

# **PHYSICAL CHEMISTRY 2014**

12<sup>th</sup> International Conference on Fundamental and Applied Aspects of Physical Chemistry

The Conference is dedicated to the 25. Anniversary of the Society of Physical Chemists of Serbia

September 22-26, 2014 Belgrade, Serbia

# ISBN 978-86-82475-30-9

Title: PHYSICAL CHEMISTRY 2014 (Proceedings)
Editors: Ž. Čupić and S. Anić
Published by: Society of Physical Chemists of Serbia, Studenski trg 12-16, 11158, Belgrade, Serbia
Publisher: Society of Physical Chemists of Serbia
For Publisher: S. Anić, President of Society of Physical Chemists of Serbia
Printed by: "Jovan" Priting and Publishing Company; 200 Copies;
Number of pages: 6+ 441; Format: B5; Printing finished in September 2014.

Text an Layout: "Jovan"

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# PHYSICAL CHEMISTRY 2014

12th International Conference on Fundamental and Applied Aspects of Physical Chemistry

Organized by The Society of Physical Chemists of Serbia

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# ROOM TEMPERATURE FERROMAGNETISM IN Cu<sup>2+</sup> DOPED TiO<sub>2</sub> NANOCRYSTALS

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# ABSTRACT

Hydrothermal synthesis of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals, using dispersion of titania nanotubes in the presence of  $Cu^{2+}$  ions as precursor, which showed room temperature ferromagnetism, is reported. Morphology of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals was characterized by transmission electron microscopy (TEM). An average size of polygonal nanocrystals mainly of square/rectangular shape was around 13 nm. The x-ray powder diffraction (XRPD) analysis of resultant powder confirmed the anatase crystal phase of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals. Room temperature ferromagnetic ordering with saturation magnetic moment of the order of 2 x  $10^{-2}$  µ<sub>B</sub> per Cu atom was observed.

# INTRODUCTION

Great practical motivation for exploring doped semiconductors leis in the possibility of combining their semiconductor characteristics with magnetic properties. Diluted magnetic impurities in semiconductors produces materials that show ferromagnetic ordering at room temperature and therefore are appealing for spintonics in which simultaneous control of charge currents and spin polarized currents will be possible. Progress in the field of diluted magnetic semiconductors (DMS) has revealed a numerous new physical phenomena, including spin coherence, a new ferromagnetism and spin-polarized photoluminescence [1,2]. Wide band metal oxide semiconductors (TiO<sub>2</sub>, ZnO) doped with Co, Cu and Ni [3-6] presents the class of diluted magnetic semiconductors that has been identified by theoretical considerations and proved as a good candidate for achieving room temperature ferromagnetism. In this paper we report on new synthetic procedure, structural and magnetic properties of  $Cu^{2+}$  doped TiO<sub>2</sub>

nanocrystals. Using as a precursor titania nanotubes instead of molecular precursors we eliminated driving force problem that arises from the increase in the activation energy for nanocrystal nucleation in the presence of the dopant ions and consequent exclusion of the  $Cu^{2+}$  ions during nanocrystal growth.

# EXPERIMENTAL

All chemicals were reagent-grade from Aldrich and used as received. Titania nanotubes were synthesized according to Kasuga et al. [7], using  $TiO_2$  powder as a precursor.  $Cu^{2+}$  doped  $TiO_2$  nanocrystals were synthesized using hydrothermal treatment (250 °C / 90 min) of a dispersion containing titania nanotubes (125 mg/50 ml) at pH=3 in the presence of  $3,11 \times 10^{-4}$  M  $Cu(NO_3)_2$ . In order to remove the excess of  $Cu^{2+}$  ions the dispersion of  $Cu^{2+}$ doped TiO<sub>2</sub> nanocrystals was dialyzed against acidified water (pH=3) at 4 °C for 3 days. Water was changed daily. Films for magnetic characterization were prepared by drop casting of dialyzed dispersions of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals onto pre-cleaned glass substrate. The films were annealed in air for 2 min at 150 °C after each drop. The shape and size of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals were characterized using JEOL 100 CX transmission electron microscope operating at 100 kV. The XRPD pattern was obtained on a Philips PW-1050 automated diffractometer. The percent of Cu<sup>2+</sup> ions in TiO<sub>2</sub> nanocrystals was determined using *inductively coupled plasma* (ICP) emission spectrometry. The field dependence of the magnetic moment was measured using a superconducting quantum interference device

