub>. The net CO<sub>2</sub> assimilation rate given  
822 by Eqn. (7) therefore needs to be parameterized with

$$823 \quad l = 0.5(1 - a_G) + a_T. \quad (13)$$

824 Equation (13) demonstrates that  $l$  can be less or greater than 0.5, subject to the  
825 magnitudes of  $a_G$  and  $a_T$ . Thus,  $A$  is described by

$$826 \quad A = V_c - \left(0.5(1 - a_G) + a_T\right)V_o - R_d. \quad (14)$$

827 Depending on the biochemical process limiting CO<sub>2</sub> assimilation,  $A$  can be described with  
828 the minimum of the three rates  $W_c$ ,  $W_j$ , and  $W_p$  (Farquhar et al., 1980), which are the  
829 carboxylation rates that can be supported under a Rubisco, electron transport or triose  
830 phosphate utilization (TPU) limitation, respectively, so that

$$831 \quad A = \min\{W_c, W_j, W_p\} \left(1 - \frac{\Gamma^*}{C_c}\right) - R_d \quad (15)$$

832 where  $\Gamma^*$  is defined as in Eqn. (8). It becomes evident that  $\Gamma^*$  varies with  $a_G$  and  $a_T$ .

833 When RuBP supply is not limiting the rate of Rubisco carboxylation,  $W_c$  is the limiting  
834 factor described by

$$835 \quad W_c = \frac{V_{cmax} C_c}{C_c + K_c \left(1 + O/K_o\right)}. \quad (16)$$

836 If the electron transport rate ( $J$ ) that drives the regeneration of RuBP is the process that  
837 limits  $A$ ,  $W_j$  is described by

$$838 \quad W_j = \frac{J}{4 + (4 + 8a_G - 4a_T + 4a_S)F} \quad (17)$$

839 Finally, at high  $\text{CO}_2$  concentrations RuBP regeneration may be controlled by the capacity  
840 for starch and sucrose synthesis from triose phosphates to regenerate inorganic  
841 phosphate for sustained ATP synthesis (Sharkey, 1985; Busch and Sage, 2017). If TPU is  
842 limiting the rate of carboxylation, we can write  $W_p$  as

$$843 \quad W_p = \frac{3T_p}{1 - 0.5(1 + 3a_G + 6a_T + 4a_S)F} \quad (18)$$

844 where  $T_p$  is the rate of triose phosphate utilization of the plant. In Eqn. (18) the  
845 denominator corresponds to the flux of carbon (scaled by  $V_c$ ) exported as triose-  
846 phosphates. While an attempts have been made to estimate the values of  $a_G$  and  $a_S$   
847 (Abadie et al., 2016; Busch et al., 2018), so far there is no information available as to the  
848 value of  $a_T$ .

849 Because carbon can leave the photorespiratory pathway in several locations, the  
850 flux of photorespiratory carbon that makes it back to the CBB cycle as 3-PGA is less than  
851 half the rate of RuBP oxygenation. We can quantify this flux mathematically, giving a  
852 rough indication of the degree of how 'wasteful' photorespiration is in terms of carbon  
853 (this, however, ignores the effect of  $\text{CO}_2$  diffusion and other beneficial aspects of  
854 photorespiration). This gives a flux of photorespiratory  $\text{CO}_2$  release that is not linked to  
855 supplying metabolites to other metabolic pathways of



856 Carbon 'wasted' by photorespiration =  $\left(0.5 - 0.5a_G - a_T - \frac{2}{3}a_S\right)F$ . (19)

857 Therefore, carbon is not 'wasted' by photorespiration if  $V_o$  does not exceed the  
858 metabolic demand for glycine, serine and CH<sub>2</sub>-THF, which is the case for

859  $a_G + 2a_T + \frac{4}{3}a_S = 1$ .

860

861

### 862 **Box 3: Measuring photorespiration**

863 As outlined in the main text, photorespiration is a multifaceted pathway that is tightly  
864 tied in with other metabolic pathways that draw metabolites from the photorespiratory  
865 pathway. As a consequence, the carbon flux through the photorespiratory pathway can  
866 change along the pathway and therefore may not be accurately represented by the  
867 oxygenation reaction of Rubisco or the CO<sub>2</sub> release from GDC. However, much could be  
868 learned about the nature of photorespiration and its costs and benefits by combining  
869 measurements of fluxes through different parts of the pathway. A range of techniques  
870 suitable to probe different aspects of photorespiration has been discussed previously  
871 (Sharkey, 1988; Busch, 2013). Other techniques that can be employed in combination to  
872 study the interactions of photorespiration with other biochemical pathways are outlined  
873 in the following.

874

#### 875 ***Estimation of the rate of Rubisco oxygenation***

876 The rate of Rubisco oxygenation, which is the main source of 2-PG, is largely determined  
877 by Rubisco kinetic properties and the CO<sub>2</sub> and O<sub>2</sub> concentrations at the site of  
878 carboxylation. While Rubisco kinetic properties can be determined *in vitro*, C<sub>c</sub> can be  
879 estimated *in vivo* from gas exchange analysis that is coupled with measurements of  
880 carbon isotope discrimination (Farquhar et al., 1982; Evans et al., 1986; Busch et al.,  
881 Accepted) or, with limitations, chlorophyll fluorescence (Epron et al., 1995; Warren,  
882 2006). An alternative strategy for estimating the rate of Rubisco oxygenation involves  
883 <sup>13</sup>CO<sub>2</sub> labeling followed by computational flux estimation (Ma et al., 2014, 2017). This  
884 “Isotopically Nonstationary Metabolic Flux Analysis” (INST-MFA) has the benefit of not  
885 requiring kinetic constants of the involved enzymes and therefore avoids some of the  
886 assumptions inherent to the other methods.

887

#### 888 ***Estimation of CO<sub>2</sub> release from GDC***

889 Carbon isotopes can be used in bulk to separate gross fluxes of CO<sub>2</sub> entering the leaf  
890 from (photo)respired CO<sub>2</sub> exiting the leaf. The efflux of photorespiratory <sup>12</sup>CO<sub>2</sub> can be  
891 measured in a <sup>13</sup>CO<sub>2</sub> atmosphere (Busch et al., 2013; Busch et al., 2017). A related  
892 method uses <sup>14</sup>CO<sub>2</sub> to separate the source pools of photorespiration into primary and  
893 stored photosynthates, allowing further details of photorespiratory CO<sub>2</sub> release to be  
894 obtained (Pärnik and Keerbergh, 1995, 2007).

895

#### 896 ***Estimation of glycine export from the photorespiratory pathway***

897 While the rate of glycine synthesis is a useful parameter to know, it does not inform us  
898 *per se* how much photorespiratory carbon leaves the photorespiratory pathway as  
899 glycine. Abadie et al. (2016) devised a technique based on isotope labelling and  
900 metabolome kinetics coupled with isotope ratio mass spectrometry and nuclear  
901 magnetic resonance (NMR) analyses to estimate how much photorespiratory glycine  
902 accumulates and is not converted to serine. This approach may be used to determine  
903 the value of  $\Gamma$ .

904

#### 905 ***Estimation of the flux through the C<sub>1</sub> metabolism***

906 Our knowledge about the magnitude of the flux through the C<sub>1</sub> metabolism to date is  
907 sparse. Attempts to measure the one-carbon fluxes associated with the  
908 photorespiratory metabolism in plants have been made using NMR techniques, and it  
909 has been shown that the C<sub>1</sub> units needed for serine synthesis originate mostly from  
910 photorespiratory carbon (Prabhu et al., 1996). Less is known about the fate of the CH<sub>2</sub>-  
911 THF produced by GDC. One of the sinks for C<sub>1</sub> units from the folate cycle is the synthesis  
912 of methionine. The flux of <sup>13</sup>C-label to methionine synthesis has been estimated with a  
913 similar NMR approach and been found to be scaled to net CO<sub>2</sub> assimilation (Gauthier et  
914 al., 2010; Abadie et al., 2017). Future studies are warranted to elucidate the overall  
915 fluxes through the folate cycle.