magnetometer (SQUID). Hysteresis loop measurement has been performed up to 60 kOe.

#### **RESULTS AND DISCUSSION**

The XRPD analysis of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals revealed the existence of a homogeneous anatase crystal phase, Figure 1. Conventional TEM image of  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals, Figure 1, inset, confirmed relatively uniform size distribution of faceted nanocrystals with average size of 13 nm.

The magnetic response for film made of 0.34 at%  $Cu^{2+}$  doped TiO<sub>2</sub> nanocrystals as a function of magnetic field strength (*H*) was followed at room temperature. The



**Figure 1.** XRPD pattern of Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals; Inset: TEM image of Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals

magnetic field was applied parallel to the film surface. The diamagnetic contribution determined from the magnetic field dependence of magnetization at high field (for H > 10 kOe) was subtracted. The field dependent magnetization after diamagnetic correction is shown in Figure 2.

As can be seen from Figure 2, a weak room temperature ferromagnetic behaviour of Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals was observed, with coercive field of  $Hc \sim 200$  Oe and saturation magnetization  $(M_s)$ of the order of 2 x  $10^{-2} \mu_{\rm B}/{\rm Cu}$ . Previous study indicated that the unexpected magnetism in such system could be the result of the interaction between the intrinsic defects and the doped ion [8]. There is couple of different carriermediated



mechanisms that could provide an explanation for the influence of oxygen vacancies and structural defects on the ferromagnetic ordering in DMSs: carrier mediated interaction such as Ruderman–Kittel–Kasuya–Yosida (RKKY) exchange, Zener double exchange or super exchange [5,8]. But, ferromagnetism in oxides with high electric resistance and low carrier densities such as the Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals can be explained in terms of the noncarrier-mediated bound magnetic polaron (BMP) model [9]. When defects (oxygen vacancies) concentration increases they overlap many dopant ions to yield BMPs. This result in ferromagnetic coupling between Cu<sup>2+</sup> ions mediated through oxygen vacancies. Li et al. proposed that substitutional Cu impurity binds easily with an oxygen vacancy (V<sub>O</sub>) in TiO<sub>2</sub> to form the complex defect Cu<sub>Ti</sub>-V<sub>O</sub>, which relaxes to form the non-regular magnetic CuO<sub>4</sub> complex, in which the Cu ion is strongly magnetic [8]. This complex is similar to the magnetic one in other 3*d* transition metal-doped TiO<sub>2</sub> rutile films [10].

### **CONCLUSION**

The polygonal Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals (d ~ 13 nm) were synthesized applying hydrothermal treatment on dispersion of titania nanotubes in the presence of Cu<sup>2+</sup> ions at pH=3. XRPD study confirmed the anatase crystal structure of Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals and absence of any impurity phases (CuO, Cu<sub>2</sub>O) in the sample. The weak ferromagnetic ordering at

room temperature with almost closed loop ( $Hc \sim 200$  Oe) and saturation magnetic moment of the order of 2 x  $10^{-2} \mu_B/Cu$  atom were observed in film made of Cu<sup>2+</sup> doped TiO<sub>2</sub> nanocrystals. The reason for observed weak ferromagnetism could be found in the interaction between the oxygen vacancy and the substitutional Cu impurity.

## ACKNOWLEDGEMENT

This work was supported by the Ministry of Education, Science and Technological Development of the Republic of Serbia (project: 172056)

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