916

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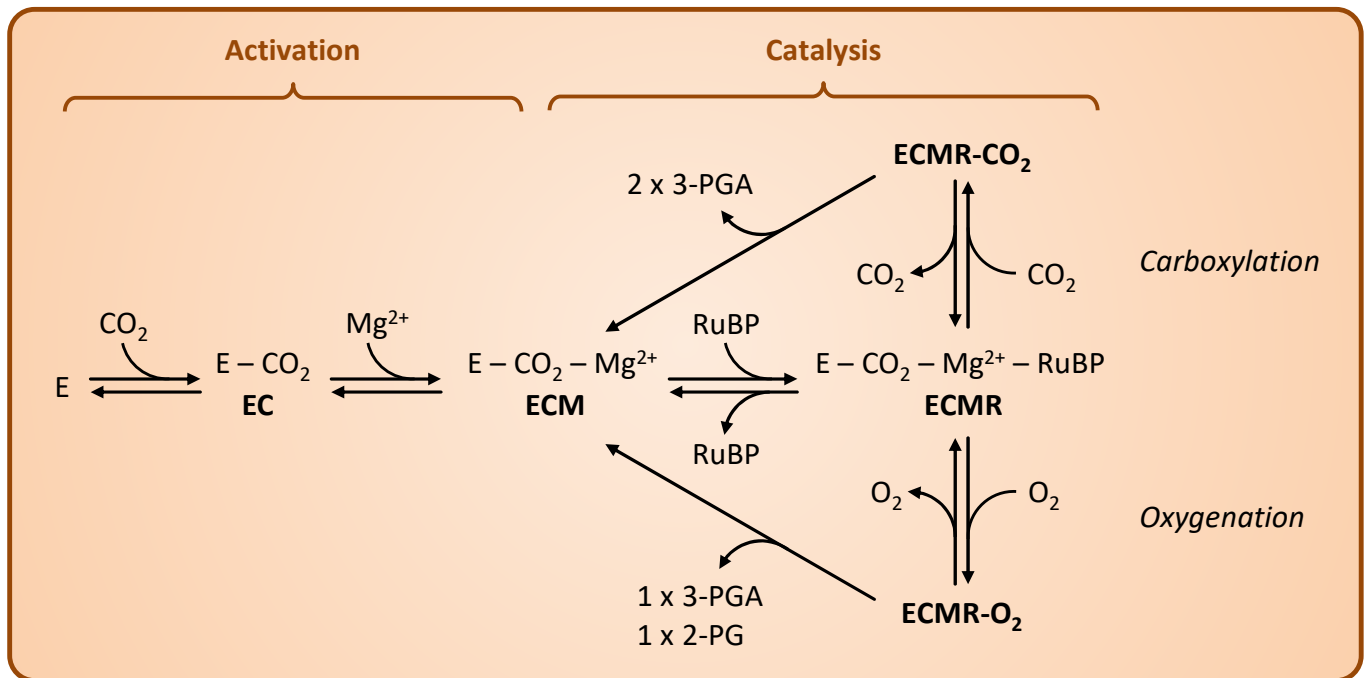
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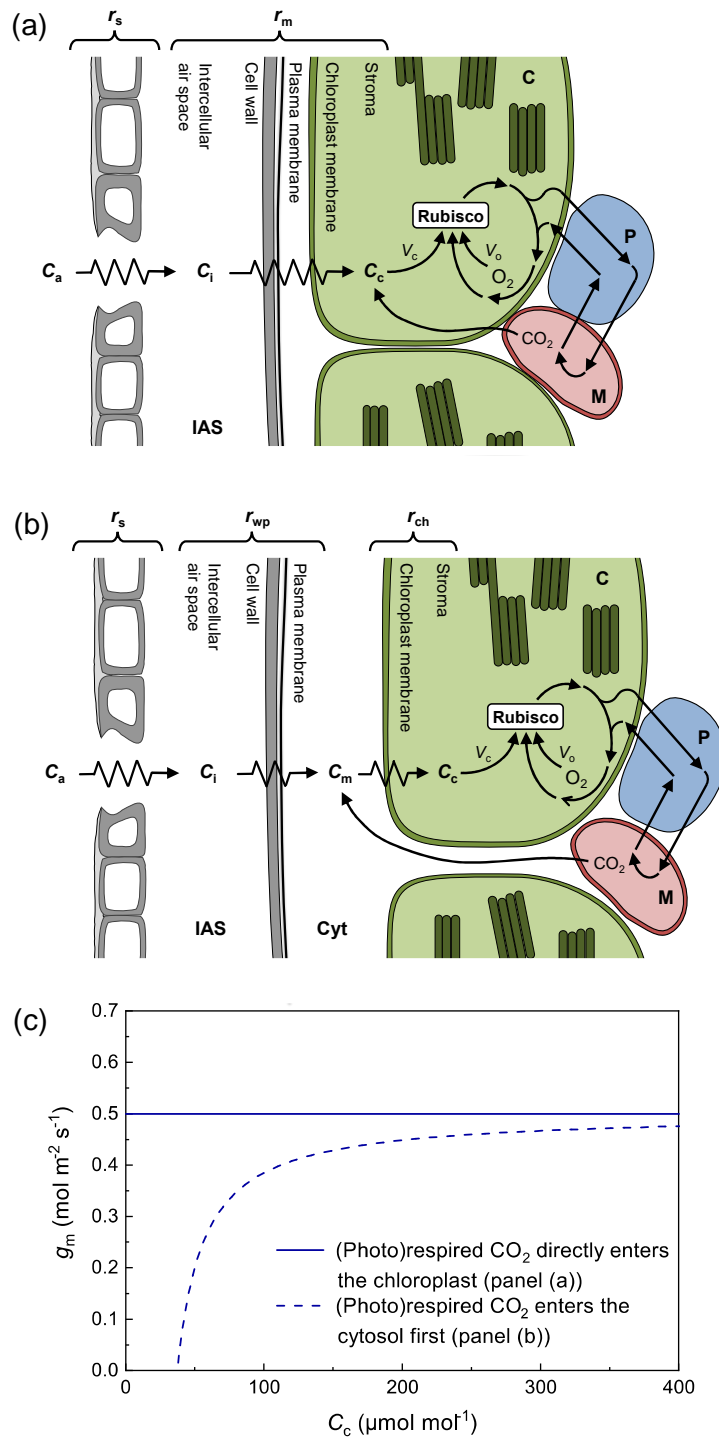




1443

1444 **Figure 1**

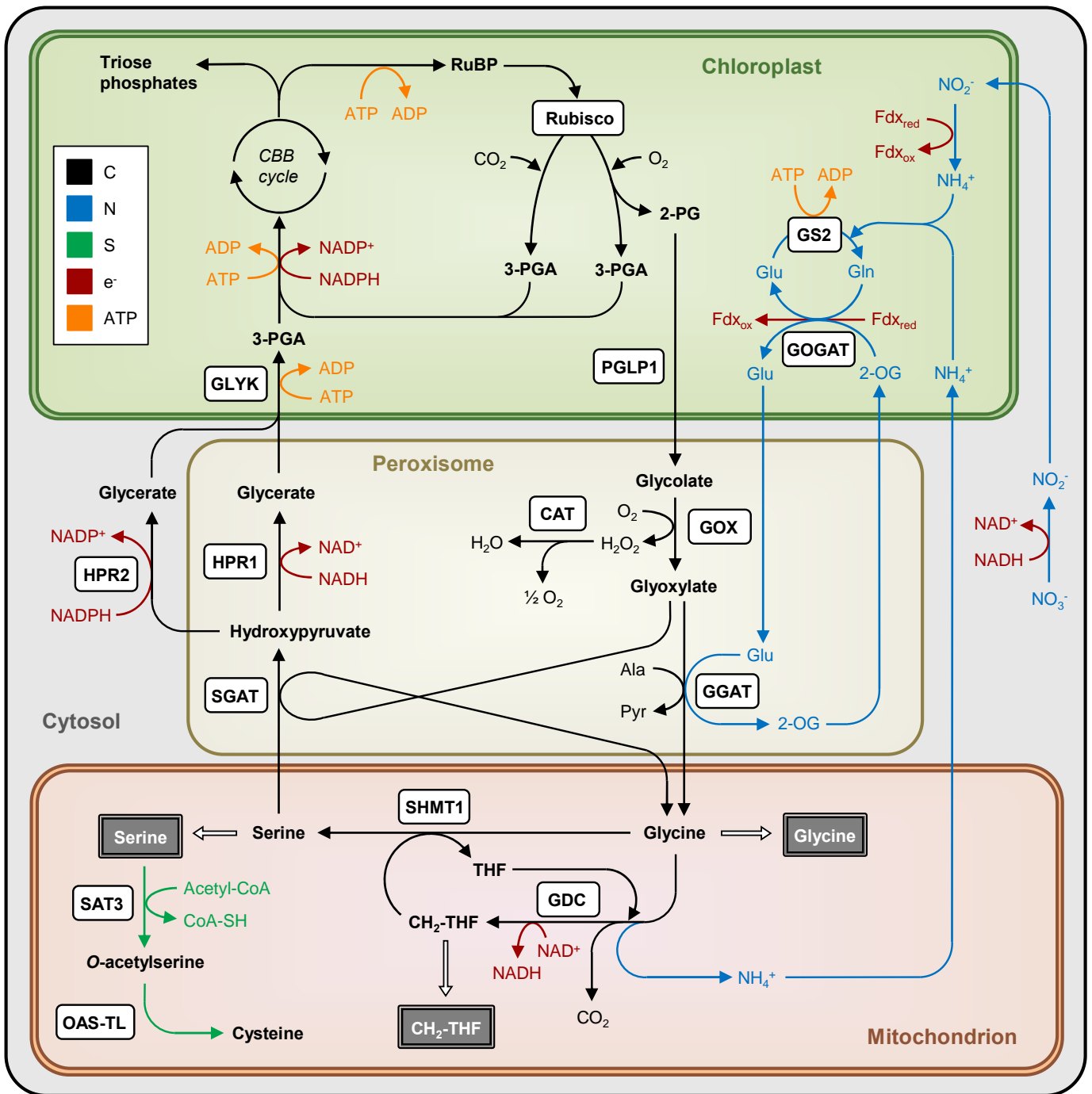
1445 **Activation and catalysis sequence of Rubisco.** Inactive Rubisco enzyme (E) becomes  
 1446 carbamated by the binding of non-substrate CO<sub>2</sub> (EC) followed by the activation through  
 1447 the binding of Mg<sup>2+</sup> (ECM). The activated Rubisco binds ribulose-1,5-bisphosphate  
 1448 (RuBP) forming the ECMR complex, which can then react either with CO<sub>2</sub> in a  
 1449 carboxylation reaction producing two molecules of 3-phosphoglycerate (3-PGA), or with  
 1450 O<sub>2</sub> in an oxygenation reaction producing one molecule of 3-PGA and one of 2-  
 1451 phosphoglycolate (2-PG). Figure adapted from Mate et al. (1996).



1452 **Figure 2**

1453 **Resistances to CO<sub>2</sub> diffusion inside the leaf affecting the CO<sub>2</sub> concentration around**  
 1454 **Rubisco.** The photorespiratory pathway involves the subcellular compartments  
 1455 chloroplasts (C), peroxisomes (P) and mitochondria (M). (a) (Photo)respired CO<sub>2</sub> is

1456 effectively released inside the chloroplast. This is the case if chloroplasts fully cover the  
1457 cell periphery and the mitochondria are located towards the inside of the cell. It is also  
1458 achieved in plants bioengineered to release photorespiratory CO<sub>2</sub> directly inside the  
1459 chloroplast. The resistances imposed by stomata ( $r_s$ ) and the mesophyll ( $r_m$ ), which  
1460 includes the cell wall and membranes, cause a progressive decline in CO<sub>2</sub> concentrations  
1461 from the ambient air ( $C_a$ ) to the intercellular air space (IAS;  $C_i$ ) and the chloroplast ( $C_c$ ).  
1462 (b) If (photo)respired CO<sub>2</sub> mixes with the CO<sub>2</sub> coming from the IAS inside the cytosol,  
1463 which is the case e.g. when chloroplasts are further apart, or mitochondria are located  
1464 between the cell wall and the chloroplasts, then  $r_m$  becomes an apparent resistance that  
1465 varies with the amount of CO<sub>2</sub> released from (photo)respiration (see text for further  
1466 information). In this case,  $r_m$  has to be separated into a cell wall and plasmalemma  
1467 component ( $r_{wp}$ ) and a chloroplast envelope and stromal component ( $r_{ch}$ ). (c) The CO<sub>2</sub>  
1468 response of the mesophyll conductance ( $g_m$ ) to CO<sub>2</sub> diffusion for the two scenarios  
1469 outlined in panels (a) and (b), modelled for equal contribution of  $r_{ch}$  and  $r_{wp}$ . Note that  
1470 when (photo)respiratory CO<sub>2</sub> enters the cytosol first, the apparent value of  $g_m$  tends to  
1471 zero when the compensation point ( $A = 0$ ) is approached. Panels (a) and (b) adapted  
1472 from von Caemmerer (2013)  
1473



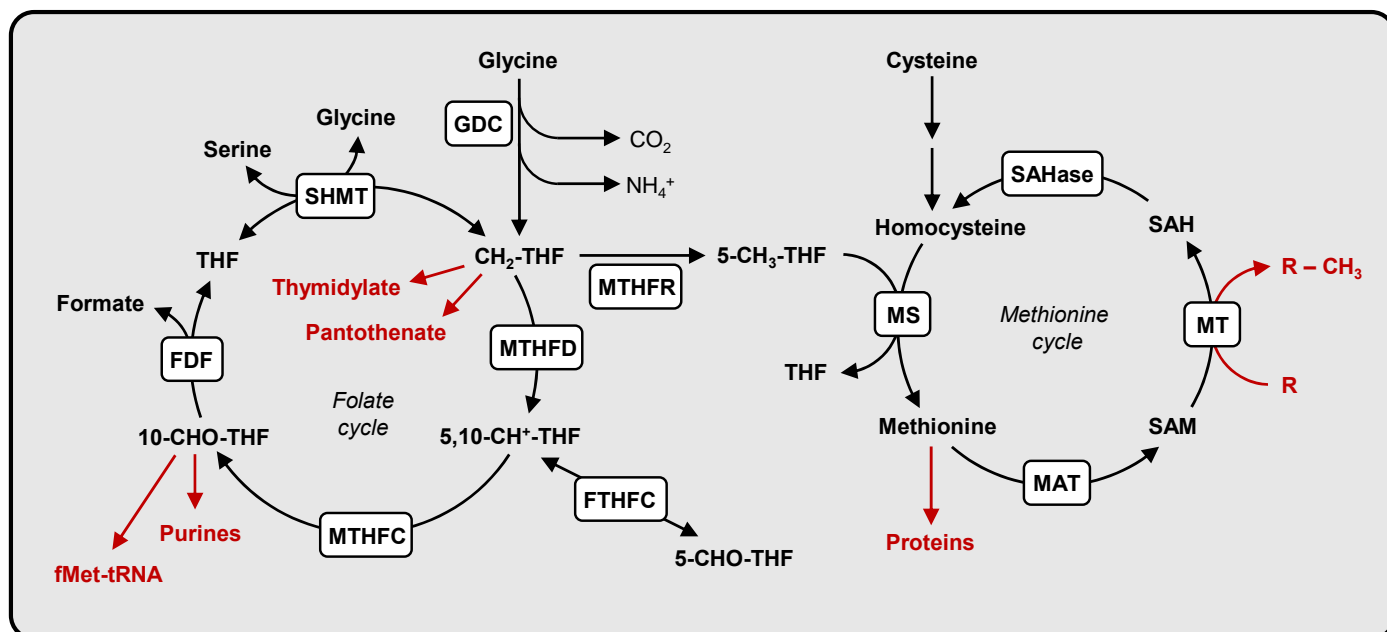
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1475 **Figure 3**

1476 **Schematic of the photorespiratory metabolism.** The movement of carbon along the  
 1477 photorespiratory pathway (black arrows) and the metabolites involved (bold font) are  
 1478 shown in the context of the nitrogen (blue) and sulfur metabolism (green). Redox

1479 reactions involving NAD, NADP, or ferredoxin (Fdx) are shown in red and ATP consuming  
1480 processes in orange. Photorespiratory carbon leaves the pathway as CO<sub>2</sub> during the  
1481 glycine decarboxylation step or is returned to the CBB cycle as 3-PGA. Carbon may also  
1482 be exported from the photorespiratory pathway (white arrows) in the form of the amino  
1483 acids glycine and serine, or as CH<sub>2</sub>-THF, which supplies one-carbon (C<sub>1</sub>) units to the C<sub>1</sub>  
1484 metabolism (grey boxes). The amount of exported carbon influences how much  
1485 photorespiratory carbon is returned to the CBB cycle. 2-PG, 2-phosphoglycolate; 3-PGA,  
1486 3-phosphoglycerate; 2-OG, 2-oxoglutarate; PGLP1, phosphoglycolate phosphatase 1;  
1487 GOX, glycolate oxidase; CAT, catalase; GGAT, glutamate:glyoxylate aminotransferase;  
1488 SGAT, serine:glyoxylate aminotransferase; GDC, glycine decarboxylase complex; SHMT1,  
1489 serine hydroxymethyltransferase 1; HPR1, hydroxypyruvate reductase 1; HPR2,  
1490 hydroxypyruvate reductase 2; GLYK, glycerate kinase; GS2, glutamine synthetase;  
1491 GOGAT, glutamine:oxoglutarate aminotransferase; SAT3, serine *O*-acetyltransferase;  
1492 OAS-TL, *O*-acetylserine (thiol) lyase (Figure adapted from Bauwe et al., 2010; Eisenhut et  
1493 al., 2019).

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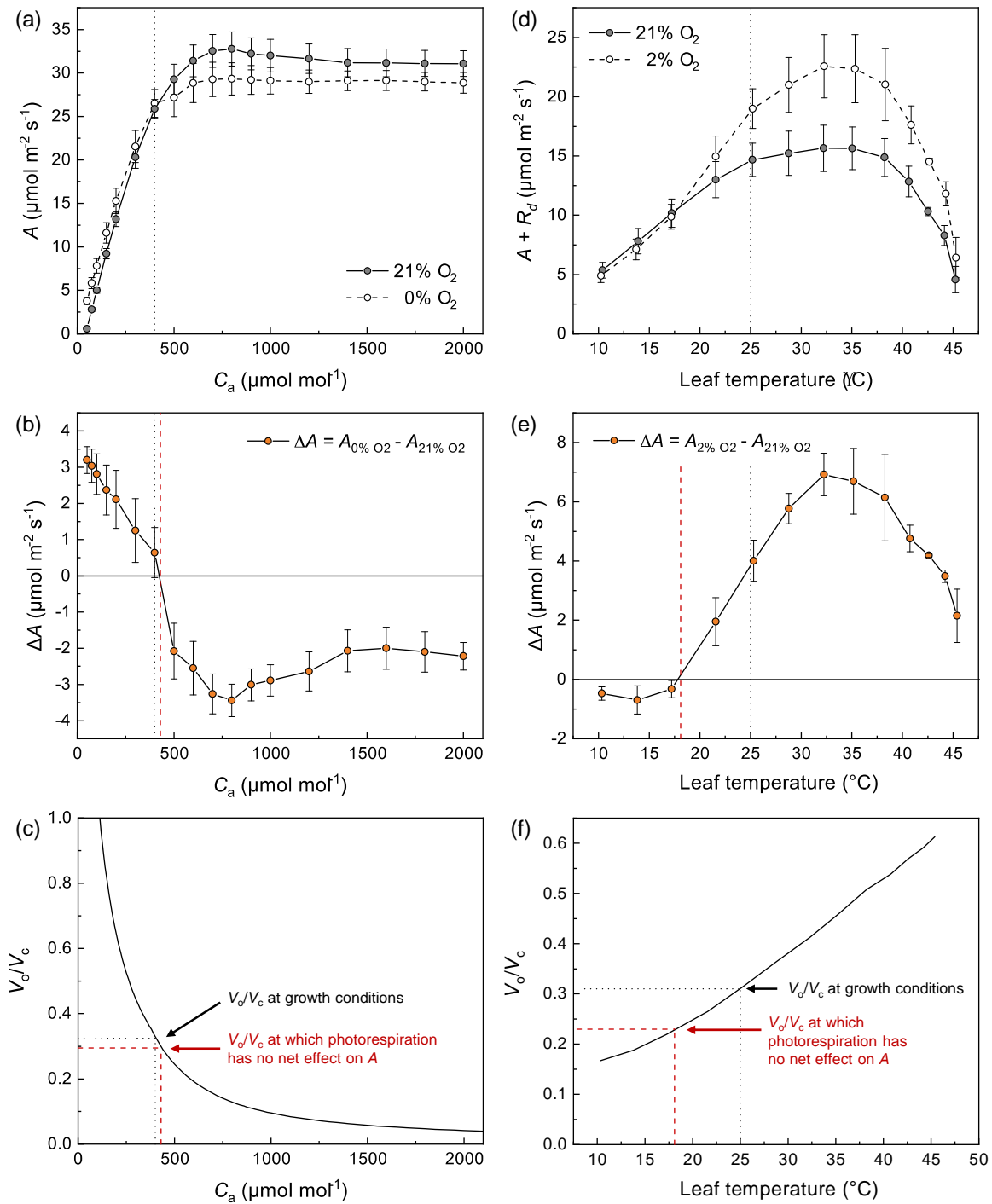


1495

1496 **Figure 4**

1497 **Metabolic uses of C<sub>1</sub> units outside the photorespiratory pathway.** CH<sub>2</sub>-THF exported  
 1498 from the photorespiratory pathway is the precursor for several derivatives of THF that  
 1499 can be interconverted and their C<sub>1</sub> units used for the biosynthesis of a wide range of  
 1500 primary and secondary metabolites (shown in red). A major sink for C<sub>1</sub> units is the  
 1501 methylation of various substrates, such as DNA, RNA, proteins, phospholipids and other  
 1502 substrates. The shown reactions occur in mitochondria, chloroplasts, and the cytosol,  
 1503 but may have a preference for a certain subcellular compartment. THF,  
 1504 tetrahydrofolate; CH<sub>2</sub>-THF, 5,10-methylene-THF; 5,10-CH<sup>+</sup>-THF, 5,10-methenyl-THF; 5-  
 1505 CHO-THF, 5-formyl-THF; 10-CHO-THF, 10-formyl-THF; 5-CH<sub>3</sub>-THF, 5-methyl-THF; SAM, S-  
 1506 adenosyl-methionine; SAH, S-adenosyl-homocysteine; fMet-tRNA, N-formylmethionine-  
 1507 tRNA; GDC, glycine decarboxylase complex; SHMT, serine:hydroxymethyltransferase;

1508 FDF, 10-formyl-THF deformylase; MTHFD, 5,10-methylene-THF dehydrogenase; MTHFC,  
1509 5,10-methenyl-THF cyclohydrolase; FTHFC, 5-formyl-THF cyclo-ligase; MTHFR, 5,10-  
1510 methylene-THF reductase; MS, methionine synthase; MAT, methionine  
1511 adenosyltransferase; MT, methyltransferase; SAHase, S-adenosyl-homocysteine  
1512 hydrolase (Figure adapted from Gorelova et al., 2017).  
1513



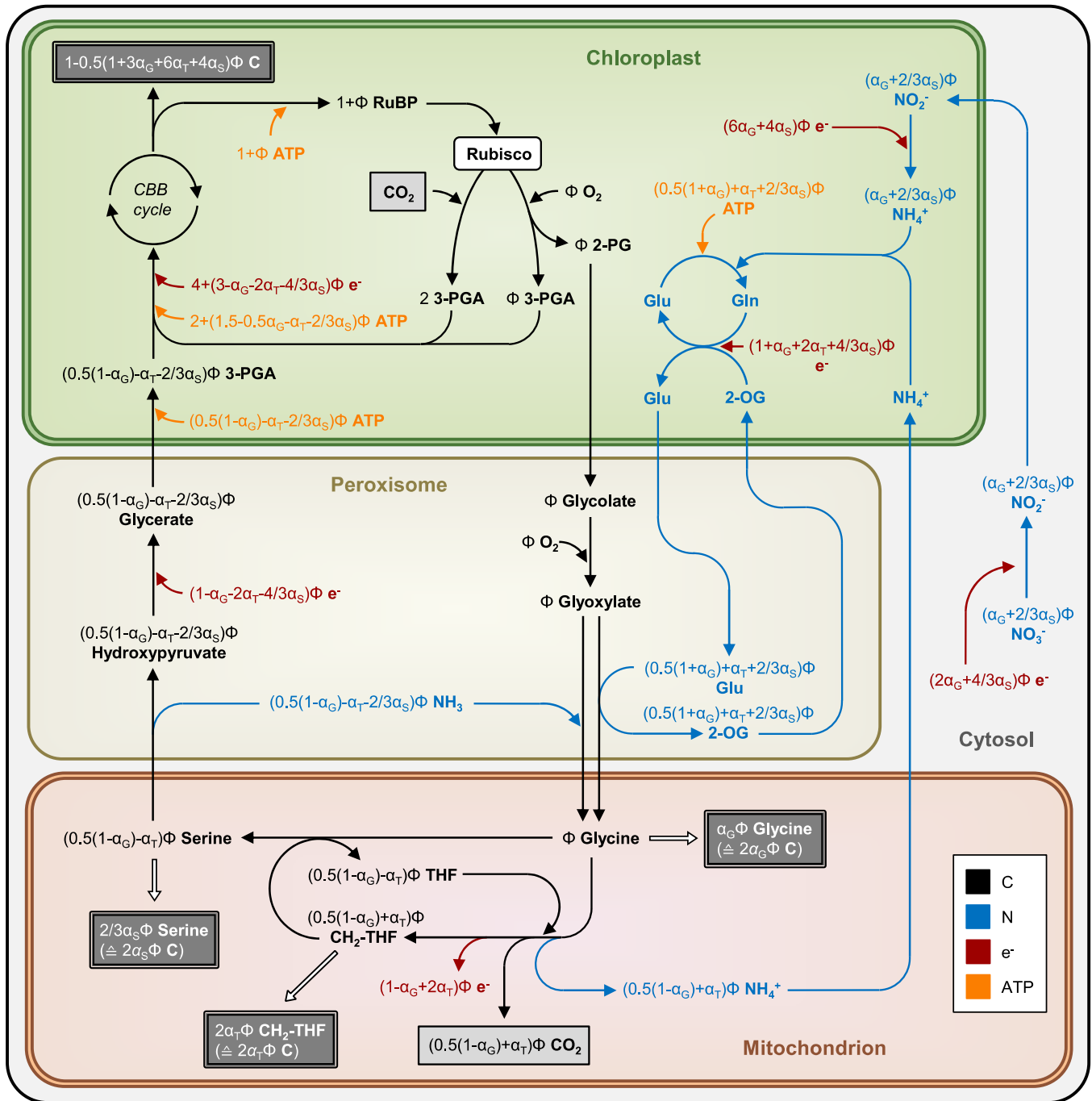
1514 **Figure 5**

1515 **Net effect of photorespiration on CO<sub>2</sub> assimilation rates.** Photorespiration can  
 1516 transiently be suppressed by decreasing  $O_2$  around the leaf.  $A$  measured during the  
 1517 instantaneous suppression of the Rubisco oxygenation reaction integrates the



1518 diffusional, biochemical, and physiological effects of photorespiration on net carbon  
1519 uptake of the leaf (calculated as  $DA$ ; see Box 1). (a)-(c) The suppression of  
1520 photorespiration under varying atmospheric  $CO_2$  concentrations ( $C_a$ ) for sunflower  
1521 (*Helianthus annuus*) leaves fertilized with  $NO_3^-$  (data from Busch et al., 2018). (d)-(f) The  
1522 suppression of photorespiration under varying leaf temperatures in leaves of sweet  
1523 potato (*Ipomoea batatas*; data from Busch and Sage, 2017). Net  $CO_2$  assimilation rates  
1524 are measured at oxygen concentrations of 21% and close to 0% (a) or 21% and 2% (b).  
1525 The absolute difference between  $A$  when photorespiration is present (21%  $O$ ) and  $A$   
1526 when it is absent (0% or 2%  $O$ ) is displayed in (b) and (e). Note that  $DA$  in (b) is  
1527 estimated from the  $A/C_a$  curves rather than  $A/C_i$  or  $A/C_c$  curves, which ensures all  
1528 diffusion resistances, and thus also the effect of the leaf anatomy, is accounted for (see  
1529 Box 1 for details). The dotted lines denote ambient conditions ( $CO_2$  concentration: 400  
1530  $\mu\text{mol mol}^{-1}$ ; growth temperature: 25°C). Red dashed lines denote the  $CO_2$  concentration  
1531 and temperature at which the net effect of photorespiration on carbon uptake is zero. In  
1532 these examples, at values of  $C_a$  above ambient, or leaf temperatures below  
1533 approximately 20°C, the net effect of photorespiration on carbon uptake is positive,  
1534 meaning that decreasing photorespiration decreases  $A$ . At lower  $C_a$ , or higher  
1535 temperatures, the overall effect of photorespiration is negative and plants would  
1536 benefit from decreasing photorespiration. (c) and (f) The ratio  $V_o/V_c$  when  
1537 photorespiration is present. At ambient  $C_a$ , approximately a quarter of Rubisco activity  
1538 goes towards oxygenation (c), corresponding to a negligible net effect on carbon uptake  
1539 (b). Similarly, at 20°C one fifth of Rubisco activity supports oxygenation (f), again with no

1540 associated costs in net carbon uptake (e). Comparing (b) with (c), and (e) with (f),  
1541 photorespiration appears to be beneficial in terms of net carbon uptake for a  $V_o/V_c$  of up  
1542 to roughly 0.25.  
1543



1544

1545 **Supplementary Figure 1**

1546 **Fluxes through the photosynthetic carbon reduction and photorespiratory pathways.**

1547 Outlined are the pathways indicating metabolites (bold font) and stoichiometries

1548 (regular font) of carbon (black) nitrogen (blue) when a proportion of the

1549 photorespiratory carbon leaves the photorespiratory pathway as amino acids and CH<sub>2</sub>-  
1550 THF. Electron and ATP requirements are indicated in red and orange, respectively. The  
1551 difference between CO<sub>2</sub> taken up by carboxylation and CO<sub>2</sub> released from  
1552 photorespiration (light gray boxes) equals the sum of the individual sinks for assimilated  
1553 carbon, indicated by double-bordered gray boxes. The proportion of 2-PG carbon  
1554 leaving the photorespiratory pathway as glycine is denoted by  $a_G$ , that leaving as CH<sub>2</sub>-  
1555 THF by  $a_T$  and that leaving as serine by  $a_S$ .  $F$  represents the ratio of the rates of RuBP  
1556 oxygenation to that of carboxylation; 3-PGA, 3-phosphoglycerate; 2-PG, 2-  
1557 phosphoglycolate. All flux magnitudes are scaled in relation to the rate of RuBP  
1558 carboxylation. Figure adapted from Busch et al. (2018).

1559